SEARCH FOR NEW DRUGS

SYNTHESIS AND ANTIBACTERIAL ACTIVITY OF 1,2,3,4-TETRAHYDRO- AND PYRIDO[1,2-*a*]BENZIMIDAZOLES

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Translated from Khimiko-Farmatsevticheskii Zhurnal, Vol. 56, No. 1, pp. 25 – 31, January, 2022.

Original article submitted September 1, 2021.

We report here the synthesis of a series of different 1,2,3,4-tetrahydro- and pyrido[1,2-*a*]benzimidazoles and assessment of their *in vitro* antibacterial activity against Gram-positive (*Bacillus cereus*) and Gram-negative (*Escherichia coli* AB1157, *Pseudomonas aeruginosa* PAO1) bacteria. Pyrido[1,2-*a*]benzimidazoles were the most effective against *Bacillus cereus*, while 1,2,3,4-tetrahydro derivatives produced the greatest level of growth inhibition of Gram-negative (*Escherichia coli* AB1157) bacteria. The antibacterial activity of 2,4-dimethyl-7,8-dinitropyrido[1,2-*a*]benzimidazole was comparable to or exceeded the efficacy of commercial tetracycline, kanamycin, levomycetin, and erythromycin formulations. A genotoxicology test (the *Allium* test) was used to study the mitosis-modifying and mutagenic actions of a series of condensed azaheterocyclic compounds with marked antibacterial effects. On the basis of the antibacterial activity of some of the condensed benzimidazole derivatives containing a nodal nitrogen atom detected here and the low toxicity of similar azaheterocycles led to the conclusion that there is potential for further searches for novel antibiotics among compounds of this class.

Keywords: pyrido[1,2-*a*] benzimidazoles, 1,2,3,4-tetrahydropyrido[1,2-*a*]benzimidazoles, antibacterial activity, mitotic and phasic indexes, anaphase-telophase analysis.

Bacterial infectious diseases remain among the leading causes of death throughout the world [1]. The antibiotics used for treating them lose efficacy over time because of the development of resistance in bacteria [2, 3]. The result is that treatment of resistant microorganisms becomes more problematic and requires the use of new drugs. Studies of the synthesis of novel biologically active compounds and studies of their antibacterial activity are constantly ongoing.

Many researchers seeking novel antibiotics have in recent years been attracted to various condensed derivatives of benzimidazole with nodal nitrogen atoms, such as pyrido-[1,2-a]benzimidazoles. This is associated with the fact that such substances have high antimicrobial activity against a variety of bacterial species [4 - 10].

Thus, series of compounds have been found to be active against Gram-negative *Escherichia coli* [4, 7–10], *Salmo-nella typhi* [7–9], and *Vibrio cholerae* [7, 8] and Gram-positive *Bacillus subtilis* [6–8], *Clostridium tetani* [6, 8], *Streptococcus pneumoniae* [6], *Staphylococcus aureus* [4, 9, 10], and *Mycobacterium tuberculosis* [5]. At the same time, the literature lacks data on the antimicrobial activity of tetrahydro derivatives of pyrido[1,2-*a*]benzimidazole and the effects of substituents in the benzene ring of this condensed heterocycle on their activity have received virtually no study.

Thus, we report here the synthesis of a wide range of condensed benzimidazole derivatives containing pyridine (I-XVII) or piperidine (XVIII-XXI) rings annealed to the imidazole moiety; we also report the first studies of their antibacterial activity. The mitosis-modifying and mutagenic actions of the substances with the greatest antibacterial effects were also studied.

