

SYNTHESIS, ANTIMICROBIAL, AND PROTISTICIDAL ACTIVITY OF 3-ARYLOXYETHYL(BENZYL)-1-CARBAMOYLMETHYL-2-IMINO BENZIMIDAZOLINE HYDROCHLORIDES

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Alkylation of 2-aminobenzimidazole by methyl iodide and benzyl- or 2-aryloxyethyl bromides produced 1-substituted 2-aminobenzimidazoles that were quaternized by chloroacetamide to previously undescribed 3-aryloxyethyl(benzyl)-1-carbamoylmethyl-2-iminobenzimidazoline hydrochlorides. These compounds were shown to possess antibacterial activity against several pathogenic Gram-positive and Gram-negative microbes (*Staphylococcus aureus*, *Escherichia coli*) combined with pronounced protistocidal activity against the protozoa *Colpoda steinii* that was on the level of the clinical reference drugs.

Keywords: 1-substituted 2-aminobenzimidazoles, 3-aryloxyethyl(benzyl)-1-carbamoylmethyl-2-iminobenzimidazoline hydrochlorides, antibacterial activity, *Staphylococcus aureus*, *Escherichia coli*, protistocidal activity, *Colpoda steinii*.

Resistance of pathogenic microorganisms to antibiotics qualified by the WHO is one of the greatest threats to human health [1-5]. About 25,000 patients per year die of multi-drug resistant infections, including nosocomial ones, even in the relatively safe European Union. The problem is exacerbated by the emergence of multi-drug resistant bacterial strains that are unaffected by the majority of commercial antibiotics. Therefore, the discovery of new natural and synthetic antibacterial drugs is one of the most important tasks of modern science. It combines the efforts of organic chemists, microbiologists, physicians, and other specialists.

Benzimidazoles are known to include many compounds with antimicrobial activity [6 – 11]. They comprise 2-unsubstituted 1- ω -aryloxyalkylbenzimidazoles that are active against *Staphylococcus aureus* and *Salmonella typhi* [12]. Recently, 2-ureidobenzimidazoles were used as examples to study the mechanism of antimicrobial action of benzimidazoles that were highly effective against Gram-positive

pathogenic microbes. The activity was related to inhibition of two key bacterial enzymes, DNA-gyrase and DNA-topoisomerase [6, 7].

We patented earlier salts of 1- β -aryloxyethyl-3-benzyl-2-iminobenzimidazolines, which exhibited high antimicrobial activity against Gram-positive bacteria (*S. aureus*, *Staphylococcus epidermidis*, *Enterococcus faecalis*, *Streptococcus* sp., *Corynebacterium* sp.). Several of them were more active against *S. aureus* than antibiotics such as ampicillin and oxytetracycline [13].

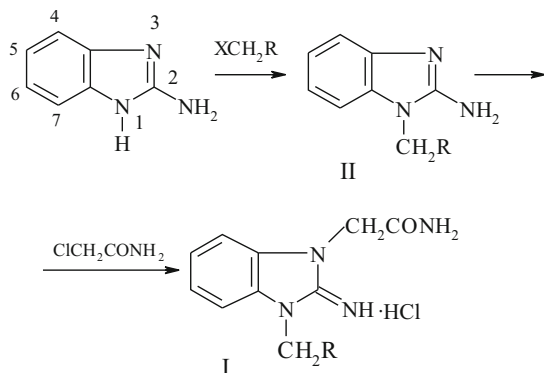
In continuation of this research, herein we report new functionalized derivatives of 2-iminobenzimidazoline, salts of 3-aryloxyethyl(benzyl)-1-carbamoylmethyl-2-iminobenzimidazolines (**I**), and their activity against *S. aureus*, *Escherichia coli*, and the ciliate *Colpoda steinii*.

Amides **Ib-i** were synthesized in 70-80% yields by quaternization of already known [13-15] or newly synthesized 1-aryloxyethyl(benzyl)-2-aminobenzimidazoles (**II**) using chloroacetamide in DMF at 120°C. The simplest 3-substituted 1-carbamoylmethyl-2-iminobenzimidazoline, 3-methyl-1-carbamoylmethyl-2-iminobenzimidazoline hydrochloride (**Ia**), was also synthesized and studied for comparison.

The structures of amides **I** were confirmed by PMR spectra and mass spectrometry.

