# ORIGINAL PAPER

# Separation and purification of bioactive botrallin and TMC-264 by a combination of HSCCC and semi-preparative HPLC from endophytic fungus Hyalodendriella sp. Ponipodef12

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Abstract Two dibenzo- $\alpha$ -pyrones, botrallin (1) and TMC-264 (2) were preparatively separated from crude ethyl acetate extract of the endophytic fungus Hyalodendriella sp. Ponipodef12, which was isolated from the hybrid 'Neva' of Populus deltoides Marsh  $\times$  P. nigra L. using a combination of high-speed counter-current chromatography (HSCCC) and semi-preparative HPLC. Botrallin (1) with 74.73 % of purity and TMC-264 (2) with  $82.29\%$  of purity were obtained through HSCCC by employing a solvent system containing n-hexane–ethyl acetate–methanol–water at a volume ratio of 1.2:1.0:0.9:1.0. It was the first time for TMC-264 (2) to be isolated from this fungus. TMC-264 (2) showed strong antimicrobial and antinematodal activity, and botrallin (1) exhibited moderate inhibitory activity on acetylcholinesterase.

Keywords Dibenzo-a-pyrones - Botrallin - TMC-264 - Endophytic fungus · Hyalodendriella sp. Ponipodef12 · High-speed counter-current chromatography - Biological activities

## Introduction

Botrallin (1) and TMC-264 (2), belonging to the class of dibenzo-a-pyrones, have been found in some fungal species

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(Kameda et al. [1974](#page-9-0); Sakurai et al. [2003a](#page-9-0); Hormazabal et al. [2005](#page-9-0); Meng et al. [2012;](#page-9-0) Mao et al. [2014\)](#page-9-0). Botrallin (1) was firstly isolated from the fungus Botrytis allii (Kameda et al. [1974](#page-9-0)), and later it was isolated from the endophytic fungi Microsphaeropsis olivacea (Hormazabal et al. [2005](#page-9-0)) and Hyalodendriella sp. Ponipodef12 (Meng et al. [2012](#page-9-0)), respectively. Botrallin exhibited a moderate inhibitory activity on acetylcholinesterase (AChE) (Hormazabal et al. [2005](#page-9-0)), and a strong inhibitory activity on tyrosine kinase (Holler et al. [1999](#page-9-0)). TMC-264 (2) was first isolated from the fermentation broth of Phoma sp. TC 1674, and was found to selectively inhibit interleukin-4 (IL-4) signaling by interfering phosphorylation of the signal transducer and activator of transcription 6 (STAT6), as well as binding of the phosphorylated STAT6 to the recognition sequence. So it might be an inhibitor of IL-4 signaling for the treatment of allergic disease (Sakurai et al. [2003a,](#page-9-0) [b\)](#page-9-0). These tremendous discoveries about botrallin (1) and TMC-264 (2) have attracted our attention to further investigate their biological activities as well as to efficiently obtain them in large amounts.

High-speed counter-current chromatography (HSCCC), a support-free liquid chromatography method, provides an advantage over the conventional column chromatography by eliminating the use of a solid support where an amount of stationary phase is limited, and the dangers of irreversible adsorption from the support are inevitably present (Chen et al. [2003;](#page-8-0) Lu et al. [2003;](#page-9-0) Ito [2005](#page-9-0)). Compared to other liquid–liquid techniques, HSCCC is advantageous because of its shorter separation time, wider range of selection of solvent systems and quantitative material recovery. It is now accepted as an efficient preparative technique, and widely used for separation and purification of various natural products (Chen et al. [2006;](#page-8-0) Wang et al. [2010](#page-9-0); Feng et al. [2013;](#page-8-0) Jin et al. [2013](#page-9-0); Mudge et al. [2013](#page-9-0); Shan et al. [2013](#page-9-0)).

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To the best of our knowledge, there was no information available about the metabolite production from the fungi Hyalodendriella speices except for our previous study about three dibenzo-a-pyrones (i.e., palmariol B, alternariol 9-methyl ether, and botrallin) isolated from the endophytic fungus Hyalodendriella sp. Ponipodef12 associated with the healthy stems of poplar hybrid 'Neva' of Populus deltoides Marsh  $\times$  P. nigra L. (Meng et al. [2012\)](#page-9-0). In this report, we described the isolation and structural identification of another dibenzo-a-pyrone, named TMC-264 from the endophytic fungus Hyalodendriella sp. Ponipodef12, and found that this fungus was a high producer of botrallin and TMC-264. The conventional methods for the separation of dibenzo-a-pyrones including botrallin and TMC-264 from the crude extracts of the fungi were performed by liquid–liquid extraction, followed by multistep silica gel column chromatography (Sakurai et al. [2003a,](#page-9-0) [b](#page-9-0); Meng et al. [2012\)](#page-9-0). However, these methods are not suitable for large-scale preparation due to the need of huge amounts of organic solvents and time consuming. The purpose of the study was to establish a method for more efficient separation and purification of botrallin and TMC-264 from the endophytic fungus Hyalodendriella sp. Ponipodef12 by HSCCC to meet the future applications of these two compounds. The separated botrallin and TMC-264 were further purified by semi-preparative HPLC. Their bioactivities, including antibacterial, antifungal, antinematodal and acetylcholinesterase inhibitory properties, were evaluated in order to provide data supporting the development and utilization of Hyalodendriella sp. Ponipodef12.

