



# Computational investigation of the selectivity mechanisms of PI3K $\delta$ inhibition with marketed idelalisib: combined molecular dynamics simulation and free energy calculation

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## Abstract

Phosphoinositide 3-kinase (PI3K) has been considered to be a potential drug target for the treatment of several human body diseases. Nowadays, great efforts have been made on the development of selective PI3K $\delta$  inhibitors because of the FDA approval of idelalisib, which is the first listed PI3K inhibitor. But serious side effects occur during the use of idelalisib that greatly promotes the development of novel PI3K $\delta$  inhibitors. Nevertheless, idelalisib is still an important milestone in the development of selective PI3K $\delta$  inhibitors, but the detailed selective binding mechanisms between idelalisib and PI3Ks have not been well elucidated. Therefore, in this study, an integrated modeling strategy combining molecular docking, molecular dynamics simulation, and free energy calculation was performed to reveal the molecular-level binding mechanisms of idelalisib and class I PI3K. First, molecular docking was carried on to obtain a reasonable binding posture of idelalisib in different PI3K isoforms. Then, key residues for selective inhibition of PI3K $\delta$  were highlighted by molecular dynamics simulation and energy calculations. Finally, idelalisib was also compared with its lead compound, IC87114, to reveal the reason for the higher potency of Idelalisib for PI3K $\delta$ . We hope that this study would provide some guidance for the rational design of selective PI3K $\delta$  inhibitors.

**Keywords** PI3K $\delta$  · selective inhibitor · Idelalisib · Molecular docking · Molecular dynamic simulation · Free energy calculation

## Introduction

The extensive research conducted over the past decades has implicated the role of various cell signaling events in numerous malignant, inflammatory, autoimmune, and

cardiovascular diseases, out of which the phosphoinositide 3-kinase (PI3K) signaling is one of the most relevant pathways because of its various vital functions, such as cell survival, proliferation, differentiation, and motility [1]. PI3K is an intracellular phosphatidylinositol kinase that is a key nodal protein in the PI3K/Akt signal transduction pathway [2]. PI3Ks belong to a lipid family which phosphorylates the inositol ring of membrane phosphoinositide, producing the second messenger, PIP3, that activates downstream signaling proteins, such as Akt (protein kinase B) [3]. The PI3Ks family is divided into three classes (I, II, and III) based on sequence homology, types of regulatory subunits, and substrates specificity. Among them, class I PI3Ks have been most widely studied, and class I PI3K further can be divided into two groups: IA class, which consists of PI3K $\alpha$ ,  $\beta$ , and  $\delta$  and IB class has only one member, PI3K $\gamma$  [4].

Since the first PI3K inhibitor, LY294002, was reported in 1994 [5], a large number of PI3K inhibitors have been developed and reported over the past few decades, most of these inhibitors are PI3K pan-inhibitors, which reversibly bind the ATP pockets of all isoforms [6]. However, current research has focused on more selective PI3K inhibitors. Comparing

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PI3K pan-inhibitors, isoform-selective inhibitors contain strong efficacy and less toxic side effects, so they are the focus of research and development of PI3K inhibitors [7]. The approval of idelalisib (CAL-101), a PI3K $\delta$ -selective inhibitor, for the treatment of chronic lymphocytic leukemia and small lymphocytic lymphoma by the U.S. Food and Drug Administration in 2014 was an important milestone for the development of selective PI3K inhibitions [8, 9]. Although idelalisib is considered as an effective drug, the unpredictable immune-mediated toxicity pattern has led to a black box warning in the USA, which limits its current role as an immunomodulator [10]. PI3K $\delta$  expression is primarily restricted to the hematopoietic system, specifically in leukocytes. The accumulated studies have highlighted the association of deregulated PI3K $\delta$  with numerous malignant, inflammatory, autoimmune, and cancer diseases, that depicts PI3K $\delta$  as a significant therapeutic target [11]. Therefore, it is not surprising that a large body of PI3K $\delta$  inhibitors have been developed in recent years, and some of them have entered the clinic trials [12].

Although serious side effects associated with idelalisib limit its clinical application, it still represents a useful tool in the study of structure selective activity relationship between PI3K $\delta$  and inhibitors [13]. But, the detailed selective binding mechanisms between idelalisib and the four PI3K isoforms have not been well elucidated. Therefore, in this study, an integrated modeling strategy, combining molecular docking, molecular dynamics (MD) simulation, and free energy calculations, was conducted to reveal the binding modes of idelalisib for class I PI3Ks at the molecular level. Firstly, molecular docking was used to get the rational binding postures within different isoforms. And then, MD and energy calculations were employed to highlight the key residues critical for PI3K $\delta$ -selective inhibition. Finally, IC87114, the lead compound of idelalisib, was also evaluated to reveal the reason for the higher potency of idelalisib for PI3K $\delta$ . We hope that our work would provide some research assistance for the rational design of novel PI3K $\delta$ -selective inhibitors.

