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Oxidative stress-mediated cytotoxicity of zirconia nanoparticles on PC12 and N2a cells

Elham Asadpour • Hamid R. Sadeghnia • Ahmad Ghorbani • Mehran Sedaghat • Mohammad T. Boroushaki

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Abstract In recent years, there is a growing interest in the application of nanoparticles like zirconium dioxide (zirconia <100 nm), for many purposes. Since a comprehensive study on the toxic effects of zirconia has not been done, we decided to investigate the effects of zirconia nanoparticles on cultured PC12 and N2a cells. In this study, cytotoxic effect of different concentrations of zirconia nanoparticles at three different time intervals were evaluated using MTT and ROS (reactive oxygen species) assays. Also, Lipid peroxidation, glutathione (GSH) content changes, and DNA damage were measured. Zirconia nanoparticles

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E. Asadpour

Anesthesiology and Critical Care Research Center, Shiraz University of Medical Sciences, Shiraz, Iran

H. R. Sadeghnia · M. T. Boroushaki (⊠) Department of Pharmacology, Faculty of Medicine, Mashhad University of Medical Sciences, Mashhad, Iran e-mail: boroushakimt@mums.ac.ir

A. Ghorbani · M. T. Boroushaki Pharmacological Research Center of Medicinal Plants, Mashhad University of Medical Sciences, Mashhad, Iran

M. Sedaghat

Department of Neurosurgery, Mashhad University of Medical Sciences, Mashhad, Iran e-mail: m_sedaghat81@yahoo.com caused a significant reduction in cell viability and GSH content of the cells, and induce a significant increase in intracellular ROS and MDA content of PC12 and N2a cells. Moreover, it increases the percentage of DNA tail of treated cells as compared with control group. Zirconia nanoparticles have cytotoxic and genotoxic effects in PC12 and N2a cells in a time and concentration-dependent manner in concentration more than 31 μ g/mL.

Keywords DNA damage · Reactive oxygen species · Viability · Zirconia nanoparticles

Introduction

Recently, tendency in using different nanoparticles in many aspects of life has been increased (Ma et al. 2011). According to different physicochemical properties of nanoparticles in comparison with traditional millimeteror micron-sized materials (mm to μ m), together with the surface properties, electronic structure, coordination and other properties, nanoscale particles show their own specific and diverse features (Tahmasebpour 2008).

The name zirconium (Zr) is derived from the Persian word, *zargon* (golden in color), which combines two Persian words, *zar* (gold) and *gun* (color) (Piconi and Maccauro 1999).

Zirconium(IV) oxide (zirconia) compounds are utilized worldwide in biological fields such as dental implants and other restorative practices (Aramouni et al. 2008; Chang et al. 2007; Uludag et al. 2008; Vigolo and Fonzi 2008), in knee (Tsukamoto et al. 2006) and hip (Garvin et al. 2009; Lappalainen and Santavirta 2005) replacement, and in middle-ear ossicular chain reconstruction surgery (Schadel et al. 1993). Moreover, it is utilized in settings such as nephrological practice, hemofiltration (Lee et al. 2010), hemodialysis (Hansen 2006), and in the design and construction of artificial kidneys (Davenport et al. 2007; Lee and Roberts 2008).

Zirconia nanoparticles are used in the production of nanobiosensor for DNA (Yang et al. 2007), phosphoprotein measurement in cell signaling studies (Rainer et al. 2008), in producing the bio-chip and information storage device (Liu et al. 2004) and also in poisons such as parathion (Parham and Rahbar 2010), organophosphate pesticides and nerve agents (Liu and Lin 2005). Recently, it is shown that zirconia nanoparticles are utilized as efficient gene delivery vehicles with target specificity for the spleen (Tan et al. 2007) and for transfection of mammalian cells (Link et al. 2007) as well as for drug delivery targets (Singh and Lillard 2009). For example, it is employed for development of ampicillin in local controlled drug delivery system (Catauro et al. 2008), itraconazole nanosuspension for oral delivery (Nakarani et al. 2010) and in the targeting delivery of bisphosphonates-based drugs such as alendronate and zoledronate (Colilla et al. 2009), which the optimized formulation has indicated better results than the marketed one (Colilla et al. 2009; Nakarani et al. 2010).

