

# Comparison of the Active Species in the RF and Microwave Flowing Discharges of $N_2$ and Ar–20 % $N_2$

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**Abstract** We report a detailed comparison between RF and microwave (HF) plasmas of  $N_2$  and Ar–20 %  $N_2$  as well as in the corresponding afterglows by comparing densities of active species at nearly the same discharge conditions of tube diameter (5–6 mm), gas pressure (6–8 Torr), flow rate (0.6–1.0 slm) and applied power (50–150 W). The analysis reveals an interesting difference between the two cases; the length of the RF plasma ( $\sim 25$  cm) is measured to be much longer than that of HF (6 cm). This ensures a much longer residence time ( $10^{-2}$  s) of the active species in the  $N_2$  RF plasma [compared to that ( $10^{-3}$  s) of HF], providing a condition for an efficient vibrational excitation of  $N_2(X, v)$  by (V–V) climbing-up processes, making the RF plasma more vibrationally excited than the HF one. As a result of high V–V plasma excitation in RF, the densities of the vibrationally excited  $N_2(X, v > 13)$  molecules are higher in the RF afterglow than in the HF afterglow. Destruction of  $N_2(X, v)$  due to the tube wall is estimated to be very similar between the two system as can be inferred from the  $\gamma_v$  destruction probability of  $N_2(X, v > 3–13)$  on the tube wall ( $2–3 \times 10^{-3}$  for both cases) obtained from a comparison between the density of  $N_2(X, v > 3–9)$  in the plasmas to that of the  $N_2(X, v > 13)$  in the long afterglows. Interestingly enough, densities of N-atoms and  $N_2(A)$  metastable molecules in the afterglow regions, however, are measured to be very similar with each other. The measured lower density of  $N_2^+$  ions than expected in the HF afterglow is rationalized from a high oxygen impurity in our HF setup since  $N_2^+$  ions are very sensitive to oxygen impurity.

**Keywords**  $N_2$  RF and HF plasmas ·  $N_2$  RF and HF afterglows · N density ·  $N_2(A)$  and  $N_2(X, v > 13)$  density ·  $N_2^+$  density · Wall destruction probability

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## Introduction

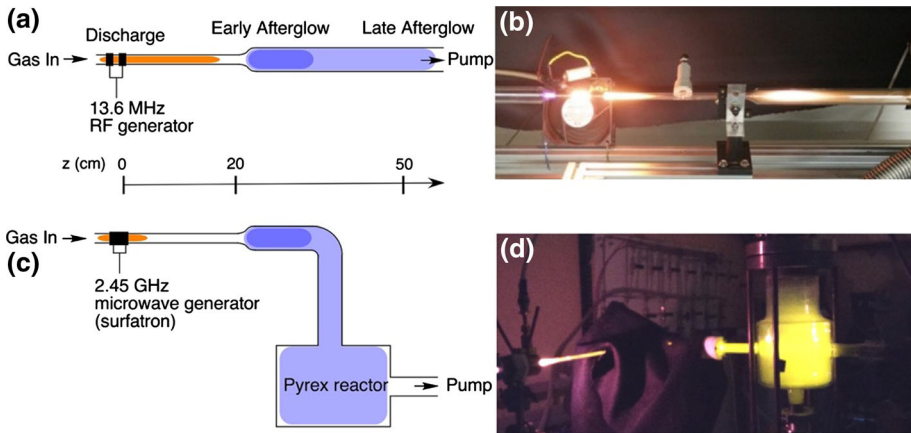
$N_2$  plasmas and corresponding afterglows have been intensively studied [1–5] with an interest in utilizing excited nitrogen species in various applications such as plasma sterilization [6–8], plasma cleaning [9–11] and surface nitridation [12–14]. Under a controlled flowing condition, an inherent nature of a longer lifetime of a certain excited  $N_2$  species makes it possible to obtain an ‘afterglow’ region with a varying density of N-atoms,  $N_2(X, \nu > 13)$  and  $N_2(A)$  metastable molecules and  $N_2^+$  ions [1]. This condition may provide a chance of developing a ‘damage-free’ plasma process in diverse application fields [7, 14].

In this respect, the exact determination of active species density in the plasma as well as in the afterglow would be a first step toward a successful use of them in application. RF and microwave (HF) plasmas in  $N_2$  and Ar– $N_2$  have been previously studied and the detailed densities of active species in the afterglow under a flowing condition have been carefully measured at two different locations of Ajou University and University of Toulouse [1, 2]. The densities of N-atoms,  $N_2(X, \nu > 13)$  and  $N_2(A)$  metastable molecules and  $N_2^+$  ions were determined using the line ratio method [1, 15] and the calibration of the N-atom density by NO titration [3, 4, 16]. These recent studies revealed important characteristics of the two plasma systems, especially in terms of plasma length, plasma temperatures and the changing densities of active species in the plasmas and the afterglows, which were explained as a result of complex kinetic processes of the active species in the  $N_2$  plasmas [1, 2, 4, 5, 16–21]. Recently, we have also reported detailed variation of active species in the RF plasmas and their afterglows of  $N_2$  and Ar–20 % $N_2$  [5]. Despite such vast reports on the detailed characteristics of the  $N_2$  and Ar– $N_2$  plasmas, a comprehensive comparison of plasma systems with different sources of plasma power (such as RF and HF) is still missing in the literature, while it can be an invaluable information for choosing the right source of plasma power in specific applications such as medical instruments sterilization [7] and iron nitriding [14] in  $N_2$  afterglows. Luckily enough, we have accumulated enough experimental data set for a quantitative comparison of the two plasma systems of RF and HF for the case of  $N_2$  plasma obtained under very similar flowing conditions. Based on this motivation, here we present our recent analysis on the differences between the two plasmas (RF and HF) and their afterglows in terms of plasma rotational and vibrational temperatures and densities of the active nitrogen species in the respective afterglows of mixed pink afterglow (PA) and late afterglow (LA). The afterglows are compared at the residence times of  $7 \times 10^{-2}$  s for the RF and  $10^{-3}$  to  $6 \times 10^{-2}$  s for the HF. We also report the destruction probability of vibrationally excited  $N_2(X, \nu)$  molecules on the RF and HF afterglow tube walls.

## Experimental Setups

The two experimental setups of RF and HF plasmas in Ajou and Toulouse, respectively, are reproduced in Fig. 1 for comparison.

