



Effect of Oat β -Glucan on the Structure and Properties of Soybean Protein Isolate During Maillard Reaction

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Abstract

Maillard reaction (MR) with oat β -glucan changed the structure of soybean protein isolate (SPI), further leading to the enhancement of its functional properties. SPI was unfolded by MR, and the SPI conjugates with high molecular weight were identified. The water solubility of SPI was improved by cross-linking with hydrophilic β -glucan, while the hydrophobicity also increased along with the unfolding of the SPI. Cross-linking with β -glucan elevated the viscosity of SPI, thus enhancing viscosity-related physiological activities, including bile acid binding ability, fat binding capacity, and hypoglycemic activity, and the functional properties increased as the β G content involved in MR increased.

Keywords Soybean protein isolate · Maillard reaction · Oat β -glucan · Functional properties

Introduction

As an ideal edible protein resource, soybean protein isolate exhibited distinct functional characteristics. Such as solubility, emulsifying property, gel property, and blistering property [1]. Due to its nutritional and functional advantages, SPI exhibits a huge application potential in the food industry. However, SPI may lose stability during different process conditions, such as a pH near its isoelectric point, an extreme ionic intensity, or a high temperature, hence resulting in reduced functional properties and further limiting its application in food industry. Therefore, some modification methods were used to improve the functional properties of SPI.

Possessing the character of green, mild, and efficiency, Maillard reaction (MR) is one of the most promising ways to improve the structure and function of protein [2]. Compared with monosaccharides and oligosaccharides, polysaccharides exhibit stronger molecular steric hindrance. The MR with polysaccharides endowed protein with more superior functional properties [2]. Xu's work showed that chitosan enhanced the emulsifying property of SPI [3]. The solubility

of pea protein isolates increased after MR with maltodextrin [4]. Pleurotus ostreatus β -glucan endowed oat protein with superior thermal stability and enhanced solubility [5]. Rice protein-glucan conjugates exhibited enhanced solubility, foaming and emulsifying properties [6].

As a linear non-starch polysaccharide, oat β -glucan (β G) exhibited many physiological activities, among which the hypoglycemic activity is the most impressive [7]. β G possesses high viscosity, water solubility, and because of many hydroxyl groups on its molecule, it can be expected to bring the improved water solubility to SPI through MR, and thus enhancing the functional property of SPI.

Research showed that the structure and functional properties of proteins conjugates may depend on the type and content of polysaccharides during MR [6, 8, 9]. In this work, MR was used to synthesize three SPI- β G conjugates. Then the structural property and functional characteristics of the conjugates were investigated. Our results will provide some foundation for the utilization of soybean protein isolate in food industry.

Materials and Methods

The material and methods section was presented as supplementary material.

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Results and Discussion

Structural Information

Infrared Spectral Characterization and Secondary Structure

As depicted in Fig. S1a, all the samples showed similar infrared characteristic peak. For example, the C=O stretching vibrations at 1631 cm^{-1} indicated amide I band, the peak at 1520 cm^{-1} was N-H of amide II bending vibration, and the absorption band around 1240 cm^{-1} was due to C-N stretching vibration and N-H deformation of amide III. The broad absorption around 3016 cm^{-1} and 3637 cm^{-1} resulted from the SPI's N-H stretching vibration [10]. Compared with SPI, the slightly broadened absorption at $3020\text{--}3682\text{ cm}^{-1}$ of conjugates was related to hydroxyl groups of β G. The enhanced absorption near 1036 cm^{-1} of SPI conjugates indicated the C-N bonds between SPI and β G [11]. The reduced absorption band at 1523 cm^{-1} of SPI conjugates revealed that amino group of SPI was consumed during MR.

The secondary structure was represented in Table S1. The α -helix and β -turn content of SPI increased slightly with the increasing β G in MR. Meanwhile, the random coils and β -sheets content of SPI displayed the opposite trend. The result may be caused by the denaturation of SPI, or by the cross-linking between SPI and polysaccharides [12]. Chen's work showed that the change in the secondary structure resulted in a significant difference in the functional properties of protein [13].

