

Accumulation, transformation, and release of inorganic arsenic by the freshwater cyanobacterium *Microcystis aeruginosa*

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Received: 10 January 2013 / Accepted: 15 April 2013 / Published online: 1 May 2013
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Abstract Arsenic (As) as a major hazardous metalloid was affected by phytoplankton in many aquatic environments. The toxic dominant algae *Microcystis aeruginosa* was exposed to different concentrations of inorganic arsenic (arsenate or arsenite) for 15 days in BG11 culture media. Arsenic accumulation, toxicity, and speciation in *M. aeruginosa* as well as the changes of As species in media were examined. *M. aeruginosa* has a general well tolerance to arsenate and a definite sensitivity to arsenite. Additionally, arsenate actively elevated As methylation by the algae but arsenite definitely inhibited it. Interestingly, the uptake of arsenite was more pronounced than that of arsenate, and it was correlated to the toxicity. Arsenate was the predominant species in both cells and their growth media after 15 days of exposure to arsenate or arsenite. However, the amount of the methylated As species in cells was limited and insignificantly affected by the external As concentrations. Upon uptake of the inorganic arsenic, significant quantities of arsenate as well as small amounts of arsenite, DMA, and MMA were produced by the algae and, in turn, released back into the growth media. Bio-oxidation was the first and primary process and methylation was the minor process for arsenite exposures, while bioreduction and the subsequent methylation were the primary metabolisms for arsenate exposures. Arsenic bioaccumulation and transformation by *M. aeruginosa* in aquatic environment should be paid more attention during a period of eutrophication.

Keywords Arsenic · Biotransformation · Speciation · Release rate · *M. aeruginosa*

Introduction

Arsenic (As) is not only a potential hazardous metalloid but also a natural environmental contaminant to which humans are routinely exposed in food, water, air, and soil (Hughes et al. 2011). In the world, more than 150 million people are exposed to drinking water containing “excessive” arsenic above the World Health Organization’s recommended limit of $10 \mu\text{g L}^{-1}$ (Nordstrom 2002; Brammer and Ravenscroft 2009). The range of As concentrations found in natural water systems is large, ranging from less than $0.5 \mu\text{g L}^{-1}$ to more than $5,000 \mu\text{g L}^{-1}$ (Smedley and Kinniburgh 2002). Arsenic concentrations can be up to 20mg L^{-1} in some polluted freshwaters (Rahman and Hasegawa 2012). Water contamination with As is a universal and urgent problem that requires immediate action to improve water quality (Rahman 2002).

Toxicity and bioaccumulation of As in organisms are dependent not only on its total amount but also on its speciation (Karadjova et al. 2008; Duncan et al. 2010). Although As can be found in the environment in several oxidation states, the trivalent As^{III} and pentavalent As^{V} oxyanions are more prevalent than the organoarsenicals in freshwater (Smedley and Kinniburgh 2002; Sharma and Sohn 2009). It is commonly accepted that, with the exception of the trivalent-methylated metabolites, such as monomethylarsonic acid (MMA^{III}) and dimethylarsinic acid (DMA^{III}), and thiolated pentavalent methylated species, such as dimethylthioarsinic acid (DMTA^{V}), the inorganic forms of As are more toxic in mammals than organoarsenicals (Naranmandura and Suzuki 2008; Dopp et al. 2010; Sun et al. 2012; Alava et al. 2012). In aquatic systems, the dominant inorganic As is incorporated into microorganisms

Responsible editor: Céline Guéguen

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such as phytoplankton and is, in turn, converted to methylarsenicals and/or high order organoarsenicals such as arsenosugars (AsS) (Francesconi et al. 1992). Algae plays an important role in As bioaccumulation and biotransformation in the aquatic environment because they show very high As bioaccumulation from surrounding water and thereby determine the amount of As available to higher organisms and their subsequent transformation. Recent studies showed that different algal species have vastly different uptake capacities for inorganic As (Caumette et al. 2011; Duncan et al. 2010; Markley and Herbert 2010).

Cyanobacteria are one of the largest and most important groups of prokaryotic autotrophs with oxygenic photosynthesis in aquatic system. As the most common toxic cyanobacterium in eutrophic freshwater, *Microcystis aeruginosa* can form harmful algal blooms causing animal poisoning and present risks to human health (Oberholster et al. 2004). Arsenic uptake differs within algal growth phases due to the dominance of different As processes in different stages (e.g., production and excretion of As^{III} during log growth phase) (Hellweger et al. 2003; Hellweger and Lall 2004; Hellweger 2005). To date, most of the studies on As uptake, accumulation, and biotransformation by phytoplankton have focused on marine organisms, especially the eukaryotic alga, for the detailed description of As^V detoxification and its metabolism with phosphorus (Maeda et al. 1992a, b; Pawlik-Skowronska et al. 2004; Levy et al. 2005; Qin et al. 2009; Yin et al. 2011b). Little is known about the comparative biotransformation of different inorganic As forms in toxic bloom alga in freshwater, particularly in the speciation transformation and release processes during cyanobacterial blooms. It is requisite to understand these relative aspects due to their potential environmental risks to other organisms and humans if the water is used as drinking water.

Therefore, the main objectives of the present study were to investigate As biotransformation and bioaccumulation in *M. aeruginosa* as well as their release to the aquatic media at various As-polluted concentrations. Growth studies were undertaken to determine the toxic effects of inorganic As on *M. aeruginosa*. Changes of As speciation in both alga and media could provide important information to understand the mechanisms of As biotransformation and to predict the possible risks of As-contaminated freshwater during cyanobacterial bloom.