The discharge quartz tubes have nearly the same inner diameters (ID) of 6 mm for the RF plasma reactor and 5 mm for the HF plasma reactor and have a length of 30 cm in both cases. In the post-discharge region, the RF discharge tube is connected to a straight quartz tube of 21 mm ID while the HF discharge tube is connected to a bent quartz tube of 18 mm ID itself connected to a 5 L-Pyrex chamber (Fig. 1). For both systems, experiments were conducted at very similar operation conditions of pressure (8 Torr), flow rate (0.5–1.0 slm)



**Fig. 1** Schematic diagrams and their pictures of the RF (a, b) and the HF (c, d) set-ups in Ajou and Toulouse, respectively, are shown. In the RF (HF) set-up, the inner diameters of the quartz tubes are 6 (5) mm in the discharge region and 21 (18) mm in the afterglow region

and applied power (50–150 W). The emission spectra across the RF reactor are obtained by means of a spectrometer (Monera 500, resolution 0.2–0.8 nm) with a PMT (Hamamatsu R928) using an optical fiber [1]. Across the HF reactor, an optical fiber is connected to an Acton Spectra Pro 2500i spectrometer (grating 600 g/mm) equipped with a Pixis 256E CCD detector (front illuminated 1024 × 256). The above similarities in the geometry of the two plasma systems allow us to compare the two RF and HF plasmas and afterglows to further understand the inherent differences in the type of plasma between the two plasmas. Plasmas are investigated by emission spectroscopy by measuring emission between the two rings of the RF discharge for the case of RF plasma and near the gap of the surfatron cavity (with a spatial resolution estimated to be 5 mm) for the case of HF plasma, respectively.  $N_2$  second positive (2nd pos),  $N_2^+$  first negative (1st neg) and  $N_2$  first positive (1st pos) intensities are recorded at these two positions.

## The $N_2$ and Ar– $N_2$ RF and HF Plasmas

In the same ranges of gas pressure (6–8 Torr), flow rate (0.6–1.0 slm) and power (50–100 W), it is observed that the plasma region, defined as the most luminous glow downstream of the RF coils and HF gap, respectively, is longer in RF than in HF. For pure  $N_2$ , the HF plasma length ( $L_p$ ) at 8 Torr increases from 1 to 6 cm when the power is increased from 50 to 100 W, while in the RF plasma,  $L_p$  varies from 18 to 25 cm for the same power variation. Such an increase of  $L_p$  of 2.45 GHz microwave plasmas in HF cavity at 100–140 W has been effectively observed in  $N_2$  plasma at 30 Torr with  $L_p$  increasing from 2.5 to 8 cm [13]. The difference is even more pronounced in Ar– $x\%N_2$  gas mixtures in which  $L_p$  increases with increasing Ar ratio.  $L_p$  is longer than the tube length (30 cm) for the RF plasma in Ar– $x\%N_2$  when  $x$  is lower than 20 %. With the HF plasma, it is even possible to decrease  $x$  down to 2 % since the HF plasma length is still confined in the tube of 30 cm length. Thus, the comparison between RF and HF is presently limited to results in pure  $N_2$  and in the Ar–20 % $N_2$  gas mixture.

For low pressure plasmas, it is known that the (minimum) cut-off electron density required to maintain a surface wave discharge depends on the square of the frequency ( $f^2$ ) of the electro-magnetic wave transferring the energy to the gas [22]. It is mentioned that the  $f^2$  variation of electron density at the plasma end is given in [22] for Ar surface wave plasma at pressure less than 1 Torr. It is extended here to the present RF (13.6 MHz) and HF (2450 MHz) plasma columns to explain our observation that the length of the RF plasma is longer than the HF one at a same gas pressure and transmitted power.

The plasma length ( $L_p$ ) is of crucial importance for the  $N_2(X, v)$  excitation by the electron–vibration (e–V), vibration–vibration (V–V) and vibration–translation (V–T) processes as previously reported and applied to results of a  $N_2$  DC flowing discharge [23]. At the end of the RF plasma ( $L_p = 30$  cm, 100 W), the residence time ( $t_R$ ) is  $10^{-2}$  s, one order of magnitude higher than at the end of the HF plasma where  $t_R = 8 \times 10^{-4}$  s ( $L_p = 6$  cm, 100 W). As a consequence, the (V–V) processes are more developed in the RF plasma than in HF one.

### The Plasma Gas Temperature

The intensity ratio (P1/P2) of the first two rotational sub-bands (labelled as P1 and P2) of the 1st pos  $N_2$  band at 775 nm is related to the rotational temperature ( $T_R$ ) of the plasma which is usually between 300 and 1000 K [19]. It is an important characteristic parameter of the  $N_2$  plasma since it is directly related to the gas temperature ( $T_g$ ), due to the efficient rotational-translational energy transfer. It is found that with an applied power between 50 and 100 W, the RF plasma at 6–8 Torr and 0.5–0.6 slm is a little warmer ( $T_g = 500$ –800 K) than the HF plasma at 8 Torr, 1 slm where  $T_g = 500$ –600 K. The uncertainty of gas temperature measurements is estimated to be 20 % [19].

### The Plasma Vibrational Temperature

The vibrational temperatures can be analyzed from the intensity ratios of the  $N_2$  2nd pos. sequence  $\Delta v = -2$  as detailed in [19]. The vibrational  $T_v(C)$  and characteristic  $\theta_{1-}(X)$  temperatures, corresponding to the excited  $N_2(C \ ^3\Pi_u)$  and  $N_2(X \ ^1\Sigma_g^+)$  ground states, respectively, are connected to each other as can be seen from the following two steps:

First, the  $f(v)$  vibrational distribution of  $N_2(X, v)$  is expressed as follows [24]:

$$f(v) = f(0) \cdot \exp^{-v \left( \frac{E_{10}}{\theta_{1(X)}} - \frac{(v-1)E_{10}\delta}{T_g} \right)} \quad (1)$$

with  $f(0) = 1 - \exp^{-\frac{E_{10}}{\theta_{1(X)}}$  for  $v < v_T$ , where  $v_T$  is the vibrational number of the Treanor minimum given by  $v_T = \frac{T_g}{280\theta_{1(X)}} + 0.5$ ,  $E_{10}$  is the energy difference between the first two vibrational levels  $E(X, 1) - E(X, 0)$  (in K) and  $\delta$  is the anharmonicity constant.