## Methods and materials

### Molecular docking

Molecular docking was performed using the *CDOCKER* module in Discovery Studio 3.5 (DS3.5) software package. The 3D structure of idelalisib was retrieved from the co-crystal structure of PI3K $\delta$ /idelalisib complex from the PDB database (PDB ID: 4XE0) [14] and then pre-treated through the *Ligand Prepare* module with the default parameters set. The other isoform structures, PI3K $\alpha$  (PDB ID: 4OVU) [15], PI3K $\beta$  (PDB ID: 2Y3A) [16], and PI3K $\gamma$  (PDB ID: 5EDS) [17], all came from PDB database. All proteins were prepared with the *Prepare Protein*

module in DS3.5 with the default parameter, namely, removed water molecules, assigned bond sequences and protonated states, added hydrogen atoms, and the CHARMM field. Afterward, idelalisib was docked into the binding pockets of PI3K $\alpha/\beta/\gamma$ , respectively. In the *CDOCKER* protocol, the parameters were set to default. Finally, the top-scored complex poses were collected for the next analysis.

### Molecular dynamics simulation

The co-crystal structure of idelalisib/PI3K $\delta$  and the idelalisib/PI3K $\alpha$ /PI3K $\beta$ /PI3K $\gamma$  complexes given from above molecule docking was used as the initial structures for MD simulation using the *SANDER* program in AMBER18 [18]. In this study, IC87114 (PDB ID: 2X38) [19], a lead compound of idelalisib, was also employed to perform MD, to compare the affinity of idelalisib for PI3K $\delta$ . The AMBER ff14SB force field was used for the protein receptors and the ligands were minimized by the general AMBER force field (gaff) [20]. Each inhibitor was optimized using the semi-empirical AM1 method in Gaussian09 [21]. Then, the atomic partial charge was obtained by fitting the electrostatic potential calculated at the HF/6-31G\* level using the RESP (constrained electrostatic potential) fitting technique. The entire system was neutralized with Na<sup>+</sup> ions and immersed in a rectangular frame of TIP3P water molecules that extend 10 Å from any solute atom in all three dimensions [22]. The particle grid Ewald (PME) scheme is used to process long-range static electricity and truncated 10 Å for VDW interaction. Before the MD simulations, each system was subjected to three-stage minimizations using the *SANDER* program [23–25]. First of all, 1000 cycles of minimizations (500 cycles of steepest descent and 500 cycles of the conjugate gradient) were conducted with the backbone carbons to constrain the initial structures (50 kcal mol<sup>-1</sup> Å<sup>-2</sup>). Then, 1000 cycles of minimizations with a weaker harmonic potential (10 kcal mol<sup>-1</sup> Å<sup>-2</sup>) were performed. Finally, the whole system was relaxed by 5000 cycles of minimizations (1000 cycles of steepest descent and 4000 cycles of the conjugate gradient) without any restrain. When the minimization is finished, each system was gradually heated from 0 to 300 K over a period of 50 ps in the NVT ensemble and then performed a 50-ps MD simulation in NPT ensemble with a temperature of 300 K and pressure 1 atm. Finally, 20-ns NPT MD simulations were performed. All bonds involving hydrogen atoms were constrained using the SHAKE algorithm, and the time step was set to 2.0 fs. Coordinates were saved every 10 ps.

### MM/GBSA free energy calculations and decomposition

The snapshots of each system extracted from the last stable 10-ns MD trace were used to calculate the binding free energy

( $\Delta G_{\text{bind}}$ ) using the MM/GBSA method according to Eq. (1): [14, 26, 27]

$$\begin{aligned}\Delta G_{\text{bind}} &= G_{\text{complex}} - G_{\text{protein}} - G_{\text{ligand}} \\ &= \Delta H + \Delta G_{\text{solvation}} - T\Delta S \\ &= \Delta E_{\text{MM}} + \Delta G_{\text{GB}} + \Delta G_{\text{SA}} - T\Delta S\end{aligned}\quad (1)$$

The protein-ligand interaction profile of each residue was calculated by MM/GBSA free energy decomposition analysis which was calculated according to Eq. (2): [28–30]

$$\Delta G_{\text{inhibitor-residue}} = \Delta G_{\text{vdw}} + \Delta G_{\text{ele}} + \Delta G_{\text{GB}} + \Delta G_{\text{SA}} \quad (2)$$

where molecular mechanical energy ( $\Delta G_{\text{MM}}$ ) is the gas phase interaction energy between the ligand and the protein, calculated by electrostatic ( $\Delta G_{\text{ele}}$ ) and van der Waals interaction ( $\Delta G_{\text{vdw}}$ ). The solvation free energy ( $\Delta G_{\text{sol}}$ ) consists of polarity ( $\Delta G_{\text{GB}}$ ) and nonpolarity ( $\Delta G_{\text{SA}}$ ).  $\Delta G_{\text{GB}}$  is calculated from the generalized Born (GB) model developed by Onufriev et al. [31].  $\Delta G_{\text{SA}}$  is calculated on the basis of SASA by using the fast LCPO algorithm with a probe radius of 1.4 Å.  $-T\Delta S$  is a change in conformational entropy after ligand binding [13].

