

Biochemical and biophysical CO₂ concentrating mechanisms in two species of freshwater macrophyte within the genus *Ottelia* (Hydrocharitaceae)

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Abstract Two freshwater macrophytes, *Ottelia alismoides* and *O. acuminata*, were grown at low (mean 5 $\mu\text{mol L}^{-1}$) and high (mean 400 $\mu\text{mol L}^{-1}$) CO₂ concentrations under natural conditions. The ratio of PEPC to RuBisCO activity was 1.8 in *O. acuminata* in both treatments. In *O. alismoides*, this ratio was 2.8 and 5.9 when grown at high and low CO₂, respectively, as a result of a twofold increase in PEPC activity. The activity of PPK was similar to, and changed with, PEPC (1.9-fold change). The activity of the decarboxylating NADP-malic enzyme (ME) was very low in both species, while NAD-ME activity was high and increased with PEPC activity in *O. alismoides*. These results suggest that *O. alismoides* might

perform a type of C₄ metabolism with NAD-ME decarboxylation, despite lacking Kranz anatomy. The C₄-activity was still present at high CO₂ suggesting that it could be constitutive. *O. alismoides* at low CO₂ showed diel acidity variation of up to 34 $\mu\text{equiv g}^{-1}$ FW indicating that it may also operate a form of crassulacean acid metabolism (CAM). pH-drift experiments showed that both species were able to use bicarbonate. In *O. acuminata*, the kinetics of carbon uptake were altered by CO₂ growth conditions, unlike in *O. alismoides*. Thus, the two species appear to regulate their carbon concentrating mechanisms differently in response to changing CO₂. *O. alismoides* is potentially using three different concentrating mechanisms. The Hydrocharitaceae have many species with evidence for C₄, CAM or some other metabolism involving organic acids, and are worthy of further study.

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· Organic acids · Photosynthesis

Abbreviations

AMP	Adenosine monophosphate
ATP	Adenosine triphosphate
Alk	Alkalinity
CAM	Crassulacean acid metabolism
CCM	Carbon dioxide concentrating mechanism
DIC	Dissolved inorganic carbon
DTT	Dithiothreitol
FW	Fresh weight
GAPDH	Glyceraldehyde 3-phosphate dehydrogenase
LDH	Lactate dehydrogenase
MDH	Malate dehydrogenase
NAD(P)-ME	NAD(P)-malic enzyme
OAA	Oxaloacetate

PEP	Phosphoenol pyruvate
PEPC	PEP carboxylase
PEPCK	PEP carboxykinase
PGK	Phosphoglycerate kinase
PPDK	Pyruvate phosphate dikinase
RuBisCO	Ribulose 1,5-bisphosphate carboxylase–oxygenase
RuBP	Ribulose 1,5-bisphosphate

Introduction

All eukaryotic photoautotrophs, plus their cyanobacterial predecessor, assimilate CO₂ via the Calvin Benson-Bassham or reductive pentose phosphate cycle where the carboxylation reaction is catalyzed by ribulose 1,5-bisphosphate carboxylase–oxygenase [RuBisCO; EC 4.1.1.39; (Raven et al. 2012)]. However, RuBisCO, has a relatively low affinity for CO₂ and will also fix oxygen competitively leading to subsequent further carbon loss via photorespiration (Bowes et al. 1971; Ogren 2003). Some terrestrial plants have a biochemical carbon dioxide concentrating mechanism (CCM), whereby carbon is fixed by the oxygen insensitive phosphoenol pyruvate carboxylase (PEPC; EC 4.1.1.31) producing a C₄ acid, oxaloacetate (OAA), that is rapidly converted into malate or aspartate. The C₄ acid is then decarboxylated, to produce elevated concentrations of CO₂ around RuBisCO, minimising photorespiration (Hatch and Slack 1966; Raghavendra and Sage 2011). Most terrestrial C₄ plants have ‘Kranz anatomy’, where the initial carboxylation by PEPC is spatially separated from subsequent decarboxylation and RuBisCO fixation, in order to prevent futile cycling (Raghavendra and Sage 2011). However, a dramatic variant of C₄ plant was discovered in a submersed monocot, *Hydrilla verticillata* (Hydrocharitaceae; Holaday and Bowes 1980; Bowes 2011) that operates an inducible single-celled C₄ metabolism with CO₂ concentrating in the chloroplast. About a decade ago, C₄ metabolism was also described in single cells of two land plants (Chenopodiaceae), *Bienertia cycloptera* and *Borszczowia aralocapsica* with RuBisCO and PEPC in different parts of the same cell (Edwards et al. 2004; Voznesenskaya et al. 2001, 2002).

Three major sub-types of C₄ plants have been described based on the decarboxylation step that liberates CO₂ from the C₄ acid compounds. In two sub-types, malate is decarboxylated to form CO₂ and pyruvate, one with NADP-malic enzyme (NADP-ME, EC 1.1.1.40) and one with NAD-malic enzyme (NAD-ME, EC 1.1.1.39). The pyruvate re-enters the cycle via pyruvate phosphate dikinase (PPDK EC 2.7.9.1) that regenerates phosphoenolpyruvate, the substrate for PEPC. In a third sub-type, PEP carboxykinase (PEPCK, EC 4.1.1.49) decarboxylates OAA to form CO₂ and PEP. In the plants with NAD-ME or PEPc kinase, aspartate rather than

malate is shuttled from the mesophyll to the bundle-sheath cells (Raghavendra and Sage 2011).

Other terrestrial plants, especially those associated with arid environments, possess crassulacean acid metabolism (CAM) where there is a temporal separation of carbon fixation by PEPC and RuBisCO. In these plants, PEPC is active at night causing malate to accumulate within the vacuole. This C₄ acid is then decarboxylated during the day to produce CO₂ that is fixed by RuBisCO. This is primarily a water-conserving mechanism minimising gaseous exchange during the day, but it also serves to conserve carbon by reducing respiratory carbon loss (Cushman and Bohnert 1999; Silvera et al. 2010).

Concentrations of CO₂ in lakes frequently exceed air equilibrium as a result of input from the catchment of CO₂ or terrestrially fixed organic carbon that is oxidised to CO₂ (Cole et al. 2007; Maberly et al. 2013). However, in productive systems the rate of carbon fixation in a unit volume of water can greatly exceed rates of carbon supply from the atmosphere, or other sources, leading to depletion of CO₂ virtually to zero (Maberly 1996) limiting productivity (Ibelings and Maberly 1998; Jansson et al. 2012). Furthermore, the rate of CO₂ diffusion in water is about 10⁴-times lower in water than in air (Raven 1970) leading to substantial transport limitation through the boundary layer surrounding objects in water (Black et al. 1981). As a consequence, the concentration of CO₂ needed to half saturate the net photosynthesis of freshwater macrophytes is roughly 8–14 times air equilibrium (Maberly and Madsen 1998).

