

Effects of Titanium Dioxide Nanoparticles in Different Metabolic Pathways in the Freshwater Microalga *Chlorella sorokiniana* (Trebouxiophyceae)

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Abstract The products that employ nanoparticles (NPs) in their composition have increased since the beginning of NP production; hence, their availability in the environment, especially in aquatic ecosystems, tends to increase. In these ecosystems, the phytoplankton is immersed in a complex matrix of nutrients, excreted materials, and other chemical compounds, which can influence the metabolic strategy of microalgae. One of the metabolic ways is mixotrophy, a situation whereby microalgae perform photosynthesis and use dissolved organic carbon at the same time. Most toxicity evaluations do not consider such a metabolic route, but this can represent a preferential metabolism in natural environments. The present study aimed at evaluating the effects of NP-TiO₂ at a log concentration range of - 3.10 to 0.89, on photosynthesis, growth, viability, and biochemical composition of the microalgae *Chlorella sorokiniana* during photoautotrophic and mixotrophic growth (glucose as the organic carbon source). The results showed lower chlorophyll *a* and photosynthetic activity in mixotrophy than in photoautotrophy, which can be due to a decreased need for photosynthesis in

mixotrophy. Photoautotrophy cultures were sensitive to NPs, reaching 39% of viability at log 0.89, while in mixotrophy, cell viability was not affected by NPs. The biochemical composition and cell density changed as a function of NP concentrations, with increase in the protein/carbohydrate ratio in both treatments. The results showed that *C. sorokiniana* is more resistant to NPs during mixotrophic growth, but with changes in biochemical composition, whereas the photoautotrophic cultures were more sensitive to the increase in NP concentrations.

Keywords TiO₂ · Biochemical composition · Phytoplankton · Mixotrophy · Cell viability

1 Introduction

Nanoparticles, defined as particles less than 100 nm in size in more than one dimension (Nowack and Bucheli 2007; Navarro et al. 2008; Yang et al. 2012a, b), are widely recognized as having versatile applications in a variety of areas as textiles, electronics, pharmaceuticals, cosmetics, anti-fouling paints, food products, and environmental remediation (Navarro et al. 2008; Cardinale et al. 2012; Melegari et al. 2013). The global investments in NPs increased from US\$ 10 billion in 2005 to US\$ 1 trillion since 2011, and the production increased from 10,000 t in 2004 to 88,000 t per year after 2010 (Navarro et al. 2008; Sharma 2009; Barreto and Lombardi 2016).

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Among the most abundant NPs manufactured, titanium dioxide (TiO₂) nanoparticles are the second highest globally produced, approximately 3000 t annually (Gottschalk et al. 2009; Zhu et al. 2011; Piccinno et al. 2012; Yang et al. 2012a, b), and with an estimated production of 2.5×10^6 tons per year until 2025 in the USA (Robichaud et al. 2009). This high production is associated with the photocatalytic activity, induced by UV light, of this NP-TiO₂, which has been used in paints, solar technologies, pharmaceuticals, cosmetics, and sunscreens (Hund-Rinke and Simon 2006; Hartmann et al. 2010; Cardinale et al. 2012; Kulacki and Cardinale 2012). The use of NP-TiO₂ as personal care products (e.g., sunscreens), coating, and paints due to their UV-light absorption efficiency, transparency to visible light that increases with decreasing particle size (Franklin et al. 2007), and environmental contamination seem to be inevitable (Zhu et al. 2011).

Kaegi et al. (2008) reported that runoff under heavy rainfall can contain concentrations as high as 3.5×10^8 nanoparticles per liter, which are discharged into aquatic ecosystems, and a concentration of less than $1 \mu\text{g L}^{-1}$ has been found in river surface waters (Sharma 2009; Dalai et al. 2013). However, once NP-TiO₂ enters aquatic habitats, these tend to interact with ions, organic matter, and organisms, as phytoplankton, but literature is contradictory regarding the toxicity of NPs-TiO₂ (Cardinale et al. 2012). Until now, it is known that the toxicity of NPs is related to their physical and chemical properties, such as particle size, shape, aggregation status, surface coating, and ionization (Nel et al. 2006; Beer et al. 2012; Yang et al. 2012a, b), and the generation of reactive oxygen species (ROS) inside the cells (Kadar et al. 2012).

In aquatic ecosystems, the phytoplankton is responsible for the primary production, using sunlight and inorganic carbon to synthesize energy-rich organic matter, a process called photosynthesis, which sustains the aquatic food web (Reynolds 2006). Due to its role as primary producers in these environments, many researches have been carried out with the objective of evaluating the entry of toxic compounds in the food chain via phytoplankton (Araujo and Souza-Santos 2013; Dalai et al. 2013).

Any change in its metabolic activity caused by toxic compounds can affect the organisms in higher trophic levels as well as the ecosystem as

a whole (Kahru and Dubourguier 2010). For example, Cardinale et al. (2012) studying the effects of NP-TiO₂ in three species of green algae reported a reduction in the primary production in the microalgae *Chlamydomonas moewussii* and *Scenedesmus quadricauda*, but an increase in *Chlorella vulgaris*, while the respiration rate reduced in *C. moewussii*, increased in *C. vulgaris*, and remained constant in *S. quadricauda*.

