ORIGINAL PAPER



# A new alkaline lipase obtained from the metagenome of marine sponge *Ircinia* sp.

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Received: 4 November 2014/Accepted: 23 April 2015/Published online: 29 April 2015 © Springer Science+Business Media Dordrecht 2015

Abstract Microorganisms associated with marine sponges are potential resources for marine enzymes. In this study, culture-independent metagenomic approach was used to isolate lipases from the complex microbiome of the sponge Ircinia sp. obtained from the South China Sea. A metagenomic library was constructed, containing 6568 clones, and functional screening on 1 % tributyrin agar resulted in the identification of a positive lipase clone (35F4). Following sequence analysis 35F4 clone was found to contain a putative lipase gene *lipA*. Sequence analysis of the predicted amino acid sequence of LipA revealed that it is a member of subfamily I.1 of lipases, with 63 % amino acid similarity to the lactonizing lipase from Aeromonas veronii (WP\_021231793). Based on the predicted secondary structure, LipA was predicted to be an alkaline enzyme by sequence/structure analysis. Heterologous expression of lipA in E. coli BL21 (DE3) was performed and the characterization of the recombinant enzyme LipA showed that it is an alkaline enzyme with high tolerance to organic solvents. The isolated lipase LipA was active in the broad alkaline range, with the highest activity at pH 9.0, and had a high level of stability over a pH range of 7.0-12.0. The activity of LipA was increased in the

presence of 5 mM Ca<sup>2+</sup> and some organic solvents, e.g. methanol, acetone and isopropanol. The optimum temperature for the activity of LipA is 40 °C and the molecular weight of LipA was determined to be  $\sim$  30 kDa by SDS-PAGE. LipA is an alkaline lipase and shows good tolerance to some organic solvents, which make it of potential utility in the detergent industry and enzyme mediated organic synthesis. The result of this study has broadened the diversity of known lipolytic genes and demonstrated that marine sponges are an important source for new enzymes.

**Keywords** Lipase · Metagenomic library · Marine sponge · Alkaline lipase · Secondary structure amino acid composition

# Introduction

Lipases, which are defined as carboxylesterases that catalyze both the hydrolysis and synthesis of long-chain acylglycerols (Gupta et al. 2004), are a group of important biocatalysts with uses in organic chemistry, pharmaceutical, food and leather industries, biodegradation of plastics and production of optically active compounds (Andree et al. 1980; Jaeger and Reetz 1998; Gombert et al. 1999; Beisson et al. 2000; Muralidhar et al. 2001; Gupta et al. 2003; Saxena et al. 2003; Kiran et al. 2008; Bajaj et al. 2010). Particularly, lipases have great value to produce biodiesel from biologically derived oil by transesterification (Watanabe et al. 2002; Bajaj et al. 2010) and esterification of free fatty acids.

The challenges of the lipase industry include the discovery of new lipases producers with good properties, for example, high stability of lipase in water-miscible organic solvents is desired in the pharmaceutical industry and for

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biodiesel production (Glogauer et al. 2011); halo-tolerant lipases are particularly useful for those industrial applications related to the production of marine products (Selvin et al. 2012); alkaline lipases in a detergent should have high activity and stability over a broad range of temperature and pH, and should also be compatible with different components in a detergent including metal ions, surfactants and oxidants (Wang et al. 1995).

The discovery of new esterases and lipases will increase the diversity of lipolytic enzymes and therefore aid the selection of suitable biocatalysts for challenging reactions (Ranjan et al. 2005). Microbial lipases have received a great deal of attention in the field of food technology, pharmaceutical sciences, chemical and detergent industries (Gupta et al. 2004; Jaeger et al. 1999). But it is known that the great majority of microorganisms in the natural environment can't be cultured (Kennedy et al. 2008), which limits the searching for new enzymes from microorganisms. Thus, metagenomics has become an increasingly important option for finding novel enzymes from complex microbiomes (Hu et al. 2010; Jeon et al. 2009; Kim et al. 2009). For example, a novel cold-adapted alkaline lipase belonging to a new family of lipolytic enzyme was isolated from an intertidal flat metagenome (Kim et al. 2009); 15 different lipolytic genes were obtained from the microbial metagenomic library of the South China Sea marine sediment and two proteins probably representing a novel family of the bacterial lipolytic enzymes were discovered (Hu et al. 2010).

