ORIGINAL PAPER

# **Ameliorative Effects of Ferulic Acid Against Lead Acetate-Induced Oxidative Stress, Mitochondrial Dysfunctions and Toxicity in Prepubertal Rat Brain**

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Abstract Epidemiological evidence has shown higher susceptibility of Children to the adverse effects of lead (Pb) exposure. However, experimental studies on Pb-induced neurotoxicity in prepubertal (PP) rats are limited. The present study aimed to examine the propensity of ferulic acid (FA), a commonly occurring phenolic acid in staple foods (fruits, vegetables, cereals, coffee etc.) to abrogate Pb-induced toxicity. Initially, we characterized Pb-induced adverse effects among PP rats exposed to Pb acetate (1,000-3,000 ppm in drinking water) for 5 weeks in terms of locomotor phenotype, activity of 5-aminolevulinic acid dehydratase (ALAD) in the blood, blood Pb levels and oxidative stress in brain regions. Further, the ameliorative effects of oral supplements of FA (25 mg/kg bw/day) were investigated in PP rats exposed to Pb (3,000 ppm). Pb intoxication increased the locomotor activity and FA supplements partially reversed the phenotype, while the reduced ALAD activity was also restored. FA significantly abrogated the enhanced oxidative stress in cerebellum (Cb) and hippocampus (Hc) as evidenced in terms of ROS generation, lipid peroxidation and protein carbonyls. Further, Pb-mediated perturbations in the glutathione levels and activity of enzymic antioxidants were also markedly restored. Furthermore, the protective effect of FA was discernible in striatum in terms of reduced oxidative stress, restored cholinergic activity and dopamine levels. Interestingly, reduced activity levels of mitochondrial complex I in Cb and enhanced levels in Hc among Pb-intoxicated rats were ameliorated by FA supplements. FA also

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Department of Biochemistry and Nutrition, CSIR-Central Food Technological Research Institute (CFTRI), Mysore 570020, India decreased the number of damaged cells in cornu ammonis area CA1 and dentate gyrus as reflected by the histoarchitecture of Hc among Pb intoxicated rats. Collectively, our findings in the PP model allow us to hypothesize that ingestion of common phenolics such as FA may significantly alleviate the neurotoxic effects of Pb which may be largely attributed to its ability to abrogate oxidative stress.

**Keywords** Ferulic acid · Lead acetate · Oxidative stress · Mitochondrial dysfunctions · Hippocampus · Prepubertal (PP) rats

#### Abbreviations

ALAD	Aminolevulinic acid dehydratase
Pb	Lead
Cb	Cerebellum
St	Straitum
ROS	Reactive oxygen species
MDA	Malondialdehyde
HP	Hydroperoxides
AChE	Acetylcholinesterase
GSH	Reduced Glutathione
HPLC	High performance Liquid chromatography
DA	Dopamine

## Introduction

Pb is a ubiquitous and a non-essential toxic heavy metal widely distributed in the environment. Human exposure to Pb occurs via food, water, air and soil. It is also prevalent in occupational settings such as smelters and Pb manufacturing industries [1]. Despite improvements in public health policies and substantial reductions in blood Pb levels

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in adults, Pb exposure remains an important health problem worldwide. Pb is a systemic toxicant affecting virtually every organ system and notably, the central nervous system [2]. Nervous system is the primary target for the low levels of Pb-exposure, and the developing brain is highly vulnerable to Pb toxicity than the mature brain [3, 4]. Several mechanisms are implicated in the pathogenesis of Pbinduced neurotoxicity, including oxidative stress mitochondrial dysfunction and excitotoxicity [5]. Oxidative stress can induce brain injury via oxidative damage to critical biomolecules, such as lipids, proteins, and DNA [6].

Pb exposure is shown to cause generation of an excessive amount of ROS and alteration of defense system in animals and causes disruption of pro/antioxidant balance in the developing rat brain [7–9]. In addition, 5-aminolevulinic acid dehydratase (ALA-D), an enzyme of heme biosynthesis that is highly sensitive to lead [10] is inhibited by low blood lead levels. On the other hand, ALA-D inhibition contributes to ROS generation and is used as one of the most-reliable indicators of lead intoxication [11].

Phytochemicals are a heterogenous group of bioactive compounds that are extensively researched for their healthpromoting potentials in humans [12]. The protective effect of several antioxidant nutrients against Pb-neurotoxicity has been investigated in experimental animals. Antioxidants such as ascorbic acid ameliorate oxidative damage induced by maternal low-level lead exposure in the hippocampus (Hc) of rat pups during gestation and lactation [13]. Recent studies have demonstrated the protective effect of quercetin, a major bioflavonoid abundant in fruits and vegetables against Pbinduced oxidative stress and impairment of synaptic plasticity [14]. Likewise previously, Gallic acid, a natural hydrolysis product of tannins (found abundantly in grapes, berries, and other fruits as well as in wine) was shown to attenuate Pb-induced oxidative and locomotor damage in rats [15]. Green tea extract and their components are partially efficacious in protection, and preventing disturbances of antioxidant defense system in the brain of Pb treated rats [16]. Other natural flavonoids such as puerarin, gossypin and curcumin, are also reported to protect against Pb-induced perturbations in hepatic antioxidant system, reduce brain oxidative stress and memory deficits in rats [2, 17, 18].

Ferulic acid (trans-4-hydroxy-3-methoxycinnamic acid; FA), one of the most-abundant phenolic compounds in the human diet, is synthesized in plants by the metabolism of phenylalanine and tyrosine. FA is present in seed plants (rice, wheat and oat), vegetables (tomato and carrot), and fruits (pineapple and orange) and possess excellent free radical scavenging and antioxidant properties [19–21]. FA possesses high antioxidant potential owing to its resonance-stabilized phenoxy radical structure. In the context of neurological disease, intravenous administration of FA has been

shown to protect against neuronal cell death induced by cerebral ischemia [22, 23]. Interestingly, FA is also known to promote neural progenitor cell proliferation in vitro/ in vivo, ameliorate stress-induced depression-like behavior in mice [24] and improves behavioral impairments and Alzheimer-like pathology in transgenic Mice [25].

The ubiquitous distribution of FA and its pleiotropic biological activity entails basic understanding on the potential neuroprotective effects under chronic Pb-intoxication in experimental animals. Accordingly, we examined the hypothesis that short-term oral supplements of FA may significantly alleviate Pb-induced phenotype, oxidative stress and neurotoxicity employing a prepubertal (PP) rat model. Initially, we recapitulated some of the adverse effects of Pb in PP rats employing 5-week regimen. Subsequently, the efficacy of oral FA supplements to ameliorate oxidative stress response in brain regions (Cb, Hc and St), cholinergic function and dopamine (DA) levels were investigated among PP rats subjected to Pb intoxication at the higher dosage. Since Pb exposure is known to cause specific effects on hippocampal function, we also examined the potential of FA to modulate histological lesions in Hc.