Degree of Glycation

As shown in Table S2, the more β G content in Maillard reaction, the higher degree of glycation (DG) for SPI conjugates can be obtained, which is caused by the more cross-linking between β G and SPI. As expected, SPI: β G 1:1 showed the highest DG (21.01%). The phenomenon revealed that higher content of carbonyl groups can react with more amino groups of SPI [14], which further affect its functional property.

SDS-PAGE

The SDS-PAGE of samples was represented in Fig. S1b. The characteristic bands of protein can be seen on the visualized gel of SPI (lane 1), and the banding pattern of SPI changed after MR. The conjugates (lanes 2 and 3) showed some darker and deeper stripes around 250 KDa, demonstrating components with high molecular weight were formed during MR of SPI and β G [15]. The color of the SPI conjugates bands at 7 and 11 S gradually became lighter as

the β G content involved in MR increased. In summary, the increase in protein molecular weight is considered to be a key indicator of conjugates formation after Maillard reaction [5].

Fluorescence Spectrum

Fluorescence spectrum is affected by the tryptophan (Trp) group of protein and could assess SPI's conformational variation and tertiary structure. MR with β G altered the structure of SPI, which influenced the exposure of Trp group. SPI's fluorescence intensity exhibited a decrease and decreased as the β G content involved in MR increased (Fig. S1c). The phenomenon may be caused by the fact that shielding of Trp group by cross-linking between polysaccharides and proteins can lead to a reduction in fluorescence intensity [16]. The λ_{max} (maximum absorption wavelength) indicates the microenvironment where Trp group is located, when $\lambda_{\text{max}} > 330\text{ nm}$ the Trp group is exposed to a polar microenvironment. The λ_{max} was 337 (SPI), 341 (SPI: β G 4:1), 342 (SPI: β G 2:1), and 344 (SPI: β G 1:1) nm, respectively. The red-shift of λ_{max} in SPI after MR suggested that SPI has been unfolded by MR and Trp group is more exposed to the solvent [17]. The polarity of the microenvironment where the fluorescence emitting group Trp is located increased, making the peptide chain more extended, and thus the spatial conformation of the protein changed. Meanwhile, the hydrophobic groups were also exposed by unfolded SPI, leading to SPI's increased surface hydrophobicity. The covalent bonding with β G altered the structure of SPI, which resulted in a series of changes in its functional properties.

Physical and Chemical Properties

Solubility

As shown in Fig. S2, the lowest solubility of samples was observed around isoelectric point, which may be attributed to their structural similarity. Meanwhile, compared with SPI, the conjugates exhibited better solubility in the whole pH range, particularly at pH values (6–9). Among them, SPI: β G 1:1 exhibited the best solubility. The hydrophilic groups, mainly hydroxyl group of β G were introduced and adsorbed on the surface of SPI, improving the affinity of SPI to water, thus increasing the interaction between water molecules and SPI [18]. Research showed that the solubility of protein-polysaccharide conjugates can be controlled by the degree of MR [19]. In our work, the more β G was introduced, the higher the DG and the better solubility. Acacia gum introduced hydrophilic groups into soy protein isolates,

enhancing the interaction between protein and water molecules, increasing the solubility of proteins [20].

Surface hydrophobicity (H_0)

H_0 is one of the important factors that affect the solubility and emulsifying property of protein. As shown in Table S2, SPI exhibited the lowest H_0 , which was caused by the fact that majority of its hydrophobic residues was buried within the dense spherical zone [21]. MR endowed SPI with enhanced H_0 , which increased with the increase of β G content in reaction. The result may be caused by the increased solubility of SPI bonded with polar groups, exposing more hydrophobic groups. Meanwhile, the red-shift of λ_{\max} in SPI after MR suggested that SPI has been unfolded by MR [17]. The hydrophobic groups were exposed by unfolded SPI, leading to SPI's increased surface hydrophobicity. Research showed that H_0 of peanut isolate was also significantly enhanced after MR with dextran or gum arabic [22].

