RESEARCH PAPER

Extraordinary high dielectric constant, electrical and magnetic properties of ferrite nanoparticles at room temperature

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Abstract Nanoparticles of spinel ferrites of basic composition $Ni_{1-x}Co_xFe_2O_4$ (0.0 $\leq x \leq 0.05$) were synthesized through modified co-precipitation method, and were characterized for structural, transport electrical and magnetic properties using XRD, HRTEM, FTIR, LCR meter and VSM techniques, respectively. XRD analysis showed that all the samples are single-phase cubic spinel in structure. The average crystallite sizes of the nanoparticles were found between 30 nm to 45 nm. Real and imaginary parts of the impedance $(Z'$ and $Z'')$ suggested coexistence of two relaxation regimes: one was introduced by electrode polarization, while the other was attributed to the coeffect of grain and grain boundary effects. The dielectric constant of the samples was found very high, which showed non-Debye relaxation

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Md. Shahabuddin - N. Ahmed Department of Physics and Astronomy, King Saud University, P.O. Box 2455, Riyadh 11451, Saudi Arabia phenomena, while conductivity of the samples exhibited a two-segment behavior with frequency. The room temperature M–H curves suggested that the samples exhibit supermagnetism, and the saturation magnetization increases with increasing $Co²⁺$ ion substitution.

Keywords Nanoparticles - Ferrites - Impedance spectroscopy - Conductivity - Dielectric constant - Magnetization

Introduction

Spinel ferrite nanoparticles (NPs) by virtue of their unique electronic, magnetic, and physical structure are an important class of materials because of their high resistivity and low eddy current losses (Son et al. [2002](#page-8-0); Sugimoto [1999;](#page-8-0) Atif and Hasanian [2006](#page-7-0)). They find wide use in ultrahigh frequency applications. As NPs, they are also considered for biomedical applications, including as contrast agents in magnetic resonance imaging (Mitchell [1997\)](#page-8-0), and in promising cancer remediation therapies such as target drug delivery, magnetohyperthermia, and magnetorheological fluids (Hafeli et al. [1997](#page-8-0)). However, the electromagnetic properties of ferrite NPs have been found to vary widely over the size, and with foreign element substitution (Hochepied et al. [2000;](#page-8-0) Morrison et al. [2004](#page-8-0)). In order to realize their full potential, understanding of processing

and refinement of ferrite NPs is essential. One of the parameters of great importance is dielectric constant/ dielectric loss which is very sensitive to cation–anion– cation bond angles, cation type and its valence state. Demonstrating its importance is an application of ferrite NPs in microwave devices, multilayer inductor chips, transformer cores, etc.

Nanocrystalline ferrite systems usually have a mixed spinel structure having chemical formula $M^{2+}_{1-\delta}Fe^{3+}_{\delta}$ $(M_{1-\delta}^{2+}Fe_{\delta}^{3+})[M_{\delta}^{2+}Fe_{\delta}^{2-}]O_4^{2-}$. The divalent metal ion M^{2+} can occupy either tetrahedral (A) or octahedral [B] sites or both sites of a spinel structure, depending upon the nature of a system. The inversion parameter, δ , is defined as the fraction of (A) sites occupied by $Fe³⁺$ cations and its value depends on method of preparation (Willard et al. [2004;](#page-8-0) Tug et al. [2002](#page-8-0)). $NiFe₂O₄$ is one of the most important magnetic materials, and therefore, is extensively used in high frequency applications via microwave devices due to their high resistivity and low dielectric losses. It is reported (Chinnasamy et al. [2001\)](#page-8-0) that nanocrystalline $NiFe₂O₄$ exhibits a mixed spinel structure with $Ni²⁺$ ions occupying both (A) and [B] sites. NiFe₂O₄ NPs with a mixed spinel structure have been shown to exhibit interesting electrical, magnetic, gas, and humidity sensing properties (Morrison et al. [2004](#page-8-0); Jacob and Khadar [2010;](#page-8-0) Brook and Kingery [1967\)](#page-7-0).