## **Materials and Methods**

## Chemicals and Reagents

Thiobarbituric acid (TBA), 1,1,3,3-tetramethoxy propane, 2',7'-dichlorofluorescein (DCF), 2',7'-dichlorofluorescein diacetate (DCFH-DA), 5-aminolevulinic acid hydrochloride and lead (II) acetate trihydrate, Ferulic acid (4-Hydroxy-3-methoxycinnamic acid) were purchased from M/s Sigma Chemical Co. St. Louis, USA. All other chemicals used were of analytical grade.

## Animals and Care

Prepubertal (PP) Wistar rats (4 week-old,  $60 \pm 5$  g) drawn from the stock colony of our animal house facility were housed in rectangular polypropylene cages kept on racks built of slotted angles, in a controlled atmosphere with a 12 h light/ dark cycle. They were acclimatized for 3 days prior to the start of the experiment. The animals were maintained on a commercial powdered diet and tap water ad libitum. Studies were conducted strictly in accordance with approved guidelines by the "Institute Animal Ethical Committee" regulated by the committee for the purpose of control and supervision of experiments on animals (CPCSEA), Ministry of Environment and Forests, Government of India, India. Handling, as well as care of animals during sacrifice, was strictly according to the standard guidelines of the Institutional Ethics Committee (registration No. 49/1,999/CPCSEA).

# **Experimental Design**

*Study 1:* For this study, male PP rats were randomly assigned to four groups (n = 6), while the first one as served as control, rats of other groups received lead acetate (PbA) daily in deionized drinking water at concentrations of 1,000, 2,000 and 3,000 ppm for 5-weeks. Terminally rats of all groups were sacrificed.

*Study 2:* Male PP rats were randomly assigned into four experimental groups (n = 6) and were designated as follows: Control, ferulic acid (FA), Pb and Pb + FA (PbFA) groups.

(1) Control group received Pb-free deionized water during the whole course of the experiment; (2) FA treated group, considered as positive control received daily oral supplements of FA at (25 mg/kg bw/day); (3) Pb group, animals received lead acetate (PbA) 3,000 ppm in deionized drinking water and (4) Pb + FA treated group, animals received PbA in drinking water and oral FA supplements. The dosage selection of Pb was based on the first study and published works. Further the criterion of selection of FA dosage was based on our preliminary dose finding studies in rats [26]. The duration of the experiment was 5 weeks and terminally rats were sacrificed, and blood samples and brain regions were processed as follows. Prior to sacrifice, rats of all groups were subjected to behavioral assessment.

In both studies, blood samples were drawn by cardiac puncture. Brain was excised and placed in 0.9 % ice-cold NaCl solution and cleaned with the physiological saline solution to remove blood cells, blotted on filter paper. The brain was immediately stored at -80 °C for later use.

## Behavioral Assessment-Motor Activity

Spontaneous motor activity was assessed by placing rats in an open-field box (76 cm  $\times$  76 cm, divided by 8 lines into 25 squares). Each rat was individually placed in the centre of the open-field and allowed to explore it for 5 min. During this time, the motor activity was recorded in the open-field [27]. The number of squares crossed with all paws (crossing) and standing on legs (rearing) was evaluated during 5-min sessions. The crossing and rearing numbers were indicators of locomotor and exploratory activity, respectively. The arena was carefully cleaned with 70 % ethanol solution between animal tests to eliminate any olfactory cue derived from the previous rat located in the box.

# Markers of Pb Intoxication

## Activity of *D*-Aminolevulinic Acid Dehydratase in Blood

The activity of blood D-ALAD was assayed according to the procedure of [28]. The assay system consisted of 0.1 mL of

heparinized blood and 0.75 mL of distilled water. After 10 min of incubation at 37 °C for complete hemolysis, 0.5 mL of standard p-aminolevulinic acid (ALA) was added to the tubes and incubated for 60 min at 37 °C. The reaction was stopped after 1 h by adding 0.5 mL of trichloroacetic acid. To the clear supernatant, an equal volume of Ehrlich's reagent was added, and the absorbance was recorded at 555 nm after 5 min.

# Blood Pb Levels

Blood samples were collected in heparinized tubes via cardiac puncture of anesthetized rats. Blood was digested with concentrated nitric acid and hydrogen peroxide. Digests were diluted to a constant volume with ultra-pure Nitric acid (2N). Blood lead levels were determined following the procedure [29] with some modifications using inductively coupled plasma optical emission spectrometer (ICPOES, Spectro Analytical Instruments, Model Ultima 2, Horiba Jobin-Yvon S.A.S, France) with axial view configuration and cross flow nebulizer coupled to a double pass-Scott type nebulization spray chamber. Lead was determined at 168.215 and 320.353 nm, which corresponds to the mostintense lines for the measurement of this element. Both wavelengths were monitored simultaneously to investigate possible spectral interferences in the quantification of the analyte. The operating conditions of the ICP OES equipment were radiofrequency power set at 1400 W, argon used for plasma generation, and plasma gas flow rates set at 14, 1.00, and 0.90 L/min for principal, auxiliary and nebulizer gas, respectively. The calibration standards were prepared from a Pb standard solution by sequential dilution in 5 % nitric acid in the range of 0.1-1 µg/L. The operating conditions used for ICP OES were selected according to the manufacturer's recommendations. The samples were appropriately diluted in 5 % HNO3 (v/v) by the factor necessary for them to stay in the same concentration range of the calibration curve.

## Histopathologic Evaluation

Brain tissue was placed in 10 % formaldehyde for 2 h. The brains were removed and placed in a new-formaldehyde solution for 24 h before being dehydrated using ethanol (70 % for 24 h, 90 % for 1 h and 100 % for 1 h) then cleaned in xylene and embedded in paraffin. Coronal Sections (5  $\mu$ m thick) were cut with a microtome (Leica RM 2025, Germany), mounted on glass slides and stained with Hematoxylin and Eosin. The sections were analyzed by observing under light microscope (Leitz, Germany) at 20× magnification.

## **Biochemical Determinations**

Preparation of Cytosol and Mitochondria: Brain regions (Cb, Hc and St) were dissected over ice. A 10 % homogenate of was prepared in ice-cold Tris-sucrose buffer (0.25 M, pH 7.4) and centrifuged at 1,000*g* for 10 min at 4 °C to obtain the nuclear pellet. Differential centrifugation (700 and 4,500*g*, 10 min, 4 °C) was employed to isolate cytosol and mitochondria from different brain regions. The crude mitochondrial pellet was washed and resuspended using HEPES buffer (HEPES 10 mM, EDTA 0.1 mM, mannitol 200 mM, sucrose 70 mM, pH 7.4; [30]. All samples were stored in aliquots at -80 °C until next use.

