The Performance of Aerobic Granular Sludge Under Diferent Aeration Strategies at Low Temperature

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Received: 10 September 2021 / Accepted: 12 January 2022 / Published online: 28 January 2022 © The Author(s), under exclusive licence to Springer Nature Switzerland AG 2022

Abstract Aeration strategy is an important factor for the formation and maintenance of aerobic granular sludge (AGS), but aeration is also the most energyconsuming part in the biological wastewater treatment system. In order to optimize the aeration strategy of AGS reactor at low temperature, short- and long-term effects of dissolved oxygen (DO) concentration and aeration intensity (AI) were investigated at 10 ℃ in this study. The results showed that the carbon and phosphorus removal performance of AGS exhibited high resistance to the short-term changes of DO

Supplementary Information The online version contains supplementary material available at [https://doi.](https://doi.org/10.1007/s11270-022-05506-y) [org/10.1007/s11270-022-05506-y](https://doi.org/10.1007/s11270-022-05506-y).

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and AI, while the nitrogen removal was greatly infuenced. The optimum DO and AI were 4 mg/L and 0.25 cm/s, corresponding to 82.7% and 81.4% of total inorganic nitrogen removal efficiencies, respectively. Long-term operation experiment showed that the properties of AGS kept stable under 4 mg/L DO concentration, but the overgrowth of flamentous bacteria and reduction of extracellular polymeric substance under 0.25 cm/s AI led to a large amount of granule disintegration, which could not be recovered with the prolonged operation. These fndings might provide guidance for the operation optimization of AGS system at low temperature.

Keywords Aerobic granular sludge · Low temperature · Dissolved oxygen · Aeration intensity · Nutrient removal

1 Introduction

Aerobic granular sludge (AGS) is an efficient biological wastewater treatment technology, which is formed by self-immobilization of microorganisms under high selective pressure (Nguyen Quoc et al., [2021](#page-9-0)). Compared with the traditional activated sludge process, AGS has the advantages of compact structure and excellent sedimentation performance, which is conductive to improving the biomass concentration of the reactors (Zitomer et al., [2007\)](#page-9-1). In addition, the special layered structure due to the difusion limitation of dissolved oxygen (DO) in AGS makes it possible to simultaneously remove nitrogen, phosphorus, and organic matter in a single system (Mosquera-Corral et al., [2005](#page-9-2)). To date, AGS technology has been successfully applied in the treatment of municipal sewage and industrial wastewater containing toxic substances, showing high pollutant removal rate, strong impact load resistance, and low operating cost (Pronk et al., [2015](#page-9-3); Sadri Moghaddam & Alavi Moghaddam, [2016;](#page-9-4) Wei et al., [2016\)](#page-9-5).

The aeration strategy is an important factor that infuences the formation and performance of AGS, by providing dissolved oxygen (DO) and shear force (Gao et al., [2013](#page-9-6); de Sousa Rollemberg et al., [2018](#page-8-0)). It has been confrmed that the cultivation and stable operation of AGS was easier to realize under high DO concentration and high shear force due to the high secretion rate of extracellular polymeric substance (EPS) (Muhammad Sajjad [2015](#page-9-7)). However, Mosquera-Corral et al. [\(2005](#page-9-2)) found that the nitrogen removal efficiency in AGS reactor increased from 8 to 45% while reducing the saturation of DO from 100 to 40%. This system ran stably for 150 days under 75% DO saturation, but collapsed soon under 40% DO saturation, which was speculated to be caused by the degradation of EPS in the core of granules. Gao et al. [\(2013](#page-9-6)) studied the impact of aeration intensity (AI) on mature AGS and found that the apparent half rate constant and the COD removal rate were similar under high and low AI. In considering that aeration is the main energy consumption part in biological wastewater treatment processes, it is necessary to optimize the aeration strategy, not only to improve the pollutant removal efficiency, but also to reduce the wastewater treatment cost. However, most relative studies were carried out at room temperature. Few studies pay attention to the optimization of aeration strategies at low temperature, in spite of the fact that the same aeration strategy can not be applied to room and low-temperature environments at the same time, because of the diference in the DO transfer rate, water viscosity, and microbial activity (Chen et al., [2017;](#page-8-1) Winkler et al., [2012\)](#page-9-8).

