

Fluoroarenes in the synthesis of benzoannulated nitrogen-containing heterocycles

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Methods for the synthesis of fluoro-containing benzazoles and benzazines from fluoroarenes (fluoroanilines, fluorobenzoic acids, fluoroacetophenones, fluorophenols, etc.) are described. The potentialities of the most important synthons are considered. Ways of constructing heterocycles with fluorine atoms in the annulated benzene fragment are discussed.

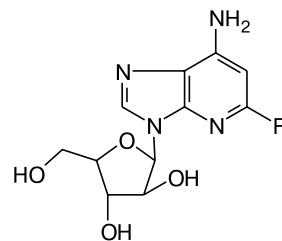
Key words: fluoroanilines, 4,5-difluoro-1,2-phenylenediamine, fluoro-containing 2-halo-benzoyl chlorides, fluoro-containing benzamides, acetophenones, phenols, quinolines, quinazolines, benzotriazines, benzofuroxans, phenoxazines, benzimidazoles, benzoxazines, benzothiazines.

Introduction

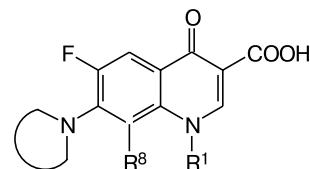
Interest in the creation of novel organofluorine compounds is due to their unique physicochemical and biological properties. Introduction of F atoms into organic molecules, especially into those fragments that are responsible for their biological activity, becomes an important aspect of pharmaceutical investigations, which in turn stimulates a search for new methodologies of the synthesis of diverse fluoro-containing compounds.^{1–8}

The volume of the fluorine atom is close to that of the hydrogen atom; when the latter is replaced by a F atom, the primary metabolism of organic substances usually remains unchanged since enzymes in biological systems often do not distinguish between these compounds (mimicry effect). However, because C—F bonds are stronger than C—H bonds, their degradation is hindered and further metabolism proceeds anomalously.^{2,9}

A well-known example of the inhibitors of nucleic exchange enzymes is 5-fluorouracil, which has become usual in oncological practice. This preparation is an antimetabolite of uracil, competes with it for thymidylate synthase, and breaks the formation of DNA. The recently obtained drug fludarabine inhibits DNA polymerase and is used to treat chronic lympholeucosis and lymphoma.¹⁰ This compound is an analog of adenosine, in which the ribose residue is replaced by the arabinose residue and the purine ring bears the fluorine atom in position 2.



Fludarabine



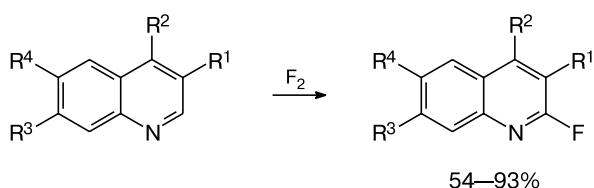
Fluoroquinolones

Fluoroquinolones (6-fluoro-4-oxoquinoline-3-carboxylic acid derivatives^{11–16}) constitute the most important class of antibacterial drugs. By inhibiting type II topoisomerase (DNA gyrase), they blockade the synthesis of RNA on the DNA matrix and, consequently, cause cells to perish.¹⁷ The specific effect of fluoroquinolones on bacteria is that they inhibit DNA gyrase of bacteria but do not bind themselves to DNA topoisomerases of host cells. Leading pharmaceutical firms have already created and brought into medical practice a large group of drugs

of this class such as pefloxacin, ciprofloxacin, levofloxacin, grepafloxacin, moxifloxacin, etc.¹⁸

In this review, we attempted to systematize the methods for the synthesis of benzazines and benzazoles containing fluorine atoms in the annulated benzene fragment with consideration for two fundamental approaches to the synthesis of fluoro-containing benzoannulated azaheterocycles. The first approach involves introduction of F atoms into a prepared heterocycle, while the second approach involves construction of cyclic systems from fluoro-containing synthons. The former approach has a number of limitations associated with the fact that selective introduction of F atoms into organic molecules is difficult. For instance, quinoline was directly fluorinated at the α -position^{19,20} (Scheme 1), while electrophilic fluorination of its benzene ring was not selective (Scheme 2).^{21,22}

Scheme 1



Reagents and conditions: 10% in nitrogen, I₂, CF₂Cl₂–CFCl₂.

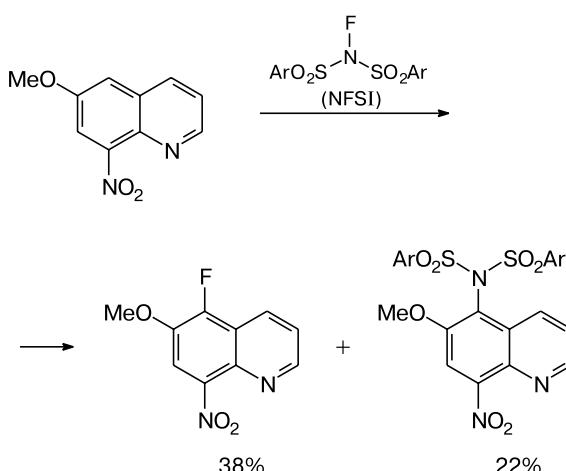
R¹ = R² = R³ = R⁴ = H (**a**); R¹ = R⁴ = H, R² = R³ = Cl (**b**);
R¹ = Br, R² = R³ = R⁴ = H (**c**); R² = Cl, R¹ = R³ = R⁴ = H (**d**);
R⁴ = Cl, R¹ = R² = R³ = H (**e**); R¹ = R⁴ = H, R² = Cl, R³ = CF₃ (**f**)

Direct fluorination at position 5 of quinoline was described in Ref. 8 (Scheme 3). However, the yield of the target product was low and sulfone imide was isolated as a by-product.

A reaction of 1-hydroxy-2-oxo-1,2,3,4-tetrahydroquinoline with diethylaminosulfur trifluoride provides an example of selective fluorination giving a 6-fluoro derivative only.²³ However, the problem of selective fluorination still persists for a wide range of heterocycles.

In this context, we believe that the synthesis of fluoro-containing benzazoles and benzazines from fluoroarenes

Scheme 3



Conditions: 130 °C, 3 h.

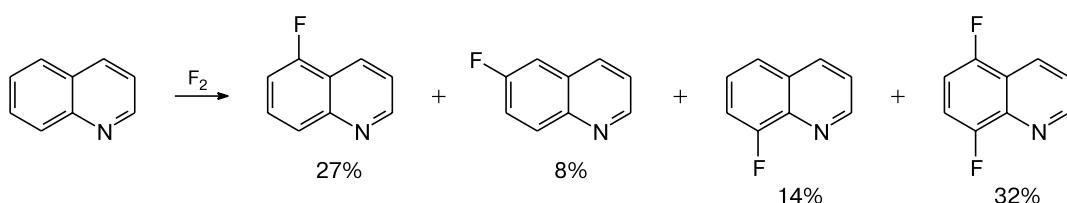
(fluoroanilines, fluorobenzoic acids, fluoroacetonophenes, fluorophenols, etc.) is more promising. Here we considered the potentialities of the most important synthons and discussed ways of constructing, on their basis, heterocycles containing fluorine atoms in the benzene fragment. These ways can be represented by Scheme 4.

1. Fluorinated anilines and their derivatives in the synthesis of benzoannulated azaheterocycles

1.1. 2-Fluoro-, 3-fluoro-, and 4-fluoroanilines

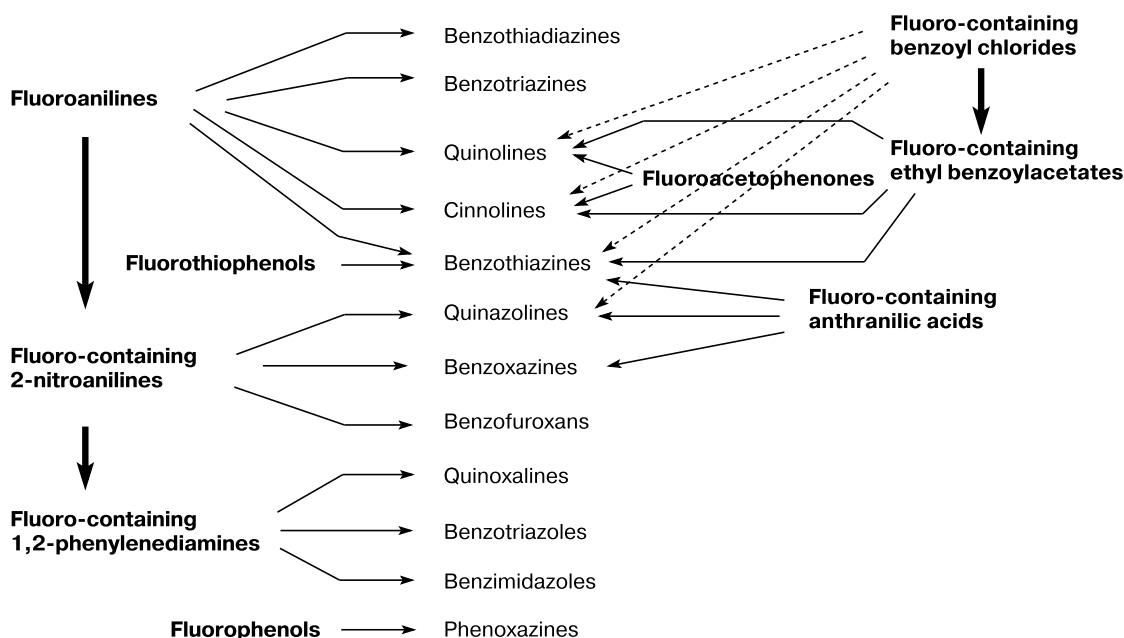
Monofluoroanilines are often used as synthons for the synthesis of benzazines containing a fluorine atom. For instance, reactions of 4-alkyl-2-fluoroanilines **1** with ethyl 2-acetylpropionate followed by cyclization of the resulting enamine **2** give 2,3,6-trialkyl-8-fluoro-4-hydroxyquinolines **3** (Scheme 5). Derivatives **3** are employed in agriculture.²⁴

Scheme 2

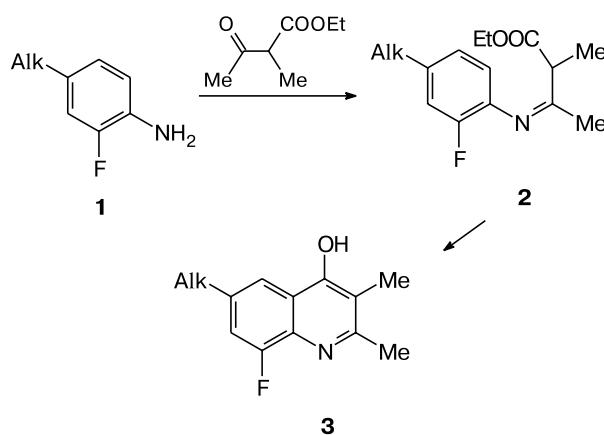


Reagents and conditions: N₂ (50 mL min⁻¹), 0 °C, H₂SO₄.

Scheme 4



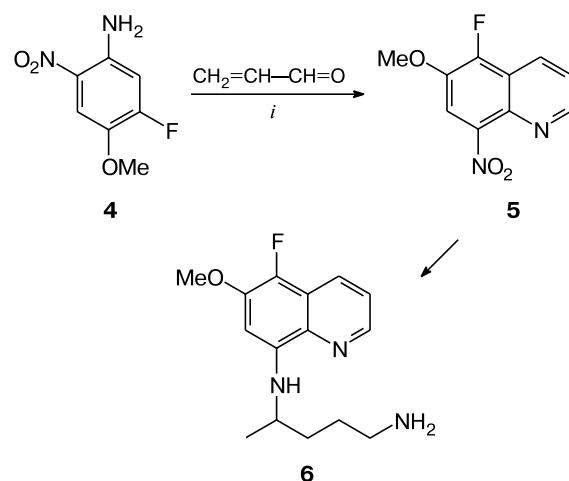
Scheme 5



The Skraup synthesis from 3-fluoroaniline **4** gives 5-fluoro-6-methoxy-8-nitroquinoline **5** (Scheme 6); compound **5** is used to prepare 5-fluoroprimaquine **6**, an analog of the known antimalaria drug.²⁵

Diazotization of 3-fluoroanilines gives the corresponding 3-fluorobenzenediazonium chlorides **7**, which in turn can be transformed into 1,3,5-triphenylformazans **8**. In the BF_3/AcOH system, the latter undergo cyclization following an electrophilic attack with cleavage of the N–N bond and release of aniline. The final products are 7-alkoxy-6-fluoro-3-phenyl-1,2,4-benzotriazines **9** (Scheme 7). Fluoro-containing 3-phenyl-1,2,4-benzotriazines are of interest because of their antiviral effect (e.g., on the smallpox virus²⁶).

Scheme 6



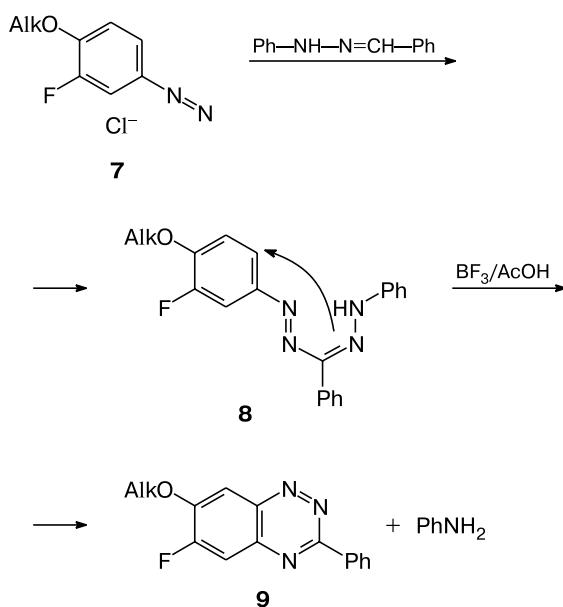
i. H_3AsO_4 , H_3PO_4 .

A reaction of 4-fluoroaniline **10** with 1,1,1-trifluoropentane-2,4-dione yields isomeric 6-fluoroquinolines **11** and **12** in the ratio 1 : 1 (Scheme 8).²⁷

Amide **13a** obtained from 4-fluoroaniline (**10**) is thionated with Lawesson's reagent. Cyclization of the resulting thioamide **13b** under UV light in the presence of $(\text{Me}_3\text{Si})_3\text{SiH}$ (4 equiv.) in degassed benzene gives 1,3-dihydrofuro[3,4-*b*]quinoline **14** (Scheme 9).²⁸

7-Fluoro-1-methyl-1,4-dihydro-9*H*-pyrazolo[4,3-*b*]quinolin-9-one (**15**), which is the protein kinase C inhibitor, is obtained by cyclization of 4-[(4-fluorophenyl)-

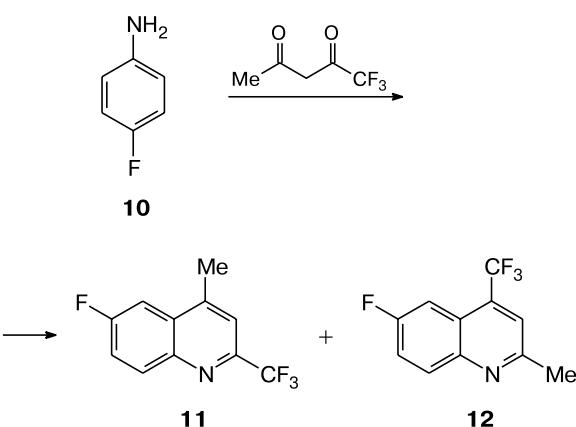
Scheme 7



amino]-1-methyl-1*H*-pyrazole-5-carboxylic acid (**16**). The latter is prepared from 4-fluoroaniline (**10**) and 4-iodo-1-methyl-1*H*-pyrazole-5-carboxylic acid (**17**) in water in the presence of a copper powder and sodium carbonate (Scheme 10).²⁹

3-(Benzoxazol-2-yl)-4-fluoroaniline (**18**, X = O) and 3-(benzothiazol-2-yl)-4-fluoroaniline (**18**, X = S), which are obtained by condensation of 5-amino-2-fluorobenzoic acid (**19**) with 2-aminophenol or 2-aminobenzenethiol in polyphosphoric acid, are used to synthesize 7-benz-

Scheme 8

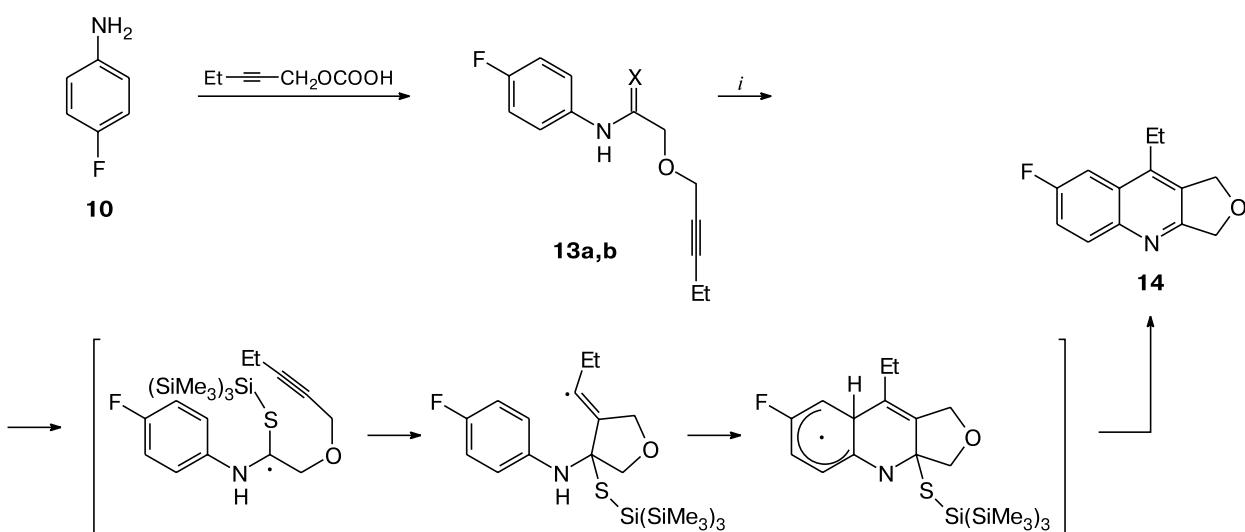


oxazolyl- and 7-benzothiazolyl-6-fluoroquinolones **20** (Scheme 11).³⁰

Diazotization of 4-fluoroaniline (**10**) gives 4-fluorobenzenediazonium chloride (**21**), which undergoes coupling with ethyl acetoacetate to form hydrazone **22**. Intramolecular cyclization of the latter in chlorobenzene in the presence of AlCl_3 leads to 3-acetyl-6-fluoro-1*H*-cinnolin-4-one (**23**). Bromination of methyl ketone **23** under UV light yields 3-bromoacetyl-6-fluorocinnolone **24**, which reacts with thioamides to give 3-thiazolyl-cinnolones **25**, which exhibit antimicrobial activity (Scheme 12).^{31,32}

Fluoro-containing cinnolin-4(1*H*)-ones, imidazo[2,1-*b*]thiazolylcinnolones, and pyrazolo[4,3-*c*]cinnolines that exhibit antibacterial and fungicidal effects are described in Ref. 33.

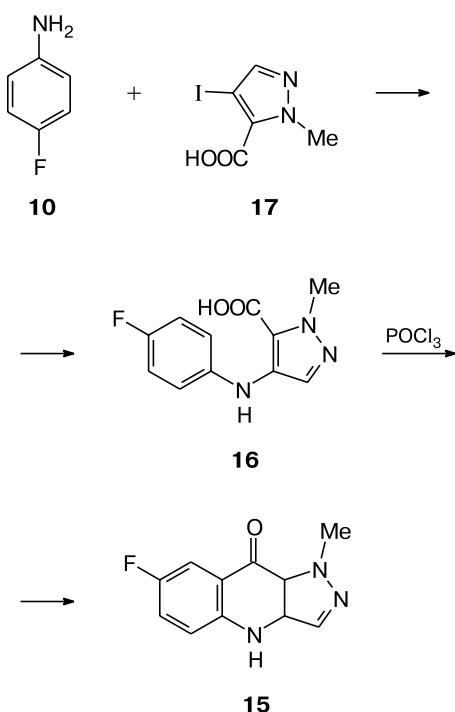
Scheme 9



i. 1) Lawesson's reagent. 2) $(\text{Me}_3\text{Si})_3\text{SiH}$, UV irradiation.

13: X = O (**a**), S (**b**)

Scheme 10



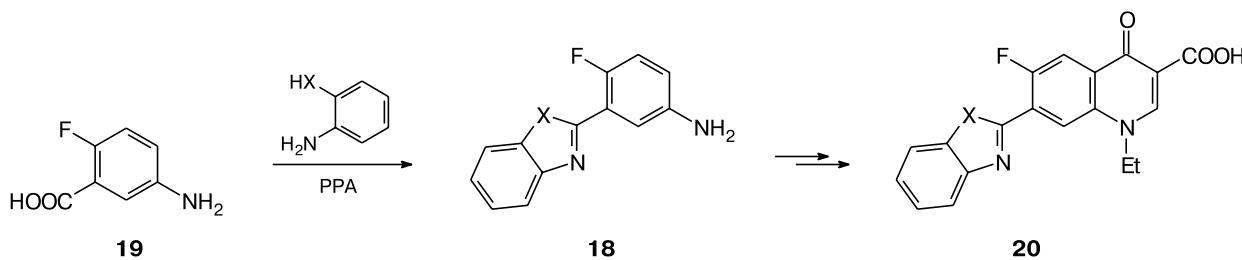
1.2. 3,4-Difluoro- and 3-chloro-4-fluoroanilines

One of the most important synthons for construction of fluoro-containing azaheterocycles is 3,4-difluoroaniline (**26**). Compound **26** can be prepared from *ortho*-difluorobenzene, a product of the difluorocarbene synthesis (Scheme 13).³⁴

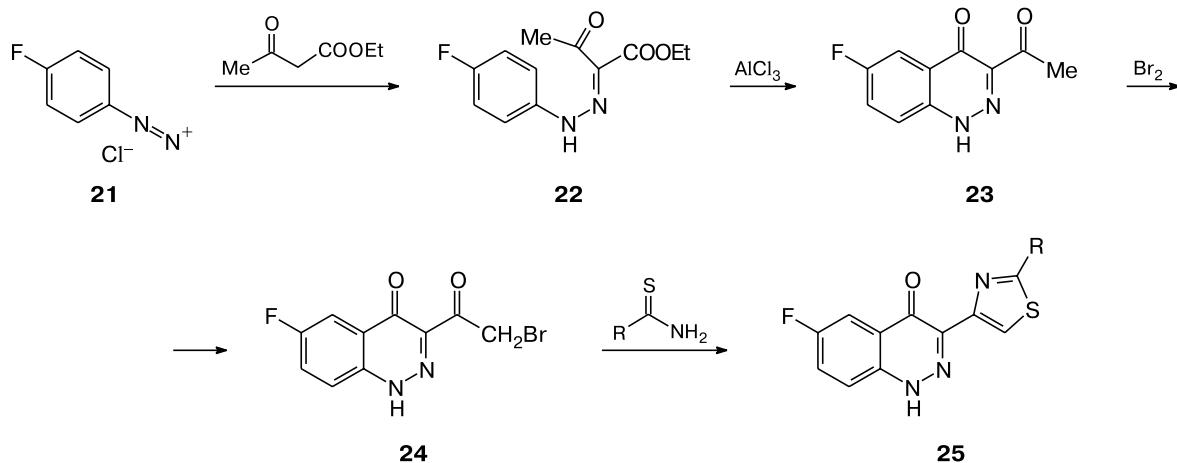
Condensation of 3,4-difluoroaniline (**26**) with ethyl ethoxymethylenemalonate gives enamine **27**; its intramolecular cyclization leads to ester **28**, which is a key intermediate in the synthesis of some fluorooxinoquinoline-carboxylic acids, including pefloxacin **29** (Scheme 14).¹⁰

A search for new drugs of the fluoroquinolone series is in progress; an important way of modifying fluoroquinolones involves introduction of various substituents into their benzene ring. For instance, prior to the synthesis of 8-substituted fluoroquinolones, a desired substituent (SMe , Me , or CH_2OAc) should be introduced into position 2 of 3,4-difluoro-*N*-benzoylaniline.³⁵ For the synthesis of fluoroquinolones, 3-chloro-4-fluoroaniline is often used instead of 3,4-difluoroaniline and the cyclization is carried out by heating enamines in polyphosphoric acid or its ester.³⁶

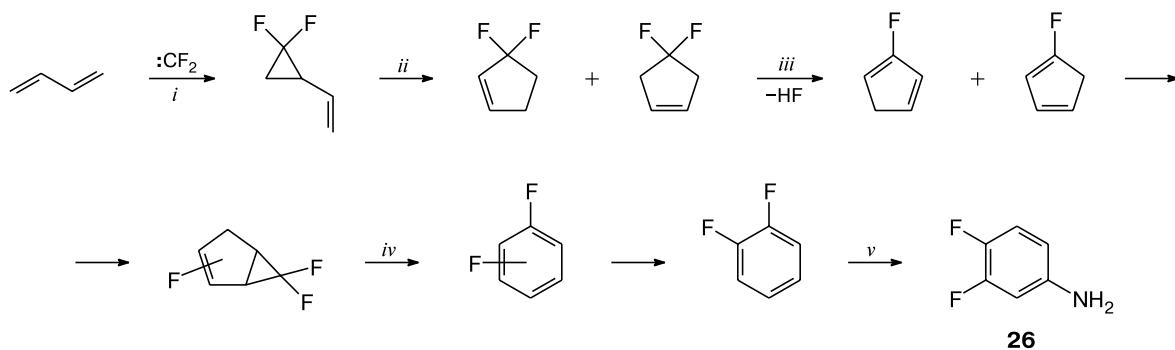
Scheme 11



Scheme 12

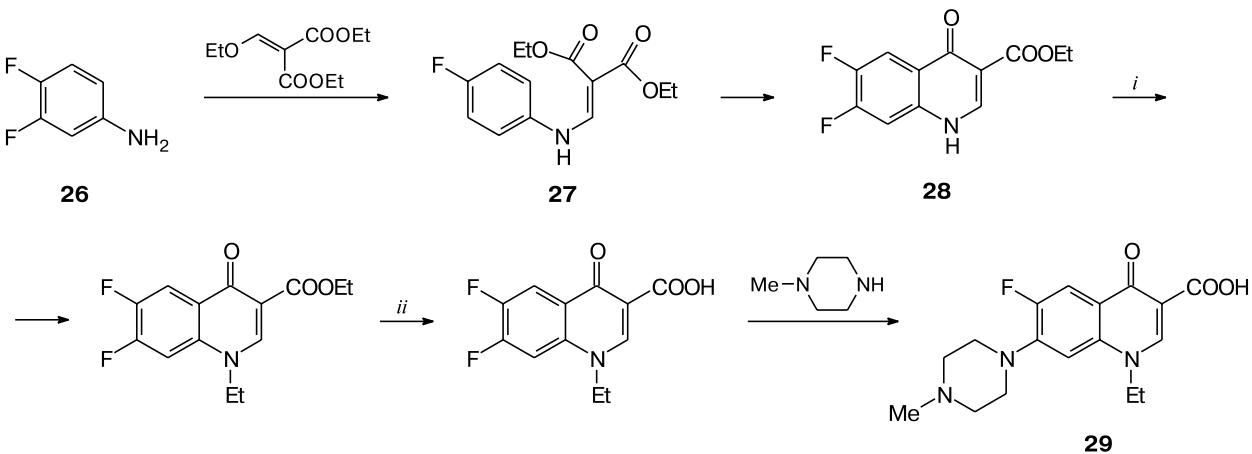


Scheme 13



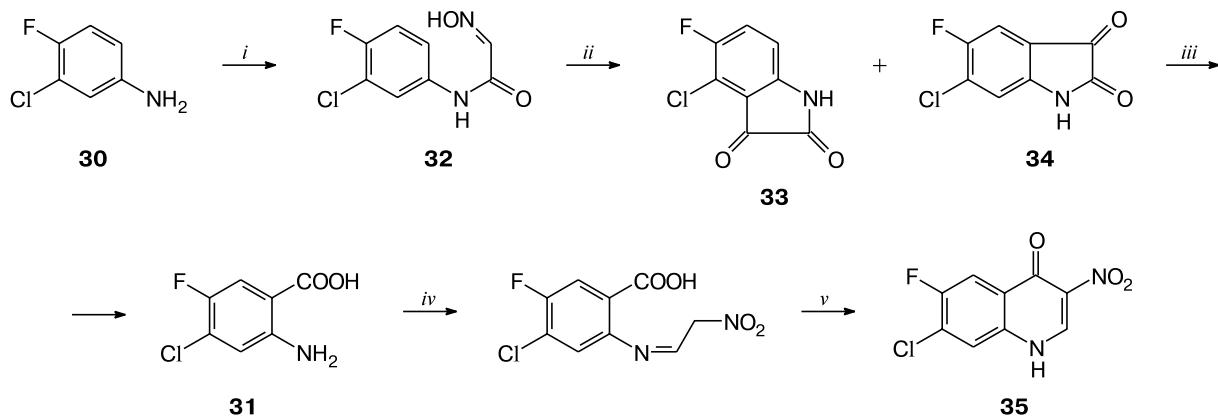
Reagents and conditions: *i.* 270 °C; *ii.* 400 °C; *iii.* 650 °C; *iv.* 650–700 °C; *v.* 1) HNO₃, H₂SO₄; 2) Sn, HCl.