Assessment of Lipid Peroxidation (LPO): LPO was assessed by measuring the formation of thiobarbituric acid reactive substances (TBARS) following the method described previously [31]. Briefly, the reaction mixture contained 0.2 mL of cytosol, 1.5 mL of acetic acid (pH 3.5, 20 %), 1.5 mL of 0.8 % TBA (0.8 % w/v) and 0.2 mL SDS (8 % w/v). The mixture was heated to boiling for 45 min and TBARS adducts were extracted into 3 mL of 1-butanol and its absorbance was measured at 532 nm and quantified as malondialdehyde (MDA) equivalents using 1, 1, 3, 3-tetramethoxypropane as the standard.

Measurement of ROS Generation: ROS generation in brain regions was assayed using dihydro dichlorofluorescein diacetate (H<sub>2</sub> DCFH-DA) as described previously [32]. The reaction mixture (1 mL) containing Locke's buffer (pH 7.4), 0.2 mL cytosol and 10  $\mu$ l of DCFH-DA (5  $\mu$ M) was incubated for 15 min at room temperature. After 30 min of further incubation, the formation of the fluorescent product DCF was measured in a spectrofluorimeter (excitation: 484 nm and emission: 530 nm). ROS formation was quantified from a DCF-standard curve and data are expressed as pmol DCF formed/min/mg protein.

Measurement of Hydroperoxide Levels: An aliquot of cytosol (100 µg protein) was added to 1 mL FOX reagent (100 µM xylenol orange; 250 µM ammonium ferrous sulphate; 100 µM sorbitol; 25 mM H<sub>2</sub>SO<sub>4</sub>) and incubated for 30 min at room temperature [33]. The mixture was centrifuged at  $600 \times g$ , and the supernatant was quantified at 560 nm in a spectrophotometer. Results were expressed as µmol hydroperoxide/mg protein.

*Protein Carbonyls:* Protein carbonyl content in the brain regions were determined as described earlier [34]. An aliquot of cytosol was incubated with 2, 4-dinitrophenyl hydrazine (DNPH) for 60 min at room temperature. Following precipitation by adding 20 % trichloroacetic acid, the pellet was washed with acetone and dissolved in 1 mL of Tris buffer containing sodium dodecyl sulphate (8 % w/v, pH 7.4). The absorbance was measured at 360 nm and expressed as nmol carbonyls/mg protein (€-22,000 nmol/cm).

*Determination of GSH:* GSH levels were quantified by a fluorimetric method [35]. An aliquot of cytosol was added to

1 mL formic acid (0.1 M) and centrifuged at  $10,000 \times g$  for 10 min. An aliquot of the supernatant was used for the assay. Concentration of GSH was calculated from the standard curve and the expressed as  $\mu g$  GSH/mg protein.

# Activities of Antioxidant Enzymes

Superoxide dismutase (SOD) activity was measured by monitoring the inhibition of quercetin autoxidation [36]. Glutathione reductase (GR) and Thioredoxin reductase (TRR) activity was measured as described previously [37, 38]. Glutathione-S-transferase (GST) activity was measured by monitoring the rate of enzyme–catalyzed conjugation of reduced glutathione with 1–chloro–2, 4–dinitrobenzene and expressed as µmol of conjugate formed per min per mg protein ( $\notin$ –9.6/mM/cm) [39]. The activity of glutathione peroxidase (GPx) was determined using t-butyl hydroperoxide as the substrate and activity expressed as nmol of NADPH oxidized per min per mg protein ( $\notin$ -6.22/mM/cm) [40].

# Mitochondrial Functions

*MTT Assay:* MTT reduction was determined as described previously [41]. Brain mitochondria were incubated in a Mannitol-Sucrose-HEPES buffer (pH 7.4) containing sodium succinate (20 mM) and 15  $\mu$ L MTT (5 mg/mL) at 37 °C for 60 min. The formazan crystals formed were dissolved in aqueous 10 % SDS-45 % DMF buffer (v/v, pH 7.4). The absorbance of the clear solution was measured at 570 nm and expressed as OD/mg protein.

Activity of ETC Enzymes: The activities of NADH-cytochrome C reductase (Complex I–III) and succinate-cytochrome C reductase (Complex II–III) were measured in brain mitochondria as described previously [42, 43]. Briefly mitochondrial protein (0.1 mg) was added to phosphate buffer (0.1 M, pH 7.4) containing NADH (0.2 mM) and KCN (1 mM). The reaction was initiated by the addition of 0.1 mM cytochrome C giving a final volume of 1 mL reaction mixture and decrease in absorbance was monitored for 3 min at 550 nm. The activity was expressed as nmol cytochrome C reduced/min/mg protein ( $\in$ -19.6/mM/cm). To determine the activity of succinate-cytochrome C reductase, the substrate succinate (20 mM) was used.

# Cholinergic Activity and Dopamine Levels

Acetylcholinesterase (AChE) activity and Butyrylcholinesterase (BChE): AChE and BChE activities were determined as described earlier [44]. To 1 mL reaction mixture containing phosphate buffer (0.1 M, pH 8.0), DTNB (10 mM), 50  $\mu$ L cytosol and 20  $\mu$ L acetylthiocholine

iodide-/butyrylthiocholine iodide (78 mM) were added, and change in absorbance was monitored for 3 min at 412 nm. The enzyme activities were expressed as nmoles of substrate hydrolyzed/min/mg protein.

Dopamine (DA) Estimation by HPLC: Dopamine (DA) levels in striatum were analyzed [45] using HPLC system (Model Shimadzu LC-10 Avp, UV-Visible detector (SPD-10 Avp), Pump (LC-10 ATvp), Shimadzu, Japan) linked to the injection valve with 20 µL sample loop (Model 7125, Rheodyne, CA, USA). Sample was injected to a reversedphase column (Make: 'Discovery' Supelco Sigma-Aldrich; C-18, 25 cm, 4.6 mm, 5 µm) accessorized with an ultraviolet detector (280 nm). Analyses were all performed at room temperature. The sample was eluted using the mobile phase 0.2 % aqueous trifluoroacetic acid and methanol (70:30, v/v) at the flow rate of 1 mL/min. The peak corresponding to DA would appear between 3.4 and 3.7 min. DA levels were calculated from an external standard curve as µg DA/mg protein, and data is represented as percent change.

# Determination of Protein

Protein concentration of cytosol and mitochondria was determined as described earlier [46] using bovine serum albumin as standard.

## Data Presentation and Statistical Analysis

Data expressed as mean  $\pm$  standard error (SE) for each experimental group were analyzed by one-way ANOVA followed by a post hoc 'Turkey' tests to compare the control and treatment groups; *P*-values <0.05 were considered as statistically significant. All statistical analysis was performed using SPSS statistical software package version 17.0.