Recent studies in nanocomponent field have revealed that zirconia nanocomposites could tolerate the severe vapor-moist environment better than zirconium component (Pezzotti et al. 2010). Then again, nanofunctionalized zirconia and barium sulfate particles as bone cement additives could improve the properties of traditional bone cements for orthopedic applications (Gillani et al. 2010). However, the introduction of these materials into clinical practice as bone cement additives has been complicated by concerns over the unknown long-term risk profile of these new structures in vivo (Gomoll et al. 2008). The large majority of people do not experience short-term side effects of metal denature. Nevertheless, prolonged exposure to dissolved metal ions released from implants can lead to poor performance of implants and may cause pathological effects (Chevalier 2006; Hallab et al. 2004; Hallab et al. 2000).

Although previously it was thought that zirconia is a neutral bioceramic metal, more recent studies have demonstrated that it is likely to cause some toxic effects. For instance, analysis of zirconium ion showed significant DNA damage induction at concentration of 5 mM and apoptosis in human (Jurkat) T-cells at concentration exceeding 0.5 mM (Caicedo et al. 2008). Similarly zirconia particles around 0.6 µm in size could induce TNF- α release and macrophage apoptosis in J774 mouse cell line (Catelas et al. 1998, 1999). As well, osteoblast-like (MG63) cells cultured on zirconia and colloidal zirconium coated disks could alter the expression of several genes which significantly up or down regulate a broad range of functional activities: such as (a) cell cycle regulation, (b) signal transduction, (c) immunity, and (d) vesicular transport (Carinci et al. 2004; Sollazzo et al. 2008). Exposure of human marrow stroma-derived mesenchymal stem cells (hMSCs) to submicron particles of zirconia, compromises cell viability through the induction of apoptosis, causing increased levels of the tumor suppressor proteins p53 and p73 (Wang et al. 2003). Additionally, exposure of zirconia (0-30 ppm) to human mesothelioma (MSTO-211H) and a rodent fibroblast (3T3) cell line for duration of 3 and 6 days inhibited the cell proliferation (Brunner et al. 2006). Zirconia was moderately apoptotic, inducing >50 % caspase-9 positive cells at 0.5 mM concentration or higher, However, zirconia did not induce any significant DNA damage below 1 mM, significant (p < 0.05) DNA damage could be observed at 5 mM, reaching 192 and 250 IDD (index of DNA damage), respectively (Caicedo et al. 2008).

Since a comprehensive study based on the probable toxic effects of zirconia nanoparticles has not been carried out and regarding the increasing production and usage of this nanoparticle, and as PC12 and N2a cell lines are widely used in neurotoxicological study and oxidative stress induced by nanoparticles (Perreault et al. 2012; Wang et al. 2011), we decided to investigate the oxidative stress induced by zirconia nanoparticles on PC12 and N2a cells.

Materials and methods

Cell lines and reagents

Commercial zirconium (IV) oxide nanopowders (Zirconia) <100 nm particle size (TEM), were purchased from Sigma-Aldrich (544760-25G, St Louis, MO). PC12 and N2a cell lines, derived from rat pheochromocytoma and mouse neuroblastoma, respectively, were purchased from Pasteur Institute (Tehran, Iran). High glucose (4.5 g/L) Dulbecco's Modified Eagles Medium (DMEM) and fetal calf serum (FCS) were purchased from Gibco (Carlsbad, CA). 3-(4,5dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium (MTT), 2',7'-dichlorodihydrofluorescein diacetate (H2DCF-DA), monochlorobimane (mBCl), and other cell culture materials were purchased from Sigma (St Louis, MO). Low melting point (LMP) and normal melting point (NMP) agarose were obtained from Fermentas (Glen Burnie, MD). Other chemicals mainly ethylene diaminetetraacetic acid disodium salt (Na2EDTA), Tris (hydroxymethyl) aminomethane (Trizma base), t-octylphenoxy poly-ethoxyethanol (Triton X-100), dimethyl sulfoxide (DMSO), sodium lauroylsarcosinate (sarkosyl, SLS), and ethidium bromide were purchased from Merck (Darmstadt, Germany). 2-Thiobarbituric acid (TBA) was purchased from Sigma-Aldrich (St Louis, USA). Trichloroacetic acid (TCA) and malondialdehyde bis-(dimethyl acetal) (MDA) were obtained from Merck (Darmstadt, Germany).

