



Facile, one-pot, four-component synthesis of a new series of imidazo[1,2-*a*]pyridines in presence of TPAB in EtOH under reflux conditions

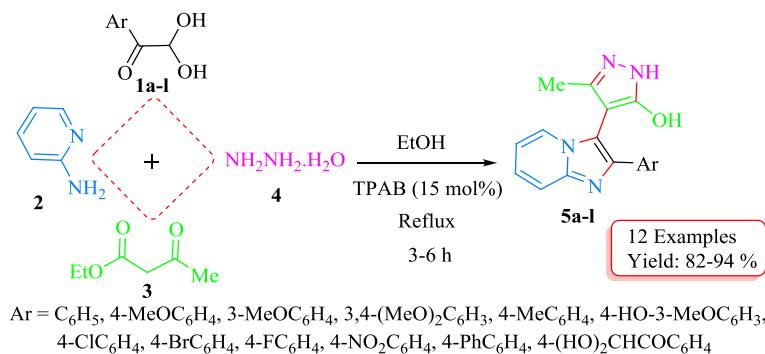
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Abstract

A convenient, regioselective, novel, and elegant one-pot, four-component reaction was designed for synthesis of a series of new of imidazo[1,2-*a*]pyridines using aryl glyoxal monohydrates, ethyl acetoacetate, hydrazine hydrate, and 2-aminopyridine in presence of tetrapropylammonium bromide under reflux in EtOH as solvent. The main advantages of this protocol include the availability and low cost of the starting materials, short reaction time, convenient operation, easy workup process, highly facile operation, nonhazardous byproducts, and high yield (82–94%).

Graphical abstract



Keywords Imidazo[1,2-*a*]pyridines · 2-Aminopyridine · Aryl glyoxal monohydrates · Ethyl acetoacetate · Hydrazine hydrate · One-pot four-component reaction · TPAB

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Extended author information available on the last page of the article

Introduction

One-pot, multicomponent reactions (MCRs) are effective, time-saving, and elaborate processes in which three or more simple and readily available components are combined together to form novel complex molecules containing features of each reagent. Therefore, these reactions have been developed as powerful, highly efficient, and useful tools for synthesis of fused heterocycle compounds in various branches of organic and medicinal chemistry [1–5], due to their minimization of cost and waste.

Imidazo[1,2-*a*]pyridines are fused [5,6]-bicyclic heterocycles with one imidazole moiety embedded with the pyridine ring, being an important class of privileged bicyclic scaffolds with widespread applications in pharmaceuticals, as organic functional materials, and for their diverse biological activities [6].

The imidazo[1,2-*a*]pyridine moiety can be found in many pharmaceuticals, natural products, bioactive compounds, biomolecules, and agrochemicals [7]. Imidazo[1,2-*a*]pyridines exhibit biological activities such as anticancer [8], antibacterial [9], antifungal [10], antiviral [11], antiinflammatory [12], antimarial [13], antiparkinsonian [14], and antituberculosis effects [15]. In addition, a few derivatives exhibit enzyme inhibition [16] and are used as drugs for Alzheimer's disease [17]. Compounds containing the imidazo[1,2-*a*]pyridine moiety are present in many natural products and marketed drugs, such as alpidem (an anxiolytic agent) [18], saripidem and necopidem (both anxiolytic agents) [19], zolpidem (an hypnotic agent used in treatment of insomnia) [20], zolimidine (an antiulcer drug) [21], miroprofen (an analgesic drug) [22], olprinone (a cardiotonic agent for treatment of acute heart failure) [23], GS812397 [with anti-human immunodeficiency virus (HIV) properties] [24], and minodronic acid (for treatment of anxiety, heart failure, and osteoporosis) [25] (Fig. 1).

All these bioactivities make the imidazo[1,2-*a*]pyridine scaffold remarkable for chemists due to its excellent pharmaceutical and medicinal applications and use in advanced organic chemistry.

Over recent years, several methods have been described for synthesis of fused bicyclic imidazo[1,2-*a*]pyridines, as summarized in a review [26].

A popular approach involves condensation reaction of 2-aminopyridines with various precursors such as α -halocarbonyl compounds [27], one-pot reaction of 2-aminopyridines, aldehydes, and nitroalkane [28], isonitrile [29], using copper catalysis, one-pot multicomponent reaction of aryl glyoxals, 2-aminopyridine, and cyclic 1,3-dicarbonyls [30], oxidative coupling between substituted 2-aminopyridine and propargylic alcohols [31], nitroolefins [32], α,β -unsaturated ketones [33], alkynes [34], copper-catalyzed oxidative coupling through C–H functionalization between pyridine derivatives and *N*-(alkylidene)-4*H*-1,2,4-triazole-4-amines [35], etc.

Nevertheless, these strategies have drawbacks in terms of harsh reaction conditions, use of excess amounts of expensive catalyst, complex workup procedures, extended reaction time, extensive by-product formation, and low yield. Hence, development of economic, practicable methods offering operational simplicity and high yield using basic chemicals as starting materials for multicomponent synthesis of this series of compounds in a single step is still desired.