#### Materials and methods

#### General analytical methods

The preparative HSCCC system was a TBE-300B (Tauto Biotech, Shanghai, China) instrument equipped with a polytetrafluoroethylene (PTFE) tube (diameter of tube was 2.6 mm, and total capacity was 280 ml) composed of three coils and a 20-ml sample loop. The HSCCC system was equipped with a TBP-5002 pump and TBD-2000 UV detector (Tauto Biotech, Shanghai, China) operating at 254 nm, and a WH500-USB workstation (Wuhao, Shanghai, China). The experimental temperature was at  $25^{\circ}$ C adjusted by HX-1050 constant temperature circulating implement (Boyikang, Beijing, China). The revolution speed of the apparatus can be regulated with a speed controller in the range between 0 and 1,000 rpm.

The semi-preparative HPLC system consisted of one K-501 pump, one K-2501 UV detector (Knauer, Berlin, Germany), one reversed-phase Luna  $C_{18}$  column (250 mm  $\times$  10 mm, 5 µm, Phenomenex, Torrance, CA,

USA), one 2-ml sample loop and a workstation (Lumtech, Beijing, China). The semi-preparative HPLC column was isocratically eluted with MeOH-H<sub>2</sub>O (60:40, v/v) in 50 min at a flow rate of 3.0 ml/min and UV detection at 234 nm. The sample injection volume was 1.0 ml.

Analytical reversed-phase HPLC–DAD data were obtained using an HPLC system (Shimadzu, Kyoto, Japan) consisted of two LC-20AT solvent delivery units, a CBM-20Alite system controller, an SIL-20A autosampler, and an SPD-M20A photodiode array detection (DAD) system. Chromatographic separations were performed at  $40^{\circ}$ C using a reversed-phase Ultimate TM XB  $C_{18}$  column (250 mm  $\times$  4.6 mm, 5 µm, Welch Materials, Inc., Ellicott, MD, USA). The mobile phase composed of methanol– water (60:40, v/v) was eluted at a flow rate of 1.0 ml/min, and the effluent was monitored at 234 nm. The LC solution multi-PDA workstation was employed to acquire and process chromatographic data.

NMR spectra were recorded on a Bruker Avance DRX-400 NMR spectrometer  $(^1H$  at 400 MHz and  $^{13}C$  at 100 MHz) using tetramethylsilane (TMS) as the internal standard, and chemical shifts were recorded as  $\delta$  values. HR–ESI–MS spectra were measured on Bruker Apex IV FTMS. The melting points (uncorrected) of the compounds were determined on an XT4-100B microscopic meltingpoint apparatus (Tianjin Tianguang Optical Instruments Company, Tianjin, China). UV spectra were recorded with a TU-1810 UV–VIS spectrophotometer (Beijing Purkinje General Instrument Company, Beijing, China). A microplate spectrophotometer (PowerWave HT, BioTek Instruments, Winooski, VT, USA) was employed to measure the light absorption value.

# Chemicals and reagents

All organic solvents used for HSCCC and preparative HPLC were of analytical grade and purchased from Beijing Chemical Company, Beijing, China. Methanol used for analytical HPLC was of chromatographic grade and purchased from Tianjin Tianhao Chemical Company, Tianjin, China. Water was purified by a Milli-Q system (TTL-30C, Tongtai, Beijing, China). Amphotericin B and 3-(4,5- Dimethylthiazol-2-yl)-2,5-dephenyl tetrazolium bromide (MTT) were purchased from Amresco (Solon, OH, USA). Streptomycin sulfate, carbendazim, acetylcholinesterase (AChE), acetylthiocholine iodide (ATCI), 5,5'-dithio bis-(2-nitrobenzoic acid) (DTNB), and 9-amino-1,2,3,4-tetrahydroacridine hydrochloride hydrate were purchased from Sigma-Aldrich (USA). Avemectin B1, which was kindly provided by Dr. Shankui Yuan at the Institute for the Control of Agrochemicals, Chinese Ministry of Agriculture, was used as the positive control with the purity of 97.2 %. It was a mixture of avermectin B1a and avermectin

<span id="page-2-0"></span>B1b in the ratio of 9.5–0.5 (w/w). All other chemicals and reagents were of analytical grade.