Freshwater macrophytes have a range of avoidance, amelioration or exploitation strategies to overcome the problem of limited inorganic carbon supply (Klavnsen et al. 2011). The most frequent CCM in freshwater macrophytes is based on the biophysical use of bicarbonate (Maberly and Madsen 2002). Bicarbonate is the most abundant form of inorganic carbon in all freshwaters where the pH is between about 6.3 and 10.1: the first and second dissociation constants of the carbonate system. Even when concentrations of CO₂ are strongly depleted as a result of photosynthetic carbon uptake, concentrations of bicarbonate can still be substantial. Freshwater concentrations of bicarbonate range from zero in acid systems to over 100 mmol L⁻¹ in soda lakes (Talling 1985). The use of bicarbonate, like other CCMs, is an active process requiring the expenditure of energy and may involve ‘polar leaves’ with localised areas of proton extrusion leading to conversion of bicarbonate to CO₂ and subsequent inward diffusion, or direct uptake of bicarbonate (Elzenga and Prins 1987).

Although much less widespread, some freshwater macrophytes also possess a type of C₄ metabolism. The best studied is that of the dioecious form of the hydrocharitaceae

H. verticillata (Bowes 2011; Holaday and Bowes 1980) that operates an inducible single-celled C_4 mechanism based on carbon fixation by PEPC and decarboxylation by NADP-ME, in addition to being able to use bicarbonate. Similar, albeit less well characterised, C_4 mechanisms appear to operate in other monocotyledons: *Egeria densa* (Hydrocharitaceae; (Browse et al. 1977; Casati et al. 2000), and in amphibious species *Eleocharis vivipara* (Cyperaceae) (Ueno et al. 1988) and *Orcuttia viscidia*, *Neostapfia colusana* and *Tuctoria greenii* [Poaceae; Keeley and Sandquist 1992]).

A number of freshwater macrophytes have also been shown to possess CAM (Keeley 1981, 1998). These include species within the genus *Isoetes* (Lycopodiophyta), and the angiosperms *Littorella uniflora* (Madsen 1987a, b), *Crassula helmsii* (Newman and Raven 1995) (Klavnsen and Maberly 2009) and *Vallisneria spiralis* (Keeley 1998). Underwater, CAM acts as a carbon-conserving mechanism that reduces the loss of respiratory carbon at night and exploits the nocturnal concentrations of CO_2 that are often higher than during the day (Klavnsen et al. 2011).

In terrestrial plants, the global frequency of C_4 is about 3 % (Edwards et al. 2004) and that of CAM about 6 %, (Silvera et al. 2010) with the remainder (91 %) being C_3 and so lacking CCMs. In contrast, about 55 % of aquatic angiosperms has a biophysical CCM based on HCO_3^- use and others have a biochemical CCM based on CAM (4 %) or C_4 (3 %; Maberly and Madsen 2002).

The Hydrocharitaceae contains a number of species with biochemical CCMs, including the C_4 syndrome (e.g. *H. verticillata*, *E. densa*) or CAM activity (e.g. *V. spiralis*), but many ecologically important species within this family have not been studied. One species-rich genus within the Hydrocharitaceae that has been little studied is *Ottelia* Pers. Here we characterized the CCMs of two species from China, *Ottelia acuminata*, (Gagne.) Dandy var. *lunensis* H. Li and *O. alismoides* (Linn.) Pers., and tested their ability to acclimate to different concentrations of CO_2 .

Materials and methods

Plant material and growth conditions

Ottelia acuminata and *O. alismoides* were both collected from Yunnan Province, China and then cultivated in a greenhouse in Wuhan Botanical Garden for several years. Seeds were germinated in a growth chamber (temperature 25 °C), and when the seedlings reached about 20 cm tall, they were transferred to 10 cm diameter plant pots containing sediment from nearby Donghu Lake and placed inside glass tanks (30 × 40 × 60 cm tall) containing about 65 L of tap water with an alkalinity of about 2 mequiv L^{-1} . The glass tanks were located in a

glasshouse on the flat roof of the laboratory in larger tanks (about 400 L) of running water to reduce diurnal changes in water temperature. The experiment was started on 11 July 2012 and finished on 29 September 2012. During the experimental period, snails and moribund leaves were removed every day. Two or three times each day, water temperature was recorded and a water sample collected to measure pH with a combination pH electrode (Metrohm 6.0238.000, Herisau, Switzerland) connected to a meter (Metrohm 718 STAT Titrino).

Two treatments were produced with four replicate tanks per treatment, each containing both species. In the ‘low CO_2 ’ treatment, the natural photosynthetic activity of the plants was allowed to deplete the inorganic carbon concentration of the water, and increase the pH. In the ‘high CO_2 ’ treatment, tank water saturated with CO_2 was added to the tanks two to three times each day to reduce the pH between 6.6 and 7.0, and thereby, increase the concentration of CO_2 . The tanks were gently stirred to mix the water after each addition of CO_2 solution. Both treatments were out of equilibrium with air CO_2 , and although CO_2 concentrations varied over time, the concentrations in the two treatments were very different.

Enzyme activity measurements

Leaves were harvested, blotted dry and quickly weighed to determine fresh weight (FW), and then frozen in a pestle and mortar with liquid nitrogen. Typically, about 0.6 g FW of leaf was extracted, and 3 mL of ice-cold extraction buffer was added for each gram FW of leaf. The extraction buffer comprised 50 mM Tris, 0.1 mM EDTA, 15 mM $MgCl_2$ and pH 8 (buffer A) plus 10 % glycerol. Following grinding to a smooth paste, the whole extract was centrifuged at 5 °C for 45 min at 12,000×g (Heraeus, Biofuge Fresco, Germany). The supernatant (the crude extract) was stored on ice prior to measuring enzyme activity.

RuBisCO activity was measured in crude extracts by coupling its activity to NADH oxidation using phosphoglycerate kinase from yeast (PGK; Sigma St Louis, MO, USA) and glyceraldehyde-3-phosphate dehydrogenase from rabbit muscle (GAPDH; Sigma). Prior to measuring activity, the extract was incubated in buffer A in the presence of 20 mM bicarbonate for 5 min. Activity was then followed using buffer A with 0.2 mM NADH (Sigma), 1 mM ATP (Sigma), 5 mM DTT (Shanghai Chemical Reagents Company, China), 5 units of PGK and 5 units of GAPDH and 1 mM ribulose 1,5-bisphosphate (Sigma). The disappearance of NADH was followed at 340 nm using a UV–Vis spectrophotometer (TU-1810PC, Purkinje General, China). The calculated carboxylase activity took account of the fact that two molecules of NADH are oxidized for every molecule of RuBP catalyzed.