In order to differentiate between transport characteristic such as grain and grain boundary in ferrites, impedance spectroscopy (IS) is one of the powerful techniques to carry out (Macdonald [1987\)](#page-8-0). The microstructure of ferrites comprises two main components; grain and grain boundary, and the correspondence between the two is important to understand the overall properties of these materials (Idrees et al. [2010\)](#page-8-0). IS can be employed to investigate the conduction mechanism in different microstructures (phases) inside a polycrystalline material. By using impedance technique, real and imaginary parts of complex electrical quantities are measured as a function of frequency of the applied field (Barsoukov and Macdonald [2005](#page-7-0)). These complex quantities include electrical impedance, dielectric permittivity, and loss tangent. The grain boundaries are found to have main control over the electrical properties of ferrites (Gopalan et al. [2008](#page-8-0)). The study of dielectric and impedance properties of $NiFe₂O₄$ NPs will be constructive for extensive range of applications.

In this paper, we present a detailed report on synthesis of Co-doped $Ni_{1-x}Co_xFe_2O_4$ ferrite NPs and study of their structural, dielectric, impedance, and magnetic properties to determine any possible change in the conduction or magnetic properties due to substitution effect.

Experimental method

Material preparation

The raw materials used to synthesize NPs of Ni_{1-} $_{x}Co_{x}Fe_{2}O_{4}$ (where $x = 0.0, 0.1, 0.2, 0.3, 0.4$ and 0.5) were nickel chloride (99 % Merck, India), cobalt (III) chloride (99 % Merck, India), iron (III) chloride (98 % Merck, India), and sodium hydroxide (98 % Merck, India). The weighted amounts of the chlorides were dissolved in a mixture of 50 ml double distilled and 25 ml ethanol. The mixed metal chloride precursor solution was introduced slowly into a boiling 1 M NaOH solution, and the reaction was carried out at 98 \degree C for 120 min. The pH value of solution was fixed at 7. The pH environment is critical to particle nucleation, rate of growth, stabilization and ultimately size of the resulting particles. Following the completion of reaction, resulting particles were rinsed in distilled water, filtered, and dried at 70 $\mathrm{^{\circ}C}$ for 36 h.

Characterization techniques

The structural characterizations of grown samples were performed by an X-ray diffractometer (XRD, PANanalytical X'Pert Pro X-ray diffractometer) with CuKa radiation ($\lambda = 1.5418 \text{ Å}$). The intensities were recorded for $20^{\circ} - 80^{\circ}$ with a step scan of 0.02° min⁻¹. For morphological studies, high-resolution transmission electron microscopy (HRTEM, Jeol 2010, JEOL USA, Inc.) was performed. Fourier inferred spectroscopy (FT-IR, Nicolet Instrument Corp.) measurements were carried out in the range of $400-4000$ cm⁻¹. For dielectric and impedance measurements, powder samples were pressed into circular disk-shaped pellets, and silver paste coating was done on the opposite faces of pellets in order to make parallel pellet capacitor geometry with ferrite material as dielectric medium. The dielectric and impedance measurements were carried out between 42 Hz to 5 MHz using LCR HI-Tester (HIOKI 3532-50) at room temperature. Room temperature magnetic hysteresis was measured by Microsense VSM (model EV9) measurement setup.

Results and discussion

Structural and morphological analysis

Figure 1 shows the X-ray diffraction pattern of $Ni₁₋$ $_{x}Co_{x}Fe_{2}O_{4}$ ferrite NPs, which confirms single-phase cubic spinel structure of samples excluding signature of any secondary phase corresponding to any structure. The average crystallite size of grown NPs was determined using Scherer's formula (Batoo and Ansari [2012\)](#page-7-0):

$$
t = \frac{k\lambda}{\beta cos\theta},\tag{1}
$$

where t is the crystallite size, β is the full width at half maximum of the peak (311), and k is an instrumental constant. The average crystallite size was found between 30 nm to 45 nm. The lattice parameter of the samples was calculated using the formula:

$$
a = \frac{(h^2 + k^2 + l^2)^{1/2}}{2\sin\theta},
$$
\n(2)

The lattice parameter of the samples was found between 8.48 and 8.57 \AA , tabulated in Table [1.](#page-3-0) The lattice parameter of undoped sample is less than that

Fig. 1 The XRD pattern of the $Ni_{1-x}Co_xFe_2O_4$ ferrite nanoparticles

reported for bulk $NiFe₂O₄$ (Phadatare et al. [2012\)](#page-8-0). The reduction in lattice parameter may be attributed to the increased degree of inversion, more surface energy, and surface tension which leads to the reduction of lattice constant (Atif and Hasanian [2006;](#page-7-0) Lee et al. [2007\)](#page-8-0). The increase in lattice parameter with increasing doping concentration is attributed to the mismatch of ionic radii where smaller ionic radii Ni^{2+} (0.64 Å) are replaced by larger ionic radii Co^{2+} (0.65 Å), and hence the lattice parameter increases (Phadatare et al. [2012;](#page-8-0) Ata-Allah and Kaiser [2004](#page-7-0)).