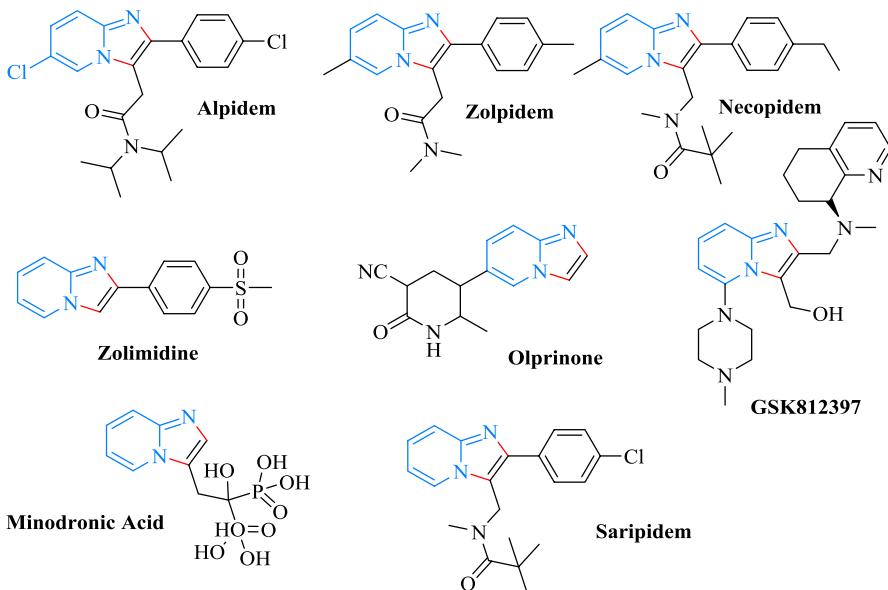


Fig. 1 Drugs possessing imidazo[1,2-*a*]pyridine unit

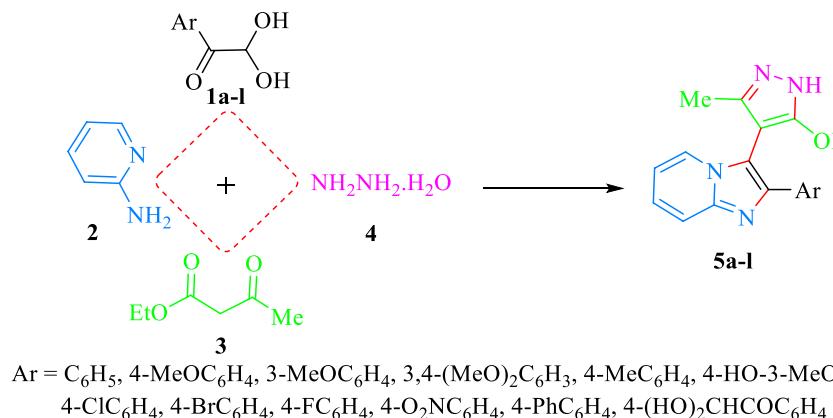
Tetrapropylammonium bromide (TPAB) was first synthesized in 1990 [36]. This quaternary ammonium salt is commonly used as a phase-transfer catalyst [37]. Recently, we used this compound as catalyst for synthesis of heterocycle scaffolds [38]. TPAB is a cheap, readily available, ecocompatible catalyst with noncorrosive nature.

Results and discussion

In continuation of our work on aryl-glyoxal-based synthesis of heterocyclic compounds using one-pot, multicomponent strategies [39–42], we report herein a convenient one-pot, four-component process for synthesis of 2,3-disubstituted imidazo[1,2-*a*]pyridine derivatives using aryl glyoxals, 2-aminopyridine, ethyl acetoacetate, and hydrazine hydrate in presence of TPAB in EtOH. This synthetic strategy is promising for synthesis of novel alkaloid structures which may exhibit pharmaceutical and biological activities (Scheme 1).

The aryl glyoxal monohydrates applied as starting material in this study were prepared by oxidation of corresponding aryl methyl ketones with SeO_2 in H_2O :dioxane under reflux condition according to standard literature process [43].

We started our study on the synthesis of a new series of imidazo[1,2-*a*]pyridine derivatives by investigating the model reaction of phenyl glyoxal monohydrate (**1a**), 2-aminopyridine (**2**), and 5-methyl-2,4-dihydro-3*H*-pyrazol-3-one (**6**) using various solvents, catalysts, and temperatures to establish the optimum reaction conditions (Table 1).



Scheme 1 Synthesis of imidazo[1,2-*a*]pyridine derivatives **5a-l** by one-pot, four-component reactions

To maintain green chemistry features, nonhazardous and safe solvents such as water, ethanol, water:ethanol, and ethylene glycol were favored as solvents in all the optimization tests. The one-pot reaction of the mixture of substrates in absence of catalyst in various solvents did not afford the desired product **5a** either at room temperature or under stirring at reflux even after 24 h (Table 1, entries 1–5).

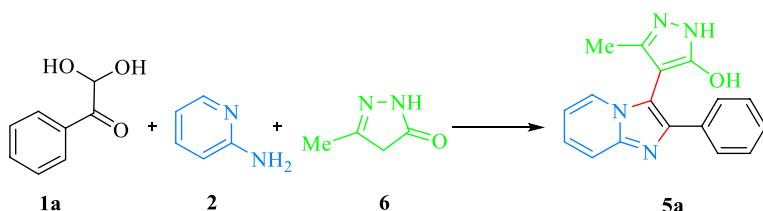
Next, the same model reaction was tested in presence of a series of Brønsted acids including *p*-toluene sulfonic acid (*p*-TSA), L-proline, L-alanine, AcOH, and sulfamic acid in refluxing H₂O or EtOH, forming the desired product in 17–35% yield (Table 1, entries 6–11).

Common basic catalysts such as Et₃N, K₂CO₃ and 1,4-diazabicyclo[2.2.2]octane (DABCO) were also investigated but unfortunately did not lead to a noticeable amount of product or increase in the yield (Table 1, entries 12–16).

When we shifted our attention to TPAB as a catalyst in water, water:acetone, or water:ethanol under reflux condition, the reaction proceeded well, and the desired product was obtained within 3 h in 65–70% yield (Table 1, entries 17–19). Finally, the optimized condition was obtained as TPAB (20 mol%) in ethanol as solvent under stirring at reflux within 3 h, which produced the desired product **5a** in 87% yield (Table 1, entry 20).

This method was successfully applied for synthesis of 3-methyl-4-(2-phenylimidazo[1,2-*a*]pyridin-3-yl)-1*H*-pyrazol-5-ol (**5a**) using phenylglyoxal monohydrate (**1a**), 2-aminopyridine (**2**), ethyl acetoacetate (**3**), and hydrazine hydrate (**4**) (Table 2). The feasibility of this four-component strategy appears to rely on the structure of 5-methyl-2,4-dihydro-3*H*-pyrazol-3-one (**6**) with the optimum reaction conditions previously set up for the three-component synthesis.

