SHORT RESEARCH AND DISCUSSION ARTICLE

# An effective treatment method for shale gas drilling cuttings solidified body

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#### Abstract



The exploration and production of shale gas technology provides a way for utilization of clean fuels. However, during the exploration process of shale gas, enormous amount of drilling cutting was generated and had to be solidified and landfilled. So the accumulation of shale gas drilling cutting solidified body (SGDS)causes severe land resource misuse and environmental complications. This study focuses on the utilization of SGDS as a raw material for the production of cement clinker, and the phase composition, microstructure, and environmental performance of the cement clinker was investigated by X-ray powder diffraction (XRD), scanning electronic microscopy (SEM), energy-dispersive X-ray spectrum analysis (EDX), and soaking test, respectively. The results show that the cement clinker obtained mainly constitutes of typical Portland cement mineral  $(C_3S, C_2S, C_3A,$  and  $C_4$ AF). The leaching test indicated that the concentration of heavy metal ions in leachate is within the limits allowed by the state "Technical specification for co-processing of solid wastes in cement kiln" (GB 30760-2014). This study therefore provides a benchmark on environmental effects resulting from drilling cuttings and utilization of resources.

Keywords Shale gas  $\cdot$  Solidified body  $\cdot$  Treatment  $\cdot$  Utilization

# Introduction

With the rapid development of China's economy, demand for energy is constantly on the rise. This has made shale gas leading as the new energy source and predominantly the main motivation in China's development economically. The straight hole and horizontal well technology for shale gas exploration is commonly used, and it is inevitable that enormous amount of drilling cuttings mixed with drilling fluid generated. Hardening agents, such as Portland cement and fly ash are added to act as a binder for purposes of stabilizing and solidifying it to reduce the contamination caused by the drilling cuttings within a shorter period

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(Leonard and Stegemann [2010;](#page-4-0) Kogbara et al. [2016](#page-4-0); Kogbara [2014;](#page-4-0) Chao-qiang et al. [2017\)](#page-4-0). The Shale gas drilling cuttings solidified body (SGDS) was generated and had to been landfilled. Since the SGDS are situated underground, the solid blocks undergo a serious of physical-chemical reaction, resulting in secondary-contamination, especially in heavy metal release and organic pollution (Antemir et al. [2010](#page-4-0); Leonard and Stegemann [2010\)](#page-4-0). And thus the need for safe disposal and recycling of SGDS is in urgent.

Previous studies of the physicochemical properties of drilling cuttings have indicated that SGDS can be used in place of sand and mineral addition for the production of building materials, such as the preparation of non-fired brick (Acchar and Marques [2016;](#page-4-0) Liu et al. [2018](#page-4-0); Dmitriy V. Oreshkin et al. [2015](#page-4-0)), concrete block (Ablieieva and Leonid [2016;](#page-4-0) Mahmoud Kassem et al. [2018\)](#page-4-0), road material (Tuncan et al. [2000;](#page-4-0) Chang-Seon Shon et al. [2016\)](#page-4-0), concrete (N.M. Wasiuddin et al. [2002;](#page-4-0) Ehsan et al. [2015\)](#page-4-0), cement (Wang et al. [2017;](#page-4-0) Bernardo et al. [2007](#page-4-0); Al-Otoom [2006\)](#page-4-0), sintered brick (Xiang-Guo Li et al. [2011](#page-4-0)), and other materials, such as lightweight ceramsite and cementitious material (Bamdad Ayati et al. [2019](#page-4-0); M. Aboutabikh et al. [2016\)](#page-4-0). Although, the research on resource utilization of drilling cuttings has a lot of aspects, but few of them could be put into practice. So, based

Hφ	Tl	As		Cd									Co	V	Zn	Mo
	2.07	6.91	24.03	0.505					119	1.10	ND.		5.19	46.7	131	6.8
0.117	ND.	4.37		1.19	ND.	ND.				ND.	ND.	0.56	5.8	8.9	236	41.2
0.121	ND.			0.196	10.2	ND.	11.9		198	0.17	ND.	1.03	3.9	21	47.1	ND.
0.319	ND.	183	30.9	1.14	154	ND.	103	87.4	796	ND.	ND.	11.39	41	232	253	16.8
0.108	ND.	7.98	ND.	2.92	132	0.27	116	90.3	209	ND.	ND.	2.58	38.7	149	179	6.76
		0.713		P <sub>b</sub> 21.1 2.45 1.67			46.7 ND	$Cr$ $Cr^{6+}$ $Cu$	Ni 80.7 6.34 9.1	36.9 16.7 140	Mn Be	Sn	Sb <sub>1</sub> 1.36			

**Table 1** Lixivium heavy metal pollutants of raw materials  $(mg/kg)$ 

ND means the item was undetected

on the results of above research, and according to the actual situations, the aim of this study is to evaluate the potential use of SGDS as a raw material for the production of Portland cement clinker on an industry scale, so as to make a resource utilization of SGDS and eliminate the potentially secondarycontamination caused by the landfill of SGDS. This study provides a way for the security and environmental disposal of SGDS.

# Materials and methods

## Raw materials

SGDS was obtained from a drilling plant in Chongqing. Slag and fly-ash were procured from a local coal-fired power plant. Cement raw meal used was obtained from a cement plant in Chongqing. The raw materials' lixivium heavy metal pollutants were presented in Table 1.

## Field experimentation and testing methods

### Field experimentation

Figure 1 shows the process flow of the production of the Portland cement clinker using SGDS.