Sceme 14



Reagents and conditions: *i.* EtBr, K₂CO₃; *ii.* HCl, AcOH, 110–120 °C.

Scheme 15

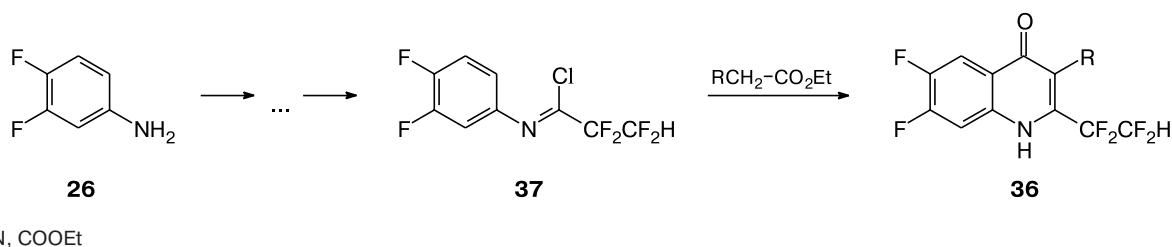


Reagents: *i.* CCl₃CH(OH)CCl₃, NH₂OH·HCl; *ii.* H₂SO₄; *iii.* H₂O₂; *iv.* HON=CHCH₂NO₂; *v.* (MeCO₂)₂O, MeCO₂Na.

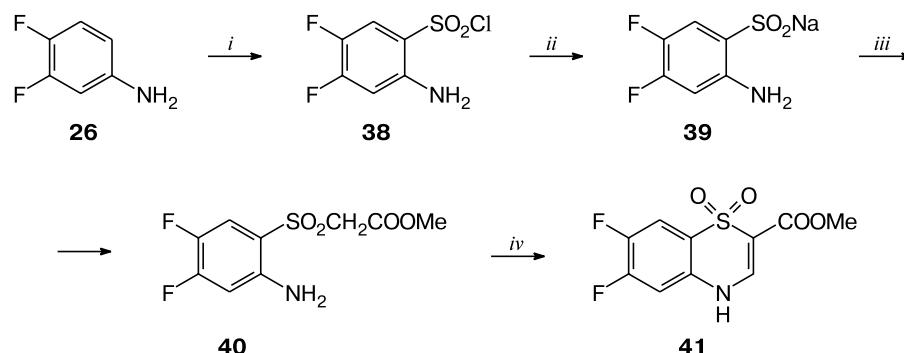
3-Chloro-4-fluoroaniline (**30**) can be easily transformed into 3-chloro-4-fluoroanthranilic acid **31**. A reaction of compound **30** with chlral hydrate and hydroxylamine

yields isonitrosoacetanilide **32**; its cyclization in H₂SO₄ leads to a mixture of indole-2,3-diones **33** and **34** (Scheme 15).³⁷ Oxidation of this mixture with H₂O₂ gives

Scheme 16



Scheme 17



Reagents and conditions: *i.* HSO_3Cl , SOCl_2 ; *ii.* Na_2SO_3 ; *iii.* $\text{BrCH}_2\text{COOMe}$, EtOH ; *iv.* HC(OEt)_3 , 140°C .

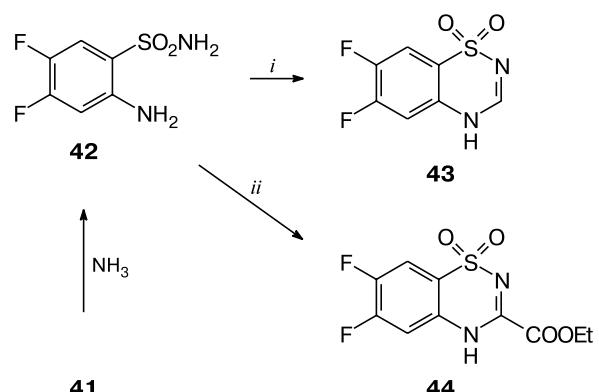
2-amino-4-chloro-5-fluorobenzoic acid (**31**), from which fluoroquinolone **35** was obtained. This way of constructing fluoroquinolones allows additional modifications of position 3 of the pyridone ring.

3,4-Difluoroaniline (**26**) is employed for the synthesis of 6,7-difluoro-2-polyfluoroalkylquinolones **36**. For this purpose, compound **26** is acylated with polyfluoroalkenoic anhydride and the resulting anilide is transformed into the corresponding imino chloride **37**. Cyclization of the latter with ethyl malonate or ethyl cyanoacetate yields quinolones **36** (Scheme 16).³⁸

Using 3,4-difluoroaniline as a starting material, one can obtain sulfur analogs of fluoroquinolones: fluoro-containing 1,4-benzothiazines and 1,2,4-benzothiadiazines. Chlorosulfonation of 3,4-difluoroaniline (**26**) gives 2-amino-4,5-difluorobenzensulfon chloride (**38**), which is reduced in a basic medium to the corresponding sodium sulfinate **39**. Alkylation of the latter with methyl bromoacetate in ethanol followed by condensation of intermediate **40** with triethyl orthoformate yields 6,7-difluoro-2-methoxycarbonyl-4*H*-1,4-benzothiazine 1,1-dioxide (**41**) (Scheme 17).³⁹

Ammonolysis of sulfonyl chloride **38** gives sulfonamide **42**; its cyclization with triethyl orthoformate or ethyl oxalate affords 1,2,4-benzothiadiazines **43** and **44** (Scheme 18).⁴⁰

Scheme 18



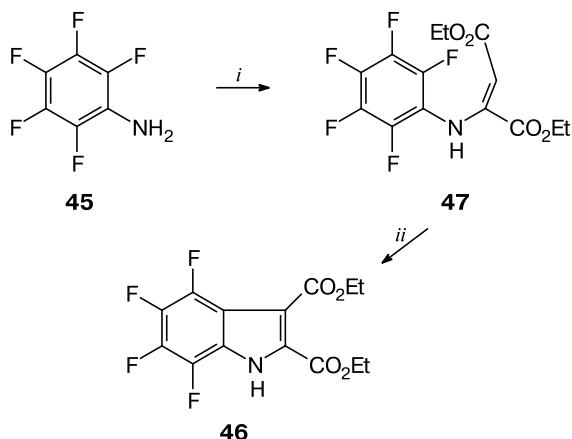
Reagents: *i.* HC(OEt)_3 ; *ii.* EtOOCCOOEt .

1.3. Polyfluorinated anilines and nitrobenzenes

The presence of a F atom in the *ortho*-position relative to the reactive group makes polyfluoroanilines and polyfluoronitrobenzenes convenient synthons for the synthesis of azaheterocycles *via* nucleophilic substitution of fluorine. For instance, pentafluoroaniline (**45**) serves as

a starting reagent for the preparation of polyfluorinated indole **46** according to Scheme 19 (see Ref. 41): amino-fumarate **47** obtained from compound **45** and diethyl acetylenedicarboxylate undergoes intramolecular cyclization into indole **46**.

Scheme 19



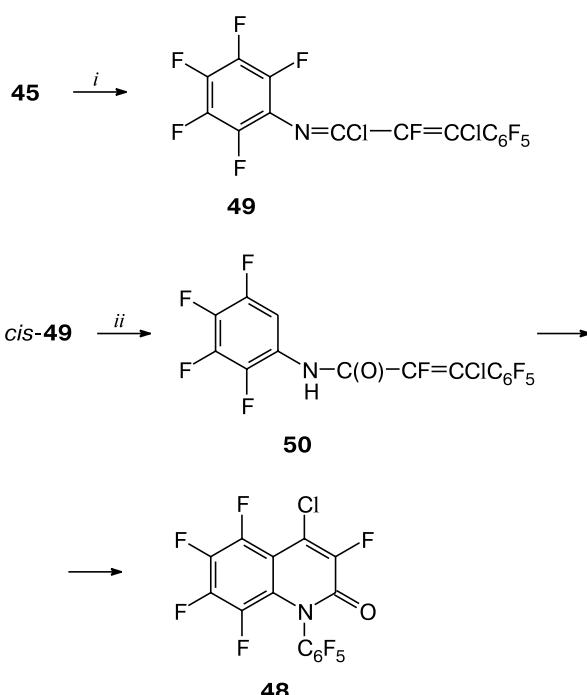
Reagents: *i.* NaH, THF, EtOOC—≡—COOEt; *ii.* NaH, DMF.

In Ref. 42, the synthesis of 4-chloro-3,5,6,7,8-pentafluoro-1-pentafluorophenylquinolin-2-one (**48**) is described. Pentafluoroaniline (**45**) reacts with *cis*-tetrachloro-2-fluoro-1-pentafluorophenylpropene in the presence of AlCl₃ to give a mixture of *cis*- and *trans*-aza dienes **49**. Hydrolysis of the mixture in a water–dioxane solution of K₂CO₃ yields the corresponding amides **50**. The amide obtained from the *cis*-isomer readily undergoes intramolecular cyclization into compound **48** (Scheme 20).

Electrochemical oxidation of pentafluoroaniline (**45**) gives a mixture of octafluorophenazine (**51**) and decafluoroazobenzene (**52**) (Scheme 21).⁴³

Diazotization of polyfluoroanilines followed by reduction of the corresponding diazonium salts affords polyfluorophenylhydrazines **53**. After their acylation, the resulting hydrazides **54** are thionated with P₂S₅ to give polyfluorophenyl thiohydrazides **55**. Finally, cyclization of the latter yields 2-substituted 5,6,8-trifluoro-

Scheme 20

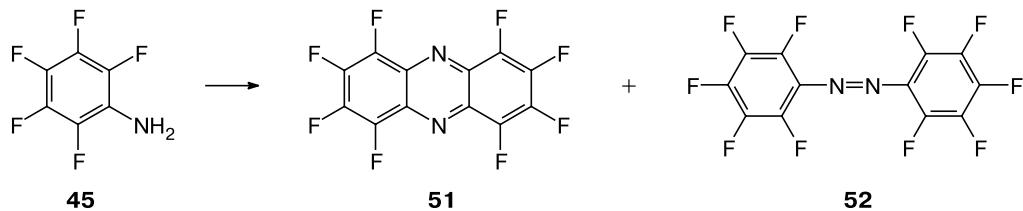


Reagents: *i.* *cis*-C₆F₅CCl=CFCCl₃, AlCl₃; *ii.* K₂CO₃.

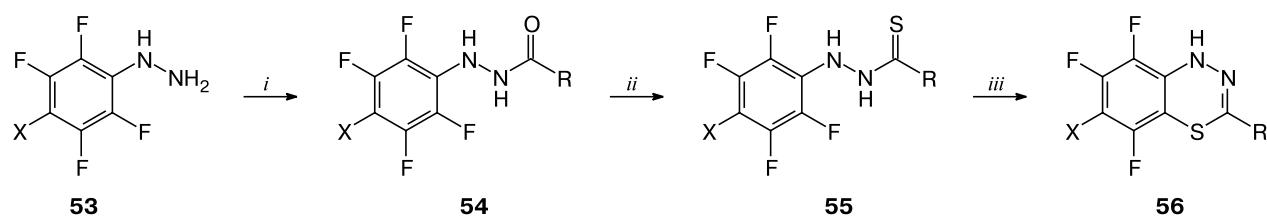
and 5,6,7,8-tetrafluoro-4*H*-1,3,4-benzothiadiazines **56** (Scheme 22).⁴⁴

2,3,4-Trifluoronitrobenzene (**57**) is a starting material for the preparation of (*S*)-(—)-7,8-difluoro-3-methyl-3,4-dihydro-2*H*-[1,4]benzoxazine (**58**), the most important intermediate in the synthesis of the antibacterial drug levofloxacin. Several routes to fluoro-containing stereoisomers of 3-methyl- and 2-methyl-2*H*-[1,4]benzoxazines were proposed.⁴⁵ For instance, compound **57** reacts with (*R*)-propane-1,2-diol in THF in the presence of NaH to give a mixture of (*R*)-3,4-difluoro-2-(2-hydroxypropoxy)-nitrobenzene and (*R*)-3,4-difluoro-2-(1-hydroxypropan-2-yloxy)nitrobenzene in the ratio 3 : 2. The mixture is transformed into mesylates **59** and **60**; this is followed by reduction of the nitro group, cyclization in the presence of potassium *tert*-butoxide, and separation of benzoxazines **58** and **61** by column chromatography (Scheme 23).

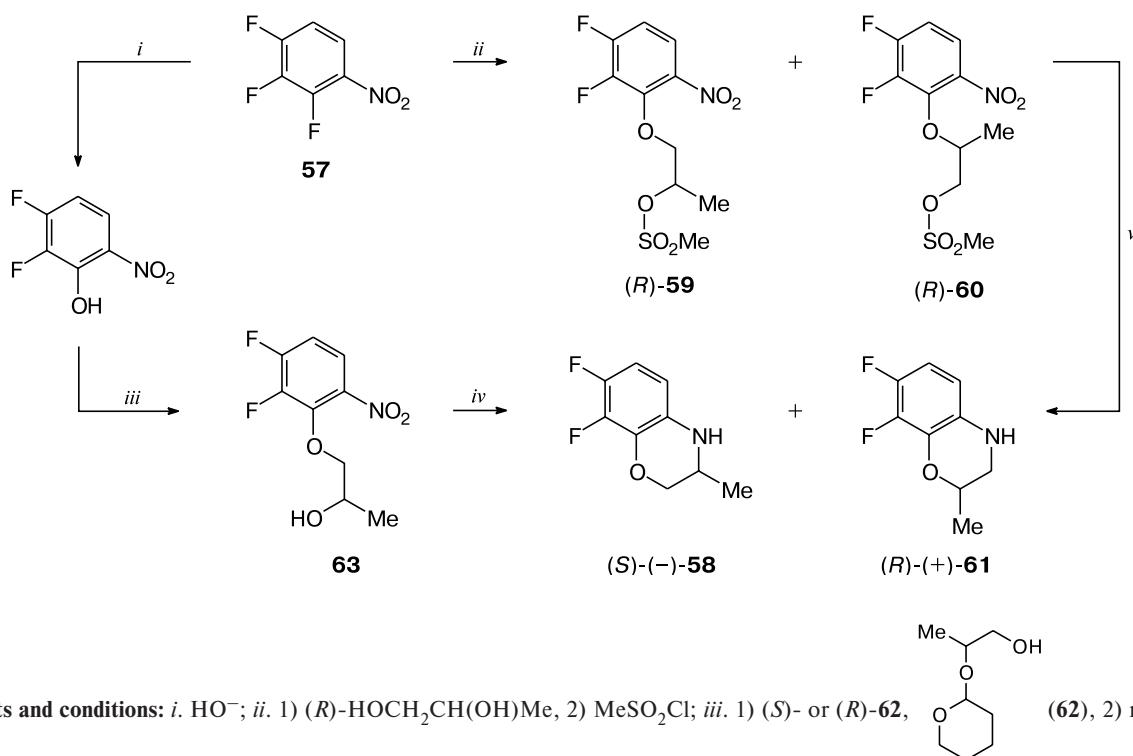
Scheme 21



Scheme 22

 $X = H, F; R = Me, Et, Ph$ Reagents and conditions: *i.* $(RCO)_2O$ or $PhCOCl$, NEt_3 ; *ii.* P_2S_5 ; *iii.* DMF , $80\text{--}120^\circ C$.

Scheme 23

Reagents and conditions: *i.* HO^- ; *ii.* $1)$ (R) - $HOCH_2CH(OH)Me$, $2)$ $MeSO_2Cl$; *iii.* $1)$ (S) - or (R) -62, $2)$ removal of the tetrahydropyranoyloxy group; *iv.* $1)$ $MeSO_2Cl$; $2)$ H_2 , Pd/C ; $3)$ Bu^4OK ; *v.* $1)$ H_2 , Pd/C ; $2)$ Bu^4OK .

Another route to [1,4]benzoxazine 58 involves condensation of 2,3-difluoro-6-nitrophenol (prepared by hydrolysis of 2,3,4-trifluoroniobenzene (57)) with (*S*)- or (*R*)-2-(tetrahydropyran-2-yloxy)propanol 62 (see Ref. 45). This is followed by elimination of the tetrahydropyranoyloxy group on ion-exchange resin, transformation of the resulting compounds 63 into mesylates, reduction of the nitro group, and cyclization in the presence of potassium *tert*-butoxide. For construction of fluoro-containing benzoxazines, a reaction of 2,3,4-trifluoroniobenzene (57) with (*R*)-2-(ethoxyethoxy)propan-1-ol or (*R*)-1-benzyloxypropan-2-ol is also used.⁴⁵

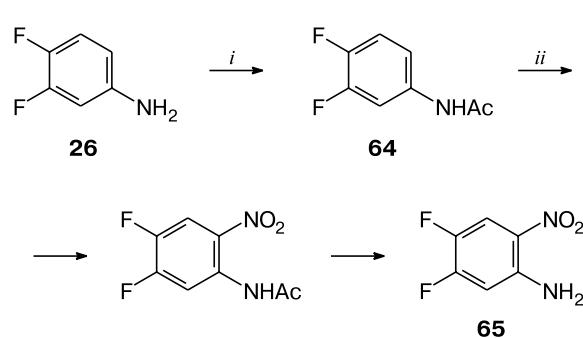
1.4. 4,5-Difluoro-2-nitroaniline

Nitration of 3,4-difluoroacetanilide 64 under usual conditions gives 4,5-difluoro-2-nitroaniline 65 (Scheme 24).

Diazotization of 4,5-difluoro-2-nitroaniline 65 followed by treatment of the diazonium salt with NaN_3 affords nitro azide 66, which can be easily transformed into 5,6-difluorobenzofuroxan (67), an important intermediate in the synthesis of fused azaheterocycles (Scheme 25).⁴⁶

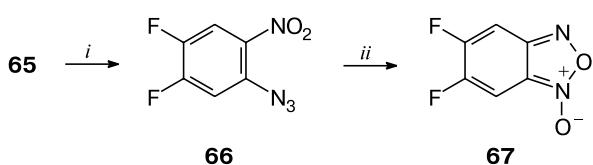
The Beirut reaction with 5,6-difluorobenzofuroxan (67) gives the corresponding 6,7-difluoroquinoxaline 1,4-dioxides 68. For instance, 2-cyano-6,7-difluoro-

Scheme 24



Reagents: *i.* Ac_2O ; *ii.* $\text{HNO}_3/\text{H}_2\text{SO}_4$.

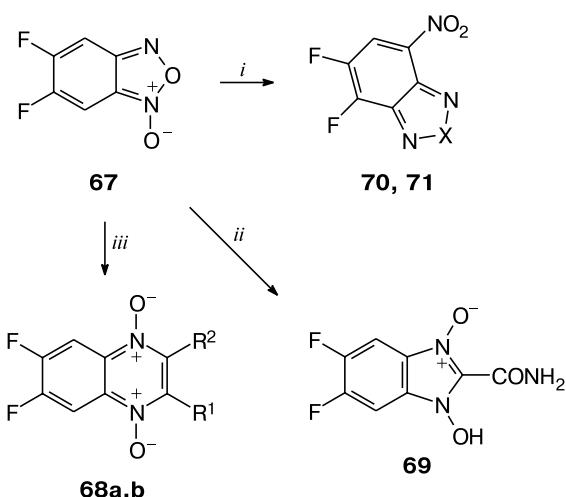
Scheme 25



Reagents: *i.* NaNO_2 , NaN_3 ; *ii.* AcOH .

3-phenylquinoxaline 1,4-dioxide (**68a**) is obtained by treating furoxan **67** with benzoylacetonitrile in the presence of triethylamine and 6,7-difluoro-2,3-dimethylquinoxaline 1,4-dioxide (**68b**) results from a reaction of furoxan **67** with ethyl methyl ketone in DMF in the presence of ammonia (Scheme 26).⁴⁷ The reaction of compound **67** with cyanoacetamide leads to 2-carbamoyl-5,6-difluoro-1-hydroxybenzimidazole 3-oxide (**69**)

Scheme 26



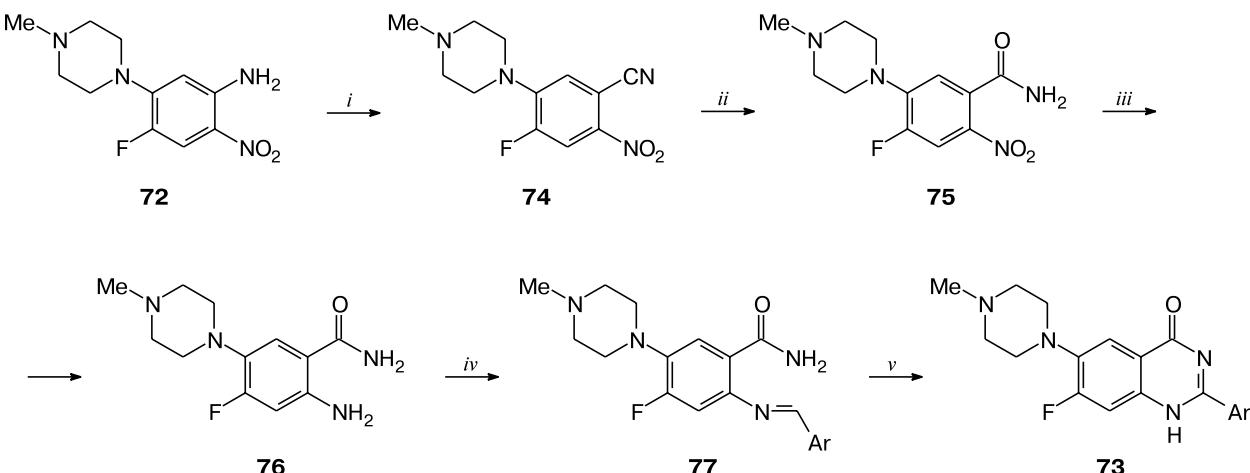
$\text{R}^1 = \text{CN}, \text{R}^2 = \text{Ph}$ (**68a**); $\text{R}^1 = \text{R}^2 = \text{Me}$ (**68b**)
 $\text{X} = \text{O}$ (**70**), $\text{N}-\text{Ar}$ (**71**)

i. NaNO_2 or ArN_2^+ ; *ii.* $\text{NCCH}_2\text{CONH}_2$; *iii.* $\text{R}^1\text{COCH}_2\text{R}^2$.