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 $I: R^{1} = R^{2} = H; II: R^{1} = CF_{3}, R^{2} = H; III: R^{1} = COOEt, R^{2} = H; IV: R^{1} = CN, R^{2} = H; V: R^{1} = NO_{2}, R^{2} = H; VI: R^{1} = CI, R^{2} = H; VI: R^{1} = CI, R^{2} = H; VI: R^{1} = CO, R^{2} = H; XII: R^{1} = CO, R^{2} = H; XII: R^{1} = COEt, R^{2} = H; XII: R^{1} = NO_{2}, R^{2} = H; XII$

EXPERIMENTAL CHEMICAL SECTION

Melting temperatures were measured on a PolyTherm A apparatus with a heating rate of 3°C/min and were not corrected. ¹H and ¹³C NMR spectra were recorded on a Bruker DRX 400 spectrometer, SF = 400 MHz, solvent DMSO-d₆, temperature 25°C. The standard for determining chemical shifts consisted of residual proton signals from the solvent in proton spectra (2.5 ppm) or DMSO-d₆ signals in carbon spectra (39.5 ppm. High-resolution mass spectra were recorded on a Bruker micrOTOF II apparatus (Bruker Deltronics) with electrospray ionization (ESI) and a mass scanning range (*m*/*z*) of 3000 Da, with samples injected by syringe. The solvent was MeCN and the solvent flow rate was 3 µl/min. The interface temperature was 180°C and the carrier gas was nitrogen (4.0 liters/min).

Pyrido[1,2-a]benzimidazoles (**I-VIII**) and 1,2,3,4-tetrahydropyrido[1,2-a]benzimidazoles (**XVIII-XXI**) were synthesized as described in [11].

Nitro derivatives of pyrido[1,2-a]benzimidazole (**XI-XIV**) were synthesized as described in [12].

Amino derivatives of pyrido[1,2-*a*]benzimidazole (**IX**, **X**, **XV-XVII**) were synthesized as described in [13].

EXPERIMENTAL BIOLOGICAL SECTION

The antibacterial activity of pyrido[1,2-*a*]benzimidazoles (**I-XVII**) and 1,2,3,4-tetrahydropyrido[1,2-*a*]benzimidazoles (**XVIII-XXI**) was studied using a standard method [14] on microorganisms widely used for developing antibiotics. Tests used two Gram-negative strains (*Escherichia coli* AB1157 and *Pseudomonas aeruginosa* PAO1, from the collection of the Institute of Molecular Genetics, provided by the Kurchatov Institute, Moscow) and one Gram-positive strain (*Bacillus cereus*, from the collection of the Department of Botany and Microbiology, Yaroslavl State University).

The antibacterial activity of pyrido[1,2-*a*]benzimidazole derivatives was assessed using commercial antibiotics used against Gram-negative and Gram-positive bacteria: gentamicin, tetracycline, kanamycin, levomycetin, and erythromycin.

Testing by the serial dilutions method was run in sterile 96-well immunology plates in volumes of 0.2 ml with final microorganism concentrations of 10^6 cfu/ml. Minimal inhibitory concentration (MIC) were determined for test compounds, at which they completely suppressed bacterial growth at 24 h of incubation at 37°C. Bacterial growth was assessed by measuring absorption at 600 nm with a Bio-Rad iMark photometric plate reader.

Mitosis-modifying and mutagenic properties of the compounds with the greatest antibacterial activity were studied using a genotoxicity test (the *Allium* test) on meristem cells from *Allium cepa* sprout root cells as described in [15]. Commercial antibiotics – tetracycline and gentamycin – were used as reference compounds.

Mutagenic activity was evaluated in terms of the ability of substances to induce chromosomal aberrations at the anaphase-telophase stage. Factors assessed were: the total number of cells (\approx 600), the total number of dividing cells at the anaphase-telophase stage, and the number of cells with chromosomal aberrations.

Mitotoxic activity and mitosis-modifying actions were assessed in the same time-based preparations as anaphase-telophase analysis. The mitosis-modifying actions of compounds were evaluated in terms of the level of mitotic activity in tissues – the mitotic index (MI, %) – and phase duration indexes. The total number of cells (\approx 600), the number of nondividing cells, and the numbers of cells at different phases of mitosis were determined.



Significant differences would experiments and controls were identified by two-factor analysis of variance and Student's test. Differences were taken as significant at p < 0.05.