Many phytoplanktonic microorganisms can assimilate organic matter from the environment at the same time as they perform photosynthesis (Baldisserotto et al. 2014). A metabolic strategy in which organisms can use autotrophy and heterotrophy concomitantly is called mixotrophy (Junttila et al. 2015). During mixotrophy, microalgae can assimilate organic matter by osmotrophy (for example, glucose, glycerol, and organic acids) or phagotrophy (predation of bacteria). Up to now, we did not find any study in the literature that takes into account the effects of NPs during mixotrophic growth of phytoplankton. However, considering that in natural aquatic environments, the final receivers of NP-TiO₂, a myriad of dissolved organic materials are present, and probably mixotrophy occurs within phytoplankton cells, investigations onto the toxicity of NPs under such a metabolic pathway can furnish important information about what actually occurs in the environment.

Most toxicity tests focus on photosynthetic and growth parameters, which may not represent the true toxic potential of the tested agent. To avoid misinterpretation, Tang and Dobbs (2007) and Barreto and Lombardi (2016) suggested the use of more specific parameters, such as biochemical composition that changes in the presence of toxic agents, and cell viability. Cell viability represents how many cells are alive, whereas total cell density shows how many cells there are, regardless of whether they are viable or not.

This work aimed at studying the physiology of *Chlorella sorokiniana* under mixotrophic and photoautotrophic conditions in cells exposed to NP-TiO₂. As physiological parameters, we monitored its growth, photosynthetic activity, cell viability, and biochemical composition. This study is a contribution to the understanding of the effects of NPs on phytoplankton in a condition that can resemble what happens in the environment.

2 Material and Methods

2.1 Characterization of Titanium Dioxide Nanoparticles

The nano-TiO₂ used in this work was acquired from Sigma-Aldrich (CAS No. 13463-67-7) for commercial use and with pre-informed characteristics. However, because these characteristics vary widely, we evaluated some of them (average particle diameter, crystallinity, morphology, specific surface area, and zeta potential), which are reported in Barreto and Lombardi (2016). In synthesis, they reported that the characterization of the NPs-TiO₂ showed crystallinity of 92% anatase and 18% rutile, specific surface area of 45.60 m² g⁻¹, and a zeta potential of 25 mV.

The mean particle diameter was determined by X-ray diffraction (LabX XRD-6000, Shimadzu, Japan); crystallinity by calculating its phases through the Gaussian area; and morphology which was observed in scanning electron microscopy (FEI Inspect F50, USA) and FEI Tecnai G2 (USA) transmission. The specific surface area was obtained in ASAP 2000 (Micromeritics, USA), while the zeta potential was determined using the ZetaPlus Zeta Potential Analyzer (BIC, USA). All the characterization was performed at Nanocharacterization and Interdisciplinary Electrochemistry and Ceramics Laboratories at the Federal University of São Carlos, São Paulo, Brazil.

2.2 Experimental Design

The freshwater microalga *Chlorella sorokiniana* was cultured in 500-mL Erlenmeyer flasks previously coated with a silanization solution to reduce the adsorption of NPs onto the flask walls. A volume of 250 mL of AAP medium (U.S. EPA 2012), with no EDTA, was used for all treatments at initial pH 7.00. The cultures were maintained in controlled conditions of temperature (24 ± 1 °C), light intensity (130 μmol photons m⁻² s⁻¹), and photoperiod (12 h light/12 h dark). Exponentially growing cells (10⁵ cells mL⁻¹ initial inoculum) were exposed for 72 h to the nominal (added) NPs-TiO₂ concentrations: 7.9 × 10⁻⁴ mg L⁻¹ (log - 3.10), 7.9 × 10⁻³ mg L⁻¹ (log - 2.10), 7.9 × 10⁻² mg L⁻¹ (log - 1.10), 7.9 × 10⁻¹ mg L⁻¹ (log - 0.10), and 7.9 mg L⁻¹ (log 0.89), with 5 × 10⁻³ mol L⁻¹ glucose to stimulate mixotrophy, and without glucose (photoautotrophic condition). This glucose concentration is reported in the literature to result in high growth and optimum algae performance (Liang et al. 2009; Kong et al. 2011). The NPs-TiO₂

concentration range was based in Mueller and Nowack (2008) that estimated nanoparticle concentrations in lakes, reporting what is referred to as environmental concentration. Reference cultures (without the addition of NPs) were performed, one without glucose, and the other with glucose. Reference cultures had a natural Ti concentration of 2.64 × 10⁻⁴ mg L⁻¹ (log - 3.58) due to the impurities present in the salts used in its preparation. Following the procedure described in Aruoja et al. (2009), NPs-TiO₂ suspensions were sonicated (Ultrasonic Sonicator, DES500, Brazil) during 30 min prior to use in order to reduce agglomeration and sedimentation of the particles.