#### Preparation of the crude sample

The endophytic fungus Hyalodendriella sp. Ponipodef12 (GenBank accession number HQ731647) was isolated from the healthy stems of the 'Neva' hybrid of Populus deltoides Marsh  $\times$  P. nigra L. in our previous study (Zhong et al. [2011\)](#page-9-0). It was stored both on PDA slants at  $4^{\circ}$ C and in 40 % glycerol at  $-70$  °C in the Herbarium of the College of Agronomy and Biotechnology, China Agricultural University (Beijing, China).

The fungus was stored on PDA slants at  $4^{\circ}$ C, and grown on PDA plates at 25  $\degree$ C for 8–10 days before being used. Then, 4–5 plugs of agar medium (0.5 cm  $\times$  0.5 cm) with fungal cultures were inoculated in each 250-ml Erlenmeyer flask containing 100 ml potato dextrose broth (PDB) medium, and incubated on a rotary shaker at 150 rpm and 25  $^{\circ}$ C for 7 days. The endophytic fungus strain Ponipodef12 was cultivated in a 500-ml Erlenmeyer flask, containing 100 g of rice (purchased from the local grocery store) and 100 ml of distilled water. The rice medium was autoclaved at 121  $^{\circ}$ C for 30 min. After sterilization, the medium was inoculated and mixed with the mycelium pellets, and incubated statically at 25 $\degree$ C for 45 days.

At the end of the incubation period, the cultures were harvested, dried, and ground. The dry materials (136 g) were extracted with MeOH (3  $\times$  300 ml). After filtration, the filtrate was concentrated under vacuum at 50  $^{\circ}$ C, the brown residue was suspended in water (100 ml) and fractionated successively with petroleum ether  $(3 \times 100 \text{ ml})$ , ethyl acetate (3  $\times$  100 ml) and *n*-butanol (3  $\times$  100 ml), to give their corresponding fractions. A total of 400 mg concentrated ethyl acetate crude fraction with the target compounds was obtained.

#### Solvent system for HSCCC

The two-phase solvent systems containing different ratios of n-hexane, ethyl acetate, methanol and water were prepared (Table 1). The partition coefficients  $(K$  values) of the pure compounds (botrallin and TMC-264) were determined by HPLC and calculated according to the ratio of the compound concentrations in the stationary phase and mobile phase according to the method described by Ito  $(2005)$  $(2005)$ .  $K =$  $C_{\text{stationary phase}}/C_{\text{mobile phase}}$ , where  $C_{\text{stationary phase}}$  is the compound concentration in the stationary phase (upper phase), and  $C_{\text{mobile phase}}$  is the compound concentration in the mobile phase (lower phase). As the volumes of the upper and lower phases of the pre-equilibrated two-phase solvent system are equal, the  $K$  values are then determined according the ratio of the peak area,  $K = A_{stationary phase}/A_{mobile phase}$ ,

Table 1 The partition coefficient  $(K$  values) of botrallin  $(1)$  and TMC-264 (2) in the two-phase solvent systems of *n*-hexane–ethyl acetate–methanol-water by HPLC analysis

No.	Ratio $(v/v)$	$K$ value				
		Botrallin (1)	TMC-264 $(2)$			
1	1.0:1.0:1.0:1.0	0.29	1.17			
$\overline{c}$	1.5:1.0:1.0:1.0	0.41	1.00			
3	1.0:1.2:1.0:1.5	1.01	5.79			
$\overline{4}$	1.0:1.0:1.0:1.5	0.45	1.71			
5	1.2:1.0:0.9:1.0	0.73	1.68			
6	1.0:1.0:0.9:1.0	1.04	2.20			
7	1.5:1.0:1.0:1.5	0.33	2.26			
8	1.0:1.5:1.0:1.0	0.66	2.14			

"Ratio" is expressed as the volume ratio of  $n$ -hexane–ethyl acetate– methanol-water. Botrallin (1) and TMC-264 (2) correspond to the peaks I and II, respectively in Fig. [1](#page-5-0)

where  $A_{\text{stationary phase}}$  is the peak area of the target compound in the stationary phase (upper phase), and  $A_{\text{mobile phase}}$  is the peak area of the target compound in the mobile phase (lower phase). Each  $K$  value was an average of three replicates. The separation factor ( $\alpha = K_2/K_1, K_2 > K_1$ ) for two compounds was also obtained (Wen et al. [2009](#page-9-0)). The HPLC analysis was performed using a reversed-phase  $C_{18}$  column (250 mm  $\times$ 4.6 mm, 5  $\mu$ m) at 40 °C. The elution system consisted of solvent A (MeOH) and solvent B  $(H<sub>2</sub>O)$  at the volume ratio of 60:40 (v/v), elution was done at 1.0 ml/min. UV detection was at 234 nm.