(Scheme 26).⁴⁸ The Boulton—Katritzky rearrangement gives nitrobenzofurazans **70** and nitrobenzotriazoles **71**.^{49,50}

4-Fluoro-2-nitroaniline **72** serves as a convenient synthon for the preparation of 2-aryl-7-fluoro-6-(4-methylpiperazin-1-yl)-4(3*H*)-quinazolin-4-ones **73** (Scheme 27).⁵¹ The diazonium salt obtained from 2-nitroaniline **72** can be transformed into 2-nitrobenzonitrile **74** using the Sandmeyer reaction. Hydrolysis of the latter in H_2SO_4 gives benzamide **75**. Reduction of the nitro group in compound **75** with SnCl_2 , followed by reactions of 2-aminobenzamide **76** with substituted

Scheme 27



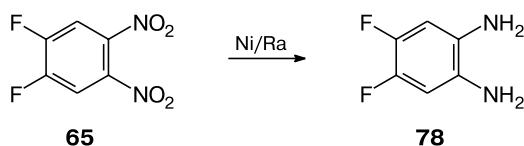
Reagents: $\text{KNO}_2/\text{H}_2\text{SO}_4$, CuCN ; *ii.* H_2SO_4 ; *iii.* SnCl_2/HCl ; *iv.* ArCHO/EtOH ; *v.* CuCl_2 .

benzaldehydes yields 2-arylideneamino-4-fluoro-5-(4-methyl-piperazin-1-yl)benzamides **77**. They undergo cyclization when heated in the presence of CuCl₂. It should be noted that 7-fluoroquinazolinones **73** exhibit antitumor activity.⁵¹

1.5. 4,5-Difluoro-1,2-phenylenediamine

Catalytic reduction of 4,5-difluoro-2-nitroaniline (**65**) gives 4,5-difluoro-1,2-phenylenediamine (**78**), an important intermediate in the synthesis of fluoro-containing benzazines and benzazoles (Scheme 28).⁵²

Scheme 28

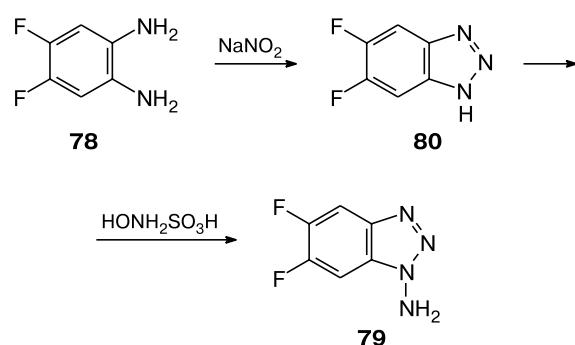


For instance, 1-amino-5,6-difluorobenzotriazole **79** is obtained from difluorophenylenediamine **78** by nitrosation in aqueous acetic acid and amination of product **80** with hydroxylamine-*O*-sulfonic acid at 65–80 °C (Scheme 29).⁵³

4,5-Difluoro-1,2-dehydrobenzene **81** is generated in room-temperature oxidation of amine **79** with lead tetraacetate in dry CH₂Cl₂ (Scheme 30). Under these conditions, 4,5-difluoro-1,2-dehydrobenzene **81** dimerizes into tetrafluorobiphenylene **82**. Using anthracene, tetraphenylcyclopentadienone, and furan as “traps” for *in situ* released **81**, one can obtain adducts **83**–**85**.⁵³

A reaction of 4,5-difluoro-1,2-phenylenediamine (**78**) with ethyl cyanoacetate gives 5,6-difluoro-1*H*-benzimid-

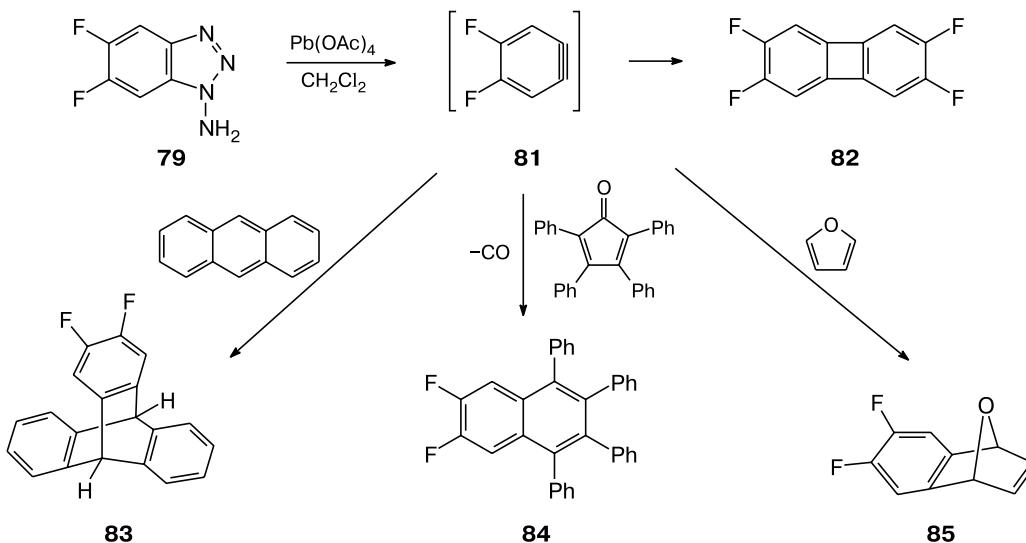
Scheme 29



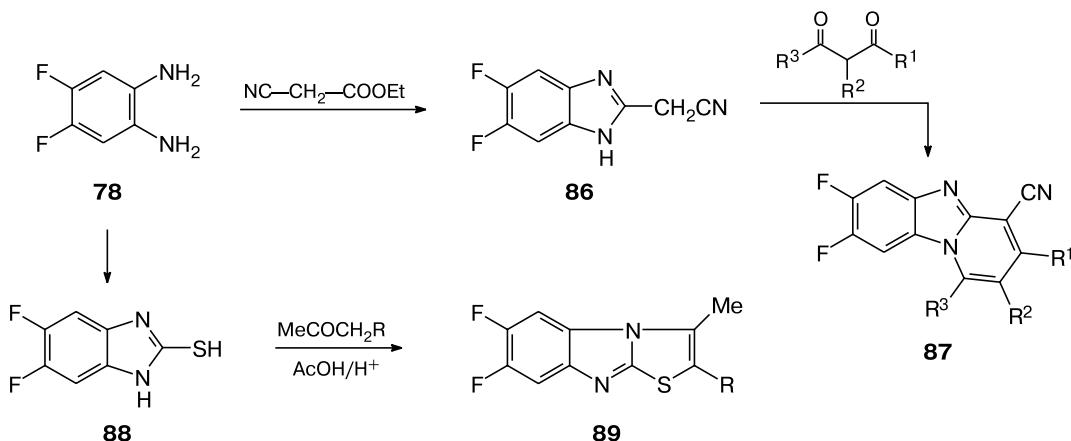
azol-2-ylacetonitrile (**86**). Its condensation with β-oxo esters yields pyrido[1,2-*a*]benzimidazoles **87** (Scheme 31). Some of compounds **87** have the effect on pathogenic *Orthopoxviruses*.⁵⁴ Condensation of diamine **78** with carbon disulfide in boiling ethanol in the presence of triethylamine gives 5,6-difluoro-2-mercaptopbenzimidazole **88**. This compound smoothly reacts with aliphatic and aromatic ketones at room temperature in the presence of catalytic amounts of H₂SO₄ to give 2-R-6,7-difluoro-3-methylbenzimidazo[2,1-*b*]thiazoles **89** (see Scheme 31).⁵⁵

6,7-Difluoroquinoxaline (**90**) prepared by cyclization of diamine (**78**) with glyoxal⁵² easily undergoes quaternization under the action of Et₃O⁺BF₄[−] (Meerwein reagent) in CH₂Cl₂. Using 1-ethyl-6,7-difluoroquinoxalinium tetrafluoroborate (**91**), one can obtain a series of fused quinoxalines (Scheme 32).⁵² For instance, reactions of quinoxalinium salt **91** with N-aryldithiocarbamic acids and β-oxo acids as enolates (bornyl acetooacetate and ethyl acetoacetate) give tetrahydrothiazolo[4,5-*b*]- and tetrahydrofuro[2,3-*b*]quinoxalines **92** and **93** (see Scheme 32).^{56,57}

Scheme 30

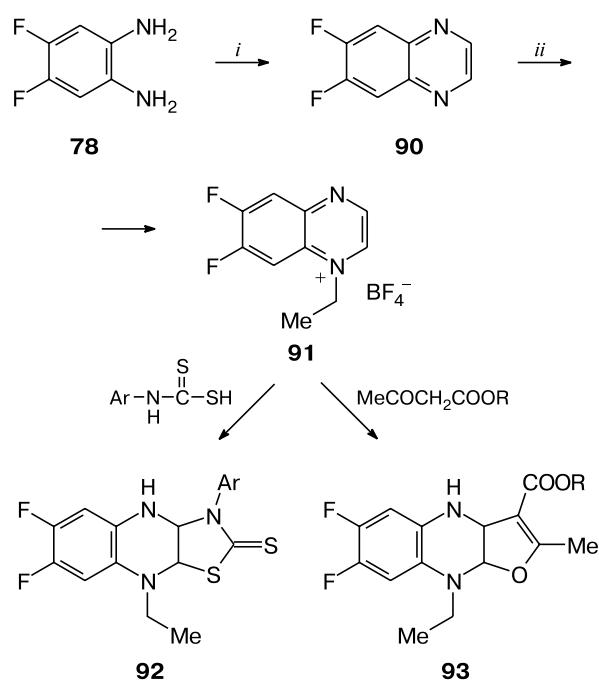


Scheme 31



R = H, Me, COMe, COPh, COOEt; R¹ = OH, Me, CH₂Br; R² = H, COOAik; R³ = H, OH, OAik, Cl, NR' R"

Scheme 32



R = Et, bornyl

i. H(O)C—C(O)H; ii. Meerwein's reagent.

1.6. 4-Fluoro-1,2-phenylenediamine

6-Fluoroquinoxalinones **94** containing a trifluoromethyl group in position 3 are known to inhibit the reverse transcriptase of HIV-1. To obtain them, 4-fluoro-1,2-phenylenediamine (**95**) is treated with hexafluoropropylene oxide (HFPO) under basic conditions, the NH fragment is protected, and quinoxalinone **96** is made to

react with lithium cyclopropylacetylide. Then the NH group in compound **97** is alkylated (or acylated) and the protective groups are removed (Scheme 33).⁵⁸

The synthesis of 3-anilinoquinoxalinone **98** from N-substituted 4-fluoro-1,2-phenylenediamine **99** involves cyclization of the latter with ethyl oxalate followed by treatment of the intermediate with POCl₃. The resulting quinoxaline **100** is used to obtain compound **98**, which is an efficient glycogen phosphorylase inhibitor (Scheme 34).⁵⁹

The synthesis of 6-fluoro-3-methyl-1*H*-quinoxalin-2-one (**101**) from N-substituted 4-fluoro-1,2-phenylenediamine **102** was proposed in Ref. 60. 4-Fluoroaniline (**10**) is acylated with α -chloropropionyl chloride, compound **103** is nitrated at the *ortho*-position relative to the amide group, and the nitro group in compound **104** is reduced. Cyclization of compound **102** in the presence of NaHCO₃ followed by oxidation of compound **105** with H₂O₂ yields product **101** (Scheme 35).

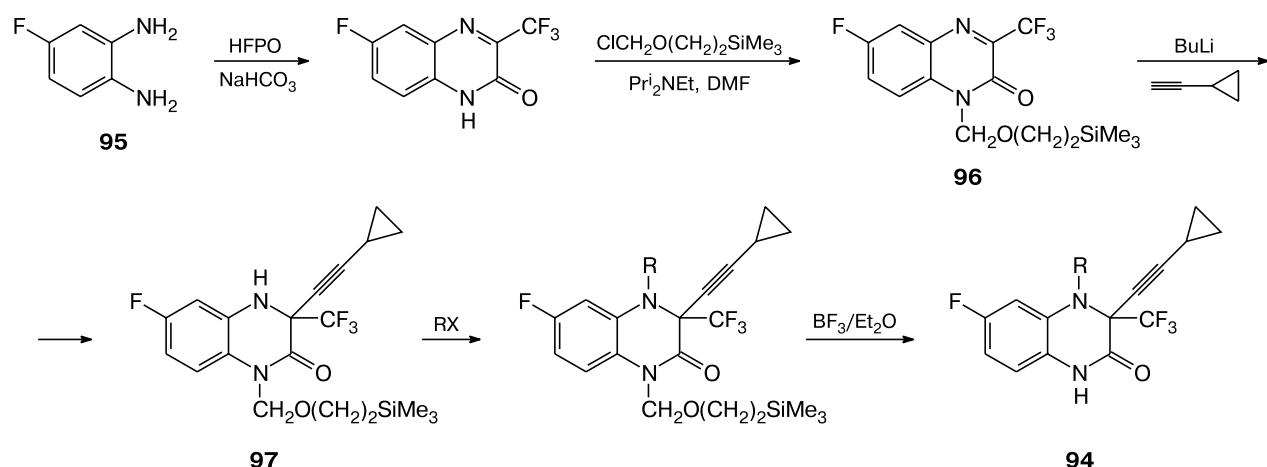
2. Fluoro-containing benzoic acid derivatives in the synthesis of benzoannulated azaheterocycles

2.1. Fluoro-containing benzoyl chlorides

Reactions of accessible fluoro-containing benzoyl chlorides with bifunctional nucleophiles provide a convenient route to a number of fluoro-containing heterocycles: fluoroquinolones, [*a*] - and [*b*] -annulated quinazolinones, and benzothiazinones.

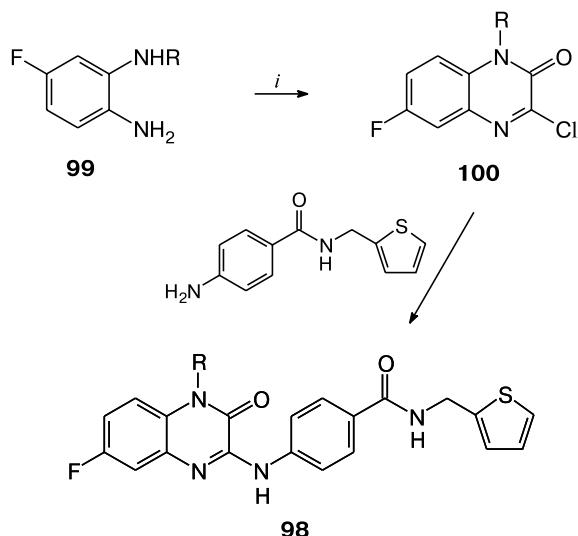
For instance, acylation of α -azahetarylacetonitriles **106** with acid chlorides **107a,b** in the presence of triethylamine gives cyano ketones **108**. Depending on the heterocyclic residue, compounds **108** can be transformed into thiazoloquinolones **109**, benzothiazoloquinolones

Scheme 33



R is allyl, cyclopropylmethyl, COOR_r, benzyl, propargyl, etc.

Scheme 34



i. 1) EtOOC—COOEt; 2) POCl₃, DMF.

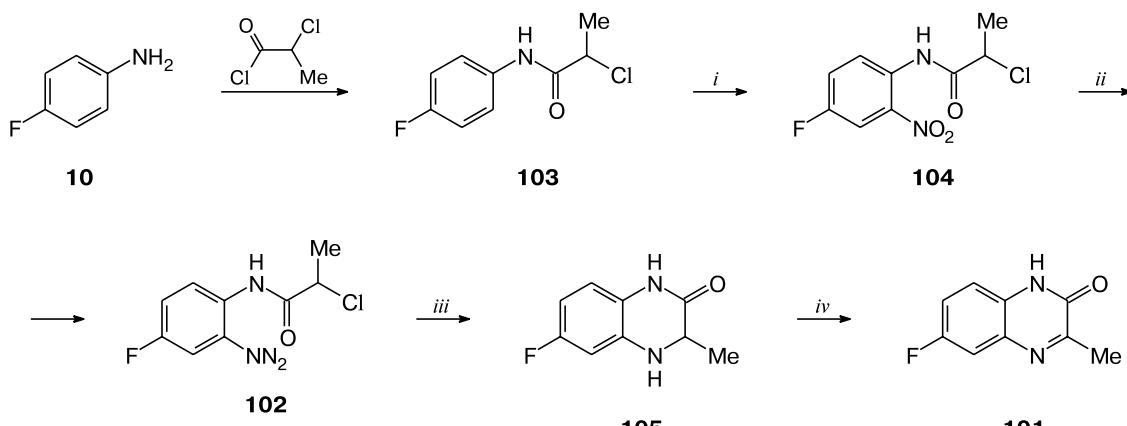
110, or benzimidazoquinolones **111** (heating in DMF), as well as into quinolinoquinolones **112** (reflux in acetonitrile in the presence of diazabicycloundec-7-ene (DBU)) (Scheme 36).^{61,62}

Quinolones **113** can be obtained by reactions of 2,4,5-trifluorobenzoyl chloride (**114**) with enamines **115** as C,N-dinucleophiles; intermediate acylated enamines resist isolation (Scheme 37).⁶³

Reactions of fluoro-containing benzoyl chlorides **116a,b** with *S*-ethylthiourea as an N,Nr-dinucleophile under mild conditions yield open-chain derivatives **117a,b**; when heated in DMF, they undergo cyclization into 1*H*-quinazolin-4-ones **118** (Scheme 38).⁶⁴

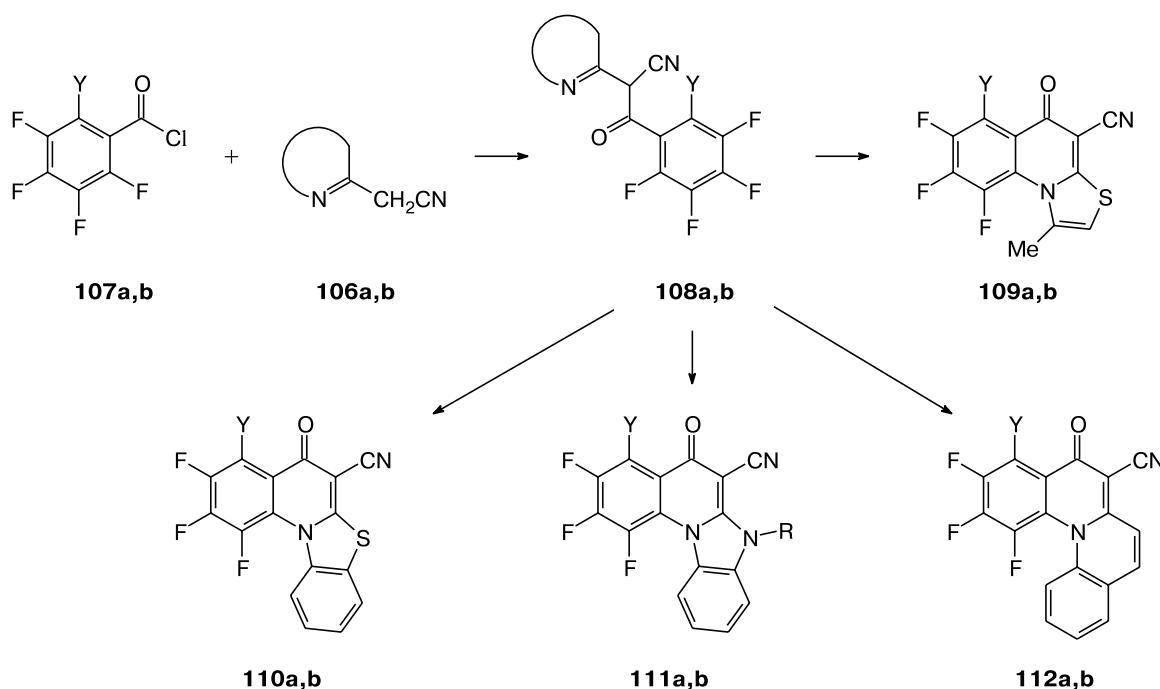
A distinctive feature of the reactions of tetrafluorobenzoyl chloride **107a** with 2-aminopyridines **119** is that acylation occurs at both endocyclic and exocyclic N atoms and that heating of diaryl derivatives **120** in boiling toluene in the presence of triethylamine gives

Scheme 35



Reagents: *i.* HNO₃/AcO or KNO₃/H₂SO₄; *ii.* Fe/AcOH, H₂O/DMF; *iii.* NaI/NaHCO₃; *iv.* H₂O₂/NaOH.

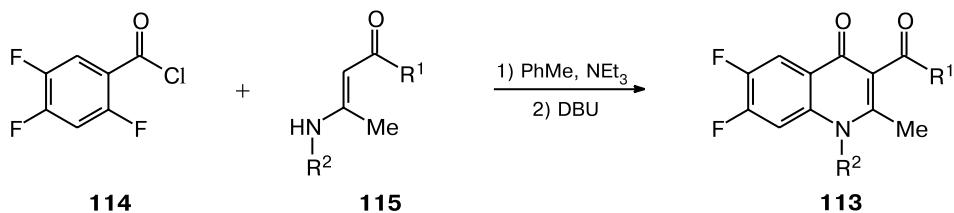
Scheme 36



106–112: Y = H (a), F (b)

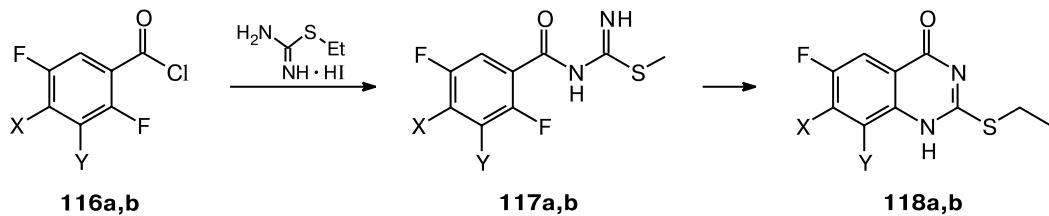
: pyridin-2-yl, quinolin-2-yl, 1*H*-benzimidazol-2-yl, 1-methylbenzimidazol-2-yl, benzothiazol-2-yl, 4-methylthiazol-2-yl.

Scheme 37



R¹ = Me, Ph; R² = Prⁿ, Ph

Scheme 38

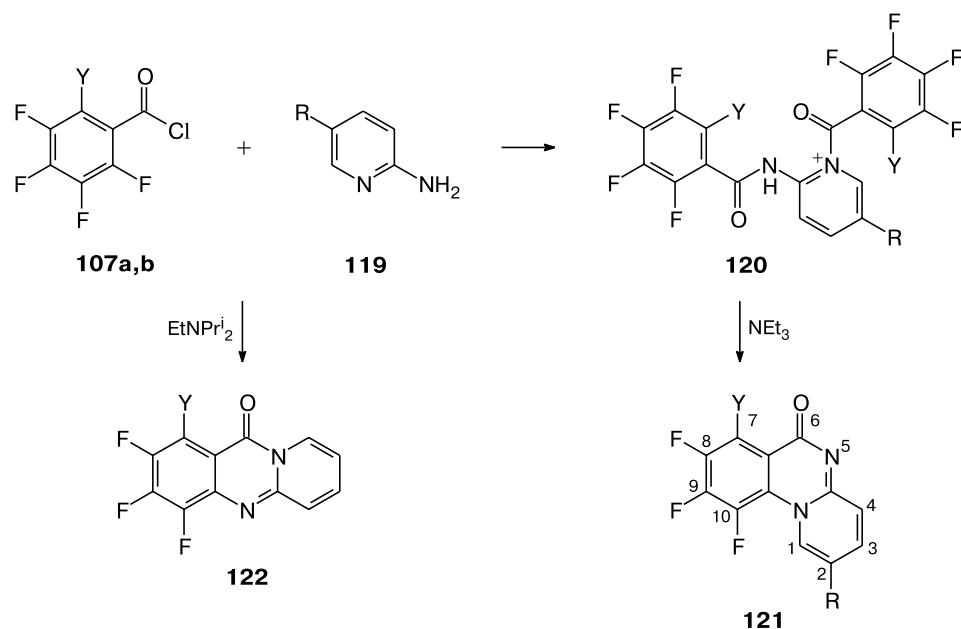


116–118: X = Y = F (a); X = Cl, Y = H (b)

pyridoquinazolinones **121** (Scheme 39).⁶⁵ Angular structure **121** (Y = H) is confirmed by the coupling constants *J*(F¹⁰,H¹) and *J*(F¹⁰,H²) in the ¹⁹F NMR spectrum and by 2D heteronuclear experiments (HetCOR and HMBC). With pentafluorobenzoyl chloride (**107b**), this reaction

yields a mixture of products **121** (Y = F) and **122** (Y = F). The use of ethyl(diisopropyl)amine as a base in reactions of polyfluorobenzoyl chlorides **107a,b** with 2-aminopyridine affords individual quinazolin-4-ones **122** (Y = H, F).⁶⁶

Scheme 39

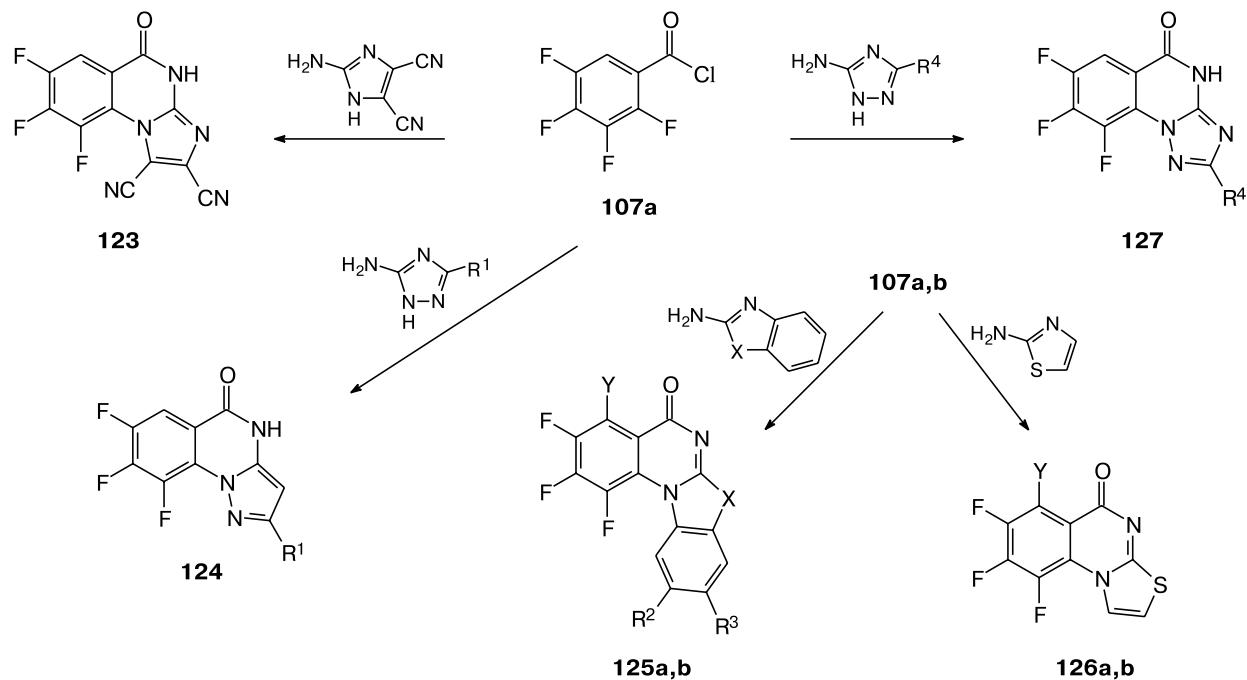


$R = \text{H, Me}; Y = \text{H, F}$

[*a*]-Annulated quinazolin-4-ones **123–127** are obtained from polyfluorobenzoyl chlorides **107a,b** and N,N'-dinucleophiles (aminoimidazoles, aminopyrazoles, aminotriazoles, or aminothiazoles). Heating of acid chlorides **107** with aminoimidazole, aminopyrazoles, or

aminotriazoles yields benzamides, the exocyclic N atom being acylated. Cyclization of benzamides into tricyclic derivatives **123**, **124**, and **127** occurs in boiling toluene in the presence of NEt_3 or in boiling acetonitrile in the presence of DBU (Scheme 40).⁶⁷ The angular structure of

Scheme 40

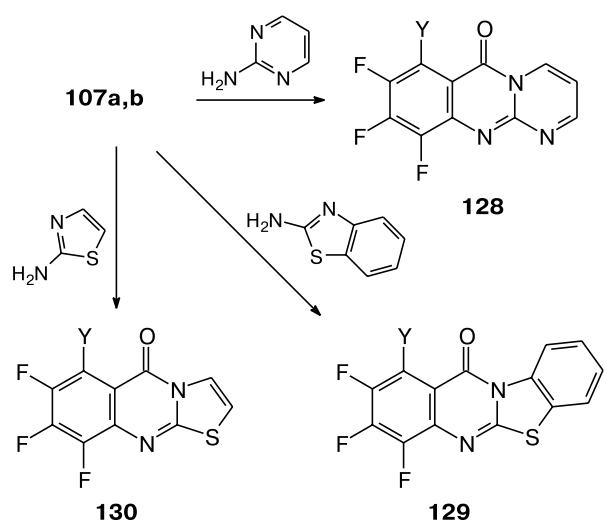


$Y = \text{H (a), F (b)}; X = \text{S, NH}; R^1 = \text{H, Ph}; R^2 = \text{H, F}; R^3 = \text{H, F, OMe}; R^4 = \text{H, CF}_3$

derivatives **125** and **126** obtained by cyclization of acylated heterylamines in diphenyl ether was confirmed by the spin-spin coupling of the F atom in position 8 of the quinazoline ring with the proton of the thiazole or benzazole fragment ($J = 4.5\text{--}6.8$ Hz).