Using metagenomic approach, novel lipolytic enzymes were isolated from sponges *Aplysina aerophoba* and *Hyrtios erecta* (Okamura et al. 2010). A lipase with novel halotolerance was also isolated from the metagenome of the marine sponge *Haliclona simulans* (Selvin et al. 2012). Marine ecosystems represent a large and as yet largely unexplored reservoir of biodiversity with respect to industrially useful biocatalysts (Kiran et al. 2014), for example, marine sponge-associated microbes were found to be an important source for marine enzymes such as lipase, urease, protease, cellulase and chitinase (Kiran et al. 2008; Han et al. 2009; Zhang et al. 2009b; Feby and Nair 2010; Shanmughapriya et al. 2010).

In this study, a metagenomic library was constructed from the marine sponge *Ircinia* sp. from the South China Sea to obtain the lipolytic genes, and the gene was heterologously expressed and the recombinant protein was characterized.

#### Materials and methods

# Sample collection and DNA extraction

Three specimens of the marine sponge *Ircinia* sp. were collected by scuba diving at a depth of ca. 10–20 m from

Yongxing Island (latitude  $112^{\circ}20'$ E, longitude  $16^{\circ}50'$ N), South China Sea in the year of 2010. The sponges was transported in an ice-cooled box immediately to the laboratory. The microbes on the sponge surface and inner cavity were removed by washing three times with sterile artificial seawater (ASW) (1.1 g CaCl<sub>2</sub>, 10.2 g MgCl<sub>2</sub>. 6H<sub>2</sub>O, 31.6 g NaCl, 0.75 g KCl, 1.0 g Na<sub>2</sub>SO<sub>4</sub>, 2.4 g Tris-HCl, 0.02 g NaHCO<sub>3</sub>, 1 L distilled water, pH 7.6). Then the sponge samples were stored at -20 °C until DNA extraction.

Three sponge cubes (~1 cm<sup>2</sup>) were dissociated in sterile ASW without Ca<sup>2+</sup> and Mg<sup>2+</sup> at 110 rpm and 20 °C for 3 h, and then ground into slurry using a mortar and pestle. Sponge slurry was centrifuged at  $100 \times g$  for 15 min to separate the skeletal components. The supernatant was freeze-dried and ground into powder, which was washed twice with 1 mL TE buffer at 12,000×g for 1 min. DNA was extracted with AllPrep DNA/RNA Mini Kit (Qiagen, Germany). The quantity and purity of DNA was determined respectively by testing the OD<sub>260</sub> and the OD<sub>260</sub>/ OD<sub>280</sub> with NanoDrop Spectrophotometer ND-1000 (Thermo Fisher Scientific, Wilmington, USA).

#### Construction of metagenomic library

The obtained DNA was partially digested with *Sau* 3AI and then separated by 0.5 % agarose gel electrophoresis to get the digested DNA fragments of 2–10 kb. The ligation of the DNA fragments with *Bam* HI-digested pUC18 (2686 bp, ampicillin-resistant, Takara), which had previously been dephosphorylated with calf intestinal alkaline phosphatase, was performed using T4 DNA ligase (NewEnglandBiolabs) for 10–12 h at 16 °C. The products were transformed into *E. coli* DH5 $\alpha$ . The transformants were plated on LB agar containing 100 µg/mL ampicillin, X-Gal and isopropyl- $\beta$ -D-thiogalactoside (IPTG). Then the white colonies were selected and transferred into 96-well plates and cultured in LB media containing 100 µg/mL ampicillin (Wei et al. 2009) at 37 °C for 24 h. The library was stored at -80 °C with 20 % glycerin in 96-well plates.