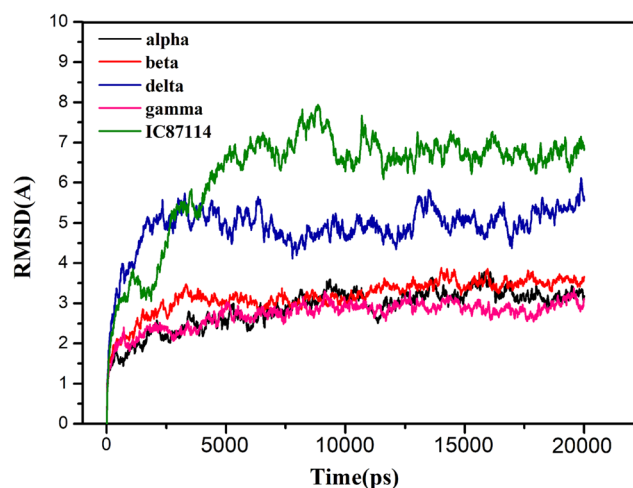
## Results and discussion

### Structural stability of the studied systems

In order to demonstrate the dynamic interaction patterns of studied complexes, 20-ns conventional MD simulations were conducted for four PI3K isoforms. Firstly, the quality of the MD simulation was assessed by monitoring the root-mean-square deviations (RMSDs) of the backbone atoms along the entire MD trajectory of each complex. As shown in Fig. 1 in the last 10 ns, the RMSD values of the stimulated systems were quite stable, which indicated that the MD simulations reached equilibrium within 10 ns.

### Isoform selectivity predicted by MM/GBSA

Subsequently, the MM/GBSA method was used to calculate the binding free energies based on a total of 1000 snapshots extracted from the last 10 ns of the MD trajectories of each system (Table 1). The predicted binding affinities of idelalisib for the four isoforms are  $-30.36$  ( $\alpha$ ),  $-38.97$  ( $\beta$ ),  $-41.65$  ( $\delta$ ), and  $-40.36$  ( $\gamma$ ) kcal/mol, respectively. The predicted free energies and the experimental data highlight the good ranking ability of MM/GBSA calculation. Notably, the MM/GBSA affinities are much stronger than the experimental values because of ignorance of the conformational entropies, accumulated studies illustrated that in most cases the MM/GBSA approach can only give good ranking results rather than accurately predicting the absolute binding free energy [32–36]. Furthermore, the energy components were calculated and the



**Fig. 1** The root-mean-square deviations (RMSDs) of the backbone atoms of the five complexes (idelalisib/PI3K $\alpha$ ,  $\beta$ ,  $\delta$ ,  $\gamma$ , and PI3K $\delta$ /IC87114)

