Secretory expression and characterization of a soluble laccase from the *Ganoderma lucidum* strain 7071-9 in *Pichia pastoris*

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Abstract Laccases are strong oxidizing enzymes that oxidize chlorinated phenols, synthetic dyes, pesticides, polycyclic aromatic hydrocarbons as well as a very wide range of other compounds with high redox potential. Based on the bias of genetic codons between fungus and yeast, we synthesized a laccase gene GlLCCI, originated from Ganoderma lucidum using optimized codons and a PCR-based two-step DNA synthesis method. The recombinant laccase, GlLCCI was successfully over-expressed in yeast, Pichia pastoris, with an alcohol oxidase1 promoter. The recombinant GILCCI has a molecular mass of approximately 58 kDa. The $K_{\rm m}$ values of GILCCI for 2-2'-azino-bis-(3-ethylbenzothiazoline-6-sulphonic acid) (ABTS) and guaiacol were 0.9665, and 1.1122 mM, respectively. The V_{max} of GILCCI for both substrates was 3,024 and 82.13 μ M $mg^{-1} min^{-1}$. When ABTS was used as a substrate, the enzyme had an optimal temperature of approximately 55°C. The enzyme was detected over pH values from 2 to 8. The enzyme was strongly activated by K⁺, Na⁺, Cu²⁺ and

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mannitol. Six amino acids (alanine, histidine, glycine, arginine, aspartate and phenylalanine) increased the catalytic ability of the enzyme. The activity of laccase was obviously inhibited by Fe^{2+} , Fe^{3+} , sodium hydrosulphite, and sodium azide. Additionally, under optimal conditions, GlLCCI decolorized 37.62 mg l⁻¹ of azo dye methyl orange (MO) in cultural medium. With a high MO degradation ability, GlLCCI may have potential in the treatment of industrial effluent containing azo dye MO.

Keywords Laccase · *Ganoderma lucidum* · Secretory expression · Protein purification · Decolorization

Introduction

Laccases (EC1.10.3.2), of the blue copper oxidase enzyme family, catalyze reactions for a broad range of substrates. Amino acid-sequence analysis of different laccases indicates that the copper-binding sites are highly conserved [1]. Substrates for laccases vary from organic to inorganic compounds, including aromatic amines, diphenols, and polyphenols catalyzed by a one-electron transfer mechanism [2]. Guaiacol, 2,6-dimethoxyphenol [3], or the electron-rich, non-phenolic substrate 2-2'-azino-bis-(3-ethylbenzothiazoline-6-sulphonic acid) (ABTS) [4] are the most commonly used substrates for enzyme activity measurement. Laccase also possesses functions wherein it decolorizes a range of synthetic dyes, such as, acidic, basic, disperse, azo, diazo, anthraquinone-based and metallic complex dyes [5]. So far, many studies have been reported on several species such as Trametes versicolor [6], Pleurotus eryngii [7], and Phanerochaete chrysosporium [8], but very limited research has been reported for the laccase of the Ganoderma lucidum strain 7071-9.

Ganoderma lucidum, a member of the white-rot basidiomycete family, enjoys special veneration in Asia and has been found to produce chemicals such as hypoglycemic polysaccharides, immunomodulatory [9] and antitumor proteins [10]. In pharmaceutical and medicinal studies and commercially cultivated isolates, it is generally named G. lucidum. This name, however, contains several laccate Ganoderma species that might differ in their bioactive compound composition. In 1996, different strains of G. lucidum were first grouped by internally transcribed spacer sequences [11]. A G. lucidum gene (GlLCC) was first cloned by Joo et al. [12]. However, studies on laccases from the G. lucidum species are limited to the analysis of laccase isoenzymes [13]. In 1999, Joan Lin Cereghino and James M. Cregg elaborated the process of heterologous protein expression in the methylotrophic yeast Pichia pastoris and described the methanol pathway in *P. pastoris* [14].