# Screening for clones with lipolytic activity and sequence analysis

Clones in the library were replicated onto LB agar plates containing 100  $\mu$ g/mL ampicillin and 1 % tributyrin, and cultured at 37 °C. The clones with a clear halo around were identified as positive ones. The insert DNA in the positive clones were sequenced with DNA sequencer ABI3730x1 (Applied Biosystems). Sequence assembly and editing were performed with the CodonCode Aligner software (CodonCode Corporation, Dedham, MA, USA). The amino acid sequences were compared with the non-redundant sequence database deposited at the NCBI using BLAST (Altschul et al. 1997), and the open reading frames (ORFs) were identified with the ORF Finder tool (http://www.ncbi. nlm.nih.gov/gorf/orfig.cgi) (Wheeler et al. 2008). The unrooted phylogenetic tree was constructed with MEGA 5.0. Multiple alignments between protein sequences were performed with the Clustal W program (Thompson et al. 1994). Signal peptide was determined using SignalP 4.0 Server (http://www.cbs.dtu.dk/services/SignalP/) analysis. Secondary structure prediction of the enzyme sequence was carried out using the Predator program (Frishman and Argos 1997). A random forest model was used to predict acidic or alkaline enzyme (Zhang et al. 2009a).

### Cloning and expression of lipA gene

The lipA gene without signal peptide sequence was amplified by PCR using the primers: 5'-GGGAATTCA TATGGGCTATACCCAAACT-3' (the Nde I site was underlined) and 5'-GGAAGGGCTCGAGTTATAAGCCT AATTT-3' (the XhoI site was underlined). The PCR product was digested with NdeI and XhoI, cloned into the NdeI/XhoI digested pET28a (+) (5369 bp, kanamycin-resistant, Novagen), and then transformed into E. coli BL21 (DE3) (Novagen). The inserted gene was examined by DNA sequence analysis. The transformed cells carrying the pET28a (+) vector containing the lipA gene were cultured until the cell density reached  $OD_{600}$  of 0.6 and induced by the addition of IPTG to a final concentration of 1 mM. The induced culture was incubated for a further overnight at 16 °C. E. coli BL21(DE3) cells were then harvested by centrifugation at  $4000 \times g$  for 20 min, washed twice with 50 mM Tris-HCl buffer (pH 8.0), and disrupted by ultrasonication (80 cycles of 3 s pulses, 200 W, with 5 s intervals). The supernatant was removed by centrifugation  $(12,000 \times g \text{ for 5 min at 4 °C})$ . The obtained inclusion body was washed with 50 mM Tris-HCl buffer (pH 8.0) and suspended in the same buffer containing 8 M urea and shaken at 100 rpm for 2 h at 4 °C to make it dissolve. After centrifugation under the same conditions  $(12,000 \times g,$ 5 min, 4 °C), the supernatant containing the His-tagged protein was loaded onto a Ni-NTA affinity His-Bind resin column (GE Healthcare Life Sciences) (Zheng et al. 2011) previously equilibrated with washing buffer (50 mM Tris-HCl, pH 8.0, 8 M urea). The protein was eluted using an imidazole step gradient (10, 20, 50, 100, 200 and 500 mM) in washing buffer. Finally, the imidazoles and urea in the purified protein were removed by dialysis in 50 mM Tris-HCl, pH 8.0 containing different concentration of urea (8, 6, 5, 4, 2, 0 M) successively and the protein solution was concentrated by polyethylene glycerol. The purified enzyme was evaluated by 12 % sodium dodecyl sulphate-polyacrylamide gel electrophoresis (SDS-PAGE) (Laemmli 1970). The protein concentration was determined by the Bradford assay (Bradford 1976).