To optimize the aeration strategy of AGS system at low temperature, the present study investigated the short- and long-term efects of DO concentration and AI on the performance of AGS at 10 ℃, respectively. The aeration strategy was frst determined by batch tests and then substituted into the reactor for verification. The nutrient removal efficiency, granule morphology, and EPS content were simultaneously detected to identify the operational stability of AGS under diferent aeration strategies. It is expected to provide useful information for further reducing the energy consumption of AGS reactors.

2 Materials and Methods

2.1 Reactor Setup

Long-term experiments were carried out in a cylindrical sequencing batch reactor (R0). The efective working volume was 3.0 L, with an inner diameter of 7.4 cm and a height of 70 cm. Compressed air was introduced via a bubble stone at the bottom of the reactor. The original aeration fux was 160 L/h, which led to an AI of 1.0 cm/s and a DO concentration of 10.5 mg/L during aeration period. The operation temperature was controlled at 10 ± 0.5 °C by a precise cryogenic thermostat throughout the experiment. A 4-h operation cycle was applied, which consisted of 60-min feeding, 152-min aeration, 3-min settling, 5-min effluent withdrawal, and 20-min idle time. The volumetric exchange ratio was 50%, corresponding to a hydraulic residence time of 8 h. The excess sludge was discharged regularly to control the sludge retention time at 30 days.

2.2 Inoculum and Wastewater

The reactor was seeded with mature aerobic granules collected from a lab-scale AGS-SBR system (Xu et al., [2018](#page-9-9)), which has been operated stably for over 300 days at $10+0.5$ °C. The mixed liquor suspended solids (MLSS) concentration of the reactor after inoculation was about 6.0 g/L. The feed to the reactor was synthetic wastewater simulating municipal sewage, which was consisted of the following: COD (NaAc) 400 mg/L, NH_4^+ -N (NH₄Cl) 50 mg/L, TP (K₂HPO₄ and KH_2PO_4) 18 mg/L, Mg^{2+} ($MgSO_4$) 8 mg/L, KCl 32.5 mg/L, and 1 mL/L of trace element solution (described by Beun et al. (2000) (2000)).

2.3 Analytical Methods

The wastewater parameters, including COD, NH_4^+ -N, NO_2^- -N, NO_3^- -N, TP, MLSS, and MLVSS, were measured according to standard method (APHA, [2005\)](#page-8-3). DO and pH were detected by HQ 30d and HI 8424 pH meters. Thermal extraction was utilized to extract EPS from sludge (Li & Yang, [2007](#page-9-10)). The contents of protein (PN) and polysaccharide (PS) in EPS were measured using modifed Lowry method (Frølund et al., [1996\)](#page-8-4) and phenol–sulfuric acid method (Gerhardt et al., [1994](#page-9-11)), respectively. The extracted EPS was qualitatively analyzed by FP-6500 fuorescence spectrophotometer (3D-EEM). The excitation wavelength was 220–450 nm and the emission wavelength was 220–650 nm. The morphology of AGS was obtained using digital camera and stereomicroscope (TM-2M45-B2).

2.4 Batch Test

After R0 (described above) had been operated stably, batch tests were conducted to investigate the shortterm efect of DO and AI on AGS performance. At the beginning of each test, mixed liquor (500 mL) was collected from R0 at the end of aeration, and then transferred to a minifed version SBR reactor with the same working volume. The minifed SBR was operated at the same condition (except for DO and AI) with R0 for 2 days. Samples were taken out regularly at the last cycle to investigate the conversion process of the pollutants. When studying the efect of DO, AI was kept steady while introducing nitrogen gas $(N_2, 99\%$ purity) into the reactor. DO was firstly aerated to the target concentrations (2, 4, 6, and 8 mg/L) within 10 min and then kept steady by changing the air to N_2 ratio, which was accomplished by DO detector, mass fow controller, and high-pressure solenoid valve. When studying the efect of AI, the aeration fux was adjusted to maintain the AI at 0.25, 0.5, and 0.75 cm/s. The variation of DO during the last cycles was also detected to identify the infuence degree of DO and AI respectively.