As indicated in Table 2, the reaction in absence of catalyst in various solvents under high-temperature condition failed to afford the product (Table 2, entries 1–3). In presence of TPAB (15 mol%), the reaction gave the product in 60% yield in water (Table 2, entry 4). Interestingly using the same amount of TPAB (15 mol%) in ethanol with stirring under reflux for 3 h gave the desired four-component product in 91% yield (Table 2, entry 5). Elevation or reduction of the molar ratio of catalyst

Table 1 Optimization of model one-pot, three-component reaction

Entry	Catalyst	Solvent (v/v%)	T (°C)	Time (h)	Yield (%) ^a
1	—	H ₂ O	RT	24	— ^b
2	—	H ₂ O	Reflux	24	Trace
3	—	EtOH	Reflux	24	Trace
4	—	H ₂ O:EtOH (1:1)	Reflux	24	Trace
5	—	Ethylene glycol	Reflux	24	—
6	<i>p</i> -TSA	H ₂ O	Reflux	24	25
7	<i>L</i> -Proline	H ₂ O	Reflux	24	32
8	<i>L</i> -Alanine	H ₂ O	Reflux	24	28
9	AcOH	H ₂ O	Reflux	24	20
10	Sulfamic acid	H ₂ O	Reflux	24	35
11	<i>p</i> -TSA	EtOH	Reflux	24	17
12	Et ₃ N	H ₂ O	Reflux	24	—
13	K ₂ CO ₃	H ₂ O	Reflux	24	14
14	K ₂ CO ₃	EtOH	Reflux	24	Trace
15	Et ₃ N	EtOH	Reflux	24	—
16	DABCO	H ₂ O	Reflux	24	37
17	TPAB	H ₂ O	Reflux	3	65
18	TPAB	H ₂ O:acetone (1:1)	Reflux	3	68
19	TPAB	H ₂ O:EtOH (1:1)	Reflux	3	70
20	TPAB	EtOH	Reflux	3	87

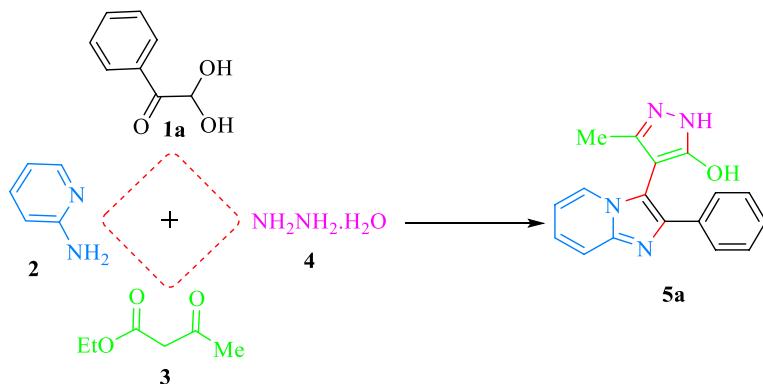
Reaction conditions: phenyl glyoxal monohydrate (**1a**, 1 mmol), 2-aminopyridine (**2**, 1 mmol), 5-methyl-2,4-dihydro-3*H*-pyrazol-3-one (**6**, 1 mmol), and catalyst (20 mol%)

^aIsolated yield

^bReaction failed to occur

or changing the temperature to room temperature with duration of 3–10 h could not improve the yield (Table 2, entries 6–9). The reusability of TPAB in synthesis of **5a** was also examined. For this purpose, the recovered catalyst was recycled up to four times; the catalytic performance results are shown in Fig. 2.

Using the optimized condition, different aryl glyoxal derivatives **1a–l** (monohydrate form), 2-aminopyridine (**2**), ethyl acetoacetate (**3**), and hydrazine hydrate (**4**) also afforded the desired products **5a–l** in high yield. The results obtained with

Table 2 Optimization of model four-component reaction for synthesis of **5a**

Entry	Catalyst (mol%)	Solvent (v/v%)	T (°C)	Time (h)	Yield (%) ^a
1	–	H ₂ O	100	24	– ^b
2	–	EtOH	80	24	–
3	–	H ₂ O:EtOH (1:1)	Reflux	24	–
4	TPAB (15%)	H ₂ O	Reflux	3	60
5	TPAB (15%)	EtOH	Reflux	3	91^c
6	TPAB (20%)	EtOH	Reflux	5	88
7	TPAB (10%)	EtOH	Reflux	5	84
8	TPAB (5%)	EtOH	Reflux	5	73
9	TPAB (15%)	EtOH	RT	10	62

Reaction conditions: phenylglyoxal monohydrate (**1a**, 1 mmol), 2-aminopyridine (**2**, 1 mmol) ethyl acetoacetate (**3**, 1 mmol), and hydrazine hydrate (**4**, 1 mmol)

^aIsolated yield

^bReaction failed to occur

^cThe bold type (entry 5) refers to the best reaction conditions

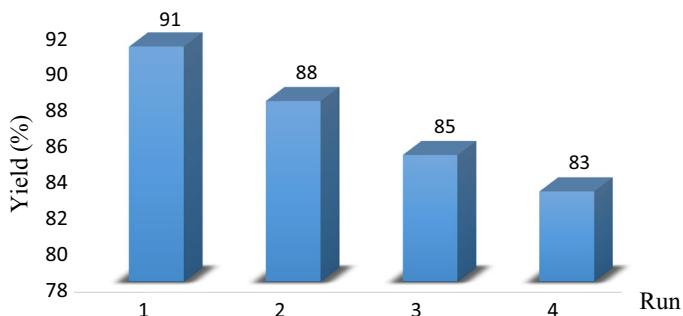
**Fig. 2** Reusability of TPAB in synthesis of imidazo[1,2-a]pyridine derivatives

Table 3 Synthesis of imidazo[1,2-*a*]pyridines **5a–l** via one-pot, four-component reaction

Entry	Aryl glyoxal monohydrates	Products	Time (h)	M.P. (°C)	Yield (%) ^b
1			4	290 (Dec.)	91
2			3	294 (Dec.)	89
3			3	168-169	85
4			3	168-169	94
5			3	294 (Dec.)	88
6			4	266-267	91
7			3	307 (Dec.)	92
8			3	300 (Dec.)	90
9			3	290 (Dec.)	86
10			6	300 (Dec.)	82
11			4	290 (Dec.)	87
12			6	324 (Dec.)	84

Table 3 (continued)