#### Analysis of enzymatic properties of LipA

The enzymatic activity of LipA was determined with pnitrophenyl palmitate (p-NPP) as substrate (Selvin et al. 2012). The substrate solution was prepared by mixing 12 µL stock solution of 4 mM p-NPP in acetonitrile/isopropanol (1/4 v/v) with an assay buffer of 50 mM Tris-HCl (pH 8.0), 0.3 % (v/v) Triton X and 1 mM CaCl<sub>2</sub> to a final volume of 230 µL. Then 230 µL of the substrate solution was pipette into a 96-well microtiter plate and the reaction was initiated by the addition of 20 µL of the enzyme solution (50 mM Tris-HCl pH 9.0, 1 mM·CaCl<sub>2</sub>). The amount of p-nitrophenol released from p-NPP was continuously monitored spectrophotometrically at 405 nm over a period of 10 min at 40 °C. All experiments were performed in triplicate. The extinction coefficients of pnitrophenol was determined to be  $1 \times 10^4 \text{ M}^{-1} \times \text{ cm}^{-1}$  at 405 nm. One unit of enzyme activity was defined as the activity required to release 1 µmol of p-nitrophenol per minute from *p*-nitrophenyl ester.

The optimum temperature of LipA was determined from 20 to 60 °C. The thermostability of the enzyme was determined by preincubating the enzyme at 4–60 °C for 120 min and then the residual activity was measured at 40 °C.

The activity of the recombinant enzyme was determined between pH 5.0–12.0. The pH stability of the enzyme was tested by preincubating the enzyme for 30 min in three different buffer systems: 20 mM disodiumhydrogen phosphate–citric acid buffer (pH 5.0–8.0), 50 mM Tris-HCl buffer (pH 9.0–10.0), 20 mM glycine-NaOH (pH 11.0–12.0).

To test the effect of metal ions on the enzymatic activity, 5 mM of CaCl<sub>2</sub>, NaCl, KCl, CuCl<sub>2</sub>, MgCl<sub>2</sub>, CoCl<sub>2</sub> and EDTA were added (Lee et al. 2006). The effects of organic solvents on the LipA activity were assessed with 15 or 30 % of methanol, ethanol, isopropanol, acetone and dimethyl sulfoxide (DMSO).

Kinetic experiments were carried out with different concentration of various *p*-nitrophenyl esters: *p*-nitrophenyl caprylate (C8), *p*-nitrophenyl decanoate (C10), *p*-nitrophenyl dodecanoate (C12), *p*-nitrophenyl myristate (C14), *p*-NPP (C16), ranging from 0.2 to 3 mM. The  $K_m$  values were determined by a linear least-squares fitting of a Lineweaver-Burke plot of the Michaelis–Menten equation.

#### Nucleotide sequence accession number

The *lip*A gene sequence of 921 bp was deposited in the GenBank database under the accession numbers KC152848.

# Results

# Screening of gene encoding lipase from the metagenomic library of sponge microbiome

Approximately 6  $\mu$ g DNA of sponge *Ircinia* sp. was used for the construction of metagenomic library. Consequently, a metagenomic library including 6568 clones (average insert size 2–10 kb) was constructed. The clones of the whole library were screened for lipolytic activity on 1 % tributyrin plates, and consequently a clone (35F4) with a clear halo around was found (Fig. 1).

The insert DNA (4602 bp) of 35F4 was sequenced. The sequence analysis revealed the presence of 921 bp *lipA* gene encoding lipase. The translated amino acid sequence of LipA showed 63 % similarity to lactonizing lipase from *Aeromonas veronii* (WP\_021231793) and lactonizing lipase from *Aeromonas veronii* (WP\_005356782), 62 % similarity to triacylglycerol lipase from *Glaciecola polaris* (WP\_007104527), 62 % similarity to lipase from *Aeromonas hydrophila* ML09-11 (YP\_008041659). The phylogenetic analysis showed that LipA is a member of subfamily I.1 and might be novel as it was not clustered with the reference sequences (Fig. 2).

