ORIGINAL ARTICLE

Synthesis and characterization of novel acyl hydrazones derived from vanillin as potential aldose reductase inhibitors

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Received: 26 July 2022 / Accepted: 3 September 2022 / Published online: 14 September 2022 © The Author(s), under exclusive licence to Springer Nature Switzerland AG 2022

Abstract

In the polyol pathway, aldose reductase (AR) catalyzes the formation of sorbitol from glucose. In order to detoxify some dangerous aldehydes, AR is essential. However, due to the efects of the active polyol pathway, AR overexpression in the hyperglycemic state leads to microvascular and macrovascular diabetic problems. As a result, AR inhibition has been recognized as a potential treatment for issues linked to diabetes and has been studied by numerous researchers worldwide. In the present study, a series of acyl hydrazones were obtained from the reaction of vanillin derivatized with acyl groups and phenolic Mannich bases with hydrazides containing pharmacological groups such as morpholine, piperazine, and tetrahydroisoquinoline. The resulting 21 novel acyl hydrazone compounds were investigated as an inhibitor of the AR enzyme. All the novel acyl hydrazones derived from vanillin demonstrated activity in nanomolar levels as AR inhibitors with IC_{50} and K_I values in the range of 94.21 \pm 2.33 to 430.00 \pm 2.33 nM and 49.22 \pm 3.64 to 897.20 \pm 43.63 nM, respectively. Compounds **11c and 10b** against AR enzyme activity were identifed as highly potent inhibitors and showed 17.38 and 10.78-fold more efectiveness than standard drug epalrestat. The synthesized molecules' absorption, distribution, metabolism, and excretion (ADME) efects were also assessed. The probable-binding mechanisms of these inhibitors against AR were investigated using molecular-docking simulations.

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Graphical abstract

Keywords Aldose reductase · Acyl hydrazones · Vanillin · Epalrestat · In silico study · ADME-Tox · Molecular docking

Introduction

The acyl hydrazone skeleton is an important intermediate used in the synthesis of interesting biologically active heterocyclic compounds and is also used as a ligand in the formation of metal complexes [[1,](#page-16-0) [2](#page-16-1)]. Hydrazone-type compounds containing azomethine protons constitute a signifcant class of compounds for novel drug research [[3\]](#page-16-2), such as anti-antimicrobial [[4\]](#page-16-3), infammatory [[5](#page-16-4)], antioxidants [\[6\]](#page-16-5), antitubercular [[7\]](#page-16-6), analgesic [\[8\]](#page-16-7), anti-candida [[9](#page-16-8)], α-glucosidase [[10\]](#page-16-9), anticancer [\[11\]](#page-16-10), and antiproliferative [\[12\]](#page-16-11) activities in medicinal felds [\[13](#page-16-12)[–15\]](#page-16-13). Acyl hydrazone fragments bound to heterocyclic systems were displayed to provide enhanced activity. This bioactivity of acyl hydrazones is explained by their tendency to form a hydrogen bond with the molecular target [\[16](#page-16-14)[–18\]](#page-16-15). In addition, it has been reported by studies in the literature that acyl hydrazones have lower toxicity than hydrazides due to the blockage effects of $NH₂$ groups. These research fndings further increased the importance of synthesizing acyl hydrazone-derived compounds [[19\]](#page-16-16).

A three-component Mannich condensation builds a class of compounds known as Mannich bases on one pot reaction of primary or secondary amine, an aldehyde reagent, and structurally diverse substrates containing at least one active hydrogen atom [[20\]](#page-16-17). Mannich base also acts as essential pharmacophores for synthesizing novel compounds in medicinal chemistry [\[21\]](#page-17-0). Examples of clinically useful Mannich bases containing aminoalkyl scaffold in their structure are drugs such as trihexyphenidyl, procyclidine, ranitidine, and biperiden [\[22](#page-17-1), [23](#page-17-2)]. Phenolic compounds are one of the most critical carbons Mannich bases because they contain active hydrogen. In the last decades, there have been many studies on the biological activities of phenolic Mannich bases, such as cytotoxic [[24](#page-17-3), [25\]](#page-17-4), anticancer [[26,](#page-17-5) [27](#page-17-6)], anticonvulsant [\[28](#page-17-7)], anti-infammatory [[29\]](#page-17-8), antifungal [\[30](#page-17-9)], and carbonic anhydrase inhibitory activities [\[31,](#page-17-10) [32\]](#page-17-11).

Morpholine and piperazines are privileged backbones that are widely used as a core structural element or substituent in efective drugs such as Noroxin (antibiotic), Clozaril (an atypical antipsychotic medication), Iressa (breast, lung, and other cancers), and Moclobemide (depression and social anxiety) [[33,](#page-17-12) [34\]](#page-17-13). They can also improve the pharmacokinetic features of molecules, such as metabolic sta-bility and solubility in water [[35\]](#page-17-14). Another core structure, tetrahydroisoquinoline (THQ), forms the main backbone of many natural products (saframycin, naphthyridinomycin/ bioxalomycin, and quinocarcin/tetrazomine) and bioactive compounds [\[36](#page-17-15)]. From this perspective, with the potential of the drugs, it would be helpful to design and synthesize some novel acyl hydrazone derivatives incorporating piperazine, morpholine, and tetrahydroquinoline and screen them for potential biological activities.

Aldose reductase (AKR1B1, AR with EC number 1.1.1.2)1 is the frst rate-limiting enzyme in the polyol pathway, belongs to the Aldo–keto reductase superfamily, and is a monomer comprising 315 amino acid residues [\[37](#page-17-16)[–41](#page-17-17)]. This overproduction of the AR and sorbitol dehydrogenase on the polyol pathway and depletion in reduced NADP⁺ and the oxidized NAD+, which are cofactors of this process, causes various metabolic processes disturbances such as the nephropathy, retinopathy, cataracts, and neuropathy [[42](#page-17-18)[–46](#page-17-19)]. The aforementioned metabolic abnormalities are the primary targets of diabetic complications in those tissues involved in insulin-independent glucose uptake and are responsible for early tissue damage in the organs [\[47](#page-17-20)[–53\]](#page-18-0).