3 Results and Discussion

3.1 Start-up of AGS-SBR Reactor

Due to the same operating conditions with source reactor, R0 entered a stable state rapidly. As shown in Table [1,](#page-2-0) the average diameter of AGS at phase I was about 4.5 mm, with a settling velocity of 75 m/h. The PN and PS contents in EPS were 79.30 and 26.86 mg/g VSS, respectively, corresponding to a PN/ PS ratio of 2.95. In order to further clarify the components of EPS in AGS, 3D-EEM was conducted. As shown in Fig. [1a](#page-3-0), 4 peaks were detected, representing humic acid–like substances (peaks A and C), tryptophan protein (peak B), and aromatic protein (peak D) (Ni et al., [2017](#page-9-12)). The high fluorescence intensity of tryptophan indicated that the granules might exhibit high hydrophobicity, which has been proved favorable for promoting the cell-to-cell interaction and then improving the stability of AGS (Campo et al., [2018](#page-8-5)).

In terms of pollutant removal performance, R0 achieved nearly 92.5% and 99.5% removal efficiencies of COD and TP at the beginning of operation, due to the high concentration of inoculated AGS. As shown in Fig. 2 , the effluent TIN concentration was about 20 mg/L, corresponding to 60% TIN removal efficiency. The nutrient removal processes in a typical cycle at day 30 was tested. The results showed that the anaerobic phosphorus release after feeding was 68 mg/L (Fig. $S1a$). During aerobic period, the phosphorus accumulating organisms (PAO) accumulated phosphorus rapidly. Almost all phosphorus was removed within 100 min of aeration. Nitrogen conversion process is illustrated in Fig. S1b. It can be seen that the concentration of NH_4^+ -N in liquid phase was only 14.5 mg/L rather than 25.0 mg/L. Considering that no anaerobic ammonium-oxidizing bacteria were detected in the source SBR, the loss of NH_4^+ -N might be caused by the adsorption of AGS, which has

Table 1 Characteristics of AGS at diferent operating periods

Operating day	DO (mg/L)	AI (cm/s)	Diameter (mm)	Settling velocity (m/h)	SVI (mL/g)	EPS		
						PN (mg/gVSS)	PS (mg/gVSS)	PN/PS
30	10.5	1.0	4.5	72	48	79.30	26.86	2.95
60	4.0	0.1	4.3	67	53	70.65	24.14	2.93
90	5.5	0.25	2.2	15	127	43.23	20.27	2.13

Fig. 1 3D-EEM of EPS at diferent operating periods: **a** phase I; **b** phase II; **c** phase III

been confrmed in previous study (He & Xu, [2018](#page-9-13)). The trend of NO_3^- -N concentration shows that the traditional denitrifying bacteria might play less role

Fig. 2 Effluent concentrations of NH_4^+ -N, NO_2^- -N, NO_3^-N , and TIN removal efficiency

in the nitrogen removal process, since the $NO₃⁻-N$ concentration in the liquid phase was 10.2 mg/L after feeding, nearly no reduction occurred, indicating that mostly of the NO_3^- -N was reduced during aeration period by denitrifying polyphosphate-accumulating organism (DPAO).

3.2 Short-Term Efect of Aeration Strategies

3.2.1 DO

DO plays an important role in the process of simultaneous nitrogen and phosphorus removal in AGS system. Under the same other conditions, the aerobic zone in AGS increases with the increase of DO concentration, and the corresponding anoxic zone and anaerobic zone decrease accordingly (Chiu et al., [2007](#page-8-6)). Therefore, appropriate DO concentration is helpful to adjust the functional zone of AGS so as to improve the removal efficiency of pollutants.

The removal processes of nitrogen and phosphorus in the last cycle of each batch experiment under diferent DO concentrations is shown in Fig. [3.](#page-4-0) It

Fig. 3 Conversion of nitrogen and phosphorus in the last cycle under diferent DO concentrations: **a** 2 mg/L; **b** 4 mg/L; **c** 6 mg/L; **d** 8 mg/L; **e** 10.5 mg/L; **f** phosphorus