In this research, levels of codon usage in *G. lucidum* were determined, and *G. lucidum* 7071-9 gene (*GlLcc*) was synthesized through PTDS method, which was a newly developed PCR-based two-step DNA synthesis method for long gene sequences, the first step needed to synthesize individual fragments of the DNA of interest: ten to twelve 60mer oligonucleotides with 20 bp overlap were mixed and a PCR reaction was carried out to produce DNA fragments that were 500 bp in length; the second step needed to synthesize the entire sequence: five to ten PCR products from the first step were combined and used as the template for a second PCR reaction, with the two outermost oligonucleotides as primers. After linked with the secretory expression vector, the gene *GlLCCI* was heterogeneously expressed in *P. pastoris* and then characterized.

Materials and methods

Codon-usage and sequence-data analysis

The original DNA sequence of *GlLCC* gene (GenBank accession No. AF185275) from *G. lucidum* was down-loaded from GenBank (ftp://ftp.ncbi.nih.gov) and were deposited in the software DNAStar v 8.02 earlier. Codon-usage processing was performed with an NCBI-related database at http://www.kazusa.or.jp/codon. The entire 1,563-bp sequence of the *G. lucidum* strain 7071-9 laccase gene was used. The DNAman v 6.0 software was used for relative adaptiveness analysis.

Strains, culture and chemicals

The *Escherichia coli* strain DH5 α (Invitrogen, USA) was used in all plasmid manipulations. All enzymes of DNA or RNA manipulation were purchased from TaKaRa (Dalian,

China). *P. pastoris* strain GS115 (his4) and plasmid pPIC9K were purchased from Invitrogen (San Diego, CA, USA). Yeast extract–peptone–dextrose (YPD), buffered minimal methanol (BMM), buffered minimal glycerol (BMG), minimal dextrose (MD), and other media were prepared following the instructions in the manufacturer's manual in the *Pichia* Expression Kit (Invitrogen, USA). Oligonucleotides of *Gl.* 60 bp each were ordered from the Shanghai Sangon Biological Engineering Technology and Service Co., Ltd (Supplementary Table 1).

Synthesis, cloning and vector construction of the laccase gene

Based on the sequence of the GlLcc gene (GenBank accession No. AF185275), we modified the gene by deleting a 5'-end fragment coding for the N-terminal 22 amino acids signal sequence and added a BamHI and a SacI site at the 5'- and 3'-end, respectively (Supplementary Fig. 1). After codon optimization, we re-synthesized the recombinant gene using the PTDS technique [15] (Fig. 1). The synthesis included two subsequent PCR steps: (1) Thirty-six overlapped PCR primers were added in three PCR tubes, consisting of 50 µl PCR reaction, 200 nM of each outer primer and 20 nM of each inner primer. The thermal cycling conditions were: one cycle of 1 min at 94°C, followed by 30 cycles of denaturation at 94°C for 30 s, annealing at 54°C for 30 s and extension at 72°C for 1.5 min. Followed by an additional 10 min at 72°C to ensured a complete extension of the PCR products; (2) Aliquots of the three PCR products were used as templates for the second-step PCR, to which only the first forward primer (P1) and the last reverse primer

One s	tep									
P1	P3	÷ –	P5	P7	+ -	P9		P11		
-	P2 +	P4	Pe		P8	+	P10	+	P12	
			Block	1 (1-500	bp)					
P13	P15	۰ -	P17	P19		P21		P23	A	
-	P14	P16		P18	P20		P22	• •	P24	
			Block	2 (480-98	Obp)					_
P25	P27	- →	P29	P31	÷ -	P33		P35	-	
-	P26	P28	- + P	30	P32		P34	• •	P36	
			Block	3 (960-15	512bp)					_
Two s	tep									
P1	Block1		Block2			_	Block3			
							_			P36
				GILCCI						

Fig. 1 The PTDS strategy for the synthesis of the *GlLCCI* gene. Oligonucleotides of *Gl.* 60 bp were assembled as the graph described. The *gray* oligonucleotides represent sense oligonucleotides while the *black* oligonucleotides represent anti-sense oligonucleotides. There was a 20 bp overlap for each of the oligonucleotides used