We investigated the synthesis, characterization, and biological activity of a series of novel acyl hydrazones to uncover novel AR inhibitors in the current work. In addition, we conducted in silico experiments, including absorption, distribution, metabolism, and excretion (ADME), density functional theory (DFT), and molecular docking, to evaluate the inhibitory mechanisms of those compounds against the target mentioned above, AR.

Experimental

Chemistry

The chemicals used in this study were supplied from Sigma Aldrich (Germany). Melting points were determined on WRS-2A Microprocessor Melting-point Apparatus and are uncorrected. IR spectra of compounds were recorded using ALPHA-P BRUKER FT-IR Spectrophotometer.¹H NMR spectra were recorded on Bruker (400 MHz) spectrometer. $13C$ NMR spectra were recorded on Bruker (100 MHz) spectrometer. Chemical shifts are reported as δ in ppm relative to tetramethylsilane (TMS) (δ 0.00 singlets) in deuterated chloroform $(CDCl₃)$. High-resolution mass spectrometry measurements were recorded on Agilent 6530 Accurate-Mass Q-TOF LC/MS.

General procedure for synthesis of compounds 2a–b

Synthesis of **2a–b** was performed according to the previously reported method [[54](#page-18-1)].

4‑Formyl‑2‑methoxyphenyl furan‑2‑carboxylate (2a)

White solid, yield 83%, mp: 101–103 °C (lit. 101–103 °C) [[54](#page-18-1)].

4‑Formyl‑2‑methoxyphenyl thiophene‑2‑carboxy‑ late (2b)

White solid, yield: 81%, mp: 89–91 °C. IR (ATR, cm⁻¹): *ν*max 3108, 3070, 2843, 1744, 1682, 1456, 1262, 1054, 854, 724. ¹H NMR (400 MHz, CDCl₃, δ/ppm): δ 10.0 (s, 1H), 8.0 (d, *J* = 3.8 Hz, 1H), 7.7 (d, *J* = 5.0 Hz, 1H), 7.5 (dt, *J* = 7.9, 1.6 Hz, 2H), 7.4 (d, *J* = 7.9 Hz, 1H), 7.2 (d, $J=4.9$ Hz, 1H), 3.9 (s, 3H). ¹³C NMR (100 MHz, CDCl₃, δ/ppm): δ 191.0 159.5, 152.2, 144.7, 135.7, 135.2, 133.9, 132.0, 128.2, 124.7, 123.6, 111.0, 56.2. HRMS (Q-TOF) m/z calcd for $C_{15}H_{14}O_5S$ [M + H]⁺: 262.0323, Found: 262.0341.

General procedure for synthesis of compounds 3a–e

Synthesis of **3a–e** was carried out according to the previously reported method [[55](#page-18-2)].

4‑Hydroxy‑3‑methoxy‑5‑(morpholinomethyl)benza‑ ldehyde (3a)

White solid; yield: 85%, mp: 99–100 °C; IR (ATR, cm^{-1}) *ν*max 2945, 2866, 2829, 2733, 1647, 1592, 1270, 1120, 868, 705; ¹H NMR (400 MHz, CDCl₃) δ 9.79 (s, 1H), 7.36 (m, 1H), 7.19 (m, 1H) 3.95 (s, 3H), 3.82 (s, 2H), 3.78 (m, 4H), 2.63 (m, 4H); ¹³C NMR (100 MHz, CDCl₃) δ 190.6, 153.5, 148.6, 128.5, 125.5, 120.3, 109.8, 66.6, 61.1, 56.0, 52.7.

4‑Hydroxy‑3‑methoxy‑5‑((4‑phenylpiperazin‑1‑yl) methyl)benzaldehyde (3b)

White solid; yield: 90%, mp: 156 °C (lit: 156–157 °C) [\[56\]](#page-18-3); IR (ATR, cm⁻¹) ν _{max} 2959, 2938, 2827, 2737, 1677, 1586, 1315, 1235, 1141, 760, 691; ¹H NMR (400 MHz, CDCl3) δ10.90 (brs, 1H), 9.81 (s, 1H), 7.39–7.38 (m, 1H), 7.32–7.27 (m, 2H), 7.22–7.17 (m, 1H) 6.95–6.86 (m, 3H), 3.96 (s, 3H), 3.89 (s, 2H), 3.28 (m, 4H), 2.80 (m,4H); 13C NMR (100 MHz, CDCl₃)δ 190.6, 153.7, 150.7, 148.7, 129.2, 128.4, 125.5, 120.6, 120.4, 116.5, 109.9, 60.8, 56.0, 52.5, 49.2.

4‑Hydroxy‑3‑methoxy‑5‑((3‑methylpiperidin‑1‑yl) methyl)benzaldehyde (3c)

Light brown solid; yield: 88%, mp: 142–144 °C; IR (ATR, cm⁻¹) ν_{max} 2946, 2922, 2853, 2748, 1651, 1592, 1271, 1147, 864, 707; ¹H NMR (400 MHz, CDCl₃) δ11.55 (s, 1H), 9.76 (s, 1H), 7.33 (m, 1H), 7.15 (m, 1H) 3.94 (s, 3H), 3.78 (m, 2H), 2.96–2.90 (m, 2H), 2.11 (t, *J*=10.5 Hz, 1H), 1.83–1.58 (m, 5H), 0.98–0.95 (m, 1H), 0.89 (d, $J=6.3$ Hz, 3H), ¹³C NMR (100 MHz, CDCl₃)δ 190.6, 154.8, 148.6, 127.8, 125.4, 120.8, 109.5, 61.2, 60.7, 55.9, 53.2, 32.2, 31.0, 25.0, 19.2.

3‑((3,4‑Dihydroisoquinolin‑2(1H)‑yl)methyl)‑4‑hy‑ droxy‑5‑methoxybenzaldehyde (3d)

Light yellow solid; yield: 88%, mp: 181–183 °C; IR (ATR, cm⁻¹) ν_{max} 3053, 2956, 2817, 2750, 1649, 1590, 1274, 749; ¹H NMR (400 MHz, DMSO) δ 9.79 (s, 1H), 7.45 (s, 1H), 7.37 (s, 1H), 7.24–6.89 (m, 4H), 3.89 (s, 2H), 3.86 (s, 3H), 3.69 (s, 2H), 2.86 (t, *J*=5.4 Hz, 2H), 2.80 (t, *J*=5.4 Hz, 2H); ¹³C NMR (101 MHz, DMSO) δ 191.0, 152.7, 148.0, 133.8, 133.5, 128.5, 127.8, 126.5, 126.3, 125.7 (2C), 122.9, 109.8, 57.5, 55.7, 54.7, 49.7, 28.2.