Reaction conditions for synthesis of **5a–k**: **1a–k** (1 mmol), **2** (1 mmol), **3** (1 mmol), **4** (1 mmol); **5l**: **1l** (1 mmol), **2** (2 mmol), **3** (2 mmol), **4** (2 mmol), and TPAB (20 mol%) in EtOH (5–10 mL) under reflux for 3–6 h

^aIsolated yield

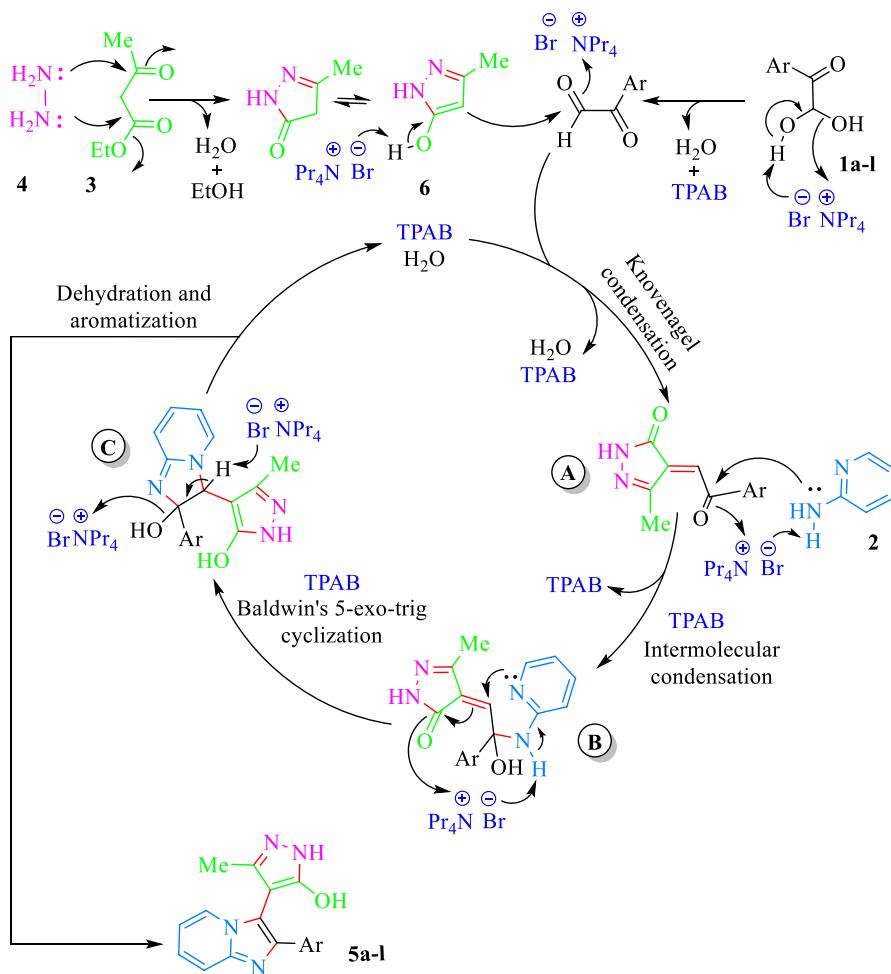
various aryl glyoxals, including the product, melting point (m.p.), and yield, are summarized in Table 3.

The structure of all the imidazo[1,2-*a*]pyridine derivatives **5a–l** was elucidated using Fourier-transform infrared (FT-IR) spectroscopy, ¹H and ¹³C nuclear magnetic resonance (NMR) spectroscopy, and mass analysis. The FT-IR (KBr) spectra of **5a–l** showed absorption bonds due to vibrations of NH and OH group at 3433–3243 cm⁻¹. In the ¹H NMR spectra, N–H and OH group of the products showed broad doublets at δ =11.20–10.92 ppm, which disappeared on addition of D₂O. The protons of aromatic rings were located at around δ =8.22–6.73 ppm. The methyl hydrogens of the pyrazole ring of the products appeared as a singlet at δ =1.83–1.58 ppm, and in the ¹³C NMR spectra as a carbon signal located at around δ =11.7–10.2 ppm. In the ¹³C NMR spectra of **5a–l**, signals located at around δ =172.5–160.1 ppm and δ =160.3–159.0 ppm were ascribed to C–OH and N–C=N group, respectively. The signals located at around δ =148.9–90.7 ppm were assigned to other carbons of aromatic rings. The analytical and spectral data of the synthesized scaffolds are all in agreement with their proposed structure.

A reasonable reaction mechanism for the synthesis of imidazo[1,2-*a*]pyridine derivatives **5a–l** by a one-pot four-component strategy is depicted in Scheme 2. In the first step, 5-methyl-2,4-dihydro-3*H*-pyrazol-3-one (**6**) forms by condensation of hydrazine hydrate (**4**) and ethyl acetoacetate (**3**), which would be stabilized by keto–enol tautomerization. In the next step, intermediate **6** in enolate form reacts with aryl glyoxal monohydrate **1a–l** in presence of TPAB to form intermediate **A**. TPAB plays an important role in the formation of intermediate **A** via Knoevenagel condensation followed by loss of one molecule of water. Subsequently, 2-aminopyridine (**2**) undergoes intermolecular condensation with the carbonyl group of intermediate **A** to form intermediate **B**, which then undergoes Baldwin 5-*exo*–trig cyclization [44] to give intermediate **C**. Finally, loss of a water molecule from intermediate **C** gives the desired imidazo[1,2-*a*]pyridine derivative **5a–l** in presence of TPAB, as shown in Scheme 2.

Conclusions

We developed a simple and efficient protocol for regioselective synthesis of a series of [5,6]-bicyclic fused imidazo[1,2-*a*]pyridines by one-pot, four-component reaction between aryl glyoxal monohydrates, 2-aminopyridine, ethyl acetoacetate, and hydrazine hydrate in EtOH under reflux conditions, using TPAB as an inexpensive, readily available, and ecocompatible catalyst, showing advantages in terms of reaction time, easy workup, and facile operation. Considering the presence of pyrazole



Scheme 2 Mechanistic rationalization for synthesis of 3-methyl-4-(2-arylimidazo[1,2-*a*]pyridin-3-yl)-1*H*-pyrazol-5-ol derivatives **5a-l**

linked to the 3-position of the imidazo[1,2-*a*]pyridine moiety, it is expected that these fused structures will show promising pharmaceutical activities. The use of an inexpensive catalyst, metal-free conditions, short reaction time, high yield, and lack of hazardous byproducts are the main advantages of this procedure.