Preparation of the two-phase solvent system and sample solution

The two-phase solvent system composed of  $n$ -hexane–ethyl acetate–methanol-water (1.2:1.0:0.9:1.0, v/v) was used for HSCCC separation. Each solvent was added to the separatory funnel according to the volume ratios and thoroughly equilibrated at room temperature overnight. The two-phases were separated and degassed by sonication prior for 20 min to use.

The sample solution was prepared by dissolving 300 mg crude extract in a 30 ml of the mixture of upper and lower phase (1:1, v/v) of the solvent system used for HSCCC separation.

## HSCCC separation procedure

The upper phase was used as the stationary phase and the lower phase was used as the mobile phase in the tail-to-head elution mode. In each separation, the multilayer-coiled column was first filled entirely with the stationary phase. The mobile phase was then pumped into the tail end of the inlet column at a flow rate of 3.0 ml/min, while the apparatus was rotated at 820 rpm. After the mobile phase front emerged and hydrodynamic equilibrium was established in the column, 10.0 ml of the sample solution containing 100 mg of the crude extract was injected. This process was repeated for three times, and the same fractions were combined and concentrated. During the separation process, the column temperature was controlled at 25 $\degree$ C, and the effluent of the column was continuously monitored with a UV detector at 254 nm. Each peak fraction was collected according to the chromatogram. The retention ratio  $(S_F)$  of the stationary phase was calculated from the volume of the stationary phase collected from the column after the separation, and the equation was as follows:

$$
S_{\rm F}(\%) = \frac{V_{\rm T} - V_{\rm S}}{V_{\rm T}} \times 100
$$

where  $S_F$  is the retention ratio of stationary phase (upper phase), and  $V<sub>S</sub>$  is the volume (i.e., 97 ml) of stationary phase flowing out.  $V_T$  is the total volume (i.e., 280 ml) of the coils containing three multilayer coil separation columns (Ito [2005\)](#page-9-0).

## Analysis and identification of HSCCC peak fractions

The ethyl acetate crude extract and peak fractions separated by HSCCC were analyzed by HPLC. The analyses were performed with a reversed-phase  $C_{18}$  column at 40 °C. The mobile phase composed of methanol–water (60:40, v/v) was eluted at a flow rate of 1.0 ml/min and the effluent monitored at 234 nm. Then each peak fraction separated by HSCCC was further purified by semi-preparative HPLC with a reversed-phase  $C_{18}$  column eluted at a flow rate of 3.0 ml/min and monitored at 234 nm.

The purified and identified botrallin (1) and TMC-264 (2) were used as the standards. The linear equation of botrallin (1) by HPLC analysis was  $Y = 5.71711 \times 10^6 X$  $-$  85792.5 ( $R^2$  = 0.9997), and that of TMC-264 (2) was  $Y = 1.62393 \times 10^7 X - 137986 (R^2 = 0.9981)$ , where Y is the peak area,  $X$  is quality ( $\mu$ g) of the sample injected for each time, and  $R$  is the correlation coefficient. The physicochemical and spectrometric data of two compounds were given as follows.

Botrallin (1). Yellow needle-like crystals (MeOH); m.p. 223–227 °C; UV (MeOH)  $\lambda_{\text{max}}$ : 234, 266 nm; The molecular formula  $C_{16}H_{14}O_7$  was assigned by HR–ESI– MS  $m/z$  319.0814  $[M + H]$ <sup>+</sup> (calcd. for C<sub>16</sub>H<sub>15</sub>O<sub>7</sub>, 319.0813); <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  (ppm): 6.25 (1H, s, OH-1), 3.67 (3H, s, OCH3-3), 6.17 (1H, s, H-4), 1.69 (3H, s, CH<sub>3</sub>-4a), 11.35 (1H, s, OH-7), 6.84 (1H, d,  $J = 2.36$  Hz, H-8), 3.92 (3H, s, OCH<sub>3</sub>-9), 7.64 (1H, d,  $J = 2.36$  Hz, H-10); <sup>13</sup>C NMR (DMSO- $d_6$ )  $\delta$  (ppm): 141.6 (C-1), 172.1 (C-2), 146.8 (C-3), 55.1 (OMe-3), 125.1 (C-4), 68.5 (C-4a),

29.2 (CH3-4a), 163.6 (C-6), 101.1 (C-6a), 163.5 (C-7), 102.5 (C-8), 165.9 (C-9), 56.2 (OCH<sub>3</sub>-9), 106.9 (C-10), 135.1 (C-10a), 130.0 (C-10b). After comparing the data with those reported in the literature (Kameda et al. [1974](#page-9-0); Hormazabal et al. [2005](#page-9-0)), the compound was identified as 1,7-dihydroxy-3,9-dimethoxy-4a-methyl-6H-dibenzo[b,d] pyran-2,6-(4aH)-dione, named botrallin.