Dispersion and characterization of the zirconia nanoparticles

The zirconia nanoparticles were suspended in media and dispersed by an ultrasonic bath for 60 min, and then the suspension was characterized by the TEM using one drop of sample on a carbon coated grid and high voltage (120 kV) imaging. The size of particles was tested using Transmission Electron Microscopy (TEM, LEO 912AB, Germany).

Cell culture and nanoparticle treatment

PC12 and N2a cell lines were maintained in DMEM containing 10 % FCS, penicillin 100 IU/mL, and streptomycin 100 mcg/mL. Cells were grown and maintained in 75 cm² cell culture flasks at 37 °C in a 5 % CO₂ humidified incubator. The suspension of zirconia nanoparticles was prepared using the culture media and dispersed for 20 min using a sonicator (Bandelin electronic, DT 510, Germany) to prevent aggregation. The cells were treated with various concentrations of nanoparticles for 12, 24, and 48 h.

Test concentration

Different concentration of nanoparticle suspension was used for various tests. For MTT and ROS assays, cells were treated with 2000, 1000, 500, 250, 125, 62, 31, 15, and 7.8 μ g/mL final concentrations of zirconia nanoparticles suspension and For remaining assays cells exposed to 2000, 1000, 500, 250, 62, 31 μ g/mL of nanoparticle suspension all for 12, 24, and 48 h.

Cell viability test

MTT was used to identify viable cells which could reduce this compound to a violet formazan (Mosmann 1983). Cells were seeded on 96-well tissue culture plates with 5×10^3 cells in 100 µL media per well. After allowing the cells to be seeded for 24 h, nanoparticle suspension (100 µL) were added to each well. At the end of exposure, MTT solution in phosphate-buffered saline (PBS, 5 mg/mL) was added to a final concentration of 0.05 %. After 2 h, the formazan precipitate was dissolved in DMSO containing 10 % glycine buffer (pH 10.5). The microplates were then gently shaken in the dark for 30 min, and the absorbance at 570 and 620 nm (background) was measured using a StatFAX303 plate reader.

All experiments were carried out as triplicate; the number of viable cells was determined as mean \pm - SEM and is presented as a percentage of non-treated control groups, which was assumed to be 100 %.

Measurement of reactive oxygen species

A fluorometric assay using intracellular oxidation of DCFH-DA was performed to measure ROS generation (Zhang et al. 2010). H₂DCF-DA readily diffuses through the cell membrane and is enzymatically hydrolyzed by intracellular esterase to non-fluorescent H₂DCF, which is then rapidly oxidized to highly fluorescent DCF (2',7'-dichlorofluorescin) in the presence of ROS. The DCF fluorescence intensity is believed to be parallel to the amount of intracellular ROS. Cells were seeded on 96-well tissue culture plates with 5×10^3 cells in 100 µL media per well and cultured so that about 70 % confluency was reached after 24 h. then they were pre-treated with different concentrations of nanoparticles, the cells were incubated with 10 µm H₂DCF-DA at 37 °C for 30 min in the dark space. Cells were washed twice with warm PBS and collected. The DCF fluorescence intensity was quantified using a FLUO-star galaxy fluorescence plate reader (Perkin Elmer 2030, Multi label reader, Finland) with excitation wavelength set at 485 nm and emission wavelength set at 530 nm. The results are given as percent's relative to the ROS content of the control cells which set as 100 %. All experiments were performed in triplicate.

MDA assay

The thiobarbituric acid reactive substances (TBARS) assay measures lipid hydroperoxides and aldehydes, such as malondialdehyde (MDA), both, in the cell culture media and cell lysate. MDA combines with thiobarbituric acid (TBA) in 1:2 ratio to form a fluorescent adduct that is measured at ex. 530 nm and em. 550 nm. TBARS are expressed as MDA equivalents.