Figure [2](#page-3-0) presents the HRTEM micrographs with inset showing average grain size distribution and selective area electron diffraction (SAED) pattern for the compositions $x = 0.0$ and 0.1, respectively. The average sizes of crystallites were found between 30 nm to 45 nm. The NPs were found highly agglomerated with less uniformity in size. The agglomeration of NPs can be because they experience a permanent magnetic moment proportional to their volume (Suryawanshi et al. [1999\)](#page-8-0). In the SAED pattern, distinct diffuse rings are observed that confirm high crystalline order of the samples. The crystallographic d value of 2.512 \AA corresponding to the lattice space of (311) was observed, which corroborates well with that obtained from XRD analysis.

FT-IR

The room temperature FT-IR spectra of the Ni_{1-} $_{x}Co_{x}Fe_{2}O_{4}$ ferrite NPs are shown in Fig. [3](#page-4-0). The spectra were recorded in the range of $400-4,000$ cm⁻¹. The absorption band observed around $1,115 \text{ cm}^{-1}$ is assigned to C–O–H, while the bands observed at 1,636 and 3,453 cm^{-1} are assigned to O–H group. The IR absorption bands of solids observed in the range of $100-1,000$ cm⁻¹ are usually assigned to vibrations of ions in the crystal lattice (Waldron [1955](#page-8-0); Wilson and Wells [1946](#page-8-0)). In case of ferrites, generally two absorption bands appear, one around 600 cm⁻¹: v_I , which is attributed to the stretching vibration of tetrahedral group Fe–O and the second around 400 cm⁻¹: v_2 , which is attributed to the octahedral group complex Fe–O. From IR spectra, it is noted that no shift occurs in band positions, v_1 and v_2 , by increasing Co content (Table [1\)](#page-3-0). It is seen that the intensity of bands increases while broadening remains same, which shows that the samples are more ordered (Ata-Allah and Kaiser [2004](#page-7-0)).

Composition (x)	Crystallite size (nm (± 1))	Lattice parameter (A)	v_1 (cm ⁻¹)	v_2 (cm ⁻¹)	$M_{\rm s}$ (emu/g)
0.0	30	8.48	602.9	414.6	23.19
0.1	35	8.50	602.9	414.6	33.18
0.2	38	8.52	602.9	414.6	38.70
0.3	40	8.53	602.9	414.6	43.97
0.4	45	8.56	602.9	414.6	44.20

Table 1 Grain size, lattice parameters, and FT-IR spectra data of $Ni_{1-x}Co_xFe_2O_4$ ferrite nanoparticles

Fig. 2 a TEM micrograph. b Grain size distribution. c Lattice planes, and d SAED pattern for composition $x = 0.0$

Impedance spectroscopy (IS)

Figure [4](#page-4-0)a, b shows the variation of real and imaginary parts of impedance $(Z'$ and Z'' , respectively) with the frequency between 42 Hz–5 MHz at room temperature. The real and imaginary parts of impedance were calculated according to the following equations (Macdonald [1987](#page-8-0); Suryavanshi et al. [1991](#page-8-0); Arcos et al. [1999](#page-7-0)):

$$
Z' = \frac{R_{g}}{1 + \omega_{g}^{2} C_{g}^{2} R_{g}^{2}} + \frac{R_{gb}}{1 + \omega_{gb}^{2} C_{gb}^{2} R_{gb}^{2}},
$$
(4)

$$
Z^{''} = \frac{-R_{\rm g}^{2}C_{\rm g}\omega_{\rm g}}{1+\omega_{\rm g}^{2}C_{\rm g}^{2}R_{\rm g}^{2}} + \frac{-R_{\rm gb}^{2}C_{\rm gb}\omega_{\rm gb}}{1+\omega_{\rm gb}^{2}C_{\rm gb}^{2}R_{\rm gb}^{2}},
$$
(5)