(P36) were added to the reaction. All thermal cycling conditions were the same as the first-step PCR. Thereafter, the PCR product was cloned and sequenced. The synthetic gene of *G. lucidum* strain 7071-9 laccase was named as *GlLCCI*. After digestion by the restriction enzymes, *GlLCCI* was cloned into the secretory expression vector pYM 7909 using a modified pPIC9K (Invitrogen) in which the *SacI* site in the *AOX 1* promoter and *XhoI* site in the G418 cassette were removed by site-directed mutagenesis. Meanwhile, a $6 \times$ His-tag was added after KEX2 protease cleavage at the site of the MF α -signal sequence (Supplementary Fig. 2).

Transformation and screening

The plasmid pYM 7909 was cloned by *E. coli* DH5 α . Three microgram of the purified plasmid pYM 7909 was linearized by digestion with enzyme *Sal*I and the transformed into competent cells of *P. pastoris* through electroporation (Bio-Rad Genepulser, Hercules, USA) at 1,750 V with a 0.2-cm cuvette. After electroporation, *P. pastoris* cells were plated on selective plates (SD, 0.8 M sorbitol, 5% glucose, and 2% agar) and incubated at 28°C for 72 h. The surviving cells were picked up and streaked on MD (1.34% YNB, 0.000004% biotin, 0.5% methanol, and 2% agar) plates. After incubation at 28°C for 48 h, clones that appeared on the MD plates were amplified through liquid BMGY and expression was induced by liquid BMM (1% methanol added), and the activity of the enzyme in supernatants was detected by 10 mM ABTS.

Enzyme expression and purification

Laccase from one positive clone exhibiting high enzymatic activity was extracted by incubating the colony in 250 ml of BMGY liquid (with 1% glucose) at 28°C overnight with continuous shaking (220 rpm). The cells were harvested by centrifugation at 4,000×g, 4°C for 3 min when the OD₆₀₀ of the culture reached 0.8–1. The cell pellets were re-suspended within BMM liquid by rotary shaking (120 rpm), methanol was added (1% final concentration) every 24 h for 3 days; CuSO₄ was also added (1 mM final concentration) to induce the expression of the enzyme. The supernatant was harvested by centrifugation (4,000×g, 4°C, and 3 min).

Ammonium sulphate was added to the clean supernatant. After a centrifugation at 12,000 rpm at 4°C for 15 min, the precipitation was filtered through Microza (filter membrane) and dialyzed supernatant was loaded on to a nickel-charged iminodiacetic acid column pre-equilibrated with 50 mM PBS buffer, containing 0.5 M NaCl, 10% glycerol, 1% Tween, 1% Triston-X100, 0.5% β -ME, DD H₂O at 4°C. The column was first washed with the same buffer containing 10 mM imidazole, and then 150 mM imidazole, finally the $6 \times$ His-tagged enzyme was eluted with the buffer containing 250 mM imidazole. Fractions were collected and dialyzed at 4°C overnight against SD, and fractions containing the laccase activity were pooled subsequently.

Enzyme characterization

The specific activity of the enzyme was measured with ABTS and guaiacol, respectively. The reagents 10 mM ABTS and 0.2 M citrate–phosphate buffer were incubated at 45°C for 10 min prior to starting the reaction with 50 μ l of purified enzyme; reactions were terminated with 50 μ l of 1 M sodium fluoride after 15 min and OD₄₂₀ was immediately measured (TECAN Tecan i-control spectro-photometer). Similarly, 1.37 mM guaiacol and 0.2 M citrate–phosphate buffer were incubated at 45°C for 10 min prior to starting the reaction with 100 μ l of purified enzyme and reactions were terminated after 12 h with 50 μ l of 1 M sodium fluoride and immediately measured the OD at 480 nm.