In reference to the preliminary experiments' results, for preparing the cement raw meal, 83.6% limestone, 10.1% sandstone, 2.5% SGDS, 2.8% coal ash, and 1.0% sulfuric acid residue were used (in mass). All the raw materials were crushed, mixed, and ground to the required fineness, typically 20% of retained on an 80 micro sieve. The homogenized raw



Fig. 1 Process flow sheet of the solidified body cement production



<span id="page-2-0"></span>Table 2 Limiting value of SGDS cement clinker heavy metal

meal is introduced into the top of the cyclone preheater and it was heated and decarbonated. After decomposition of the raw meal, it was fed into a rotary kiln in which the raw meal is heated to 1500 °C. After firing, the clinker exits the kiln and is cooled from 1200 °C to 60 °C in a cooler. And then the clinker is ground together with slag, cinder, and desulfurization gypsum in a mill to produce Portland cement.

#### Test methods

The clear crystalline minerals of the clinker were identified by an X-rd (X-ray powder diffraction), conducted in a Panalytical (X'Pert PRO diffractometer) with a speed current of 60 mA and the voltage of 35 kV. The acquired spectrum was scrutinized in X'Pert High Score and software Plus MDI Jade 5.0.

The use of scanning electronic microscopy (SEM) and energy-dispersive X-ray spectrum analysis (EDX), the morphologies and elements of clinker were analyzed by scanning photo-electron microscope (ASTEREO SCAN440, Leica Cambridge Ltd).

According to the requirement standards for the production of lixivium which detects its admissible indexes and refers to Chinese national standard "Technical specification for coprocessing of solid wastes in cement kiln" (GB 30760-2014) to determine the environmental performance of the product. The heavy metal content in leachates were analyzed by ICP-MS methods,

# Results and discussion

### Heavy metal analysis

According to the requirements of national quality, "Environmental protection technical specification for coprocessing of solid wastes in cement kiln" (HJ 662-2013), the limiting value of heavy metal kiln feeding has specific requirements. Therefore, according to the article, the heavy metal content contained in solidified body cement clinkers is presented in Table 2. It is indicated that the heavy metal content contained in the clinker produced from SGDS is within the limits set by national standard.

### XRF analysis

It is clear from the Table 3 that the chemical composition of the SGDS is similar with the composition of commercially available cement. However, the Cl,  $SO_3$ , and  $K_2O$  contained in SGDS were lower than commercially available cement, making the SGDS an idea raw material for the production of the Portland cement clinker.

### Microstructure analysis

As shown in Fig. [2,](#page-3-0) that the mineral phases of the cement clinker produced from SGDS mainly consist of  $C_3S$ ,  $C_2S$ ,

Table 3 Chemical composition



<span id="page-3-0"></span>

Fig. 2 Mineral phases of the cement clinker produced from SGDS

 $C_3A$ , and  $C_4AF$ , and the diffraction peak intensity and position were brought into correspondence with typical commercially available cement. In addition, Fig. 2 indicated that the major crystalline state of the cement clinker produced from SGDS is arranged in a lamellar fashion  $C_3S$  (spot B and spot C) and near-spherical particles  $C_2S$  (spot A) around the mineral phases. This may owe the addition of SGDS which can promote the formation of liquid phase in the raw meal. The early appearance of liquid phase reduces the burning temperature of cement clinker prepared with SGDS. The burnability was improved of the raw meal which contains SGDS, and is help for the synthesis of Belite. In addition, from Table [3,](#page-2-0) we can conclude that the calcium to silicon ratio of  $C_2S$  and  $C_3S$ unblended raw meal are 1.731 and 1.829, respectively; the cement clinkers produced from SGDS are 1.863 and 1.817, respectively, which is in accordance with the proposed theoretical analysis. Another cause could be the presence of alkaline liquid phase and valence bond strong  $AIO_4^5$  and  $FeO_4^5$ 

Table 4 Lixivium content of heavy metal pollutants Unit: mg/kg

Item	Cu —	Zn Cd Ni As Cr Pb			- Mn
Standard value 100 500 1.5 100 40 150 100 600					
Detection value 87.9 279.4 0.51 31.9 6.38 47.6 54.7 316.9					

tetrahedron which are not conducive to the formation and growth of  $C_3S$  crystals, so that the  $C_3S$  mineral of clinker is close to  $C_2S$ .

#### Environmental safety analysis

The results of leaching property of the cement clinker produced from SGDS are presented in the Table 4.

The dates in Table 4 clearly indicate all indexes within the required limit for the China standard "Technical specification for co-processing of solid wastes in cement kiln" (GB 30760- 2014) (Liu et al. [2018;](#page-4-0) Shu et al. [2016\)](#page-4-0). In conclusion, drilling cuttings solidified body-based cement clinker cannot pollute the environment.

# Conclusions

This paper presents a case study of resource utilization of the SGDS in industrial scale. And it is indicated that the SGDS could be used as a raw material for the production of Portland cement clinker in a cement production line with a daily output of 10,000 tons using the ingredient ratio: 83.6% limestone, 10.1% sandstone, 2.5% solidified body, 2.8% coal ash, and 1.0% sulfuric-acid residue, resulting a daily consumption of

<span id="page-4-0"></span>about 250 tons of SGDS. The cement clinker obtained mainly constitutes of typical Portland cement minerals  $(C_3S, C_2S,$  $C_3A$ , and  $C_4AF$ ) and the leaching tests showed that the concentration of heavy metal ions in leachate of the cement clinker is within the limits allowed by the state national standard. Our strategy has provided a convenient reference for the controlling of the pollution induced by the SGDS and the resource utilization of it in a practical way.

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