In addition, the sequence alignment showed that it belongs to  $\alpha/\beta$  hydrolase superfamily, with conserved signature sequences (Gly-X-Ser-X-Gly) of lipolytic enzymes in the ORF (101–105, Fig. 3). The 35F4 LipA catalytic triad was predicted to be formed by Ser<sup>103</sup>, Asp<sup>250</sup> and His<sup>272</sup> and the nucleophilic Ser residue appears in the conserved pentapeptide Gly-X-Ser-X-Gly. Two Asp residues of 35F4 LipA (Asp<sup>232</sup>and Asp<sup>274</sup>) form a calcium



Fig. 1 Identification of a lipolytic clone, 35F4, that forms a hydrolysis halo in the tributyrin plate assay

binding pocket that is conserved in lipases of subfamilies I.1 (Arpigny and Jaeger 1999). 35F4 LipA had two Cys residue (Cys<sup>205</sup> and Cys<sup>256</sup>) and they were predicted to form a disulfide bridge (Nardini et al. 2000).

Also, the secondary structure of 35F4 was predicted and the content of amino acid residues in helix, sheet and random coil regions were computed (Zhang et al. 2009a). 35F4 LipA was predicted to be an alkaline enzyme using the random forest model (Zhang et al. 2009a).

#### The enzymatic characteristics of LipA

After removing signal peptide sequence, LipA was expressed in *E. coli* BL21 (DE3). However, an insoluble inclusion body was formed. 8 M urea was added to make the inclusion body dissolve, and the recombinant LipA with a C-terminal histidine tag was purified by His-Bind resin affinity chromatography. The protein size was determined to be about 30 kDa by SDS-PAGE (12 %) (Fig. 4).

The temperature profile of enzyme activity was showed in Fig. 5a. The enzyme was active over a wide range of temperature, retaining over 50 % of its relative activity between 30 and 55 °C (Fig. 5a) with the optimum temperature at 40 °C. After incubation for 120 min, the LipA retained 90, 42.39, 30.54 and 10 % of relative activity at 4, 40, 50 and 60 °C, respectively (Fig. 5b).

Over a pH range of 5–12, LipA displayed more than 60 % relative activity at pH values from 7.0 to 10.0, with the maximum activity at pH 9.0 (Fig. 5c). The lipase retained above 60 % of its relative activity after 30 min incubation at pH 7.0–12.0 in the pH stability study (Fig. 5d). These results showed that LipA is an alkaline lipase.

The effect tests of metal ions revealed that the relative activity of LipA was greatly inhibited by the addition of 5 mM Cu<sup>2+</sup> and Co<sup>2+</sup>, respectively (Table 1), and in contrast, the relative activity of LipA was increased to 120.86 and 112.24 % with the addition of 5 mM Ca<sup>2+</sup>, Mg<sup>2+</sup> respectively (Table 1), indicating the LipA activity was stimulated particularly by Ca<sup>2+</sup>. The activation observed with calcium is consistent with the inhibitory effect of EDTA, which preferentially binds calcium (Table 1). The presence of 5 mM Na<sup>+</sup> enhanced LipA activity (Table 1) as well. The most likely explanation for this is it improves the solubility of pNPP within the emulsion. Evidence of this increase in solubility is given by the fact that at 1.5 M NaCl the pNPP emulsion remains transparent even without heating (Glogauer et al. 2011).

The addition of 15 or 30 % methanol or acetone greatly increased the activity of LipA. The increase of enzyme activity was also observed in 15 % isopropanol, while 30 % isopropanol inhibited the activity. In the other two solvents, ethanol and DMSO (15 and 30 %), the activity of LipA was inhibited (Table 2).



**Fig. 2** Phylogenetic analysis of 35F4 LipA and closely related proteins. The protein sequences are retrieved from GenBank (NCBI). The phylogenetic tree is generated using MEGA 5. The *scale* represents the number of amino acid substitutions per site. *Open* 

The activity of purified LipA was examined by using p-nitrophenyl esters with five different length acyl chains (C8, C10, C12, C14, C16) as substrate. Lipase activities and K<sub>m</sub> of the enzyme toward different p-nitrophenyl esters were showed in Table 3. The purified enzyme showed the highest activity towards p-nitrophenyl myristate (C14).

## Discussion

Lipases which are active and stable in alkaline media have a great potential for applications in the detergent industry (Wang et al. 1995). For example, lipases that will work under alkaline conditions as fat stain removers are desirable (Cherif et al. 2011).