## Results

Effect of Pb Acetate

## Effect on Locomotor Activity

The effect of Pb exposure (low dose -1,000 ppm, mid dose -2,000 ppm and high dose -3,000 ppm) on motor activity in the open field test is shown in Fig. 1. While Pb exposure at low and mid dose had no significant effect on the crossing and rearing activity in open field test, a marked increase in crossing (62 %) and a decrease in rearing (40 %) activity was evident at the highest dose (Fig. 1a, b).

#### Effect on Hb and ALAD Activity Levels

The Hb levels were not significantly affected at the low and mid dose of Pb, while only a marginal decrease was observed at the high dose (Fig. 2a). However, we could observe a significant diminution in the blood ALAD activity (65 %) at all the concentrations in relation to the control group (Fig. 2b).

# Effect of Pb per se on Oxidative Stress Markers

Pb intoxication resulted in consistent induction of oxidative stress in brain regions of PP rats as evidenced by enhanced generation of ROS levels. High dose induced a robust increase in reactive oxygen species formation in all regions (Cb-25 %; Hc-90 %; St-46 %). However, the increase in MDA levels were not consistent and showed no dose-dependency (Table 1). Further, high dose of Pb also caused only marginal depletion in reduced GSH levels in Hc and St, while significant increase in protein carbonyl levels was evident in all regions (Cb-26 %; Hc-19 % and St-30 %) excepting cortex (Table 2).

# Ameliorative Effects of Ferulic Acid

# Effect on Locomotor Activity

The locomotor activity measured as crossing activity among Pb exposed rats was only partially restored by FA supplements (Fig. 1c, d), while no effect was evident with respect to rearing activity.

Effect of FA on Hb and ALAD Activity and Blood Pb Levels

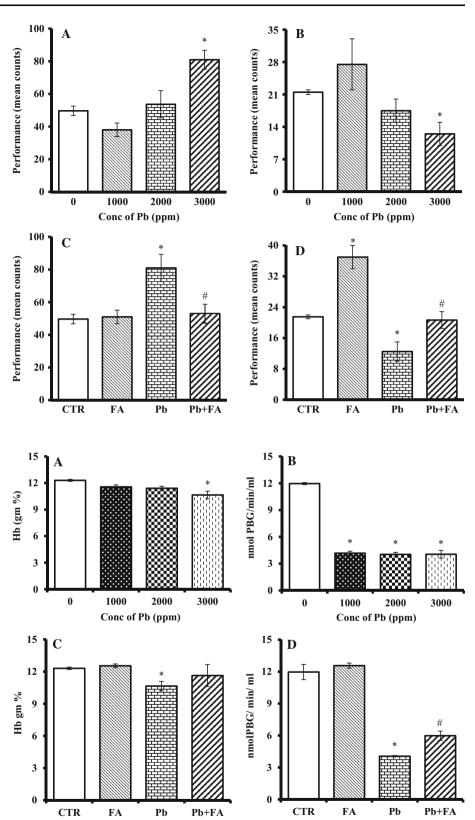
The modulatory effect of FA on Pb-induced alterations in blood Hb and ALAD activity are shown in Fig. 2. FA had no effect on the Pb-induced decrease in Hb levels (Fig. 2c). Pb exposure significantly decreased blood ALAD activity (65 %) in relation to the control group. FA alone had no effects on this variable, but the co-treatment (Pb + FA) partially reversed the effect (Fig. 2d). As anticipated rats exposed to Pb showed significantly higher blood levels of the metal than the controls. However, FA treatment had no significant effect on the blood Pb levels (Data not shown).

Ameliorative Effects of FA on Oxidative Markers Modulatory Effect on Oxidative Markers in Cerebellum and Hippocampus

Pb caused differential degree of oxidative stress in brain regions. While the extent of reactive oxygen species generation was marginal in Cb, it was robust (93 %) in Hc clearly suggesting the higher susceptibility of this brain region. Interestingly, the ROS levels in both regions were restored to normal levels by FA supplements (Fig. 3a).

Fig. 1 Effect of lead acetate (1,000-3,000 ppm in drinking water) in prepubertal rats evaluated as number of crossings (a) and rearing activity (**b**) in a open field box; Modulatory efficacy of Ferulic acid (FA) supplements on Pbinduced effect on crossing (c) and rearing (d) activity among rats exposed to Pb (3,000 ppm in drinking water). Values are mean  $\pm$  SE (n = 6); Data analyzed by one way ANOVA ( $P \le 0.05$ ) followed by Turkey's test (\* compared to control; \* compared to Pb 3,000), CTR-Control; Pb-lead acetate-3,000 ppm, FA-ferulic acid 25 mg/kg bw/day oral for 5 weeks

Fig. 2 Effect of lead acetate (1,000, 2,000 and 3,000 ppm in drinking water) on hemoglobin (a) and activity of aminolevulinic acid dehydratase-ALAD (b) in prepubertal rats; Modulatory effect of Ferulic acid (FA) supplements on hemoglobin level (c) and activity of ALAD (d) in prepubertal rats exposed to lead (Pb-3,000 ppm in drinking water). Values are mean  $\pm$  SE (n = 6); Data analyzed by one way ANOVA (P < 0.05) followed by Turkey's test. (\* compared to control; # compared to Pb 3,000), CTR-Control; Pb-lead acetate-3,000 ppm, FA-ferulic acid 25 mg/kg bw/day oral for 5 weeks



Further Pb exposure decreased the hydroperoxide levels in Cb (18 %) and increased the levels in Hc (18 %), while FA supplements normalized the levels (Fig. 3b). Furthermore,

Pb caused moderate elevation in MDA levels in both Cb (33 %) and Hc (30 %) while FA supplements completely normalized the MDA levels (Fig. 3c). Similarly, Pb caused

	Brain regions				
	Cortex	Cerebellum	Hippocampus	Striatum	
ROS <sup>a</sup>					
CTR	$8.71\pm0.2$	$13.55 \pm 0.2$	$7.56 \pm 0.1$	$6.65\pm0.1$	
Pb 1,000	$9.29\pm0.2$	$16.88 \pm 0.4*$	$9.97\pm0.4$	$10.32 \pm 0.0*$	
Pb 2,000	$9.04 \pm 0.4$	$17.75 \pm 0.5^{*}$	$8.82 \pm 0.8^{*}$	$8.35 \pm 0.1*$	
Pb 3,000	$8.39\pm0.3$	$16.77 \pm 0.2^{*}$	$14.61 \pm 0.8^*$	$9.22 \pm 0.5^{*}$	
MDA <sup>b</sup>					
CTR	$2.13\pm0.1$	$1.84 \pm 0.1$	$1.10 \pm 0.1$	$1.31\pm0.0$	
Pb 1,000	$3.07 \pm 0.1*$	$2.42 \pm 0.1^{*}$	$1.35 \pm 0.0^{*}$	$1.33\pm0.1$	
Pb 2,000	$2.21\pm0.1$	$2.28 \pm 0.1*$	$1.18\pm0.0$	$1.07 \pm 0.0*$	
Pb 3,000	$1.90 \pm 0.2^{*}$	$2.45 \pm 0.0^{*}$	$1.44 \pm 0.1^{*}$	$1.65 \pm 0.1*$	