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I, II: R=H (a); 2-FC₆H₄ (b);
 2-ClC₆H₄ (c); C₆H₅OCH₂ (d);
 4-CH₃C₆H₄OCH₂ (e);
 4-OCH₃C₆H₄OCH₂ (f);
 4-ClC₆H₄OCH₂ (g);
 4-BrC₆H₄OCH₂ (h);
 C₁₀H₇OCH₂ (i).

EXPERIMENTAL CHEMICAL PART

PMR spectra of the synthesized compounds in DMSO-d₆ or CDCl₃ were recorded on a Varian Unity-300 (300 MHz) spectrometer. Chemical shifts of protons were given relative to the residual deuterated solvent resonance. Mass spectra were measured on a Finnigan MAT INCOS 50 instrument with direct sample introduction into the source at ionization energy 70 eV. Elemental analyses of amines II and salts I agreed with those calculated.

2-Amino-1-methylbenzimidazole (IIa) was prepared by the literature method [14]; **2-amino-1-(2-phenoxyethyl)-benzimidazole (IIc)**, by the literature method [15]; **1-[2-(4-methylphenoxy)ethyl]-2-aminobenzimidazole (II d)** and **1-[2-(4-chlorophenoxy)ethyl]-2-aminobenzimidazole (II g)**, by the literature method [13].

2-Amino-1-(2-fluorobenzyl)benzimidazole (II b). A mixture of 2-aminobenzimidazole (1.31 g, 10 mmol) and

TABLE 1. Yields of Amide Hydrochlorides I and Their Melting Points and PMR Spectra