Specific activity was defined as units per milligram of protein. One unit of laccase activity was defined as the enzyme amount required to catalyze 1 µmol of substrate per min. Suitable amounts of enzyme were used to obtain absorbance of 0.1-1 after approximation of time was used in all photometric measurements applying common test conditions. Protein concentrations were determined using the Bradford method (Bio-Rad protein assay dye reagent protocol). The enzyme molecular mass was estimated by 12% sodium dodecyl sulfate-polyacrylamide gel (SDS-PAGE) electrophoresis compared with the protein markers (Watson Bio-tech, Shanghai, China). Gel staining was performed with Coomassie Brilliant R-250. All assays were performed at least in triplicates. The graphs represented the averages of the values obtained and the error bars show the ranges of values.

The thermo dependence of enzymatic activity was measured between 25 and 45°C with ABTS. Thermostability assays were performed by incubating aliquots of the enzyme at 25, 35, 45 and 55°C for 10 min. Aliquots were removed in a time course and enzyme activity measured with ABTS as described above at 55°C for 15 min. The effect of pH on enzyme activity was determined with ABTS by assaying in the 0.2 M citrate–phosphate buffer ranging from pH 2 to 7. The pH stability was examined by assaying with 1.5 mM ABTS after incubating the enzyme for 24 h in the 0.2 M citrate–phosphate buffer at 4°C. Relative activity was measured as described above.

Effects of metallic ions were examined at two different concentrations (1 and 10 mM). Aliquots of enzyme 0.2 M citrate–phosphate buffer (pH 2.6) containing FeSO₄, FeCl₃, NaCl, CaCl₂, KCl, MgCl₂, ZnSO₄, AlCl₃, MnSO₄, CuSO₄,

SDS, or EDTA were cultured for 1.5 h at 4°C. Activities of the treated enzyme was measured at 55°C for 15 min with 10 mM ABTS. Effects of organic and inorganic compounds were measured in 0.1 and 1 mM two different concentrations after incubation at 4°C for 1.5 h with the following four compounds: sodium thiosulfate, sodium azide, mannitol, and sodium hydrosulphite, or six L-amino acids: alanine, histidine, glycine, arginine, aspartate and phenylalanine. The relative activity was measured by methods described above.

MO decolorization was carried out in citrate–phosphate buffer (pH 4.0, 0.2 M) containing MO at 65 mg⁻¹ l and 0.025 mM ABTS as a redox mediator. After adding 140 µl purified enzyme, enzyme activity of the reaction mixture (400 µl with 5 mM Cu²⁺) was measured at 25, 35, 45 and 55°C from 3 to 48 h. To study the metal ions effects on MO decolorization, different concentrations of K⁺, Mg²⁺, Cu²⁺ and Zn²⁺ were used in a total volume of 400 µl in a test tube. The reaction mixtures were incubated at 35°C for 3 h and the dye decolorization was measured at 470 nm.

Kinetic studies

Kinetic studies were performed using ABTS concentrations from 0.5 to 4 mM. The absorbance at 420 nm was monitored for 15 min at 55°C and guaiacol concentrations from 0.0343 to 0.343 mM, the absorbance at 480 nm was monitored for 12 h at 55°C. The K_m and V_{max} were calculated using the Enzyme Kinetics computer program GraphPad Prism (version 4.0, GraphPad Software Inc., San Diego, California).

Results

Parameterization of relative adaptiveness and codon usage in *Ganoderma lucidum*

Codons from *G. lucidum* were analyzed to determine the codon usage. The total coding GC content in the final modified gene is estimated at 53.79%. The relative adaptiveness of codon usage between *G. lucidum* and *P. pastoris* were calculated (mean difference = 24.76%). After gene modification, a new gene of *GlLCCI* that was more appropriate for expression of *P. pastoris* was designed.

Reconstruction, transformation and screening

Following two-step overlapped PCR amplification of the entire ORF of *GlLCCI*, a unique band of 1,512 bp was obtained (not shown) and sequenced to be correct. The reconstructed pYM 7909 was assessed by *Bam*HI and *SacI*, and two right bands of 1,512 and 9,258 bp were shown on

an agarose gel. The positively transformed *P. pastoris* cells, which contained correct reconstructed plasmids, were successfully selected on MD screen plate. Following activity tests of the crude laccase, 12 out of 1,500 *P. pastoris* colonies were shown to be active with ABTS and one of the vigorous colonies was chosen for further studies.