N‑ethyl‑4‑(5‑formyl‑2‑hydroxy‑3‑methoxybenzyl) piperazine‑1‑carboxamide (3e)

White solid, yield 80%, mp: 86–87 °C; IR (ATR, cm⁻¹) ν_{max} 2978, 2948, 2823, 2735, 1735, 1681, 1589, 1237, 1138, 863, 694; ¹H NMR (400 MHz, CDCl₃) δ 9.79 (s, 1H), 7.36 (m,

1H), 7.18 (m, 1H), 4.15 (q, *J*=7.1 Hz, 2H), 3.95 (s, 3H), 3.83 (m, 1H), 3.56 (m, 4H), 2.58 (m, 4H), 1.27 (t, *J*=7.1 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 190.6, 155.2, 153.4, 148.6, 128.5, 125.4, 120.4, 109.9, 61.6, 60.8, 56.0, 52.2, 43.4, 43.3, 14.6.

General procedure for synthesis of compounds 6a–c

In a 50 mL round-bottom fask, the related secondary amine (10 mmol) and triethylamine (11 mmol, 1.11 g, 1.53 mL) were dissolved in 20 mL of THF and the solution was then put into an ice bath. Ethylchloroacetate (10 mmol, 1.23 g) in 20 mL of THF was added to this solution dropwise and stirred for three hours at room temperature. After completion, the solvent was removed under reduced pressure. It was washed with cold water to remove the triethylammonium chloride salt from the oily mixture obtained. Then the crude product was dissolved in 20 mL of ethanol and hydrazinium hydroxide (80%, 25 mmol) was added to the solution. The reaction mixture was refuxed for two hours. The solvent was removed under reduced pressure and the product was washed with cold water. The crude product was used for next step without any purifcation.

2‑(4‑Phenylpiperazin‑1‑yl)acetohydrazide (6a)

White solid, yield 88%, mp: 76–78 °C (lit. 75 °C) [[57\]](#page-18-4).

2‑(3,4‑Dihydroisoquinolin‑2(1H)‑yl)acetohydrazide (6b)

White solid, yield 92%, mp: 85–87 \degree C [\[58](#page-18-5)]. ¹H NMR (400 MHz, DMSO) δ 8.95 (s, 1H), 7.22–6.87 (m, 4H), 4.25 (brs, 2H), 3.61 (s, 2H), 3.10 (s, 2H), 2.82 (t, *J*=5.8 Hz, 2H), 2.71 (t, *J*=5.9 Hz, 2H). 13C NMR (101 MHz, DMSO) δ 168.3, 134.6, 133.9, 128.4, 126.2, 125.9, 125.4, 59.7, 55.2, 50.6, 28.6.

2‑Morpholinoacetohydrazide (6c)

White solid, yield 75%, mp: 95–98 °C (lit. 99–101 °C) [\[59](#page-18-6)].

General procedure for synthesis of compounds 7a–c, 11a–c, 12a–c, and 13a–c

The corresponding aldehyde **2a–b, 3a–e** (10 mmol) and acetohydrazide derivative **6a–c** (10 mmol) were dissolved in absolute ethanol (20 mL), and 4–5 drops of acetic acid was added. Reaction mixture was refuxed for 1–2 h. Reaction was monitored by TLC. After completion, half of the solvent volume was removed under reduced pressure. The mixture was left in the freezer overnight, and the formed solid was filtered off. The crude product was recrystallized from ethanol.

2‑Methoxy‑4‑((2‑(2‑(4‑phenylpiperazin‑1‑yl)acetyl) hydrazono)methyl)phenylfuran‑2‑carboxylate (7a)

Beige solid, yield 77%, mp: 178–180 °C, IR (ATR, cm⁻¹) ν_{max} 3205, 3065, 2936, 1729, 1657, 1597, 1232, 1070, 745; 1 H NMR (400 MHz, CDCl3) δ 10.21 (s, 1H), 8.25 (s, 1H), 7.69 (s, 1H), 7.57 (s, 1H), 7.42 (d, *J*=3.4 Hz, 1H), 7.33–7.28 (m, 3H), 7.25–7.14 (m, 2H), 7.01–6.84 (m, 2H), 6.61 (dd, *J*=3.4, 1.6 Hz, 1H), 3.89 $(s, 3H), 3.35-3.33$ (m, $J=10.8$ Hz, 6H), 2.82 (brs, 4H). ¹³C NMR (101 MHz, CDCl₃) δ 166.0, 156.2, 151.8, 150.9, 148.0, 147.3, 143.6, 141.2, 132.7, 129.2, 123.1, 121.8, 120.2, 119.8, 116.3, 112.3, 109.9, 61.0, 56.2, 53.6, 49.3. HRMS (Q-TOF) m/z: $[M+H]^{+}$ calcd for $C_{25}H_{26}N_{4}O_{5}$, 463.1981; found 463.1977.

4‑((2‑(2‑(3,4‑Dihydroisoquinolin‑2(1H)‑yl)acetyl) hydrazono)methyl)‑2‑methoxyphenyl furan‑2‑car‑ boxylate (7b)

Beige solid, yield 81%, mp: 105–107 °C, IR (ATR, cm⁻¹) *ν*max 3145, 3028, 2922, 1742, 1668, 1546, 1270, 1077, 746; ¹H NMR (400 MHz, CDCl₃) δ 10.30 (s, 1H), 8.13 (s, 1H), 7.68 (s, 1H), 7.56 (s, 1H), 7.40 (d, *J*=3.4 Hz, 1H), 7.25–7.10 (m, 5H), 7.05 (d, *J*=6.0 Hz, 1H), 6.60 (dd, *J*=3.5, 1.7 Hz, 1H), 3.86 (s, 3H), 3.79 (s, 2H), 3.38 (s, 2H), 3.00 (t, *J*=5.6 Hz, 2H), 2.91 (t, $J = 5.7$ Hz, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 166.5, 156.2, 151.7, 147.8, 147.3, 143.6, 141.1, 133.5, 128.8, 126.7, 126.6, 126.0, 123.0, 121.8, 119.8, 112.3, 109.9, 61.0, 56.2, 51.6, 29.2. HRMS (Q-TOF) m/z: $[M+H]^{+}$ calcd for $C_{24}H_{23}N_3O_5$, 434.1716; found 434.1710.