Compound	R	mp, °C	Yield, %	Empirical formula	PMR, δ, ppm
Ia	H	281 – 282	67	C ₁₀ H ₁₂ N ₄ O · HCl	3.75 (s, 3H, CH ₃), 4.95 (s, 2H, NCH ₂ CO), 7.28 – 7.36 [m, 3H, 5-H, 6-H, CONH ₂ (1H)], 7.42 (d, 1H, 7-H), 7.55 (d, 1H, 4-H), 7.75 (s, 1H, CONH ₂), 9.05 (s, 2H, =N ⁺ H ₂)
Ib	2-FC ₆ H ₄	268 – 271	81	C ₁₆ H ₁₃ FN ₄ O · HCl	5.00 (s, 2H, NCH ₂ CO), 5.55 (s, 2H, NCH ₂ Ar), 7.10 – 7.25 [m, 7H, 5-H, 6-H, 3'-H, 4'-H, 5'-H, 6'-H + CONH ₂ (1H)], 7.35 (d, 1H, 7-H), 7.45 (d, 1H, 4-H), 7.90 s (1H, CONH ₂), 9.50 (s, 2H, =N ⁺ H ₂)
Ic	2-ClC ₆ H ₄	255 – 256	74	C ₁₆ H ₁₅ ClN ₄ O · HCl	5.20 (s, 2H, NCH ₂ CO), 5.55 (s, 2H, NCH ₂ Ar), 7.15 – 7.30 [m, 7H, 5-H, 6-H, 3'-H, 4'-H, 5'-H, 6'-H + CONH ₂ (1H)], 7.45 (d, 1H, 7-H), 7.55 (d, 1H, 4-H), 7.80 (s, 1H, CONH ₂), 9.55 (s, 2H, =N ⁺ H ₂)
Id	C ₆ H ₅ OCH ₂	261 – 262	78	C ₁₇ H ₁₈ N ₄ O ₂ · HCl	4.25 (t, 2H, NCH ₂ CH ₂ OAr), 4.65 (t, 2H, CH ₂ O), 4.90 s (2H, NCH ₂ CO), 6.80 (d, 2H, 3'-H, 5'-H), 6.84 (t, 1H, 4'-H), 7.16 – 7.40 [m, 6H, 5-H, 6-H, 7-H, 2'-H, 6'-H + CONH ₂ (1H)], 7.60 (d, 1H, 4-H), 7.84 (s, 1H, CONH ₂), 9.35 (s, 2H, =N ⁺ H ₂)
Ie	4-CH ₃ C ₆ H ₄ OCH ₂	254 – 255	69	C ₁₈ H ₂₀ N ₄ O ₂ · HCl	2.45 (s, 3H, CH ₃), 4.26 (t, 2H, NCH ₂ CH ₂ OAr), 4.70 (t, 2H, CH ₂ O), 5.00 (s, 2H, CH ₂ CO), 6.85 (d, 2H, 3'-H, 5'-H), 7.15 – 7.30 [m, 5H, 5-H, 6-H, 2'-H, 6'-H, CONH ₂ (1H)], 7.40 (d, 1H, 7-H), 7.55 (d, 1H, 4-H), 7.92 (s, 1H, CONH ₂), 9.65 (s, 2H, =N ⁺ H ₂)
If	4-OCH ₃ C ₆ H ₃ OCH ₂	227 – 228	67	C ₁₈ H ₂₀ N ₄ O · HCl	3.65 (c, 3H, Me), 4.25 (t, 2H, NCH ₂ CH ₂ OAr), 4.68 (t, 2H, CH ₂ O), 5.05 (c, 2H, CH ₂ CO), 6.60 – 6.70 (m, 4H, 2'-H, 3'-H, 5'-H, 6'-H), 7.00 (s, 1H, CONH ₂), 7.05 – 7.25 (m, 2H, 5-H, 6-H), 7.35 (d, 1H, 7-H), 7.50 (d, 1H, 4-H), 7.95 (s, 1H, CONH ₂), 9.70 (s, 2H, =N ⁺ H ₂)
Ig*	4-ClC ₆ H ₃ OCH ₂	257 – 260	72	C ₁₇ H ₁₇ ClN ₄ O ₂ · HCl	4.28 (t, 2H, NCH ₂ CH ₂ OAr), 4.62 (t, 2H, CH ₂ O), 4.88 (s, 2H, CH ₂ CO), 6.82 (d, 2H, 3'-H, 5'-H), 7.24 – 7.34 (m, 4H, 5-H, 6-H, 2'-H, 6'-H), 7.38 (d, 1H, 7-H), 7.48 (s, 1H, CONH ₂), 7.64 (d, 1H, 4-H), 7.86 (s, 1H, CONH ₂), 9.26 (s, 2H, =N ⁺ H ₂)
Ih	4-BrC ₆ H ₃ OCH ₂	264 – 266	83	C ₁₇ H ₁₇ BrN ₄ O ₂ · HCl	4.30 (t, 2H, NCH ₂ CH ₂ OAr), 4.75 (t, 2H, CH ₂ O), 5.00 (s, 2H, CH ₂ CO), 6.75 (d, 2H, 3'-H, 5'-H), 7.05 (s, 1H, CONH ₂), 7.15 – 7.25 (m, 4H, 5-H, 6-H, 2'-H, 6'-H), 7.35 (d, 1H, 7-H), 7.45 (d, 1H, 4-H), 7.85 (s, 1H, CONH ₂), 9.70 (s, 2H, =N ⁺ H ₂)
Ii**	α-C ₁₀ H ₇ OCH ₂	278 – 279	88	C ₂₁ H ₂₀ N ₄ O ₂ · HCl	4.50 (t, 2H, NCH ₂ CH ₂ OAr), 4.90 (t, 2H, CH ₂ O), 5.05 (c, 2H, CH ₂ CO), 6.88 (d, 1H, 8'-H), 7.25 – 7.45 (m, 8H, 5-H, 6-H, 2'-H, 3'-H, 4'-H, 5'-H, 6'-H, 7'-H), 7.65 – 7.77 [m, 2H, 7-H, CONH ₂ (1H)], 7.80 – 8.05 [m, 2H, 4-H, CONH ₂ (1H)], 9.86 (s, 2H, =N ⁺ H ₂)

* Mass spectrum, *m/z* (*I*_{rel}, %): 287 (5.7), 190 (57), 173 (10), 146 (28.8), 119 (8.5), 91 (8.5), 77 (6.0), 44 (10), 36 (100).