GILCCI purification and kinetic properties

We purified the enzyme and examined its biochemical properties. The enzyme was stable at 4°C. SDS-PAGE analysis of the recombinant GILCCI enzyme preparation showed that it was at least 85% pure and had an apparent molecular mass of 58 kDa (Fig. 2), which is smaller than that of three laccase isozymes derived from a different strain of *G. lucidum* [13]. The $K_{\rm m}$ values of GILCCI for ABTS and guaiacol were 0.9665 and 1.1122 mM, respectively, and the $V_{\rm max}$ values of the GILCCI at 55°C with a pH 2.6 for both substrates is 3,024 and 82.13 μ M mg⁻¹ min⁻¹, respectively (Table 1).

Effects of pH and temperature on laccase activity and stability

The optimal pH value of the purified GlLCCI enzyme was 2.6 (Fig. 3b), reaching an undetectable level of laccase activity at pH greater than 8.0. When a pH of 9.0 was reached, GlLCCI activity was completely depleted. However, the activity of GlLCCI was still stable after incubation for 24 h at 4°C in 0.2 M citrate–phosphate buffer at pH ranging from 2 to 7; and 65% of activity persisted at pH



Fig. 2 SDS-PAGE of recombinant *GlLCCI* expressed via *P. pastoris. Lane Marker* protein molecular mass marker, *Lane 1* unpurified GlLCCI, *Lane 2*, GlLCCI purified with nickel-charged iminodiacetic acid column at 4°C

 Table 1 Kinetic constants of laccase from Ganoderma lucidum

Substrate	λ_{\max} (nm)	K_m (mM)	$V_{\rm max}~(\mu {\rm M}~{\rm mg}^{-1}{\rm min}^{-1})$	k_{cat} (s ⁻¹)	$k_{\rm cat}/K_{\rm m}~({\rm mM}^{-1}{\rm s}^{-1})$
ABTS	420	0.9665 ± 0.046	$3,024 \pm 0.079$	17.29	17.89
Guaiacol	480	1.1122 ± 0.001	82.13 ± 0.083	0.0098	0.0088

The activity of the enzyme was measured with ABTS and guaiacol, respectively. The enzyme was tested in 0.2 M Na₂HPO₄-citric acid buffer (pH 2.6) at 55°C. Values are means \pm SD of three individual experiments



Fig. 3 Optimal temperature and pH, thermostability and pH stability of GlLCCI. **a** Thermodependence of purified GlLCCI was determined at various temperatures from 35°C to 70°C in increments of 5°C. **b** Effects of pH dependence were measured with 0.2 M Na₂HPO₄– citric acid buffer from pH 2 to 8. **c** Effect of pH on the stability of the recombinant GlLCCI, assayed after incubating the enzyme for 24 h at 4°C. The *gray bar* indicates that the enzyme was more stable at the

7 (Fig. 3d). The thermodependence of activity results showed that a maximum activity with ABTS was observed at 55°C and the enzyme still showed activity at 70°C (Fig. 3a). Slight loss of the activity was shown during the preincubation stage of the enzyme for 1 h at with temperature at 25°C. However, the activity decreased rapidly when pre-incubating at 55°C for 20 min (Fig. 3c).

Effects of metallic ions on enzymatic activity

The effect of metallic ions on laccase activity was assessed with ABTS. Two metal ions concentrations of 1 and 10 mM were chosen for experiments. We noted that laccase activity remained stable in the presence of Mg^{2+} , Mn^{2+} and EDTA at a low concentration of 1 mM. However, 10 mM Mg^{2+} could also increase the enzymatic activity. It is noteworthy that 10 mM K⁺ or Na⁺ enhanced laccase activity to 159.71 and 171.79%, which were higher



optimum pH 2.6. The activity of the enzyme remained at 65% when incubated at pH 7. **d** The thermostability of purified GlLCCI was assayed via incubation of the same volume of purified enzyme at 25, 35, 45 or 55°C with different durations (10-min increment) and determined the remaining activity with ABTS as described above. The specific activity corresponding to 100% activity level was 141.28 U/mg. Values are means \pm SD of three individual experiments

than with Cu^{2+} . Compared with the control, when the concentration of Cu^{2+} increased from 1 to 10 mM, the laccase activity increased by 24.14 and 49.14%, respectively (Table 2). This may be attributed to the sufficient occupation of copper binding sites with copper ions [16]. Both concentrations of Fe²⁺, Fe³⁺, Al³⁺, Zn²⁺ and SDS have been noted to decrease the activity of laccase. With increasing concentration of EDTA, the activity of laccase decreased to 45.85% at 10 mM, and had little effect at 1 mM.