PEPC activity was measured using buffer A, with 20 mM bicarbonate and 1 mM phosphoenol pyruvate (Sigma) to produce oxaloacetate that is in turn coupled to malate dehydrogenase (MDH, Sigma) activity using an excess of this enzyme. The reaction mixture, therefore, also contained 0.2 mM NADH and 5 units of MDH. Activity was continuously followed by recording a decrease of absorbance at 340 nm.

NAD-ME activity was measured spectrophotometrically in buffer A supplemented with 1 mM NAD (Biosharp, Japan), 10 mM malate (Energy Chemical, Shanghai, China), 1 mM MnCl_2 and 5 mM dithiothreitol. NADP-ME activity in crude extracts was measured spectrophotometrically in buffer A containing 1.5 mM NADP (Sigma), 10 mM malate, 1 mM MnCl_2 and 5 mM dithiothreitol. Activities of NAD-ME and NADP-ME were followed continuously by recording an increase of absorbance at 340 nm.

PPDK activity was measured spectrophotometrically, at 340 nm, in the opposite direction to the one operating in C_4 photosynthesis, by following pyruvate formation and NADH disappearance using lactate dehydrogenase (LDH, Amresco, Biochemicals and Life Science Research Products). The reaction was carried out in buffer A supplemented with 5 mM PEP, 1.2 mM AMP (Sigma), 1 mM pyrophosphate, 2.5 mM dithiothreitol, 0.2 mM NADH and 2 units of LDH.

All activities were maximal activities for the studied growth conditions, but are not *in vivo* activities. All activities were measured at 25 °C.

CAM activity

The daily change in titratable acidity was calculated as the difference between the minimum and maximum amount of titratable acidity per unit fresh mass. The minimum amount of acid was measured on plants collected at the end of the pH-drift experiment (July) or collected towards the end of the light period on 14–16 August and 27–29 September 2012. The maximum amount of acid was assayed after incubation of material in the dark at 25 °C for 18 h in 1 mmol L^{-1} equimolar NaHCO_3 and KHCO_3 at a concentration of CO_2 of about 700 $\mu\text{mol L}^{-1}$ (pH about 6.4). About 0.2 g fresh mass of material was quickly blotted, carefully weighed, roughly chopped into 10 mL plastic stoppered tubes and frozen at –20 °C. Prior to analysis, 5 mL of deionised water was added, and the tubes were boiled for 15 min, cooled and stored in a refrigerator. Titratable acidity was assayed on an aliquot from each tube by end point titration to pH 8.3 using $\sim 0.01 \text{ mol L}^{-1}$ NaOH, standardized by Gran titration against 0.1 mol L^{-1} HCl. Measurements were made in triplicate, and results are expressed as $\mu\text{equiv g}^{-1}$ FW.

pH-drift

The ability of leaves to use bicarbonate was assessed in pH-drift experiments (Maberly and Spence 1983). Leaves were cleaned by gentle rubbing to remove the marl deposit from their upper surface. They were then rinsed for about 10 min in one of the two test media: equimolar concentration of NaHCO_3 and KHCO_3 at total HCO_3^- concentrations of 0.1 or 1.0 mmol L^{-1} . The leaves were placed in 30 mL test tubes with ground glass stoppers containing 25 mL of solution and about 5 mL air. The tubes were incubated in a growth cabinet at a constant temperature of 25 °C and receiving about 75 $\mu\text{mol photon m}^{-2} \text{ s}^{-1}$ (photosynthetically available radiation) from fluorescent tubes, measured with a cosine corrected sensor (Li-Cor LI-192SA). The pH was measured after 24 h and roughly every 6–12 h thereafter until a maximum pH was reached. The final alkalinity in the solution was measured by Gran titration with a standard solution of HCl.

Kinetics of O_2 evolution

Rates of net photosynthesis were measured as O_2 evolution at 25 °C at a photon irradiance of 120 $\mu\text{mol photon m}^{-2} \text{ s}^{-1}$. Leaves (0.2–0.5 g FW) were collected from the growth tanks, and cleaned by gentle rubbing to remove the marl deposit from their upper surface. They were then rinsed for about 10 min in a solution of 20 mM Tricine, pH 7. The leaves were then placed in a glass and Perspex chamber, the volume of which was 120 mL, in Tricine buffer bubbled briefly with nitrogen (starting O_2 concentration 60–70 % air saturation). The chamber was sealed and the O_2 concentration measured with an optical oxygen electrode (YSI Pro ODO Yellow Spring Instruments, USA) calibrated in air at 100 % humidity and 25 °C. Incremental small volumes (6–90 μL) of 2 mol L^{-1} Na/ KHCO_3 stock were added to generate a range of inorganic carbon concentrations from 0.1 to 3.8 mmol L^{-1} . The output of the electrode was logged on a computer and linear regressions of concentration against time were used to calculate rates of oxygen exchange. The kinetic response was fitted to the Michaelis–Menten equation.

Soluble protein, chlorophyll and leaf area

The soluble protein concentration of crude extracts was assayed using the Bio-Rad (Hercules, CA, USA) reagent using bovine serum albumin as a standard (Bradford 1976). The content of chlorophyll *a* and *b* in the leaves of *Ottelia* was determined on 0.1–0.5 g fresh leaf material ($n = 3–6$). Chlorophyll was extracted overnight at 4 °C with 95 % ethanol, and chlorophyll concentration was calculated from absorbance measured in a spectrophotometer (TU-1810PC)

using the equations of (Brain and Solomon 2007). Projected (1-sided) leaf area was calculated from digital photographs using AreaAna software (Huazhong University of Sciences and Technology, China).

Results

Growth conditions

The temperature was identical (around 29 °C) in the low and high CO₂ treatments and relatively constant (Table 1). The pH in the low CO₂ treatment was more than one pH unit greater than in the high CO₂ treatment. Precipitation of calcium carbonate on the leaves of both species of *Ottelia* in the low CO₂ treatment caused the alkalinity to be on average nearly 1 mequiv L⁻¹ lower than in the high CO₂ treatment. The bicarbonate concentration in the low CO₂ treatment was consequently also lower than that in the high CO₂ treatment. The CO₂ concentration was 80-fold lower in the low vs the high CO₂ treatment.

Soluble protein, chlorophyll and leaf area

Growth in low or high CO₂ did not have a statistically significant effect on the soluble protein, chlorophyll and leaf area of *O. alismoides*, although the ratio of chlorophyll *a* to chlorophyll *b* was slightly higher at high vs low CO₂ ($p < 0.05$; Table 2). For *O. acuminata*, the chlorophyll content per unit FW was 1.7-fold higher in leaves grown at low CO₂ compared to leaves grown at high CO₂ ($p < 0.01$; Table 2).

Enzyme activities

The activity of RuBisCO on a protein basis was similar in both species and did not vary with the CO₂ growth conditions (Fig. 1a). In *O. alismoides*, PEPC activity was twofold higher in low CO₂ compared to high CO₂ leaves (Student's *t* test, $p < 0.001$), but was constant in *O. acuminata* (Fig. 1b). Consequently the ratio of PEPC to

RuBisCO activity increased significantly from 2.8 to 5.9 in *O. alismoides* (Student's *t* test, $p < 0.001$), while it remained constant at about 1.8 in *O. acuminata* (Fig. 1c).