For  $N_2(X)$ ,  $E_{10} = 3396$  K and  $\delta = 6.22 \times 10^{-3}$ . The population of  $N_2(X, v)$  is then written as  $[X, v] = f(v) N_0$ , where  $N_0$  is the  $N_2$  density.

Second, by considering that the  $N_2(C, v')$  states are mainly produced by electron collisions on the  $N_2(X, v)$  ground states, following the Franck–Condon principle (vertical direct excitation by electron collisions), the stationary  $[C, v']$  population is written as follows:

$$[C, v'] = \sum_v [X, v] n_e k_e(C) q(X, v - C, v') / v(C, v') \quad (2)$$

where  $q(X, v - C, v')$  is the Franck–Condon factor [24],  $n_e$  is the electron density,  $k_e(C)$  is

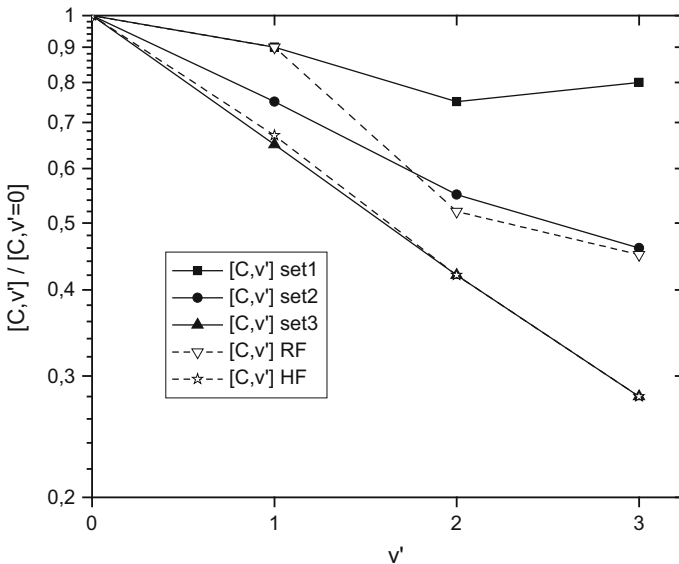
the electron excitation rate which depends on the electron excitation cross-sections and on the electron energy distribution and  $v(C, v') = v_R(C, v') + N_0 k_Q(C, v')$ , ( $v_R(C, v')$  and  $k_Q(C, v')$  are the radiative loss frequency [25] and the collisional quenching rate by the  $N_2$  molecules [26]), respectively. The thresholds between the  $N_2(C, v' = 0-3)$  and the  $N_2(X, v = 0-8)$  varying from 9 to 11.5 eV and the value of  $k_c(C)$  is approximated to be constant [27]. For the pressure range used in the present work,  $N_0 k_Q(C, v')$  can be neglected in front of  $v_R(C, v')$ .

As previously described [19], the experimental relative distribution  $[C, v' < 4]$  is obtained by measuring the band head intensities of the  $N_2$  2nd pos. sequence  $\Delta v = -2$ . If a Boltzmann distribution of  $[C, v']$  is verified, the vibrational temperature  $T_v(C)$  can be obtained.

The vibrational distribution of the  $N_2$  ground state  $[X, v]$  and the corresponding vibrational temperature  $\theta_1(X)$  are obtained from the experimental values of  $T_g$  and of the  $[C, v' < 4]$  distributions by combining Eqs. (1) and (2).

Figure 2 shows the calculated  $[C, v' < 4]$  distributions normalized to  $[C, v' = 0]$  for the  $\theta_1(X)$  values of 5, 8 and  $20 \times 10^3$  K which are compared with the experimental determined values ( $[C, v' < 4]/[C, v' = 0]$ ) for the RF and HF plasmas. From the comparison,  $\theta_1(X) = 8 \times 10^3$  K and  $\theta_1(X) = 5 \times 10^3$  K are chosen for the RF and HF plasmas, respectively. Note that the  $[C, v' < 4]$  calculated distributions have been found to be weakly influenced by the  $T_g$  variations between 500 and 800 K. It is observed a high value of the  $N_2(C, v = 1)$  density of the RF plasma which has been previously interpreted as a result of collisions between electrons and the  $N_2(A)$  metastable molecules [27, 28].

Table 1 reproduces the values of  $T_g$  (see “The plasma gas temperature” section),  $\theta_1(X)$ ,  $v_T$  and  $N_2(X, v = v_T)$  for the  $N_2$  and Ar-20 % $N_2$  plasmas at 6 Torr, 100 W for RF where  $[N_2]_{RF} = 7 \times 10^{16} \text{ cm}^{-3}$  and at 8 Torr, 100 W for HF where  $[N_2]_{HF} = 1.5 \times 10^{17} \text{ cm}^{-3}$ .



**Fig. 2** Vibrational distributions of  $N_2(C, v' < 4)$  versus  $v'$  for set1 [ $\theta_1(X) = 2 \times 10^4$  K], set2 [ $\theta_1(X) = 8 \times 10^3$  K] and set3 [ $\theta_1(X) = 5 \times 10^3$  K]. The experimental points are *triangles* for RF and *stars* for HF

**Table 1** Values of  $T_g$ ,  $T_v(C)$ ,  $\theta_1(X)$ ,  $\nu_T$  and  $N_2(X, \nu = \nu_T)$  for the RF plasma at 6 Torr, 100 W and for the HF plasma at 8 Torr, 100 W

Plasma gas	Discharge type	$T_g$ (K)	$\theta_1(X)$ ( $10^3$ K)	$\nu_T$	$N_2(X, \nu = \nu_T)$ ( $10^{15}$ cm $^{-3}$ )
N <sub>2</sub>	RF	800	8	9	4
N <sub>2</sub>	HF	500	5	8–9	3
Ar–20 %N <sub>2</sub>	RF	700	8	7	0.9
Ar–20 %N <sub>2</sub>	HF	600	6	8	0.8

We find a higher  $\theta_1(X)$  temperature in RF than in HF. This result agrees well with our suggestion of a more vibrationally excited RF plasma than the HF one.

## The N<sub>2</sub> and Ar–20 %N<sub>2</sub> RF and HF Afterglows

### Afterglows Characteristics

Characteristics of the afterglows can be studied by measuring the emission bands of N<sub>2</sub> 1st pos. (chosen band at 580 nm), N<sub>2</sub> 2nd pos. (chosen band at 316 nm) and N<sub>2</sub><sup>+</sup> 1st neg. (chosen band at 391.4 nm).