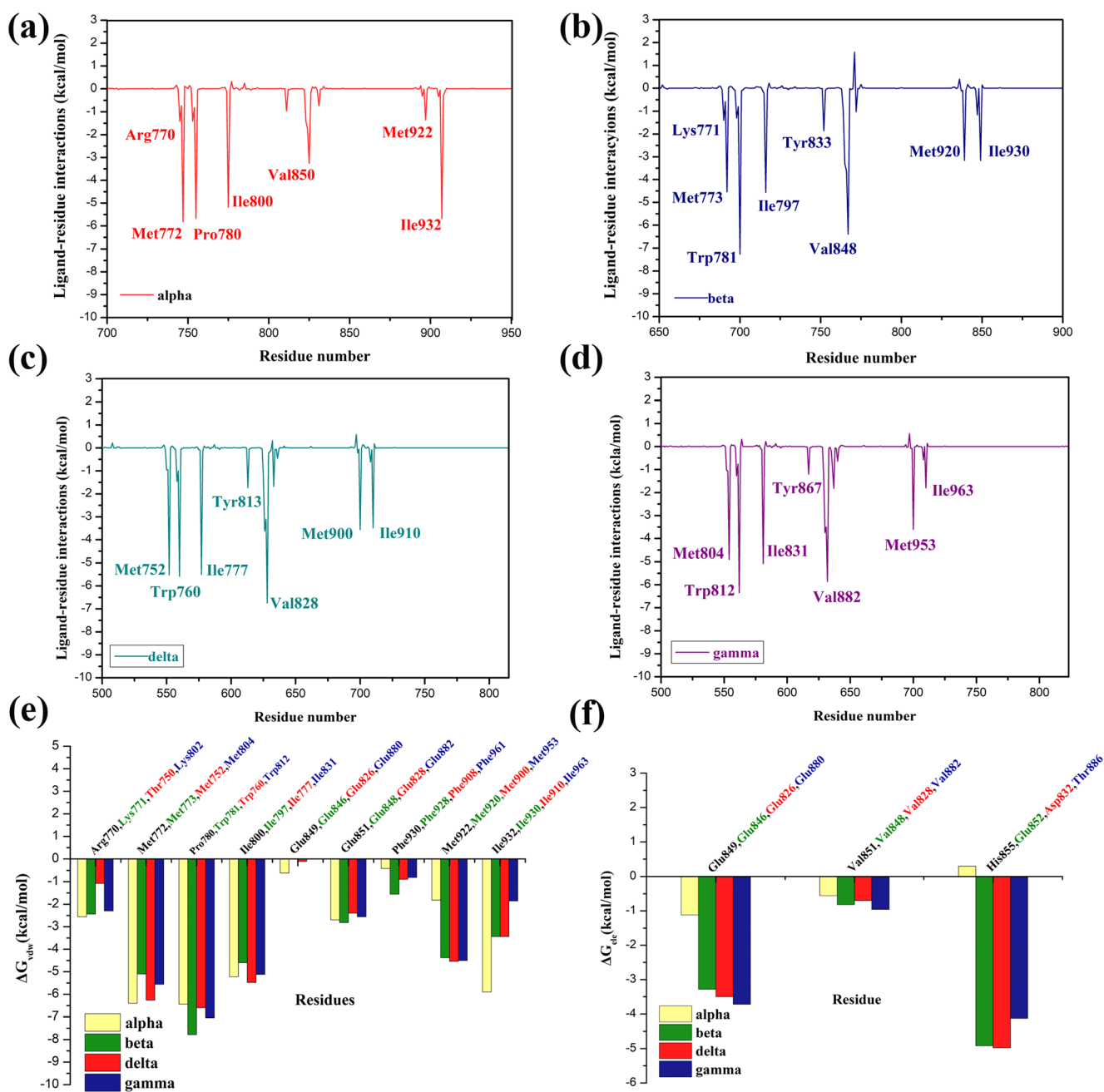
results were summarized in Table 1. As a whole, the van der Waals ( $\Delta E_{\text{vdw}}$ ) and electrostatic ( $\Delta E_{\text{ele}}$ ) terms are quite favorable for inhibitor binding to proteins, while the polar solvation term is unfavorable for binding in all complexes. In particular, the  $\Delta E_{\text{vdw}}$  is the largest contributor to the binding affinities between idelalisib and PI3K $\delta$ . The nonpolar components were defined as  $\Delta E_{\text{vdw}} + \Delta G_{\text{SA}}$ , and the values are  $-47.18$  kcal/mol for PI3K $\alpha$ ,  $-49.98$  kcal/mol for PI3K $\beta$ ,  $-51.81$  kcal/mol for PI3K $\delta$ , and  $-49.09$  kcal/mol for PI3K $\gamma$ , respectively. The nonpolar components are also consistent with the experiment results and it indicates that the nonpolar plays a critical role in determining the binding specificity of PI3K $\delta$ . Considering the hydrophobicity of the ATP-binding pocket of PI3K, it is not surprising to understand why idelalisib has stronger inhibition to PI3K $\delta$  than other isoforms. It is noteworthy that PI3K $\delta$  shows lower  $\Delta E_{\text{ele}}$  than PI3K $\beta$  and PI3K $\gamma$ , suggesting that the electrostatic term is propitious for PI3K $\beta$  and PI3K $\gamma$ . The above results preliminarily showed that increasing the van der Waals interaction for PI3K $\delta$  and decreasing the electrostatic interaction for other isoforms may improve the PI3K $\delta$ -selective binding for inhibitors.

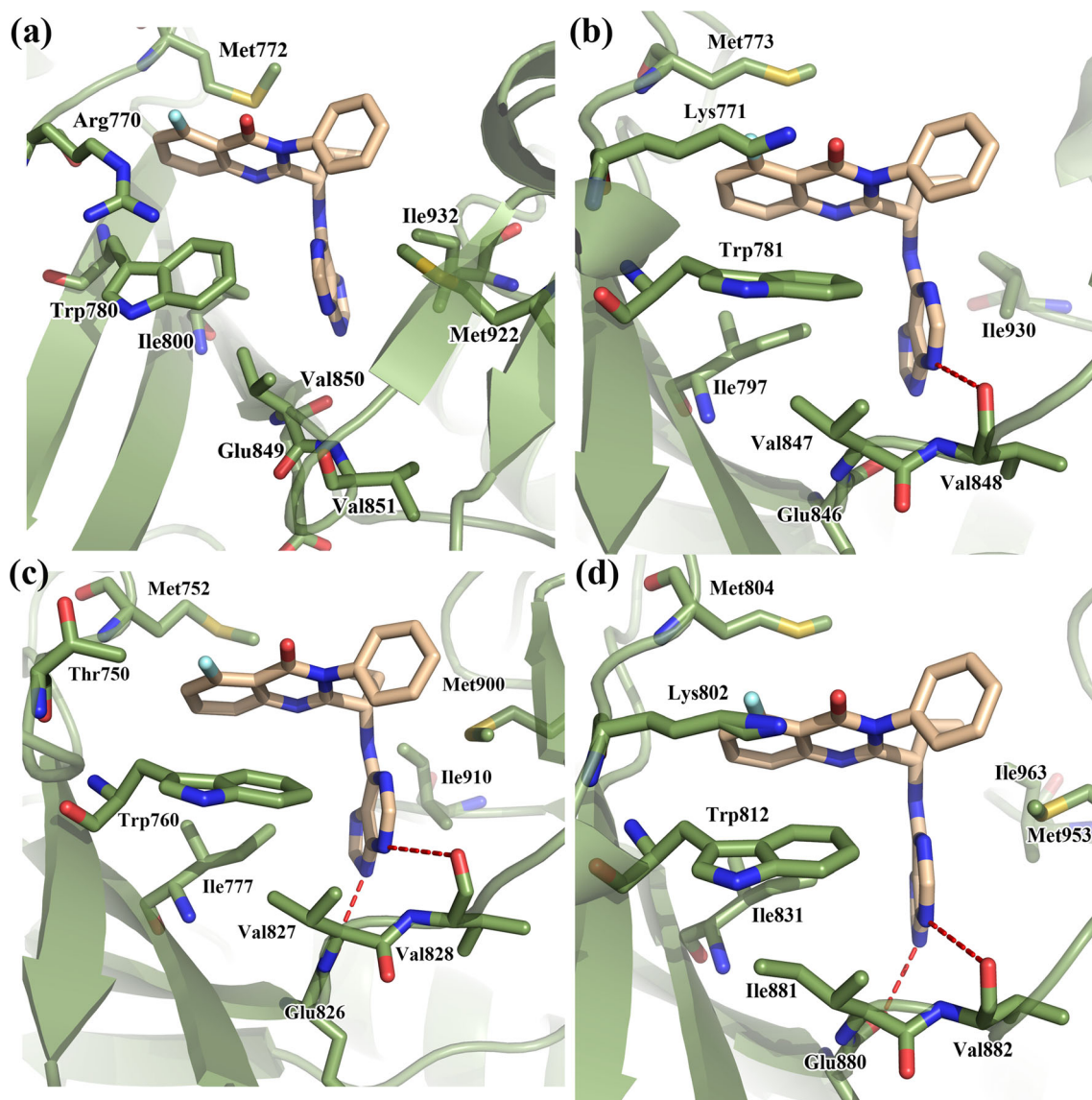
### The mechanisms of isoform-selective inhibition