Experimental

Materials and methods

The chemicals used in this work were obtained from Arcos and Merck companies and used without purification. Melting points were measured on a Philip Harris C4954718 apparatus and are uncorrected. Reaction progress was monitored by thin-layer chromatography (TLC) on Merck silica plates. Infrared spectra were recorded on a Thermo Nicolet Nexus 670 FT-IR instrument using KBr discs. ^1H and ^{13}C NMR spectra were recorded on a Bruker Advance AQS 300 MHz spectrometer at 300 and 75.5 MHz, respectively. Chemical shifts were measured in dimethyl sulfoxide ($\text{DMSO}-d_6$) as solvent relative to tetramethylsilane (TMS) as internal standard. Mass spectra and high-resolution mass spectra were recorded on a Kratos MS 25RF spectrometer; relative abundance of fragments is quoted in parentheses after m/z values.

General procedure for synthesis of 3-methyl-4-(2-arylimidazo[1,2-*a*]pyridin-3-yl)-1*H*-pyrazol-5-ol derivatives (**5a–k**)

Hydrazine hydrate (**4**, 1 mmol) was added dropwise to a solution of ethyl acetoacetate (**3**, 1 mmol) in ethanol (5 mL). The reaction mixture was stirred at room temperature and monitored by TLC (using chloroform:methanol 20:1 as eluent). After reaction completion (5–10 min), white precipitate of 5-methyl-2,4-dihydro-3*H*-pyrazol-3-one (**6**) formed. To the reaction mixture, aryl glyoxal monohydrate **1a–k** (1 mmol), 2-aminopyridine (**2**, 1 mmol), and TPAB (15 mol%) were added, and the mixture was refluxed for appropriate time as mentioned in Table 3. After reaction completion as monitored by TLC (using chloroform:methanol 3:1 as eluent), the reaction mixture was cooled to room temperature and the precipitate filtered, washed with water, and cold ethanol to give the desired products **5a–k** in high yield (82–94%).

Synthesis of 4,4'-(1,4-phenylenebis(imidazo[1,2-*a*]pyridine-2,3-diyl)) bis(3-methyl-1*H*-pyrazol-5-ol) (**5l**)

Hydrazine hydrate (**4**, 2 mmol) was added dropwise to a solution of ethyl acetoacetate (**3**, 2 mmol) in ethanol (10 mL). The reaction mixture was stirred at room temperature and monitored by TLC (using chloroform:methanol 20:1 as eluent). After reaction completion (5–10 min), 5-methyl-2,4-dihydro-3*H*-pyrazol-3-one (**6**) formed as white precipitate. To the reaction mixture, bis-phenylene glyoxal monohydrate (**1l**, 1 mmol), 2-aminopyridine (**2**, 2 mmol), and TPAB (15 mol%) were added, and the mixture was refluxed for 6 h. After reaction completion as monitored by TLC (using chloroform:methanol 3:1 as eluent), the reaction mixture was cooled to room temperature, and the orange precipitate filtered, washed with water, and cold ethanol to give the desired product **5l** in 84% yield.

Physical and spectral data for compounds 5a–l

3-Methyl-4-(2-phenylimidazo[1,2-*a*]pyridin-3-yl)-1*H*-pyrazol-5-ol (5a)

White solid; yield 91% (264 mg); m.p. 290 °C (dec.). FT-IR (KBr) ν : 3243, 2976, 2923, 1633, 1606, 1571, 1505, 1440, 1337, 1283, 1236, 1149, 1106, 1031, 983, 738, 694 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆) δ (ppm): 11.78 (bs, 1H, OH, D₂O exch.), 10.36 (bs, 1H, NH, D₂O exch.), 7.85 (d, *J*=6.9 Hz, 1H, Ar), 7.78 (d, *J*=7.2 Hz, 2H, Ar), 7.61 (d, *J*=9 Hz, 1H, Ar), 7.37–7.26 (m, 4H, Ar), 6.88 (t, *J*=6.9 Hz, 1H, Ar), 1.76 (s, 3H, Me). ¹³C NMR (75.5 MHz, DMSO-*d*₆) δ (ppm): 160.2, 160.1, 144.7, 142.3, 135.4, 132.0, 129.7, 128.6, 127.9, 126.7, 125.9, 124.3, 112.9, 91.3, 10.2. LRMS (EI, 70 eV) *m/z* (%): 290 (M⁺, 100), 275 (5), 262 (13), 244 (5), 233 (29), 218 (19), 205 (22), 194 (6), 179 (3), 167 (2), 154 (4), 140 (2), 128 (8), 115 (5), 105 (7), 91 (3), 78 (36), 65 (4), 51 (3). HRMS (ESI): *m/z* (M)⁺ calcd. for C₁₇H₁₄N₄O⁺: 290.1168; found: 290.1163.

4-(2-(4-Methoxyphenyl)imidazo[1,2-*a*]pyridin-3-yl)-3-methyl-1*H*-pyrazol-5-ol (5b)

White solid; yield 89% (285 mg); m.p. 294 °C (dec.). FT-IR (KBr) ν : 3433, 3110, 3036, 3005, 2907, 2837, 2782, 2613, 1614, 1512, 1459, 1385, 1299, 1255, 1174, 1111, 1032, 973, 837, 788, 754 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆) δ (ppm): 11.05 (bs, 2H, OH, NH, D₂O exch.), 7.83 (d, *J*=6.6 Hz, 1H, Ar), 7.72 (d, *J*=6.9 Hz, 2H, Ar), 7.58 (d, *J*=8.7 Hz, 1H, Ar), 7.24 (t, *J*=6.9 Hz, 1H, Ar), 6.93–6.82 (m, 3H, Ar), 3.76 (s, 3H, OMe), 1.78 (s, 3H, Me). ¹³C NMR (75.5 MHz, DMSO-*d*₆) δ (ppm): 160.2, 159.0, 144.7, 142.4, 140.3, 129.4, 127.9, 127.2, 125.8, 124.0, 114.7, 113.7, 111.9, 91.4, 56.2, 10.3. LRMS (EI, 70 eV) *m/z* (%): 320 (M⁺, 100), 305 (31), 290 (5), 277 (5), 262 (17), 248 (6), 234 (4), 219 (10), 205 (6), 192 (5), 181 (2), 170 (2), 153 (5), 142 (2), 131 (3), 115 (3), 103 (3), 91 (2), 78 (10), 65 (2), 51 (3). HRMS (ESI): *m/z* (M)⁺ calcd. for C₁₈H₁₆N₄O₂⁺: 320.1273; found: 320.1275.