The analysis of the N<sub>2</sub> 1st pos. vibrational distribution reveals that the N<sub>2</sub> afterglows are a mixture of pink (PA) and late (LA) afterglows. In a pure LA, the 1st pos. emission is only due to the 3 body recombination of nitrogen atoms. To evaluate the contribution of the LA emission ( $a_{N+N} = 1$  for pure LA and  $a_{N+N} = 0$  for pure PA), the  $a_{N+N}$  value is obtained from a comparison between experimental and theoretical vibrational intensities distributions of the  $\Delta\nu = -4$  sequence (see [1–4, 16]).

For the experimental conditions of 8 Torr, 0.5 slm and 100 W in the RF set-up with the tube ID of 21 mm at  $z = 34$  cm (see Fig. 1), a pure N<sub>2</sub> PA is observed. It is followed by an increasing contribution of the LA with increasing  $z$ . At  $z = 51$  cm (17 cm downstream from the starting point of the 21 mm quartz tube), the  $a_{N+N}$  value of 0.5 is obtained at the same experimental conditions.

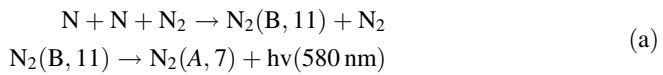
For the HF afterglow,  $a_{N+N}$  of 0.3 is obtained at 3 cm downstream from the starting point of the 18 mm quartz tube ( $z = 18$  cm, Fig. 2) at 8 Torr, 1 slm and 150 W. It increases up to 0.8 at  $z = 55$  cm. In this case, a pure LA is obtained just before the 5 L-reactor (see Fig. 1).

The measured larger  $z$  values for the RF afterglows (the mixed PA and LA of  $0 < a_{N+N} < 1$ ) than for the HF indicate longer afterglow times. For the RF afterglow, it is calculated to be  $7 \times 10^{-2}$  s at  $z = 51$  cm (Fig. 1) by assuming a laminar flow in the 21 mm quartz tube.

As for the HF afterglow, it is estimated to be  $1.5 \times 10^{-3}$  s at  $z = 18$  cm (Fig. 2) for a post-discharge jet with low radial expansion having the same gas velocity than in the discharge tube of 5 mm ID, as calculated in [29]. At  $z > 20$  cm, a laminar flow is assumed in the tube of dia. 18 mm, giving a residence time of  $6 \times 10^{-2}$  s at  $z = 55$  cm.

### Densities of N, N<sub>2</sub>(A), N<sub>2</sub>(X, v > 13) and N<sub>2</sub><sup>+</sup>

In the LA (a<sub>N+N</sub> = 1), the N-atoms recombine as follows [1–4, 16]:



When a<sub>N+N</sub> < 1 (that is, a mixture of LA and PA), the a<sub>N+N</sub> fraction of the measured I<sub>580</sub><sup>m</sup> intensity can be related to the square of the N-atom density as follows:

$$a_{\text{N+N}} I_{580}^m = k_3 [\text{N}]^2 \tag{3}$$

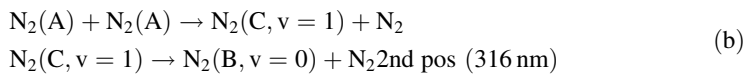
The calibration coefficient k<sub>3</sub> is obtained by the use of NO titration, which also gives an independent access to the N-atom density.

In the RF N<sub>2</sub> plasma, it is found that the N-atom density slowly increases from 1 to 2 × 10<sup>15</sup> cm<sup>-3</sup> as the RF power increases from 50 to 130 W under the condition of 8 Torr, 0.5 slm and z = 51 cm (residence time of 7 × 10<sup>-2</sup> s) where a<sub>N+N</sub> = 0.5. In Ar–20 %N<sub>2</sub>, the N-atom density is about 2 × 10<sup>15</sup> cm<sup>-3</sup> at 10–50 W.

For the HF case (see Table 2), the N-atom density slowly increases from 1.0 to 1.2 × 10<sup>15</sup> cm<sup>-3</sup> for the residence times between 1.5 × 10<sup>-3</sup> and 6 × 10<sup>-2</sup> s. Thus, it can be concluded that the N-atom density is about the same in the afterglow regions after the RF and HF plasmas considering the uncertainty on N-atom density to be 30 % [4].

Densities of the other active species (N<sub>2</sub>(A), N<sub>2</sub>(X, v > 13) and N<sub>2</sub><sup>+</sup>) are then obtained by the line-ratio method as developed in [1].

For the N<sub>2</sub>(A) density determination, the main N<sub>2</sub>(C) excitation is considered from the following pooling reaction:



Then, the following line-ratio is considered:

$$a_{\text{N+N}} I_{580}^m / I_{316}^m = k_4 ([\text{N}] / [\text{N}_2\text{A}])^2 \tag{4}$$

The k<sub>4</sub> coefficient is calculated from the rate coefficients of reactions (a) and (b) as reported in [1–4, 16] for the two RF and HF studied afterglows.

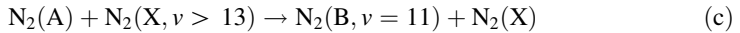
**Table 2** Active specie densities and γ<sub>v</sub> values after the N<sub>2</sub> HF and RF plasmas at 8 Torr, 1–0.5 slm and 100 W at various afterglow times

	HF			RF	
Afterglow time (10 <sup>-3</sup> s)	1.5	20	35	60	70
a <sub>N+N</sub>	0.3	0.55	0.45	0.8	0.5
N (10 <sup>15</sup> cm <sup>-3</sup> )	1.0	1.1	1.1	1.2	2
N <sub>2</sub> (A) (10 <sup>11</sup> cm <sup>-3</sup> )	1–5	1	1–2		2
N <sub>2</sub> (X, v > 13) (10 <sup>13</sup> cm <sup>-3</sup> )	6–10	8	7		20
N <sub>2</sub> <sup>+</sup> (10 <sup>8</sup> cm <sup>-3</sup> )	0.2–1.4	0.2	0.4		20
γ <sub>v</sub>		5 × 10 <sup>-3</sup>	3 × 10 <sup>-3</sup>		2 × 10 <sup>-3</sup>

It is concluded that a nearly constant value of  $[N_2(A)] = 1\text{--}2 \times 10^{11} \text{ cm}^{-3}$  is obtained as the RF power increases from 50 to 100 W in  $N_2$  and from 10 to 50 W in Ar–20 % $N_2$  at the residence time of  $7 \times 10^{-2}$  s.