Materials and methods

Organism and culture conditions

The tested bloom cyanobacterium (unicellular *M. aeruginosa*, FACHB 905) was obtained from the Institute of Hydrobiology, Chinese Academy of Sciences, which was isolated from the eutrophic water of Dianchi Lake (Wu et al. 2007). Batch

culture experiments of *M. aeruginosa* were conducted using BG11 growth media in 250-mL Erlenmeyer flasks with a 16:8 h light:dark cycle under a light intensity of 115 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$ at 25 °C. Test flasks were rotated and shaken three times daily by hand to ensure sufficient gas exchange. The alga culture experiment was run for 15 days.

All stock vessels and culture flasks were soaked in 10 % HNO₃ for at least 24 h and then rinsed with Milli-Q water, oven dried, autoclaved, and handled under sterile conditions prior to use. Analytical reagent-grade chemicals and Milli-Q water were used for incubate preparations. The standard stock solutions of arsenate, arsenite, MMA, and DMA were prepared by Na₃AsO₄·12H₂O (Fluka, p.a.), NaAsO₂ (Alfa Aesar), NaCH₄AsO₃ (Fluka), and NaC₂H₆AsO₂ (Sigma-Aldrich), respectively.

Growth rate inhibition bioassays

The 72-h growth rate inhibition bioassays were used to determine the toxicity of inorganic As, As^V as Na₃AsO₄·12H₂O (Fluka, p.a.), and As^{III} as NaAsO₂ (Alfa Aesar). Cells in exponential growth phase, with the initial cell density of 10⁶ cells mL⁻¹, were added separately into increasing concentrations of As^V (5 μM to 20 mM) and As^{III} (5 μM to 500 μM). Fresh algal samples that were transferred to a 10-mm quartz cell were used to measure the turbidity at wavelength of 682 nm using a Thermo ultraviolet visible light spectrophotometer. The number of cells per milliliter was followed every 24 h over a 72-h exposure, which was counted traditionally using a hemocytometer and a microscope. The specific growth rate (μ) of the cells was calculated with the following equation: $\mu = \ln(C_t / C_0) / t$, where μ is the cell-specific growth rate (per day), C_t and C_0 represent the cell density at time t and the initial of the experiment, respectively, and t is the duration of the arsenic exposure (Zeng et al. 2009). The specific growth rate inhibition (I) was calculated from the μ for the sample and μ_0 for the control with the equation: $I\% = (\mu_0 - \mu) / \mu_0 \times 100$.

Inorganic As treatment of *M. aeruginosa* cultures

The inoculum of cells from the stock cultures was designed to give an initial cell density of 10⁵ cells mL⁻¹. Each culture, just after inoculation with *M. aeruginosa*, was treated with arsenate or arsenite at six concentrations (0, 5, 10, 20, 50, and 100 μM) for a 15-day uptake period to examine As bioaccumulation, biotransformation, and release by living cells. Cultures without As were used as controls. Periodically, each flask was sampled eight times by removing 5-mL aliquots of culture solution beginning 24 h after inoculation and then every 2 days thereafter. Additionally, separate flasks with all treatments in BG11, but no algae, were also monitored to detect the stability of As^{III} and As^V at the onset and end of the experiments.

Each sample was analyzed for optical density, total As, and As speciation in the media. Chlorophyll a quantification was performed every 2 days in *M. aeruginosa* starting 8 days after the beginning of the experiment in order to attain sufficient cell density. For total As determination, chlorophyll a and speciation analysis of media samples, alga were sampled and centrifuged to create an algal pellet. Then, the algal pellet was used for analysis of chlorophyll a, while the supernatant (media) was used for analysis of total As and As speciation. For total As and As speciation analyses in the media, the supernatant was filtered through 0.45- μm cellulose acetate syringe filters. Samples for total As measurements were acidified to a final concentration of 1 % acid with guaranteed reagent 10 M nitric acid for storage. These samples were then analyzed for total As concentration using inductively coupled plasma mass spectrometry (ICP-MS, 7500a Agilent). Supernatant that was analyzed for As speciation was stored at $-20\text{ }^{\circ}\text{C}$ for no more than 10 days. Arsenic species were analyzed using anion exchange chromatography coupled to an inductively coupled plasma mass spectrometer as a detector. At the end of the experiment, two algal suspensions of 25 mL were centrifuged into a pellet. Both algal pellets were washed twice with sterile Milli-Q water ($18.2\text{ m}\Omega\text{ cm}^{-2}$; Millipore) and then immersed in an ice-cold phosphate buffer (1 mM K_2HPO_4 , 5 mM MES, and 0.5 mM $\text{Ca}(\text{NO}_3)_2$) for 15 min to completely remove apoplastic As. Eventually, one was dried in an oven at $60\text{ }^{\circ}\text{C}$ and the other was freeze-dried at a vacuum freeze dryer (Labconco, USA). Next, both of them were digested separately. Then, the oven-dried alga samples were used to analyze total As and the freeze-dried alga for As speciation (Zhang et al. 2009). To get better insights into the cellular concentrations of As speciation and possible biotransformation in alga, the cell concentrations of different As species (As^{III} , As^{V} , MMA, and DMA) were measured in the dissolved and intracellular fractions following a 15-day exposure.