*circle* represents the closely related proteins with 35F4 LipA; *X with dots* represents the proteins obtained through metagenomic approach (Glogauer et al. 2011; Lee et al. 2006; Kim et al. 2009; Jeon et al. 2009)

In this study, the obtained lipase (LipA) from the metagenomic library of marine sponge Ircinia sp. was active in the broad alkaline range, with the highest activity at pH 9.0, and had a high level of stability over a pH range of 7.0–12.0 (retaining above 60 % activity), indicating it is an alkaline enzyme. This phenomenon of marine derived lipases preferring alkaline pH as optimum was also reported previously through metagenomic approach and culturing of isolated microorganisms. A new lipase, LipEH 166, isolated from an intertidal flat metagenome had the highest activity at pH 8.0, and maintained more than 80 % activity in the pH range of 5–11 (Kim et al. 2009). Another lipase from the marine bacterium Oceanobacillus sp. PUMB02 was optimally active at pH 8.0 and was stable at alkaline pH (Kiran et al. 2014). Also a detergent compatible lipase produced by marine Bacillus smithii BTMS

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Fig. 3 Characteristic residues in multiple sequence alignment between 35F4 LipA and lipases from subfamilies I.1. Conserved residues of the active site (as), cysteine residues forming the disulfide bridge (db) and aspartic residues involved in the calcium binding domains (cb) are annotated. *Shaded* regions indicate similar amino acids in all the aligned sequences. The accession number and the organism name of the aligned sequences are (from *top to the bottom*): Lipase from *Proteus mirabilis* ATCC 29906 (Genebank:ZP\_03839679); Lipase

11 was found to have pH 8.0 as optimum condition for maximal activity and was active over pH 7.0–10.0 (Lailaja and Chandrasekaran 2013). These alkaline lipases have desirable properties with regard to pH optimum and pH stability for a potential application in the detergent industry, supporting the faith that marine ecosystem is a large

from *Aeromonas hydrophila* ML09-11 (Genebank: YP\_00804165); Lipase from *Yersinia enterocolitica* subsp. *Palearctica* (Genebank:CBY26912); Lipase from uncultured bacterium (Genebank: AEK97793); Lipase from *Pseudomonas fragi* [Genebank:CAC07191]; 35F4 LipA; Triacylglycerol lipase from *Glaciecola polaris* (GeneBank: WP\_007104527); Lactonizing lipase from *Aeromonas veronii* (GeneBank: WP\_021231793); Lactonizing lipase from *Aeromonas veronii* (GeneBank: WP\_005356782)

and as yet largely unexplored reservoir of biodiversity with respect to industrially useful biocatalysts.

Because enzyme instability at pH extreme is one of the main bottlenecks in extending the application of it, the stability mechanism of acidic and alkaline enzymes is quite important and interesting not only academically but also industrially. The stability of acidic and alkaline enzymes had been studied in the biophysical and biotechnological research areas, and it has been investigated whether acidophily and alkaliphily can be detected at the amino acid level (Dubnovitsky et al. 2005; Kelch et al. 2007; Geierstanger et al. 1998; Settembre et al. 2004; Shirai et al.



**Fig. 4** SDS-PAGE of purified LipA. *M* Molecular weight markers; *1*: Induced cell lysate; *2*: Flow through proteins; *3*: Purified LipA. The size of the Molecular weight markers is 97.2, 66.4, 44.3, 29.0, 20.1, 14.3 kDa (from *top* to *bottom*)

1997). It was proposed that the propensity of the individual residues to participate in different secondary structures might be a general stability mechanism for their adaptation to pH extremes. Base on it, a random forest model in conjunction with a feature named secondary structure amino acid composition was used to discriminate acidic and alkaline enzymes (Zhang et al. 2009a). Using this model, LipA was predicted to be an alkaline enzyme, which is consistent with the experimental data, supporting the proposed adaptation mechanism of enzymes to pH extremes. The new lipase could be further used to achieve a better understanding of enzyme mechanism and structure–function relationships.