can be observed that DO concentration had a signifcant efect on the conversion process of nitrogen, especially nitrite oxidation and denitrifcation. When DO concentration decreased from 10.5 to 2 mg/L, the accumulation rate of $NO₃⁻-N$ in liquid phase decreased gradually, and the concentration of NO_3^- -N in effluent decreased from 20 to 3.04 mg/L, with a slight increase of NO_2^- -N (0–0.5 mg/L). Linear fitting of NO_3^- -N concentration during aeration (Table S1) shows that when DO concentration was above 8 mg/L, the nitrite oxidation rate had no signifcant change, for the nitrate accumulation rate was both 0.12 mgNO_3^- -N·L⁻¹ min⁻¹. NH₄⁺-N could be efectively removed in the studied DO concentrations, except for 2 mg/L, at which NH_4^+ -N in effluent increased to about 9 mg/L. The reason for this result can also be explained from the regulation of DO on AGS functional zoning: functional fora related to nitrogen removal in AGS mainly distributed in the peripheral aerobic and anoxic zones, with $AOB \rightarrow NOB \rightarrow$ denitrifying bacteria from surface to inert core (Guimarães et al., [2017](#page-9-14)). The penetration depth of DO in AGS decreased alongside liquid DO concentration, resulting in the compression of aerobic zone and the outward migration of anoxic zone. The NOB was afected frstly while the oxidation rate of NO_2^- -N decreased. More NO_2^- -N was directly reduced to N_2 rather than oxidized to NO_3^- -N. In addition, due to the short-cut denitrifcation process which required fewer carbon source, more NOx could be reduced under the same COD concentration, thereby improving the TIN removal efficiency (Yang & Yang, 2011). The highest TIN removal efficiency in this experiment was 82.7%, which was obtained at 4 mg/L DO. This was consistent with the partial nitrifcation study of Wang et al. ([2020\)](#page-9-16), in which the ammonium oxidation rate was kept stable when the DO concentration was greater than 4 mg/L at 15℃. When DO concentration decreased to 2 mg/L, the aerobic zone decreased dramatically and the activity of AOB was inhibited, resulting in the residue of NH_4^+ -N. Therefore, although the effluent NO_3^- -N concentration continued to decline, the removal efficiency of TIN decreased accordingly.

In terms of phosphorus removal, AGS system showed strong resistance to the change of DO, for more than 96% of phosphorus removal could be obtained at each cycle (as shown in Fig. [3f](#page-4-0)). In AGS systems, the biological removal of phosphorus is mainly performed by PAO and DPAO. Unlike traditional PAO, DPAO can use NO_2^- -N or NO_3^- -N instead of DO as electronic donors to accumulate phosphorus under anoxic conditions (Lee & Yun, [2014\)](#page-9-17). Considering the performance of NO_3^- -N reduction, DPAO might account for a large proportion in the total microbial community, which could adapt to various DO environments. It is noteworthy that although phosphorus could be removed adequately, its accumulation rates decreased with the decrease of DO concentration. This might be due to the compression of aerobic zone and the outward migration of anoxic zone in granules, which led to the decrease of phosphorus uptake rate of traditional PAO. It can also be found that when DO concentration in the system was at a low level, NO_3^- -N decreased continuously in the frst 40 min of aeration. Considering that there was only a small amount of carbon source residue in the system at this period, the cause for this phenomenon might be the faster denitrifying phosphorus removal rate than the nitrite oxidation rate.

3.2.2 AI

The effect of AI on AGS is not only reflected in promoting foc aggregation and maintaining granular stability, but also closely related to the removal rate of pollutants (He et al., [2018\)](#page-9-18). As reported, the diffusion coefficients of NaAc, DO, and ammonium are 2.5×10^{-9} , 1.67×10^{-9} , and 1.01×10^{-9} m² s⁻¹, respectively (Liu & Liu, [2006](#page-9-19)). The mass transfer resistance of substrates in AGS is much higher than that in liquid phase, so a higher AI is needed to improve the mass transfer rate of substrates (Zhu et al., [2015\)](#page-9-20). Figure S2 shows the variation of DO concentration in typical cycles under diferent AI. It can be seen that the rising rate of DO concentration increased with the increase of AI, which might be related to the increase of the gas–liquid oxygen transfer rate (Garcia-Ochoa & Gomez, [2009\)](#page-9-21). Data also show that when AI was 1.0 and 0.75 cm/s, DO in the systems could both reach saturation state, but 0.75 cm/s AI took longer time to achieve. When AI was 0.5 and 0.25 cm/s, DO in the system was increased throughout the aeration period, which could reach 8.45 and 5.94 mg/L eventually.