Chlorophyll a measurements

Chlorophyll a was extracted from the algal pellet using 5 mL of cold, neat ethanol overnight at $4\text{ }^{\circ}\text{C}$ (Knudson et al. 1977). The extract was filtered using cellulose filters (0.45 μm) and then measured with a Thermo ultraviolet visible light spectrophotometer for light absorbance at 649 and 665 nm for chlorophyll a, using the previously empirical correction equation: chlorophyll a ($\mu\text{g L}^{-1}$) = $13.7A_{665\text{ nm}} - 5.76A_{649\text{ nm}}$ (Knudson et al. 1977).

Total As determination

For the estimation of the intracellular total As, the samples of alga were prepared and analyzed according to the method of Yin et al. (2011a). About 0.02 g oven-dried alga samples

were digested with 5 mL concentrated HNO_3 (65 %, guaranteed reagent, Fairfield, OH, USA) overnight and heated in a microwave-accelerated reaction system (CEM Microwave Technology Ltd., Matthews, NC, USA). The details were similar to the report by Xue et al. (2012).

The total As in the media and alga were measured by ICP-MS operating in the helium gas mode to remove possible interference of $^{40}\text{Ar}^{35}\text{Cl}$ on m/z 75. The masses monitored were ^{75}As with ^{115}In and ^{103}Rh as internal standards for signal stability.

HPLC–ICP-MS for As speciation analysis

Approximately 0.02-g freeze-dried alga samples were placed in 15-mL centrifuge tubes and treated in 5 mL of 1 % HNO_3 overnight. Then, they were digested in a microwave-accelerated reaction system (the working program was similar to the measurement of intracellular As). The digests were filtered through 0.45- μm cellulose acetate syringe filters. The alga extracts as well as the solutions were analyzed by HPLC–ICP-MS (Agilent LC1100 series and Agilent ICP-MS 7500a; Agilent Technologies) for As speciation with anion exchange columns as detailed by Zhu and Zhang et al. (Zhu et al. 2008; Zhang et al. 2012). Chromatographic columns were obtained from Hamilton and consisted of a precolumn (11.2 mm, 12–20 μm) and a PRP-X100 10- μm anion exchange column (250 \times 4.1 mm). As species (As^{III} , As^{V} , DMA, and MMA) were separated with a mobile phase of 10 mM $\text{NH}_4\text{H}_2\text{PO}_4$ and 10 mM NH_4NO_3 , adjusted to pH 6.2 with nitric acid or ammonia, and run isocratically at 1.0 mL min^{-1} . Germanium (Ge) as an internal standard was mixed continuously with the postcolumn solution through a peristaltic pump. Signals at m/z 75 (As) and 72 (Ge) were collected with a dwell time of 300 and 100 ms, respectively.

Quality control and data analysis

For quality control, two standard solutions of 5 and 50 $\mu\text{g L}^{-1}$ As were analyzed every 10 samples, and results were accepted only if recovery was in an acceptable range (e.g., 90–110 %). The deviation of the instrument was below 1.3 % during the analysis. Total As was certified by the Laver, a standard reference materials (GBW08521 from the National Research Center for Standard Materials in China), which was used to represent phytoplankton tissues. The measured recovery ($n=3$) was 105 % with the concentration of $43\pm 1\text{ mg kg}^{-1}$, which was similar to the certified concentration ($41\pm 3\text{ mg kg}^{-1}$).

The software OriginPro 8.5.0 SR1 (OriginLab Corporation, 1991–2010) was used to measure the concentration of each compound identified. The limits of detection of the instrument were, respectively, 0.6, 0.5, 0.5, and 0.6 $\mu\text{g kg}^{-1}$ for As^{III} ,

DMA, MMA, and As^V with anion exchange column. All data were subjected to analysis of variance (ANOVA) using windows-based SPSS 15.0 and GraphPad Prism 6.0.

Results

Algal growth and inorganic As toxicity

The toxic effect of inorganic As on the algal cells (*M. aeruginosa*) was determined. Algal-specific growth inhibition rate increased with the increasing concentration of As. This algae showed great tolerance to As^V with 72-h IC₅₀ value of 3,582 μM and to As^{III} with 72-h IC₅₀ of 133 μM, indicating that As^{III} was more toxic than As^V to *M. aeruginosa*. 72-h IC₅₀ was many orders of magnitude greater than the expected environmental As concentrations.

Variation of alga cell density and chlorophyll a concentrations of *M. aeruginosa* exposed to different concentrations of inorganic As were shown in Fig. 1. There was an insignificant difference of alga cell density and chlorophyll a between the control and different As^V concentration treatments during 15 days of culture (Fig. 1). Both cell density and chlorophyll a increased gradually with the increase of

incubation time in the log phase of growth, indicating that unnoticeable growth-stimulating or toxic effects of As^V on this alga at the tested concentrations occurred. No significant difference of alga cell density and chlorophyll a was detected between the control and 5–20 μM As^{III} treatments (Fig. 1). Nevertheless, 50 μM As^{III} treatment inhibited the alga growth at the preliminary stage. It regenerated slightly on day 7 and was multiplied on subsequent days, although both cell count and chlorophyll a content decreased obviously compared to the control or other treatments. Approximately complete death of alga at 100 μM As^{III} after 7 days demonstrated its sensitivity to As^{III} stress.