RESULTS AND DISCUSSION

Pyrido[1,2-*a*]benzimidazoles (**I-VIII**) were synthesized by reductive intramolecular heterocyclization of quaternary N-(2-nitro(phenyl)aryl) pyridinium salts (Scheme 1). The electron source was an electric current, use of which significantly reduced the cost of synthesis and excluded formation of toxic wastes. The extraction procedure was also simplified, as the reaction material at the end of the reaction contained only the reaction product, ethanol, and the background electrolyte, which was 6% HCl. Syntheses were run at a temperature of 40°C by passage of 4.5 F/mol of electricity through an electrolytic cell. The yields of products **I-VIII** were 85 - 96%.

Nitration and reduction reactions yielded pyrido-[1,2-*a*]benzimidazoles (**XI-XVII**), which contained both an electron-donor or electron-acceptor substituent or substituents of different electronic nature. The nitrating agent was a mixture of KNO₃ and H₂SO₄. Reactions were run at 30°C for 3 h. Nitro products **XI-XIV** were extracted with yields of 89 – 94%. Reduction of nitro compounds **V**, **VIII**, and **XI-XIV** was with titanium (III) chloride, which produced contaminant-free amines **IX**, **X**, **XV-XVII** with yields of 91 – 97%.

1,2,3,4-Tetrahydropyrido[1,2-a]benzimidazoles (**XVIII-XXI**) were also synthesized by reductive intramolecular cyclization (Scheme 2). By comparison with pyrido[1,2-a]-benzimidazoles (**I-VIII**), these reactions required passage of half the quantity of electricity. Electrolyte was 8% HCl containing N-(2-nitro-4-R¹-phenyl)piperidine. Heterocycles **XVIII-XXI** were obtained with yields of 89 – 97%.

This yielded a large series of condensed benzimidazole derivatives with a nodular nitrogen atom (structures I-XXI,

Compound	R^1	R^2	HRMS, $m/z [M + H]^+$	mp, °C	Yield, %
Ι	Н	Н	Calculated 169.0766. C ₁₁ H ₉ N ₂ Found 169.0758	176 – 178 [16]	92
II	CF_3	Н	Calculated 237.0639. $C_{12}H_8F_3N_2$ Found 237.0637	233 – 235 [17]	96
III	COOEt	Н	Calculated 241.0977. $C_{14}H_{13}N_2O_2$ Found 241.0971	179 – 182 [17]	91
IV	CN	Н	Calculated 194.0713. C12H8N3 Found 194.0712	242 - 244 [17]	92
V	NO_2	Н	Calculated 214.0617. C ₁₁ H ₈ N ₃ O ₂ Found 214.0611	290 - 294 [18]	85
VI	Cl	Н	Calculated 203.0376. C ₁₁ H ₈ ClN ₂ Found 203.0371	212 - 214 [13]	95
VII	CN	CH_3	Calculated 222.1031. $C_{14}H_{12}N_3$ Found 222.1037	227 – 229 [17]	91
VIII	NO_2	CH_3	Calculated 242.0930. $C_{13}H_{12}N_3O_2$ Found 242.0921	214 - 217	89
IX	NH_2	Н	Calculated 184.0875. $C_{11}H_{10}N_3$ Found 184.0868	178 – 182 [18]	93
X	NH_2	CH_3	Calculated 212.1188. $C_{13}H_{14}N_3$ Found 212.1183	134 - 137	91
XI	CF_3	Н	Calculated 282.0492. $C_{12}H_7F_3N_3O_2$ Found 282.0485	225 – 228 [12]	92
XII	COOEt	Н	Calculated 286.0830. $C_{14}H_{12}N_3O_4$ Found 286.0822	172 – 175 [12]	94
XIII	NO_2	Н	Calculated 259.0469. C11H7N4O4 Found: 259.0463	276 – 278 [12]	91
XIV	NO_2	CH_3	Calculated 287.0781. $C_{13}H_{11}N_4O_4$ Found: 287.0772	195 – 199	89
XV	CF_3	Н	Calculated 252.0749. $C_{12}H_9F_3N_3$ Found 252.0743	233 – 235 [19]	97
XVI	COOEt	Н	Calculated 256.1087. $C_{14}H_{14}N_3O_2$ Found 256.1082	201 - 203	93
XVII	NH_2	Н	Calculated 199.0984. C11H11N4 Found 199.0981	241 - 243 [13]	91
XVIII	Н	-	Calculated 173.1079. $C_{11}H_{13}N_2$ Found 173.1068	90 - 93 [11]	91
XIX	CF_3	-	Calculated 241.0953. $C_{12}H_{12}F_3N_2$ Found 241.0947	129 – 132 [11]	97
XX	CN	-	Calculated 198.1032. $C_{12}H_{12}N_3$ Found 198.1021	169 – 172 [11]	92
XXI	Cl	-	Calculated 207.0689. C ₁₁ H ₁₂ ClN ₂ Found 207.0684	147 – 149 [11]	89