Effects of organic and inorganic compounds on the enzyme activity

The effect of different chemical compounds on enzymatic activity was shown in Fig. 4. The most potent inhibitor for the enzyme was sodium hydrosulphite; when concentration

Treatment	Relative activity				
	Concn (1 mM)	Concn (10 mM)			
None	100	100			
KCl	112.23	159.71			
NaCl	115.49	171.79			
CaCl ₂	111.35	148.25			
MgCl ₂	97.86	144.06			
MnSO ₄	97.69	58.19			
AlCl ₃	49.58	29.64			
ZnSO ₄	91.60	39.14			
CuSO ₄	124.14	149.44			
FeSO ₄	14.42	12.12			
FeCl ₃	15.09	12.96			
EDTA	99.90	73.25			
SDS	13.20	11.6			

Table 2 Effects of metallic ions on laccase activity

Assays were performed in 0.2 M citrate-phosphate buffer (pH 2.6). Control indicates no chemicals added in the reaction system. For each assay, equal amounts of both laccases, based on their activity with ABTS, were used. All values represent the mean of triplicate measurements with a sample mean deviation of less than 3%



Fig. 4 Effects of organic and inorganic compounds effects were measured in gradient concentrations. *1* sodium thiosulfate, 2 sodium azide, 3 mannitol, 4 sodium hydrosulphite, 5 L-alanine, 6 L-histidine, 7 L-glycine, 8 L-arginine, 9 L-aspartate, and *10* L-phenylalanine. The relative activity was measured as described above. Values are means \pm SD of three individual experiments. The significant (*P* < 0.05) and extremely significant (*P* < 0.01) data were annotated as * and **

increased from 0.1 to 1 mM, enzymatic activity decreased from 39.75 to 2.21%. Sodium thiosulfate and sodium azide also inhibited enzymatic activity at 1 mM concentration, but had minimal effect at 0.1 mM. Mannitol increased enzymatic activity to 115.62% at a concentration of 1 mM, but had no effect at 0.1 mM. All of the 1 mM six L-amino acids (alanine, histidine, glycine, arginine, aspartate and phenylalanine) stimulated the activity of the laccase to variable extents, but at the low concentration of 0.1 mM, the effect was not significant.



Fig. 5 Metal ions, temperature and time effects on decolorization of MO. **a** Among the four metal ions used in the reactive system, Cu^{2+} activated the decolorization of MO most effectively at 5 mM, 8.8% of MO was degradable in 3 h. Five multimolar Mg²⁺, 15 mM Zn²⁺ and 20 mM K⁺ decolorized MO by 6.07, 2.77 and 2.41% respectively. **b** During 48 h incubation at four different temperatures, maximum of 57.48% decolorization of MO was observed at 35°C. All values represent the mean of triplicate measurements with a sample mean deviation of less than 3%

Decolorization of methyl orange

The effect of metal ions on laccase mediated decolorization of azo dye MO is shown in Fig. 5. In the presence of metal ions, 5 mM Cu²⁺ caused 8.8% degradation of MO in 3 h, which appeared to be the most effective. Five multimolar Mg^{2+} , 15 mM Zn²⁺ and 20 mM K⁺ could also enhanced decolorization to a various extent (Fig. 5a). During 48 h incubation at four different temperatures, maximum of 57.48% decolorization of MO was observed at 35°C (Fig. 5b). Without ABTS as redox mediator, no decolorization was detected.