The  $N_2(A)$  density at the residence time of  $1.5 \times 10^{-3}$  s is  $1\text{--}5 \times 10^{11}$  and  $5 \times 10^{12} \text{ cm}^{-3}$ , respectively, after the  $N_2$  and Ar–20 % $N_2$  HF plasmas under the condition of 8 Torr, 1 slm and 100–150 W. As shown in Table 2, the  $N_2(A)$  density is nearly constant for the residence times up to  $3.5 \times 10^{-2}$  s. By this method, the  $N_2(A)$  density is estimated to be in the range  $10^{11}\text{--}10^{12} \text{ cm}^{-3}$ .

Then, the  $N_2(X, \nu > 13)$  density is deduced by considering that the  $N_2(B, \nu = 11)$  state in the PA is produced by the following main reaction:



The ratio of  $a_{N+N} I_{580}^m$  (LA) and  $(1 - a_{N+N}) I_{580}^m$  (PA) is as follows:

$$(a_{N+N}/1 - a_{N+N}) [A] [X, \nu > 13] k_c = [N]^2 k_a [N_2] \quad (5)$$

Here, the  $N_2(X, \nu > 13)$  density is determined from  $a_{N+N}$  values, the N-atoms and the  $N_2(A)$  densities.

A nearly constant  $[N_2(X, \nu > 13)]$  density of  $1\text{--}2 \times 10^{14} \text{ cm}^{-3}$  is obtained at 50–100 W for pure RF  $N_2$  and at 10–50 W for the Ar–20 % $N_2$  gas mixture.

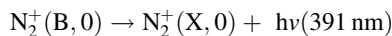
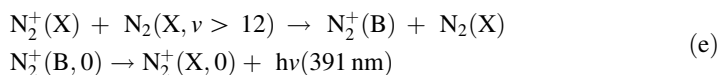
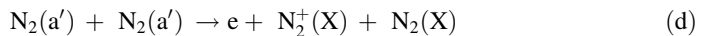
At the residence time of  $1.5 \times 10^{-3}$  s after the HF plasma (8 Torr, 1 slm and 100–150 W), the  $[N_2(X, \nu > 13)]$  density is  $5\text{--}6 \times 10^{13} \text{ cm}^{-3}$  for both  $N_2$  and Ar–20 % $N_2$ .

Such  $[N_2(X, \nu > 13)]$  density can be considered as an estimated value which can vary depending on the  $k_c$  rate coefficient.

The measured N-atoms and  $N_2(X, \nu > 13)$  densities above are in the same order of magnitude with the values reported in [14] in which  $[N_2(X, \nu > 4)] = 3 \times 10^{13} \text{ cm}^{-3}$  and  $[N] = 6 \times 10^{14} \text{ cm}^{-3}$  are obtained in  $N_2$  and Ar– $N_2$  afterglow of HF plasmas at 7 Torr, 3 slm and 200 W.

It should be noted that the  $N_2(X, \nu > 13)$  densities found in the present RF and HF afterglows are about one order of magnitude lower than the  $N_2(X, \nu = 7\text{--}9)$  densities in the corresponding plasmas (see Table 1). This result is used to determine the rate of destruction of these vibrational molecules on the tube wall in the following section.

For the  $N_2^+$  ions, the following dominant reactions in PA are considered [1]:



where reaction (d) is for the Penning ionization in the afterglow and reaction (e) indicates the excitation of  $N_2^+(X)$  by vibrational molecules  $N_2(X > 12)$ .

The  $N_2^+$  density is deduced from the  $N_2(X, \nu > 12)$  and  $N_2(A)$  densities by considering the line-ratio  $I_{391}/I_{316}$  from reactions (e) and (b):

$$I_{391}/I_{316} = k_5 \left( [N_2^+] [N_2(X, \nu > 12)] / [N_2(A)]^2 \right), \quad (6)$$

The  $k_5$  coefficient is calculated from the rate coefficients of reactions (e) and (b) [4, 16].

By taking  $[N_2(X, \nu > 12)] = [N_2(X, \nu > 13)]$ , it is found that the  $N_2^+$  density increases from  $5 \times 10^8$  to  $2 \times 10^9 \text{ cm}^{-3}$  for the case of the RF  $N_2$  plasma, when the RF power is



raised from 70 to 100 W. In Ar–20 %N<sub>2</sub> RF gas mixture, the N<sub>2</sub><sup>+</sup> ion density remains at  $5 \times 10^8 \text{ cm}^{-3}$  at 10–50 W. After the HF afterglows, the N<sub>2</sub><sup>+</sup> density is in the range  $(0.2\text{--}1.4) \times 10^8 \text{ cm}^{-3}$  as reported in Table 2.

The observed strong variations of the N<sub>2</sub><sup>+</sup> density originate from the fact that it is very sensitive to air impurity. By applying the intensity line-ratio method in comparison of I<sub>320</sub> of the NO<sub>β</sub> band to a<sub>N+N</sub> I<sub>580</sub>, an O-atom density of  $9 \times 10^{12} \text{ cm}^{-3}$  (that is, 30 ppm at 8 Torr) and N<sub>2</sub><sup>+</sup> density of  $10^9 \text{ cm}^{-3}$  have been previously determined in the RF afterglow [1]. In the HF afterglow, the O-atom density is in the order of  $10^{14} \text{ cm}^{-3}$  (that is, 400 ppm or  $4 \times 10^{-4}$  [O]/[N<sub>2</sub>] ratio) and the N<sub>2</sub><sup>+</sup> density is  $10^8 \text{ cm}^{-3}$ . These results clearly show that N<sub>2</sub><sup>+</sup> ions produced in the afterglow by the Penning ionization N<sub>2</sub>(a') + N<sub>2</sub>(a') reaction are strongly correlated to the amount of O-atom impurity.

### Destruction Probability of N<sub>2</sub>(X, v > 3–13) on the Tube Walls

Densities of the N<sub>2</sub>(X, v > 3) vibrational levels are found to be of 3 and  $4 \times 10^{15} \text{ cm}^{-3}$  in the respective RF and HF plasmas and about  $10^{14} \text{ cm}^{-3}$  for N<sub>2</sub>(X, v > 13) in the afterglows. By assuming a “plateau” of N<sub>2</sub>(X, v > v<sub>T</sub>), which means that a nearly constant vibrational density is reached after the Treanor minimum with v<sub>T</sub> = 3 and before a drop of high N<sub>2</sub>(X, v) density by (V–T) vibration–translation reactions, typically at v > 35 as described in [30], we evaluate a γ<sub>v</sub> destruction probability of the N<sub>2</sub>(X, v > 3–13) vibrational levels on the tube walls.