Cells were processed as above in 6 well plates (10⁶ cells per well) and after seeding were pre-treated with different concentrations of nanoparticles.

Cell media sample preparation

After collecting the Cell media from each well, 500 μ L of sample media were added to 400 μ L TCA 15 % plus 800 μ L TBA 0.67 %/BHT 0.01 % in 5 mL amber vials. Then samples were vortexed and heated for 20 min in a 95 °C water bath. Thereupon 3 mL butanole were added to the samples' vials and the attained phases gently mixed. Finally, 200 μ L of the butanole phase (top) were transferred to 96 well plates.

Cell lysate sample preparation

The 96-well plates were washed with cold PBS. Cells were scraped into 1 mL TCA2.5 %. Then cell suspensions were centrifuged at 13000 g for 2 min and samples were processed as above.

Cell media and cell lysate plates were read in fluorescence mode, ex = 530 nm, em = 550 nm. The MDA amounts calculated according to the standard curve which was obtained with various concentration of MDA from 400 to 0.007 nmol/mL.

GSH determination

Cells were seeded in 96-well tissue culture plates containing 5×10^3 cells in 200 μ L media per well.

After treating the cells with different concentrations of nanoparticles, fluorescent probe (mBCl, monochlorobimane) was dissolved in dimethyl sulfoxide. Working solutions in PBS with concentration equivalent to 20 times the required final concentration were prepared immediately before use and added directly to the cultures in 96-well plates. Fluorescence was recorded before (blank) and after adding the probe in a FLUO-star galaxy fluorescence plate reader at the following excitation/emission wavelengths \pm band pass: $360 \pm 40/460 \pm 40$ nm. Data were obtained by subtracting blank values and were presented as a percentage of control cells fluorescence (Sebastia et al. 2003).

Single cell gel electrophoresis (SCGE, Comet) assay

The alkaline SCGE assay was conducted based on the method described previously (Asadpour et al. 2014a; Ghorbani et al. 2015). Briefly, PC12 and N2a cells (3×10^5) were incubated with five different concentrations (2000, 1000, 500, 250, 62, 31 µg/mL) of zirconia nanoparticles. After removing the medium, the cells were washed three times with cold PBS, harvested and centrifuged at 3000 rpm for 5 min at 4 °C. The pellets were then re-suspended in PBS at a cell density of 1×10^5 . For the comet assay, 100 µL NMP agarose was quickly layered on conventional slides, covered with a cover slip, and then the slides were placed on ice to allow agarose to get form of gel. 10 μ L of the nucleus suspension, prepared as above, was mixed with 100 µL LMP agarose, and the mixture was quickly layered over the NMP agarose layer after removal of the cover slip. Finally, another layer of LMP agarose was added on top. The slides were immersed immediately in a chilled lysing solution (pH 10) made up of 2.5 M NaCl, 100 mM Na₂EDTA, 10 mM Trizma, 1 % sarkosyl, 10 % DMSO, and 1 % Triton X-100, and kept at 0 °C in the dark overnight. Then, the slides were placed on a horizontal gel electrophoresis platform and covered with a prechilled alkaline solution made up of 300 mM NaOH and 1 mM Na₂EDTA (pH > 13). They were left in the solution in the dark at 0 °C for 40 min, and then electrophoresed at 0 °C in the dark for 30 min at 25 V and approximately 300 mA. The slides were rinsed gently three times with 400 mM Trizma solution (adjusted to pH 7.5 by HCl) to neutralize the excess alkali, stained with 50 µL of 20 µg/mL ethidium bromide, and covered with a cover slip. For comet analysis, 150 nuclei were randomly selected from three replicated slides (50 nuclei per slide), examined and photographed through a fluorescence microscope (Nikon, Kyoto, Japan), at ×400 magnification equipped with an excitation filter of 520–550 nm and a barrier filter of 580 nm. Undamaged cells resemble an intact nucleus without a tail, and damaged cells have the appearance of a comet. The percent of DNA in the comet tail (% tailed DNA), which is an estimation of DNA damage, was analyzed using the computerized image analysis software (CASP software). The experiments were carried out in triplicate.