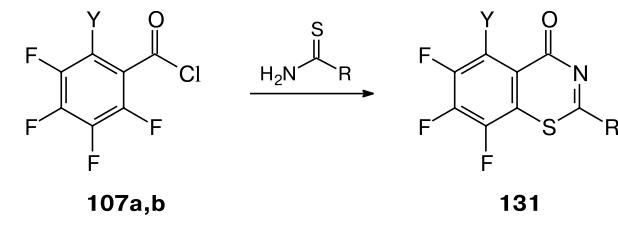
The synthesis of [b]-annulated 4(3*H*)-quinazolinones is described in Ref. 66. Compounds **128** -- **130** are obtained by room-temperature cyclocondensation of 2-aminoazaheterocycles with polyfluorobenzoyl chlorides **107** in CH_2Cl_2 in the presence of ethyl(diisopropyl)amine (Scheme 41). The linear structure of compound **129** was proved by X-ray diffraction analysis; structure **128** is confirmed by the coupling constant $^3J_{\text{C},\text{H}}$ for the carbonyl C atom and the proton of the pyrimidine ring.

Scheme 41



Heating of polyfluorobenzoyl chlorides **107a,b** with thioamides gives compounds **131** (Scheme 42); *N*-acylated intermediates cannot be isolated under these conditions.⁶⁸ It should be noted that this synthesis of various 2-substituted benzothiazinones is limited since many thioamides are not easily accessible.

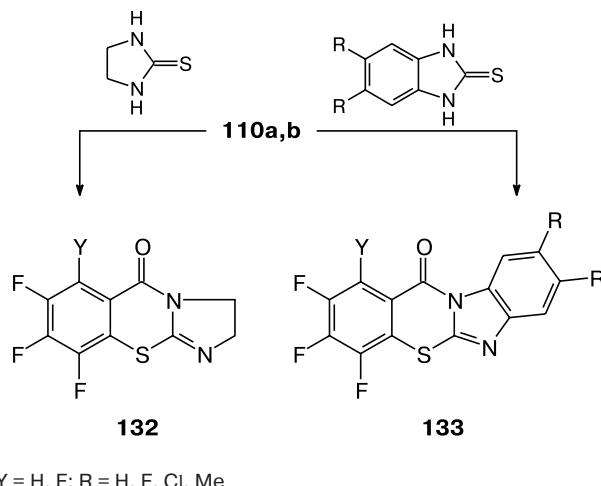
Scheme 42



Y = H (**a**), F (**b**), R = 4-Cl-C₆H₄, 4-Me-C₆H₄, 2-pyridyl

Reactions of acid chlorides **107a,b** with imidazolidine-2-thione in pyridine at 80 °C afford benzothiazinone **132**. Analogously, benzimidazole-2-thiones react with compounds **107a,b** under heating in toluene to give tetracyclic derivatives **137** (Scheme 43).⁶⁹

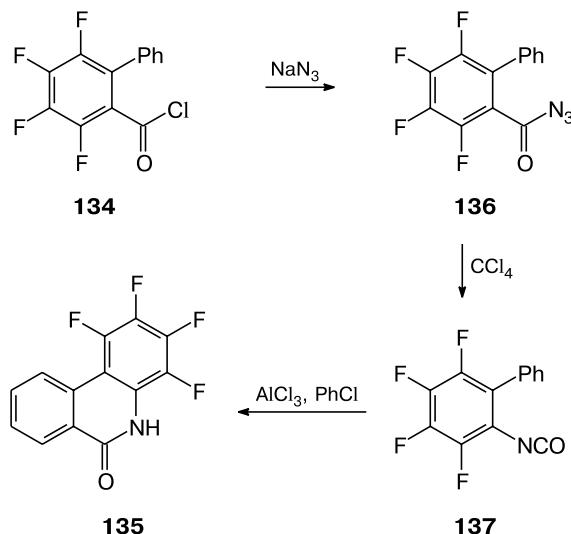
Scheme 43



Y = H, F; R = H, F, Cl, Me

Tetrafluorobenzoyl chloride **134** containing a phenyl substituent is employed for the synthesis of phenanthridin-6-one **135** (Scheme 44).⁷⁰ For this purpose, 3,4,5,6-tetrafluoro-2-phenylbenzoyl azide **136** (prepared from acid chloride **134**) is subjected to the Curtius rearrangement into isocyanate **137** and the latter undergoes cyclization into phenanthridin-6-one **135** on heating in the presence of AlCl₃ (see Scheme 44).

Scheme 44



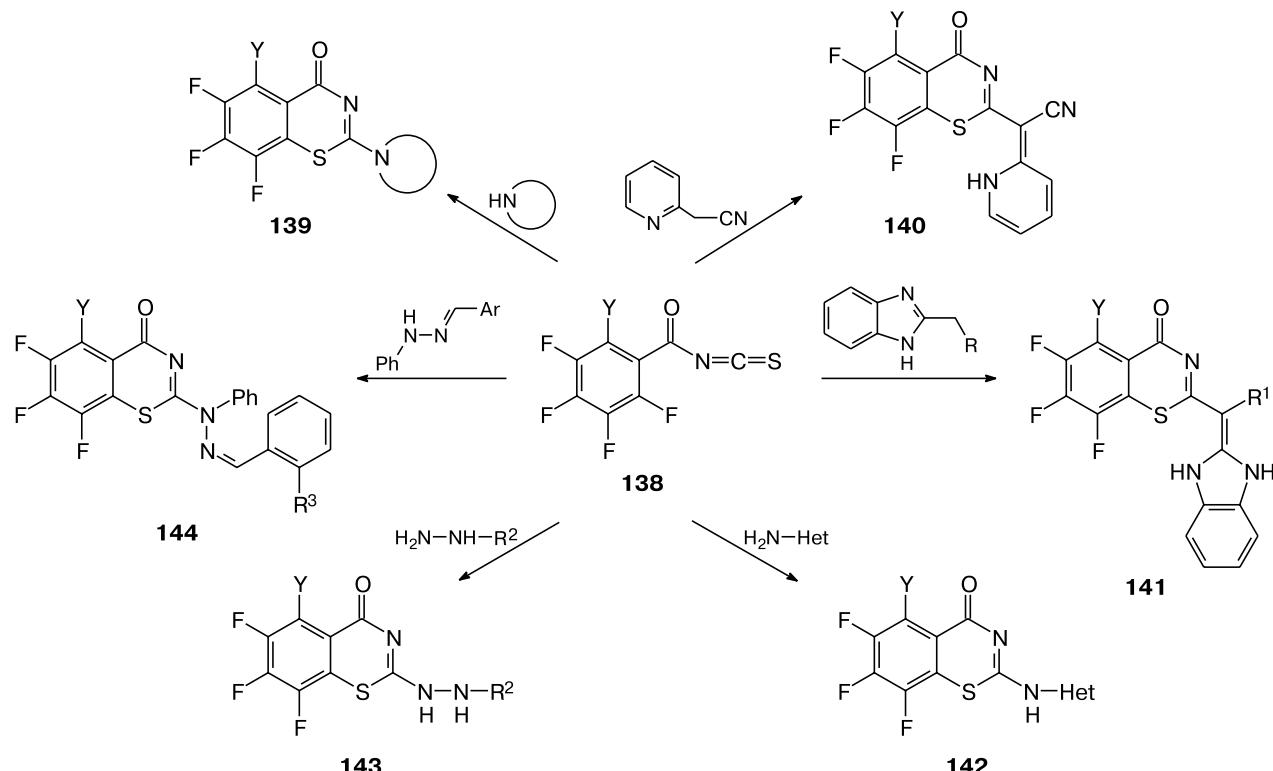
2.2. Polyfluorobenzoyl isothiocyanates

Polyfluorobenzoyl isothiocyanates **138** (prepared from appropriate benzoyl chlorides and ammonium thiocyanate) readily react with N- and C-nucleophiles at the C atom of the N=C bond, giving derivatives of thiocarbamic acid. Adducts of nucleophiles with isothiocyanates undergo intramolecular cyclization with replacement of the F atom to give 2-substituted [1,3]benzothiazin-4-ones **139**–**144** (Scheme 45).⁷¹ This method allows wide variation of the substituent in position 2 of benzothiazinones.

2.3. Fluoro-containing benzamides and benzonitriles

The synthesis of quinazolinones **145** from polyfluorobenzamides **146** involves heating of the latter with oxalyl chloride in the presence of amines and intramolecular cyclization of the resulting ureas **147** in the presence of potassium bis(trimethylsilyl)amide and catalytic amounts of 18-crown-6. Amination of quinazolinedione **145** with *O*-(2,4-dinitrophenyl)-hydroxylamine in the presence of NaH yields compound **148** (Scheme 46).⁷²

Scheme 45

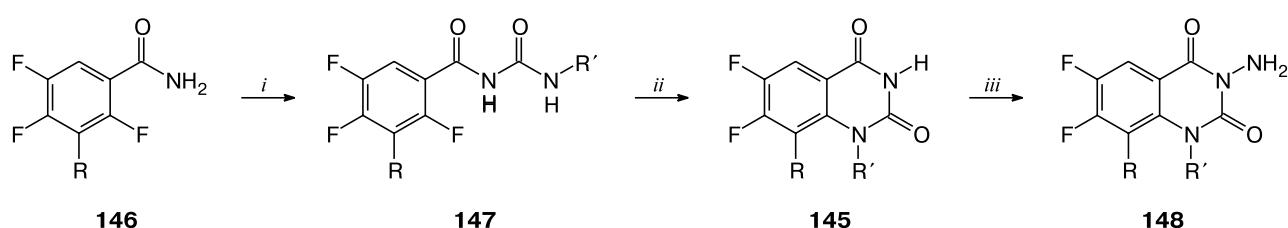


R¹ = COPh, CN; R² = 4-NO₂Ph, 2,4-dimethylpyridin-2-yl; R³ = OH, H

N—: morpholino, 4-ethoxycarbonylpiperazin-1-yl, hexamethyleneimino;

Het is pyridin-2-yl, 6-methylpyridin-2-yl, pyrimidin-2-yl, 4,6-dimethylpyrimidin-2-yl, 5-methylpyrazol-3-yl, 5-phenylpyrazol-3-yl.

Схема 46

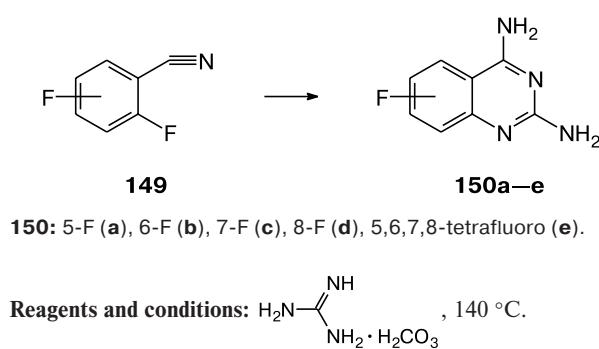


R' = CHMe₂, cyclo-C₃H₅, Ph; R = H, Me, OMe

Reagents: *i.* (COCl)₂, R'NH₂; *ii.* (Me₃Si)₂NK, 18-crown-6; *iii.* O₂N-C₆H₃(NO₂)₂-O-NH₂, NaH.

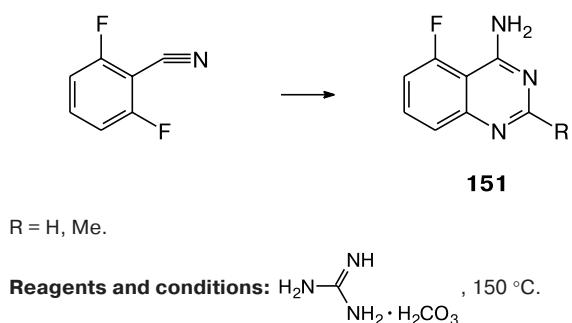
Reactions of *o*-fluorobenzonitriles with N,N'-di-nucleophiles seem to be one of the most rational synthetic approaches to fluoro-containing 4-aminoquinazolines. Indeed, reactions of 2,3-difluoro-, 2,4-difluoro-, 2,5-difluoro-, or 2,6-difluorobenzonitrile **149** and penta-fluorobenzonitrile with guanidine carbonate in dimethyl-acetamide at 140 °C give fluoro-containing 2,4-diamino-quinazolines **150a–e** in good yields (Scheme 47).⁷³

Scheme 47



Analogously, a reaction of 2,6-difluorobenzonitrile with formamidine acetate or acetamidine acetate in *N,N*-dimethylacetamide under nitrogen at 150 °C leads to 4-amino-5-fluoroquinazolines **151** (Scheme 48).⁷⁴

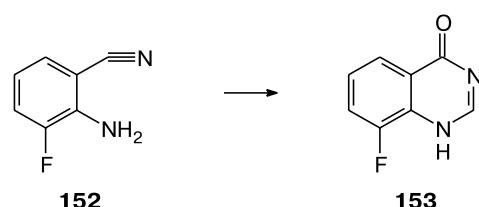
Scheme 48



Treatment of 2-amino-3-fluorobenzonitrile (**152**) with formic acid in the presence of H_2SO_4 gives 8-fluoro-1*H*-quinazolin-4-one (**153**) in 88% yield (Scheme 49).⁷⁵ Apparently, the formation of the quinazolinone involves condensation of intermediate 2-aminobenzamide with formic acid.

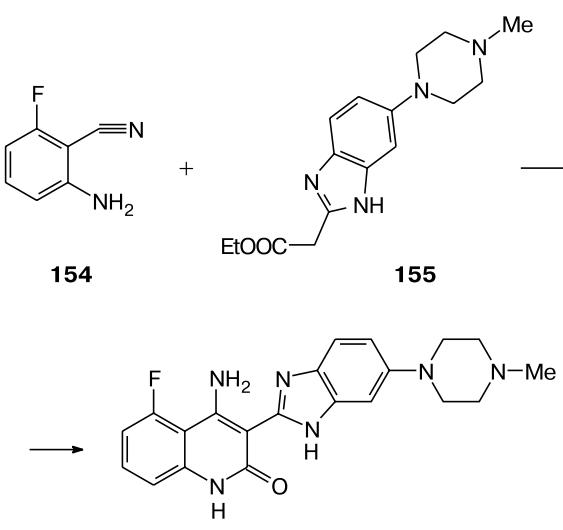
A reaction of 2-amino-6-fluorobenzonitrile (**154**) with ethyl 5-(4-methylpiperazin-1-yl)-1*H*-benzimidazol-2-ylacetate (**155**) in the presence of potassium bis(trimethylsilyl)amide produces 4-amino-5-fluoro-3-[5-(4-methylpiperazin-1-yl)-1*H*-benzimidazol-2-yl]-1*H*-quinolin-2-one (**156**) (Scheme 50).⁷⁶ Compound **156** is a dihydrofolate

Scheme 49



Reagents and conditions: HCOOH , H_2SO_4 , 110 °C.

Scheme 50



Reagents and conditions: $\text{KN}(\text{SiMe}_3)_2$, 40–62 °C, 1 h.

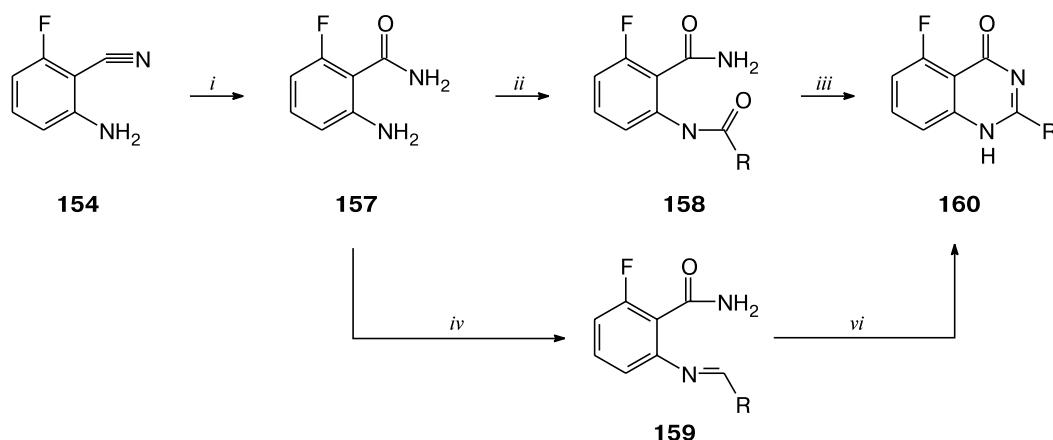
reductase inhibitor and has an effect on multiple myeloma.^{77,78}

Acid hydrolysis of nitrile **154** leads to 2-amino-6-fluorobenzamide (**157**), which allows widely varying the substituent in position 2 of quinazolinones. Two approaches to the synthesis of 5-fluoroquinazolin-4-ones were proposed. These involve reactions of amide **157** with acid chlorides (anhydrides) or aromatic (heterocyclic) aldehydes followed by cyclization of intermediates **158** and **159** into quinazolin-4-ones **160** (Scheme 51).⁷⁹

2.4. Polyfluorobenzoylacetic and acrylic acid esters

Fluoro-containing ethyl benzoylacetates **161** are important synthons for the preparation of [*a*]-annulated fluoroquinolones **162** and **163** (Scheme 51). Esters **161** are synthesized by base-catalyzed reactions of fluoro-benzoyl halides **164** with ethyl malonate followed by hydrolysis of the resulting malonate **165** and decarboxyla-

Scheme 51



R = Me, Ph, 2-, 3-, 4-pyridyl, 2-furyl, 2-hydroxyphenyl, 2-hydroxy-5-nitrophenyl, 4-methoxyphenyl, 2,4-dihydroxyphenyl, 4-fluorophenyl, 4-trifluoromethylphenyl.

Reagents and conditions: i. H_2SO_4 , 55 °C; ii. RC(O)Cl or $(\text{RCO})_2\text{O}$; iii. KOH ; iv. RCHO ; vi. CuCl_2 .

tion.⁸⁰ Compound **161** can also be obtained directly by reactions of fluorobenzoyl halides **164** with potassium ethyl malonate in the presence of Et_3N and MgCl_2 (Scheme 52).²⁵

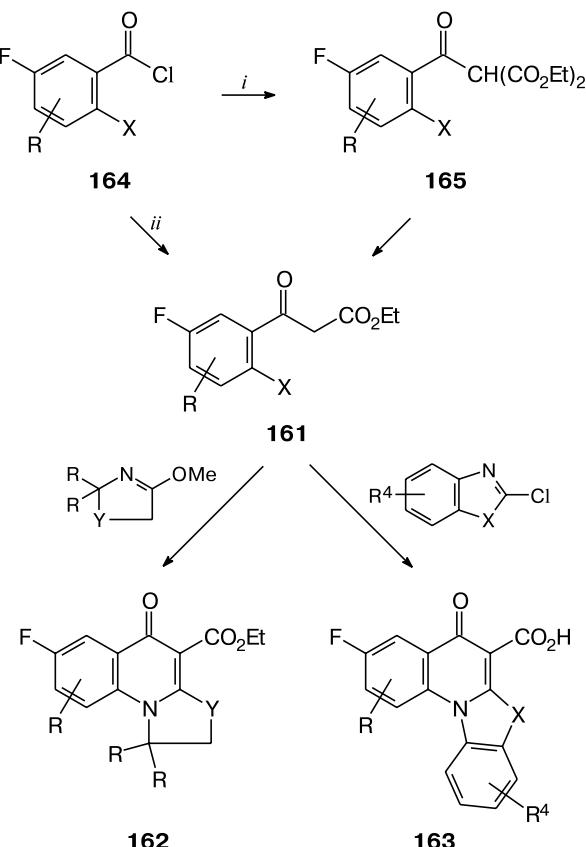
Condensation of compounds **161** with imino ethers gives tricyclic fluoroquinolones **162** (see Ref. 80). A reaction of compound **161** with 2-chlorobenzothiazole or 2-chlorobenzoxazole in diphenyl ether in the presence of NaH at 160 °C leads to benzothiazolo[3,2-*a*]quinolones and benzoxazolo[3,2-*a*]quinolones **163** (see Scheme 52).⁸⁰

The synthesis of fluoroquinolone derivatives from 3,4-difluoroaniline is discussed in Section 1. The second method of constructing the quinolone system, which employs fluoro-containing ethyl acetates **161** as the starting reagents, has a greater synthetic scope. Condensation of fluoro-containing ethyl acetates with diazonium salts and hydroxylamine affords aza and oxo analogs of quinolonecarboxylic acids (Schemes 53, 54). For instance, ethyl pentafluorobenzoylacetate (**166**) reacts with benzenediazonium chloride in the presence of KOH at 0 °C to give hydrazone **167**; treatment of this compound with a weak solution of KOH yields 3-ethoxycarbonyl-5,6,7,8-tetrafluoro-1-phenylcinnolin-4-one (**168**) (see Scheme 53).⁸¹ 3-Ethoxycarbonyl-6,7-difluoro-1-methylcinnolin-4-one is obtained analogously from ethyl 4,5-difluorobenzoylacetate; in this case, the cyclization of intermediate hydrazone is promoted by heating with NaH .⁸²

A reaction of ester **166** with hydroxylamine leads to oxime **169**, which undergoes cyclization into 5,6,7,8-tetrafluoro-1,2-benzoxazin-4-one **170** (see Scheme 54).⁸³

Ethyl 3-ethoxy-2-polyfluorobenzoylacrylate **171** is the most important synthon for the synthesis of fluoroquinolones and their annulated analogs. This ester is

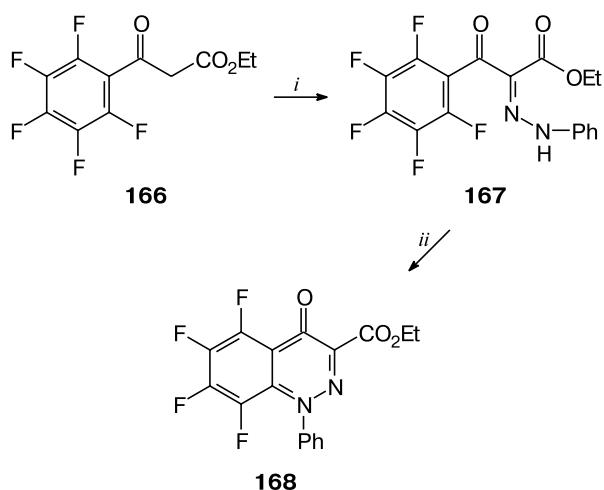
Scheme 52



X = S, O; R = H, Me; R = H, F, Cl, NR^1R^2 , AlkS, AlkO, etc.;
 $\text{R}^4 = \text{COOH}, \text{CN}, \text{AlkO}, \text{AlkS}, \text{NH}_2, \text{AlkNH}$

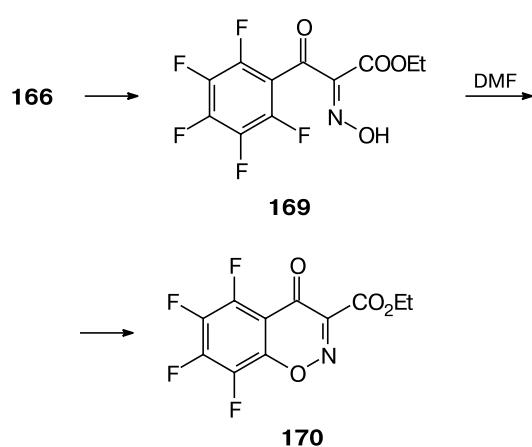
Reagents and conditions: i. $\text{CH}_2(\text{COOEt})_2$; ii. $\text{KOOC-CH}_2-\text{COOEt/NEt}_3/\text{MgCl}_2$.

Scheme 53



i. 1) KOH, 2) PhN₂Cl; *ii.* KOH.