2‑Methoxy‑4‑((2‑(2‑morpholinoacetyl)hydrazono) methyl)phenyl furan‑2‑carboxylate (7c)

Beige solid, yield 75%, mp: 114–116 °C, IR (ATR, cm^{-1}) *ν*max 3153, 3108, 2955, 1743, 1658, 1572, 1267, 1067, 745; ¹H NMR (400 MHz, CDCl₃) δ 10.15 (s, 1H), 8.22 (s, 1H), 7.68 (s, 1H), 7.53 (s, 1H), 7.40 (d, *J* = 3.3 Hz, 1H), 7.18 (m, 2H), 6.60 (d, *J*= 4.9 Hz, 1H), 3.86 (s, 3H), 3.77 (t, *J* = 4.4 Hz, 4H), 3.21 (s, 2H), 2.62 (t, *J* = 4.4 Hz, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 166.0, 156.2, 151.7, 148.0, 147.3, 143.5, 141.1, 132.7, 123.0, 121.7, 119.9, 112.3, 109.9, 66.8, 61.4, 56.2, 53.9. HRMS (Q-TOF) m/z: $[M + H]^{+}$ calcd for $C_{19}H_{21}N_{3}O_{6}$, 388.1509; found 388.1503.

2‑Methoxy‑4‑((2‑(2‑(4‑phenylpiperazin‑1‑yl)acetyl) hydrazono)methyl)phenyl thiophene‑2‑carboxylate (8a)

White solid, yield 79%, mp: 175–177 °C, IR (ATR, cm−1) *ν*max 3193, 3050, 2929, 1734, 1654, 1587, 1253, 1071, 755; ¹H NMR (400 MHz, CDCl₃) δ 10.21 (s, 1H), 8.24 (s, 1H), 8.00 (d, *J*=3.8 Hz, 1H), 7.69 (d, *J*=5.0 Hz, 1H), 7.57 (s, 1H), 7.38–7.25 (m, 3H), 7.26–7.13 (m, 3H), 7.03–6.90 (m, 3H), 3.89 (s, 3H), 3.30–3.28 (m, 6H), 2.81 (t, *J*=4.0 Hz, 4H). 13C NMR (100 MHz, CDCl3) δ 166.0, 159.9, 151.9, 150.9, 148.0, 141.6, 134.9, 133.7, 132.2, 129.2, 128.1, 123.1, 121.8, 120.2, 116.3, 110.0, 61.0, 56.3, 53.6, 49.3. HRMS (Q-TOF) m/z: $[M + H]$ ⁺ calcd for C₂₅H₂₆N₄O₄S, 479.1753; found 479.1747.

4‑((2‑(2‑(3,4‑Dihydroisoquinolin‑2(1H)‑yl)acetyl) hydrazono)methyl)‑2‑methoxyphenyl thio‑ phene‑2‑carboxylate (8b)

White solid, yield 82%, mp: 110–112 °C, IR (ATR, cm^{-1}) *ν*max 3172, 3073, 2915, 1734, 1661, 1561, 1251, 1089, 734; ¹H NMR (400 MHz, CDCl₃) δ 10.34 (s, 1H), 8.13 (s, 1H), 7.99 (d, *J*=4.7 Hz, 1H), 7.67 (d, *J*=5.0 Hz, 1H), 7.55 (s, 1H), 7.25–7.12 (m, 6H), 7.05 (d, *J*=6.2 Hz, 1H), 3.86 (s, 3H), 3.81 (s, 2H), 3.39 (s, 2H), 3.00 (t, *J*=5.5 Hz,

2H), 2.92 (t, $J = 5.7$ Hz, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 166.4, 159.9, 151.8, 147.9, 141.6, 134.9, 133.7, 132.3, 128.8, 128.1, 126.7, 126.6, 126.1, 123.1, 121.8, 109.9, 61.0, 56.2, 51.6, 29.1. HRMS (Q-TOF) m/z: $[M + H]$ ⁺ calcd for $C_{24}H_{23}N_3O_4S$, 450.1488; found 450.1483.

2‑Methoxy‑4‑((2‑(2‑morpholinoacetyl)hydrazono) methyl)phenyl thiophene‑2‑carboxylate (8c)

White solid, yield 77%, mp: 140–142 °C, IR (ATR, cm^{-1}) *ν*max 3177, 3053, 2956, 1735, 1655, 1572, 1247, 1082, 740; 1 H NMR (400 MHz, CDCl₃) δ 10.18 (s, 1H), 8.20 (s, 1H), 7.99 (d, *J*=4.0 Hz, 1H), 7.68 (d, *J*=4.0 Hz, 1H), 7.21–7.16 (m, 3H), 3.85 (s, 3H), 3.77 (t, *J*=4.4 Hz, 4H), 3.22 (s, 2H), 2.62 (t, $J=4.4$ Hz, 4H). ¹³C NMR (10o MHz, CDCl₃) δ 166.0, 160.0, 151.8, 148.1, 141.6, 135.0, 133.7, 132.6, 132.5, 128.1, 123.1, 121.7, 110.0, 66.8, 61.4, 56.2, 53.9. HRMS (Q-TOF) m/z: $[M+H]^{+}$ calcd for $C_{19}H_{21}N_{3}O_{5}S$, 404.1280; found 404.1274.