Pyruvate phosphate dikinase (PPDK), a key enzyme in two of the three decarboxylation types of C₄, showed a similar pattern of change to PEPC (Fig. 1d). The CO₂ concentration during growth did not affect PPDK activity in *O. acuminata*, but triggered a 1.9-fold increase in *O. alismoides* at low compared to high CO₂ that was highly significant (Student's *t* test, $p < 0.001$). There was a significant correlation between activity of PEPC and PPDK (Fig. 2a). The activity of the widespread decarboxylating enzyme NADP-ME was very low in both species. The activity of NADP-ME at low and high CO₂ concentration during growth did not change in *O. alismoides*, but decreased at low CO₂ in *O. acuminata* (Student's *t* test, $p < 0.05$; Fig. 1e). Activity of NADP-ME did not correlate with changes in activity of PEPC (Fig. 2b). In contrast to NADP-ME, activities of NAD-ME (Fig. 1f) were very high and up to 27-fold greater than the activity of PEPC. In *O. acuminata* NAD-ME activity was slightly, but significantly, greater in the low CO₂-grown compared to high CO₂-grown leaves (Student's *t* test, $p < 0.05$). The pattern in *O. alismoides* was similar, but difference between high and low CO₂ treatments was not significant (Student's *t* test, $p = 0.08$). The activity of NAD-ME increased with that of PEPC in *O. alismoides* (Fig. 2c). We were not able to measure PEPC spectrophotometrically as malate dehydrogenase was present in the crude extract and would interfere with the assay.

CAM capacity

The CAM capacity of the two species was assessed initially by measuring diel change in acidity. Across the two species and growth conditions for CO₂, acidity levels varied between 14 and 24 μequiv g⁻¹ FW in the light and between 22 and 44 μequiv g⁻¹ FW in the dark (Fig. 3). There was a statistically significant difference between light and dark acidity levels in *O. alismoides* at low CO₂ of about 24 μequiv g⁻¹ FW (Student's *t* test, $p < 0.05$; Fig. 3). *O. alismoides* at high CO₂ showed a small diel change in acidity that was not statistically significant, but there was no evidence for diel acidity variation in *O. acuminata* in either condition.

In *O. alismoides*, the capacity to undertake CAM was re-measured in August and September. In August, a similar pattern was obtained and in leaves grown at low CO₂ there was a statistically significant (Student's *t* test, $p < 0.01$) diel change in acidity of 34 μequiv g⁻¹ FW (Fig. 3b), slightly greater than in July. However, in September there was no indication of a diel acidity change in either CO₂ treatment (Fig. 3c). Even in the absence of a diel change in

Table 1 Conditions in the two growth treatments

Conditions	Low CO ₂	High CO ₂
Temperature (°C)	29 (27–31)	29 (27–31)
pH ^a	8.27 (7.43–9.19)	6.99 (6.71–7.37)
Alkalinity (mequiv L ⁻¹)	1.21 (0.82–1.74)	2.08 (1.94–2.23)
CO ₂ (μmol L ⁻¹)	5 (0.1–19)	401 (156–748)
HCO ₃ (mmol L ⁻¹)	1.05 (0.37–1.69)	1.98 (1.66–2.23)

Mean values are given with ranges in parentheses

^a Calculated as a geometric mean

Table 2 Characteristics of the *Ottelia* species grown at low and high CO₂ concentration

Species	Chla + b/FW (mg g ⁻¹)		Chla:Chlb		Protein/FW (mg g ⁻¹)		Specific leaf area (1-sided cm ² g ⁻¹ FW)	
	Low	High	Low	High	Low	High	Low	High
<i>O. acuminata</i>	1.26 (0.14)	0.74 (0.12)	2.74 (0.19)	3.03 (0.38)	2.11 (0.10)	1.97 (0.10)	95.5 (26.2)	72.2 (16.2)
<i>O. alismoides</i>	0.95 (0.32)	0.65 (0.36)	2.75 (0.08)	2.90 (0.08)	1.62 (0.22)	1.34 (0.21)	100.5 (7.9)	85.5 (22.1)

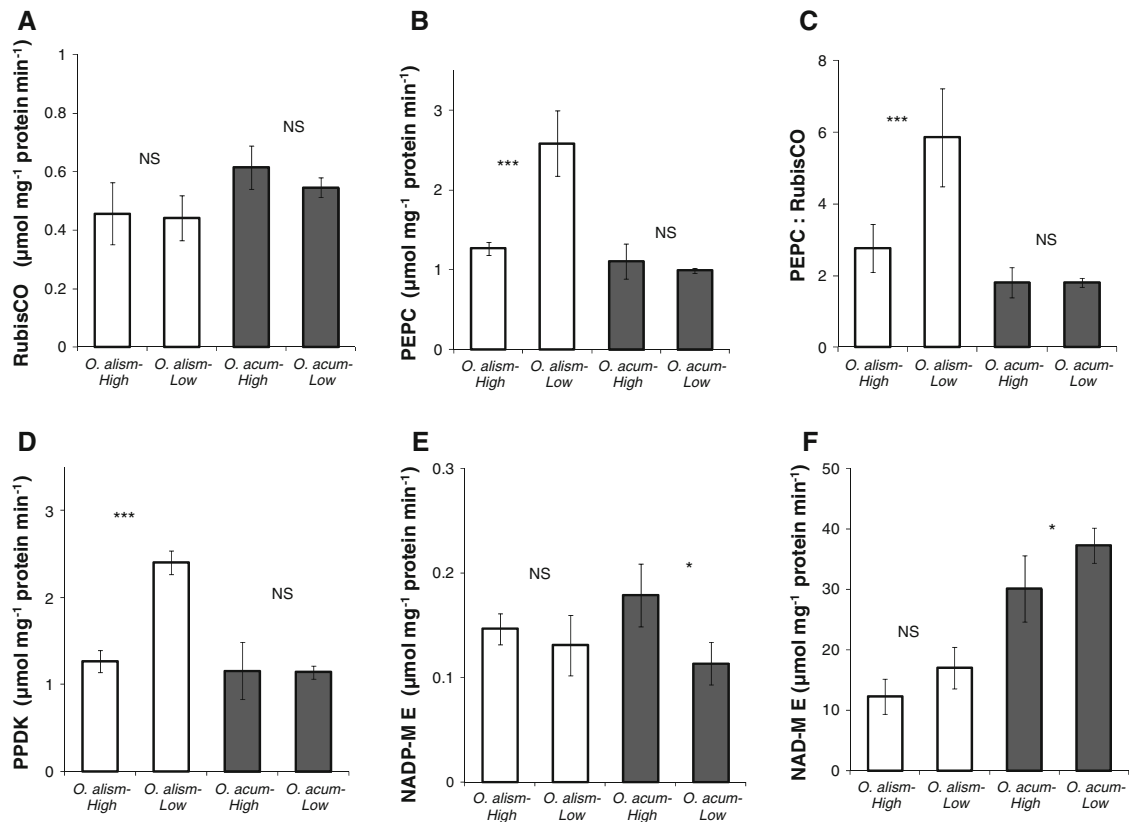


Fig. 1 Comparison of the activity of C₃ and C₄ metabolic enzymes in crude extracts from *O. alismoides* (light grey bars) and *O. acuminata* (dark grey bars) grown under low and high CO₂ concentrations for RuBisCO: ribulose 1,5-bisphosphate carboxylase-oxygenase. **a** PEPC: PEP carboxylase. **b** Ratio of PEPC to RuBisCO activity.