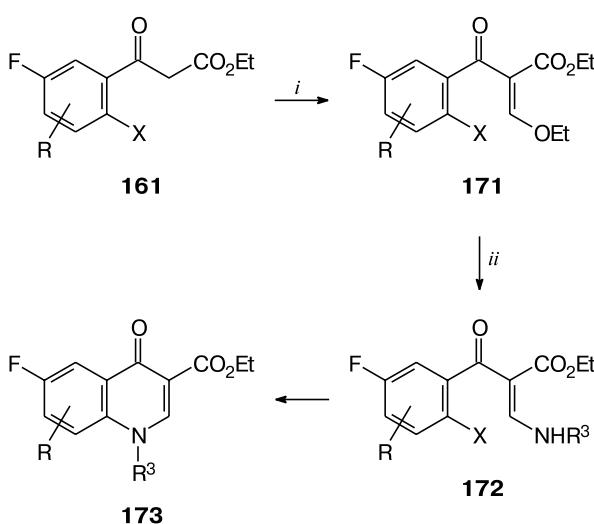
Scheme 54



prepared by condensation of ethyl aroylacetate **161** with ethyl orthoformate in acetic anhydride (Scheme 55).²⁵ Replacement of the ethoxy group in compounds **171** under the action of primary amines or hydrazines followed by base-catalyzed intramolecular cyclization of intermediates **172** yields the corresponding ethyl 6-fluoro-4-oxoquinoline-3-carboxylates **173** (see Scheme 55).²⁵

Room-temperature reactions of ethyl 3-ethoxy-2-[tetra(penta)fluorobenzoyl]acrylates **174** with hydrazides (**175**, X = O), thiohydrazides (**175**, X = S), or amidrazones (**175**, X = NH) in ethanol lead to acrylates **176**. Their intramolecular cyclization gives 1,3,4-thiadiazino-[6,5,4-*i,j*]-, 1,3,4-oxadiazino[6,5,4-*i,j*]-, and 1,2,4-triazino[5,6,1-*i,j*]annulated quinolones **177**. For instance, reflux of acrylate **176a** (X = S) in toluene for 2 h yields thiadiazinoquinolones **177a** (X = S); bicyclic intermediates cannot be isolated under these conditions (Scheme 56).^{84,85} Cyclization of acrylates **176** (X = O, NH) in toluene for

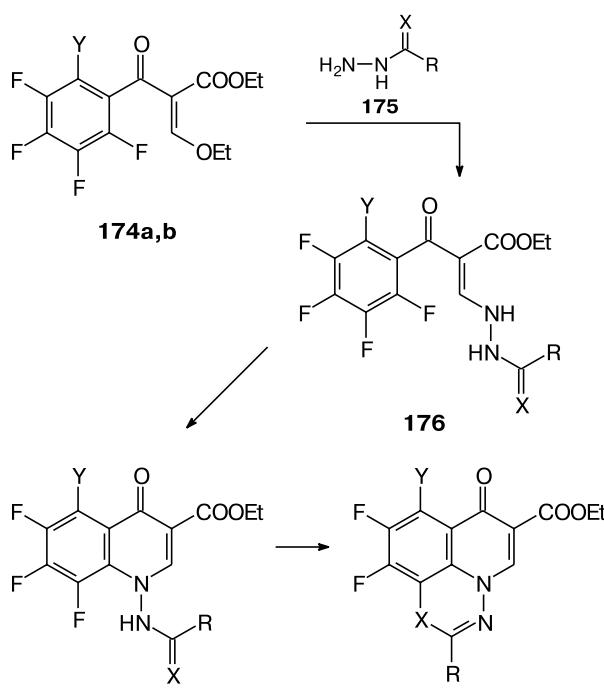
Scheme 55



i. HC(OEt)₃, Ac₂O; *ii.* R³NH₂

2–3 h gives bicyclic derivatives **178**. Heating of the latter in acetic anhydride produces triazinoquinolones **177** (X = NH), while heating in boiling toluene in the presence of K₂CO₃ or in acetonitrile in the presence of KF gives oxadiazinoquinolones **177** (X = O) (Scheme 56).^{86–88}

Scheme 56



174, 176–178: Y = H (**a**), F (**b**); X = S, O, NH; R = Ar, Het, cycloalkylimino

Room-temperature reactions of ethyl 3-ethoxy-2-[tetra(penta)fluorobenzoyl]acrylates **174a,b** with heterylhydrazines in toluene yield compounds **179a–g** (Scheme 57).⁸⁹ The tendency of the latter toward amino-imine tautomerism allows cyclization with [i,j]-annulation of quinolines to the triazine ring. Pentacyclic derivatives **180a–g** are obtained by cyclization of acrylates **179a–g** in boiling acetonitrile in the presence of DBU, triethylbenzylammonium chloride (TEBAC), or KF, or by heating in dioxane in the presence of NaH. When acrylate **179a** was heated with TEBAC in acetonitrile, bicyclic derivative **181a** was isolated (Scheme 57).⁸⁹

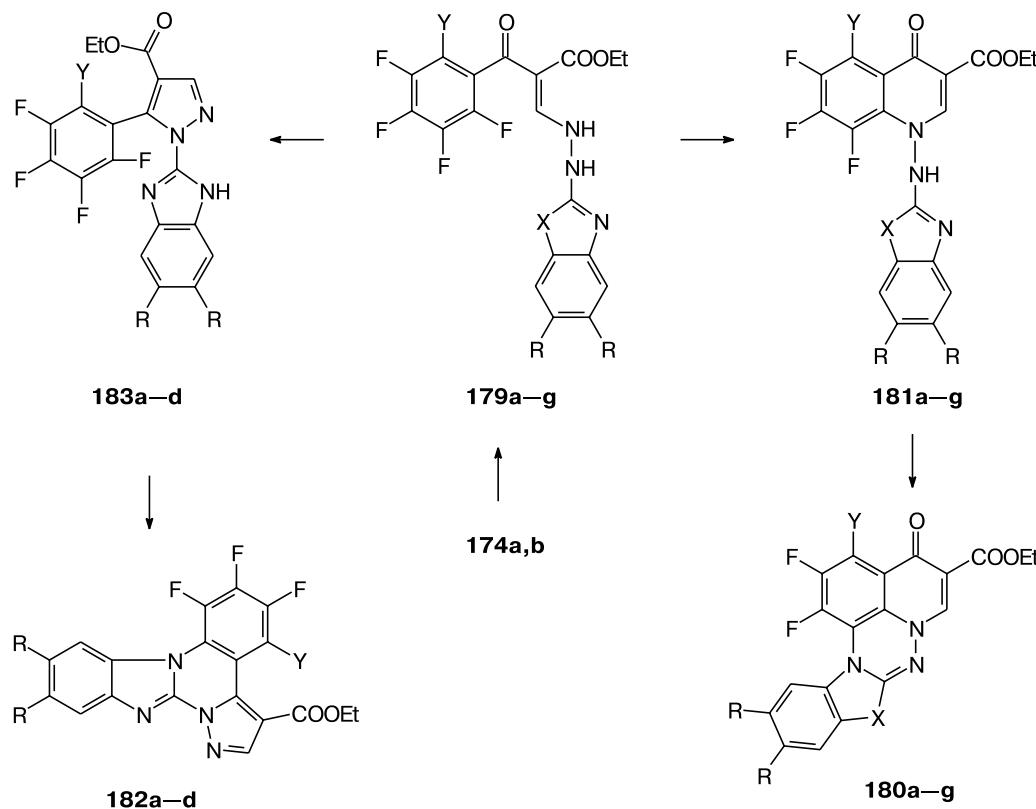
Alternative transformations of 3-benzazolylhydrazino-2-polyfluorobenzoylacrylates **179** involve the carbonyl group. For instance, reflux of acrylates **179** with KF in acetonitrile gives benzimidazo[1,2-*a*]pyrazolo[1,5-*c*]-quinazolines **182a–d**. Pyrazolyl intermediates **183a–d** can be isolated after short-time heating of acrylates **179a–d** in acetonitrile; heating of compounds **183** with KF for 4 h yields products **182a–d** (Scheme 57).^{90,91}

Analogously, room-temperature reactions of acrylates **174** with aminoazoles **184** in ethanol give ethyl 3-azolyl-amino-2-polyfluorobenzoylacrylates **185**, which undergo cyclization in boiling acetonitrile in the presence of KF. For instance, 1-substituted quinolone **186** is obtained from ethyl 2-pentafluoro-3-(pyrazol-3-yl)aminobenzoylacrylate **185b**. Cyclization of tetrafluorobenzoyl derivatives **185a,c** follows a different pathway leading to pyrazolo[1,5-*a*]pyrimidines **187a,b** (Scheme 58).⁹²

The synthesis of difluoro-1,2,4-oxadiazino[i,j]-quinolone **188** involves cyclization of acrylate **189** in DMF in the presence of K_2CO_3 (Scheme 59). Compound **189** is prepared by heating of ethyl 3-ethoxyacrylate **174a** with acetamide oxime **190** in acetic anhydride (see Scheme 59).⁹³

Methyl 3-methoxy-2-pentafluorobenzoylacrylate (**191**) reacts with 4-amino-5-phenyl-4*H*-1,2,4-triazole-3-thiol (AMT) to give compound **192**. Heating of the latter in DMF at 120 °C in the presence of K_2CO_3 yields quinolone **193** (Scheme 60).⁹⁴

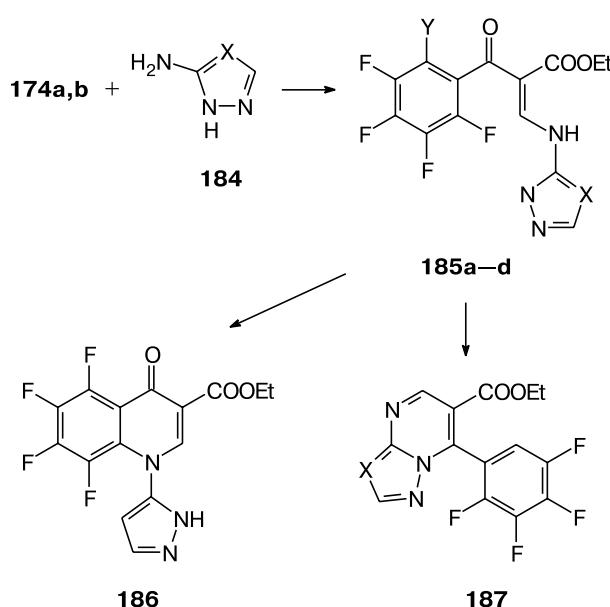
Scheme 57



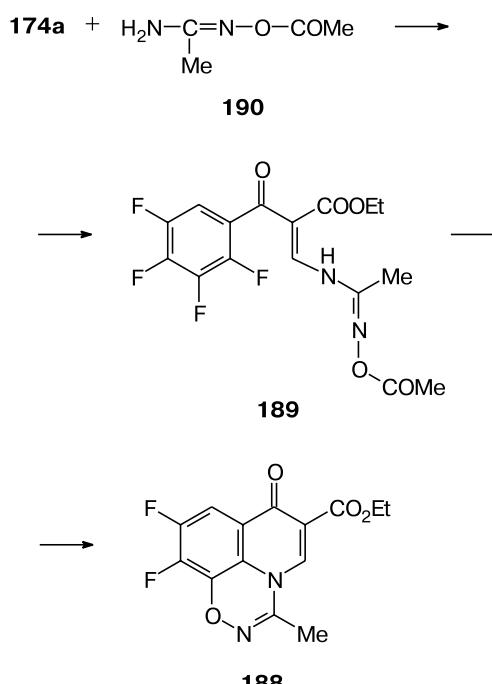
179–181: X = NH, R = F, Y = F (**a**), X = NH, R = F, Y = H (**b**), X = NH, R = Y = H (**c**), X = NH, Y = F, R = NO₂ (**d**), X = NH, Y = F, R = Br (**e**); X = S, R = H, Y = F (**f**), R = H, Y = H (**g**)

182, 183: Y = H, R = H (**a**), R = Br (**b**), R = F (**c**), Y = F, R = F (**d**)

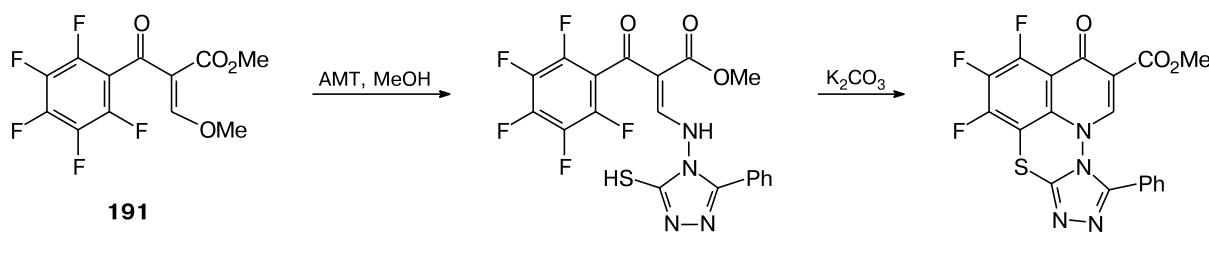
Scheme 58

**184:** X = CH (**a**), N (**b**);**185:** X = CH, Y = H (**a**), F (**b**), X = N, Y = H (**c**), F (**d**)

Scheme 59



Scheme 60



2.5. Fluoro-containing derivatives of anthranilic and 2-nitrobenzoic acids

Derivatives of benzoic acids bearing an amino or nitro group in the *ortho*-position are widely used as synthons for the preparation of fluoro-containing benzazines.

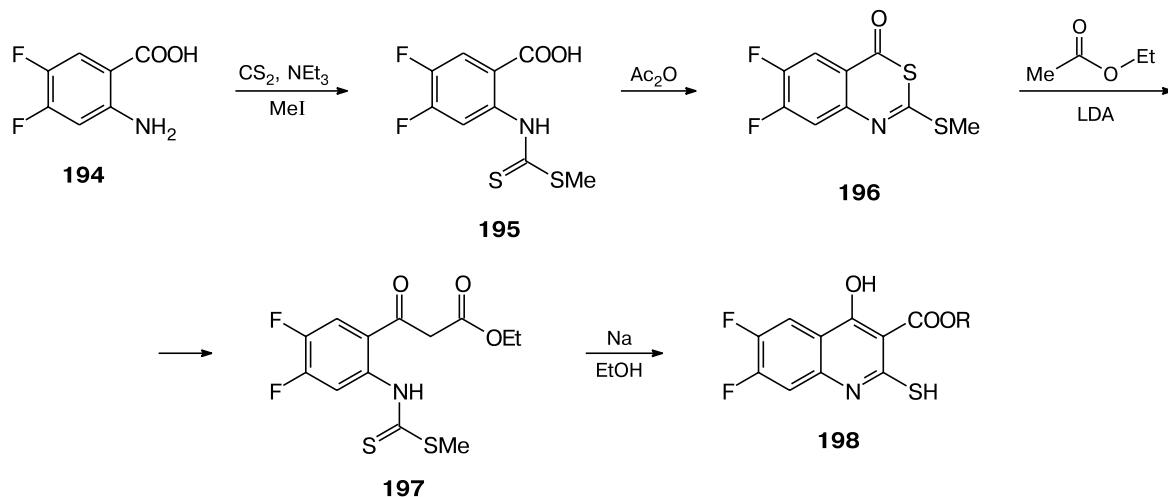
One of the key synthons is 4,5-difluoroanthranilic acid **194**. A reaction of compound **194** with carbon disulfide in the presence of triethylamine followed by methylation with methyl iodide gives dithiocarbamate **195**; treatment of the latter with acetic anhydride affords 4*H*-[3,1]-benzothiazin-4-one **196** in high yield. Treatment of compound **196** with a lithium salt of ethyl acetate gives β -oxo ester **197**, which undergoes cyclization into 2-mercaptopquinoline **198** under the action of sodium ethoxide in ethanol (Scheme 61).⁹⁵

Anthranilic acid is known to be a convenient synthon for construction of the quinazoline system.⁹⁶ Using fluoroanthranilic acids, one can also synthesize fluoro-containing quinazolines according to well-developed

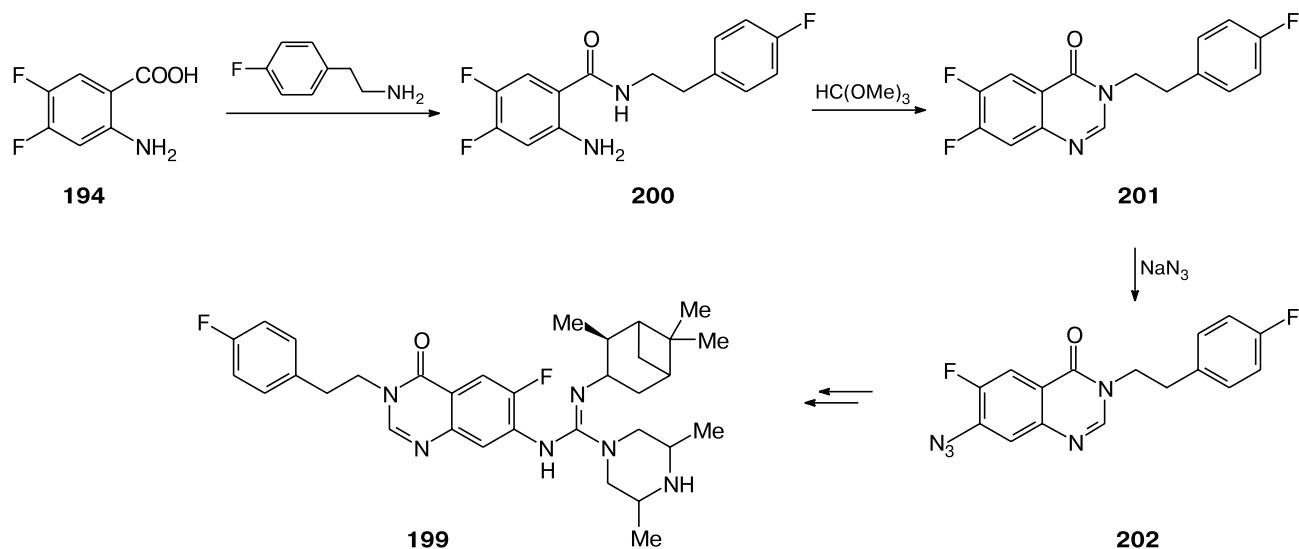
procedures. This approach allows a greater variation of the substituents in the heterocycle compared to the synthesis of quinazolines from benzoyl chlorides (Section 2.1). This can be illustrated with the synthesis of 6-fluoro-7-guanidinoquinazolinones **199** (Scheme 62). 2-Cyclization of aminobenzamide **200** (prepared from acid **194**) with trimethyl orthoformate gives 3-substituted quinazo-linone **201** transformed further into azide **202**. The latter is treated with (1*S*,2*S*,3*S*,5*R*)-2,6,6-trimethylbicyclo-[3.1.1]heptan-3-yl isocyanate in THF in the presence of PPh₃ and then with (6*S*,2*R*)-2,6-dimethylpiperazine. The resulting quinazoline **199** is biologically active and can be used to treat obesity and type II diabetes.⁹⁷

A reaction of *o*-phenylenediamine with 5-fluoroanthranilic acid (**203**) in the presence of polyphosphoric acid yields 2-(2-aminophenyl)benzimidazole **204**. The latter can react with a second molecule of acid **203** to give 6-(2-aminophenyl)benzimidazo[1,2-*c*]quinazoline **205** (Scheme 63).⁹⁸

Scheme 61



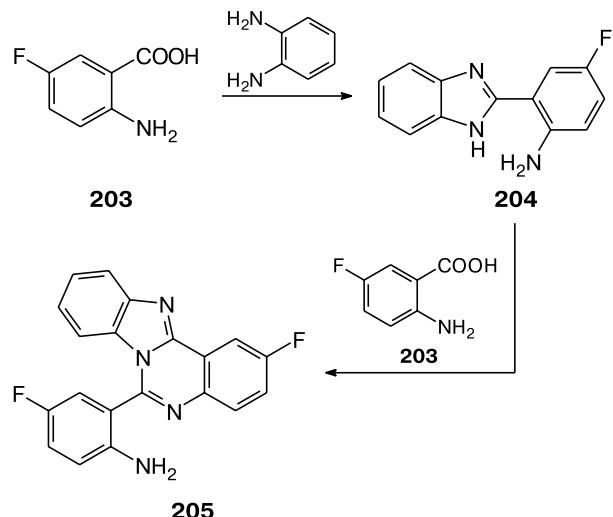
Scheme 62



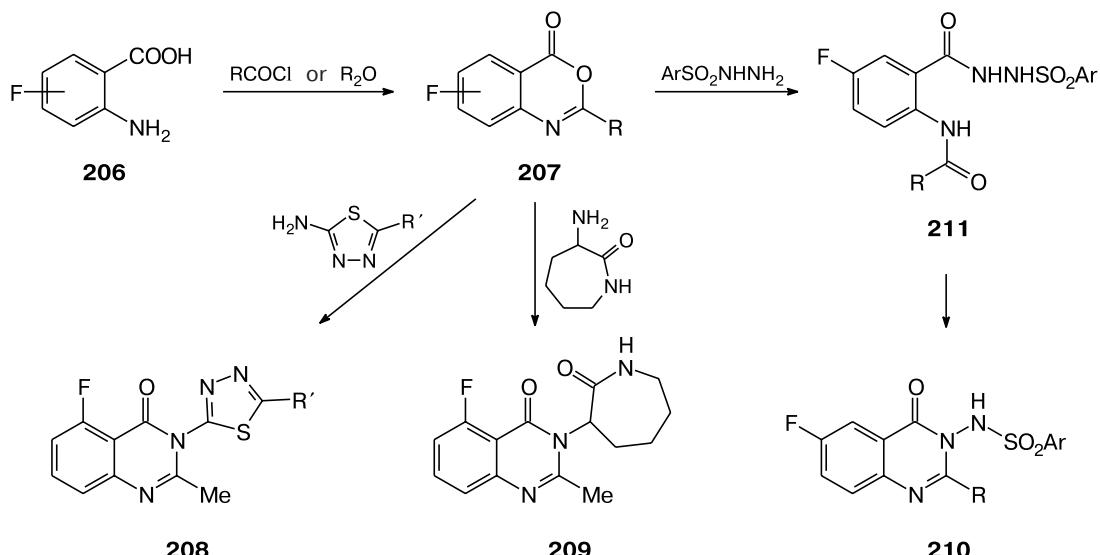
Heating of monofluoroanthranilic acids **206** with acetic anhydride or carboxylic acid chlorides leads to 2-substituted [3,1]benzoxazin-4-ones **207**. Reactions of 5-fluoro-2-methyl-3,1-benzoxazin-4-one with 2-amino-1,3,4-thiadiazoles and (*D,L*)- α -amino- ϵ -caprolactam give 3-substituted quinazolin-4(3*H*)-ones **208** and **209**, respectively (Scheme 64).⁹⁹ Fusion of arenesulfonylhydrazides with 6-fluoro-3,1-benzoxazines **207** at 130 °C for 30 min or keeping of the reagents in DMF at room temperature for 22 h affords a number of sulfonamides **210** (see Ref. 100). The reaction proceeds through open-chain intermediates **211**, which can be isolated under mild reaction conditions.

In some cases, construction of fluoroquinazolinones involves the formation of intermediates containing an isothiocyanate group. For instance, cyclization of methyl anthranilate **212** with isothiocyanates and that of substituted 2-(methoxycarbonyl)phenyl isothiocyanates **213**

Scheme 63

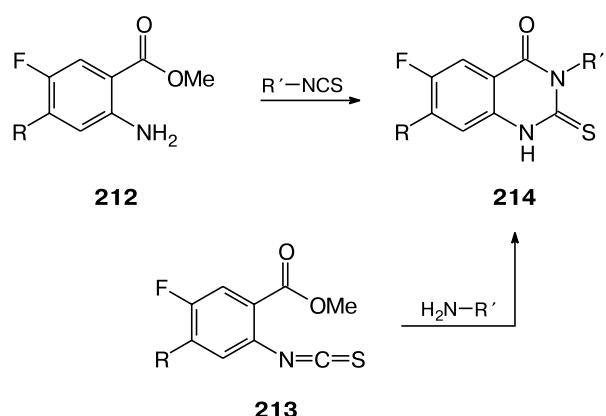


Scheme 64



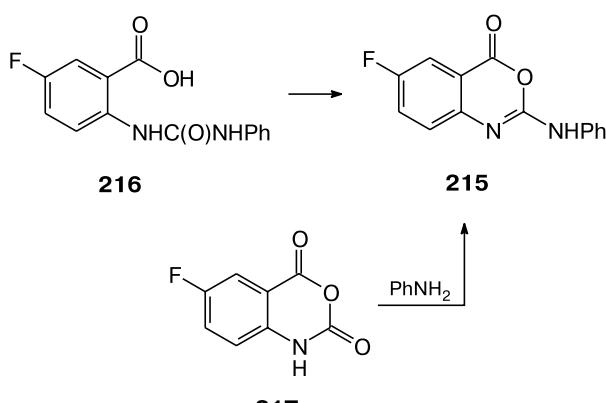
R = Me, PhCH_2 , PhCH_2CH_2 , thiophen-2-ylmethyl

Scheme 65



R = H, Cl, COOMe; R' = $\text{Et}_2\text{NCH}_2\text{CH}_2$, cyclohexyl, CH_2Ph , $2-\text{H}_2\text{NC}_6\text{H}_4$

Scheme 66

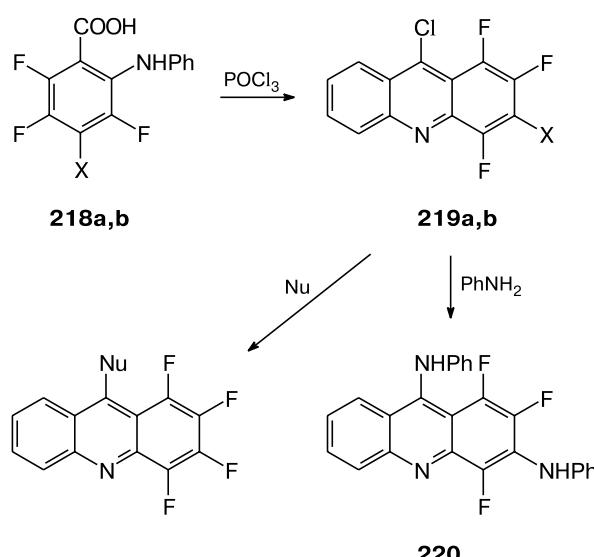


with primary amines lead to 4-oxo-1,2,3,4-tetrahydroquinazoline-2-thiones 214 (Scheme 65).¹⁰¹

It is reported that 6-fluoro-2-phenylamino-4*H*-3,1-benzoxazin-4-one 215 obtained by cyclization of compound 216 or by condensation of benzoxazinedione 217 with aniline (Scheme 66) is used to treat obesity.¹⁰²

Reactions of 2-anilinopolyfluorobenzoic acids 218 with POCl_3 give 9-chloro-1,2,4-trifluoro- and 9-chloro-1,2,3,4-tetrafluoroacridines 219a,b, in which the Cl atom is replaced by a nucleophile. Compound 219b can be easily transformed into compound 220 in the presence of aniline (Scheme 67).¹⁰³

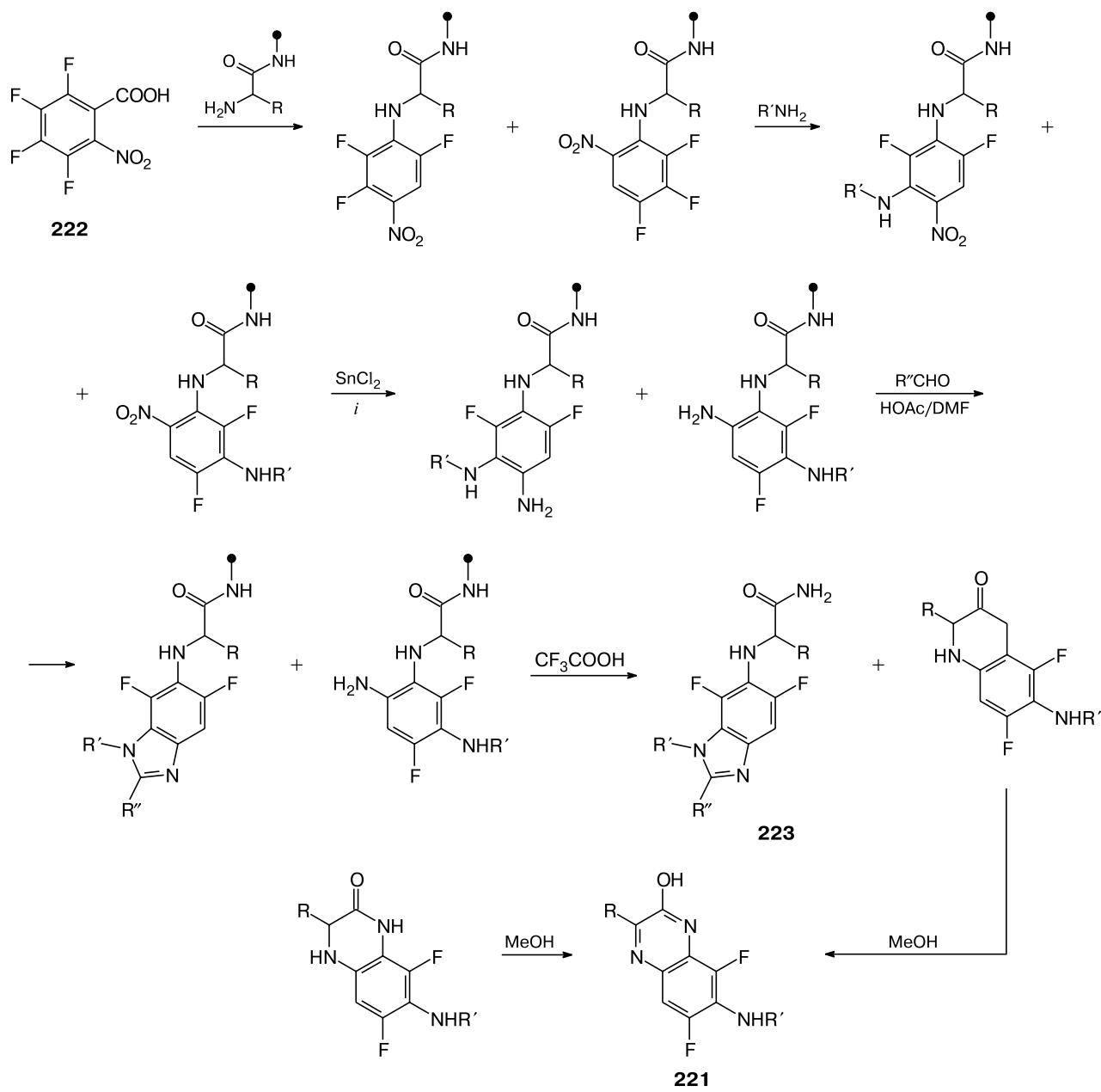
Scheme 67



218, 219: X = H (a), X = F (b)

Nu = MeNH , Me_2N , PhNH , NH_2NH , PhNNHNH , $\text{C}_6\text{F}_5\text{NHNH}_2$.