After 72 h of exposure to NP-TiO₂, the experimental parameters were determined. The hydrogen ion concentration was determined with a pH meter (Gehaka, PG1800, Brazil), chlorophyll *a* concentration (mg L⁻¹) by in vivo fluorescence using a fluorimeter (Turner Designs, Model Trilogy, USA), and cell density (cell mL⁻¹) and viability (% of viable cells) determined with a cytometer Muse® Cell Analyzer (USA). Specific growth rates (μ) were calculated through a graphic representation of the natural logarithm of chlorophyll *a* concentration per milliliter as a function of time. The linear regression from the straight line (exponential growth phase) was calculated, and the angular coefficient represents the specific growth rate.

The maximum fluorescence of photosystem II was obtained in 20-min dark-adapted cells using a pulse amplitude modulated fluorimeter, PHYTO-PAM (Heinz Walz Effeltrich, Germany). This parameter can be used to infer about the physiological status of photosynthetic microalgae (Lombardi and Maldonado 2011).

Proteins were determined according to the methodology of Bradford (1976) with extraction based in Rausch (1981), while carbohydrates followed the method of Albalasmeh et al. (2013). All biochemical composition data are reported in picograms per cubic micrometer (pg μm⁻³) instead of per cell because mixotrophic cells are at least twice the photoautotrophic ones.

2.3 Data Analysis

This study was run with three experimental replicates for each treatment, and significant differences between means of each variable were tested by one-way ANOVA and Tukey's post hoc analysis using Assisat 7.7 beta software. Graphs were plotted using the software Origin Pro (version 8.5.0).

3 Results

The nanoparticle characterization demonstrated the predominance of sphere and cuboid morphologies with a diameter between 10 and 50 nm, crystallinity of 82% anatase, specific surface area of $45.60 \text{ m}^2 \text{ g}^{-1}$, and zeta potential of 25 mV (Barreto and Lombardi 2016).

The values of pH are shown in Table 1; among the photoautotrophic cultures, treatment without NPs-TiO₂ showed the highest pH value (7.57), whereas in the other treatments, no statistical differences in the pH with the increase in NPs-TiO₂ were observed. Comparing the two types of metabolism, with exception of the treatments without NP, at each concentration, the pH was generally higher in the mixotrophic metabolism than in the photoautotrophic metabolism ($p > 0.05$).

Figure 1 reports that chlorophyll *a* concentration in the photoautotrophic cultures increased with the increase in nanoparticle concentration, reaching the maximum ($1.25 \times 10^{-7} \text{ } \mu\text{g cell}^{-1}$) in $7.9 \times 10^{-1} \text{ mg L}^{-1}$ NP-TiO₂ (log -0.10). However, in the highest NP-TiO₂ concentration tested (7.9 mg L^{-1} ; log 0.89), the chlorophyll *a* concentration decreased to $4.24 \times 10^{-8} \text{ } \mu\text{g cell}^{-1}$. During mixotrophy, there was no change ($p = 0.2047$) in the concentration of nanoparticles, all of which were lower in relation to photoautotrophic cultures.

As expected, the maximum fluorescence of photosystem II (Fig. 2) was higher in photoautotrophy than in mixotrophy, confirming a reduction in photosynthetic activity during mixotrophic growth conditions. NPs-TiO₂ does not appear to affect the photosynthesis in both photoautotrophy and mixotrophy cultures, since no statistically significant differences ($p = 0.085$ for photoautotrophy and $p = 0.0608$ for mixotrophy) were

Table 1 Values of culture pH (72 h NPs-TiO₂ exposed cultures). Values are means \pm standard deviation of three replicates (same letters indicate statistical similarity). Statistical analysis was performed separately for the treatments either photoautotrophy or mixotrophy. Ref = reference culture

Treatment	Photoautotrophy	Mixotrophy
Ref (log -3.58)	7.57 \pm 0.05 ^a	6.25 \pm 0.03 ^b
$7.9 \times 10^{-4} \text{ mg L}^{-1}$ (log -3.10)	6.24 \pm 0.02 ^a	6.54 \pm 0.03 ^b
$7.9 \times 10^{-3} \text{ mg L}^{-1}$ (log -2.10)	6.24 \pm 0.03 ^a	6.89 \pm 0.05 ^b
$7.9 \times 10^{-2} \text{ mg L}^{-1}$ (log -1.10)	6.39 \pm 0.01 ^a	6.92 \pm 0.02 ^b
$7.9 \times 10^{-1} \text{ mg L}^{-1}$ (log -0.10)	6.42 \pm 0.02 ^a	6.99 \pm 0.03 ^b
7.9 mg L^{-1} (log 0.89)	6.55 \pm 0.02 ^a	7.05 \pm 0.04 ^b

detected among the different concentrations tested in this study.