The conversion processes of nitrogen and phosphorus are shown in Fig. [4.](#page-6-0) It can be seen that although the changes of AI had little effect on ammonium

Fig. 4 Conversion of nitrogen and phosphorus in the last cycle under diferent AI: **a** 0.25 cm/s; **b** 0.5 cm/s; **c** 0.75 cm/s; **d** phosphorus

oxidation rate and nitrate accumulation rate due to the high DO concentration (> 5.9 mg/L), the NO_3^- -N concentration in effluent decreased gradually. The TIN removal efficiency increased from 60 to 81.4% while the AI decreased from 1.0 to 0.25 cm/s. This result might be caused by the increase of denitrifying rate at the frst 20 min of aeration, for the concentration of NO_3^- -N continued to decrease at this period. In previous research, the variation of DO concentration was considered to be the main reason for the efect of AI on the pollutant removal of AGS (He et al., [2019\)](#page-9-22). However, when comparing the efects of DO and AI at the same time, it can be observed that even if DO concentration was close, the denitrifcation capacity of the system was diferent under diferent AI. Such as Fig. [3d](#page-4-0) and Fig. [4b,](#page-6-0)

the fnal DO concentrations of the two systems were 8 and 8.5 mg/L, but the concentrations of $NO₃⁻-N$ in the effluent were 17.33 and 12.54 mg/L, and the corresponding TIN removal efficiency was 65.34% and 74.92%, respectively. There might be two reasons for this phenomenon. One of them is the increase rate of DO was different between these two conditions. In the experiment of DO efect, the DO in system was not controlled until it reached the desired concentration. That is to say, the duration of DO concentration below 8 mg/L in aeration stage was less than 5 min. However, in the AI efect experiment, DO did not reach 8 mg/L until 90 min of aeration, resulting in higher denitrifcation rate in the majority aeration time than the former. The other reason might be that the mass transfer rate under DO experiment

was higher, so the solid–liquid surface of AGS could catch more DO and substrate than low AI system. Therefore, DO could penetrate deeper in AGS under high AI, leading to the decrease of the anoxic zone and denitrification efficiency.

In the aspect of phosphorus removal, the efect of AI showed similar trends with DO, that is, the phosphorus accumulation rate decreased with the AI. It is noteworthy that in the DO efect experiments, the aerobic phosphorus uptake rate had little change before DO concentration dropped to 4 mg/L, and the phosphorus could be fully removed within 160 min. However, in the AI efect experiments, the phosphorus uptake rate slowed down obviously with the decrease of AI, and the time of the complete removal of phosphorus was delayed to 180, 200, and 220 min, respectively, even though the DO concentration was higher (Fig. [4d\)](#page-6-0). This phenomenon indicated that the mass transfer rate controlled by AI might have a greater impact on the pollutant removal than DO.

3.3 Long-Term Efect of Aeration Strategies

The long-term effects of DO and AI on AGS were sequenced conducted under 4.0 mg/L DO concentration and 0.25 cm/s AI, which have been proved to be the most conducive condition for the simultaneous removal of pollutants by short-term experi-ments, respectively. As shown in Fig. [2](#page-3-1) (phase II, days 30–60), the removal performance of COD, NH_4^+ -N, and TP remained stable at this stage, while the average TIN removal efficiency increased to 82% soon after the reduction of DO, accompanied by the significant decrease of effluent $NO₃⁻-N$. Granules could be divided into two types: white and pale yellow. Diferent from the study conducted under room temperature, in which AGS soon disintegrated due to the overgrowth of flamentous bacteria when the DO saturation decreased from 100 to 40% (Mosquera-Corral et al., [2005\)](#page-9-2), the reduction of DO in this study had little efect on the stability of AGS, but the white to yellow ratio changed obviously. The succession of microbial community structure might be the main reason for the diferent granular morphology. As reported by Barr et al. ([2010\)](#page-8-7), the white granules in AGS system was dominated by *Candidatus Accumulibacter*, which is a typical DPAO, while the yellow granules was dominated by *Candidatus Competibacter*, which is a typical denitrifying

glycogen-accumulating organisms (DGAO). It has been confrmed that PAOs have an advantage over GAOs at low DO levels, as PAOs have a higher DO affinity (Carvalheira et al., [2014\)](#page-8-8). Therefore, the change of white to yellow ratio might be due to the enrichment of DPAO. The content of PN and PS in EPS also kept stable at this period (Table [1](#page-2-0)). However, the 3D-EEM spectra shows that the fuorescence intensity of humic acid–like substances increased signifcantly, indicating that the degree of cell lysis inside granules increased with the increase of anaerobic zone (Burdon, [2001](#page-8-9)).