N′**‑(4‑hydroxy‑3‑methoxy‑5‑(morpholinomethyl) benzylidene)‑2‑(4‑phenylpiperazin‑1‑yl) acetohy‑ drazide (9a)**

White solid, yield 88%, mp: 187–189 °C, IR (ATR, cm⁻¹) ν_{max} 3187, 3060, 2942, 1660, 1594, 1258, 1077, 761; ¹ H NMR (400 MHz, CDCl3) δ 10.06 (s, 1H), 8.08 (s, 1H), 7.30–7.26 (m, 3H), 7.00 (s, 1H), 6.94–6.87 (m, 3H), 3.90 (s, 3H), 3.75–3.72 (m, 6H) 3.25 (brs, 4H), 2.76 (t, J = 4.0 Hz, 4H), 2.58 (brs, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 165.8, 150.9, 149.7, 148.8, 148.3, 129.2, 124.6, 121.8, 120.6, 116,2, 109.4, 66.7, 61.0, 56.1, 53.6, 52.8, 49.3. HRMS (Q-TOF) m/z: $[M+H]^{+}$ calcd for $C_{25}H_{33}N_{5}O_{4}$, 468.2611; found 468.2605.

2‑(3,4‑Dihydroisoquinolin‑2(1H)‑yl)‑N'‑(4‑hy‑ droxy‑3‑methoxy‑5‑(morpholinomethyl)ben‑ zylidene) acetohydrazide (9b)

White solid, yield 91%, mp: 209–211 °C, IR (ATR, cm⁻¹) *ν*_{max} 3186, 3056, 2960, 1656, 1592, 1268, 1080, 742; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 10.13 (s, 1H), 7.99 (s, 1H), 7.26 (s, 1H), 7.21–7.16 (m, 3H), 7.05 (s, 1H), 6.97 (s, 1H), 3.90 (s, 3H), 3.78 (s, 2H), 3.75 (t, *J*=4.0 Hz, 4H) 3.72 (s, 2H), 3.36 (s, 2H), 2.99 (t, *J*=5.5 Hz, 2H), 2.90 (t, *J*=5.5 Hz, 2H), 2.58 (brs, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 166.0, 149.7, 148.7, 148.3, 133.8, 133.4, 128.8, 126.7, 126.6, 126.0, 124.7, 121.8, 120.6, 109.4, 66.7, 61.0, 56.2, 52.8, 51.6, 29.2. HRMS (Q-TOF) m/z: $[M+H]^{+}$ calcd for $C_{24}H_{30}N_{4}O_{4}$, 439.2345; found 439.2339.

N′**‑(4‑hydroxy‑3‑methoxy‑5‑(morpholinomethyl) benzylidene)‑2‑morpholinoacetohydrazide (9c)**

White solid, yield 85%, mp: 183–185 °C, IR (ATR, cm⁻¹) *ν*max 3206, 3045, 2956, 1664, 1591, 1267, 1078; 1 H NMR (400 MHz, CDCl₃) δ 9.98 (s, 1H), 8.05 (s, 1H), 7.22 (s, 1H), 6.95 (s, 1H), 3.86 (s, 3H), 3.72 (t, *J*= 4.4 Hz, 8H), 3.15 (s, 2H), 2.56 (t, *J*=4.4 Hz, 8H). 13C NMR (100 MHz, CDCl3) δ 165.7, 149.7, 148.8, 148.2, 124.6, 121.8, 120.6, 109.3, 66.9, 66.6, 61.4, 61.1, 56.0, 53.8, 52.8. HRMS (Q-TOF) m/z: $[M + H]^{+}$ calcd for $C_{19}H_{28}N_{4}O_{5}$, 393.2138; found 393.2132.

N′**‑(4‑hydroxy‑3‑methoxy‑5‑((4‑phenylpiperazin‑1 ‑yl)methyl)benzylidene)‑2‑(4‑phenylpiperazin‑1‑yl) acetohydrazide (10a)**

White solid, yield 91%, mp: 208–210 °C, IR (ATR, cm^{-1}) *ν*max 3192, 2997, 2935, 1659, 1594, 1225, 1081, 762; ¹ H NMR (400 MHz, CDCl₃) δ 10.06 (s, 1H), 8.11 (s, 1H), 7.33–7.27 (m, 5H), 7.05 (s, 1H), 6.97–6.90 (m, 6H), 3.94 (s, 3H), 3.81 (s, 2H), 3.28–3.26 (m, 10H), 2.79 (t, $J = 4.4$ Hz, 8H). ¹³C NMR (100 MHz, CDCl₃) δ 165.8, 150.9, 150.8, 149.9, 148.9, 148.3, 129.2, 124,6, 121.8, 120.9, 120.4, 120.2, 116.5, 116.2, 109.4, 61.1, 60.8, 56.1, 53.6, 52.5, 49.3, 49.2. HRMS (Q-TOF) m/z: $[M+H]$ ⁺ calcd for $C_{31}H_{38}N_6O_3$, 543.3084; found 543.3080.

2‑(3,4‑Dihydroisoquinolin‑2(1H)‑yl)‑*N*′**‑(4‑hydrox y‑3‑methoxy‑5‑((4‑phenylpiperazin‑1‑yl)methyl) benzylidene)acetohydrazide (10b)**

White solid, yield 93%, mp: 214–216 °C, IR (ATR, cm−1) *ν*_{max} 3203, 3067, 2916, 1666, 1595, 1226, 1078, 739; ¹H NMR (400 MHz, CDCl₃) δ 10.14 (s, 1H), 8.01 (s, 1H), 7.31–7.26 (m, 3H), 7.22–7.17 (m, 3H), 7.05 (s, 1H), 7.01 (s, 1H), 6.94–6.90 (m, 3H), 3.93 (s, 3H), 3.80 (s, 4H), 3.38 (s, 2H), 3.26 (brs, 4H), 3.00 (t, *J*=5.5 Hz, 2H), 2.91 (t, *J*=5.5 Hz, 2H), 2.76 (brs, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 166.0, 150.8, 149.8, 148.7, 148.3, 133.8, 133.5, 129.2, 128.8, 126.7, 126.6, 126.0, 124.6, 120.9, 120.3, 116.4, 109.4, 61.0, 60.8, 56.2, 56.1, 52.5, 51.6, 49.2, 29.3. HRMS (Q-TOF) m/z: $[M + H]^{+}$ calcd for $C_{30}H_{35}N_{5}O_{3}$, 514.2818; found 514.2810.