The same procedure as in [31] for N-atoms is taken here for the calculations of the γ<sub>v</sub> recombination probability of the N<sub>2</sub>(X, v > 3–13) molecules on the tube wall.

It is written as follows:

$$[X, v > 3 - 13]_z = [X, v > 3 - 13]_{z=0} \cdot \exp[-v_{X,v}z/v], \quad (7)$$

where  $v_{X,v} = \gamma_v \langle c \rangle / 2R$ ,  $\langle c \rangle$  is the thermal gas velocity ( $\langle c \rangle = 5 \times 10^4 \text{ cm s}^{-1}$ ) and R the tube radius.

$z = 0$  is taken at the RF and HF plasma end, which is at  $z = 25 \text{ cm}$  in RF and at  $z = 6 \text{ cm}$  in HF, respectively.

For such positions, the flow velocity in the RF set-up can be directly calculated from the gas flow rate and from the tube i.d. (21 mm) for a laminar flow.

From Eq. 7, it is found  $\gamma_v = 2 \times 10^{-3}$  after the RF plasma at  $z = 25 \text{ cm}$  (residence time of  $7 \times 10^{-2} \text{ s}$ ). At  $z > 20 \text{ cm}$  after the HF plasma, it is obtained:  $\gamma_v = 5 \times 10^{-3}$  at the residence time of  $2 \times 10^{-2} \text{ s}$  and  $\gamma_v = 3 \times 10^{-3}$  at the residence time of  $3.5 \times 10^{-2} \text{ s}$  where a laminar flow is expected.

We note that the same range of γ<sub>v</sub> values are obtained after the RF and HF plasmas (γ<sub>v</sub> = 2–5 × 10<sup>−3</sup>) at long times (7 and 3.5 × 10<sup>−3</sup> s) when the parts of LA and PA are in equality.

The present γ<sub>v</sub> values are about 2–5 times higher than the previous published value [32] for the first N<sub>2</sub>(X, v = 1) state: γ<sub>1</sub> = 10<sup>−3</sup> on Pyrex and quartz tube walls. They are largely higher than the γ<sub>N</sub> values of N-atoms recombination on quartz and Pyrex tube walls: γ<sub>N</sub> = 10<sup>−5</sup>–10<sup>−4</sup> [31]. Such results could come from vibrational relaxation of the N<sub>2</sub>(X, v) molecules on the tube wall by N<sub>2</sub>(X, v) + wall → N<sub>2</sub>(X, v–n) + wall, with n ≥ 1, faster than for the N-atoms with N + wall → 1/2 N<sub>2</sub> + wall.

The loss frequency of the N<sub>2</sub>(X, v > 3–13) on the tube wall is in the range of 50–150 s<sup>−1</sup>.

It can be compared to the following volume losses:  $10 \text{ s}^{-1}$  for the reaction  $\text{N}_2(\text{X}, \nu > 13) + \text{O}$  with a rate of  $10^{-13} \text{ cm}^3 \text{ s}^{-1}$  [33] with  $[\text{O}] = 10^{14} \text{ cm}^{-3}$ ;  $1\text{--}5 \text{ s}^{-1}$  for the reaction  $\text{N}_2(\text{X}, \nu) + \text{N} \rightarrow \text{N}_2(\text{X}, \text{w}) + \text{N}$ , with a rate  $k(\nu = 10\text{--}15, \text{w} = 0, 15) = 1\text{--}5 \times 10^{-15} \text{ cm}^3 \text{ s}^{-1}$  [34] and  $[\text{N}] = 10^{15} \text{ cm}^{-3}$ .

As for the (V–V) and (V–T) collisions, it is given in [35] the characteristic frequencies at 5 Torr, 500 K:  $\nu(\text{V–V}) = 2 \text{ } 10^3 \text{ s}^{-1}$  for  $\text{N}_2(1) + \text{N}_2(0) \rightarrow \text{N}_2(0) + \text{N}_2(1)$  and  $\nu(\text{V–T}) = 4 \text{ } 10^{-3} \text{ s}^{-1}$  for  $\text{N}_2(1) + \text{N}_2 \rightarrow \text{N}_2(0) + \text{N}_2$ .

Outside the high V–V frequency, these frequencies are low in comparison to the  $\text{N}_2(\text{X}, \nu > 3\text{--}13)$  loss frequency on the tube wall:  $50\text{--}150 \text{ s}^{-1}$ .

## Conclusion

To conclude, the presently analyzed  $\text{N}_2$  RF and HF flowing plasmas with nearly the same discharge parameters (tube diameters, gas pressure, flow rates and electrical power) produce long time afterglows with about the same N atoms and  $\text{N}_2(\text{A})$  densities. The main difference between RF and HF appears in the plasma length, which is longer in RF than in HF. As a consequence, the residence time of the active species is increased in the RF plasma, allowing in comparison to HF plasmas an enhancement of  $\text{N}_2(\text{X}, \nu)$  excitation by (V–V) collisions. Such an effect is detected as a higher density of vibrationally excited  $\text{N}_2(\text{X}, \nu)$  molecules in the RF plasma and particularly in the afterglow where a pure pink afterglow is observed in RF, ‘not in HF’.

If about equal densities of N and  $\text{N}_2(\text{A})$  are found in the long time RF and HF afterglows, it appears a higher  $\text{N}_2(\text{X}, \nu > 13)$  and  $\text{N}_2^+$  densities in the RF afterglow which can be related to high  $\text{N}_2(\text{V–V})$  excitation until the plasma end. The present results concern first the N-atom density calibrated by NO titration with an accuracy of 30 %. Then the other active species densities:  $\text{N}_2(\text{A})$ ,  $\text{N}_2(\text{X}, \nu > 13)$  and  $\text{N}_2^+$  are obtained by a line-ratio method, giving the order of magnitude of density, depending on kinetic rates of the chosen dominant reactions.