In order to understand the isoform-selective inhibition and reveal the key residues involved in the specific binding process, the binding free energies of the four complexes were decomposed into the individual residue contributions. And, the results were tabulated in Table S1. Here, the non-bonded interaction was defined as  $\Delta G_{\text{nonpolar}} = \Delta E_{\text{vdw}} + \Delta G_{\text{SA}}$  and the polar contributions  $\Delta G_{\text{polar}} = \Delta E_{\text{ele}} + \Delta G_{\text{GB}}$ . The key interaction spectra were illustrated in Figs. 2 and 3. At first glance, the four complexes share three similar strong interaction patterns with the active site residues, as shown in Fig. 2,

**Table 1** Binding free energies (kcal mol<sup>-1</sup>) and experimental data of the different complexes

Complex	$\Delta E_{\text{vdw}}$	$\Delta E_{\text{elec}}$	$\Delta G_{\text{SA}}$	$\Delta G_{\text{GB}}$	$\Delta G_{\text{pred}}$	PIC <sub>50</sub>
$\alpha$ /idelalisib	-43.59 ± 0.23	-19.37 ± 2.31	-3.59 ± 0.14	36.19 ± 1.16	-30.36 ± 0.75	3.09
$\beta$ /idelalisib	-46.09 ± 1.57	-45.16 ± 3.08	-3.89 ± 0.21	56.17 ± 2.57	-38.97 ± 1.30	3.25
$\delta$ /idelalisib	-47.85 ± 0.99	-31.36 ± 1.81	-3.96 ± 0.12	41.52 ± 1.08	-41.65 ± 0.57	5.60
$\gamma$ /idelalisib	-45.19 ± 1.00	-40.58 ± 2.83	-3.90 ± 0.18	49.31 ± 2.22	-40.36 ± 0.63	4.05
$\delta$ /IC87114	-48.59 ± 1.01	-26.16 ± 1.59	-3.84 ± 0.14	39.27 ± 0.85	-39.31 ± 0.13	3.30

**Fig. 2** The interaction spectrum of **a** idelalisib-PI3K $\alpha$  residues; **b** idelalisib-PI3K $\beta$  residues; **c** idelalisib-PI3K $\delta$  residues; **d** idelalisib-PI3K $\gamma$  residues; **e** the nonpolar contributions ( $\Delta G_{\text{vdw}}$ ) for idelalisib/PI3K $\alpha$ ,  $\beta$ ,  $\delta$ ,  $\gamma$ ; **f** the polar contributions ( $\Delta G_{\text{elec}}$ ) for idelalisib/PI3K $\alpha$ ,  $\beta$ ,  $\delta$ ,  $\gamma$



**Fig. 3** The binding modes of **a** idelalisib-PI3K $\alpha$ ; **b** idelalisib-PI3K $\beta$ ; **c** idelalisib-PI3K $\delta$ ; **d** idelalisib-PI3K $\gamma$  (idelalisib colored in pink and H-bond colored in red)

they are Met (772 for  $\alpha$ , 773 for  $\beta$ , 752 for  $\delta$ , 804 for  $\gamma$ ), Ile (800 and 932 for  $\alpha$ , 797 and 930 for  $\beta$ , 777 and 910 for  $\delta$ , 831 and 963 for  $\gamma$ ). It indicates that these residues could be critical to the bio-affinity of these inhibitors. Then, the four complexes will be analyzed one by one.