*TMC-264* (2). Yellow amorphous powder (CHCl<sub>3</sub>); m.p. 164–166 °C; UV (MeOH)  $\lambda_{\text{max}}$ : 237, 267 nm; The molecular formula C<sub>16</sub>H<sub>13</sub>Cl O<sub>7</sub> was assigned by HR-ESI–MS  $m/z$ 353.04291 [M + H]<sup>+</sup> (calcd. for C<sub>16</sub>H<sub>14</sub>ClO<sub>7</sub>, 353.04226); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm): 1.89 (3H, s, CH<sub>3</sub>-1), 3.35 (1H, br s, OH-1), 3.95 (3H, s, OCH<sub>3</sub>-3), 11.15 (1H, s, OH-7),  $6.58$  (1H, d,  $J = 2.16$  Hz, H-8),  $3.92$  (3H, s, OCH<sub>3</sub>-9), 7.60  $(H, d, J = 2.16 \text{ Hz}, H-10);$ <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm): 72.3 (C-1), 29.2 (CH<sub>3</sub>-1), 146.0 (C-2), 146.7 (C-3), 60.7 (OCH3-3), 172.0 (C-4), 141.5 (C-4a), 163.7 (C-6), 101.1  $(C-6a)$ , 164.5  $(C-7)$ , 103.4  $(C-8)$ , 166.8  $(C-9)$ , 56.1  $(OCH<sub>3</sub>-9)$ , 107.1 (C-10), 134.1 (C-10a), 128.9 (C-10b). After comparing the data with those reported in the literature (Sakurai et al. [2003a](#page-9-0), [b\)](#page-9-0), the compound was determined to be 2-chloro-4,6 dihydro-1,7-dihydroxy-3,9-dimethoxy-1-methyl-1H-diben $z \circ b \circ d$  pyran-4,6-dione, named TMC-264. The stereochemistry of C-1 remains to be determined.

#### Biological activity

### Antibacterial activity assay

Four Gram-negative (Xanthomonas vesicatoria ATCC 11633, Ralstonia solanacearum ATCC 11696, Pseudomonas lachrymans ATCC 11921, and Agrobacterium tumefaciens ATCC 11158), and two Gram-positive (Bacillus subtilis ATCC 11562 and Staphylococcus haemolyticus ATCC 29970) bacteria were selected for the antibacterial activity assay. DMSO was used as the negative control, and streptomycin sulfate was used as the positive control. A modified broth dilution-colorimetric assay using the chromogenic reagent 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide (MTT) was used to detect the antibacterial activity of the compounds (Langfied et al. [2004](#page-9-0)). This assay was described in detail in our previous report (Meng et al. [2012](#page-9-0)). The median inhibitory concentration  $(IC_{50})$  was obtained using the linear relation between the inhibitory probability and concentration logarithm (Sakuma [1998](#page-9-0)). The  $IC_{50}$  values were calculated from three individual determinations. The percentage  $(\%)$  of bacterial growth inhibition was determined by the formula:

$$
[(Ac - At)/Ac)] \times 100
$$

where  $A_c$  was an average of three replicates of the light absorption at 510 nm of the negative controls, and  $A_t$  was

an average of three replicates of the light absorption at 510 nm of the samples. The median inhibitory concentration  $(IC_{50})$  was calculated using the linear relation between the inhibitory probability and concentration logarithm.

#### Antifungal activity assay

Antifungal activity assay was similar to that of antibacterial activity which was described in detail in our previous report (Meng et al. [2012\)](#page-9-0). Two fungal species Magnaporthe oryzae P131 and Candida albicans ATCC 2538 were selected for antifungal activity. For M. oryzae P131, the negative control was 5 % acetone, and the positive control was carbendazim. For C. albicans ATCC 2538, the negative control was DMSO, and the positive control was amphotericin B. The  $IC_{50}$  values were calculated from three individual determinations. The percentage of spore germination inhibition was determined by the expression:

Spore germination inhibition  $(\%)$  $= [(Gc - Gt)/Gc] \times 100$ 

where  $G_c$  is an average of three replicates of germinated spore number in the negative control, and  $G_t$  is an average of three replicates of germinated numbers in the treated sets. The  $IC_{50}$  value calculation for the spore germination inhibition was the same as that for antibacterial activity assay.