A new result obtained with these RF and HF afterglows concerns the  $\gamma_\nu$  destruction probability of the  $\text{N}_2(\text{X}, \nu \geq 3\text{--}13)$  vibrational levels on the quartz tube wall, which is more than one order of magnitude higher than for the N-atoms, certainly as the result of a more easy climbing down of  $\text{N}_2(\text{X}, \nu)$  on the tube wall.

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## References

1. Ricard A, Oh SG, Guerra V (2013) Line-ratio determination of atomic oxygen and  $\text{N}_2$  metastable absolute densities in an RF nitrogen late afterglow. *Plasma Sources Sci Technol* 22(3):035009. doi:10.1088/0963-0252/22/3/035009
2. Zerrouki H, Ricard A, Sarrette JP (2013) Determination of N and O-atom and  $\text{N}_2(\text{A})$  metastable molecule densities in the afterglows of  $\text{N}_2$  and  $\text{N}_2\text{--O}_2$  microwave discharges. *Contrib Plasmas Phys* 53(8):599–604. doi:10.1002/ctpp.201300008
3. Ricard A, Oh SG (2014) Densities of active species in  $\text{N}_2$  and  $\text{N}_2\text{--H}_2$  RF pink afterglow. *Plasma Sources Sci Technol* 23(4):045009. doi:10.1088/0963-0252/23/4/045009

4. Zerrouki H, Ricard A, Sarrette JP (2014) Determination of N and O-atoms and N<sub>2</sub>(A) metastable molecule densities in the afterglows of N<sub>2</sub>-H<sub>2</sub>, Ar-N<sub>2</sub>-H<sub>2</sub> and Ar-N<sub>2</sub>-O<sub>2</sub> microwave discharges. *Contrib Plasmas Phys* 54(10):827–837. doi:[10.1002/ctpp.201400001](https://doi.org/10.1002/ctpp.201400001)
5. Ricard A, Oh SG, Jang J, Kim YK (2015) Quantitative evaluation of the densities of active species of N<sub>2</sub> in the afterglow of Ar-embedded N<sub>2</sub> RF plasma. *Curr Appl Phys* 15(11):1453–1462. doi:[10.1016/j.cap.2015.08.013](https://doi.org/10.1016/j.cap.2015.08.013)
6. Philip N, Saoudi B, Crevier MC, Moisan M, Barbeau J, Pelletier J (2002) The respective roles of UV photons and oxygen atoms in plasma sterilization at reduced gas pressure: the case of N<sub>2</sub>-O<sub>2</sub> mixtures. *IEEE Trans Plasma Sci* 30(4):1429–1436. doi:[10.1109/TPS.2002.804203](https://doi.org/10.1109/TPS.2002.804203)
7. Villegier S, Sarrette JP, Ricard A (2005) Synergy between N and O atom action and substrate surface temperature in a sterilization process using a flowing N<sub>2</sub>-O<sub>2</sub> microwave post discharge. *Plasma Process Polym* 2(9):709–714. doi:[10.1002/ppap.200500040](https://doi.org/10.1002/ppap.200500040)
8. Pointu A-M, Ricard A, Dodet B, Odic E, Larbre J, Ganciu M (2005) Production of active species in N<sub>2</sub>-O<sub>2</sub> flowing post-discharges at atmospheric pressure for sterilization. *J Phys D Appl Phys* 38(12):1905–1909. doi:[10.1088/0022-3727/38/12/009](https://doi.org/10.1088/0022-3727/38/12/009)
9. Ricard A, Jaoul C, Gaboriau F, Gherardi N, Villegier S (2004) Production of N, H, O, and C atoms in flowing microwave discharges. *Surf Coat Technol* 188–189:287–293. doi:[10.1016/j.surfcoat.2004.08.171](https://doi.org/10.1016/j.surfcoat.2004.08.171)
10. Ricard A, Monna V (2002) Reactive molecular plasmas. *Plasma Sources Sci Technol* 11(3A):A150. doi:[10.1088/0963-0252/11/3A/322](https://doi.org/10.1088/0963-0252/11/3A/322)
11. Molchan IS, Thompson GE, Skeldon P, Trigoulet N, Chapon P, Tempez A, Malherbe J, Lobo Revilla L, Bordel N, Belenguer P, Nelis T, Zahri A, Therese L, Guillot P, Ganciu M, Michler J, Hohl M (2009) The concept of plasma cleaning in glow discharge spectrometry. *J Anal At Spectrom* 24(6):734–741. doi:[10.1039/B818343K](https://doi.org/10.1039/B818343K)
12. Kaluri SR, Hess DW (1996) Nitrogen incorporation in thin oxides by constant current N<sub>2</sub>O plasma anodization of silicon and N<sub>2</sub> plasma nitridation of silicon oxides. *Appl Phys Lett* 69(8):1053–1055. doi:[10.1063/1.116928](https://doi.org/10.1063/1.116928)
13. Ricard A, Hubert J, Michel H (1992) Correlations between active plasma species and steel surface nitriding in microwave post-discharge reactors. In: Capitelli M, Gorse C (eds) *Plasma technology: fundamentals and applications*. Springer, Boston, pp 125–142. doi:[10.1007/978-1-4615-3400-6\\_9](https://doi.org/10.1007/978-1-4615-3400-6_9)
14. Ricard A, Czerwicz T, Belmonte T, Bockel S, Michel H (1999) Detection by emission spectroscopy of active species in plasma-surface processes. *Thin Solid Films* 341(1–2):1–8. doi:[10.1016/S0040-6090\(98\)01529-6](https://doi.org/10.1016/S0040-6090(98)01529-6)
15. Zhu X-M, Pu Y-K (2010) Optical emission spectroscopy in low-temperature plasmas containing argon and nitrogen: determination of the electron temperature and density by the line-ratio method. *J Phys D Appl Phys* 43(40):403001. doi:[10.1088/0022-3727/43/40/403001](https://doi.org/10.1088/0022-3727/43/40/403001)
16. Zerrouki H, Ricard A, Sarrette JP (2014) Determination of N and O-atoms, of N<sub>2</sub>(A) and N<sub>2</sub>(X, v > 13) metastable molecules and N<sup>2+</sup> ion densities in the afterglows of N<sub>2</sub>-H<sub>2</sub>, Ar-N<sub>2</sub>-H<sub>2</sub> and Ar-N<sub>2</sub>-O<sub>2</sub> microwave discharges. *J Phys Conf Ser* 550(1):012045. doi:[10.1088/1742-6596/550/1/012045](https://doi.org/10.1088/1742-6596/550/1/012045)
17. Kang N, Lee M, Ricard A, S-g Oh (2012) Effect of controlled O<sub>2</sub> impurities on N<sub>2</sub> afterglows of RF discharges. *Curr Appl Phys* 12(6):1448–1453. doi:[10.1016/j.cap.2012.04.009](https://doi.org/10.1016/j.cap.2012.04.009)
18. Boudam MK, Saoudi B, Moisan M, Ricard A (2007) Characterization of the flowing afterglows of an N<sub>2</sub>-O<sub>2</sub> reduced-pressure discharge: setting the operating conditions to achieve a dominant late afterglow and correlating the NO β UV intensity variation with the N and O atom densities. *J Phys D Appl Phys* 40(6):1694–1711. doi:[10.1088/0022-3727/40/6/019](https://doi.org/10.1088/0022-3727/40/6/019)
19. Britun N, Gaillard M, Ricard A, Kim YM, Kim KS, Han JG (2007) Determination of the vibrational, rotational and electron temperatures in N<sub>2</sub> and Ar-N<sub>2</sub> RF discharge. *J Phys D Appl Phys* 40(4):1022. doi:[10.1088/0022-3727/40/4/016](https://doi.org/10.1088/0022-3727/40/4/016)
20. Nassar H, Pellerin S, Musiol K, Martinie O, Pellerin N, Cormier JM (2004) N<sub>2</sub><sup>+</sup>/N<sub>2</sub> ratio and temperature measurements based on the first negative N<sub>2</sub><sup>+</sup> and second positive N<sub>2</sub> overlapped molecular emission spectra. *J Phys D Appl Phys* 37(14):1904–1916. doi:[10.1088/0022-3727/37/14/005](https://doi.org/10.1088/0022-3727/37/14/005)
21. Sadeghi N, Foissac C, Supiot P (2001) Kinetics of N<sub>2</sub>(A <sup>3</sup>Σ<sub>u</sub><sup>+</sup>) molecules and ionization mechanisms in the afterglow of a flowing N<sub>2</sub> microwave discharge. *J Phys D Appl Phys* 34(12):1779. doi:[10.1088/0022-3727/34/12/304](https://doi.org/10.1088/0022-3727/34/12/304)
22. Chaker M, Moisan M, Zakrzewski Z (1986) Microwave and RF surface wave sustained discharges as plasma sources for plasma chemistry and plasma processing. *Plasma Chem Plasma Process* 6:79–96
23. Massabieaux B, Plain A, Ricard A, Capitelli M, Gorse C (1983) Excitation of vibrational and electronic states in a glow discharge column in flowing N<sub>2</sub>. *J Phys B At Mol Phys* 16(10):1863. doi:[10.1088/0022-3700/16/10/021](https://doi.org/10.1088/0022-3700/16/10/021)