 Table 1
 Induction of oxidative stress measured as ROS generation and lipid peroxidation in cytosol of brain regions in prepubertal rats exposed to lead (Pb) acetate in drinking water for 5 weeks

Values are mean  $\pm$  SE (n = 6)

Data analyzed by one-way ANOVA followed by post hoc Turkey test

\* P < 0.05. Significances were determined by making comparisons between CTR versus Pb 1,000, 2,000 and 3,000 ppm

a pmol DCF/min/mg protein

<sup>b</sup> nmol MDA/mg protein

Table 2         Status of reduced glutathione levels and protein carbonyls in cytosol of brain regions in prepubertal rats exposed to lead (Pb) acetate in
drinking water for 5 weeks

	Brain regions	Brain regions				
	Cortex	Cerebellum	Hippocampus	Striatum		
Reduced glutathione	a					
CTR	$3.72\pm0.1$	$3.85\pm0.1$	$3.41 \pm 0.1$	$3.95\pm0.0$		
Pb 1,000	$3.89 \pm 0.1$	$3.87\pm0.0$	$3.51\pm0.0$	$3.30\pm0.0$		
Pb 2,000	$3.89 \pm 0.1$	$3.87\pm0.0$	$3.26 \pm 0.2$	$3.42\pm0.3$		
Pb 3,000	$3.76 \pm 0.1$	$3.29 \pm 0.1*$	$3.08 \pm 0.0^{*}$	$3.28 \pm 0.0^{*}$		
Protein Carbonyls <sup>b</sup>						
CTR	$10.59\pm0.7$	$7.53\pm0.5$	$14.86 \pm 0.4$	$12.80\pm0.2$		
Pb 1,000	$9.86 \pm 0.1$	$8.11 \pm 0.4$	$15.65 \pm 0.6$	$12.35\pm0.4$		
Pb 2,000	$10.00\pm0.1$	$8.11\pm0.6$	$16.09 \pm 0.7*$	$14.75 \pm 1.7$		
Pb 3,000	$9.64 \pm 0.2$	$9.49 \pm 0.8*$	$17.30 \pm 0.0^{*}$	$17.12 \pm 0.2^{*}$		

Values are mean  $\pm$  SE (n = 6)

Data analyzed by one-way ANOVA followed by post hoc Turkey test

\* P < 0.05. Significances were determined by making comparisons between CTR vs Pb 1,000, 2,000 and 3,000 ppm

<sup>a</sup> µgGSH/mg protein

<sup>b</sup> nmol carbonyls/mg protein

a significant increase in protein carbonyl levels in Hc (16 %), while FA supplement marginally decreased the levels (Fig. 3d).

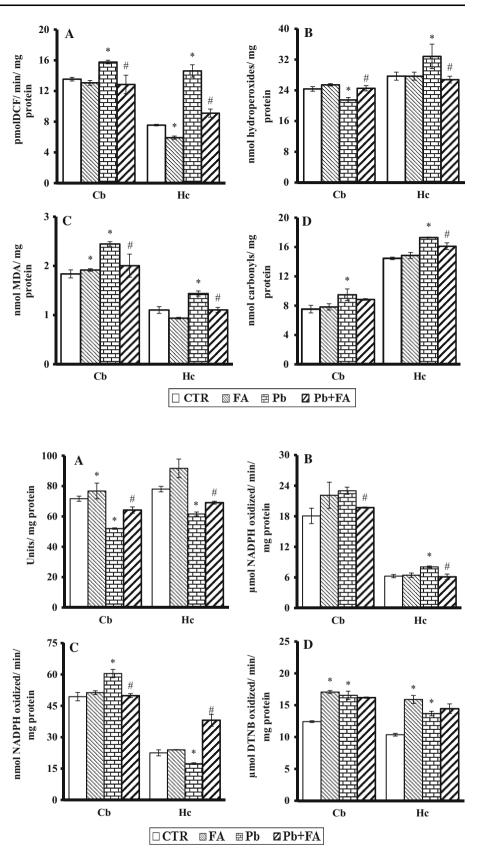
Effect on Antioxidant Enzyme Activities in Cerebellum and Hippocampus

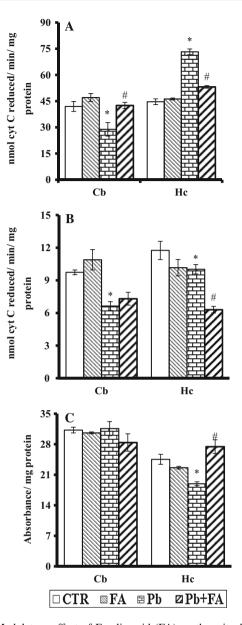
Pb exposure resulted in decreased SOD activity in Cb (27 %) and Hc (21 %), while FA supplement partially restored the levels (Fig. 4a). Further, Pb exposure resulted

in increased GR activity in Cb (27 %) and Hc (29 %). However, FA supplements restored the elevated activity levels to normalcy in both the brain regions (Fig. 4b). Likewise, Pb exposure caused increased GPx activity in Cb (22 %) and decreased the activity levels in Hc (23 %). However, FA supplements enhanced the levels in Hc by 1.2 fold and restored the levels in Cb (Fig. 4c). Interestingly, Pb exposure caused enhanced TRR activity in both

Fig. 3 Modulatory effect of Ferulic acid (FA) supplements on ROS generation (a), Hydroperoxides (b), Malondialdehyde (c) and protein carbonyl content (d) in brain regions of prepubertal rats exposed to lead (Pb-3,000 ppm in drinking water). Values are mean  $\pm$  SE (n = 6); Data analyzed by one way ANOVA (P < 0.05) followed by Turkey's test (\* compared to control; # compared to Pb 3,000), CTR-Control; Pb-lead acetate-3,000 ppm, FA-ferulic acid 25 mg/kg bw/day oral for 5 weeks. Cb cerebellum, Hc hippocampus

Fig. 4 Modulatory effect of Ferulic acid (FA) on the activity levels of Superoxide dismutase (a), Glutathione reductase (b), Glutathione peroxidase (c) and thioredoxin reductase (d) in brain regions of prepubertal rats exposed to lead (Pb-3,000 ppm in drinking water). Values are mean  $\pm$  SE (n = 6); Data analyzed by one way ANOVA  $(P \le 0.05)$  followed by Turkey's test (\* compared to control; \* compared to Pb 3,000), CTR-Control; Pb-lead acetate-3,000 ppm, FA-ferulic acid 25 mg/kg bw/day oral for 5 weeks. Cb cerebellum, Hc hippocampus