### The binding mechanism of PI3K $\alpha$ complex

From Fig. 3 a and c, we found that the displacement of idelalisib in the PI3K $\alpha$  cavity is greater than that in the PI3K $\delta$  cavity; the predicted binding free energy of idelalisib/PI3K $\delta$  ( $pIC_{50} = 5.60$ ) is  $-41.65$  kcal/mol, which is greater than  $-30.36$  kcal/mol ( $pIC_{50} = 3.09$ ) of PI3K $\alpha$ . Table 1 shows that both complexes form strong van der Waals interactions with the corresponding residues.

Herein, the residues with binding energies below  $-2$  kcal/mol are identified to the key residues. For PI3K $\alpha$ , these residues include Arg770, Met772, Pro780, Ile800, Val850, and Ile932, while in PI3K $\delta$ , these residues are Met752, Trp760, Ile777, Glu826, Val827, Val828, Met900, and Ile910 (Fig. 2 a and c). As shown in Fig. 3c, Met900 of PI3K $\delta$  to be parallel to the quinazolinone group of idelalisib, rather than perpendicular to the group like the corresponding Met922 of PI3K $\alpha$  (Fig. 3a), that led to a stronger van der Waals interaction with Met900 ( $-4.54$  kcal/mol) than Met922 ( $-1.82$  kcal/mol) (Fig. 2e). Moreover, comparing with Thr750 of PI3K $\delta$ , Arg770 of PI3K $\alpha$  would cover the tryptophan surface formed in the PI3K $\alpha$  cavity to produce a steric blockage, which prevents the inhibitor from socking into the ATP-binding pocket, it

is not conducive to the combination of idelalisib and PI3K $\alpha$  (Fig. 3 a and c) [12].

On the other hand, the polar contributions were taken into consideration, as shown in Table 1, the  $\Delta E_{\text{ele}}$  value of PI3K $\delta$  is much higher than PI3K $\alpha$ , indicating that the hydrogen bond (H-bond) interactions may play an important role in the bonding of idelalisib and PI3K $\delta$ . Therefore, the H-bond occupancy was calculated and the results were summarized in Table 2. Glu826 and Val828 of PI3K $\delta$  both formed H-bonds with the inhibitor, while the corresponding Val851 and Glu849 of PI3K $\alpha$  lost these H-bonds (Figs. 2f and 3a). As shown in Table 2, the averaged H-bond occupancies between Val828 and Glu826 of PI3K $\delta$  are 60.5% and 29.7%. Val828 and Glu826 of PI3K $\delta$  have been identified two core residues, which could both form characteristic H-bonds in almost all of the PI3K $\delta$ /inhibitors complexes [14, 19].

### The binding mechanism of PI3K $\beta$ complex

As shown in Table 1, the nonpolar contribution of idelalisib/PI3K $\beta$  ( $\Delta E_{\text{vdw}} + \Delta G_{\text{SA}} = -49.98$  kcal/mol) complexes is similar to that of idelalisib/PI3K $\delta$  ( $\Delta E_{\text{vdw}} + \Delta G_{\text{SA}} = -51.81$  kcal/mol), suggesting that the nonpolar interactions play an equal role in both systems. For PI3K $\beta$ , the key residues consist of Lys771, Met773, Trp781, Ile797, Tyr833, Glu846, Val847, Val848, Met920, and Ile930. Comparing the  $\Delta G_{\text{nonpolar}}$  of PI3K $\beta$  and  $\delta$ , Met752 and Ile777 of PI3K $\delta$  exhibited more contribution than the corresponding Met773 and Ile797 of PI3K $\beta$  ( $-6.26$  and  $-5.48$  versus  $-5.10$  and  $-4.60$  kcal/mol) (Table S1). As shown in Fig. 2e, the contribution of Trp781 of PI3K $\beta$  to  $\Delta E_{\text{vdw}}$  ( $-7.78$  kcal/mol) is higher than corresponding residues of PI3K $\delta$  (Trp760), which almost dominate the nonpolar contribution in PI3K $\delta$  ( $-6.60$  kcal/mol, Fig. 2e), it indicates that the residue has a greater effect on the binding of the inhibitor to PI3K $\beta$ . As mentioned in our previous work, these residues could form the “hydrophobic pocket,” and it is significant to the binding affinity