LipA activity was found to be enhanced in the presence of  $Ca^{2+}$ , which was similar to the increases observed in the lipases from the metagenome of the marine sponge *Haliclona simulans* (Selvin et al. 2012) and the lipase isolated from the tidal flat sediment metagenomic library (Lee et al. 2006).  $Ca^{2+}$  ion is regarded as a ligand between amino acid residues in the enzyme active site (Salameh and Wiegel 2007) and the electrostatic interaction between calcium and fatty acids produced by the hydrolysis of the substrate leads to the "clearing" of the active site thereby allowing another substrate molecule to access the site (Voget et al. 2003).



**Fig. 5** Enzymatic characteristics of LipA. **a** The effect of temperature on LipA. Enzymatic activity is assayed toward *p*-NPP (C16) in 50 mM Tris-HCl (pH 9.0) for 10 min; **b** Thermostability. The thermostability of purified LipA is determined by preincubating the LipA in 50 mM Tris-HCl buffer (pH 9.0) at 4 °C (*filled diamond*), 40 °C (*filled circle*), 50 °C (*filled triangle*), 60 °C (*filled square*) for 120 min; **c** The effect of pH on LipA. Enzymatic activity is measured

at 40 °C in buffers ranging from pH 5.0–12.0 for 10 min; **d** The pH stability. The activity is determined after preincubating the LipA at 40 °C for 30 min in three different buffer systems: 20 mM disodiumhydrogen phosphate-citric acid buffer (pH 5.0–8.0), 50 mM Tris-HCl buffer (pH 9.0–10.0), 20 mM glycine-NaOH (pH 11.0–12.0). In each of the panels, the activity under optimum condition is set as 100 %

Similarly, LipC12 belonging to the subfamily I.1 of bacterial lipases was isolated through metagenomic approach and found to be calcium ion dependent (Glogauer et al. 2011).

In this study, the LipA activity was increased in the presence of organic solvents, such as methanol, isopropanol and acetone. Similarly, the stability of the lipase (Lpc53E1)

Table 1 Effect of metal ions and EDTA on the activity of LipA

Additives	Relative activity (%)	
None	100	
EDTA	$74.70 \pm 6.59$	
$K^+$	$94.37 \pm 4.24$	
Na <sup>+</sup>	$104.39 \pm 2.78$	
Ca <sup>2+</sup>	$120.86 \pm 5.69$	
$Mg^{2+}$	$112.24 \pm 1.75$	
Co <sup>2+</sup>	$6.04 \pm 1.87$	
Cu <sup>2+</sup>	$19.46 \pm 2.19$	

The lipase activity in the presence of the additives was compared with the control whose activity was taken as 100 %. Reaction conditions: Substrate *p*-nitrophenyl palmitate (*p*-NPP); Temperature 40 °C; pH 9.0

Results were presented as mean  $\pm$  standard deviation, n = 3

Table 2 Effect of organic solvents on the activity of LipA

Relative activity (%)			
Concentration			
30 %			
100			
$173.19 \pm 4.78$			
$36.66 \pm 4.52$			
$97.63 \pm 4.34$			
159.65 ± 3.27			
$56.67 \pm 5.17$			

The lipase activity in the presence of an organic solvent was compared with the control whose activity was taken as 100 %. Reaction conditions: Substrate *p*-nitrophenylpalmitate (*p*-NPP); Temperature 40 °C; pH 9.0

The values are mean  $\pm$  standard deviation, n = 3

 Table 3
 Activity and K<sub>m</sub> of purified LipA using *p*-nitrophenyl esters as substrates