N′**‑(4‑hydroxy‑3‑methoxy‑5‑((4‑phenylpiperazi n‑1‑yl)methyl)benzylidene)‑2‑morpholinoaceto hydrazide (10c)**

White solid, yield 93%, mp: 188–190 °C, IR (ATR, cm^{-1}) *ν*_{max} 3210, 3068, 2946, 1662, 1596, 1226, 1076, 758; ¹H NMR (400 MHz, CDCl₃) δ 10.00 (s, 1H), 8.10 (s, 1H), 7.28–7.24 (m, 3H), 7.01 (s, 1H), 6.92–6.86 (m, 3H), 3.91 (s, 3H), 3.81–3.75 (m, 6H), 3.24 (brs, 4H), 3.19 (s, 2H), 2.73 (brs, 4H), 2.60 (t, $J=4.4$ Hz, 4H). ¹³C NMR (100 MHz, CDCl3) δ 165.7, 150.8, 149.9, 148.9, 148.3, 129.2, 124.5, 121.8, 120.9, 120.3, 116.4, 109.3, 66.9, 61.5, 60.8, 56.1, 53.9, 52.4, 49.2. HRMS (Q-TOF) m/z: $[M+H]^{+}$ calcd for $C_{25}H_{33}N_5O_4$, 468.2611; found 468.2603.

N′**‑(4‑hydroxy‑3‑methoxy‑5‑((3‑methylpiperidin‑1 ‑yl)methyl)benzylidene)‑2‑(4‑phenylpiperazin‑1‑yl) acetohydrazide (11a)**

Beige solid, yield 82%, mp: 134–136 °C, IR (ATR, cm−1) *ν*max 3208, 3064, 2926, 1657, 1593, 1232, 1076, 760; ¹ H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 10.04 (s, 1H), 8.06 (s, 1H), 7.30–7.24 (m, 3H), 6.99 (s, 1H), 6.95–6.87 (m, 3H), 3.90 (s, 3H), 3.69 (d, *J*=7.2 Hz, 2H), 3.25–3.23 (m, 7H), 2.89–2.85 (m, 2H), 2.76 (t, *J*=4.4 Hz, 4H), 2.05 (t, *J*=10.0 Hz, 1H), 1.75–1.69 (m, 4H), 0.95–0.86 (m, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 165.8, 150.9, 150.6, 149.1, 148.2, 129.2, 124.0, 121.6, 121.2, 120.1, 116.2, 109.1, 61.2, 61.0, 60.8, 56.1, 53.6, 53.2, 49.3, 32.3, 31.0, 19.4. HRMS (Q-TOF) m/z: $[M+H]^{+}$ calcd for $C_{27}H_{37}N_{5}O_{3}$, 480.2975; found 480.2967.

2‑(3,4‑Dihydroisoquinolin‑2(1H)‑yl)‑*N*′**‑(4‑hydrox y‑3‑methoxy‑5‑((3‑methylpiperidin‑1‑yl)methyl) benzylidene)acetohydrazide (11b)**

Beige solid, yield 85%, mp: 170–172 °C, IR (ATR, cm−1) *ν*max 3201, 3070, 2923, 1661, 1591, 1227, 1081, 738; ¹ H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 10.10 (s, 1H), 7.97 (s, 1H), 7.22 (s, 1H), 7.19–7.14 (m, 3H), 7.03 (d, *J*=4.4 Hz, 1H), 6.95 (s, 1H), 3.88 (s, 3H), 3.76 (s, 2H), 3.67 (d, *J*=7.2 Hz, 2H), 3.33 (s, 2H), 2.98–2.85 (m, 7H), 2.03 (t, *J*=10.0 Hz, 1H), 1.74–1.72 $(m, 4H), 0.97–0.86$ $(m, 4H)$. ¹³C NMR (100 MHz, CDCl₃) δ 166.1, 150.5, 148.9, 148.2, 133.9, 133.5, 128.8, 126.6, 126.0, 124.0, 121.5, 121.2, 109.1, 61.3, 61.0, 60.8, 56.2, 56.0, 53.2, 51.6, 32.3, 31.0, 29.3, 25.1, 24.2, 19.4. HRMS (Q-TOF) m/z: $[M+H]^+$ calcd for $C_{26}H_{34}N_4O_3$, 451.2709; found 451.2704.

N'‑(4‑hydroxy‑3‑methoxy‑5‑((3‑methylpiperi‑ din‑1‑yl)methyl)benzylidene)‑2‑morpholinoaceto hydrazide (*11c***)**

Beige solid, yield 81%, mp: 130–132 °C, IR (ATR, cm⁻¹) *ν*max 3208, 3074, 2928, 1663, 1592, 1246, 1078; ¹ H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 9.97 (s, 1H), 8.03 (s, 1H), 7.18 (s, 1H), 6.91 (s, 1H), 3.84 (s, 3H), 3.74 (t, *J*=4.4 Hz, 4H), 3.63 (d, *J*=7.2 Hz, 2H), 3.13 (s, 2H), 2.94–2.54 (m, 7H), 2.00 (t, *J*=10.0 Hz, 1H), 1.70–1.65 (m, 4H), 0.91–0.82 (m, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 165.6, 150.6, 149.1, 148.2, 123.9, 121.5, 121.3, 66.8, 61.5, 61.4, 61.3, 60.8, 56.0, 53.8, 53.2, 32.3, 31.0, 25.1, 19.3. HRMS (Q-TOF) m/z: [M+ H]⁺ calcd for $C_{21}H_{32}N_4O_4$, 405.2502; found 405.2492.

N′**‑(3‑((3,4‑dihydroisoquinolin‑2(1H)‑yl) methyl)‑4‑hydroxy‑5‑methoxybenzylidene)‑2‑(4 ‑phenyl piperazin‑1‑yl)acetohydrazide (12a)**

White solid, yield 94%, mp: 232–234 °C, IR (ATR, cm⁻¹) ν_{max} 3209, 3063, 2944, 1668, 1592, 1228, 1079, 750; ¹ H NMR (400 MHz, CDCl3) δ 10.05 (s, 1H), 8.12 (s, 1H), 7.33–7.28 (m, 2H), 7.18–7.12 (m, 4H), 7.09 (s, 1H), 7.02–6.90 (m, 5H), 3.93–3.92 (m, 5H), 3.80 (s, 2H), 3.29–3.27 (m, 5H), 2.99 (t, *J*=5.5 Hz, 2H), 2.91 (t, *J*=5.5 Hz, 2H), 2.80 (t, *J*=4.4 Hz, 4H). 13C NMR (100 MHz, CDCl3) δ 165.8, 150.9, 150.2, 149.0, 148.4, 133.3, 132.8, 129.3, 128.7, 126.7, 126.6, 126.0, 121.8, 121.1, 120.2, 116.2, 109.5, 61.1, 60.4, 56.1, 55.1, 53.6, 49.9, 49.3, 28.4. HRMS (Q-TOF) m/z: $[M+H]^{+}$ calcd for $C_{30}H_{35}N_5O_3$, 514.2818; found 514.2811.