TABLE 1. Characteristics of Pyrido[1,2-a]benzimidazoles (I-XVII) and 1,2,3,4-Tetrahydropyrido[1,2-a]benzimidazoles (XVIII-XXI)

Synthesis and Antibacterial Activity

Table 1), some being compounds not previously described in the literature.

The structures of pyrido[1,2-a]benzimidazole derivatives (I-XXI) were confirmed by ¹H and ¹³C NMR spectroscopy and high-resolution mass spectrometry (HRMS) (Tables 1 and 2).

Data on the antibacterial activity of substances I-XXI are presented in Table 3.

As shown by the data in Table 3, MIC for most pyrido-[1,2-a]benzimidazoles (I-XVII) was in the range 500 –

1000 μ g/ml. Introduction of acceptor (compounds II-V, XI, XII) and donor (IX, XVII) substituents and substituents with different electronic natures (XV, XVI) into the benzene ring of unsubstituted heterocycle I did not in most cases increase antimicrobial activity. Only compounds containing a chlorine atom (VI) and two nitro groups (XIII) produced 100% inhibition of bacterial growth of *P. aeruginosa* and *Bacillus* sp. Respectively, at 62.5 μ g/ml. The presence of two methyl

TABLE 2. ¹H and ¹³C NMR Spectral Data for Pyrido[1,2-*a*]benzimidazoles (**I-XVII**) and 1,2,3,4-Tetrahydropyrido[1,2-*a*]benzimidazoles (**XVIII-XXI**)*