24. Gordiets B, Hamedov SS, Shelepin IA (1975) Vibrational kinetics of harmonic oscillators under essentially nonequilibrium conditions. *Sov Phys JETP* 40:640
25. Gilmore FR, Laher RR, Espy PJ (1992) Franck–Condon factors, r-centroids, electronic transition moments, and Einstein coefficients for many nitrogen and oxygen band systems. *J Phys Chem Ref Data* 21(5):1005–1107. doi:[10.1063/1.555910](https://doi.org/10.1063/1.555910)
26. Pancheshnyi SV, Starikovskaia SM, Starikovskii AY (2000) Collisional deactivation of  $N_2(C\ ^3\Pi_u, v = 0, 1, 2, 3)$  states by  $N_2, O_2, H_2$  and  $H_2O$  molecules. *Chem Phys* 262(2–3):349–357. doi:[10.1016/S0301-0104\(00\)00338-4](https://doi.org/10.1016/S0301-0104(00)00338-4)
27. De Benedictis S, Dilecce G (1995) Vibrational relaxation of  $N_2(C, v)$  state in  $N_2$  pulsed RF discharge: electron impact and pooling reactions. *Chem Phys* 192(2):149–162. doi:[10.1016/0301-0104\(94\)00370-P](https://doi.org/10.1016/0301-0104(94)00370-P)
28. Mavadat M, Ricard A, Sarra-Bournet C, Laroche G (2011) Determination of ro-vibrational excitations of  $N_2(B, v')$  and  $N_2(C, v')$  states in  $N_2$  microwave discharges using visible and IR spectroscopy. *J Phys D Appl Phys* 44(15):155207. doi:[10.1088/0022-3727/44/15/155207](https://doi.org/10.1088/0022-3727/44/15/155207)
29. Blanchard H, Sarrette JP, Villeger S, Baudel P, Ricard A (2004) Density of oxygen atoms in high pressure (10–50 torr) flowing microwave post-discharges for elastomer treatments. *Eur Phys J Appl Phys* 26(03):221–226. doi:[10.1051/epjap:2004036](https://doi.org/10.1051/epjap:2004036)
30. Capitelli M, Gorse C, Ricard A (1982) Non equilibrium dissociation and ionization of  $N_2$  in decaying plasmas. *J Phys (Paris)* 43:L417–L423
31. Villeger S, Sarrette JP, Rouffet B, Cousty S, Ricard A (2008) Treatment of flat and hollow substrates by a pure nitrogen flowing post discharge. *Eur Phys J Appl Phys* 42(01):25–32. doi:[10.1051/epjap:2007177](https://doi.org/10.1051/epjap:2007177)
32. Marinov D, Lopatik D, Guaitella O, Hubner M, Ionikh Y, Ropcke J, Rousseau A (2012) Surface vibrational relaxation of  $N_2$  studied by  $CO_2$  titration with time-resolved quantum cascade laser absorption spectroscopy. *J Phys D Appl Phys* 45(175201):33
33. Guerra V, Loureiro J (1997) Self-consistent electron and heavy particle kinetics in a low-pressure  $N_2-O_2$  glow discharge. *Plasmas Sources Sci Technol* 6:373–385
34. Armenise I, Capitelli M, Garcia E, Gorse C, Lagana A, Longo S (1992) Deactivation dynamics of vibrationally excited nitrogen molecules by nitrogen atoms. *Chem Phys Lett* 200:597–604
35. Capitelli M, Gorse C, Ricard A (1986) Coupling of vibrational and electronic energy distributions in discharge and post-discharge conditions. In: *Topics in current chemistry*, vol 39. Non equilibrium vibrational kinetics. Chapter 11, pp 315–337