2‑(3,4‑Dihydroisoquinolin‑2(1H)‑yl)‑*N*′**‑(3‑((3,4‑dihy droisoquinolin‑2(1H)‑yl)methyl)‑4‑hydroxy‑5‑meth‑ oxybenzylidene)acetohydrazide (12b)**

White solid, yield 92%, mp: 215–217 °C, IR (ATR, cm^{-1}) *ν*max 3188, 3063, 2905, 1658, 1589, 1226, 1077, 737; 1 H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 10.13 (s, 1H), 8.02 (s, 1H), 7.28 (s, 1H), 7.20–7.13 (m, 6H), 7.07–7.00 (m, 3H), 3.91 (s, 5H), 3.80 (s, 2H), 3.78 (s, 2H), 3.38 (s, 2H), 3.02–2.96 (m, 4H), 2.92–2.87 (m, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 166.0, 150.2, 148.8, 148.4, 133.9, 133.5, 133.3, 128.8, 128.7, 126.7, 126.6, 126.6, 126.0, 124.5, 121.7, 121.1, 109.5, 61.0, 60.5, 56.2, 56.1, 55.2, 51.6, 49.9, 29.3, 28.5. HRMS (Q-TOF) m/z: $[M+H]^{+}$ calcd for $C_{29}H_{32}N_4O_3$, 485.2553; found 485.2548.

N′**‑(3‑((3,4‑dihydroisoquinolin‑2(1H)‑yl) methyl)‑4‑hydroxy‑5‑methoxybenzylidene)‑2‑mor‑ pholin oacetohydrazide (12c)**

White solid, yield 90%, mp: 220–222 °C, IR (ATR, cm^{-1}) *ν*_{max} 3199, 3069, 2936, 1660, 1565, 1229, 1081, 745; ¹H NMR (400 MHz, CDCl₃) δ 9.97 (s, 1H), 8.13 (s, 1H), 7.30 (s, 1H), 7.20–7.12 (m, 3H), 7.08 (s, 1H), 7.02–7.00 (m, 1H), 3.93 (s, 5H), 3.80–3.78 (m, 6H), 3.21 (s, 2H), 2.98–2.91 (m, 4H), 2.63 (bs, 4H). ¹³C NMR (100 MHz, CDCl3) δ 165.6, 150.2, 149.0, 148.4, 133.2, 132.7, 128.7, 126.7, 126.6, 126.1, 124.5, 121.8, 121.0, 114.8, 109.5, 66.9, 61.5, 60.3, 56.1, 55.1, 53.9, 49.9, 28.4. HRMS (Q-TOF) m/z: $[M + H]^{+}$ calcd for $C_{24}H_{30}N_{4}O_{4}$, 439.2345; found 439.2340.

Ethyl 4‑(2‑hydroxy‑3‑methoxy‑5‑((2‑(2‑(4‑phenyl‑ piperazin‑1‑yl)acetyl)hydrazono)methyl)benzyl) piperazine‑1‑carboxylate (13a)

White solid, yield 91%, mp: 181–183 °C, IR (ATR, cm^{-1}) *ν*max 3186, 3056, 2922, 1698, 1657, 1594, 1239, 1089, 760; ¹H NMR (400 MHz, CDCl₃) δ 10.09 (s, 1H), 8.06 (s, 1H), 7.27–7.23 (m, 3H), 6.97 (s, 1H), 6.91–6.84 (m, 3H), 4.12 (q, *J*=7.1 Hz, 2H), 3.89 (s, 3H), 3.51 (brs, 4H), 3.22 (m, 8H), 2.73 (t, *J*=4.4, 4H), 2.50 (brs, 4H), 1.23 (t, *J*=7.1, 3H). 13C NMR (100 MHz, CDCl3) δ 165.8, 155.2, 150.9, 149.6, 148.7, 148.3, 129.2, 124.7, 121.6, 120.7, 120.1, 116.1, 109.4, 61.6, 61.0, 60.8, 56.1, 53.6, 52.1, 49.2, 43.4, 14.6. HRMS (Q-TOF) m/z: $[M+H]^+$ calcd for $C_{28}H_{38}N_6O_5$, 539.2982; found 539.2975.

Ethyl 4‑(5‑((2‑(2‑(3,4‑dihydroisoquinolin‑2(1H)‑yl) acetyl)hydrazono)methyl)‑2‑hydroxy‑3‑methoxy benzyl)piperazine‑1‑carboxylate (13b)

White solid, yield 93%, mp: 188–191 °C, IR (ATR, cm^{-1}) *ν*_{max} 3183, 3034, 2936, 1703, 1655, 1596, 1267, 1082, 750; ¹H NMR (400 MHz, CDCl₃) δ 10.15 (s, 1H), 7.99 (s, 1H), 7.25 (s, 1H), 7.20–7.14 (m, 3H), 7.04 (d, *J*=5.6 Hz, 1H), 6.97 (s, 1H), 4.14 (q, *J*=7.1 Hz, 2H), 3.90 (s, 3H), 3.78 (s, 2H), 3.72 (s, 2H), 3.54 (brs, 4H), 3.36 (s, 2H), 2.98 (t, *J*=5.5 Hz,, 2H), 2.91 (t, $J=5.5$, 2H), 2.54 (brs, 4H), 1.26 (t, $J=7.1$, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 166.0, 155.2, 149.5, 148.6, 148.3,

133.7, 133.4, 128.8, 126.7, 126.6, 126.0, 124.7, 121.6, 120.7, 109.5, 61.6, 61.0, 60.8, 56.2, 56.1, 52.2, 51.6, 43.4, 29.2, 14.6. HRMS (Q-TOF) m/z: $[M+H]^{+}$ calcd for $C_{27}H_{35}N_{5}O_{5}$, 510.2716; found 510.2712.