Compound	R^1	R^2	¹ H NMR spectrum, δ, ppm	¹³ C NMR spectrum, δ, ppm
I	Н	Н	[16]	[16]
II	CF ₃	Н	17]	[17]
III	COOEt	Н	[17]	[17]
IV	CN	Н	[17]	[17]
V	NO ₂	Н	[18]	112.4, 113.4, 115.5, 115.8, 118.1, 128.1, 132.4, 133.3, 144.2, 146.6, 151.5
VI	Cl	Н	[13]	110.8, 113.4, 116.9, 118.1, 120.5, 127.2, 127.4, 129.7, 130.8, 144.7, 148.8
VII	CN	CH_3	[17]	[17]
VIII	NO ₂	CH ₃	2.29 (s, 3H, CH ₃), 2.51 (s, 3H, CH ₃), 7.24 (s, 1H, H ₃), 8.06 (dd, 1H, H ₈ , J 9.0, 2.1 Hz), 8.25 (d, 1H, H ₉ , J 9.0 Hz), 8.51 (d, 1H, H ₆ , J 2.2 Hz), 8.64 (s, 1H, H ₁)	17.3, 18.2, 113.2, 115.5, 115.7, 121.8, 122.7, 126.7, 133.5, 143.3, 145.4, 150.8
IX	NH ₂	Н	[18]	100.9, 110.1, 111.6, 112.4, 116.5, 121.7, 126.8, 129.1, 146.5, 148.1, 148.3
Х	NH ₂	CH ₃	$\begin{array}{l} 2.27 \; (s, 3H, CH_3), 2.46 \; (s, 3H, CH_3), 5.09 \; (s, 2H, NH_2), \\ 6.66 \; (dd, 1H, H_8, J \; 8.6, 2.0 \; Hz), \; 6.84 \; (d, 1H, H_6, J \; 2.0 \; Hz), \\ 7.06 \; (s, 1H, H_3), \; 7.81 \; (d, 1H, H_9, J \; 8.6 \; Hz), \; 8.49 \; (s, 1H, H_1) \end{array}$	17.5, 18.2, 101.1, 111.6, 112.3, 119.2, 121.8, 122.2, 125.4, 130.4, 146.1, 147.6, 147.8
XI	CF ₃	Н	[12]	[12]
XII	COOEt	Н	[12]	[12]
XIII	NO_2	Н	[12]	[12]
XIV	NO ₂	CH ₃	2.32 (s, 3H, CH ₃), 2.50 (s, 3H, CH ₃), 7.46 (s, 1H, H ₃), 8.41 (s, 1H, H ₆), 8.95 (s, 1H, H ₁), 9.22 (s, 1H, H ₉)	17.4, 18.2, 112.1, 113.4, 115.1, 117.2, 128.1, 128.4, 133.4, 133.6, 141.6, 145.0, 152.4
XV	CF ₃	Н	[19]	97.0, 110.2, 112.8 q (J 6.0 Hz), 117.0 q (J 33.0 Hz), 124.1 q (CF 3, J 272 Hz), 126.2, 126.7, 129.5, 132.1, 135.3, 140.9, 147.8
XVI	COOEt	Н	1.38 (t, 3H, CH ₃ , J 7.0 Hz), 4.36 (quin, 2H, CH ₂ , J 6.9 Hz), 6.40 (s, 2H, NH ₂), 6.80 (t, 1H, H ₂ , J 6.7 Hz), 7.36 – 7.38 m (1H, H ₃), 7.39 (s, 1H, H ₉), 7.50 (d, 1H, H ₄ , J 9.3 Hz), 8.26 (s, 1H, H6), 8.60 (d, 1H, H1, J 6.7 Hz)	14.2, 60.2, 95.4, 109.9, 110.6, 117.3, 121.7, 126.8, 129.6, 133.5, 135.3, 146.2, 148.2, 167.6
XVII	NH ₂	Н	[13]	94.2, 101.3, 108.8, 115.8, 121.4, 125.0, 125.6, 133.0, 136.7, 138.0, 145.3
XVIII	Н	-	[11]	[11]
XIX	CF_3	-	[11]	[11]
XX	CN	-	[11]	[11]
XXI	Cl	-	[11]	[11]

* Table 2 shows NMR spectral data for new condensed heterocycles (VIII, X, XIV, XVI) and previously unpublished ¹³C NMR spectral data for some of the compounds. References to reports containing spectral data are given for other compounds.

groups in the pyridine ring of study compounds had greater influences. Thus, compounds **VII**, **VIII**, **XIII**, and **XIV** inhibited the growth of *Bacillus* sp. at significantly lower concentrations than structures **IV**, **V**, **X**, and **XI**. The antibacterial activity of 2,4-dimethyl-7,8-dinitropyrido- [1,2-*a*]benzimidazole (**XIV**) against *Bacillus* sp. was comparable with or greater than that of the efficacy of commercial tetracycline, kanamycin, levomycetin, and erythromycin formulations. Comparison of pyrido[1,2-*a*]benzimidazoles (**I**, **II**, **IV**, and **VI**) and 1,2,3,4-tetrahydropyrido[1,2-*a*]benzimidazoles (**XVIII-XXI**) showed that the latter had more marked antibacterial activity, especially against *E. coli*.

Thus, this study established that the nature of the substituent in the benzene ring had no significant effects on the antimicrobial activity of compounds. MIC was more dependent on the structure of the heterocycle annealed to the imidazole. Pyrido[1,2-a]benzimidazoles produced greater inhibition of the growth of Gram-positive *Bacillus cereus* bacteria. The presence of methyl groups in the pyridine fragment increased the antimicrobial action of compounds. 1,2,3,4-Tetrahydropyrido[1,2-*a*]benzimidazoles were more effective against Gram-negative *Escherichia coli* AB1157.