where the terms R_g and C_g represents the resistance and the capacitance of grain, while R_{gb} and C_{gb} represent the corresponding terms for grain boundary volume. The resistance values for grain and grain boundary have been calculated from the intercepts on real part of $Z(Z)$ axis, whereas the capacitance values have been calculated from the frequency peaks of semicircle arc. Since the value of $Z' = -Z''$ at maximum, we have (Macdonald [1992](#page-8-0); Arcos et al. [1999;](#page-7-0) Zaki and Mansour [2006\)](#page-8-0):

$$
C_{g} = \frac{1}{R_{g}\omega_{g}},\tag{6}
$$

$$
C_{\rm gb} = \frac{1}{R_{\rm gb}\omega_{\rm gb}},\tag{7}
$$

Fig. 3 The FTIR spectra for $Ni_{1-x}Co_xFe_2O_4$ ferrite nanoparticles

Fig. 4 a, **b** The variation of Z' and Z'' with frequency for Ni_{1-x} $Co_xFe₂O₄$ ferrite nanoparticles

A frequency dependent Z' plateau is observed initially in low frequency side followed by a nearly positive slope at high frequency side, indicating a crossover from low frequency relaxation behavior to high frequency dispersion phenomenon. It is seen that Z' value decreases rapidly above 12 kHz with increasing value of frequency. Figure 4b shows a broad imaginary impedance peak (Z'') which moves to high frequency side (step like) with little negative slope and finally disappears above 12 kHz and becomes almost frequency independent. The observed decrease in both Z' and Z'' can be attributed to the increase in conduction mechanism due to hopping of electrons, which increases with increasing frequency of the applied field (Bauerle [1969](#page-7-0); Srivastava et al. [2009](#page-8-0); Ortega et al. [2008\)](#page-8-0).

Nyquist plots for various compositions of $Ni₁$ $_{x}Co_{x}Fe_{2}O_{4}$ ferrite NPs in the frequency range of 42 Hz– 5 MHZ at room temperature are presented in Fig. 5. It is seen that each sample exhibits almost a complete semicircle arc, and the radius of semicircle gradually increases with increasing doping concentration except for 10 % doping where it decreases. The lower total resistance at this composition promotes electron hopping, which is known to be mechanism for both conduction and polarization in ferrites. The grain boundary resistances are larger than grain resistances. This is ascribed to the fact that atomic arrangement near

Fig. 5 The variation of Z'' versus Z' for $Ni_{1-x}Co_xFe_2O_4$ ferrite nanoparticles

grain boundary region is disordered, which increases electron scattering (Dutta et al. [2004\)](#page-8-0). It can be seen that none of the composition exhibit two semicircles corresponding to the grain and grain boundary in Cole–Cole plot, except for $x = 0.1$, which shows emergence of one more semicircle at low frequency region. The second semicircle may be attributed to the electrode polarization. Therefore, we propose that the semicircles corresponding to the grain and grains boundaries overlap and their individual effects cannot be separately evaluated due to sufficiently high grain boundary resistance values, which increase drastically with the progressive increment in doping concentration contrary to the grain resistance values.

Dielectric properties

Figure 6a–c shows real, imaginary part of the dielectric constant, and dielectric loss (ε' , ε'' , and tan δ) plotted against frequency at room temperature. It is seen that all the dielectric parameters (ε' , ε'' , and tan δ) show a strong dispersion with increasing value of frequency and the response becomes almost frequency independent for real and imaginary parts of the dielectric constant at higher frequency, while $tan\delta$ continuously decreases with increasing value of frequency. The dielectric constant of the samples is found exceptionally high while loss factor has been found less than 1 which makes present material as good choice for high frequency/microwave applications. Further, it is seen that no relaxation or Debye peaks are observed in any composition for both real and imaginary parts of dielectric constant, but a weak relaxation phenomenon is noticed in dielectric loss for the compositions, $x = 0.4$ and 0.5, respectively. The observed relaxation behavior can be attributed to the ferromagnetic resonance where matching of electron hopping frequency and the frequency of applied field (Chen et al. [2010\)](#page-8-0) happens.