c PPK: pyruvate phosphate dikinase. **d** NADP-ME: NADP-malic enzyme. **e** NAD-ME: NADP-malic enzyme. **f** Means and standard deviation are presented. Statistical differences between high and low CO₂ treated plants are designated as follows: NS not significant. **p* < 0.05, ***p* < 0.01, ****p* < 0.001

acidity, there was a substantial amount of acidity on all measuring occasions, 26–44 µequiv g⁻¹ FW, at the end of the dark period.

pH-drift

The pH-drift experiments provided clear evidence for bicarbonate use in both species. The final concentration of bicarbonate was relatively constant and low with values between 0.06 and 0.09 mmol L⁻¹ in the low CO₂ treatment and 0.06 and 0.11 mmol L⁻¹ in the high CO₂ treatment (Table 3). The final CO₂ concentration was very low, in the range of 3–26 nmol L⁻¹ and tended to be lower when measured at the higher concentration of bicarbonate which

is again consistent with use of bicarbonate and lower than would be expected from C₄ photosynthesis alone based on CO₂ uptake. For example, assuming a low C₄ CO₂ compensation point of 3 ppm in air, at 25 °C this would be equivalent to 100 nmol L⁻¹, roughly 4- to 30-times higher than the final CO₂ concentrations in the drift experiments.

There were small differences in the final CO₂ concentration between CO₂ treatments at the low bicarbonate test concentration, especially in *O. acuminata*, with lower final CO₂ concentrations in the low CO₂ treatment. There were substantial, but reproducible shifts in alkalinity despite rinsing the leaves several times in the test medium prior to the experiment. In the lower alkalinity experiment alkalinity increased, but in the higher alkalinity experiments,

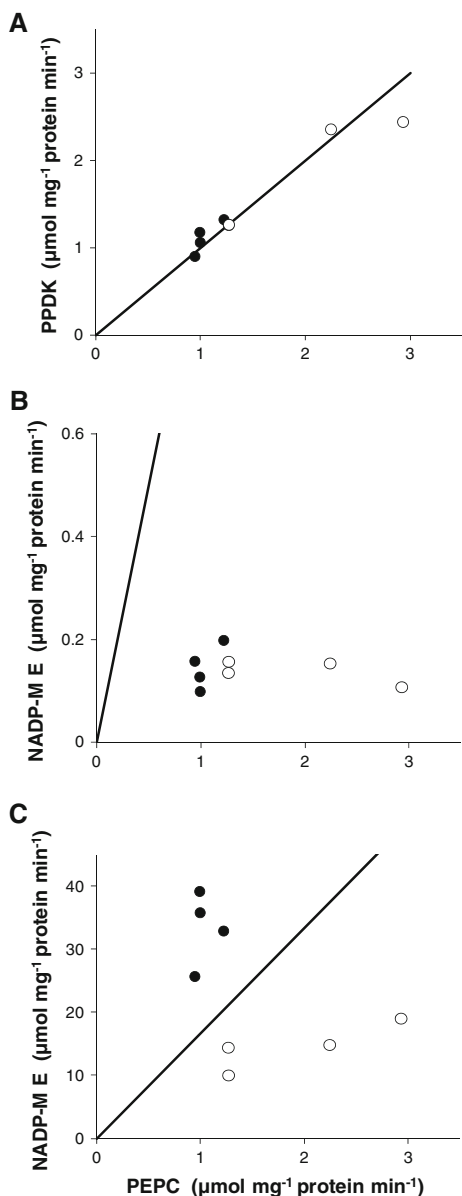


Fig. 2 Correlations between activities of **a** PPKDK, **b** NADP-ME and **c** NAD-ME and activity of PEPC for *O. alismoides* (open circle) and *O. acuminata* (closed circle) grown at low and high concentrations of CO₂. Each point represents the mean activity from one tank. 1:1 activity is represented by a solid line

alkalinity was unchanged in the presence of *O. acuminata*, but reduced in the presence of *O. alismoides*.

Kinetics of O₂ evolution

Oxygen exchange was measured as a function of dissolved inorganic carbon (DIC) concentration at pH 7 in both species and both treatments (Fig. 4). In *O. alismoides*, the kinetic responses of leaves from the low and high CO₂ treatments were not significantly different (variance ratio test; $F_{2,12} = 1.82, p = 0.20$). Using the combined data, the

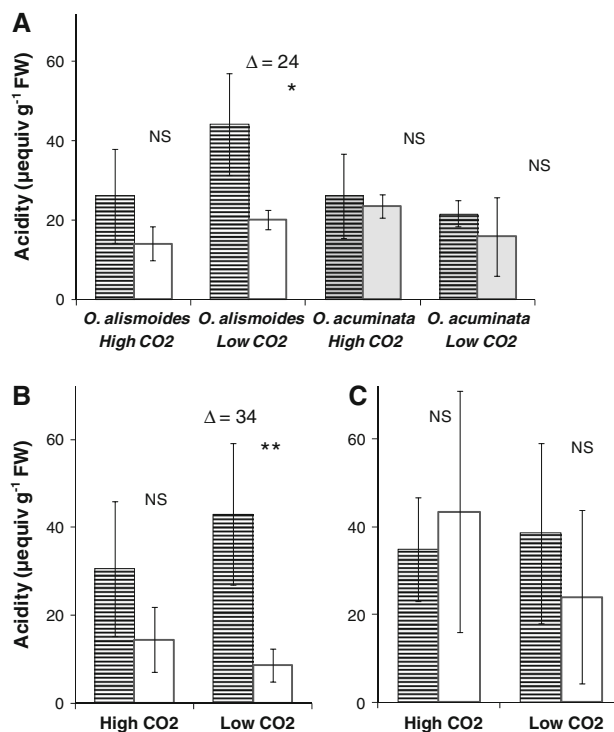


Fig. 3 Acidity of extracts from *O. alismoides* (white bars) and *O. acuminata* (grey bars) grown under low and high CO₂ concentrations, measured at the end of the dark period (hatched bars) and in the light (open bars) in July (a), August (b) and September (c). Means and standard deviation are presented. Statistical differences between high and low CO₂ treated plants are designated as follows: NS not significant, * $p < 0.05$. Where there is a significant difference between light and dark acidity, the difference is given as open triangle