Accumulation of As by *M. aeruginosa*

The intracellular As concentrations of the alga after 15-days As^{III} exposures were significantly higher than that after As^V exposures (Fig. 2). In the control, As concentration in the alga was only 4 μg g⁻¹ DW, indicating the original algae were not contaminated by As. During 15 days of incubation period, total As concentrations accumulated by *M. aeruginosa* were proportional to the increasing concentrations of As added to the growth media (*P*<0.05). Compared to the maximum As accumulation (96.6 μg g⁻¹ DW) in the As^V treatments, the

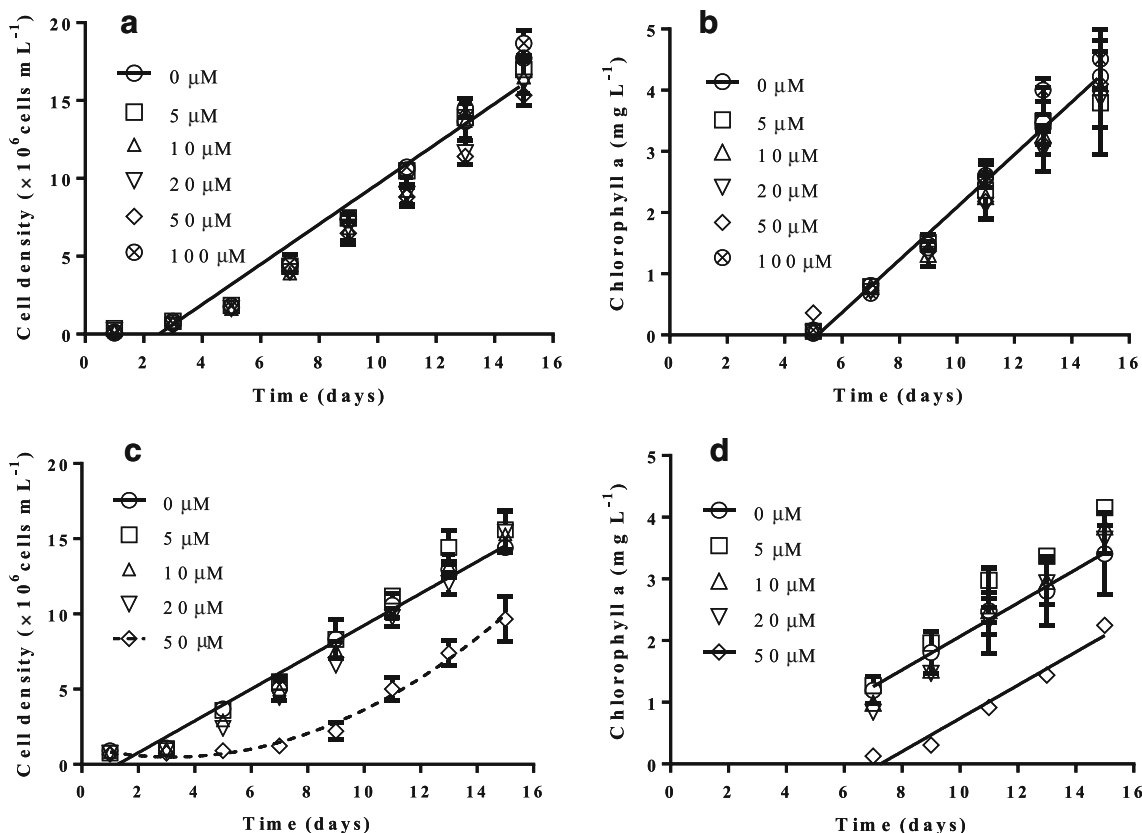


Fig. 1 Cell density and chlorophyll a of *M. aeruginosa* under different As^V or As^{III} concentrations. Data are mean ± SD (*n*=3). **a** Cell density in As^V exposures. **b** Chlorophyll a in As^V exposures. **c** Cell density in As^{III} exposures. **d** Chlorophyll a in As^{III} exposures

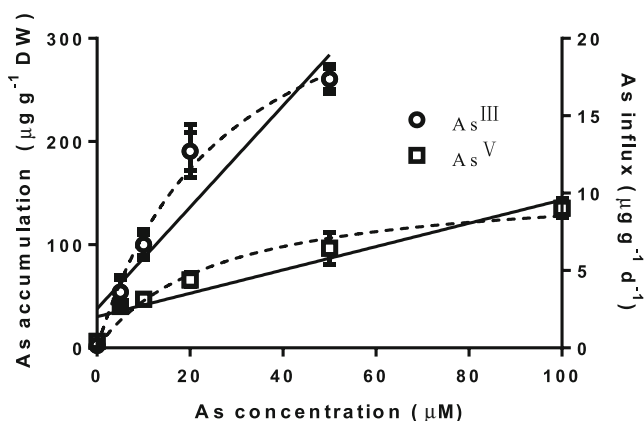


Fig. 2 Arsenic accumulation and influx into *M. aeruginosa* following 15 days of exposures to As^{V} (open circles) or As^{III} (open squares) (mean values \pm SD, $n=3$). Concentrations are shown following the subtraction of the background As concentration. Linear regression (solid line) was used to determine the uptake rate constants ($\mu\text{g g}^{-1} \text{day}^{-1}$), and nonlinear regression (Michaelis–Menten) fits were used to derive metal binding characteristics

intracellular As amount was higher in As^{III} exposures with the maximum value of $260 \mu\text{g g}^{-1} \text{DW}$. The percentage of the intracellular As contents to the total As supplemented in the media was up to 12 % for 100- μM As^{V} exposures and 21 % for 50- μM As^{III} exposures after 15 days. No data about the intracellular concentration was shown in 100- μM As^{III} exposures due to the death of algal cells after 5 days (Fig. 2).