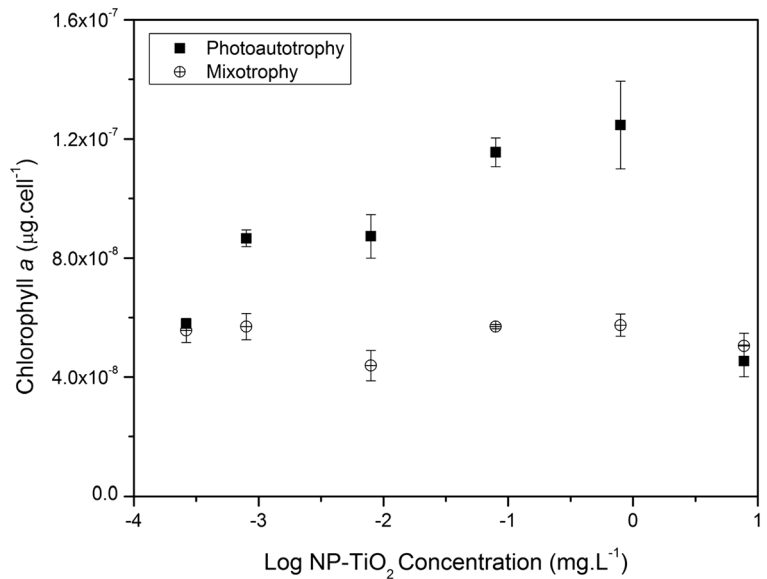
The specific growth rates are shown in Fig. 3. No statistical difference ($p = 0.0079$) was obtained among the treatments during photoautotrophic growth, whereas during mixotrophy, the treatments with nanoparticles showed growth rates smaller than the control with glucose; this was higher than the photoautotrophic control (no NPs-TiO₂ addition).

In photoautotrophic cultures, the 72-h cell density increased in the first two NPs-TiO₂ concentrations (Fig. 4), reaching the maximum value of $5.42 \times 10^6 \text{ cells mL}^{-1}$, decreasing thereafter. This suggests a toxic potential of the NPs-TiO₂ to *C. sorokiniana*. Cell density in the mixotrophic cultures decreased with the increase in NP concentrations with the highest density in the control (with glucose, but without NPs).

Cell viability describes the percentage of live cells in the sample, and as Fig. 5 shows, in the mixotrophic condition, cell viability was not affected by NP-TiO₂, with 97 to 99% of the cells remaining alive. However, in the photoautotrophic cultures, cell viability decreased with the increase in the NPs-TiO₂ concentration ($p < 0.001$), exhibiting 39% of viable cells in the highest concentration tested.

Figure 6 shows the concentration of carbohydrates (Fig. 6a) and proteins (Fig. 6b) in *C. sorokiniana* after 72 h of NP exposure, which are reported per unit cell volume. Considering carbohydrates, no significant differences ($p > 0.05$) among treatments were obtained for the photoautotrophic conditions, except in the highest NP concentration ($7.9 \times 10^{-4} \text{ mol L}^{-1}$; log 0.89), doubling its value in relation to the control ($p < 0.0001$). However, under mixotrophy, a decrease in carbohydrate concentrations was obtained for the three higher concentrations of NPs ($7.9 \times 10^{-2} \text{ mg L}^{-1}$, log -1.10; $7.9 \times 10^{-1} \text{ mg L}^{-1}$, log -0.10; and 7.9 mg L^{-1} , log 0.89). For proteins, the photoautotrophic cultures always had higher values than the mixotrophic condition. In addition, this figure shows that protein synthesis was affected by the NP-TiO₂ concentration, being approximately two times higher in both photoautotrophic and mixotrophic cultures at $7.9 \times 10^{-2} \text{ mg L}^{-1}$ (log -1.10) of NP-TiO₂ and above. The P/C ratios (Fig. 6c) were higher in the photoautotrophic cultures; once again, the highest values were present in the three highest NP concentrations ($7.9 \times 10^{-2} \text{ mg L}^{-1}$ and above).

Fig. 1 Concentration of chlorophyll *a* ($\mu\text{g cell}^{-1}$) in *C. sorokiniana* at 72 h of exposed to NP-TiO₂ in photoautotrophic and mixotrophic conditions. Error bars mean standard deviation from the mean ($n = 3$)



4 Discussion

The reduction in cell viability at $7.9 \times 10^{-1} \text{ mg L}^{-1}$ (log -0.10) NP concentrations under photoautotrophic condition, but no effect under mixotrophy obtained in the present study, is a clear demonstration that different metabolic pathways are differently affected by the NPs. Similar to the present results, Barreto and Lombardi (2016) obtained reduction in cell viability in the green alga *Scenedesmus bijugus* grown under

photoautotrophic metabolism. The authors used the same NPs as we did, which had a predominance of the anatase phase. However, Barreto and Lombardi (2016) obtained a decrease in the photosynthetic yield for *S. bijugus*, but for *C. sorokiniana*, photosynthetic yield was not affected by the NPs.