Statistical analysis

The results are presented as the mean \pm SEM. The values were compared using the one- and two-way analysis of variance (ANOVA) followed by Bonferroni post tests for multiple comparisons. The p < 0.05 were considered to be statistically significant.

Results

Characterization of zirconia nanoparticles

The size of the zirconia nanoparticles was varied <100 nm as shown in the images of Transmission Electron Microscopy (TEM) with a size distribution between 8 and 50 nm, multivariate shapes and molecular weight of 123.22 g/mol (Fig. 1).

Cytotoxicity of the zirconia nanoparticles

Assessment of zirconia nanoparticles cytotoxicity through the MTT assay exhibited the dose-dependent toxic effects on cell viability in a concentration range of 31–2000 µg/mL (with IC₅₀ value for the PC12 cell line of about 66.68, 53.32, and 40.82 µg/mL and IC₅₀ value for the N2a cell line of about 45.49, 33.06, and 26.27 µg/mL following 12, 24, and 48 h exposure, respectively). Comparison of PC12 cells viability between 12 h and the two other time exposures showed a significant difference from 62 to 2000 µg/mL (p < 0.05 for 62 µg/mL between 12 and 24 h and p < 0.001 for others), but this significant differences were not seen between 24 and 48 h except for the 15 and 31 µg/mL (p < 0.001). These results were



Fig. 1 Transmission electron microscopy (TEM) micrograph of zirconia nanoparticles suspended in media (DMEM) and dispersed by an ultrasonic bath

different in N2a cells since the difference between 12 and 24 h was started from 250 µg/mL (p < 0.05 for 250 and 500 µg/mL and p < 0.001 for upper concentrations), between 12 and 48 h from 31 µg/mL (p < 0.001 for 31–125 µg/mL and p < 0.05 for upper concentrations) and between 24 and 48 h from 31 µg/mL (p < 0.001) (Fig. 2).

Measurement of reactive oxygen species

ROS generation was elucidated in PC12 and N2a cell lines at all three exposure times from concentration of 15 µg/mL and continues up to concentration of 2000 μ g/mL (p < 0.01 for 15–31 μ g/mL of N2a after 48 h exposure and p < 0.001 for all other concentrations). Comparison between different exposure times did not show a time-dependent manner for ROS generation since in PC12 cell line it was obvious only at concentration of 7, 15, and 2000 µg/mL between 12 h and the two other exposure times (p < 0.0001) and in N2a cell line between 12 and 24 h was significant for 62–1000 μ g/mL (p < 0.01), between 12 and 48 h only for 7 and 2000 μ g/mL (p < 0.01) and between 24 and 48 h only for 500 μ g/mL (p < 0.01). This increase was not significantly different between various duration of exposure but only it showed a concentration-dependent manner (Fig. 3).

Fig. 2 Effects of zirconia nanoparticles on a PC12, **b** N2a cell viability after 12, 24, and 48 h treatment. The cell viability (quantified by MTT assay) is shown and discussed as percentage of control group (zirconia nanoparticles 0 μ g/mL). Mean and SEM of three independent experiments are shown. *p < 0.05, **p < 0.01 and ***p < 0.001 as compared to control









Effect of zirconia nanoparticles on MDA generation in PC12 and N2a cells

Zirconia nanoparticles lipid peroxidation effect was increased in both cell lines gradually from lower concentration toward upper ones. But the amount of MDA in the cell lysate is almost 10 times lower than the cell culture media. This effect is also time dependent especially between 12 and 48 h exposure times. The significant difference between 12 and 48 h, at PC12 cell line sample of culture media, started from 31 µg/ mL (p < 0.01) but in cell lysate sample the difference is seen between 12 and 24 h from 62 µg/mL (p < 0.001) and between 12 and 48 h from 31 µg/mL (p < 0.001). At N2a cell line sample of culture media, the lipid peroxidation effect made a significant difference between 12 and 48 h of culture media from 31 μ g/ mL (p < 0.01) and in cell lysate between 12 h and the two other exposure times from 62 μ g/mL (p < 0.01) (Fig. 4).