The mitosis-modifying and mutagenic properties of compounds VI, VII, XIII, XIV, and XVIII, which had the greatest antibacterial activity among the study compounds, were studied using a genotoxicity test (the *Allium* test).

Study compounds VI, VII, XIII, XIV, and XVIII were shown, like tetracycline and gentamicin, to decrease the number of dividing cells as compared with controls (6.1%) at all concentrations tested. Mitotic index was 5.6%, 5.7%, 3.9%, 3.1%, and 5.8% for substances VI, VII, XIII, XIV, and XVIII respectively. Nitrogen-containing compounds XIII and XIV were found to have the greatest effects on mitotic index. The cytotoxicity of these compounds was comparable with that of tetracycline (MI 3.3%, C = 0.01 mg/ml) and was greater than that of gentamicin (MI 6.0%, C =

TABLE 3. Antimicrobial Activity of Pyrido[1,2-a]benzimidazoles (I-XVII) and 1,2,3,4-Tetrahydropyrido[1,2-a]benzimidazoles(XVIII-XXI)

Compound	\mathbf{p}^1	\mathbf{P}^2	MIC, µg/ml			
Compound	К	К	P. aeruginosa	E. coli	Bacillus sp.	
Ι	Н	Н	500	500	250	
II	CF ₃	Н	1000	1000	500	
III	COOEt	Н	1000	1000	500	
IV	CN	Н	1000	1000	500	
V	NO_2	Н	1000	1000	1000	
VI	Cl	Н	62.5	500	500	
VII	CN	CH ₃	500	500	62.5	
VIII	NO ₂	CH ₃	1000	1000	250	
IX	NH ₂	Н	1000	500	1000	
X	NH ₂	CH ₃	500	500	500	
XI	CF ₃	Н	500	500	500	
XII	COOEt	Н	500	500	500	
XIII	NO_2	Н	1000	1000	62.5	
XIV	NO_2	CH ₃	1000	500	7.8	
XV	CF ₃	Н	1000	1000	500	
XVI	COOEt	Н	1000	500	500	
XVII	NH ₂	Н	500	500	500	
XVIII	Н	-	500	125	500	
XIX	CF ₃	-	1000	500	500	
XX	CN	-	500	500	250	
XXI	Cl	-	500	250	500	
Gentamicin			15.6	15.6	31.2	
Tetracycline			15.6	7.8	7.8	
Levomycetin			15.6	15.6	15.6	
Erythromycin			500	500	125	

Synthesis and Antibacterial Activity

		Phase			
Compound	Concentration, mg/ml	Prophase index, $\% \pm m$	Metaphase index, $\% \pm m$	Anaphase-telophase index, $\% \pm m$	
Control	-	56.6 ± 2.5	19.7 ± 2.1	23.7 ± 2.5	
VI	0.1	$48.1 \pm 2.3*$	$28.3 \pm 2.4*$	23.6 ± 2.0	
	0.001	$50.5\pm1.5*$	$26.8\pm2.4*$	22.7 ± 2.4	
VII	0.01	$47.4 \pm 2.1*$	$30.3 \pm 1.1*$	22.3 ± 1.5	
	0.001	$49.2\pm0.9*$	$29.6 \pm 1.4 *$	21.2 ± 0.7	
XIII	0.01	$46.8\pm0.9*$	$34.6\pm0.9*$	$18.6\pm0.9*$	
	0.001	$47.5 \pm 1.2*$	$31.2 \pm 1.4*$	21.3 ± 0.8	
XIV	0.01	$45.2 \pm 1.2*$	$36.1 \pm 2.1*$	$18.7\pm0.9*$	
	0.001	$46.7 \pm 1.5*$	$33.6 \pm 1.2*$	$19.7 \pm 1.4 *$	
XVIII	0.01	$51.2 \pm 0.8*$	$24.9\pm2.0*$	23.9 ± 1.3	
	0.001	$53.1 \pm 0.7*$	$22.8\pm1.2*$	24.1 ± 1.2	
Tetracycline	0.01	$52.4 \pm 1.3*$	$24.8\pm2.0*$	22.8 ± 2.1	
	0.001	55.1 ± 1.8	21.7 ± 1.9	23.2 ± 1.4	
Gentamicin	0.01	$50.6 \pm 1.2*$	$23.3\pm1.1*$	26.1 ± 1.5	
	0.001	54.9 ± 1.6	18.9 ± 1.3	26.2 ± 1.5	