Kinetic constants for As uptake were calculated by fitting hyperbolic models to the data of As accumulation using the nonlinear regression module (Kahle and Zauke 2002; Clason et al. 2003). Arsenic influx can be well described by the Michaelis–Menten equation with R^2 values of 0.940 for As^{V} and 0.973 for As^{III} (Fig. 2, dotted line). The Michaelis–Menten calculations suggested that the maximum uptake rate V_{max} (micrograms per gram per day \pm SE) for As^{III} (27.53 ± 3.185) was twofold greater than the V_{max} for As^{V} (10.66 ± 0.997). The half saturation concentrations

(K_m (micromolar) \pm SE) were 27.66 ± 6.484 for As^{III} and 25.15 ± 6.241 for As^{V} (not significantly different with $P=0.26$, two-tailed t test).

Arsenic speciation in *M. aeruginosa*

Four As species including As^{V} , As^{III} , MMA, and DMA were detected in algal cells in different As^{III} treatments (Fig. 3). However, no MMA was observed in As^{V} treatments, while the range of 0.10 to 0.15 $\mu\text{g g}^{-1}$ for MMA was observed in cells in As^{III} treatments. Analysis of As speciation showed that As^{V} was the predominant species in As^{III} or As^{V} exposures, accounting for 78–93 % of total As, while As^{III} , the other inorganic form, was the minor species that accounted for 6–21 % of the total As. Additionally, small amounts of organoarsenicals accounted for <1 % of total As in different treatments. This alga accumulated relatively more As^{V} from As^{III} solutions than that from As^{V} solutions. Arsenate accumulation in cells increased linearly with increasing concentrations of As^{III} and As^{V} added in the test media. The cellular concentrations of DMA were similar in both As^{III} and As^{V} exposures, and reached a plateau at 10 μM concentrations (about $0.75 \mu\text{g g}^{-1}$). In contrast to As^{V} accumulation, As^{III} accumulation in As^{V} treatments ranged largely from 8.9 to 20.3 $\mu\text{g g}^{-1}$ with a significant difference from the control. However, in As^{III} treatments, the tendency of As^{III} accumulation was similar to As^{V} concentrations in cells, which remarkably increased with increasing addition of As^{III} .

Arsenic speciation changes in the growth media

Although small amounts of As^{V} were found in the growth media containing As^{III} (about 15 %), neither of the tested As species showed any transformation in the algae-free controls after 15-day incubation (data not shown). This confirmed their relative stability in the growth media throughout the test period. Only trace amounts of As^{III} were present in test

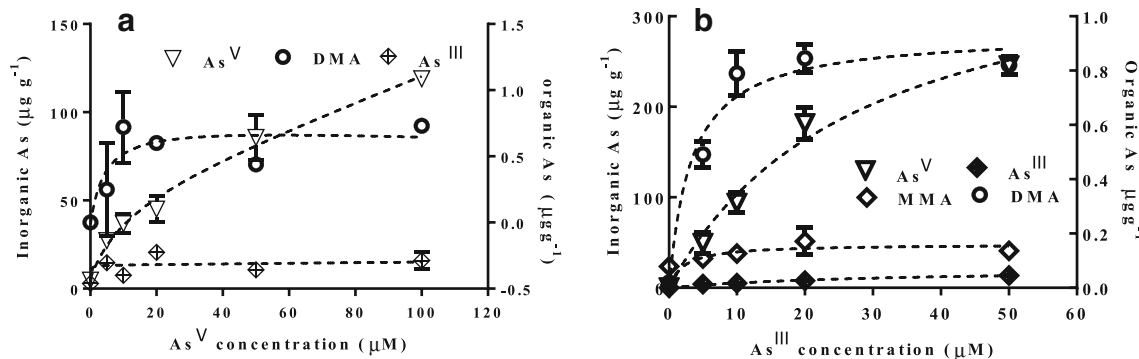


Fig. 3 Arsenic speciation in *M. aeruginosa* after 15 days of exposures to arsenate (a) or arsenite (b) (mean values ($\mu\text{g g}^{-1} \text{dw}$) \pm SD, $n=3$); MMA monomethylarsonous acid, DMA dimethylarsinous acid. Regression(s) have been fit to reflect “One site-specific binding with Hill

slope” using GraphPad Prism. a The coefficients of regression (R^2) for As^{V} , As^{III} , and DMA are 0.9770, 0.4620, and 0.5889, respectively. b The coefficients of regression (R^2) for As^{V} , As^{III} , MMA, and DMA are 0.9695, 0.9758, -0.3265, and 0.9403, respectively

media containing *M. aeruginosa* and decreased with increasing time during the first 5 days in As^V exposures (Fig. 4). As^{III} reduced from the initial As^V and increased with increasing initial As^V concentrations. However, no As^{III} was detected in the controls (no algae), indicating that the presence of As^{III} in the growth media containing *M. aeruginosa* was due to biological reduction. Reduction to As^{III} in the media also occurred insignificantly on other days (Fig. 4). As for As^V treatments, As^V concentrations in the media decreased with increasing time. Although the lowest values were observed on the 13th day, As^V remained relatively stable and predominated in the media over the 15-day period. Methylated As species including DMA and MMA were also detected at lower concentrations in these media. Additionally, both of them increased with time and had insignificant differences among different concentration treatments.