maximum net rate of O₂ evolution was 27.2 (SD = 1.2) μmol O₂ g⁻¹ FW h⁻¹ which is equivalent to 56 and 39 μmol O₂ mg⁻¹ Chla h⁻¹ at high and low CO₂, respectively. The K_{1/2} for DIC was 1.29 (SD = 0.15) mmol L⁻¹ which at pH 7 is equivalent to 0.199 mmol L⁻¹ CO₂ and 1.090 mmol L⁻¹ HCO₃⁻. In *O. acuminata*, the kinetic responses of leaves from the two treatments were significantly different (variance ratio test; $F_{2,12} = 5.00, p < 0.05$). At the low CO₂ treatment, the maximum rate of O₂ evolution was 44.0 (SD = 3.9) μmol O₂ g⁻¹ FW h⁻¹ (48 μmol O₂ mg⁻¹ Chla h⁻¹), and the K_{1/2} for DIC was 1.64 (SD = 0.34) mmol L⁻¹. At the high CO₂ growth treatment, the maximum rate of O₂ evolution was 37.8 (SD = 4.9) μmol O₂ g⁻¹ FW h⁻¹ (69 μmol O₂ mg⁻¹ Chla h⁻¹), and the K_{1/2} for DIC was 2.36 (SD = 0.62) mmol L⁻¹. At pH 7, the K_{1/2} at low and high CO₂ growth treatments are equivalent to 0.253 and 0.363 mmol L⁻¹, respectively, for CO₂ and 1.386 and 1.994 mmol L⁻¹, respectively, for HCO₃⁻. The maximum rate of O₂ evolution was greater in *O. acuminata* than in *O. alismoides*, but the values of K_{1/2} were between 1.27- and 1.82-fold greater in *O. acuminata* than in *O. alismoides*. If the maximum rate of O₂ evolution is expressed on a protein

Table 3 Carbon concentrations at the end of the pH drift experiments

Species	[CO ₂] treatment	Starting Alk (mM)	Final Alk (mM)	Max pH	C _T (mM)	CO ₂ (nM)	HCO ₃ ⁻ (mM)	C _T /Alk
<i>O. acuminata</i>	Low	0.1	0.36 (0.13)	10.21 (0.13)	0.117 (0.056)	7.6 (2.1)	0.060 (0.024)	0.32 (0.05)
	Low	1.0	1.04 (0.01)	10.72 (0.01)	0.278 (0.011)	3.0 (0.3)	0.074 (0.004)	0.27 (0.01)
	High	0.1	0.17 (0.11)	9.64 (0.30)	0.084 (0.042)	28.9 (17.8)	0.062 (0.024)	0.53 (0.11)
	High	1.0	1.06 (0.06)	10.68 (0.02)	0.322 (0.025)	4.1 (0.4)	0.091 (0.007)	0.30 (0.01)
<i>O. alismoides</i>	Low	0.1	0.31 (0.13)	10.01 (0.28)	0.119 (0.047)	15.8 (13.0)	0.071 (0.035)	0.4 (0.13)
	Low	1.0	0.84 (0.06)	10.64 (0.04)	0.224 (0.018)	3.4 (0.60)	0.068 (0.007)	0.27 (0.02)
	High	0.1	0.37 (0.18)	10.01 (0.21)	0.175 (0.121)	26.4 (30.6)	0.111 (0.089)	0.45 (0.15)
	High	1.0	0.83 (0.01)	10.52 (0.06)	0.290 (0.035)	7.2 (2.5)	0.107 (0.022)	0.35 (0.04)

Mean values are given with SD in parentheses

basis, however, the rates in the two species are very similar.

Discussion

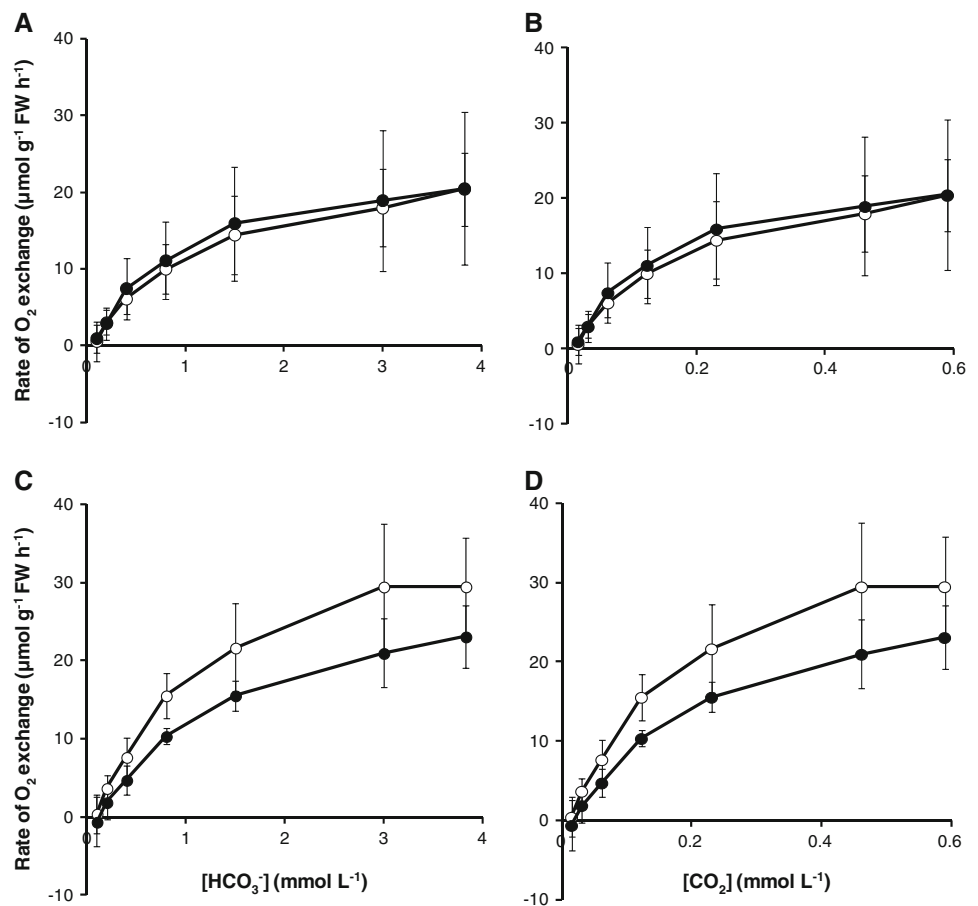
Comparison of carbon concentrating mechanisms in *Ottelia* with other aquatic and terrestrial plants