4-(2-(3-Methoxyphenyl)imidazo[1,2-*a*]pyridin-3-yl)-3-methyl-1*H*-pyrazol-5-ol (5c)

White solid; yield 85% (272 mg); m.p. 168–169 °C. FT-IR (KBr) ν : 3407, 3206, 2943, 2840, 1606, 1503, 1459, 1435, 1390, 1343, 1288, 1246, 1219, 1042, 787, 754 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆) δ (ppm): 11.00 (bs, 2H, OH, NH, D₂O exch.), 7.85 (d, *J*=6.3 Hz, 1H, Ar), 7.62 (d, *J*=9 Hz, 1H, Ar), 7.36 (d, *J*=8.1 Hz, 2H, Ar), 7.30–7.22 (m, 2H, Ar), 6.90–6.82 (m, 2H, Ar), 3.72 (s, 3H, OMe), 1.79 (s, 3H, Me). ¹³C NMR (75.5 MHz, DMSO-*d*₆) δ (ppm): 160.2, 159.6, 144.6, 142.1, 140.4, 136.6, 130.8, 128.8, 126.3, 124.3, 123.8, 120.3, 118.4, 113.1, 111.2, 91.2, 55.9, 10.2. LRMS (EI, 70 eV) *m/z* (%): 320 (M⁺, 100), 305 (17), 290 (10), 275 (3), 263 (20), 247 (8), 231 (5), 219 (19), 205 (14), 192 (5), 159 (4), 145 (2), 127 (2), 109 (3), 98 (5), 78 (19), 67 (4), 57 (8). HRMS (ESI): *m/z* (M)⁺ calcd. for C₁₈H₁₆N₄O₂⁺: 320.1273; found: 320.1278.

4-(2-(3,4-Dimethoxyphenyl)imidazo[1,2-*a*]pyridin-3-yl)-3-methyl-1*H*-pyrazol-5-ol (5d**)**

White solid; yield 94% (329 mg); m.p. 168–169 °C; FT-IR (KBr) ν : 3367, 3097, 2995, 2940, 2836, 1589, 1523, 1464, 1434, 1392, 1257, 1237, 1178, 1142, 1024, 757 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆) δ (ppm): 11.00 (bs, 2H, OH, NH, D₂O exch.), 7.84 (d, *J*=6.6 Hz, 1H, Ar), 7.60 (d, *J*=9 Hz, 1H, Ar), 7.45 (s, 1H, Ar), 7.32–7.23 (m, 2H, Ar), 6.93 (d, *J*=8.4 Hz, 1H, Ar), 6.86 (t, *J*=6.6 Hz, 1H, Ar), 3.75 (s, 3H, OMe), 3.68 (s, 3H, OMe), 1.70 (s, 3H, Me). ¹³C NMR (75.5 MHz, DMSO-*d*₆) δ (ppm): 172.5, 160.3, 148.9, 144.5, 142.4, 140.4, 128.0, 126.1, 124.0, 120.4, 118.5, 115.7, 113.4, 112.6, 112.0, 91.4, 56.6, 55.2, 10.2. LRMS (EI, 70 eV) *m/z* (%): 350 (M⁺, 44), 335 (8), 320 (100), 305 (23), 291 (6), 277 (10), 262 (31), 247 (8), 234 (7), 219 (19), 205 (17), 192 (7), 179 (3), 160 (3), 145 (2), 130 (3), 115 (3), 105 (4), 93 (3), 78 (19), 67 (2), 57 (5). HRMS (ESI): *m/z* (M)⁺ calcd. for C₁₉H₁₈N₄O₃⁺: 350.1379; found: 350.1378.

3-Methyl-4-(2-(*p*-tolyl)imidazo[1,2-*a*]pyridin-3-yl)-1*H*-pyrazol-5-ol (5e**)**

White solid; yield 88% (268 mg); m.p. 294 °C (dec.). FT-IR (KBr) ν : 3405, 3182, 3133, 3028, 2923, 2862, 2738, 1609, 1520, 1476, 1443, 1419, 1387, 1332, 1237, 1100, 823, 747 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆) δ (ppm): 10.92 (bs, 2H, OH, NH, D₂O exch.), 7.82 (d, *J*=7.2 Hz, 1H, Ar), 7.66 (d, *J*=7.5 Hz, 2H, Ar), 7.58 (d, *J*=9.6 Hz, 1H, Ar), 7.26 (t, *J*=7.5 Hz, 1H, Ar), 7.15 (d, *J*=7.8 Hz, 2H, Ar), 6.87 (t, *J*=6.6 Hz, 1H, Ar), 2.29 (s, 3H, Me), 1.58 (s, 3H, Me). ¹³C NMR (75.5 MHz, DMSO-*d*₆) δ (ppm): 160.2, 160.1, 144.7, 142.4, 140.2, 136.9, 132.5, 130.4, 128.6, 128.1, 126.1, 124.1, 112.5, 91.4, 22.4, 10.3. LRMS (EI, 70 eV) *m/z* (%): 304 (M⁺, 100), 289 (18), 273 (2), 261 (3), 247 (14), 232 (7), 219 (5), 205 (2), 195 (2), 183 (2), 169 (2), 154 (2), 144 (4), 130 (3), 119 (5), 103 (2), 91 (5), 78 (14), 65 (2), 51 (3). HRMS (ESI): *m/z* (M)⁺ calcd. for C₁₈H₁₆N₄O⁺: 304.1324; found: 304.1319.