GSH changes induced by zirconia nanoparticles

As a reference for the fluorescence studies, GSH was analyzed by Tietze's method. These data indicated that PC12 and N2a cells are practically devoid of GSH after different times of exposure to zirconia nanoparticles. Both cell lines showed almost same level of GSH depletion during three different exposure times from almost 85 % at 12 h exposure in concentration of 31 μ g/mL to nearly 35 % at 48 h exposure in 2000 μ g/mL. The GSH depletion seen here was concentration dependent without any remarkable time-dependent manner (Fig. 5).

Effect of zirconia nanoparticles on DNA damage in PC12 and N2a cells

DNA tail percentage was measured by SCGE as an indicator of DNA damage. The results showed that exposure of PC12 and N2a cells to zirconia



Time (hour)

was measured against the standard curve provided with different concentration of MDA (zirconia nanoparticles 0 μ g/mL). *p < 0.05, **p < 0.01, and ***p < 0.001, as compared with control group of 12, 24, and 48 h, respectively

Time (hour)

Fig. 4 Effects of zirconia nanoparticles on malondialdehyde (MDA), in **a** PC12 cell culture media content, **b** N2a cell culture media content, **c** PC12 cell lysate content and **d** N2a cell lysate content. Cells were treated with different concentrations of zirconia nanoparticles for 12, 24 and 48 h. The MDA content



Fig. 5 Fluorescence labeling curves of **a** PC12, **b** N2a cell cultures with mBCl at the three indicated exposure duration with different concentrations of zirconia nanoparticles. Fluorescence values are shown as a percentage of control cells fluorescence. N = 9, from three independent cultures. *First column* indicates control cells, and after that, columns for each exposure time indicate cells exposed to the GSH-depleting media. (mBCl, monochlorobimane), *p < 0.05, **p < 0.01, and ***p < 0.001, as compared with control group of 12, 24, and 48 h respectively

nanoparticles significantly increased DNA fragmentation (DF), comparing to the control cells. Similar to GSH, here, both cell lines showed parallel manner regarding treatment with zirconia nanoparticles since the DNA tail percentage was enlarged from almost 10 % at 12 h exposure in concentration of 31 μ g/mL to nearly 70 % at 48 h exposure in 2000 μ g/mL. DNA damage measurement revealed a concentration-dependent mode in both cell lines (Fig. 6).

Discussion

In the present study for the first time, we have mechanistically shown that zirconia nanoparticles persuade cytotoxic and genotoxic effects on PC12 and N2a cell lines at specific concentration. Recently, there are growing interest in the application of nanometal oxides such as zirconia in tissue engineering and implants especially orthopedic and dental implants. It has been shown that metal-based nanoparticles could penetrate through the blood brain barrier into the central nervous system of exposed animals, via blood circulation or the nervous olfactorius after intravenous injection or deposition on the olfactory mucosa following intranasal instillation (Kreyling et al. 2002; Liu et al. 2010; Lockman et al. 2004; Oberdorster et al. 2004). Recently the potential risk of copper oxide nanoparticles on the function of nervous system has been investigated (Xu et al. 2009; Zhao et al. 2009). It also stated that nanoparticle-based exposure of PC12 and N2a cells resulted in dosedependent neurotoxicity in cultured cells (Perreault et al. 2012; Wang et al. 2011).

The in vitro cytotoxicity of zirconia nanoparticles exhibited the dose-dependent reduction in cell viability over a concentration range of $31-2000 \,\mu g/mL$ which is in consistent with the previous results quantified its cytotoxicity by MTS assay (Asadpour et al. 2014b). Our findings suggest the toxicity difference of zirconia nanoparticles with respect to the cell types. PC12 cells were found to be more tolerant to zirconia-induced cytotoxicity than N2a cells. The reduction in cell viability is more evident in N2a than PC12 cell line since N2a cell viability decreased to lower amount in all three different assessed time durations. In N2a cell line the cytotoxic effect after 24 h was stayed almost constant, this could prove the more sensitivity of this line comparing to PC12 which the cytotoxic effect is continued up to 48 h exposure. The cell viability reduction could be the consequence of physical or chemical characteristics of nanoparticles that persuade to cell injury or death by disturbing effect on the cell membrane or the intracellular organelles (Ghasempour et al. 2015). These nanoparticles' adverse effects can be ATP decline through mitochondrial respiratory chain damage, formation of reactive oxygen species (ROS), and oxidative stress contributing to elevated chance of cell injury or death (Xia et al. 2006).