Rheological Property

Proteins' rheological properties in food are tightly associated with its functional characteristics, which could directly affect the food quality and shelf-life [23]. The viscosity of SPI hardly changed within the range of shear rates of 0.1–100 s^{-1} , approaching an ideal Newtonian fluid (Fig. S3). However, the SPI conjugates showed shear thinning behavior of a pseudoplastic fluid, which was a result of the disrupting of irregularly curled polymer and its parallel alignment during shear [24]. Compared to SPI, SPI conjugates exhibited higher viscosity, which further affected the relevant characteristics.

Physiological Activities

The excellent physiological activity of β G is related to its viscosity, and since MR enhanced the viscosity of SPI, can it be expected that oat β -glucan will enhance or even confer SPI-related activity? Bile acid binding ability (BAB), fat binding capacity (FB), and glucose availability (GA) of SPI conjugates were shown in Table S2. The BAB of proteins has been associated with its cholesterol lowering. The BAB of SPI was 10.76%, and increased as the β G content involved in MR increased, and SPI: β G 1:1 showed a significant increase in BAB (47.70%). Whey protein isolate bounded to the bile salts through hydrogen bond interactions, and the increase in surface hydrophobicity indicated higher bile acid binding through hydrophobic interactions [25]. β G introduced hydrophilic groups to SPI during MR, enhancing the solubility and hydrophobicity (H_0), which will result in the increased BAB of SPI. The MR product

of glucose and L-lysine could adsorb bile acids and reduce plasma cholesterol [26].

The fat binding capacity of chitosan is closely related to its lipid-lowering effect [27]. SPI conjugates' FB was improved by MR and increased as the β G content increased. Research revealed that hydroxyl group showed a positive effect on the fat binding capacity of macromolecules [28]. Meanwhile, the SPI helix structure was unfolded, exposing more hydrophobic groups and leading to the increased H_0 [29]. The exposed hydrophobic regions on the protein surface can effectively interact with the hydrophobic portions of the fats, leading to improved FB [30].

In vitro digestion tests, the glucose availability was 20.34 (SPI), 12.58 (SPI: β G 4:1), 11.76 (SPI: β G 2:1) and 11.20 (SPI: β G 1:1) mmol/L, respectively. The result indicated that MR endowed SPI with enhanced hypoglycemic ability, and the ability increased as the β G content involved in MR increased. Generally, the enhancement of intestinal viscosity and reduction of pancreatic amylase activity can lower the digestibility and absorption of starch [31]. In the present study, the increase in hypoglycemic ability might be associated with the enhanced viscosity of SPI conjugates.

Carbohydrate digestion and uptake could be significantly reduced by the inhibition of α -glucosidase activity (I α G), which result in a lower blood glucose levels and an inhibition of postprandial hyperglycemia [32]. The I α G of SPI conjugates was depicted in Table S2. SPI barely demonstrated α -glucosidase activity, but the activity of SPI conjugates increased remarkably, among them SPI: β G 1:1 showed the best I α G (35.45%). Research showed that the I α G of chitosan was enhanced by MR with glucose [33], indicating MR was a resultful approach to increase SPI's hypoglycemic effect.

Crossing with β -glucan increased the length of SPI's molecular chain and thus increased the viscosity of SPI conjugates. This may be one of the reasons why viscosity-related properties such as bile acid binding ability, fat binding capacity, glucose availability and the inhibition of α -glucosidase activity were enhanced.

Conclusion

The content of β -glucan showed a great effect on the structure and function of SPI during the Maillard reaction. As β -glucan content increased, the degree of glycation of the Maillard reaction was elevated, and the water solubility and H_0 of the SPI conjugates increased. The increase in DG meant that more β -glucan was bonded to SPI, which led to a higher viscosity of SPI conjugates and consequently a significant increase in viscosity-related physiological activities, especially, hypoglycemic activity. All the functional

properties of SPI were positively correlated with the content of β G. Our result provided some theoretical information for the effective utilization of SPI.

Supplementary Information The online version contains supplementary material available at <https://doi.org/10.1007/s11130-023-01092-4>.

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Data Availability Data will be made available on request.

Declarations

Conflict of Interest The authors declare that there are no conflicts of interest or personal relationships that could have appeared to influence the work reported in this paper.

Ethical Approval This study does not involve any human or animal testing.

Consent to Participate All authors have read and approved the MS; and, that all are aware of its submission.

Consent for Publication All authors agree to publish.

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