In practice, high AI in biological reactor operation usually results in huge energy consumption. Therefore, reducing AI can not only improve nutrient removal, but also reduce operating costs. During days 60–90, the long-term efect of AI on AGS system was investigated without DO controlling. As shown in Fig. [2](#page-3-1) (phase III), the performance of AGS could be divided into two phase. The frst was days 60–70, during which granules kept stable and no obvious change occurred. TIN removal efficiency decreased gradually with the recovery of NOB activity, which might be due to the increase of DO concentration. However, from day 70, flamentous bacteria began to accumulate in the system and quickly occupied a dominant position. The surface of AGS turned into fufy and the settling speed decreased to 15 m/h within 5 days, which directly led to a large amount of sludge discharged with the effluent. It has been reported that low temperature was not favorable for the amplifcation of flamentous bacteria, due to a higher temperature coefficient than floc-forming bacteria (Liu & Liu, [2006\)](#page-9-19). However, the special layered structure of AGS and the decrease of mass transfer rate increased the substrate gradient inside granules (including DO), which might provide a suitable environment for flamentous bacteria (Castellanos et al., [2021\)](#page-8-10). At the end of phase III, a signifcant granule disintegration occurred, which directly caused the collapse of the system. As a chain reaction, the nutrient removal performance deteriorated rapidly. Ammonium oxidation rate and nitrite oxidation rate decreased simultaneously, which caused the accumulation of NH_4^+ -N and NO_2^- -N. The TIN, COD, and TP removal efficiencies dropped to 55% , 78% , and 65% respectively, which could not be recovered with the prolong of operation. It is noteworthy that, during studying the efect of AI on AGS system at room temperature, granules exhibited similar performance under 0.58 and 0.14 cm/s (Gao et al., [2013](#page-9-6)). In addition, Yuan and Gao ([2010\)](#page-9-23) found that the optimal DO (achieved by reducing AI) for AGS was 2.5 mg/L at 30℃. However, in this study, the fuidization of AGS was poor when the AI was below 0.25 cm/s, which might be due to the higher water viscosity at low temperature. Besides, the inhibitation of low temperature on functional microorganisms could also lead to the different performance of AGS. In order to figure out the main reason for system collapse except for flamentous bacteria overgrowth, the characteristics of EPS at day 90 were investigated. It can be seen that the PN and PS contents in EPS decreased 38.81% and 16.03% compared to day 60, which resulted in a 27.3% decrease of PN/PS. The fuorescence intensity of tryptophan and aromatic-like proteins decreased accordingly, and the peak E, which belonged to soluble microbial products, appeared at this period. As described by Franca et al. ([2018\)](#page-8-11), the degradation of EPS could cause the formation of cavity structure and reduce the stability of AGS. Besides, a lower EPS secretion rate under low AI has also been confrmed (He et al., [2019](#page-9-22)). Therefore, the decrease of EPS content, especially PN, might also be a main reason for the collapse of AGS while reducing AI. How to select the optimized AI that can simultaneously improve the nutrient removal ability and maintain the stability of AGS at low temperature needs further research.

4 Conclusion

 In this study, short- and long-term efects of aeration strategies on AGS at low temperature were investigated. The results showed that the AGS had strong resistance to the short-term changes of DO and AI. The optimized conditions were 4 mg/L DO and 0.25 cm/s AI, corresponding to 82.7% and 81.4% of TIN removal efficiencies, respectively. The longterm reduction of DO had little effect on the performance of AGS. However, the reduction of AI caused the excessive growth of flamentous bacteria and the decrease of EPS content, which eventually led to the disintegration of granules. These results suggest that the relatively high AI and low DO might be more suitable for AGS operation, which could be realized through adjusting the aeration mode, such as changing microporous aerator to perforated aerator.

Funding This research was supported fnancially by the Key Research and Development Program of Shandong Province (No. 2020CXGC011202) and the Natural Science Foundation of Shandong Province (No. ZR2021QE274).

Data Availability The data that support the findings of this study are available from the corresponding author upon reasonable request. All data generated or analyzed during this study are included in this published article and its supplementary information fles.

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