Our results also indicated that the cytotoxicity appears to be mediated by oxidative stress, a common mechanism which has been well documented for nanoparticles-induced toxic effects (Jeong et al. 2011; Wu et al. 2010). This ROS generation could provoke inflammation or pro-apoptotic cell signaling through caspase activation and cytochrome c release from mitochondria and the following oxidative stress Fig. 6 Comet assay images and %Tail DNA induced by zirconia nanoparticles in a PC12, b N2a after 12, 24, and 48 h exposure time, including Comet assay images. All data are represented as the mean \pm SEM of three independent experiments *p < 0.05, **p < 0.01, and ***p < 0.001, as compared with control group of 12, 24, and 48 h, respectively



produced by ROS could lead to external or internal cell damage (Jeong et al. 2011; Simon et al. 2000). More studies in this area are warranted to discover the

cellular mechanism of ROS damage. Here, we have considered that the cellular oxidative stress induced by zirconia nanoparticles were indicated by elevated ROS generation, reduced GSH levels, increased lipid peroxidation and oxidative DNA damage.

Lipid peroxidation has been considered as a crucial key in the nanoparticles-induced cytotoxicity and DNA damage (Asadpour et al. 2014a; Blair 2008) as it started from 31.25 µg/mL. An increase in the level of thiobarbituric acid reactive species that correlated reversely with the cell viability of the cells, is suggesting that zirconia nanoparticles could induce membrane damage through lipid peroxidation. The glutathione redox cycle has an important role in detoxifying the generated ROS and other free radicals (Khodagholi et al. 2012). From the results, the GSH depletion was seen from 31.25 µg/mL after 12 h exposure, but this effect was more obvious at two highest concentrations and following 48 h exposure. Since the GSH is markedly reduced by zirconia nanoparticles' exposure, it is reasonable to deem that cytotoxic effect of this nanoparticle, at least partially, is mediated via oxidative properties. Moreover, cell membrane injury triggered by lipid peroxidation can induce continuous enzyme release like lactate dehydrogenase which in exchange initiate a chain reaction of more cell loss (Wang et al. 2011). This is in agreement with previous studies which revealed the lactate dehydrogenase level boost in cell medium after the exposure of cells with nanoparticles (Hussain et al. 2005; Lin et al. 2006).

it is well documented that excessive generation of ROS could induce DNA damage by changing the mitochondrial membrane permeability or by oxidative alteration of DNA molecule, directly (Circu and Aw 2010). Our investigation showed that great DNA damage is evident following treatment of the cells by concentrations above 31.25 µg/mL. Amorphous zirconia increases both cytotoxicity and genotoxicity in human osteosarcoma cell line but in another study it is indicated nano zirconia reveals no genotoxic effect on human peripheral blood lymphocytes and cultured human embryonic kidney cells (Demir et al. 2013; Di Virgilio et al. 2014). This could suggest that particular physical characteristics of zirconia nanoparticles may not be the more vigorous DNA damage mediator and its effect on diverse cell lines is poles apart.

In the present study, we have evaluated the cytotoxic and genotoxic effect of zirconia nanoparticles without the existence of any positive control like hydrogen peroxide for comparing possibility of cytotoxic effect and with no facility for comparing. Moreover, we did not have access to any standard drugs to counteract with probable toxic effects of zirconia nanoparticles.

These data highlight the prominence of using numerous assays to cover the wide-ranging interactions between cell and nanoparticles and to define nontoxic nanoparticles concentrations especially in neuropathic disease like Parkinson and Alzheimer.

Conclusion

In conclusion, the present study addressed the potential toxicity of zirconia nanoparticles in PC12 and N2a cells. The results showed that zirconia nanoparticles could induce cell damage via oxidative stress induction which found to be dependent on the concentration used and time of exposure. Further studies are required to address the molecular mechanisms of the cell death induced by zirconia nanoparticles, in vitro or in vivo.

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Compliance with ethical standards

Conflict of interest The authors report no conflict of interest. The authors alone are responsible for the content and writing of the paper.

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