Significant changes of As speciation occurred in the media containing As^{III} (Fig. 5). With the onset of exponential algal growth, As^{III} concentration decreased rapidly with time and almost completely disappeared on the seventh day due to the biologically and chemically mediated oxidation of As^{III} to As^V. However, As^V immediately appeared in the initial stage in the media and then became dominant on the seventh day. At the end of the experiment, 99 % of total As remaining in the solution was presented as As^V. Two other As species, MMA and DMA, which were measured at the same time on the fifth

day, increased similarly with increasing time during 15-day incubation in the media containing different concentrations of As^{III}. Additionally, DMA showed higher concentrations in the media than MMA. In our experiments, there were significant differences between the sum of two reduced transformation products and total As concentrations remaining in the media at the end of the exposure period (two-way ANOVA, $P < 0.01$), indicating that there were other major unidentified As species released into the solution. Supportively, there was a little loss of volatile As speciation, which was verified by ancient Gutzeit method in our experiment.

As can be seen from Fig. 6a, the release rate of As^{III}, averaging from 4.7×10^{-3} to $4.3 \times 10^{-1} \mu\text{M g}^{-1} \text{ day}^{-1}$, dramatically increased with increasing As^V concentration. This suggested that a significant fraction of As^{III} was produced and released by *M. aeruginosa* after As^V exposure. The average release rate of MMA reached peak value with $2.6 \times 10^{-2} \mu\text{M g}^{-1} \text{ day}^{-1}$ at $10 \mu\text{M As}^{\text{V}}$ concentration, but DMA was at $20 \mu\text{M As}^{\text{V}}$ concentration with $7.7 \times 10^{-2} \mu\text{M g}^{-1} \text{ day}^{-1}$. Notably, only As^{III} was found in solution on the first 5 days during As^V exposure, of which average release rate was higher than that of organoarsenicals, especially on the second day. As can be seen from Fig. 6b, all the average release rates of As^V, DMA, and MMA increased with increasing As^{III} concentration during 15-day incubations, particularly at $50 \mu\text{M As}^{\text{III}}$ concentration. This suggested the coexistence of bio-oxidation and methylation of As metabolism.

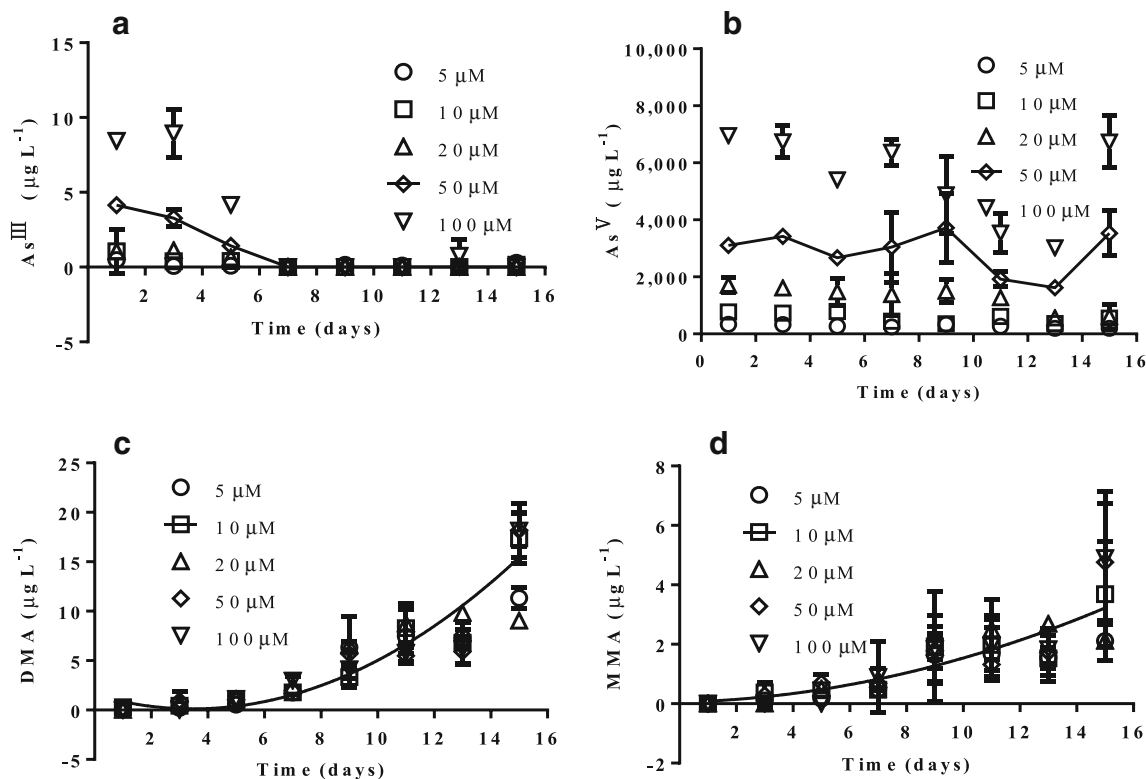


Fig. 4 Arsenic speciation in the media during growth of *M. aeruginosa* after treatment with different concentrations of As^V. Data are mean ± SD ($n = 3$)