Scheme 68

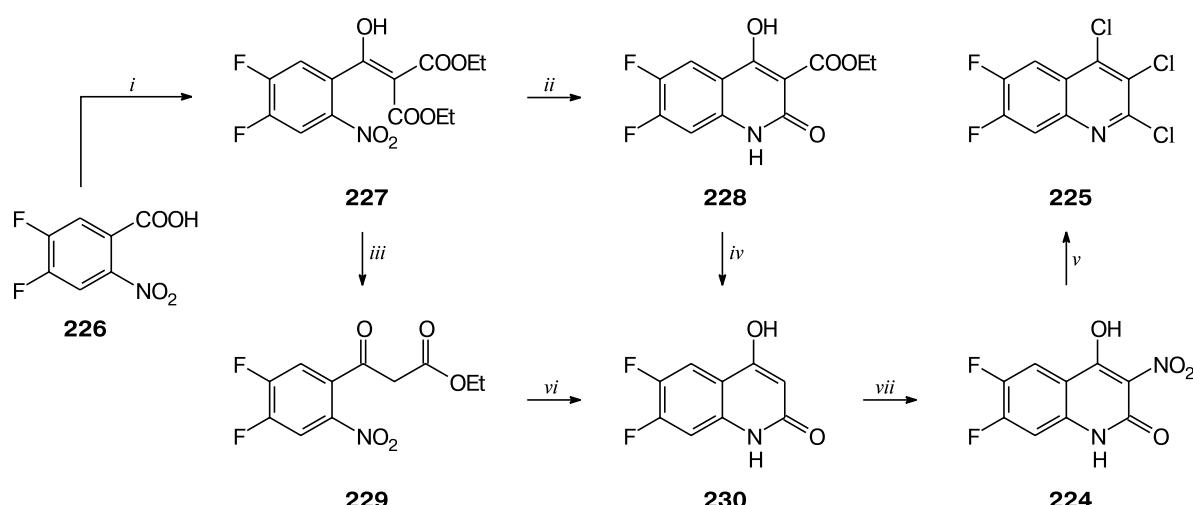


The solid-state synthesis of fluoro-containing 2-hydroxyquinoxalines **221** from 2,3,4,5-tetrafluoro-6-nitrobenzoic acid (**222**) was considered in Ref. 104 (Scheme 68); another product is 4,6-difluorobenzimidazole **223**.

The synthesis of 6,7-difluoro-4-hydroxy-3-nitroquinolin-2-one (**224**) and 2,3,4-trichloro-6,7-difluoroquinoline (**225**) from 4,5-difluoro-2-nitrobenzoic acid (**226**) is shown in Scheme 69. Malonate **227** is smoothly transformed into 4-hydroxyquinolone **228** in the presence of

NaBH_4 in a basic medium. Treatment of compound **228** with *p*-toluenesulfonic acid gives ethyl 3,4-difluoro-6-nitrobenzoylacetate (**229**). Catalytic reduction of the nitro group in the latter on Pd/C in ethanol results in cyclization into 6,7-difluoro-4-hydroxy-quinolin-2(1*H*)-one (**230**) in high yield. Hydrolysis and decarboxylation of compound **228** also gives 4-hydroxy-quinolin-2(1*H*)-one **230**. Nitration of compound **230** yields 3-nitroquinolone **224**, which is transformed into 2,3,4-trichloroquinoline **225** under the action of POCl_3 (Scheme 69).¹⁰⁵

Scheme 69



Reagents and conditions: *i.* SOCl_2 , $\text{CH}_2(\text{COOEt})_2$, Mg; *ii.* NaBH_4 , Pd/C, NaOH; *iii.* $p\text{-TsOH}$, H_2O ; *iv.* 3 N HCl, EtOH; *v.* POCl , NEt_3 , 100 °C; H_2 ; *vi.* EtOH, Pd/C; *vii.* HNO_3 , AcOH , 100 °C.

3. Fluoro-containing acetophenones in the synthesis of benzoannulated azaheterocycles

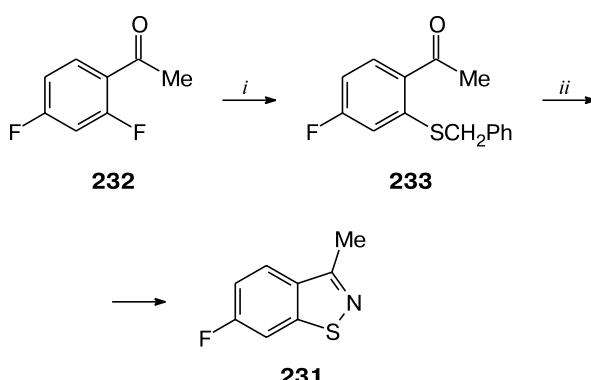
Fluoro-containing acetophenones are convenient synthons for construction of various benzoannulated azaheterocycles. For instance, 6-fluorobenzo[*d*]isothiazole **231** is obtained from 2,4-difluoroacetophenone (**232**).¹⁰⁶ A reaction of compound **232** with benzyl mercaptide gives sulfide **233** through selective replacement of the F atom in position 2. When treated with thionyl chloride, sulfide **233** is transformed into sulphenyl chloride, which reacts with ammonia to give sulfenamide undergoing spontaneous cyclization into benzoisothiazole **231** (Scheme 70).

Reactions of pentafluoroacetophenone (**234**) with aromatic amines in the presence of K_2CO_3 gives, through replacement of the F atom at the C(2) atom by the aniline residue, derivatives **235**–**237**. With highly basic amines ($\text{p}K_a > 4$), the yields of these products are sufficiently high. Heating of 2-arylaminoo-3,4,5,6-tetrafluoroacetophenones **235**–**237** in a mixture of sulfuric and acetic acids produces the corresponding 1,2,3,4-tetrafluoro-9-methylacridines **238**–**241** in high yields (Scheme 71).¹⁰⁷ The formation of 9-methyl-3-phenylamino- or 3-phenylamino-9-trifluoromethylacridine on heating of pentafluoroacetophenone **234** or perfluoroacetophenone with aniline was reported in Ref. 108.

The synthesis of fluoroquinolones and their analogs from ethyl 3-pentafluorobenzoylpyruvate (**242**) is well covered in the literature. Compound **242** is prepared by the Claisen condensation of pentafluoroacetophenone **234** and ethyl oxalate in the presence of LiH (Scheme 72).¹⁰⁹ Reactions of pyruvate **242** with 2-aminophenol and *o*-phenylenediamine in methanol at 20 °C afford 3-pentafluorobenzoylmethylidene-1,2,3,4-tetrahydrobenzoxazin-2-one (**243a**) and 3-pentafluorobenzoylmethylidene-1,2,3,4-tetrahydroquinoxalin-2-one (**243b**), respectively. Intramolecular cyclization of compound **243a** in DMSO at 200 °C without any base or at 80 °C in the presence of triethylamine yields 4,5,6-trifluoro-3*H*-pyrido[3,2,1-*k,l*]-phenoxazin-3-one (**244**) (Scheme 72).¹¹⁰ Product **243b** is transformed into quinolone **245** on heating with triethylamine in DMSO (see Scheme 72).¹¹¹

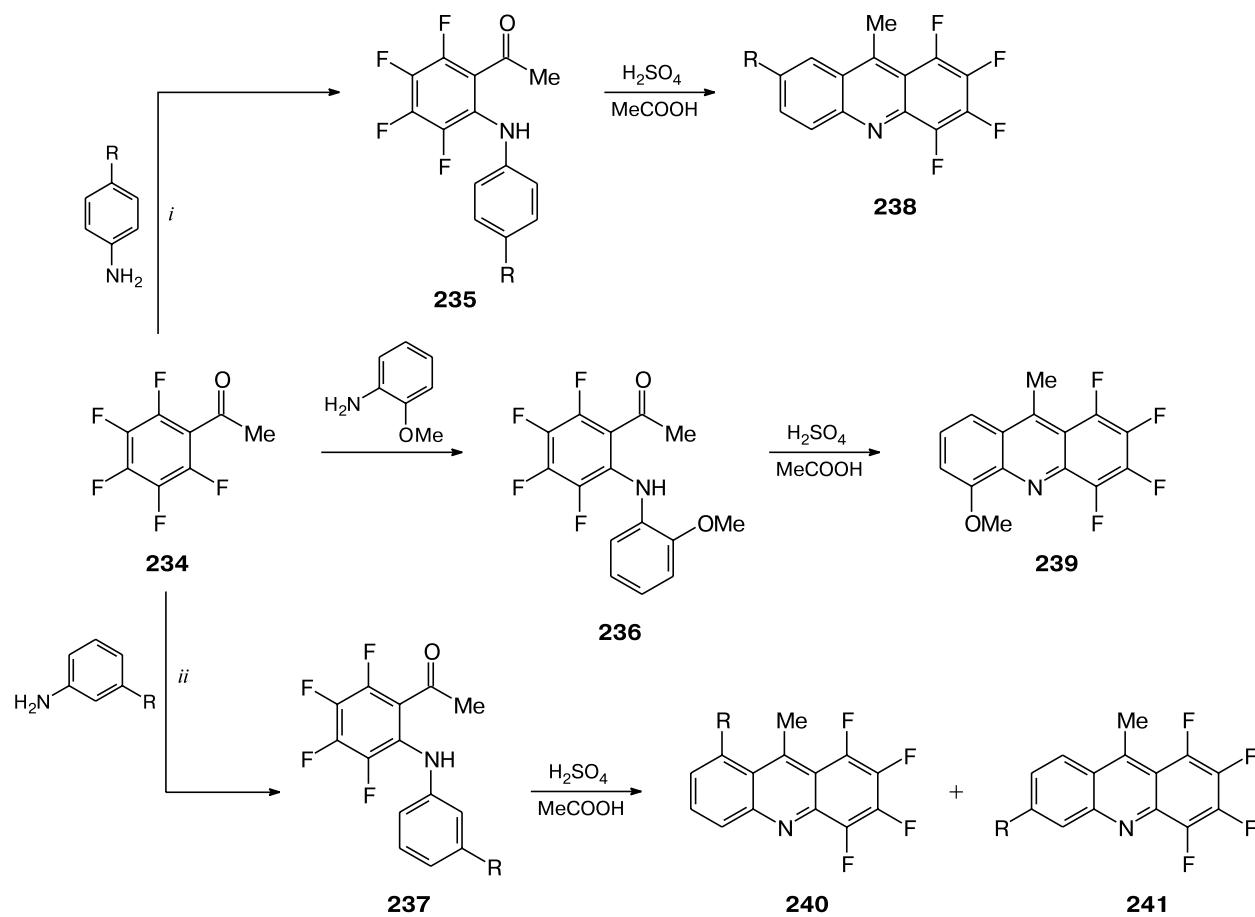
Hydrolysis of ester **242** leads to acid **246**, which reacts with primary amines to give 2-alkylamino(anilino)-3-pentafluorobenzoylacrylic acids **247a,b**. In the

Scheme 70



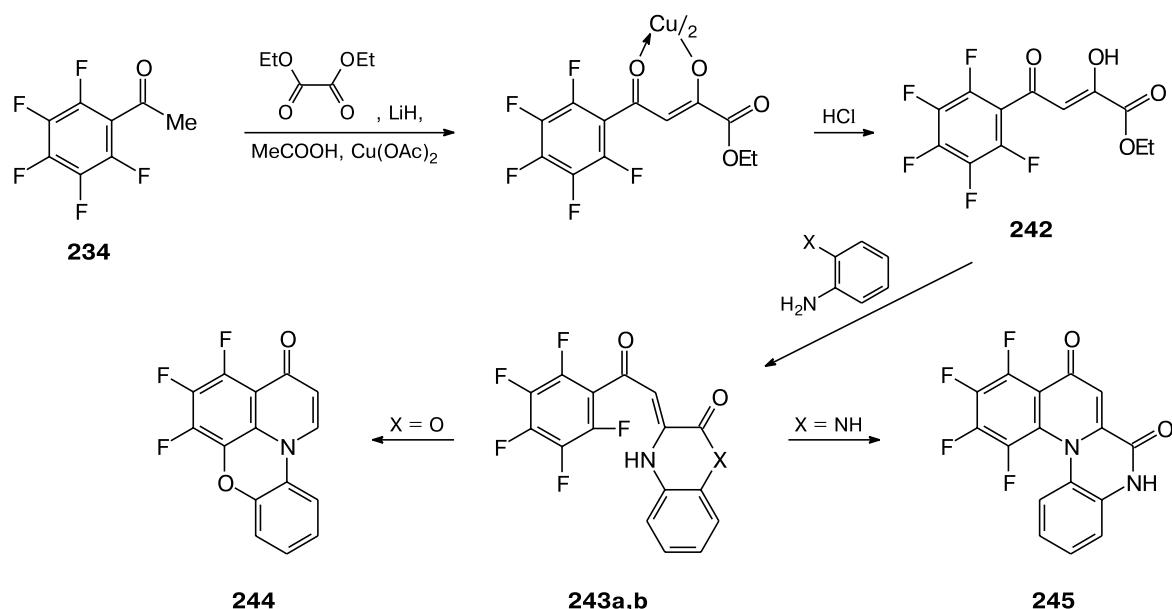
i. PhCH_2SH , KOBu^1 , THF; *ii.* 1) SOCl_2 ; 2) NH_3 , THF.

Scheme 71



i. R = Me, OMe, Br, NMe₂; ii. R = Me, OMe.

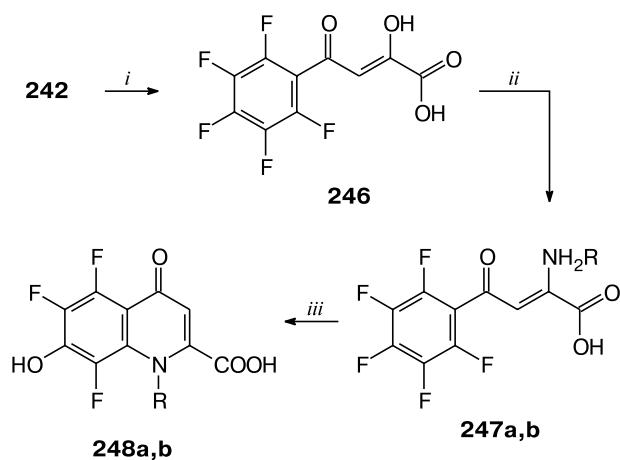
Scheme 72



243: X = O (a), NH (b)

presence of KOH, compounds **247a,b** are transformed into 1-substituted 5,6,8-trifluoro-7-hydroxy-4-oxoquinoline-2-carboxylic acids **248a,b** (Scheme 73).¹⁰⁹

Scheme 73



247, 248: R = C₆H₁₁ (**a**), Prⁱ (**b**)

Reagents and conditions: *i.* HCl, 35–40 °C; *ii.* H₂NR, Δ; *iii.* KOH, 90–95 °C.

A reaction of pyruvate **242** with arenediazonium chloride gives ethyl 3-(arylhydrazino)pentafluorobenzoylpyruvate **249**. When heated in DMSO or refluxed in chloroform with a double excess of triethylamine, compound **249** undergoes cyclization into cinnolones **251** (Scheme 74).¹¹² The latter react with a double excess of

o-phenylenediamine and *o*-aminophenol in boiling alcohols to give complex heterocyclic ensembles of hydrazones (**252**, X = NH, O). In a reaction of ester **249** (Ar = *p*-methoxyphenyl) with diethylenetriamine in ethanol at 20 °C, intramolecular nucleophilic substitution of the F atom is accompanied by an attack on the α-dicarbonyl fragment leading to polycyclic derivative **250** (Scheme 74).¹¹³

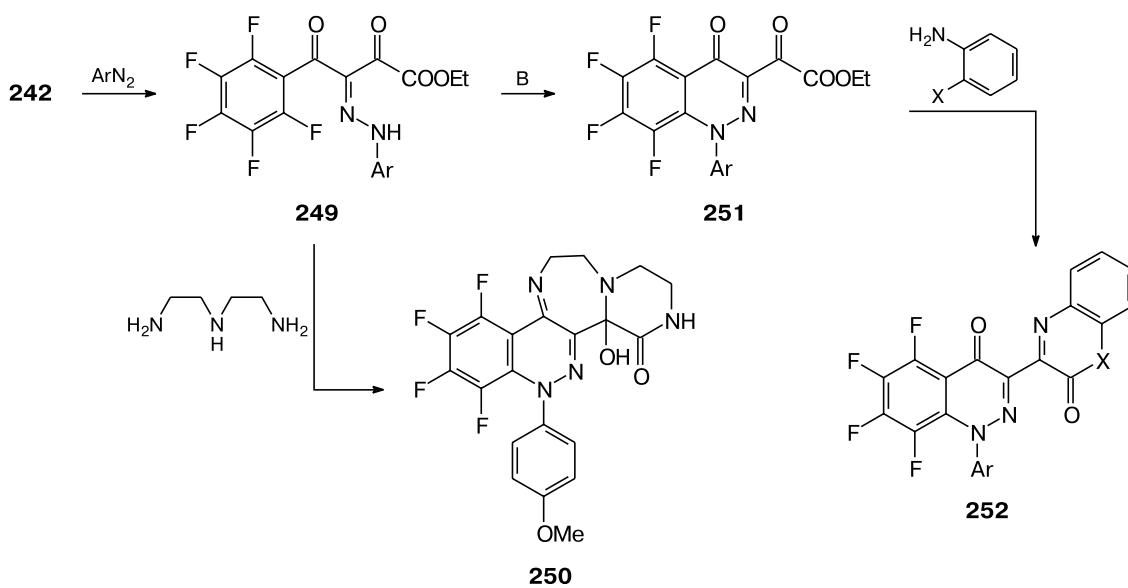
4. Fluoro-containing benzaldehydes and benzophenones in the synthesis of benzoannulated azaheterocycles

A number of fluoro-containing benzazoles and benzazines are obtained from appropriate benzaldehydes and benzophenones. For instance, a microwave-promoted reaction of 2-bromo-5-fluorobenzaldehyde (**253**) with phenylhydrazine gives hydrazone **254**, which undergoes cyclization into 1-phenyl-1*H*-indazole **255** on heating in the presence of CuI and K₂CO₃ (Scheme 75).¹¹⁴

Condensation of fluoro-containing *ortho*-halobenzaldehydes with C,N-dinucleophiles affords various quinolines. For instance, room-temperature reactions of pentafluorobenzaldehyde (**256b**) with substituted anilines give 1,2,3,4-tetrafluoroacridines **257** in quantitative yields and condensation of 2-chloro-3,4,5,6-tetrafluorobenzaldehyde (**256a**) with 5-amino-1,2-azoles yields azolo[*b*]-quinolines **258** (Scheme 76).¹¹⁵

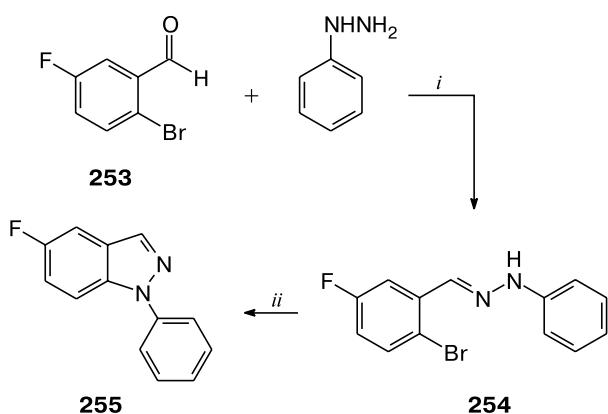
Analogously, pentafluorobenzaldehyde (**256b**) reacts with 6-amino-1,3-dimethyluracil in glacial acetic acid to give 5,6,7,8-tetrafluoro-1,3-dimethylpyrimidino[6,5-*b*]-quinoline-2,4-dione (**259**) (Scheme 77).¹¹⁶

Scheme 74



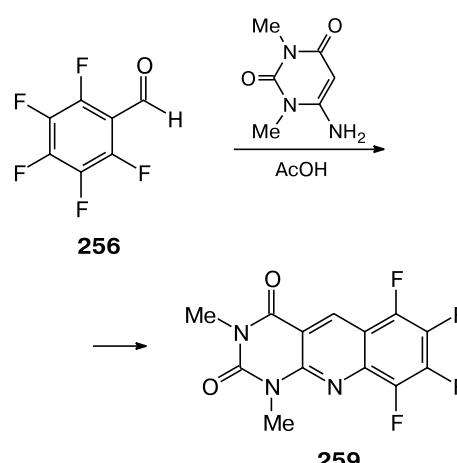
B is a base; Ar = *p*-MeOC₆H₄, X = NH, O

Scheme 75

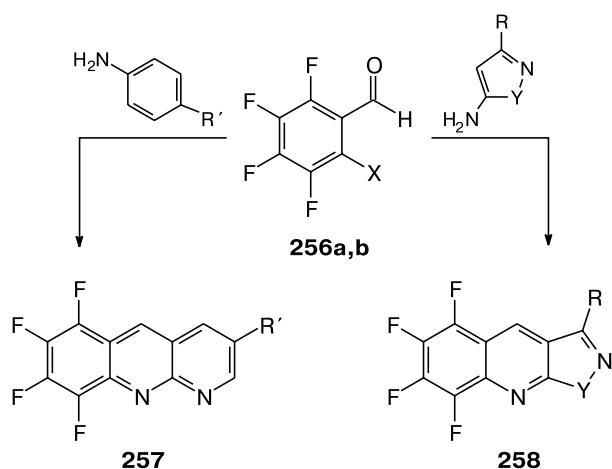


Reagents and conditions: *i.* MW, 160 °C; *ii.* K₂CO₃, CuI, MW, 160 °C.

Scheme 77



Scheme 76



272: X = Cl (**a**), F (**b**);

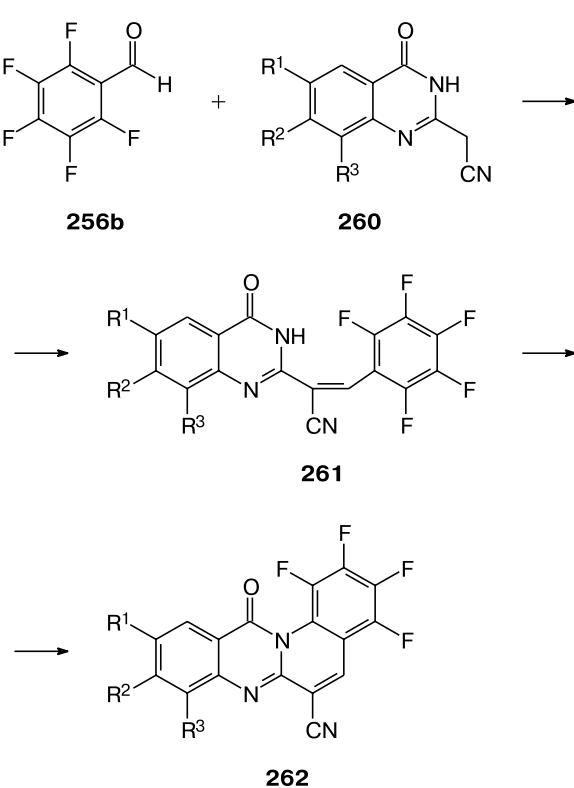
Y = NMe, NPh, O, S; R = Me, CMe, Ph; Rr = OMe, Me, H, F, Br.

Reactions of pentafluorobenzaldehyde (**256b**) with quinazolinylacetonitriles **260** yield compounds **261**, which undergo cyclization into polycyclic systems **262** on heating with NEt₃ in DMF (Scheme 78).¹¹⁷

Base-catalyzed condensation of 2-amino-5-fluorobenzaldehyde (**263**) with 2-acetyl-1-methyl-5-nitroimidazole leads to 2-(1-methyl-5-nitroimidazol-2-yl)quinoline (**264**). This compound reacts with H₂O₂ in acetic acid to give derivative **265** transformed further into compound **266** under the action of POCl₃ (Scheme 79).¹¹⁸ Condensation of aldehyde **263** with 2-acetyl-1-methyl-5-nitrobenzimidazole gives quinoline **267**.¹¹⁹

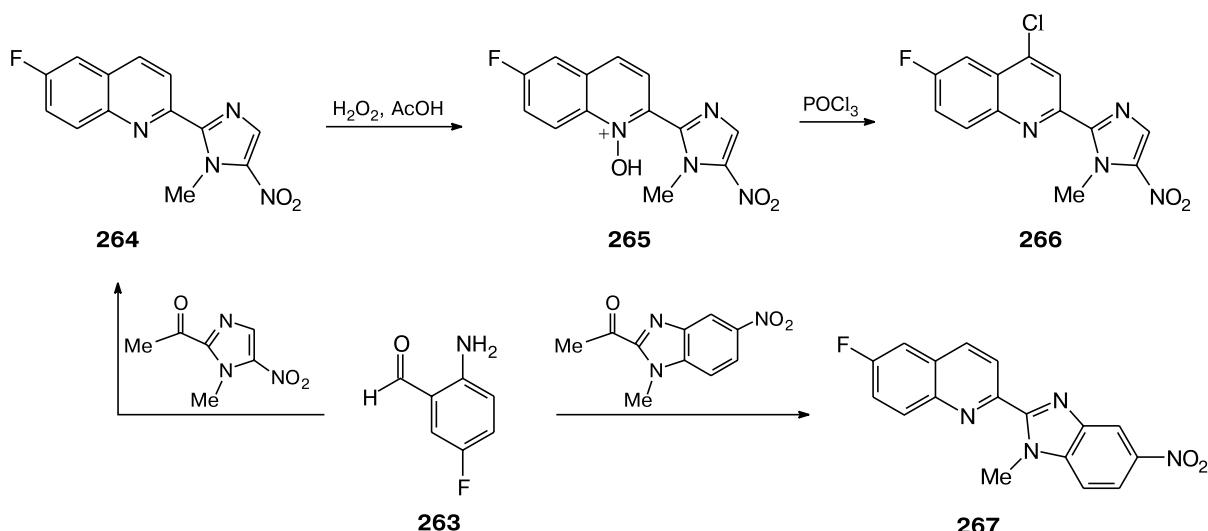
The patented^{120,121} synthesis of 2-chloro-6-fluoroquinolines **268** and benzo[b]naphthyridines **269** from benzaldehydes **270** is shown in Scheme 80. Nitration of com-

Scheme 78

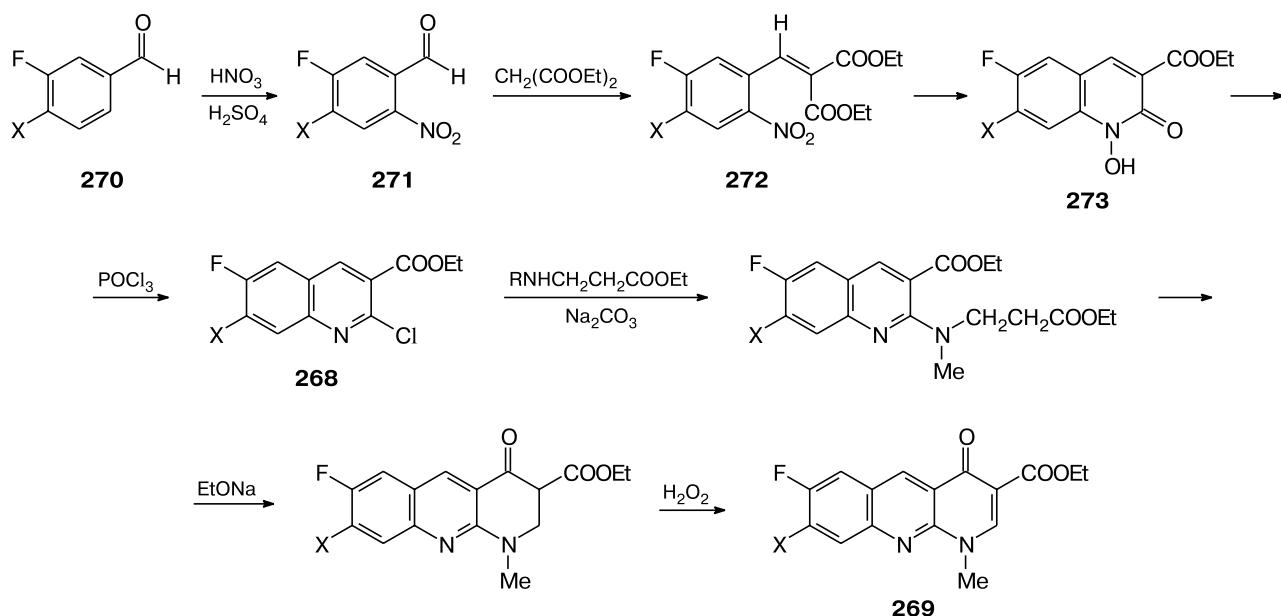


ound **270** with a mixture of nitric and sulfuric (or acetic) acids gives 2-nitrobenzaldehyde **271**. Condensation of this compound with ethyl malonate in the presence of a base (alkoxides, Na₂CO₃, or NaH) leads to ester **272**. Cyclization of malonates **272** followed by treatment of intermediates **273** with POCl₃ yields 2-chloro-6-fluoroquinolines **268**, from which benzo[b]naphthyridines **269** are obtained.