In photoautotrophic microalgae cultures, pH above 7.00 generally indicates photosynthetic activity by the cells, which removes carbon dioxide, therefore reducing the carbonic acid content and increasing pH (Reynolds

Fig. 2 Maximum fluorescence of photosystem II (F_v/F_m) of microalga *C. sorokiniana* at 72 h exposed to NP-TiO₂ in photoautotrophic and mixotrophic conditions. Error bars mean standard deviation from the mean ($n = 3$)

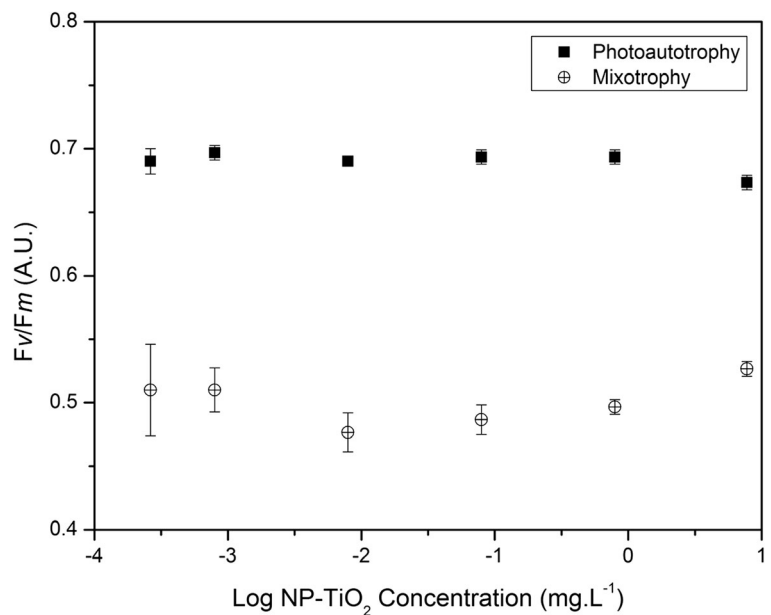
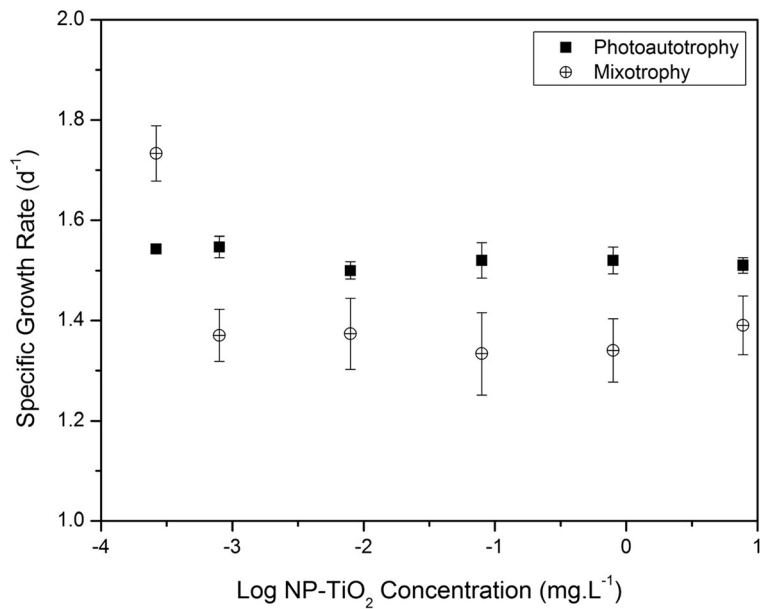


Fig. 3 Specific growth rates (day^{-1}) of microalga *C. sorokiniana* exposed to NPs- TiO_2 in photoautotrophic and mixotrophic conditions. Error bars mean standard deviation from the mean ($n = 3$)



2006), but in mixotrophy, the pH increases due to the symport uptake of protons and glucose (Komor and Tanner 1974; Juntilla et al. 2015). Considering that our initial culture pH was 7.00, values below this in the photoautotrophic cultures indicate that NP- TiO_2 affected the pH of the culture medium, since the cells were doing photosynthesis (Fig. 2).

It is known that the physicochemical surface properties of nanoparticles are dependent on environmental factors, including pH (Navarro et al. 2008; Sharma

2009). Nanoparticles of TiO_2 are expected to have a negative surface charge at $\text{pH} > 7.00$, but a positive surface charge at $\text{pH} < 6.00$ (Ridley et al. 2006). Considering that the cell wall exhibits a negative charge and the pH in this study was around 6.00, we expected that most NPs must be adhered to the cell wall, as described by Navarro et al. (2008).

The increase in chlorophyll *a* concentration in the photoautotrophic cultures can be due to a shading effect caused by the NPs that may have adsorbed onto the cell

Fig. 4 Cell density (cell mL^{-1}) at 72 h of exposed to NPs- TiO_2 in photoautotrophic and mixotrophic conditions. Error bars mean standard deviation from the mean ($n = 3$)