4-(2-(4-Hydroxy-3-methoxyphenyl)imidazo[1,2-*a*]pyridin-3-yl)-3-methyl-1*H*-pyrazol-5-ol (5f**)**

White solid; yield 91% (306 mg); m.p. 266–267 °C. FT-IR (KBr) ν : 3351, 3094, 3025, 2925, 2726, 2572, 1600, 1576, 1531, 1487, 1465, 1436, 1395, 1285, 1254, 1217, 1172, 1138, 1112, 1035, 819, 786, 754, 683, 640 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆) δ (ppm): 10.95 (bs, 2H, OH, NH, D₂O exch.), 9.05 (s, 1H, OH, D₂O exch.), 7.81 (d, *J*=6.9 Hz, 1H, Ar), 7.57 (d, *J*=8.7 Hz, 1H, Ar), 7.42 (s, 1H, Ar), 7.26–7.17 (m, 2H, Ar), 6.85 (t, *J*=6.6 Hz, 1H, Ar), 6.73 (d, *J*=8.1 Hz, 1H, Ar), 3.70 (s, 3H, OMe), 1.80 (s, 3H, Me). ¹³C NMR (75.5 MHz, DMSO-*d*₆) δ (ppm): 160.6, 160.2, 147.8, 146.5, 144.5, 142.8, 140.4, 126.7, 123.8, 120.7, 118.8, 117.0, 115.8, 144.7, 111.6, 91.5, 56.7, 11.7. LRMS (EI, 70 eV) *m/z* (%): 336 (M⁺, 100), 320 (16), 306 (12), 293 (3), 277 (6), 264 (5), 247 (3), 235 (4), 221 (3), 207 (4), 193 (2), 181 (1), 168 (4), 161 (2), 153 (1), 139 (1), 131 (1), 118 (1), 110 (1). HRMS (ESI): *m/z* (M)⁺ calcd. for C₁₈H₁₆N₄O₃⁺: 336.1222; found: 336.1227.

4-(2-(4-Chlorophenyl)imidazo[1,2-*a*]pyridin-3-yl)-3-methyl-1*H*-pyrazol-5-ol (5g**)**

White solid; yield 92% (298 mg); m.p. 317 °C (dec.). FT-IR (KBr) ν : 3427, 3174, 3130, 3027, 2927, 2860, 2740, 1610, 1542, 1509, 1464, 1443, 1418, 1387, 1329, 1283, 1238, 1098, 974, 831, 743 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆) δ (ppm): 11.10 (bs, 2H, OH, NH, D₂O exch.), 7.85 (d, *J*=6.6 Hz, 1H, Ar), 7.79 (d, *J*=8.4 Hz, 2H, Ar), 7.60 (d, *J*=9 Hz, 1H, Ar), 7.41 (d, *J*=8.4 Hz, 2H, Ar), 7.29 (t, *J*=7.5 Hz, 1H, Ar), 6.89 (t, *J*=6.9 Hz, 1H, Ar), 1.80 (s, 3H, Me). ¹³C NMR (75.5 MHz, DMSO-*d*₆) δ (ppm): 160.1, 160.0, 144.9, 141.1, 140.3, 134.3, 132.3, 129.7, 127.5, 126.5, 124.6, 118.3, 113.2, 91.0, 10.6. LRMS (EI, 70 eV) *m/z* (%): 326 [(M+2)⁺, 45], 324 (M⁺, 100), 309 (18), 295 (3), 281 (3), 267 (14), 254 (4), 243 (2), 231 (9), 218 (3), 205 (5), 186 (1), 174 (1), 162 (3), 144 (2), 128 (2), 116 (1), 103 (2), 89 (1), 78 (10), 65 (1), 52 (1). HRMS (ESI): *m/z* (M)⁺ calcd. for C₁₇H₁₃ClN₄O⁺: 324.0778; found: 324.0783.

4-(2-(4-Bromophenyl)imidazo[1,2-*a*]pyridin-3-yl)-3-methyl-1*H*-pyrazol-5-ol (5h**)**

White solid; yield 90% (331 mg); m.p. 300 °C (dec.). FT-IR (KBr) ν : 3430, 3178, 3130, 3028, 2926, 2858, 2738, 1610, 1505, 1540, 1465, 1441, 1418, 1385, 1327, 1238, 1100, 1077, 1007, 826, 744 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆) δ (ppm): 11.00 (bs, 2H, OH, NH, D₂O exch.), 7.85 (d, *J*=6.6 Hz, 1H, Ar), 7.72 (d, *J*=8.4 Hz, 2H, Ar), 7.62–7.53 (m, 1H, Ar), 7.28 (t, *J*=7.5 Hz, 2H, Ar), 6.89 (t, *J*=6.6 Hz, 1H, Ar), 1.80 (s, 3H, Me). ¹³C NMR (75.5 MHz, DMSO-*d*₆) δ (ppm): 160.5, 160.1, 144.6, 141.1, 134.6, 132.9, 130.7, 130.1, 127.8, 126.5, 124.5, 120.9, 113.3, 91.0, 11.5. LRMS (EI, 70 eV) *m/z* (%): 370 [(M+2)⁺, 100], 368 (M⁺, 90), 353 (16), 341 (1), 324 (9), 311 (10), 300 (2), 288 (2), 274 (9), 262 (12), 243 (3), 231 (17), 218 (5), 205 (8), 184 (2), 168 (1), 154 (1), 137 (7), 126 (1), 115 (1), 103 (2), 89 (1), 78 (17), 63 (1), 51 (2). HRMS (ESI): *m/z* (M)⁺ calcd. for C₁₇H₁₃BrN₄O⁺: 368.0273; found: 368.0279.