Scheme 79



Scheme 80



R = Me, Et, cyclo-*o*-C₃H₅; **270**: X = F, Cl

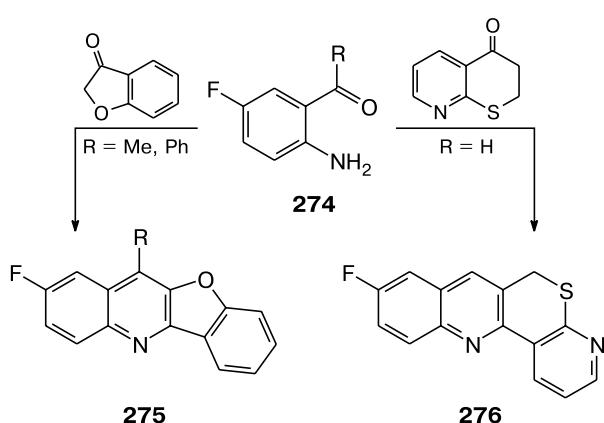
Replacement of the halogen atom X by the heterocyclic residues gives benzo[1,8]naphthyridines **269**, which exhibit antimicrobial activity.^{122–125}

Condensation of 2-amino-5-fluorobenzaldehyde, acetophenone, or benzophenone **274** with 3(2*H*)benzofuranones or 3,4-dihydrothiopyran-4(2*H*)-one leads to benzofuro[3,2-*b*]quinolines **275** and 6*H*-pyrido[3*r*,2*r*:5,6]thiopyran-4,3-*b*]quinolines **276** (Scheme 81).¹²⁶

Fluoro-containing benzophenone **277** reacts with aniline in the presence of K₂CO₃ to give compound **278**, which can undergo cyclization into acridine **279** (Scheme 82).

Hydrolysis of the ether bond in the latter followed by intramolecular replacement of the F atom in DMF in the presence of KF completes the formation of the polycyclic system 6,7,8-trifluoro[1]benzopyran-2,3,4-*k,l*]acridine (**280**) (see Scheme 82).¹²⁷

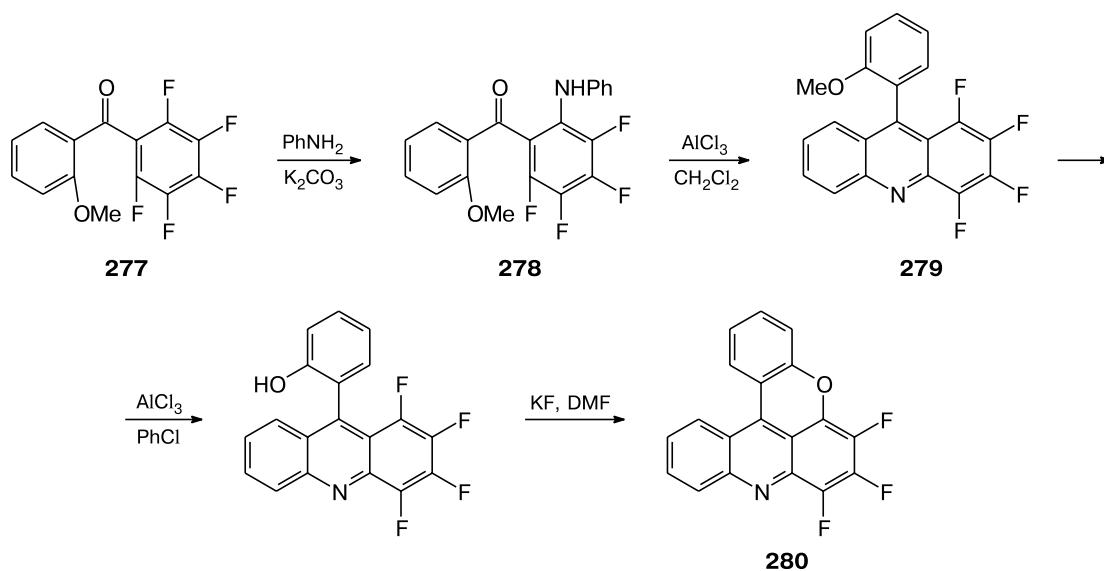
Scheme 81



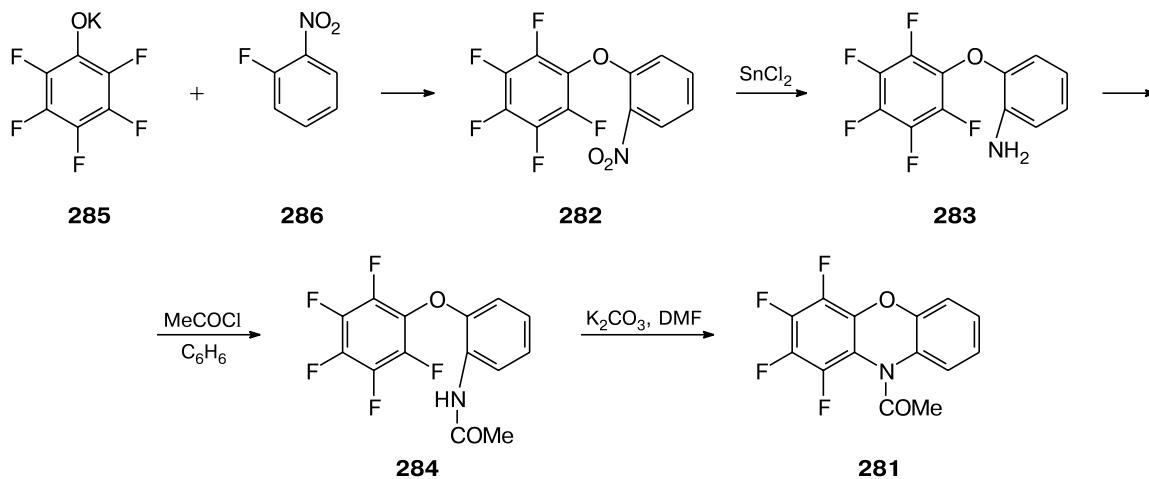
5. Fluoro-containing phenols and benzenethiols in the synthesis of benzoannulated azaheterocycles

Fluoro-containing phenols and benzenethiols find use in the synthesis of phenoxyazines and [1,4]benzothiazines. Polyfluorodiphenyl ethers can serve as the starting materials for the preparation of polyfluorophenoxyazines. For instance, 10-acetyl-1,2,3,4-tetrafluorophenoxyazine (281) is obtained from pentafluoro-2-nitrodiphenyl ether 282. Acid-catalyzed reduction of the latter with SnCl_2 gives 2-amino derivative 283. *N*-Acyl derivative 284 undergoes cyclization into 10-acetyl-1,2,3,4-tetrafluorophenoxyazine (281) on heating with anhydrous K_2CO_3 (Scheme 83). Ether 283 can be prepared either from potassium pentafluorophenolate (285) and 2-fluoronitrobenzene (286) or

Scheme 82



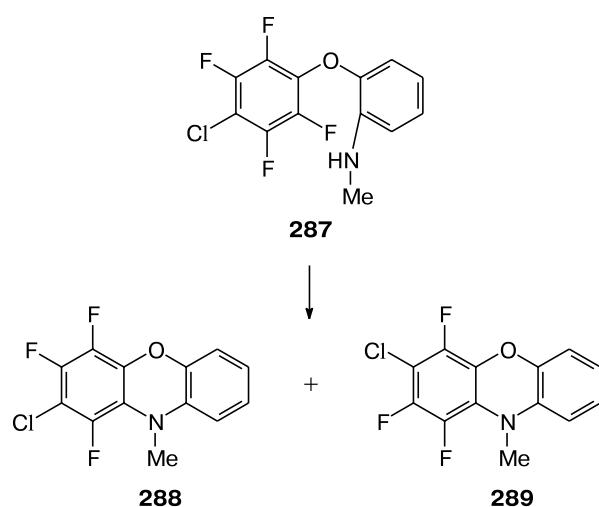
Scheme 83



from polyfluorobenzene and 2-aminophenol in pyridine in the presence of a base.¹²⁹

Another example of the synthesis from polyfluorinated diphenyl ether **287** is provided by the formation of polyfluoro-10-methylphenoxazine **288** on heating of compound **287** in DMF for 3 h; the content of by-product **289** is 5% (Scheme 84).¹³⁰

Scheme 84

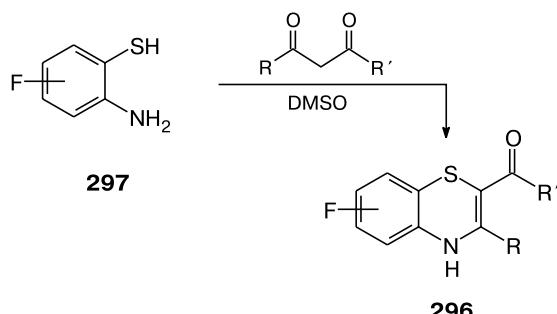


2,3,4,5-Tetrafluorobenzenethiol (**290**) is used to obtain sulfur analogs of fluoroquinolones (**291** and **292**). Benzenethiol **290** is chlorinated and the resulting sulfenyl chloride reacts with ethyl 3-cyclopropylaminoacrylate in pyridine at 60 °C to give enamine **293**. Oxidation of the

latter with chloroperoxybenzoic acid yields sulfoxide **294**; with a threefold excess of the oxidant, the oxidation product is sulfone **295**. Room-temperature cyclization of intermediates **294** and **295** in THF in the presence of NaH affords compounds **291** and **292**, respectively (Scheme 85).¹³¹

The single-step synthesis of fluoro-containing 4*H*-[1,4]benzothiazines **296** is effected by cyclization of 2-aminofluorobenzenethiols **297** with β -diketones in DMSO (Scheme 86).^{132,133}

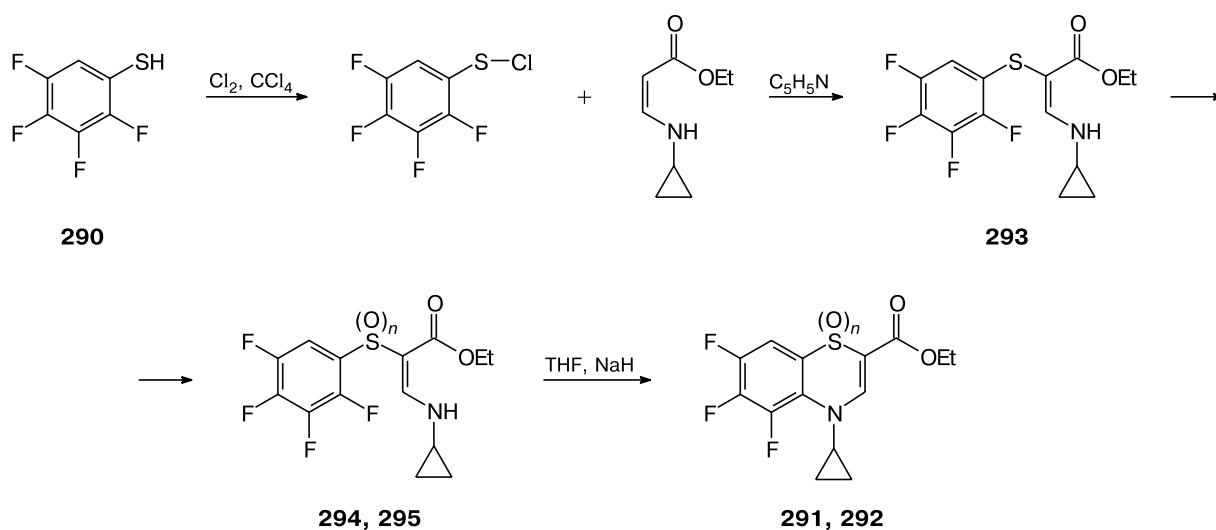
Scheme 86



$R = Me, Rr = Me, Ph, p\text{-FC}_6H_4, p\text{-ClC}_6H_4, EtO, MeO, p\text{-MeOC}_6H_4, p\text{-MeC}_6H_4; R = Rr = Ph$

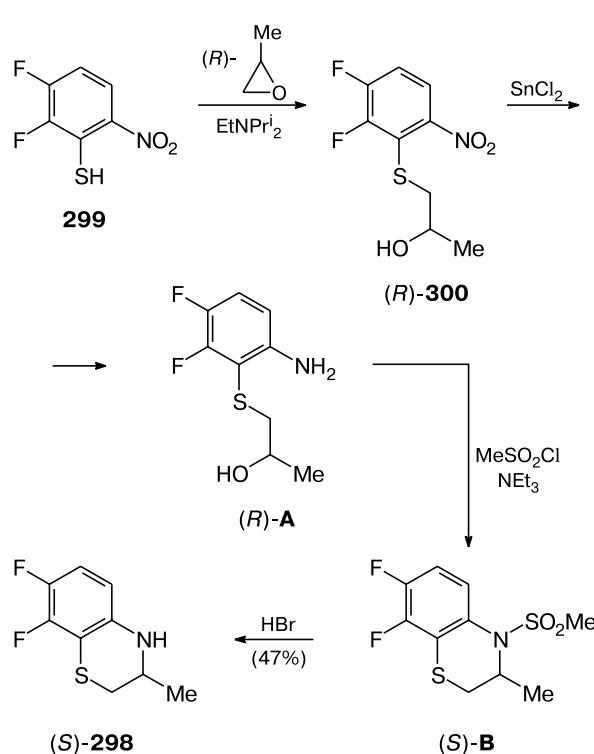
Fluoro-containing 2- and 3-methylbenzothiazines are used to obtain sulfur analogs of levofloxacin. An approach to optically active benzothiazines is shown in Scheme 87. (*S*)-7,8-Difluoro-3-methyl-3,4-dihydro-2*H*-[1,4]benzothiazine (**298**) is synthesized from 3,4-difluoro-2-mercaptopnitrobenzene **299** and (*R*)-propylene oxide via intermediate (*R*)-**300** (see Scheme 87).⁴⁵

Scheme 85



$n = 1$ (**291**, **294**), 2 (**292**, **295**)

Scheme 87



6. Fluoro-containing arenes with other substituents

Apart from the synthons described above, other fluoroarenes containing substituents that can be involved in construction of azaheterocycles are of interest.

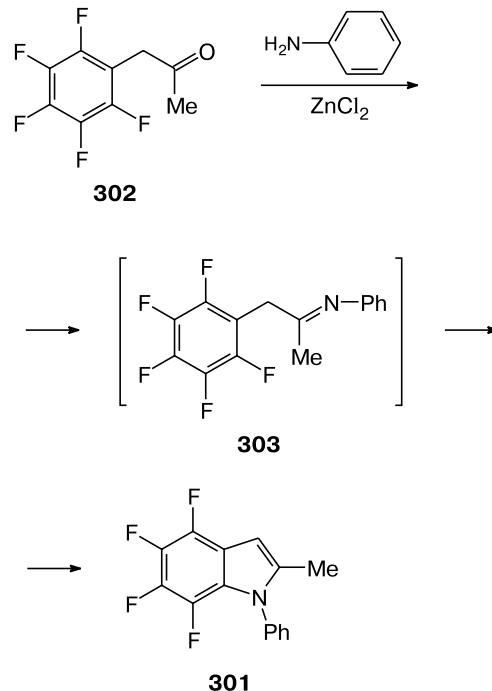
For instance, 4,5,6,7-tetrafluoro-2-methyl-1-phenylindole (**301**) is obtained by reaction of ketone **302** with an excess of aniline in the presence of anhydrous ZnCl₂ followed by intramolecular cyclization of the Schiff base **303** (Scheme 88).¹³⁴

4,5,6,7-Tetrafluoroindole (**304**) is obtained as shown in Scheme 89. The synthesis involves reduction of pentafluorophenylacetonitrile (**305**) to amine **306**, intramolecular cyclization with replacement of the F atom, and smooth oxidation of dihydro derivative **307** with MnO₂ in benzene (Scheme 89).¹³⁵

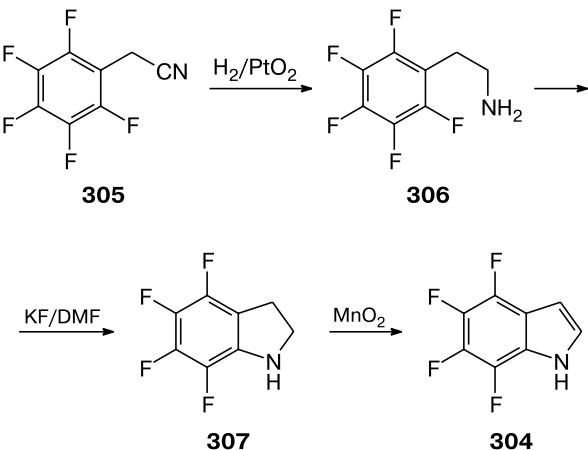
5,6,7,8-Tetrafluoro-1,4-dihydrocinnoline (**308**) is generated by heating of 1-amino-4,5,6,7-tetrafluoro-3-hydroxyindoline (**309**) (ring extension probably involves nitrene). In turn, compound **309** is prepared through intramolecular cyclization of hydrazine **310**. Dihydro derivative **308** is oxidized into 5,6,7,8-tetrafluorocinnoline (**311**) (Scheme 90; see Ref. 136).

Reactions of pentafluorophenylglyoxal (**312**) with alicyclic 2-hydroxyamino oximes **313** in methanol give a mixture of annulated 2-(pentafluorophenyl)pyrazine

Scheme 88



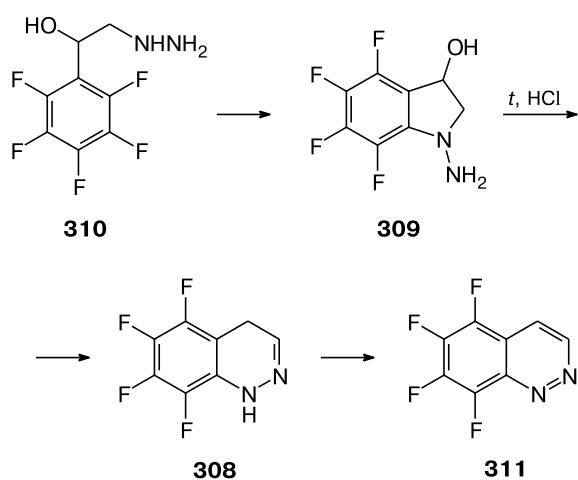
Scheme 89



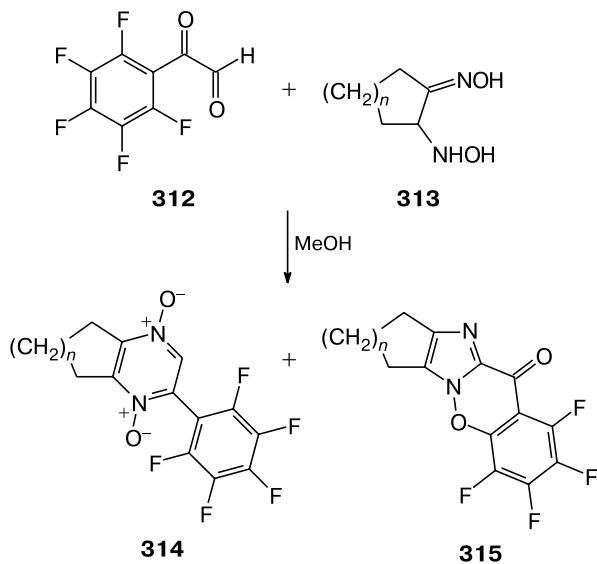
1,4-dioxides **314** and tetrafluoro-10*H*-imidazo[1,2-*b*]-[1,2]benzoxazin-10-ones **315** (Scheme 91).¹³⁷

Pentafluorobenzyl bromide (**316**) reacts with benzimidazole-2-thione in the presence of NaH to give fused tetrafluoro derivative **317** (Scheme 92).¹³⁸ It should be noted that the cyclization pathway in this reaction differs from that in Scheme 43 and no analog of compound **133a** is formed. Apparently, benzimidazole-2-thione is initially alkylated at the S atom, while in the cyclocondensation with polyfluorobenzoyl chloride **107**, benzimidazole-2-thione is acylated at the N atom.

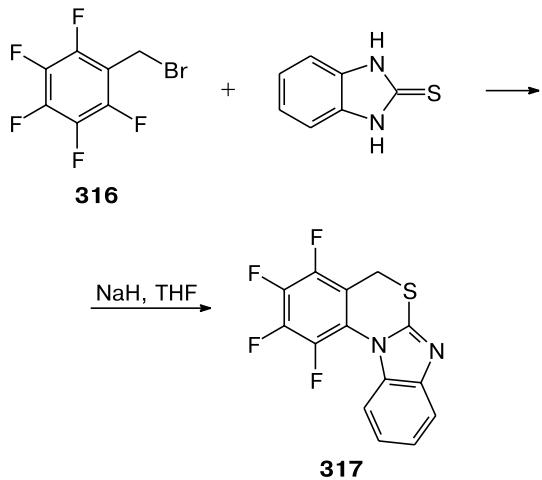
Scheme 90



Scheme 91

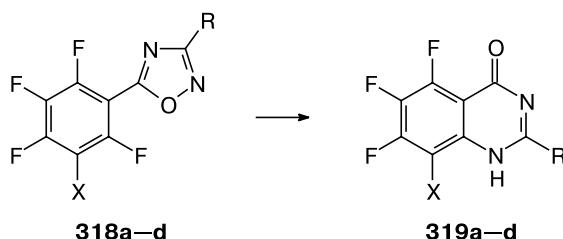


Scheme 92



Fluorinated quinazolin-4-ones can be conveniently obtained using a photochemical method.¹³⁹ Irradiation ($\lambda = 313$ nm) of 5-polyfluoroaryl-3-phenyl- or polyfluoroaryl-3-methyl-1,2,4-oxadiazoles **318** in dry methanol or acetonitrile in the presence of triethylamine gives the corresponding quinazolinones **319** (Scheme 93).

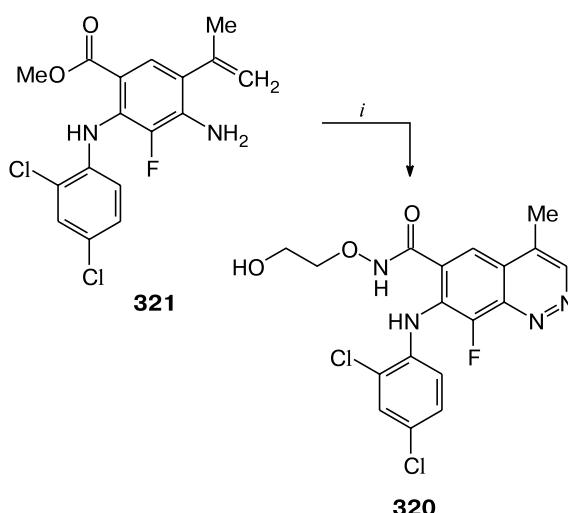
Scheme 93



$R = Ph$, $X = H$ (**a**), F (**b**); $R = Me$, $X = H$ (**c**), F (**d**)

The synthesis of 8-fluorocinnoline **320** involves cyclization of methyl 4-amino-2-(2,4-dichlorophenylamino)-3-fluoro-5-propen-2-ylbenzoate (**321**) under the action of $NaNO_2$ and H_2SO_4 , hydrolysis of the ester bond, and a reaction with *O*-(2-vinyloxyethyl)hydroxylamine (Scheme 94).¹⁴⁰ Hydrolysis of the ether group is carried out in a solution of HCl.

Scheme 94

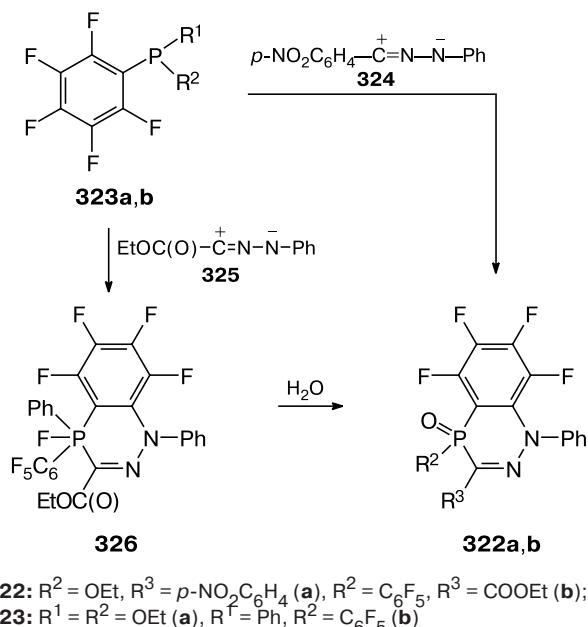


i. 1) $NaNO_2$, H_2SO_4 ; 2) 1 M LiOH; 3) $\text{CH}_2=\text{CH}-\text{O}-\text{CH}_2-\text{NH}_2$; 4) 1 M HCl.

4-Ethoxy-5,6,7,8-tetrafluoro-3-(4-nitrophenyl)-4-oxo-1-phenyl-1,4-dihydrobenzo[e]-1,2,4-diazaphosphorine (**322a**) can be obtained by a reaction of diethyl pentafluorophenylphosphinate (**323a**) with *C*-(4-nitro-

phenyl)-*N*-phenylnitrile imine (**324**) under mild conditions (20 °C) (Scheme 95).¹⁴¹ A reaction of polyfluorophenylphosphine **323b** with *C*-ethoxycarbonyl-*N*-phenylnitrile imine (**325**) under the same conditions gives polyfluorophosphorane **326** further hydrolyzed to 5,6,7,8-tetrafluoro-1,4-dihydrobenzo[*e*]-1,2,4-diazaphosphorine **322b** (see Scheme 95).¹⁴¹

Scheme 95



Conclusion

To sum up, the reviewed methods of constructing fluoro-containing azaheterocycles from fluoroarenes demonstrate that heterocyclization of fluoroarenes opens a route to a wide range of heterosystems. The great synthetic potential of fluoro-containing compounds makes them convenient synthons for the preparation of fluorinated heterocycles which are of interest for medicinal chemistry.