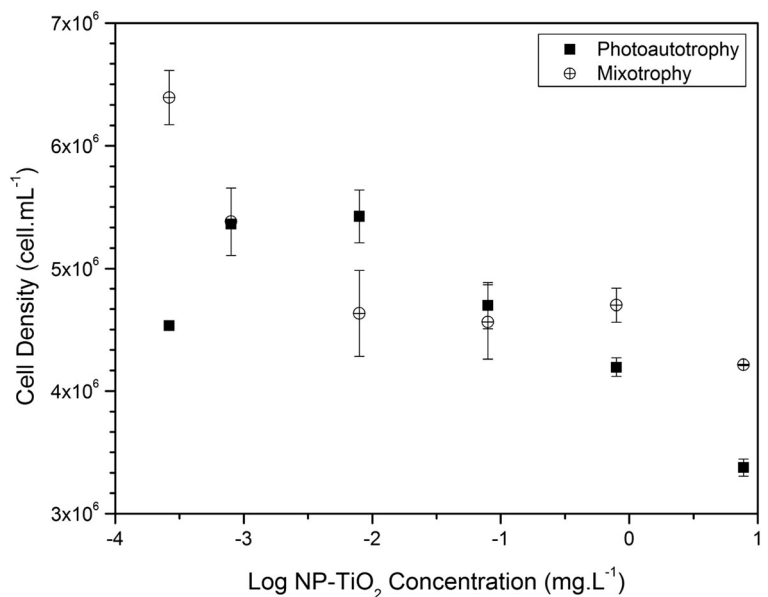
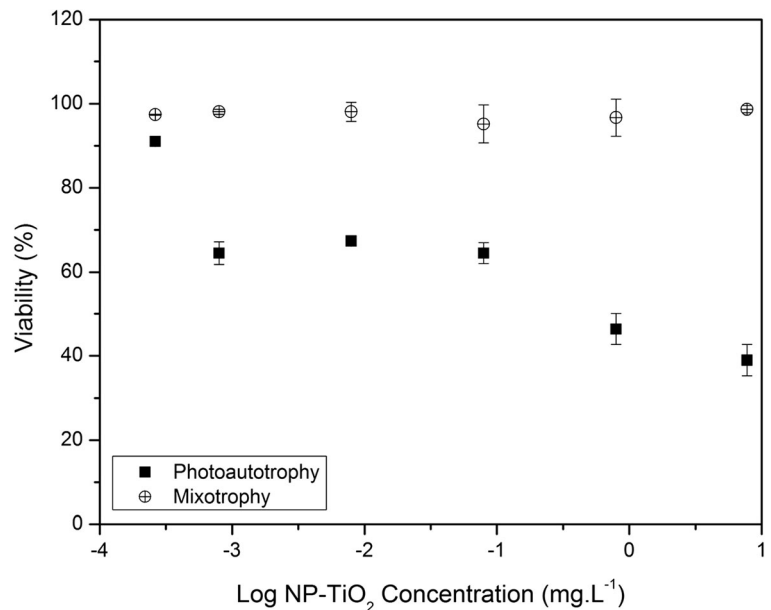


Fig. 5 The percentage of viable cell at 72 h of exposed to NPs-TiO₂ in photoautotrophic and mixotrophic conditions. Error bars mean standard deviation from the mean ($n = 3$)



wall of the microalgae. According to Kulacki and Cardinale (2012), the adhesion of NPs to the microalgae surface reduces the availability of light for each cell in the culture, and this stimulates chlorophyll production by the cell as an attempt to overcome the shading effect (Navarro et al. 2008; Sharma 2009; Cardinale et al. 2012; Melegari et al. 2013). Recalling the photoautotrophic culture pH (6.24–6.55) and the consequent positive charge of the NPs in these cultures together with the negative charge of the cell wall, we can state that in the photoautotrophic growth, the cells would probably be more coated with NPs-TiO₂ than in the mixotrophic condition, whose pH varied within 6.54–7.05. In more neutral pH, the mixotrophic cells would inversely be less coated with NPs and chlorophyll would not increase, which in fact was detected. Even with the nanoparticles reducing light availability to the cell, the photosynthetic activity (F_v/F_m) was not affected, remaining with values close to 0.7, indicating that the algae was healthy, possibly with no stress or nutrient limitation (Kumar et al. 2014). However, the effect of shading in the concentration of 7.9 mg L⁻¹ ($\log = 0.10$) of NPs-TiO₂ could have caused a decline in cell density and chlorophyll *a*.

Differently, in the mixotrophic cultures the chlorophyll *a* concentration and the photosynthetic activity (F_v/F_m around 0.5) together with cell growth results confirmed that *C. sorokiniana* was using another source of energy and carbon (Giovanardi et al. 2014; Juntala

et al. 2015). Consequently, the microalga reduced its need for light to be used in photosynthesis, and produced less chlorophyll *a* than in photoautotrophy (Perez-Garcia et al. 2011; Alkhamis and Qin 2016).

In photoautotrophic cultures, the specific growth rate was not affected by the presence of NPs-TiO₂. This result is in agreement with others in literature. Kulacki and Cardinale (2012) also observed no significant effects of NPs-TiO₂ in 10 phytoplanktonic species belonging to Cyanobacteria, Bacillariophyta, Chlorophyta, and Charophyta. According to Barreto and Lombardi (2016), specific growth rates reflect the general microalgae metabolism and it is not a parameter as sensitive as cell viability to detect the effects of nanoparticles.