4-(2-(4-Fluorophenyl)imidazo[1,2-*a*]pyridin-3-yl)-3-methyl-1*H*-pyrazol-5-ol (5i**)**

White solid; yield 86% (265 mg); m.p. 290 °C (dec.). FT-IR (KBr) ν : 3174, 3127, 3029, 2927, 2862, 2745, 1608, 1519, 1477, 1420, 1388, 1331, 1298, 1233, 1155, 1099, 840, 803, 748 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆) δ (ppm): 11.02 (bs, 2H, OH, NH, D₂O exch.), 7.86–7.78 (m, 3H, Ar), 7.60 (d, *J*=9 Hz, 1H, Ar), 7.27 (t, *J*=7.8 Hz, 1H, Ar), 7.19 (t, *J*=8.7 Hz, 2H, Ar), 6.88 (t, *J*=6.6 Hz, 1H, Ar), 1.78 (s, 3H, Me). ¹³C NMR (75.5 MHz, DMSO-*d*₆) δ (ppm): 163.5, 160.1, 144.8, 141.5, 140.3, 131.9, 130.1, 127.8, 126.3, 123.9, 116.7, 114.9, 112.7, 91.1, 10.3. LRMS (EI, 70 eV): *m/z* (%)=308 (M⁺, 100), 293 (14), 279 (3), 264 (2), 251 (17), 236 (8), 223 (11), 212 (2), 199 (1), 187 (1), 172 (3), 158 (2), 145 (4), 133 (2), 118 (1), 103 (1), 90 (1), 78 (17), 63 (2), 51 (4). HRMS (ESI): *m/z* (M)⁺ calcd. for C₁₇H₁₃FN₄O⁺: 308.1073; found: 308.1075.

3-Methyl-4-(2-(4-nitrophenyl)imidazo[1,2-*a*]pyridin-3-yl)-1*H*-pyrazol-5-ol (5j**)**

Yellow solid; yield 82% (275 mg); m.p. 300 °C (dec.). FT-IR (KBr) ν : 3418, 3110, 3028, 2888, 2783, 2614, 1602, 1513, 1453, 1346, 1280, 1108, 858, 753 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆) δ (ppm): 11.20 (bs, 2H, OH, NH, D₂O exch.), 8.22 (d, *J*=8.7 Hz, 2H, Ar), 8.03 (d, *J*=8.4 Hz, 2H, Ar), 7.90 (d, *J*=6.6 Hz, 1H, Ar), 7.65 (d, *J*=9 Hz, 1H, Ar), 7.34 (t, *J*=7.8 Hz, 1H, Ar), 6.94 (t, *J*=6.6 Hz, 1H, Ar), 1.83 (s, 3H, Me). ¹³C NMR (75.5 MHz, DMSO-*d*₆) δ (ppm): 160.8, 160.0, 146.5, 145.1, 142.1, 139.9, 128.9, 128.6, 127.2, 126.7, 125.2, 123.3, 115.3, 90.7, 10.4. LRMS (EI, 70 eV) *m/z* (%): 335 (M⁺, 100), 324 (12), 314 (1), 304 (8), 288 (10), 274 (18), 262 (6), 244 (2), 231 (7), 218 (4), 205 (4), 192 (1), 177 (1), 168 (2), 150 (1), 140 (1), 123 (1), 113 (1), 103 (1), 92 (1), 78 (1), 67 (1), 56 (1). HRMS (ESI): *m/z* (M)⁺ calcd. for C₁₇H₁₃N₅O₃⁺: 335.1018; found: 335.1015.

4-(2-([1,1'-Biphenyl]-4-yl)imidazo[1,2-*a*]pyridin-3-yl)-3-methyl-1*H*-pyrazol-5-ol (5k**)**

White solid; yield 87% (319 mg); m.p. 290 °C (dec.). FT-IR (KBr) ν : 3415, 3030, 2875, 2778, 2607, 1613, 1502, 1448, 1386, 1344, 1281, 1111, 847, 761, 740, 694 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆) δ (ppm): 11.15 (bs, 2H, OH, NH, D₂O exch.), 7.90 (t, *J*=8.1 Hz, 3H, Ar), 7.68–7.58 (m, 5H, Ar), 7.45 (t, *J*=6.9 Hz, 2H, Ar), 7.36–7.25 (m, 2H, Ar), 6.89 (t, *J*=6.9 Hz, 1H, Ar), 1.83 (s, 3H, Me). ¹³C NMR (75.5 MHz, DMSO-*d*₆) δ (ppm): 160.4, 160.2, 144.9, 141.9, 140.1, 139.1, 134.5, 130.5, 128.8, 128.6, 126.4, 124.3, 123.9, 118.2, 116.0, 113.1, 112.1, 91.3, 11.7. LRMS (EI, 70 eV) *m/z* (%): 366 (M⁺, 100), 351 (15), 337 (2), 320 (8), 309 (14), 294 (7), 281 (8), 262 (5), 243 (3), 231 (10), 218 (4), 205 (7), 194 (2), 183 (4), 165 (2), 154 (3), 141 (1), 130 (3), 115 (1), 103 (1), 91 (1), 78 (15), 67 (1), 52 (1). HRMS (ESI): *m/z* (M)⁺ calcd. for C₂₃H₁₈N₄O⁺: 366.1481; found: 366.1489.

4,4'-(1,4-Phenylenebis(imidazo[1,2-*a*]pyridine-2,3-diyl)) bis(3-methyl-1*H*-pyrazol-5-ol) (5l**)**

Pale-yellow solid; yield 84% (422 mg); m.p. 324 °C (dec.). FT-IR (KBr) ν : 3388, 3215, 3043, 2926, 1609, 1508, 1439, 1388, 1343, 1239, 1104, 1046, 978, 843, 743 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆) δ (ppm): 11.01 (bs, 4H, 2×OH, 2×NH, D₂O exch.), 7.85 (d, *J*=6.6 Hz, 2H, Ar), 7.76 (s, 4H, Ar), 7.60 (d, *J*=8.7 Hz, 2H, Ar), 7.28 (t, *J*=8.1 Hz, 2H, Ar), 6.89 (t, *J*=6.6 Hz, 2H, Ar), 1.79 (s, 6H, 2×Me). ¹³C NMR (75.5 MHz, DMSO-*d*₆) δ (ppm): 160.3, 160.1, 144.7, 142.0, 140.4, 134.1, 128.0, 125.8, 123.9, 118.1, 115.9, 113.1, 91.3, 11.6. LRMS (EI, 70 eV) *m/z* (%): 502 (M⁺, 15), 369 (32), 354 (10), 313 (11), 299 (10), 285 (10), 260 (10), 236 (23), 222(10), 211 (11), 152 (11), 135 (10), 123 (13), 109 (19), 97 (37), 83 (52), 71 (53), 69 (58), 57 (86), 55 (67), 43 (100), 41 (57). HRMS (ESI): *m/z* (M)⁺ calcd. for C₂₈H₂₂N₈O₂⁺: 502.1866; found: 502.1861.

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