**Table 2** The occupancy analysis for the important H-bonds of the different complexes

Complex	Donor	Acceptor	Occupancy (%)
PI3K $\alpha$ /idelalisib	idelalisib	Glu849	5.8 $\pm$ 5.59
PI3K $\beta$ /idelalisib	Val848	idelalisib	57.4 $\pm$ 6.88
	idelalisib	Glu846	14.7 $\pm$ 5.14
PI3K $\delta$ /idelalisib	Val828	idelalisib	60.5 $\pm$ 6.24
	idelalisib	Glu826	29.7 $\pm$ 4.72
PI3K $\gamma$ /idelalisib	Val882	idelalisib	52.7 $\pm$ 5.62
	idelalisib	Glu880	22.9 $\pm$ 7.13
PI3K $\delta$ /IC87114	Val828	IC87114	42.1 $\pm$ 4.80
	IC87114	Glu826	20.0 $\pm$ 5.50

rather than the selectivity for this series of compounds [37]. Besides, Met900 and Ile910 of PI3K $\delta$  (corresponding to Met920 and Ile930 of PI3K $\beta$ ) have basically the same force, which is why the nonpolar contribution between PI3K $\beta$ /PI3K $\delta$  and the inhibitor are very similar. On the other hand, Val848 of PI3K $\beta$  forms an H-bond with the ligand (Fig. 3b); the occupancy is slightly lower than Val828 of PI3K $\delta$ , while the H-bond occupancy for Glu846 of PI3K $\beta$  falls below 15%, suggesting that this H-bond interaction was almost lost in PI3K $\beta$  system (Table 2).

### The binding mechanism of PI3K $\gamma$ complex

The predicted binding free energy of PI3K $\delta$  (pIC50 = 5.60) is  $-41.65$  kcal/mol, which is slightly greater than  $-40.36$  kcal/mol (pIC50 = 4.05) of PI3K $\gamma$ , that is, the selective rate between PI3K $\delta$ / $\gamma$  for idelalisib is worse than that between  $\alpha$  or  $\beta$ . As shown in Fig. 2d, the key residues of PI3K $\gamma$  include Met804, Trp812, Ile831, Glu880, Val882, Met953, and Ile963. Figure 3 c and d show that both PI3K $\delta$  and PI3K $\gamma$  formed two H-bonds with corresponding residues, but the H-bond occupancies of Glu880 and Val882 of PI3K $\gamma$  are both smaller than that of Glu826 and Val828 in PI3K $\delta$  (Table 2). Comparing with PI3K $\gamma$ , Ile910 of PI3K $\delta$  shows a stronger van der Waals interaction than Ile963 of PI3K $\gamma$  ( $-3.44$  versus  $-1.86$  kcal/mol). Similar to PI3K $\alpha$ , Met900 and Ile910 of PI3K $\delta$  could form hydrophobic regions with equal force, while in PI3K $\gamma$ , the nonpolar contribution of Ile963 is much lower than that of Met953, leading to an obvious decrease in inhibitor binding improvement in this “hydrophobic pocket.” In addition, the same as  $\alpha$  and  $\beta$  is that Lys802 will cover the surface of the formed tryptophan, and the obstruction of this space is not conducive to the binding of the inhibitor to PI3K $\gamma$  (Fig. 3d). Therefore, the corresponding Thr750 plays a very important role in enhancing the selectivity of PI3K $\delta$ . From Fig. 2e, the van der Waals interactions of Met752, Ile 777, and Ile910 of PI3K $\delta$  are stronger than Met804, Ile831, and Ile963 of PI3K $\gamma$ . Generally speaking, the main residues of PI3K $\delta$  contribute the more favorable interaction to idelalisib than PI3K $\gamma$ , but these binding free energies are not obviously different that is why idelalisib shows the poor selectivity rate to PI3K $\gamma$ .

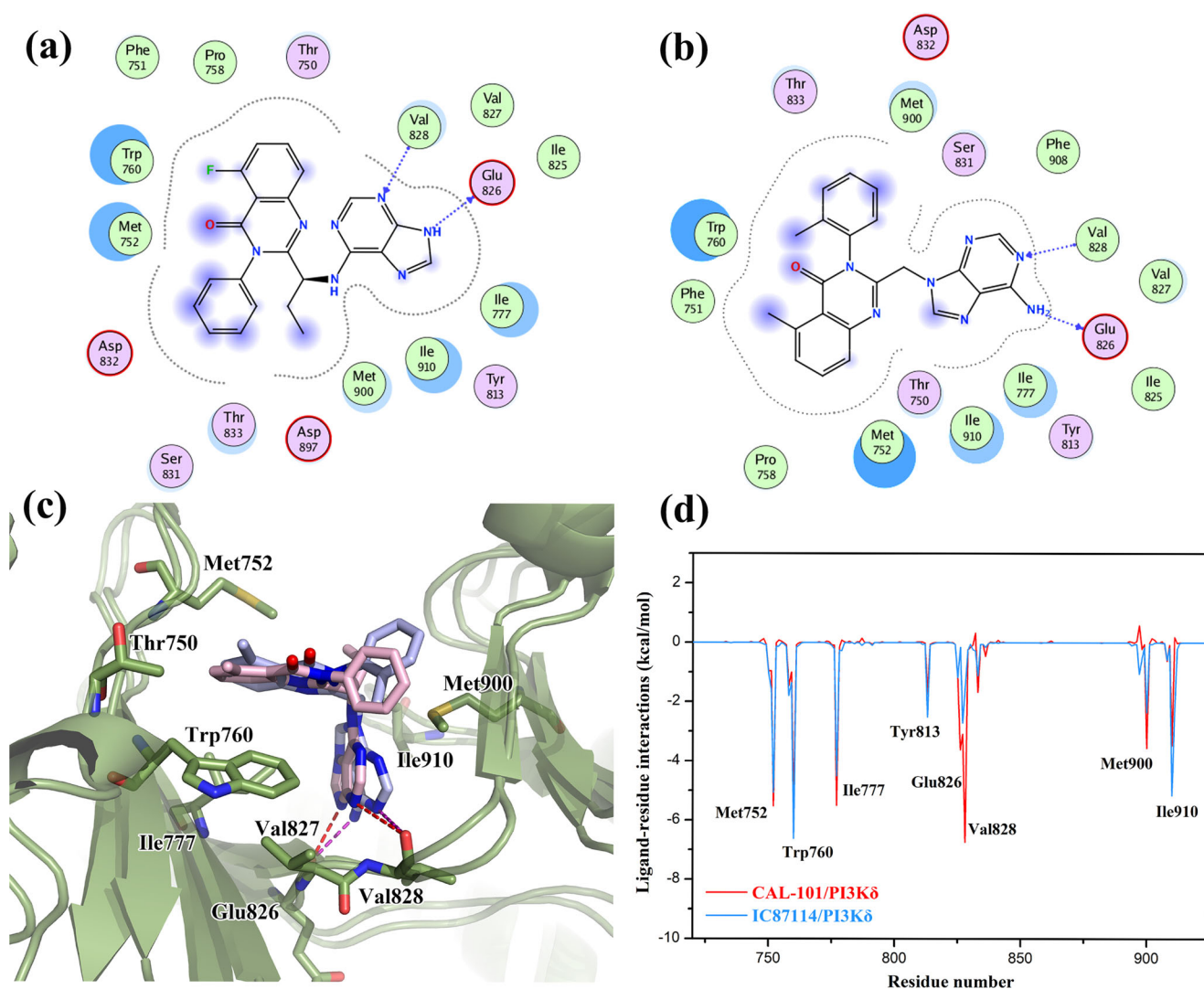
### The stronger binding affinity mechanisms of idelalisib to PI3K $\delta$