**Fig. 5** Modulatory effect of Ferulic acid (FA) on the mitochondrial enzyme activity-Complex I–III (**a**) and Complex II–III (**b**), and MTT (**c**) in brain regions of prepubertal rats exposed to lead (Pb-3,000 ppm in drinking water). Values are mean  $\pm$  SE (n = 6); Data analyzed by one way ANOVA ( $P \le 0.05$ ) followed by Turkey's test (\* compared to control; <sup>#</sup> compared to Pb 3,000), CTR-Control; Pb-lead acetate-3,000 ppm, FA-ferulic acid 25 mg/kg bw/day oral for 5 weeks. *Cb* cerebellum, *Hc* hippocampus

Cb and Hc. However, FA supplement had no effect on the elevated TRR activity levels (Fig. 4d). Among Pb exposed rats, GST activity was enhanced in Cb (45 %), and Hc (14 %) and FA supplements further enhanced the activity in Hc (Data not shown).

Effect of FA on Mitochondria in Cerebellum and Hippocampus

Pb exposure caused a significant decrease in the activity of complex I–III in Cb (31 %) and elevated the activity levels in Hc (64 %). However, FA supplements restored the activity levels to normalcy in both Cb and Hc (Fig. 5a). In contrast, Pb exposure caused a significant decrease in the activity of complex II–III in Cb (32 %) and Hc (15 %). However, FA supplements had no effect on diminished activity levels in Cb, but FA supplements further decreased the activity levels in Hc (Fig. 5b). Mitochondria of Pb exposed rats exhibited a significant reduction in the formation of formazan on exposure to MTT in Hc (23 %), and FA supplement offered significant protection in Hc (Fig. 5C).

# Modulatory Effect of FA in Striatum

Data on the modulatory effect of FA against Pb (3,000 ppm)-induced perturbations in oxidative stress markers, protein carbonyls and activities of antioxidant enzymes in striatum of PP rats is presented in Table 3. FA supplements significantly diminished the ROS and HP levels, while there was no effect on MDA levels. The marginally depleted GSH levels among Pb rats were restored to normalcy by FA and a similar restorative effect was also observed in protein carbonyl levels. In contrast, Pb exposure resulted in marked diminution in the activity levels of SOD (43 %) and FA supplement further decreased the SOD levels (33 %). Similarly, Pb exposure resulted in increased activity levels of GR (89 %), GPx (47 %), TRR (27 %) and GST (20 %) levels. FA supplements restored the activity levels of GR and GPx, while no effect was evident in the activity levels of GST. On the other hand, the activity levels of TRR were elevated in Pb and FA supplemented rats. Interestingly, Pb caused a significant increase in the activity levels of complex I-III (45 %) and decreased the complex II-III (26 %) in striatum of PP rats. However, FA supplements restored the activity levels to normalcy suggesting the protective effect of FA in mitochondria. Further, FA supplements among Pb exposed rats, significantly offset the extent of MTT reduction as evident in the MTT reduction assay (Table 4).

Effect of FA on Cholinergic Function and Dopamine Levels in Striatum

Pb intoxication resulted in a marginal increase in the activity levels of AChE in striatum, while, FA supplements restored the activity levels (Fig. 6a). However, Pb exposure increased the BChE levels (by 16 %) and FA supplements further elevated the activity levels (Fig. 6b). Further, DA levels were significantly increased among Pb exposed rats, while the enhanced levels were completely restored with FA supplements (Fig. 6c).

Brain regions				
	Control	FA	Pb	Pb + FA
Markers				
ROS <sup>a</sup>	$6.65\pm0.08$	$5.03 \pm 0.34*$	$9.72 \pm 0.24*$	$7.98 \pm 0.53^{\#}$
GSH <sup>b</sup>	$3.95\pm0.01$	$3.72\pm0.01$	$3.18 \pm 0.03^{*}$	$3.84\pm0.03$
HP <sup>c</sup>	$40.9\pm0.85$	$34.3 \pm 2.05^{*}$	$32.8 \pm 1.42^*$	$32.1\pm0.93$
MDA <sup>d</sup>	$1.31\pm0.00$	$1.11 \pm 0.0*$	$1.30 \pm 0.1$	$1.15\pm0.0$
Protein carbonyls <sup>e</sup>	$12.80\pm0.18$	$9.62 \pm 0.39^{*}$	$17.69 \pm 0.41^{*}$	$14.03 \pm 0.79^{\#}$
Enzymes				
$SOD^{f}$	$130.49 \pm 4.2$	$97.08 \pm 6.5^{*}$	$74.52 \pm 3.4^{*}$	$50.21 \pm 17.1^{\#}$
$GR^{g}$	$4.6\pm0.47$	$5.0 \pm 0.53$	$8.7 \pm 0.83^{*}$	$4.1 \pm 0.03^{\#}$
GPx <sup>h</sup>	$23.5\pm1.53$	$28.7 \pm 0.09*$	$34.5 \pm 1.65*$	$29.0 \pm 1.12^{\#}$
TRR <sup>i</sup>	$13.3 \pm 1.02$	$14.1 \pm 0.96$	$16.9 \pm 0.12^{*}$	$20.3\pm0.13^{\#}$
GST <sup>j</sup>	$84.2\pm0.95$	$100.8 \pm 2.56^*$	$101.0 \pm 0.61*$	$97.7 \pm 0.42^{\#}$

**Table 3** Modulatory effect of ferulic acid (FA) on the status of oxidative stress markers, reduced glutathione, activities of antioxidant enzymes and glutathione-s-transferase activity in striatum of prepubertal rats exposed to lead (Pb) acetate in drinking water for 5 weeks

Values are mean  $\pm$  SE (n = 6)

Data analyzed by one-way ANOVA followed by post hoc Turkey test

\* P < 0.05. (\* compared to control; # compared to Pb 3,000), CTR-Control; Pb-lead acetate-3,000 ppm, FA-ferulic acid 25 mg/kg bw/day oral for 5 weeks

<sup>a</sup> pmol DCF/min/mg protein

<sup>b</sup> µgGSH/mg protein (GSH)

<sup>c</sup> nmol hydroperoxides/mg protein (Hydroperoxides)

<sup>d</sup> nmol MDA/mg protein

<sup>e</sup> nmol carbonyls/mg protein (protein carbonyls)

<sup>f</sup> Units/mg protein (SOD)

<sup>g</sup> µmol NADPH oxidized/min/mg protein (GR)

<sup>h</sup> nmol NADPH oxidized/min/mg protein (GPx)

<sup>i</sup> µmol DTNB oxidized/min/mg protein (TRR)

<sup>j</sup> nmol GS-DNB/min/mg protein (GST)