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References

1. *The Fluoro Compounds. The Synthesis and Application*, Ed. N. Ishikawa, Mir, Moscow, 1990.
2. G. G. Furin, *Ftorsoderzhashchie geterotsiklicheskie soedineniya: sintez i primenenie* [Fluorine-Containing Heterocyclic Compounds: Synthesis and Applications], Nauka, Novosibirsk, 2001, 340 pp. (in Russian).
3. V. I. Saloutin, Ya. V. Burgart, O. N. Chupakhin, *Ftorsoderzhashchie trikarbonilnye soedineniya* [Fluorine-Containing Tricarbonyl Compounds], UrO RAN, Ekaterinburg, 2002, 142 pp. (in Russian).
4. E. V. Nosova, N. N. Mochulskaya, S. K. Kotovskaya, G. N. Lipunova, V. N. Charushin, *Heteroatom Chem.*, 2006, **17**(6), 579.
5. G. M. Brooke, *J. Fluorine Chem.*, 1997, **86**, 1.
6. A. Pace, S. Buscemi, N. Vivona, *Org. Prep. Proc. Int.*, 2005, **37**, 447.
7. A. Pace, S. Buscemi, N. Vivona, *Org. Prep. Proc. Int.*, 2007, **39**, 1.
8. S. D. Taylor, C. C. Kotoris, G. Hum, *Tetrahedron*, 1999, **55**, 12431.
9. V. G. Granik, *Osnovy meditsinskoi khimii* [Fundamentals of Medicinal Chemistry], Vuzovskaya Kniga, Moscow, 2001, 230 pp. (in Russian).
10. V. S. Mokrushin, G. A. Vavilov, *Osnovy khimii i tekhnologii bioorganicheskikh i sinteticheskikh lekarstvennykh veshchestv* [Fundamentals of the Chemistry and Technology of Bioorganic and Synthetic Drugs], GOU VPO UGTU-UPI, Ekaterinburg, 2004, 481 pp. (in Russian).
11. L. L. Shen, *Quinolone Antibacterial Agents*, American Society for Microbiology, Washington, 1993.
12. *The Quinolones*, Ed. T. V. Andriole, Academic Press, New York, 1988.
13. D. Bouzard, *Recent Progress in the Chemical Synthesis of Antibiotics*, Springer-Verlag, Berlin, 1990, 249 pp.
14. *Quinolone Antibacterial Agents*, Eds J. S. Wolfson, D. C. Hooper, American Society for Microbiology, Washington, 1989.
15. *Quinolone Antibacterial Agents*, Eds D. C. Hooper, J. S. Wolfson, 2nd ed., American Society for Microbiology, Washington, 1993.
16. E. N. Padeiskaya, V. P. Yakovlev, *Ftorkhinolony* [Fluoroquinolones], Bioinform, Moscow, 1995, 395 pp. (in Russian).
17. N. I. Fadeeva, M. V. Shulgina, G. G. Glushkov, *Khim.-Farm. Zh.*, 1993, **27**, 4 [*Pharm. Chem. J.*, 1993, **27** (Engl. Transl.)].
18. E. N. Padeiskaya, *Antibiotiki i khimioterapiya* [Antibiotics and Chemotherapy], 1998, **43**, 38 (in Russian).
19. PCT Int Appl WO 19 456 (1996), *Chem. Abstr.*, 1996, **125**, 114513z.
20. R. D. Chambers, M. Parsons, G. Sandford, C. J. Skinner, M. J. Atherton, J. S. Moilliet, *J. Chem. Soc., Perkin Trans. I*, 1999, 803.
21. R. D. Chambers, D. Holling, G. Sandford, A. S. Batsanov, S. A. Howard, *J. Fluorine Chem.*, 2004, **125**, 661.
22. R. D. Chambers, D. Holling, G. Sandford, H. Puschmann, J. A. K. Howard, *J. Fluorine Chem.*, 2002, **117**, 99.
23. JP Patent 06 135 856 (1994), *Chem. Abstr.*, 1995, **122**, 881142k.

24. PCT Int Appl WO 7460 (2004), *Chem. Abstr.*, 2004, **140**, 128288b.
25. P. M. Neill, R. C. Storr, B. K. Park, *Tetrahedron*, 1998, **54**, 4615.
26. S. K. Kotovskaya, G. A. Zhumabaeva, N. M. Perova, Z. M. Baskakova, E. F. Belanov, N. I. Bormotov, S. M. Balakhnin, O. A. Serova, V. N. Charushin, O. N. Chupakhin, *Khim.-Farm. Zh.*, 2007, **41**, No. 12, 5 [*Pharm. Chem. J.*, 2007, **41** (Engl. Transl.)].
27. J. S. Sloop, C. L. Beemgardner, W. D. Loehle, *J. Fluorine Chem.*, 2002, **118**, 135.
28. W. Du, D. P. Curran, *Org. Lett.*, 2003, **5**, 1765.
29. JP Patent 55 376 (2003); *Chem. Abstr.*, 2003, **138**, 205051g.
30. T. O. Richardson, V. P. Shanbhag, S. Adairk, S. J. Smith, *J. Heterocycl. Chem.*, 1998, **35**, 1301.
31. S. L. Vingkar, A. S. Bohade, B. G. Khadse, *Indian Drugs*, 2001, **38**, 347.
32. S. L. Vingkar, A. S. Bohade, B. G. Khadse, *Indian Drugs*, 2001, **38**, 573.
33. S. R. Pattan, M. S. Ali, Y. S. Pattan, V. V. K. Reddy, *Indian J. Heterocycl. Chem.*, 2004, **14**, 157.
34. O. M. Nefedov, A. I. Ioffe, L. G. Menchikov, *Khimiya karbenov [The Chemistry of Carbenes]*, Khimiya, Moscow, 1990, 256 pp. (in Russian).
35. G. Jones, in *Comprehensive Heterocyclic Chemistry II*, Eds. A. R. Katritzky, C. W. Rees, E. V. Scriven, vol. 5, Pergamon, Oxford, 1996, 167–243.
36. G. A. Mokrushina, S. G. Alekseev, V. N. Charushin, O. N. Chupakhin, *Zh. Vses. Khim. O-va im. D.I. Mendeleva*, 1991, **36**, 447 [*Mendelev. Chem. J.*, 1991, **36** (Engl. Transl.)].
37. R. Krishnan, S. A. Lang, M. M. Siegel, *J. Heterocycl. Chem.*, 1986, **23**, 1801.
38. A. Ya. Aizikovich, V. N. Charushin, O. N. Chupakhin, *Khim.-Farm. Zh.*, 1996, **30**, No. 8, 43 [*Pharm. Chem. J.*, 1996, **30** (Engl. Transl.)].
39. V. I. Vysokov, V. N. Charushin, G. A. Afanasyeva, O. N. Chupakhin, *Mendelev. Commun.*, 1993, 159.
40. V. I. Vysokov, V. N. Charushin, O. N. Chupakhin, T. K. Pashkevich, *Zh. Org. Khim.*, 1998, **34**, 455 [*Russ. J. Org. Chem.*, 1998, **34** (Engl. Transl.)].
41. G. M. Brooke, R. J. D. Rutherford, *J. Chem. Soc. (C)*, 1967, 1189.
42. I. V. Kolesnikov, A. G. Ryabichev, T. D. Petrova, V. E. Platonov, *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1988, 1651 [*Bull. Acad. Sci. USSR, Div. Chem. Sci.*, 1988, **37** (Engl. Transl.)].
43. A. G. Hudson, A. E. Pedler, J. C. Tatlow, *Tetrahedron Lett.*, 1968, **17**, 2143.
44. A. J. Elliott, V. S. Gibson, *Can. J. Chem.*, 1975, **53**, 1484.
45. EP Patent 0 368 419 (1990); *Chem. Abstr.*, 1990, **112**, 95565d.
46. O. N. Chupakhin, S. K. Kotovskaya, N. M. Perova, Z. M. Baskakova, V. N. Charushin, *Khim. Geterotsikl. Soedin.*, 1999, 520 [*Chem. Heterocycl. Compd.*, 1999, **35** (Engl. Transl.)].
47. S. K. Kotovskaya, V. N. Charushin, O. N. Chupakhin, E. O. Kozhevnikova, *Zh. Org. Khim.*, 1998, **34**, 399 [*Russ. J. Org. Chem.*, 1998, **34** (Engl. Transl.)].
48. S. K. Kotovskaya, N. M. Perova, V. N. Charushin, O. N. Chupakhin, *Mendelev Commun.*, 1999, 76.
49. S. K. Kotovskaya, S. A. Romanova, V. N. Charushin, O. N. Chupakhin, Abstracts of Papers, XVII Mendeleevskii srezd po obshchei i prikladnoi khimii [XVII Mendeleev Conf. on General and Applied Chemistry], Kazan, 2003, 454 (in Russian).
50. S. K. Kotovskaya, S. A. Romanova, V. N. Charushin, M. I. Kodess, *Zh. Org. Khim.*, 2004, **40**, 1214 [*Russ. J. Org. Chem.*, 2004, **40** (Engl. Transl.)].
51. R. J. Abdel-Jalil, H. M. Aldocum, M. T. Ayoub, *Heterocycles*, 2005, **65**, 2061.
52. V. N. Charushin, G. A. Mokrushina, A. M. Shevelin, O. M. Chasovskikh, A. A. Shcherbakov, G. G. Alexandrov, O. N. Chupakhin, *Zh. Org. Khim.*, 1998, **34**, 123 [*Russ. J. Org. Chem.*, 1998, **34** (Engl. Transl.)].
53. V. N. Charushin, S. A. Romanova, S. K. Kotovskaya, O. N. Chupakhin, Proc. Symposium “Advances in Synthetic, Combinatorial and Medicinal Chemistry (ASCMC)”, Moscow, 2004, 41.
54. S. K. Kotovskaya, Z. M. Baskakova, N. M. Perova, S. A. Romanova, V. N. Charushin, O. N. Chupakhin, *Azotistye geterotsikly i alkaloidy [Nitrogen Heterocycles and Alkaloids]*, 2001, **2**, IRIDIUM Press, Moscow, 162 (in Russian).
55. S. K. Kotovskaya, N. M. Perova, Z. M. Baskakova, S. A. Romanova, O. N. Chupakhin, V. N. Charushin, *Zh. Org. Khim.*, 2001, **37**, 598 [*Russ. J. Org. Chem.*, 2001, **71** (Engl. Transl.)].
56. V. N. Charushin, G. A. Mokrushina, G. M. Petrova, G. G. Alexandrov, O. N. Chupakhin, *Mendelev Commun.*, 1998, 133.
57. G. A. Mokrushina, G. M. Petrova, O. M. Chasovskikh, V. N. Charushin, O. N. Chupakhin, *Proc. Int. Memorial I. Postovsky Conf.*, 1998, Ekaterinburg, 97.
58. M. Patel, R. J. McHugh, B. C. Cordova, R. M. Klabe, S. Erichson-Viilanen, G. L. Trainor, J. D. Rodgers, *Bioorg. Med. Chem. Lett.*, 2000, **10**, 1729.
59. J. Dudash, Y. Zhang, J. B. Moore, R. Look, Y. Liang, M. P. Beavers, B. R. Conway, P. J. Rybczynski, K. T. Demarest, *Bioorg. Med. Chem. Lett.*, 2005, **15**, 4790.
60. X. Li, D. Wang, J. Wu, W. Xu, *Synth. Commun.*, 2005, **35**, 2553.
61. G. N. Lipunova, E. V. Nosova, P. V. Vasil'eva, V. N. Charushin, *Izv. Akad. Nauk, Ser. Khim.*, 2003, 436 [*Russ. Chem. Bull., Int. Ed.*, 2003, **52**, 457].
62. Yu. M. Volovenko, A. G. Nemazanyi, I. G. Ryabokon', F. S. Babichev, *Ukr. Khim. Zh.*, 1988, **54**, 295 (in Russian).
63. M. Vales, V. Lokshin, G. Pepe, A. Samat, R. Guglielmetti, *Synthesis*, 2001, **16**, 2419.
64. A. A. Layeva, E. V. Nosova, G. N. Lipunova, T. V. Trashakhova, V. N. Charushin, *J. Fluorine Chem.*, 2007, **128**, 748.
65. E. V. Nosova, G. N. Lipunova, M. I. Kodess, P. V. Vasil'eva, V. N. Charushin, *Izv. Akad. Nauk, Ser. Khim.*, 2004, 2216 [*Russ. Chem. Bull., Int. Ed.*, 2004, **53**, 2314].
66. M. J. Deetz, J. P. Malerich, A. M. Beatty, B. D. Smith, *Tetrahedron Lett.*, 2001, **42**, 1851.
67. G. N. Lipunova, E. V. Nosova, A. A. Layeva, M. I. Kodess, V. N. Charushin, *Zh. Org. Khim.*, 2005, **41**, 1092 [*Russ. J. Org. Chem.*, 2005, **41** (Engl. Transl.)].
68. E. V. Nosova, G. N. Lipunova, A. A. Layeva, N. N. Mochul'skaya, V. N. Charushin, in *Karbonilnye soedineniya v sinteze geterotsiklov [Carbonyl Compounds in Heterocyclic Synthesis]*, Saratov, 2004, 209 (in Russian).
69. G. N. Lipunova, E. V. Nosova, G. A. Mokrushina, E. G. Ogloblina, G. G. Alexandrov, V. N. Charushin, *Zh. Org. Khim.*, 2003, **39**, 270 [*Russ. J. Org. Chem.*, 2003, **39** (Engl. Transl.)].
70. T. V. Fomenko, T. N. Gerasimova, E. P. Fokin, *Izv. Sib. Otd. Akad. Nauk SSSR, Ser. Khim. Nauk*, 1977, **1**, No. 2, 99 (in Russian).

71. E. V. Nosova, G. N. Lipunova, A. A. Layeva, L. P. Sidorova, V. N. Charushin, *Zh. Org. Khim.*, 2006, **42**, 1555 [*Russ. J. Org. Chem.*, 2006, **42** (Engl. Transl.)].
72. T. P. Tran, E. L. Ellsworth, B. M. Watson, J. P. Sanchez, S. H. D. Hollis, J. R. Rubin, M. A. Stier, Y. Yip, D. Q. Nguyen, P. Bird, R. Singh, *J. Heterocycl. Chem.*, 2005, **42**, 669.
73. A. Pathak, C. H. Panos, C. C. Okeke, *J. Heterocycl. Chem.*, 1988, **25**, 1173.
74. J. B. Hynes, A. Tomazie, C. A. Parrish, O. S. Fetzer, *J. Heterocycl. Chem.*, 1991, **28**, 1357.
75. G. A. Roth, J. J. Tai, *J. Heterocycl. Chem.*, 1996, **33**, 2051.
76. PCT Int. Appl. WO 46 590 (2005); *Chem. Abstr.*, 2005, **143**, 7732b.
77. PCT Int. Appl. WO 47 244 (2005); *Chem. Abstr.*, 2005, **143**, 7710k.
78. US Patent 261 307 (2005); *Chem. Abstr.*, 2005, **143**, 4779691.
79. A. A. Layeva, E. V. Nosova, G. N. Lipunova, T. V. Trashakhova, V. N. Charushin, *Izv. Akad. Nauk, Ser. Khim.*, 2007, 1758 [*Russ. Chem. Bull., Int. Ed.*, 2007, **56**, 1821].
80. G. A. Mokrushina, E. V. Nosova, G. N. Lipunova, V. N. Charushin, *Zh. Org. Khim.*, 1999, **35**, 1447 [*Russ. J. Org. Chem.*, 1999, **35** (Engl. Transl.)].
81. A. T. Prudchenko, G. S. Shchegoleva, V. A. Barkhash, N. N. Vorozhtsov, *Zh. Obshch. Khim.*, 1967, **37**, 2487 [*J. Gen. Chem. USSR*, 1967, **37** (Engl. Transl.)].
82. W. J. Coates, in *Comprehensive Heterocyclic Chemistry II*, Eds A. R. Katritzky, C. W. Rees, E. V. Scriven, Pergamon, Oxford, 1996, vol. **6**, 1.
83. G. S. Shchegoleva, A. K. Petrov, V. A. Barkhash, N. N. Vorozhtsov, *Khim. Geterotsikl. Soedin.*, 1970, 278 [*Chem. Heterocycl. Compd. USSR*, 1970, **16** (Engl. Transl.)].
84. G. N. Lipunova, E. V. Nosova, V. N. Charushin, L. P. Sidorova, O. M. Chasovskikh, *Mendeleev Commun.*, 1998, 131.
85. G. N. Lipunova, L. P. Sidorova, E. V. Nosova, N. M. Perova, V. N. Charushin, G. G. Alexandrov, *Zh. Org. Khim.*, 1999, **35**, 1729 [*Russ. J. Org. Chem.*, 1999, **35** (Engl. Transl.)].
86. G. N. Lipunova, E. V. Nosova, N. N. Mochulskaya, A. A. Andreiko, O. M. Chasovskikh, V. N. Charushin, *Izv. Akad. Nauk, Ser. Khim.*, 2002, 613 [*Russ. Chem. Bull., Int. Ed.*, 2002, **51**, 663].
87. G. N. Lipunova, E. V. Nosova, V. N. Charushin, O. M. Chasovskikh, *Khim. Geterotsikl. Soedin.*, 2001, 1396 [*Chem. Heterocycl. Compd.*, 2001, **37** (Engl. Transl.)].
88. E. V. Nosova, L. P. Sidorova, G. N. Lipunova, N. N. Mochulskaya, O. M. Chasovskikh, V. N. Charushin, *Khim. Geterotsikl. Soedin.*, 2002, 1060 [*Chem. Heterocycl. Compd.*, 2002, **38** (Engl. Transl.)].
89. G. N. Lipunova, G. A. Mokrushina, E. V. Granovskaya, O. M. Chasovskikh, V. N. Charushin, *Mendeleev Commun.*, 1996, 15.
90. G. N. Lipunova, G. A. Mokrushina, E. V. Nosova, L. I. Rusinova, V. N. Charushin, *Mendeleev Commun.*, 1997, 109.
91. E. V. Nosova, G. N. Lipunova, G. A. Mokrushina, O. M. Chasovskikh, L. I. Rusinova, V. N. Charushin, G. G. Alexandrov, *Zh. Org. Khim.*, 1998, **34**, 436 [*Russ. J. Org. Chem.*, 1998, **34** (Engl. Transl.)].
92. G. N. Lipunova, E. V. Nosova, M. I. Kodess, V. N. Charushin, Yu. A. Rozin, O. M. Chasovskikh, *Zh. Org. Khim.*, 2001, **37**, 604 [*Russ. J. Org. Chem.*, 2001, **37** (Engl. Transl.)].
93. H. Miao, V. Ceccetti, O. Tabarrini, A. Fravolini, *J. Heterocycl. Chem.*, 2000, **37**, 297.
94. G. Q. Hu, Z. Q. Zhang, W. L. Huang, H. B. Zhang, S. T. Huang, *Chinese Chem. Lett.*, 2004, **15** (1), 23.
95. M. Matsuoka, J. Segawa, Y. Makita, S. Ohmachi, T. Kashima, K. Nakamura, M. Hattori, M. Kitano, M. Kise, *J. Heterocycl. Chem.*, 1997, **34**, 1773.
96. T. Undheim, T. Benneche, in *Comprehensive Heterocyclic Chemistry II*, Eds A. R. Katritzky, C. W. Rees, E. V. Scriven, Pergamon, Oxford, 1996, vol. **6**, 93–231.
97. PCT Int. Appl. WO 99 818 (2003); *Chem. Abstr.*, 2004, **140**, 16739f.
98. L. L. Zaika, M. M. Joullie, *J. Heterocycl. Chem.*, 1966, **3**, 289.
99. C. Parkanyi, H. L. Yuan, B. H. E. Stromberg, A. Evenzahav, *J. Heterocycl. Chem.*, 1992, **29**, 749.
100. Y. Zhou, D. E. Murphy, Z. Sun, V. E. Gregor, *Tetrahedron Lett.*, 2004, **45**, 8049.
101. A. V. Ivachenko, S. M. Kovalenko, O. G. Drush-Lyak, *J. Comb. Chem.*, 2003, **5**, 775.
102. Zayavka RF [RF Application] 1 121 983/04 (2001); *Byull. Izobret.*, 2003, **32**, 159 (in Russian).
103. N. A. Orlova, L. L. Dmitrieva, T. N. Gerasimova, E. P. Fokin, *Izv. Sib. Otd. Akad. Nauk SSSR, Ser. Khim. Nauk*, 1976, **3** (7), 109 (in Russian).
104. Y. F. Ji, X. D. Pan, X. Y. Wei, *Synlett.*, 2004, 1607.
105. J. C. Jung, S. Oh, W. K. Kim, W. K. Park, J. Y. Kong, O. S. Park, *J. Heterocycl. Chem.*, 2003, **40**, 617.
106. D. M. Fink, P. M. Strupczewski, *Tetrahedron Lett.*, 1993, **34**, 6525.
107. T. N. Gerasimova, L. L. Golumbovskaya, I. I. Baturina, E. P. Fokin, *Izv. Sib. Otd. Akad. Nauk SSSR, Ser. Khim. Nauk*, 1973, **2** (4), 88 (in Russian).
108. T. N. Vasilevskaya, I. I. Baturina, M. I. Kollegova, T. N. Gerasimova, V. A. Barkhash, *Zh. Org. Khim.*, 1971, **7**, 1230 [*J. Org. Chem. USSR*, 1971, **7** (Engl. Transl.)].
109. V. I. Saloutin, Y. V. Burgart, C. O. Kappe, O. N. Chupakhin, *Heterocycles*, 2000, **52**, 1411.
110. S. G. Perevalov, Z. E. Skryabina, V. I. Saloutin, *Zh. Org. Khim.*, 1997, **33**, 1418 [*J. Org. Chem. USSR*, 1997, **33** (Engl. Transl.)].
111. V. I. Saloutin, I. T. Bazilr, Z. E. Skryabina, S. N. Shurov, S. G. Perevalov, *Zh. Org. Khim.*, 1995, **31**, 718 [*J. Org. Chem. USSR*, 1995, **31** (Engl. Transl.)].
112. Y. V. Burgart, A. S. Fokin, O. G. Kuzueva, O. N. Chupakhin, V. I. Saloutin, *J. Fluorine Chem.*, 1998, **92**, 101.
113. A. S. Fokin, Ya. V. Burgart, V. I. Saloutin, *V Molodezhnaya nauchnaya shkola-konferentsiya po organicheskoi khimii [V Youth Scientific School-Conf. on Organic Chemistry]*, Ekaterinburg, 2002, 461 (in Russian).
114. C. Pabba, H.-J. Wang, S. R. Mulligan, Z.-J. Chen, T. M. Stark, *Tetrahedron Lett.*, 2005, **46**, 7553.
115. M. A. Abramov, E. Ceulemans, C. Jackers, M. van der Auweraer, W. Dehaen, *Tetrahedron Lett.*, 2001, **57**, 9123.
116. J. B. Jiang, J. Roberts, *J. Heterocycl. Chem.*, 1985, **22**, 159.
117. O. V. Khilya, T. A. Volovenko, A. V. Turov, Yu. M. Volovenko, *Khim. Geterotsikl. Soedin.*, 2004, **8**, 1226 [*Chem. Heterocycl. Compd.*, 2004 (Engl. Transl.)].
118. A. Shafiee, K. Parang, M. Khazan, F. Ghasemian, *J. Heterocycl. Chem.*, 1992, **29**, 1859.
119. F. Alimohammadi, S. Abedifard, A. Shafiee, *J. Heterocycl. Chem.*, 1994, **31**, 1037.

120. FR Patent 2 703 681 (1992); *Chem. Abstr.*, 1995, **122**, 56035v.
121. US Patent 5 442 070 (1995); *Chem. Abstr.*, 1993, **119**, 117234w.
122. US Patent 4 970 213 (1990); *Chem. Abstr.*, 1991, **114**, 62083k.
123. US Patent 5 004 745 (1991); *Chem. Abstr.*, 1991, **114**, 62083k.
124. EP Patent 431 991 (1991), *Chem. Abstr.*, 1992, **116**, 6538s.
125. PCT Int. Appl. WO 07 121 (1993); *Chem. Abstr.*, 1993, **119**, 280414g.
126. S. Yamaguchi, K. Tsuzuki, Y. Sannomiya, Y. Ohhira, Y. Kawase, *J. Heterocycl. Chem.*, 1989, **26**, 285.
127. T. N. Gerasimova, L. L. Dmitrieva, E. P. Fokin, *Izv. Sib. Otd. Akad. Nauk SSSR, Ser. Khim. Nauk*, 1976, **3** (7), 106 (in Russian).
128. T. V. Mikhлина, Е. F. Kolchina, T. N. Gerasimova, E. P. Fokin, *Izv. Sib. Otd. Akad. Nauk SSSR, Ser. Khim. Nauk*, 1984, **2** (No. 1), 113 (in Russian).
129. E. F. Kolchina, I. Yu. Kargapolova, T. N. Gerasimova, *Izv. Akad. Nauk, Ser. Khim.*, 1986, 1855 [*Bull. Acad. Sci. USSR, Div. Chem. Sci.*, 1986, **35**, 1685 (Engl. Transl.)].
130. E. F. Kolchina, T. N. Gerasimova, *Izv. Akad. Nauk, Ser. Khim.*, 1993, 1101 [*Russ. Chem. Bull.*, 1993, **42**, 1055 (Engl. Transl.)].
131. T. P. Culbertson, *J. Heterocycl. Chem.*, 1991, **28**, 1701.
132. R. R. Gupta, R. Kumar, R. K. Gautam, *J. Fluorine Chem.*, 1985, **28**, 381.
133. P. R. Gupta, A. Thomas, R. K. Gautam, V. Gupta, *J. Fluorine Chem.*, 1989, **44**, 1.
134. G. M. Brooke, *Tetrahedron Lett.*, 1968, **37**, 4049.
135. R. Filler, S. M. Woods, A. F. Freudenthal, *J. Org. Chem.*, 1973, **38**, 811.
136. V. P. Petrov, V. A. Barkhash, *Khim. Geterotsikl. Soedin.*, 1970, 381 [*Chem. Heterocycl. Compd.*, 1970 (Engl. Transl.)].
137. S. A. Amitina, I. V. Elrtsov, T. V. Rybalova, Yu. V. Gatilov, I. A. Grigorrev, *Izv. Akad. Nauk, Ser. Khim.*, 2004, 636 [*Russ. Chem. Bull., Int. Ed.*, 2004, **53**, 1700].
138. W. R. Dolbier, G. Burkholder, K. A. Abboud, *J. Org. Chem.*, 1994, **59**, 7688.
139. S. Buscemi, A. Pace, A. P. Piccionello, I. Pibiri, N. Vivona, *Heterocycles*, 2004, **6**, 1619.
140. PCT Int. Appl. WO 51 302 (2005); *Chem. Abstr.*, 2005, **143**, 43893b.
141. V. I. Namestnikov, Yu. G. Trishin, S. K. Bel'skii, *Zh. Org. Khim.*, 1998, **68**, 1398 [*Russ. J. Org. Chem.*, 1998, **68** (Engl. Transl.)].

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