Discussion

During the stage of codon-usage adaptive modification, various biological parameters, such as expression level, mutation frequency, protein length and protein translation signals, are affected by codon bias [17]. The bias in the usage of the Ala, Leu, Arg and Gly codons is closer between *G. lucidum* and *P. pastoris*, and this suggest that host codon bias should be correlated to the expression for

maximal heterologous gene expression [18]. After analysis of gene adaptation, we modified the original sequence and synthesized the 1512-bp GlLCCI gene rapidly through PTDS [15], which offered a simple, rapid, high-fidelity and low-cost alternative method of synthesizing and assembling long DNA fragments. P. pastoris was used to secrete foreign protein as the low levels of native proteins [19]. An inducible expression system is particularly important because the strongly regulated promoter (AOX1) is induced by this promoter [20]. Given the secretive signal for "MVKFQSLLSCVTLLFAASAHAG" in pYM 7909, the GILCCI gene successfully expresses in the heterologous host P. pastoris. Compared with the reported molecular weights of most of fungal laccases, ranged from 55 to 90 kDa including carbohydrates [21], a 58 kDa of GILCCI are quite similar to those of other fungal laccases.

Although GILCCI is stable over a wide acidic pH range, the optimal pH value of the purified enzyme was 2.6, which was quite similar to other laccase enzymes from Agaricus blazei (pH 2) [22], Albatrellus dispansus (pH 4) [23], Lentinula edodes (pH 4) [16], and Hericium erinaceum (pH 5) [24], the optimal pH value of the purified enzyme was 2.6 and GILCCI is stable over a wide acidic pH range, maintaining over 65% activity following incubation in buffers ranging from pH 2 to 7 for 24 h, which indicates that the pH-stable GILCCI may be a good candidate for the development of a biotechnological tool. The specific activity of GILCCI on ABTS was 1413-fold greater than that of guaiacol, suggested that the GILCCI had better affinity to ABTS compared with the laccase of Pleurotus sajor-caju [1]. Unfortunately, the V_{max} value of GlLCCI on guaiacol was relatively low [25].

Results of ion studies with the enzyme were similar to those reported for other laccases that were highly sensitive to Fe^{2+} and Fe^{3+} [26]; this may be attributed to the ability of the ions of iron to interact with the electron transportation system of laccase. When a concentration of EDTA was increased, enzyme activity had decreased. This failure to restore activity may be due to an inability to maintain the proper conformation necessary for the reincorporation of metal ions that may be required to ensure structural stability, but not catalytic activity. Surprisingly, high concentrations of K^+ and Na⁺ did not inhibit GILCCI activity but promoted laccase activity distinctly, which digressed with the research data of a laccase from the Melanocarpus albomyces [27]. This can be attributed to the ions of K^+ and Na^+ may be required for laccase catalytic activity, but not for structural stability. A 10 mM high concentration of Mn²⁺, Al³⁺, and Zn²⁺ suppressed enzymatic activity obviously; this may because these ions had been incorporated with the enzyme wherein the active sites were blocked at the stage of pre-incubation.

After an independent-samples t test, we found that all the 1 mM six L-amino acids tested increased enzyme activity of laccase apparently. GILCCI was more sensitive to 1 mM histidine, which demonstrated an ability to increase activity by 26%. Alanine, at 1 mM, enhanced activity by 13% but had no significant effect at a concentration of 0.1 mM. In the research reported by O'Callaghan [28], the appropriate addition of alanine into culture has increased the unitary output of the laccase. The mechanism of L-amino acids increased laccase activity level may correlate with their chemical structure of L-amino acids residue. These chemical structures increased the dielectric constant of water, diminished polarity effect and the Coulomb force which made electriferous substrate and enzyme easier to attract each other.

In conclusion, the results from this study clearly showed that laccase GILCCI from the *G. lucidum* strain 7071-9 is a new member of the growing family of laccase enzymes. We report the purification and characterization of the recombinant laccase, GILCCI from *P. pastoris*, which will be a new expression system. The strong ability of its decolorization of MO suggested that GILCCI can be a potential enzyme for the removal of color from reactive textile dye effluent.

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