Table 4 Modulatory effect of ferulic acid (FA) on the mitochondrial function in striatum of prepubertal rats exposed to lead (Pb) acetate in drinking water for 5 weeks

Parameters	Control	FA	Pb	Pb + FA
Complex I–III <sup>a</sup>	$50.88 \pm 3.08$	$66.36 \pm 3.46*$	$73.66 \pm 0.64*$	$58.36 \pm 5.75^{\#}$
Complex II–III <sup>b</sup>	$8.33 \pm 0.78$	$8.96\pm0.15$	$6.17 \pm 0.14^{*}$	$8.46 \pm 0.39^{\#}$
MTT <sup>c</sup>	$30.4 \pm 1.18$	$26.7\pm0.64$	$21.8 \pm 0.93^*$	$29.6 \pm 0.45^{\#}$

Values are mean  $\pm$  SE (n = 6)

Data analyzed by one-way ANOVA followed by post hoc Turkey test

\* P < 0.05. (\* compared to control; # compared to Pb), CTR-Control; Pb-lead acetate-3,000 ppm, FA-ferulic aci, 25 mg/kg bw/day oral for 5 weeks

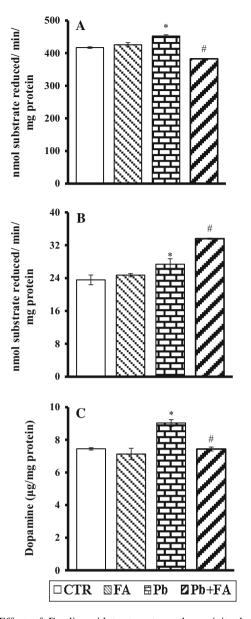
<sup>a</sup> nmol cytochrome C reduced/min/mg protein

<sup>b</sup> nmol cytchrome C reduced/min/mg protein

<sup>c</sup> Absorbance/mg protein (MTT)

Effect of FA on Pb-Induced Neuronal Damage in the Hippocampus

In order to determine implications of Pb-associated oxidative stress in the hippocampal region, we performed histopathologic analysis with H&E staining. Figure 7 represents the cornu ammonis area CA1 and the dentate gyrus (DG) regions of the Hc among control; Pb treated and Pb + FA treated groups. The hippocampi of the control



**Fig. 6** Effect of Ferulic acid treatment on the activity levels of acetylcholinesterase (**a**), butyrylcholinesterase (**b**) and dopamine levels (**c**) in striatum among rats exposed to lead (Pb-3,000 ppm in drinking water). Values are mean  $\pm$  SE (n = 6); Data analyzed by one way ANOVA ( $P \le 0.05$ ) followed by Turkey's test (\* compared to control; <sup>#</sup> compared to Pb 3,000), CTR-Control; Pb-lead acetate-3,000 ppm, FA-ferulic acid 25 mg/kg bw/day oral for 5 weeks

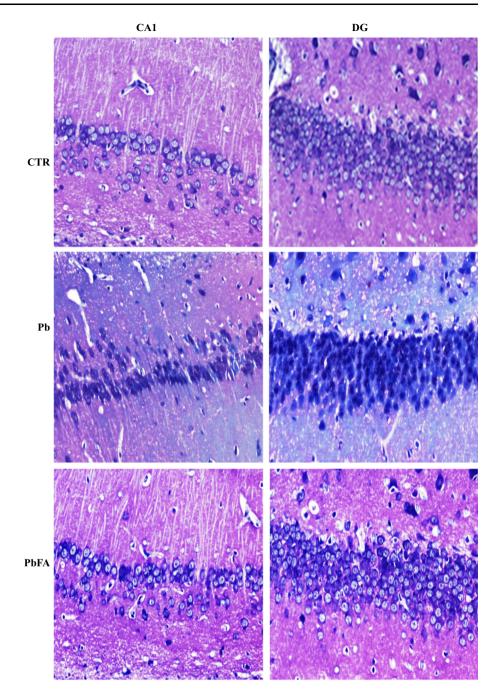
group had a normal architecture and damaged cells were almost nonexistent in CA1and DG regions. The number of damaged neurons among the Pb-treated group was markedly increased in the CA1 and DG regions of the hippocampi compared to that of the control group. Interestingly, we found significantly reduced number of damaged neurons in CA1 and DG regions among the Pb + FA group compared to that of Pb-exposed group.

#### Discussion

Pb toxicity is a persistent public health problem throughout the world and children are more susceptible than adults owing to their hand to mouth activity, increased respiratory rates and higher gastrointestinal absorption per unit body weight [5]. Epidemiological studies have shown that Pb exposure is associated with significant deficits in intelligence quotient of children and is associated with attention deficit hyperactivity disorder [47]. To alleviate Pb-associated toxic effects, several approaches such as chelation, antioxidant nutrients and their combination have been attempted [13–15]. It is in this context, that we examined the potential of FA supplements to ameliorate Pb-induced toxic effects employing a short-term dosing regimen. The criteria for using PP rats was based on the evidence that children exhibit enhanced susceptibility to Pb exposure and the brain is still in the process development of new inter neuronal connections which continue till the adult architecture is established by about 6-7 weeks. Hence, first we assessed the pattern of susceptibility of PP rats to varying dosages of Pb in terms of phenotype and induction of oxidative stress in brain regions. Further, with the objective of developing a pharmacological strategy, the propensity of FA, a ubiquitous naturally occurring antioxidant to ameliorate Pb-induced toxicity was investigated at the higher dosage.

In the present study, Pb exposure caused a significant reduction in the growth of rats, which is in agreement with previous reports [7]. Further, in the co-treatment paradigm, FA only marginally improved the reduction in the body weight. Pb exposure caused significant inhibition of ALAD activity, as shown previously [48]. ALAD is one of the most-reliable indicators of lead intoxication [11], whose inhibition contributes to the development of oxidative stress due to the accumulation of ALA, which is a substrate for ALAD. In fact, ALA may rapidly oxidize to generate ROS such as superoxide ion, hydroxyl radical, and hydrogen peroxides [49]. ALA is a potential endogenous source of free radicals and has been shown to cause oxidative damage to DNA in CHO cells in vitro through the formation of 8-OHdG adduct [50]. Further, ALAD inhibition decreases the biosynthesis of heme, which is the prosthetic group of many enzymes. We found that Pbinduced inhibition of ALAD activity was partially reversed by FA treatment. In contrast, FA supplements had no significant effect on the elevated blood Pb levels. Previously, several workers have adopted various strategies employing chelators to treat Pb poisoning and suggested that combinational therapy of antioxidants along with chelating agents may be a better treatment strategy than monotherapy to counter Pb-induced oxidative stress [51].