# Acetylcholinesterase inhibitory activity assay

The assay was carried out as described by Ellman et al. [\(1961](#page-8-0)) using a microplate spectrophotometer with some modifications. In the 96-well plates, 50  $\mu$ l of pH 7.4 phosphate buffered saline (PBS) solution,  $10 \mu l$  of  $1.0 \text{ U/ml}$ enzyme (AChE) solution,  $10 \mu l$  of a serially diluted solutions of the compounds, and 9-amino-1,2,3,4-tetrahydroacridine hydrochloride hydrate as the positive control were added. After the mixture was maintained at  $37^{\circ}$ C for 15 min, 20 µl of acetylthiocholine iodide (ATCI) solution was added. After the mixture was maintained at 37  $^{\circ}$ C for another 15 min, 30  $\mu$ l of 1.0 % SDS solution and 30  $\mu$ l of DTNB solution, the absorbance was measured at 415 nm immediately. pH 7.4 PBS solution was used as the negative control. The percentage (%) of acetylcholinesterase inhibitory activity was determined using the following expression:

$$
[(Ac - At)/Ac)] \times 100
$$

where Ac was an average of three replicates of the light absorption at 415 nm of the negative controls, and At was an average of three replicates of the light absorption at 415 nm of the samples. The  $IC_{50}$  value calculation for the acetylcholinesterase activity was the same as that for antibacterial activity assay.

#### Antinematodal activity assay

This assay was carried out as described by Meng et al. [\(2012](#page-9-0)) with some modifications. Three nematode species (i.e., Bursaphelenchus xylophilus, Caenorhabditis elegans, and Panagrellus redivivus) were tested for antinematodal activity of the two compounds. 5 % acetone–water solution was used as the negative control. Avemectin B1 was used as the positive control. Five replicates were carried out for each treatment. The nematodes were considered to be dead when they did not move by treating with a fine needle as the physical stimuli. The mean percentage of mortality was then calculated, and the  $IC_{50}$  value calculation for the antinematodal activity was the same as that for antibacterial activity assay.

## Results and discussion

Separation and structural identification of botrallin and TMC-264

The crude ethyl acetate extract from endophytic fungus Hyalodendriella sp. Ponipodef12 was first analyzed by reversed-phase HPLC (Fig. [1](#page-5-0)). Two main peaks (i.e., peaks I and II) were satisfactorily separated with methanol–water (60:40, v/v) as the solvent system. Accordingly, two compounds, which correspond to peaks I and II, respectively in Fig. [1,](#page-5-0) were obtained from the ethyl acetate extract by repeated column chromatography over normal and reverse silica gel. After comparing their physicochemical and spectrometric data with those reported in the literature (Sakurai et al. [2003b;](#page-9-0) Hormazabal et al. [2005](#page-9-0)), they were identified as botrallin (1) and TMC-264 (2), whose chemical structures are shown in Fig. [2.](#page-5-0) Both compounds belong to dibenzo-a-pyrones. This is the first time for TMC-264 (2) to be isolated from the endophytic fungus Hyalodendriella sp. Ponipodef12 as well as being obtained as natural product for the second time. In this work, rice medium was employed instead of potato dextrose broth (PDB) used in our previous study (Meng et al. [2012](#page-9-0)) to lead to difference of the secondary metabolites. This is an interesting topic about biosynthetic regulation of the dibenzo- $\alpha$ -pyrones in endophyitc fungus Hyalodendriella sp. which needs to be investigated in detail.

Selection of two-phase solvent system for HSCCC

Successful separation by HSCCC largely depends upon the selection of a suitable two-phase solvent system, and the two-phase solvent system was selected according to the partition coefficients  $(K$  values) of each target component and an optimum range of K value should be from  $0.5$  to  $2.0$  <span id="page-5-0"></span>Fig. 1 HPLC chromatogram of the ethyl acetate extract from the endophytic fungus Hyalodendriella sp. Ponipodef12. The retention times of peaks I and II were 7.38 and 10.40 min, respectively





Fig. 2 The structures of botrallin (1) and TMC-264 (2)

(Han et al.  $2012$ ). A smaller K value elutes the solute closer to the solvent front with lower resolution while a larger  $K$  value tends to give better resolution but broader peaks due to a longer elution time (Ito [2005](#page-9-0)). In addition, the value of separation factor  $(\alpha)$  between two compounds should be higher than 1.5 (Wen et al. [2009](#page-9-0)).

In this experiment, the  $K$  values for botrallin  $(1)$  and TMC-264 (2) in several two-phase solvent systems containing n-hexane–ethyl acetate–methanol-water were measured and shown in Table [1.](#page-2-0) Among these solvent systems, the most appropriate  $K$  values were 0.73 for botrallin  $(1)$  and  $1.68$  for TMC-264  $(2)$  at a ratio of 1.2:1.0:0.9:1.0 (v/v), which was selected to further isolate and purify the two target compounds from the crude extract of the endophytic fungus Hyalodendriella sp. Ponipodef12 by HSCCC in the present study. Under this condition, the separation factor  $(\alpha)$  of the two compounds was calculated as 2.30 which should be acceptable.