from the marine sponge was increased upon the exposure to solvents, e.g. methanol, isopropanol and acetone (Selvin et al. 2012). Generally, enzymes, being proteins, lose their activity at organic co-solvent concentrations higher than 10-20 % (Gupta et al. 1997). Stability and activity of enzymes in organic solvents depend not only on the properties and concentration of the organic solvents, but also on the nature of the enzymes (Torres and Castro 2004). It is quite well known that the lipase activity can be increased in the presence of organic solvents, for example, Candida rugosa lipase is activated by organic solvents, which keep the lid of the enzyme in the open conformation, facilitating the access of the substrate to the active site (Grochulski et al. 1993; Colton et al. 1995). Thus the increased LipA activity suggests that the enzyme may possess a lid that is converted into an open conformation in the presence of organic solvents. These findings are particularly significant, due to the fact that organic solvents have been used in biodiesel production through biocatalysis (Glogauer et al. 2011). It was observed that B. smithii lipase with enhanced activity in the presence of 1-5 % organic solvents could effectively catalyze methyl-ester synthesis between fatty acids and methanol to obtain methyl esters of long-chain fatty acids which could be used as diesel fuels (Lailaja and Chandrasekaran 2013). In the present study, the activity of LipA was enhanced in 30 % methanol or 30 % acetone, suggesting that the enzyme holds potential for application in transesterification and ester synthesis.

LipA was found to have substrate specificity for *p*-nitrophenyl myristate (C14) followed by *p*-nitrophenyl dodecanoate (C12), *p*-nitrophenyl decanoate (C10) and *p*-NPP (C16). A low  $K_m$  value represents a high affinity. For most industrially relevant enzymes,  $K_m$  values range between  $10^{-1}$  and  $10^{-5}$  M, although the  $K_m$  values of the enzyme generally vary widely (Fullbrook 1996). The marine derived *B. smithii* lipase had a value of 0.1 mM for the hydrolysis of *p*-nitrophenyl butyrate (Lailaja and Chandrasekaran 2013). In this study, the reaction kinetics of LipA with *p*-nitrophenyl esters with different length acyl chains indicated that the enzyme had the lowest  $K_m$  value of 0.42 mM for the hydrolysis of *p*-nitrophenyl myristate (C14).

<i>p</i> -Nitrophenyl esters	Activity (U/mg)	K <sub>m</sub> (mM)	V <sub>max</sub> [µg(product)/min mg(protein)]	
<i>p</i> -NP caprylate(C8)	$5.07\pm0.79$	$1.35 \pm 0.026$	$0.68 \pm 0.142$	
p-NP decanoate(C10)	$7.53\pm0.99$	$0.91\pm0.019$	$0.978 \pm 0.149$	
p-NP dodecanoate(C12)	$12.17 \pm 1.37$	$0.65 \pm 0.021$	$1.02 \pm 0.121$	
p-NP myristate(C14)	$17.59 \pm 1.83$	$0.42 \pm 0.012$	$13.3 \pm 2.79$	
p-NP palmitate(C16)	$5.26 \pm 1.04$	$1.27\pm0.034$	$0.79 \pm 0.097$	

One unit of enzyme activity was defined as the amount of activity required to release 1 µmol of *p*-nitrophenol/min from *p*-nitrophenyl ester. Reaction conditions: Temperature 40 °C; pH 9.0

The values are mean  $\pm$  standard deviation, n=3

In conclusion, a metagenomic library of marine sponge Ircinia sp. from South China Sea was constructed. A lipolytic-active positive clone, 35F4, was discovered using an activity-based method from the metagenomic library of marine sponge Ircinia sp. The sequence analysis revealed that the predicted amino acid sequence displayed 63 % similarity to the lactonizing lipase from Aeromonas veronii (WP\_021231793). Alignment of the protein sequence of LipA with known lipase families showed that LipA is a member of subfamily I.1 lipases. And LipA was predicted to be an alkaline enzyme by sequence/structure analysis, which was consistent with the characterization of the heterologously expressed enzyme that LipA was active and stable in the broad alkaline pH range. LipA showed high tolerance to some organic solvents, including methanol, acetone and isopropanol. So LipA has the potential applications in the detergent industry and enzyme mediated organic synthesis and could be used to understand the mechanism of acidophily and alkaliphily of enzymes. This study demonstrated that marine sponges are an important source for new enzymes.

Acknowledgments This work was supported by the National Natural Science Foundation of China (NSFC) (31000062).

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