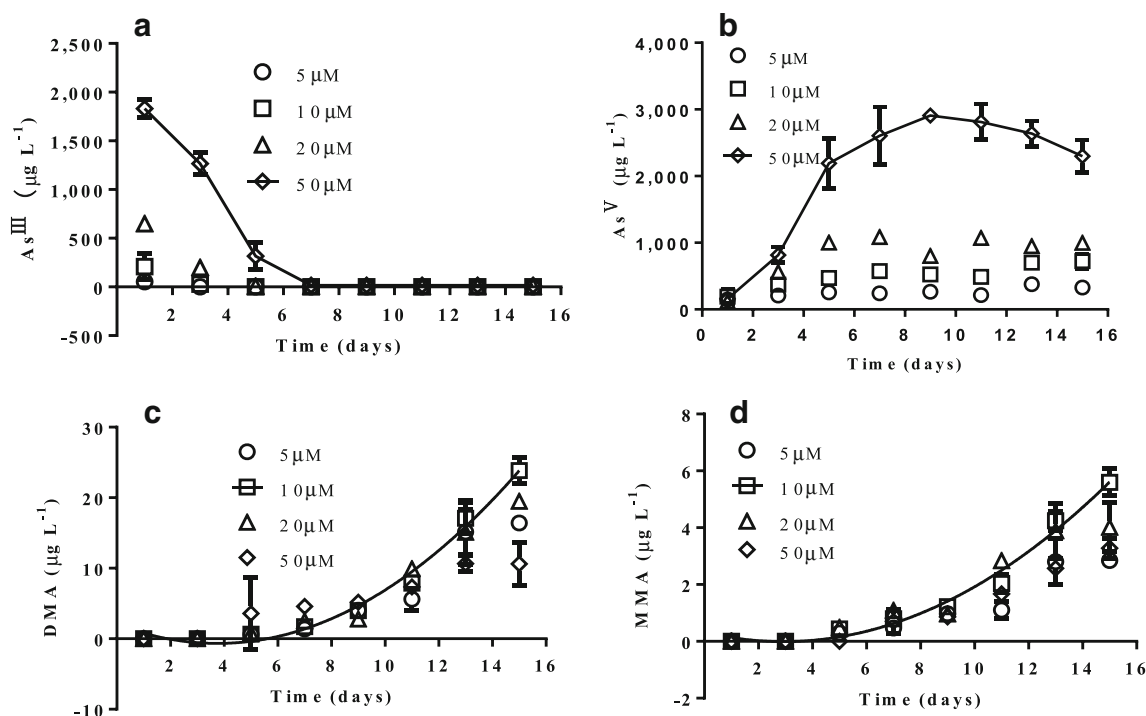


Fig. 5 Arsenic speciation in the media during growth of *M. aeruginosa* after treatment with different concentrations of As^{III} . Data are mean \pm SD ($n=3$)

Discussion

Accumulation of As by *M. aeruginosa*

In this study, the accumulation of As by *M. aeruginosa* typically increased with increasing As^V and As^{III} concentrations in the test media. The active uptake was considered to play major roles in As accumulation in alga (Maeda et al. 1985). Surely, similar studies have already documented the findings in *Synechococcus leopoliensis* and *Chlorella vulgaris* (Budd and Craig 1981; Maeda et al. 1992a, b).

The V_{max} for As^{III} was twice more than that of As^V , suggesting that as an analog of phosphate structure, As^V uptake in *M. aeruginosa* cells is achieved via fewer

available sites (i.e., lower binding site density) in such excess phosphate regimes (Guo et al. 2011). As^{III} can be easily transported via aquaporins in some plants because it exists predominantly as an undissociated neutral molecule at pH 6.0 (Zhao et al. 2009). Therefore, the influx rate of As^{III} was greater than that of As^V in solutions with higher phosphate concentrations in this study (Fig. 2) (Huang et al. 2011). Subsequently, As^{III} exposures significantly elevated intracellular As concentrations compare to As^V exposures.

Additionally, *M. aeruginosa* exhibited a stronger capacity of As bioaccumulation compared to other freshwater algae (Levy et al. 2005; Yin et al. 2011b). It was supported by the results that intracellular As in *M. aeruginosa* was more than 7 % of total As supplemented in the media on the 15th day. Surely, previous reports have stated that much greater

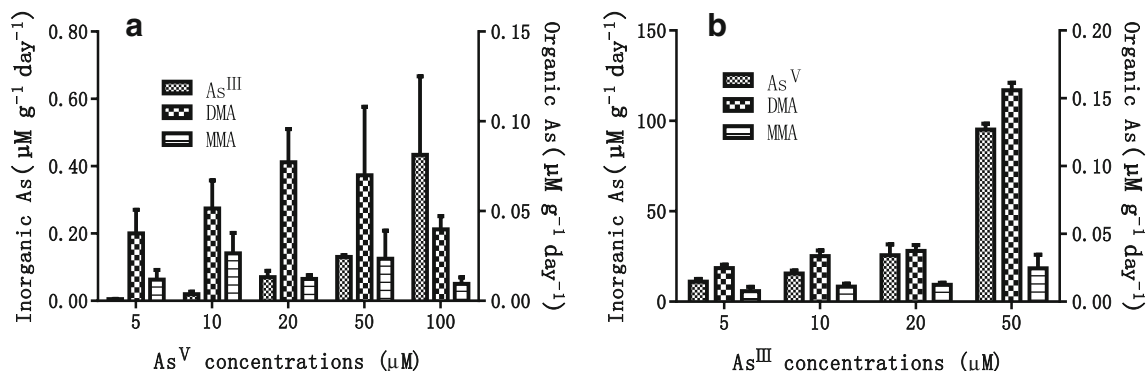


Fig. 6 The average release rate of different As species to the media in As^V (a) and As^{III} (b) exposures. Data are mean \pm SD ($n=3$)