Separation of botrallin and TMC-264 from crude extract by HSCCC

Under the optimized condition with a solvent system containing n-hexane–ethyl acetate–methanol-water at a volume ratio of 1.2:1.0:0.9:1.0, about 100 mg of the crude ethyl acetate extract was separated by HSCCC for each time. Two fractions (i.e., the fractions of peaks I and II shown in Fig. 3) of 25 ml each were collected according to the HSCCC chromatogram over 2 h. This process was



<span id="page-6-0"></span>Fig. 4 HPLC chromatograms of the HSCCC fractions and purified compounds. a peak fraction I from HSCCC; b peak fraction II from HSCCC; c botrallin (1) purified from peak fraction I by semipreparative HPLC; d TMC-264 (2) purified from peak fraction II by semi-preparative HPLC



Microorganism	Botrallin (1)		TMC-264 $(2)$		$CK^+$	
	MIC (µg/ml)	$IC_{50}$ ( $\mu$ g/ml)	MIC (ug/ml)	$IC_{50}$ ( $\mu$ g/ml)	MIC (µg/ml)	$IC_{50}$ ( $\mu$ g/ml)
A. tumefaciens ATCC 11158	150	$81.65 \pm 4.25$	50	$11.12 \pm 2.07$	50	$21.01 \pm 1.41$
B. subtilis ATCC 11562	200	$104.27 \pm 4.93$	25	$13.89 \pm 2.05$	50	$27.32 \pm 2.62$
P. lachrymans ATCC 11921	200	$99.06 \pm 4.01$	25	$5.89 \pm 0.91$	50	$18.23 \pm 1.68$
R. solanacearum ATCC 11696	150	$86.39 \pm 3.87$	25	$16.49 \pm 2.14$	50	$22.30 \pm 2.48$
S. haemolyticus ATCC 29970	200	$91.62 \pm 4.19$	25	$15.94 \pm 1.85$	50	$22.95 \pm 2.17$
X. vesicatoria ATCC 11633	150	$83.17 \pm 3.77$	50	$15.39 \pm 2.02$	50	$16.55 \pm 2.10$
M. oryzae P131	200	$97.51 \pm 4.25$	10	$7.53 \pm 0.42$	20	$6.25 \pm 0.19$
C. albicans ATCC 2538	200	$112.31 \pm 4.71$	75	$55.55 \pm 4.98$	50	$31.56 \pm 3.54$

Table 2 Antimicrobial activity of botrallin (1) and TMC-264 (2)

MIC minimum inhibitory concentration,  $IC_{50}$  median inhibitory concentration,  $CK^+$  positive controls for bacteria, M. oryzae and C. albicans were streptomycin sulfate, carbendazim and amphotericin B, respectively

repeated for three times, and the same fractions were combined and concentrated. The retention ratio  $(S_F)$  of the ethyl acetate crude extract in the stationary phase was 65.36 %. The HSCCC chromatogram is shown in Fig. [3](#page-5-0).

The collected HSCCC fractions were concentrated and further analyzed by HPLC which gave the chromatograms shown in Fig. [4.](#page-6-0) The fractions I (28.5 mg) and II (23.5 mg) were analyzed to contain botrallin with 74.73 % of purity and TMC-264 with 82.29 % of purity, respectively. The fractions with the retention time from 24 to 45 min in HSCCC were analyzed by HPLC as other minor compounds in the crude extract. The two main dibenzo- $\alpha$ pyrones prepared by HSCCC were further purified by semipreparative HPLC. Finally, 17.0 mg of botrallin (1) and 14.8 mg of TMC-264 (2) were purified from *Hyaloden*driella sp. Ponipodef12 using a combination of HSCCC and semi-preparative HPLC. Though it was efficient to prepare these two compounds with the method of HSCCC and semi-preparative HPLC, we should admit that the method in this work did not save the organic solvents. In recent years, the elution-extrusion, gradient elution, ondemand solvent preparation modes in HSCCC have been developed and applied to the preparation of natural products (Pereira da Silva et al. [2009](#page-9-0); Sethi et al. [2009](#page-9-0); Wu et al. [2012\)](#page-9-0). These techniques should be favorable for our subsequent work.

## Biological activities

# Antimicrobial activity

Both botrallin (1) and TMC-264 (2) were tested with microplate-MTT colorimetric assay for their antifungal and antibacterial activity shown in Table 2. TMC-264 (2) exhibited stronger antimicrobial activity than botrallin (1). In a general way, the halogens favor biological properties of the compounds (Muller et al. [2007](#page-9-0); Hagmann [2008\)](#page-8-0). It is

Table 3 Inhibitory activity of botrallin (1) and TMC-264 (2) on acetylcholinesterase



MIC minimum inhibitory concentration,  $IC_{50}$  median inhibitory concentration,  $CK^+$  positive control for acetylcholinesterase inhibitory activity was 9-amino-1,2,3,4-tetrahydroacridine hydrochloride hydrate

possible that chlorine atom in TMC-264 attributes to the strong antimicrobial activity that remains to be confirmed.