IC87114 is the first reported PI3K $\delta$  inhibitor, and idelalisib was developed based on the rational modification of IC87114. As shown in Table 1, the inhibitory activity of idelalisib increases from 3.30 to 5.60 relative to IC87114; therefore, in this study, IC87114, as a lead compound, was also submitted the MD simulation to find the key residues associated with the

binding affinity to PI3K $\delta$ . The 2D structures of the two inhibitors and the energy of the key residues are shown in Fig. 4. Compared with IC87114, the thieno[3,2-*d*]-fused analogues retain the high valence of PI3K $\delta$  for the mother core to a large extent. Previous studies indicated that the small substituent of isoquinolinone is essential for the activity of PI3K $\delta$  that led to the discovery of chloroisoquinolinone motif [38]. The contribution of key amino acids to the energy of the inhibitor shows that Ile777, Val826, Val828, and Met900 are all favorable to idelalisib (Fig. 4d). Comparing the two structures, a hydrophobic group, ethyl, was introduced to idelalisib, which significantly enhances the hydrophobic interaction to the Met900 (Fig. 4 a and b). As illustrated in Fig. 4c, idelalisib and IC87114 both form two H-bonds with Glu826 and Val828, while the H-bond occupancies of IC87114 are lower than that of PI3K $\delta$  (Table 2). These H-bond interactions may make idelalisib stronger binding affinity to PI3K $\delta$  than IC87114.

## Conclusion

In the present work, the MD simulations and binding free energy calculations were employed to investigate the selective binding mechanisms between idelalisib and PI3Ks. Firstly, the predicted binding free energies are in good agreement with the bio-activity data. And then, the contribution of individual residue to binding affinity was evaluated through the free energy decomposition analysis. The H-bond interactions between idelalisib and Glu826/Val828 play critical roles in the selective inhibition to PI3K $\delta$ . Besides, the hydrophobic interaction with Met752, Trp760, Ile777, Met900, and Ile910 could improve the stability of the inhibitor at the binding site of PI3K $\delta$ . Comparing with other PI3K isoforms, the residue equivalent to Thr750 of PI3K $\delta$  is lysine or arginine. This unique structure characteristic may account for the high PI3K $\delta$  selectivity. In a word, the specific binding of PI3K $\delta$  is determined by the



**Fig. 4** The 2D presentations of interaction between **a** idelalisib and PI3K $\delta$ ; **b** IC87114 and PI3K $\delta$ ; **c** the alignment of the binding pocket of idelalisib (colored in pink) and IC87114 (colored in purple); **d** the PI3K $\delta$ -residues interaction spectrum of idelalisib (colored in red) and IC87114 (colored in blue)

additive contributions provided by these multiple “key” residues; our studies may provide some guidance for the development of novel selective PI3K $\delta$  inhibitors.

**Authors' contributions** Z.J., L.H., and J.J. developed the study concept and design. Z.H. and S.H. performed the modeling studies. Z.H., Y.L., and C.Y. carried out the data analysis. Z.J., Z.H., and Y.L. drafted the manuscript, C.Y., L.H., and J.J. approved the manuscript.

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## Compliance with ethical standards

**Conflict of interest** The authors declare that they have no conflict of interest.

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