Ottelia acuminata and *O. alismoides* both have the carboxylating, PEP regenerating and decarboxylating enzymes needed to operate a C₄ pathway. In both species and under both growth treatments, the activity of PEPC was greater than that of RuBisCO and in plants adapted to low CO₂, PEPC:RuBisCO ratios were 5.9 and 1.8 for *O. alismoides* and *O. acuminata*, respectively. The ratio for *O. alismoides* is similar to those reported for *H. verticillata*, and the ratio for *O. acuminata* is identical to that of *E. densa* (Table 4) both of which are regarded as C₄ aquatic plants (Bowes 2011; Casati et al. 2000). The PEPC:RuBisCO ratio of *O. alismoides* is slightly lower than in some terrestrial C₄ plants, but very similar to the single-celled C₄ plants *B. aralocaspica* and *B. cycloptera* (Voznesenskaya et al. 2001, 2002) (Table 4). In contrast, terrestrial C₃ plants and aquatic plants lacking a biochemical concentrating mechanism have PEPC:RuBisCO ratios substantially less than 1 (Table 4). PPDK regenerates PEP, the substrate for PEPC. In the two species of *Ottelia*, activities of PPDK were equivalent to those of PEPC and so should be able to support PEPC activity. The high activities in *Ottelia* are similar to those in terrestrial C₄ plants (Table 4), although, the ratio of PPDK to PEPC in *Ottelia* is greater. Of the two potential decarboxylating enzymes, the activity of NAD-ME was 130 times greater than NADP-ME in *O. alismoides* and 340 times greater in *O. acuminata*. NAD-ME is a mitochondrial enzyme that can act as the decarboxylating enzyme in the terrestrial single-celled C₄ plants *B. aralocaspica* and *B. cycloptera* (Voznesenskaya et al. 2001,

2002) (Table 4). The apparent decarboxylation by NAD-ME in *Ottelia*, if confirmed, would be the first report of an aquatic plant belonging to the NAD-ME C₄-subtype. Casati et al. (Casati et al. 2000) assumed that NADP-ME was the decarboxylation pathway in *E. densa* as the activity of this enzyme increased on transfer to low CO₂ conditions. However, NAD-ME activity was not measured, and the activity of NADP-ME was about half that of PEPC so, it is not impossible that NAD-ME is also involved in decarboxylation in this species. *H. verticillata* is also assumed to belong to the NADP-ME sub-group (Bowes 2011; Bowes et al. 2002), but much more evidence is available to support this contention, since physiological characteristics changed in parallel to NADP-ME activity, and oxygen inhibition measurements are consistent with high concentrations of CO₂ being generated in the chloroplast where NADP-ME is located (Magnin et al. 1997; Reiskind et al. 1997). In *H. verticillata*, the ratio of NAD-ME to NADP-ME is about five, much less than that found in the two species of *Ottelia* studied here (Table 4). However, further work is needed to confirm that *Ottelia* is operating NAD-ME C₄ photosynthesis. It has been reported that versions of the enzymes used in the variants of C₄ photosynthesis can occur in C₃ plants. For example, in the C₃ *Arabidopsis* there are one or more isoforms of PEPC, PEPCK, NAD-ME, NADP-ME and PPDK which have different functions (Aubry et al. 2011). Analysis of enzyme activity might help to temper future claims of C₄ photosynthetic metabolism based solely on genomics or transcriptomics and detailed studies of biochemical turnover using short-term labelling with ¹⁴C-labelled inorganic carbon should be investigated in the future.

Preliminary examination of leaf sections for both species under the light microscope has shown no evidence for Kranz anatomy (data not shown) so it is possible that *Ottelia* is also operating a single-cell C₄ mechanism. However, unlike *Hydrilla* and *Egeria*, the leaves of both *Ottelia* species are four cells thick, so RuBisCO and PEPC could be localized in different types of cell.

Fig. 4 Rate of net photosynthesis as a function of the concentration of HCO_3^- (a, c) or CO_2 (b, d) in *O. alismoides* (a, b) and *O. acuminata* (c, d) grown under low (open circle) and high (closed circle) CO_2 concentrations



The C_4 system in *O. alismoides* and *O. acuminata* is not abolished at high CO_2 ($400 \mu\text{mol CO}_2 \text{ L}^{-1}$; over 30-fold air equilibrium) unlike in the two other well-studied C_4 freshwater macrophytes, *E. densa* and *H. verticillata*. The C_4 syndrome may be constitutive in *Ottelia* as it is in the marine macroalga *Udotea flabellum* (Reiskind and Bowes 1991; Reiskind et al. 1988) although the effect on *Ottelia* of other environmental factors such as low temperature or light has not been tested.

Both species of *Ottelia* studied here showed an ability to use bicarbonate as an exogenous carbon source based on the pH-drift experiments and also the rates of oxygen evolution as a function of inorganic carbon concentration, despite the use of a buffer to maintain constant pH. Bicarbonate use is a widespread feature in freshwater angiosperms (Maberly and Madsen 2002), however, its combination in a species able to show CAM has not been reported before as far as we are aware.

Distribution of biochemical CCMs in terrestrial and aquatic plants

Several lines of evidence show that in the terrestrial environment, C_4 photosynthesis became widespread around 11

to 5 million years ago during periods of hot and arid conditions and that it is polyphyletic and arose at least 62 times (Sage et al. 2011). C_3 – C_4 metabolism has been described in several species in the genera *Moricandia*, *Panicum* and *Flaveria*. C_3 – C_4 intermediates are important because they are viewed as possible evolutionary intermediates between the C_3 and C_4 photosynthetic pathways (Peisker 1986). In all known *Flaveria* C_3 – C_4 intermediates, both RuBisCO and PEP carboxylase are not entirely compartmentalized between mesophyll and bundle-sheath cells, as is observed in C_4 species (Moore et al. 1988; von Caemmerer 2000). Different *Flaveria* C_3 – C_4 intermediates fix between 15 and 85 % of atmospheric CO_2 into C_4 acids during short-term exposure to $^{14}\text{CO}_2$; however, transfer of label to the C_3 cycle does not occur at the rates normally observed in C_4 species (Monson et al. 1986). Our results showed that there was a statistically significant difference between light and dark acidity levels in *O. alismoides* at low CO_2 of about 24 – $34 \mu\text{equiv g}^{-1} \text{FW}$ in July and August. However, in September there was no indication of a diel acidity change in either CO_2 treatment. Even in the absence of a diel change in acidity, there was a substantial amount of acidity on all measuring occasions, at the end of the dark period. C_3 – C_4 intermediate photosynthesis could

Table 4 Comparison of activities of photosynthetic enzymes in aquatic and terrestrial plants