#### Inhibitory activity on acetylcholinesterase

The inhibitory activity of botrallin (1) and TMC-264 (2) on acetylcholinesterase (AChE) is shown in Table 3. The  $IC_{50}$ values were 58.04  $\mu$ g/ml for botrallin (1) and 78.80  $\mu$ g/ml for TMC-264 (2), respectively. The inhibitory activity of botrallin (1) on AChE was in agreement with the previous report (Hormazabal et al. [2005](#page-9-0)). One possible approach for the treatment of Alzheimer's disease is the restoration of the level of acetylcholine (ACh) through the inhibition of AChE with the inhibitors (Huang et al. [2013\)](#page-9-0). Both botrallin (1) and TMC-264 (2) from the endophyte  $Hyalo$ dendriella sp. Ponipodef12 might be as the potential candidates for the treatment of Alzheimer's disease.

#### Antinematodal activity

Antinematodal activity of botrallin (1) and TMC-264 (2) on the nematodes Bursaphelenchus xylophilus, Caenorhabditis elegans, and Panagrellus redivivus by using the microplate assay expressed as the median inhibitory concentration  $(IC_{50})$  values are depicted in Table [4](#page-8-0).

<span id="page-8-0"></span>Table 4 Antinematodal activity of botrallin (1) and TMC-264 (2)



MIC minimum inhibitory concentration,  $IC_{50}$ , median inhibitory concentration,  $CK^+$  positive control for antinematodal activity was avermectin  $B<sub>1</sub>$ 

As shown in Table 4, both compounds had moderate antinematodal activity on three nematodes. For each nematode, TMC-264 (2) showed stronger antinematodal activity than botrallin (1). This phenomenon is similar to that of their antimicrobial activity. It is possible that the stronger antinematodal activity of TMC-264 (2) might be due to the presence of chlorine atom in its structure which should be further verified. The mechanisms of inhibitory effects on acetylcholinesterase, bacteria, fungi, and nematodes of these two compounds also need to be further investigated.

The  $IC_{50}$  values of botrallin (1) for biological activity assay in this work were different from those in our previous report (Meng et al. [2012\)](#page-9-0). The phenomena may be attributed to the modified methods as well as the differences of experimental conditions in this study.

## **Conclusions**

Endophytic fungi are potential sources of bioactive natural compounds for their application in medicine, agriculture, and food industry (Strobel [2003](#page-9-0); Guo et al. 2008; Zhou et al. [2010](#page-9-0); Kharwar et al. [2011](#page-9-0); Zhao et al. [2011;](#page-9-0) Gutierrez et al. 2012; Kumar and Kaushik [2012](#page-9-0)). The preparative separation of natural compounds from endophytic fungi by classical methods is tedious, time consuming, and requires multiple chromatographic steps (Shan et al. [2013](#page-9-0)). In this work, two dibenzo- $\alpha$ -pyrones, botrallin (1) and TMC-264 (2) were successfully obtained from crude ethyl acetate extract of the endophytic fungus Hyalodendriella sp. Ponipodef12 using a combination of HSCCC and semipreparative HPLC. This is the first report on the application of HSCCC for the preparative separation of dibenzo- $\alpha$ pyrones from the cultures of endophytic fungus Hyalodendriella sp. Ponipodef12. The present study will provide a basis for a large preparation of dibenzo-a-pyrones, and also demonstrates that HSCCC is an efficient technique in preparatively separating bioactive compounds from fungi.

TMC-264 (2) was isolated from the endophytic fungus Hyalodendriella sp. Ponipodef12 for the first time as well as being obtained as the natural product for the second time. TMC-264 (2) showed strong antimicrobial and antinematodal activity, and botrallin (1) exhibited moderate inhibitory activity on acetylcholinesterase, which indicates these two compounds have their potential applications. To the best of our knowledge, this is the first report on the antimicrobial, antinematodal and acetylcholinesterase inhibitory activities of TMC-264 (2). The results indicate the potential of the endophytic fungus Hyalodendriella sp. Ponipodef12 as a source of active dibenzo- $\alpha$ -pyrones. Some issues such as the mechanisms of action of botrallin (1) and TMC-264 (2), isolation of the other active compounds from this fungus, and efficient strategies for increasing content and yield of botrallin and TMC-264 in fermentation culture of Hyalodendriella sp. Ponipodef12 need to be further investigated.

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