Ac conductivity

Real conductivity is usually adopted for studying ion motion in ionic oxides or gases. The real conduction in present study has been calculated using Eq. (Abbas et al. [2007](#page-7-0)):

$$
\sigma' = \frac{d\cos\theta}{A|Z|},\tag{8}
$$

Fig. 6 a–c The real, imaginary part of the dielectric constant, and dielectric loss (ε' , ε'' and tan δ) versus frequency for Ni_{1-x} $Co_xFe₂O₄$ ferrite nanoparticles

where d is the sample thickness, A is top electrode area, θ is impedance phase angle, and |Z| is absolute impedance value. The real conductivity spectra are

Fig. 7 The variation of ac conductivity with frequency with *inset* showing variation of exponent with composition for $Ni₁$ $_{x}Co_{x}Fe_{2}O_{4}$ ferrite nanoparticles

hence presented in Fig. 7 as a function of frequency (42 Hz–5 MHz) at room temperature. It may be seen that at high frequency, the conductivity increases exponentially and moves to higher side of frequency window. According to jumping relaxation model (Abbas et al. [2007](#page-7-0); Patange et al. [2009](#page-8-0)), at low frequencies, where the conductivity is constant, transport takes place on infinite paths (dc conductivity). The conductivity in low frequency region is associated with the successful hops to its neighborhood vacant site due to available long time period; such successive jumps result in a long-range transitional motion of ions contributing to dc conductivity. At higher frequencies, the conductivity increases strongly with frequency. The electrical transport phenomenon is dominated by contributions from hopping between infinite clusters (ac conductivity). Two competing relaxation processes can be realized at

Fig. 8 Room temperature hysteresis for $Ni_{1-x}Co_xFe_2O_4$ ferrite nanoparticles

this stage which can be termed as unsuccessful, i.e., when jumping ion jumps back to initial position, and successful, i.e., when neighborhood ions become relaxed with respect to ions position. The increase in ratio of successful to unsuccessful ion hopping results in more dispersive conductivity at higher frequencies as observed in present study (Abbas et al. 2007).

The exponent ' n ' has been calculated as a function of composition by plotting ln σ versus ln ω according using Eq. ([5\)](#page-3-0), which represents straight line with slope equal to the exponent 'n' and intercept equal to $\ln B$ on vertical axis at ln $\omega = 0$.

$$
\sigma_{ac}(\omega, T) = B\omega^n,\tag{9}
$$

where constants B and n are composition and temperature dependent parameters. The inset in Fig. [8](#page-6-0) shows the variation of 'n' as a function of Co^{2+} ion doping. In present study, value of parameter n has been found between 0.104 and 0.177, which corresponds to short range transitional hopping of charges or ac conductivity (Bauerle 1969).

Magnetic properties

Figure [8](#page-6-0) shows the hysteresis loops for first four grown ferrite nanoparticle samples at room temperature. Interestingly, from the hysteresis loops, it is seen that the coercivity and the remanence magnetization are zero for all samples. The samples are not saturated even with the applied magnetic field of 10 KOe, which show the characteristics of superparamagnetism (Ghasemi et al. [2012\)](#page-8-0). The variation of magnetization of ferrite NPs show that up to a specific magnetic field, magnetic moments of core align with the applied field and beyond this field; any increase in the applied field has an effect on the shell of particles, and therefore, the slope of magnetization decreases with increasing applied field (Abbas et al. 2007). Further, from Fig. [8](#page-6-0) it is noted that the magnetization increases with increasing doping concentration of $Co²⁺$ ions. This increase in magnetization is due to replacement of Co^{2+} ions of magnetic moment 2 μ_B with lower magnetic moment $1\mu_B$ cations such as Ni²⁺ ions. A systematic change in the shape of virgin curve is observed, which shows transition from a pinning type magnet in case of pure $NiFe₂O₄$ ferrite due to a large spin disordering (Kodama et al. [1996](#page-8-0)). The tendency for softening and transition to a nucleation type magnetic is seen in the increasing steepness of virgin

curves with increasing doping concentration and becomes maximum for $x = 0.4$ (not shown in graph). Also, this increased value of saturation magnetization can be due to partial replacement of $Fe³⁺$ ions with $Co³⁺$ ions (Ghasemi et al. [2012;](#page-8-0) Thakur et al. [2012](#page-8-0)).

Conclusion

Single phase, high crystalline ordered $\text{Ni}_{1-x}\text{Co}_x\text{Fe}_2\text{O}_4$ ferrite NPs were synthesized through modified citrategel method. The dielectric properties show a non-Debye relaxation mechanism, with 10 % doping having maximum value of dielectric constant. The dielectric loss has been found less than 1 in the grown NPs, which makes synthesized material a good choice for high frequency applications. The hysteresis curves of samples show that the grown ferrite NPs exhibit superparamagnetism. The saturation magnetization was found to increase with progressive substitution of $Co²⁺ ions.$

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