Environment/species	Type	RuBiCO	PEPC	PPDK	NADP-ME	NAD-ME	PEPC: RuBisCO	References
Aquatic								
14 species	C ₃	187	29	–	–	–	0.2	Farmer et al. (1986)
<i>Hydrilla verticillata</i> ^a	C ₄ NADP-ME	23–45	116–330	30–41	23–30	104–175	6.6 ^b	Holaday and Bowes (1980; Magnin et al. 1997; Salvucci and Bowes 1983)
<i>Egeria densa</i> ^a	C ₄ NADP-ME	71	130		60		1.8	Casati et al. (2000; Salvucci and Bowes 1983)
Eleocharis								
<i>Isoetes howellii</i>	CAM	256	36	110	37	2	0.1	Keeley (1998)
<i>Crassula aquatica</i>	CAM	392	178	208	78	2	0.5	Keeley (1998)
<i>Ottelia acuminata</i> ^a	C ₄ ?	55	100	114	11	3,740	1.8	This study
<i>Ottelia alismoides</i> ^a	C ₄ NAD-ME/ CAM?	45	264	246	13	1,740	5.9	This study
Terrestrial								
<i>Suaeda heterophylla</i>	C ₃	424	18	15	nd	168	0.04	Voznesenskaya et al. (2001)
<i>Borszczowia aralocaspica</i>	Single-cell C ₄	130	768	511	145	226	5.9	Voznesenskaya et al. (2001)
<i>Bienertia cycloptera</i>	Single-cell C ₄	258	1,368	101	20	510	5.3	Voznesenskaya et al. (2002)
Average	C ₄ NADP-ME	60–240	780–1,440	240–480	600–960	<60	7.4 ^b	Kanai and Edwards (1999)
Average	C ₄ NAD-ME	60–180	720–1,500	24–540	<60	300–1,080	9.3 ^b	Kanai and Edwards (1999)
Average	C ₄ PEP-CK	60–240	1,020–1,620	120–240	<60	60–180	7.3 ^b	Kanai and Edwards (1999)
<i>Mesembryanthemum crystallinum</i>	CAM	80	1,039	44	62	112	13.0	Winter et al. (1982)

Activities ($\mu\text{mol mg}^{-1} \text{Chla h}^{-1}$) were measured between 22 and 30 °C

nd not determined

^a Adapted to low CO₂ conditions

^b Ratio based on average of the range of activities for PEPC and RuBisCO

be a possible metabolism involving organic acids besides CAM.

Aquatic C₄ photosynthesis is probably more ancient than that of terrestrial C₄ and is also likely to be polyphyletic. The marine macroalga *U. flabellum* (Chlorophyta, Udoteaceae) performs C₄ metabolism, but PEPCK is believed to carry out the dual role of carboxylation and decarboxylation (Reiskind and Bowes 1991). It has recently been proposed that another marine macroalga *Ulva prolifera* (Chlorophyta, Ulvophyceae) has C₄ metabolism based on the presence of PEPC and PPDK (Xu et al. 2012). Within microalgae, C₃–C₄ metabolism has been described in some marine diatoms (Bacillariophyta that arose about 180 million years ago) such as *Thalassiosira weissflogii* (Reinfelder 2011; Reinfelder et al. 2000),

but appears to be absent in others such as *T. pseudonana* and *Phaedodactylum tricorutum* (Haimovich-Dayana et al. 2013; Roberts et al. 2007).

Within aquatic angiosperms, C₄ photosynthesis appears to be largely restricted to the Hydrocharitaceae, a monocotyledonous family of 18 genera and about 120 species that is believed to have an Oriental origin about 65 million years ago (Chen et al. 2012). The genus *Stratiotes* is believed to be the first diverging lineage of the Hydrocharitaceae and two clades have been recognized: Clade A includes *Hydrilla*, *Najas*, *Vallisneria* and the seagrasses *Halophila*, *Thalassia* and *Enhalus*, and Clade B includes *Ottelia*, *Egeria*, *Elodea* and *Lagarosiphon* (Chen et al. 2012). Both clades contain species with C₄ activity: within Clade A in *Hydrilla* (Bowes et al. 2002; Bowes 2011) and

possibly in the seagrass *Halophila* (Koch et al. 2013), within Clade B in *Egeria* (Casati et al. 2000) and *Ottelia* (this study). Outwith the Hydrocharitaceae, although within the order Alismatales, the aquatic angiosperm (seagrass) *Cymodocea* (Cymodoceaceae) may also have some evidence for C₄ metabolism (Koch et al. 2013), but this requires further investigation.

There is also a high incidence of CAM-like activity, or at least evidence for elevated concentrations of organic acids, in the Hydrocharitaceae. The report here of CAM in *O. alismoides* at low CO₂ contrasts with the data from (Webb et al. 1988) where a diel change of only 7 $\mu\text{equiv g}^{-1}$ FW has been found in the amphibious *O. ovalifolia*; however, our results suggest that CAM is facultative in *O. alismoides* and apparently absent in *O. acuminata* so this is not necessarily contradictory. *V. americana* and *V. spiralis* (Clade B) show evidence for CAM with diel changes in acidity of up to 42 and 51 $\mu\text{equiv g}^{-1}$ FW, respectively (Keeley 1998; Webb et al. 1988). Diel changes in acid contents in other Hydrocharitaceae such as species from the genera *Egeria*, *Elodea* and *Lagarosiphon* are relatively low (Keeley 1998; Webb et al. 1988). Earlier studies reported fixation of ¹⁴C into C₄ acids in species of freshwater Hydrocharitaceae within the genera: *Egeria*, *Elodea* and *Lagarosiphon* (Brown et al. 1974; Browse et al. 1977; Degroote and Kennedy 1977; Salvucci and Bowes 1983) and also in the marine *Halophila* (Beer 1989) although, there is little evidence for turnover in pulse-chase experiments. Thus, the precise role of these acids and their relationship to C₄, CAM and C₃–C₄ intermediates or other functions such as pH-regulation remains to be elucidated within the Hydrocharitaceae in species that do not appear to operate C₄ or CAM.

Comparison of *O. acuminata* and *O. alismoides*

Ottelia alismoides is an annual plant and is widespread in tropical and warmer regions of Asia and Australia (Cook and Urmikoni 1984). It can grow in still or flowing water to a depth of about 1 m. It is also found elsewhere as a non-native such as in Louisiana in the south of the USA (<http://plants.usda.gov>). In contrast, *O. acuminata* is a perennial with a restricted distribution, being confined to western China where it grows in still and flowing water to a depth of 5 m. Both species are able to use bicarbonate in addition to CO₂ as an inorganic carbon source for photosynthesis, but *O. alismoides* appears to have greater flexibility in its CCMs with apparently facultative CAM and constitutive C₄ metabolism. It is tempting to suggest that this flexibility of CCM operation may be linked to its annual growth cycle with the requirement to produce seeds at the end of a growing season, its distribution in shallow tropical waters where ecological success is likely to be favoured by high growth rates, and also that an efficient and effective carbon

uptake system may increase its potential to invade other non-native habitats.

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