accumulation can occur in *C. vulgaris* and *Synechocystis* from an As-polluted environment (Maeda et al. 1992a, b; Yin et al. 2011a). Consequently, As bioaccumulation in alga bloom should be given more attention due to a potential hazard to animals and humans.

Arsenic speciation in *M. aeruginosa*

The distribution of intracellular As speciation (Fig. 3) clearly demonstrated that As^V was the predominant species followed by As^{III}, DMA, and MMA after 15 days of algae exposure to As^{III} or As^V at concentrations of 5–100 μM. The findings also clearly demonstrated that bio-oxidation of As^{III} to As^V was the predominant transformation process in algal cells in freshwater enriched with As^{III} and phosphate. Similar findings were also observed in *C. vulgaris* and *Synechocystis* (Suhendrayatna et al. 1999; Yin et al. 2011a). Maeda et al. (1992a) similarly found that 95 % of As was accumulated by freshwater algae, *C. vulgaris*, as inorganic species, and dimethylated As was the major methylated As compound. In our study, significant changes of intracellular As speciation occurred after 15 days of exposure to different concentrations of As^V or As^{III} (Fig. 3). With the increase of exposure of As concentrations, intracellular As^V concentrations as the dominant species increased dramatically. However, for methylated As, the contents seemed to be limitation to the alga, just as Maeda inferred in the survey of *C. vulgaris* (Maeda et al. 1992b). Arsenite concentrations in cells showed a similar tendency after As^{III} treatments, but it fluctuated strongly in As^V treatments. Nevertheless, the accumulation of As^{III} in cells occupied a higher proportion for As^V treatments compared to As^{III} treatments. These results demonstrated that *M. aeruginosa* have the ability to biotransform As^V into reduced As species as a precursor for methylation (Hellweger and Lall 2004).

Arsenic speciation changes in the growth media

Rapid changes of As speciation that occurred in the growth media containing inorganic As species indicated that *M. aeruginosa* was able to oxidate, reduce, methylate, and release As to the growth media. Although some of the additional As was incorporated into cells, much of it was apparently metabolized and rapidly released to the surrounding water. The reduction of As^V to As^{III} was rapid in the initial stage, which exhibited positive correlations with the exposure concentrations of As^V, just as Sanders and Cullen documented (Sanders and Windom 1980; Cullen et al. 1994). The result of a peak of As^{III} in the media occurred in the initial stage and preceded algal bloom as reported by Hellweger (2005). It is generally known that the amount of organo-As species in the media depends

on the rates of methylation/demethylation and release processes. The discovery of MMA in the growth media with little or no MMA in cells also supported the methylation of inorganic As as well as a rapid and ready excretion of MMA and DMA in As^V exposures.

As^{III} oxidation rates in the media containing As^{III} dominated by the alga were significant (Fig. 6), especially at 50 μM As^{III} concentration. The average release rate of As^{III} increased with increasing As^V concentration, indicating that As^V reduction by algae was enhanced with increasing As concentrations in the media. The observed reduction rates of *M. aeruginosa* with 2.0×10^{-2} to 4.3×10^{-1} μM g⁻¹ day⁻¹ were similar to net reduction rates previously estimated from green alga *Chlorella* sp. in culture (Knauer and Hemond 2000).

Generally, bio-oxidation was the first and primary transformation mechanism for As^{III} biotransformation, while bioreduction together with subsequent methylation may play an important role in As^V biotransformation. Additionally, the results also showed that the alga was more sensitive to As^{III}. Similar results were also often reported (Carbonell et al. 1998; Levy et al. 2005; Duester et al. 2011; Yin et al. 2011a). These similar findings supported that this algal bloom species from As-polluted water could tolerate higher As concentrations in freshwater environments and might become the dominant species in As-contaminated waters. Therefore, As toxicity depends on not only As speciation in the media but also the specific mechanism of bioaccumulation and biotransformation of the algal species of interest.

Conclusion

The aims of this study were to investigate the bioaccumulation and biotransformation of inorganic As in *M. aeruginosa* as well as the changes of As speciation in the media caused by the dominant species in cyanobacterial bloom. *M. aeruginosa* came from As-polluted lake has high tolerance to As^V with little influence on its growth. Although different As species transformations occurred during As^V or As^{III} exposures, As^V was the prominent species in the media, eventually with unremarkable DMA and MMA. It can be concluded that the transformations of As species in *M. aeruginosa* were seldom affected by external As concentrations, especially in high concentrations of As pollution.

Acknowledgments We thank the anonymous reviewers for their helpful comments. This work was supported by International Science & Technology Cooperation Program of China (2011DFB91710), the National Nature Science Foundation of China (Project Nos. 21277136 and 41271484), and the Fujian Provincial Education Department Foundation (JA12214).

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