Fig. 7 Hematoxylin and eosin stained sections of hippocampi of rats (5 week study). Control (CTR); Pb exposure (Pb 3,000 ppm); Pb + Ferulic acid (25 mg/kg bw). Degenerating neurons with shrunken and dark nuclei were markedly increased in cornu ammonis CA1 and DG hippocampal regions of Pbexposed rats. The numbers of degenerating neurons were decreased among Pb + FA rats



Evidence of oxidative damage associated with the presence of Pb in the brain clearly suggests a possible role of free radicals in the pathogenesis of Pb neurotoxicity [7, 52]. Oxidative stress can be induced as a result of a depressed antioxidant system, increased ROS production, or both. FA, a phenolic acid with pleiotropic biological activity and exhibits a wide range of pharmacological effects including anti-ageing, anti-inflammatory, anticancer, antidiabetic, antiapoptotic and neuroprotective [19]. It is an effective scavenger of free radicals and has been approved in certain countries as a food additive to prevent

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LPO [53]. It also acts as an antioxidant against peroxyl radical generator [54]. Hence, we reasoned that it may be relevant to understand the neuroprotective effects of FA in vivo under Pb intoxication. In the PP model, we evidenced significant elevation of ROS and MDA levels in Cb, Hc and St of Pb administered rats. Interestingly, FA supplements at the administered dose caused a significant decline of ROS and MDA levels in Cb and Hc. Accumulating evidence suggests that Pb causes oxidative stress and that MDA levels strongly correlate with Pb concentration in the brain of rats [55–57]. It is well established that Pb

alters the lipid metabolism and enhances lipid peroxidation and increases brain TBARS through the inhibition of SOD in all brain regions [58]. Consistent with this the neuroprotective effect of epigallocatechin gallate (EGCG), the green tea polyphenol against Pb-induced neuronal damage is reported both in animal and cell model [59].

Earlier studies have shown that pre and neonatal Pb exposure altered the expression of antioxidant enzymes in the brain [9]. In the present model, Pb exposure caused significant perturbations in antioxidant enzyme activities such as SOD, catalase, and GPx, and changes in the concentrations of glutathione. SOD is considered as a first line of defense mechanism against ROS generation. Interestingly, FA supplements resulted in varying degree of restoration of SOD activity in Cb, which is in accordance with earlier results that FA effectively antagonizes the oxidative and nitrosative stress and inflammation as evidenced by down-regulated nitrite, LPO, IL-1 $\beta$ , TNF- $\alpha$ , and up-regulated GSH and SOD [60]. In the present study, Pb exposure resulted in altered GPx activity in Cb and Hc, which also corroborated with previous findings of decreased activity and protein expression of GPx in the frontal cortex, Cb and Hc. Furthermore, Pb exposure also resulted in altered activity levels of GR, TRR and GST, while FA supplements significantly restored the antioxidant enzyme activities clearly suggesting that its protective effects may be mediated through regulating oxidant/antioxidant defense, inflammatory and apoptotic signaling pathways. In the present study, Pb caused increased protein carbonyl levels in brain regions, viz., Cb, Hc, and St suggesting elevated protein oxidation as reported by previous workers [15]. FA supplements reduced the protein carbonyl levels only in the striatal region of the brain suggesting a specific effect of FA, which merits further study. In contrast, previously the antioxidant, gallic acid failed to reverse the Pb-induced protein carbonyl levels in rats [15].

Cerebral cortex and basal ganglia are brain areas involved in motor control and are intensely affected by Pb poisoning [3, 61, 62]. In view of this, in animal models, locomotor activity in open-field is often employed to assess the Pb-induced behavioral phenotype. PP rats exposed to Pb presented increased motor activity as observed by the increased number of crossings and decreased rearing movements, which are related to locomotor and exploratory activities, respectively. Interestingly, FA, supplements could partially reverse the locomotor and exploratory movements. The neuronal damage induced by Pb exposure is known to result in a reduction of the catecholaminergic transmission, either by inhibition of the DA synthesis and its release in the synaptic cleft or oxidative damages to the postsynaptic membranes [63, 64]. Since dopamine systems are important for central regulation of motor activity, we speculate that the observed behavioral phenotype reflects the effect of Pb on dopamine levels and its consequent oxidative stress.

Oral administration of FA has been demonstrated to increase the number of newly generated cells in the DG of the Hc of corticosterone treated mice, indicating that FA enhances the proliferation of neural stem/progenitor cells (NSC/NPCs) in vitro and in vivo [24]. Further FA was shown to increase cAMP response element binding protein (CREB) phosphorylation and brain-derived neurotrophic factor (BDNF) mRNA level in the Hc and ameliorative effect on the stress-induced depression-like behavior of mice. In the present model, evidence obtained data from histopathologic analysis indicated that supplementation of FA protected hippocampal neurons in the CA1 and DG regions against Pb-induced damage. Although speculative this specific effect of FA may be largely responsible for the ameliorative effect of FA against Pb-induced oxidative stress in the PP rat brain and may alleviate the associated cognitive deficits.

Despite intensive research efforts, the cellular mechanisms underlying the clinical manifestation of low-level Pb-induced neurotoxicity have remained elusive. Earlier studies have shown the role of neurotransmitter systems in Pb-induced behavioral perturbations and alterations in the properties of glutamatergic, cholinergic, and dopaminergic neurotransmitter function and signal transduction [64–66]. In the present study, Pb exposure resulted in increased cholinergic activity and enhanced DA levels in the striatal region, as reported previously [67]. Similarly, the locomotor activity and exploratory behavior were altered significantly among Pb-exposed animals corresponding to the alterations observed in cholinergic and aminergic systems [65]. Interestingly, FA supplements significantly altered both cholinergic and dopaminergic activity in PP rats. Further, recent studies have also shown that FA treatment reversed reserpine-induced behavioral abnormalities and decreased norepinephrine, serotonin and DA levels in the Hc and frontal cortex [60] and alleviated both learning/ memory deficits in vascular dementia model of rats [68].

Collectively our findings clearly indicate that FA supplements during PP stages in rats could significantly ameliorate Pb-induced oxidative stress damage and reverse the altered antioxidant/mitochondrial enzyme activities. Although the precise mechanisms responsible for the neuroprotective efficacy of FA against Pb intoxication during PP stage warrants further investigation, our data for the first time, demonstrate its efficacy to attenuate Pbmediated oxidative stress in brain regions and neurotoxicity. These findings emphasize the need for studies to understand the molecular events and the protective pathways which are responsible for the neuroprotective action of FA supplements under conditions of Pb-intoxication. Since oral FA supplementation is effective in reducing oxidative damage in brain regions and induces modifications in pathophysiological status under Pb-exposure, we propose its use as a potential complementary compound in the treatment of low-level Pb intoxication. Since FA is ubiquitously present in a variety of commonly consumed foods, increased consumption of FA-rich foods may reduce the neurotoxic implications of Pb among children.

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**Conflict of interest** The authors declare that there are no conflict of interest.

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