This is in contrast to what occurs in mixotrophic cultures with no NPs, where the addition of an organic carbon source stimulates the growth of microalgae and higher growth rates are obtained in comparison with photoautotrophic conditions (Li et al. 2014; Rosemberg et al. 2014; Juntala et al. 2015). In the presence of nanoparticles, the growth rates in the mixotrophic cultures were lower than in the photoautotrophic ones. This difference indicates that the nanoparticles affected the rate of cell division of *C. sorokiniana*, confirming the results of Linkous et al. (2000) and Cardinale et al. (2012). In addition, during mixotrophy, the added nanoparticles can catalyze redox reactions with glucose, making the organic source less available

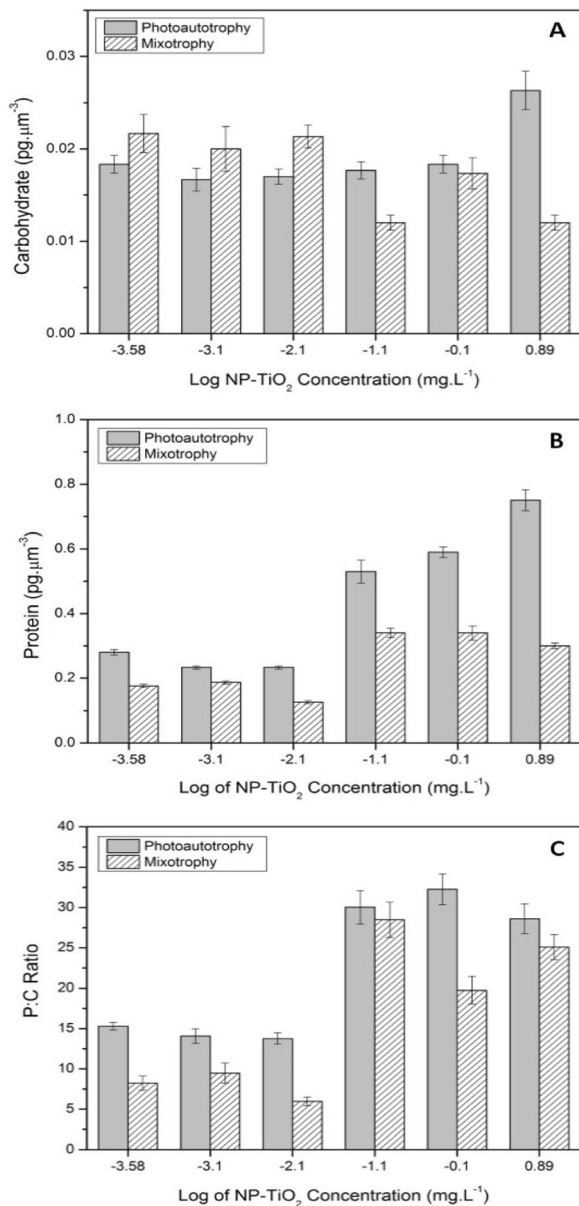


Fig. 6 Biochemical composition of microalga *C. sorokiniana* at 72 h exposed to NP-TiO₂ in photoautotrophic and mixotrophic conditions. **a** Carbohydrate content (pg μm^{-3}). **b** Protein content (pg μm^{-3}). **c** P/C ratio. Error bars means standard deviation from the mean ($n = 3$)

to the microalgae, thus reducing their growth rates (Zhan 2003; Sadiq et al. 2011).

The reduction in total cell density in both photoautotrophic and mixotrophic treatments can be a consequence of chlorophyll *a* reduction due to shading and unavailability of the added glucose due to redox reactions with the nanoparticles (Sadiq et al. 2011; Kulacki

and Cardinale 2012). According to Tang and Dobbs (2007) and Barreto and Lombardi (2016), the use of total cell density to measure toxic effects in microalgae is not as reliable as cell viability. In fact, looking at viable cells, no variation under mixotrophy and a decrease under photoautotrophy conditions were obtained, revealing the higher resistance of *C. sorokiniana* to NPs under mixotrophic metabolism. This observation cannot be made looking at the total cell density data, since it decreased for both metabolic conditions, confirming the findings of Tang and Dobbs (2007) and Barreto and Lombardi (2016) who showed that cell viability, which discriminates live and dead cells, was much more sensitive as a response to the toxic effects of NPs than total cell density.

In the present work, the microalgae cultured in mixotrophy were more resistant to the NPs-TiO₂ than the photoautotrophic cells. In the mixotrophic situation, no changes in the viability occurred and it was kept always close to 100%, albeit the reduced values of specific growth rates under NP exposure. This indicates that cell division was affected, but not the algal health. Whereas in photoautotrophy, as the concentration of NP increased, the percentage of living cells decreased, showing the negative effects of NPs-TiO₂ in the microalgae population. Barreto and Lombardi (2016) observed similar results related to the viability of *S. bijugus* exposed to NPs-TiO₂, and Melegari et al. (2013) related cell viability of *C. reinhardtii* cultivated with copper oxide nanoparticles.