KOH (0.67 g, 10 mmol) (calculated as 85% KOH) in DMSO (15 mL) was stirred at 35°C for 15 min, treated with 2-fluorobenzylbromide (2.08 g, 11 mmol) at a rate such that the temperature remained < 50°C, held for 30 min at 50°C, cooled, and treated stepwise with H₂O (40 mL). The resulting voluminous precipitate of amine **IIb** was filtered off and washed with KOH solution (5%, 2 mL) and H₂O. Yield 1.74 g (72%). White plates, mp 184–185°C (MeCN). C₁₄H₁₂FN₃. PMR spectrum (300 MHz), δ , ppm (CDCl₃)¹: 4.65 (br.s, 2H, NH₂), 5.19 (s, 2H, CH₂), 7.03–7.16 (m, 6H, 3'-H, 4'-H, 5'-H, 6'-H, 5-H, 6-H), 7.26 (d, 1H, 7-H), 7.44 (d, 1H, 4-H)

2-Amino-1-(2-chlorobenzyl)benzimidazole (IIc) was synthesized analogously to amine **IIb** from 2-aminobenzimidazole and 2-chlorobenzylbromide in 77% yield. Beige crystals, mp 157–159°C (EtOH). C₁₄H₁₂ClN₃. PMR spectrum (300 MHz), δ , ppm (CDCl₃): 4.70 (br.s, 2H, NH₂), 5.22 (s, 2H, CH₂), 7.00–7.22 (m, 6H, 3'-H, 4'-H, 5'-H, 6'-H, 5-H, 6-H), 7.25 (d, 1H, 7-H), 7.48 (d, 1H, 4-H).

2-Amino-1-[2-(4-methoxyphenoxy)ethyl]benzimidazole (IIIf). A solution of 2-aminobenzimidazole (1.33 g, 10 mmol) and KOH (1.3 g, 20 mmol) (calculated as 85% KOH) in H₂O (1.5 mL) was stirred vigorously, treated with Me₂CO (10 mL) and in portions with 2-(4-methoxyphenoxy)ethylbromide (2.54 g, 11 mmol) so that the temperature remained <40°C, stirred for 2.5 h at 45°C, and cooled. The acetone layer was separated, evaporated to half the volume, and poured into H₂O (20 mL). The resulting precipitate was filtered off and washed with H₂O (5 mL). Yield, 1.90 g (67%). Pale-pink crystals, mp 150–151°C (*i*-PrOH). C₁₆H₁₇N₃O₂. PMR spectrum (300 MHz), δ , ppm (CDCl₃):

3.72 (s, 3H, Me), 4.24 (t, 2H, NCH₂), 4.34 (t, 2H, OCH₂), 5.05 (s, 2H, NH₂), 6.74–6.80 (m, 4H, ArO), 7.05–7.18 (m, 3H, 5-H, 6-H, 7-H), 7.45 (d, 1H, 4-H).

2-Amino-1-[2-(4-bromophenoxy)ethyl]benzimidazole (IIh) was prepared analogously to **IIIf** from 2-aminobenzimidazole and 2-(4-bromophenoxy)ethylbromide in 79% yield. Colorless crystals, mp 167–168°C (*i*-PrOH). C₁₅H₁₄BrN₃O. PMR spectrum (300 MHz), δ , ppm (CDCl₃): 4.25 (t, 2H, NCH₂), 4.38 (t, 2H, OCH₂), 5.20 (br.s, 2H, NH₂), 6.74 (2H, 3'-H, 5'-H), 6.82–6.95 (m, 2H, 5-H, 6-H), 7.20–7.38 (m, 3H, 2'-H, 6'-H, 7-H), 7.45 (d, 1H, 4-H).

2-Amino-1-[2-(1-naphthoxy)ethyl]benzimidazole (IIIi) was prepared from 2-aminobenzimidazole and 2-(1-naphthoxy)ethylbromide analogously to **IIb** in 79% yield. Colorless crystals, mp 234°C (nitromethane). C₁₉H₁₇N₃. PMR spectrum (300 MHz), δ , ppm (CDCl₃): 4.48 (s, 4H, 2CH₂), 4.90 (br.s, 2H, NH₂), 6.80 (d, 1H, 8'-H), 7.06–7.20 (m, 3H, 5-H, 6-H, 2'-H), 7.30–7.48 (m, 5H, 3'-H, 4'-H, 5'-H, 6'-H, 7'-H), 7.78 (d, 1H, 7-H), 8.05 (d, 1H, 4-H).

Hydrochlorides of 3-aryloxyethyl(benzyl)-1-carbamoylmethyl-2-iminobenzimidazolines (I). General method. A solution of 1-substituted 2-aminobenzimidazole (**II**, 10 mmol) in DMF (20 mL) at 80°C was treated with chloroacetamide (0.94 g, 10 mmol) and held at 120°C for 1 h. The precipitate that formed on cooling was filtered off, washed with EtOH, and recrystallized from DMF (**Ia-c** and **Ii**) or a mixture (1:1) of DMF and EtOH (**Id-h**). Table 1 presents the yields, melting points, and PMR spectra of the synthesized compounds.