TABLE 4. Phase Indexes in the Meristem of Allium cepa Root Shoots Treated with Study Compounds

* Significantly different from control, p < 0.05

TABLE 5.	Frequencies of Chromosome	Mutations in C	ells of Allium	cepa Root	Meristem on	Exposure to	Study	Substances	(Number	of
Anaphase-7	Felophases Examined = 200)									

Compound	Concentration	Chromosome aberrations, $\% \pm m$	Chromosome aberrations and delays, $\% \pm m$
Control	-	0.7 ± 0.1	1.1 ± 0.2
VI	0.01	$2.1 \pm 0.1*$	$3.1 \pm 0.2*$
	0.001	$1.8 \pm 0.3*$	$2.9 \pm 0.2*$
VII	0.01	$2.4 \pm 0.2*$	$3.5 \pm 0.1*$
	0.001	$1.9 \pm 0.2*$	$3.0 \pm 0.1*$
XIII	0.01	$2.4\pm0.1*$	$4.5 \pm 0.1*$
	0.001	$1.8 \pm 0.2*$	$3.0 \pm 0.1*$
XIV	0.01	$4.1 \pm 0.1*$	$5.1 \pm 0.1*$
	0.001	$2.8 \pm 0.1*$	$3.7 \pm 0.1*$
XVIII	0.01	$1.4 \pm 0.2*$	$1.7 \pm 0.1*$
	0.001	$1.1 \pm 0.1*$	1.6 ± 0.2
Tetracycline	0.01	$1.0 \pm 0.1*$	$1.4 \pm 0.2*$
	0.001	0.7 ± 0.1	1.3 ± 0.1
Gentamicin	0.01	0.8 ± 0.2	1.3 ± 0.1
	0.001	0.7 ± 0.3	1.2 ± 0.1

* Significantly different from control, p < 0.05

0.01 mg/ml). Compounds VI, VII, and XVIII had relatively low toxicity at a concentration of 0.01 mg/ml. All study compounds produced some degree of change in the durations of mitosis phases as compared with controls, and there was an overall tendency to decreases in the numbers of cells at the prophase and anaphase-telophase stages and an increase in the numbers of cells in metaphase (Table 4).

Anaphase-telophase analysis showed that study azaheterocycles induced genetic lesions in *Allium cepa* cells (Table 5). Compound **XVIII**, containing an unsubstituted heterocycle, had the smallest effect. The largest numbers of chromosome aberration as compared with controls were obtained with compounds **XIII** and **XIV**, containing two nitro groups. At a concentration of 0.01 mg/ml, these compounds increased the number of genetic lesions by factors of 4.1 and 4.6 as compared with controls; at 0.001 mg/ml, increases were by factors of 2.7 and 3.4 respectively. The remaining study compounds displayed no significant mutagenic effects. Positive control substance tetracycline also increased mutation frequency at C = 0.001 mg/ml. Gentamicin produced no increase in the number of chromosome aberrations.

Thus, compounds VI, VII, XIII, XIV, and XVIII had mutagenicity and mitosis-modifying activity comparable to those of other members of this class of condensed heterocycles.

The antibacterial activity of several of the benzimidazole derivatives with nodular nitrogen atoms studied here (VI, VII, XIII, XIV, and XVIII), along with the low toxicity of thee azaheterocycles, leads to the conclusion that there is potential for further searches for novel antibiotics among compounds of this class.

This study was carried out in the framework of the Yaroslavl State University Development Program, project No. P2-K-1-G-1/2021.

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