The present results showed that cell viability, considered an expression of algal health, was impaired in photoautotrophic cell in NP-TiO₂ concentration as low as 7.9×10^{-4} mg L⁻¹ (log -3.10) while cell density was only affected in the NP-TiO₂ concentration of 7.9×10^{-2} mg L⁻¹ (log -1.10), emphasizing the need to use the cell viability as a more sensitive parameter to identify the toxicity of nanoparticles.

The differences observed in the content of total intracellular carbohydrates and proteins at concentrations of NPs-TiO₂ from 7.9×10^{-2} to 7.9 mg L⁻¹ (log -1.10 to log 0.89) indicate that the particles affected the microalga *C. sorokiniana*. The increase in carbohydrates in the photoautotrophic cultures at the highest NP-TiO₂ concentration tested (~1.4 times higher at 7.9 mg L⁻¹; log 0.89) can be considered a signal of cellular stress. As reported in literature, the accumulation of storage compounds in microalgae indicates they are facing a physiologically problematic situation, as commonly reported

for trace metal stress. Chia et al. (2013) reported that *Chlorella vulgaris* exposed to Cd (1.12×10^{-2} mg L⁻¹; log -1.95) had its carbohydrate content increased ~3 times in comparison with the controls. Miao et al. (2009) observed an increase in the production of polymeric substances in *Thalassiosira weissflogii*, exposed to silver engineered nanoparticles at concentrations between 1.08×10^{-10} and 1.08×10^{-4} mg L⁻¹ (log -9.96 to log -3.96). Granum et al. (2002) found an increase higher than four times in the carbohydrate content in *Skeletonema costatum* cultured in media with nitrogen depletion; Yang et al. (2012a, b) reported that light intensity and nitrogen concentration had significant effects on polysaccharide production in *Microcystis aeruginosa*. The opposite effect observed for the mixotrophic growth (~1.6 times lower carbohydrates at 7.9×10^{-2} to 7.9 mg L⁻¹; log -1.10 to log 0.89) should also be related to a stressing condition. Cardinale et al. (2012) reported that *Chlorella* spp. (photoautotrophic growth) exposed to NPs-TiO₂ (50 to 300 mg L⁻¹; log 1.69 to log 2.47) showed an increased respiration rate. It is known that mitochondrial respiration consumes carbohydrates, and this can end up resulting in decreased intracellular carbohydrates, as we observed. Nevertheless, we should mention that the respiration rate increase would be expected to occur in both photoautotrophic and mixotrophic metabolisms; thus, further research is needed to understand the differences in carbohydrate metabolic routes under different metabolisms.

The increase in intracellular protein concentration for both mixotrophic and photoautotrophic metabolisms at NP-TiO₂ of 7.9×10^{-2} mol L⁻¹ (log -1.10) and above can be related to a detoxification mechanism, as reported in Miao et al. (2009). These authors hypothesized that microalgae also synthesize proteins as a way of protecting them from the action of nanoparticles, such as phytochelatins for metal ion (Kaplan et al. 1995; Perales-Vela et al. 2006). However, more research is needed to verify this hypothesis. It is known that plants and microalgae can produce peptides that bind metal ions thus decreasing its internal availability for the cell (Kaplan et al. 1995).

The P/C ratio reports on the physiological status of the cell, and values lower than 1 indicate a nutrient-limited environment according to Ganf et al. (1986), Kilham et al. (1997), and Rocha et al. (2015). Therefore, based on the P/C ratios obtained in this research, microalgae were not suffering from nutrient limitation,

evidencing that NPs do not affect the uptake of nutrients by the cells. In addition, the doubled P/C ratio at NP-TiO₂ at 7.9×10^{-2} mg L⁻¹ (log -1.10) and above found in both photoautotrophy and mixotrophy is an indication that the NPs induced the synthesis of proteins in these cultures, and whether this is a matter of detoxification or other cellular compensation process still needs further investigation. It is becoming clear from literature (Barreto and Lombardi 2016) that the effects of the NPs on microalgae vary according to the species and to the NP tested. For example, Barreto and Lombardi (2016) did not find differences in the composition of carbohydrates and proteins or in the P/C ratio in cultures of *S. bijugus* treated with NPs-TiO₂; Cherchi et al. (2015) reported that NPs-TiO₂ induced changes in intracellular composition and nutrient stoichiometry in cultures of Cyanobacteria *Anabaena variabilis*.

5 Conclusion

The present work demonstrated that during mixotrophy, the microalgae *Chlorella sorokiniana* were more resistant to NP-TiO₂ than in photoautotrophy. This conclusion is based on the cell viability parameter, discriminating living from dead cells. This can implicate that under natural environmental conditions, where a myriad of organic substances occur, those species that can benefit from mixotrophy will possibly have higher survival rates than those that cannot. As time goes, a selection and possible reduction of biodiversity can take place. In addition, energy imbalance throughout the aquatic food chain can become a problem since the intracellular biochemical composition was affected by NP-TiO₂ at environmentally important concentrations, as demonstrated by the reduction in the production of proteins during mixotrophy.

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