TABLE 2. Antibacterial Activity of Amide Hydrochlorides **Ia-i**

Compound	Minimum inhibiting concentration (MIC), $\mu\text{g/mL}$	
	<i>Staphylococcus aureus</i> , ATCC 25923	<i>Escherichia coli</i> , ATCC 25922
Ia	> 100	> 100
Ib	> 100	> 100
Ic	> 100	> 100
Id	> 100	> 100
Ie	50	> 100
If	100	50
Ig	50	> 100
Ih*	50	> 100
Ii	50	> 100
Ampicillin	6.3	6.3/-**
Furazolidone	3.1	12.5
Kanamycin	3–6	100

* MIC in *Staphylococcus epidermidis*, *Enterococcus faecalis*, *Streptococcus* sp., and *Corynebacterium* sp., 50 $\mu\text{g/mL}$;

** “6.3/-” represents the different sensitivity to ampicillin of the standard strain and the clinical isolate that is resistant to the antibiotic; “-”, a lack of antibacterial activity.

TABLE 3. Protistocidal Activity of Amides **I** Against Protozoa *Colpoda steinii*

Com-pound	Concentration, $\mu\text{g/mL}$											
	1000	500	250	125	62.5	31.3	15.6	7.8	3.9	2.0	1.0	0.5
Ia	+	+	+	+	-	-	-	-	-	-	-	-
Ib	+	+	+	+	-	-	-	-	-	-	-	-
Id	+	+	+	+	-	-	-	-	-	-	-	-
Ie	+	+	+	+	+	-	-	-	-	-	-	-
If	+	+	+	+	-	-	-	-	-	-	-	-
Ig	+	+	+	+	+	-	-	-	-	-	-	-
Ih	+	+	+	+	+	-	-	-	-	-	-	-
Ii	+	+	+	+	-	-	-	-	-	-	-	-
Baycox	+	+	+	+	+	-	-	-	-	-	-	-

* Measurements from two parallel tests are given for each compound; “+” denotes the death of all protozoa.

EXPERIMENTAL BIOLOGICAL PART

Antimicrobial activity of **Ia-i** was studied using standard strains *S. aureus* ATCC 25923 and *E. coli* ATCC 25922 and literature methods [16, 17] for double serial dilutions in Luria—Bertani (LB) liquid agar growth medium. A suspension of bacteria (2 mL) at a concentration of 5×10^5 microbes/mL was placed into tubes with a solution (2 mL) of the test compound at various concentrations. Thus, the calculated microbe load was 250,000 microbes per mL. The tubes were incubated in a thermostat for 18 h at 37°C. The controls were tubes containing growth medium with bacteria at a concentration of 250,000 microbes per mL (positive control). The negative controls were tubes with growth medium (without bacteria). The test organisms were bacteria strains *E. coli* 078 (field strain) and *S. aureus* P-209. Table 2 lists the antibacterial activity of amides **I**.

Protistocidal activity of amides **I** against the protozoa *C. steinii* was studied using the previously developed method [18].

The activity of the compounds was determined in decreasing dilutions from 1000.0 to 0.5 µg/mL with a constant load of 3-day protozoa culture. The viability control of the culture was medium used to prepare the dilutions. The reference drug was Baycox (Bayer AG), which is a widely used anticoccidial. Table 3 presents the protistocidal activities.

The antibacterial activity of amides **I** was significantly less than that of the reference drugs and the 1-β-aryloxyethyl-2-iminobenzimidazolines reported previously [13]. Amides **Ie** and **Ig-i** were the most active against *S. aureus*; amide **If**, which contained a 3-aryloxyethyl substituent, against *E. coli*. Amide **Ig** with a *p*-chlorophenoxyethyl group had the most pronounced protistocidal activity, which was at the level of the known protistocidal drug Baycox (Bayer AG).

Thus, the results led to the conclusion that phenoxyethyl-(benzyl)-substituted 2-iminobenzimidazolines were promising for discovering antimicrobial drugs, including for veterinary use, with combined (antibacterial and protistocidal) activity for treating diseases caused by simultaneous infection by bacteria and protozoa.

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