Ethyl 4‑(2‑hydroxy‑3‑methoxy‑5‑((2‑(2‑mor‑ pholinoacetyl)hydrazono)methyl)benzyl)pipera‑ zine‑1‑carboxylate (13c)

White solid, yield 88%, mp: 163–165 °C, IR (ATR, cm^{-1}) *ν*max 3187, 3055, 2924, 1699, 1656, 1590, 1236, 1087, 764; ¹H NMR (400 MHz, CDCl₃) δ 9.99 (s, 1H), 8.06 (s, 1H), 7.23 (s, 1H), 6.96 (s, 1H), 4.11 (q, *J*=7.1 Hz, 2H), 3.89 (s, 3H), 3.73–3.71 (m, 5H), 3.51 (brs, 4H), 3.16 (s, 2H), 2.57 (t, *J*=4.4 Hz, 4H), 2.52 (brs, 4H), 1.23 (t, *J*=7.1, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 165.6, 155.2, 149.6, 148.8, 148.3, 124.7, 121.7, 120.7, 109.4, 66.9, 61.6, 61.4, 60.8, 56.1, 53.8, 52.15, 43.5, 14.6. HRMS (Q-TOF) m/z: $[M+H]$ ⁺ calcd for $C_{22}H_{33}N_5O_6$, 464.2509; found 464.2503.

Biological studies

According to prior research, the AR purifcation procedure was carried out utilizing the $(NH_4)_2SO_4$ precipitation DE-52 cellulose ion-exchange column, Sephadex G-100 gel fltration column, and 2'5'-ADP Sepharose-4B affinity column [[60–](#page-18-7)[65\]](#page-18-8). The Bradford technique at 595 nm was used to determine the protein content of the samples [\[66](#page-18-9)[–68](#page-18-10)]. SDS-PAGE technique was employed to ensure enzyme purity [\[69–](#page-18-11)[71\]](#page-18-12). AR activity was assessed spectrophotometrically using DL-glyceraldehyde and NADPH reduction at 340 nm [$72-74$]. Activity (%) novel acyl hydrazones generated from vanillin compounds and standard inhibitor epalrestat plots were used to calculate the IC_{50} values, inhibitory concentrations that reduce enzyme activity by 50%. Three diferent inhibitory doses were applied to determine K_I values and inhibition types [[75](#page-18-15), [76](#page-18-16)].

In silico studies

The Maestro ver. 13.1 [[77\]](#page-18-17), Protein Preparation Wizard [\[78](#page-18-18)], SiteMap [\[79](#page-18-19)], Receptor Grid Generation [\[80](#page-18-20)], LigPrep [\[81](#page-18-21)], QikProp [\[82](#page-19-0)], Prime MM-GBSA [[83\]](#page-19-1), and Jaguar [\[84](#page-19-2)] tools are implemented in Small-Molecule Drug Discovery Suite 2022-1 for Mac (Schrödinger, LLC, NY, USA) and were

used to perform molecular docking, ADME, and DFT calculations. All compounds, including novel acyl hydrazones and the reference ligand EPR, were sketched in the 2D-sdf format using ChemDraw ver. 19.1 for Mac [[85\]](#page-19-3) (PerkinElmer, Inc., Waltham, MA, USA), and ligand production was performed using the LigPrep tool [\[86–](#page-19-4)[88\]](#page-19-5). The QikProp module was used to estimate ADME-related parameters for these substances as described in previous studies [\[89–](#page-19-6)[91](#page-19-7)]. The Protein Data Bank provided the X-ray structure of the template 4JIR [[92\]](#page-19-8) (Resolution: 2.00 Å; *R*-Values free and work: 0.210 and 0.160, respectively; Species: Homo sapiens) and wass prepared using the Protein Preparation Wizard [[93–](#page-19-9)[95](#page-19-10)]. The Receptor Grid Generation tool [\[96](#page-19-11)[–98](#page-19-12)] was used to create the docking grid box. The extra-precision (XP) approach [[99–](#page-19-13)[101](#page-19-14)] was used to perform molecular-docking simulations. Also, the VSGB energy model [\[102](#page-19-15)[–104](#page-19-16)] and OPLS4 force feld [\[105,](#page-19-17) [106](#page-19-18)] were used to calculate MM-GBSA binding energies [\[107](#page-19-19), [108](#page-19-20)], which predict relative binding afnities for these novel acyl hydrazones. The novel acyl hydrazones were also analyzed via Becke's three-parameter exchange potential and Lee–Yang–Parr correlation functional (B3LYP) using a 6-31G∗∗ basic set level. With singlepoint calculations, the implicit solvation model of Poisson Boltzmann Finite was used. The electrostatic potentials were computed using the molecule's van der Waals contact surface area [[109,](#page-20-0) [110](#page-20-1)].

Statistical studies

Analysis of the data and drawing of graphs were realized using GraphPad Prism ver. 8 for Mac (GraphPad Software, La Jolla California USA). The inhibition constants were calculated by SigmaPlot ver. 12 for Windows (Systat Software, San Jose California USA). The fit of enzyme inhibition models was compared using the extra sum-of-squares *F* test and the AICc approach. The results were exhibited as mean \pm standard error of the mean (95%) confidence intervals). Differences between datasets were considered statistically significant when the *p* value was less than 0.05.

Results and discussion

Chemistry

The synthesis pathway of designed molecules was carried out using reagents and conditions as presented in Schemes [1](#page-10-0) and [2](#page-10-1). Briefy, compounds **2a** and **2b** were synthesized according to the Schotten Baumann reaction, and compounds **3a-e** were synthesized according to the Mannich Reaction, with good yields.

In Scheme [2,](#page-10-1) hydrazides (**6a–c**) were synthesized from esters formed by the reaction of the related cyclic secondary amine with ethyl chloroacetate. In the fnal step, the synthesized aldehydes (**2a–b**, **3a–e**) were treated with the synthesized hydrazides (**6a–c**), producing target compounds (**7a–c** to **13a–c**) with yields ranging from 75 to 90%. The melting points of the known intermediates were compared with the values in the literature. The structures of the newly synthesized compounds were characterized by IR, 1H NMR, and 13C NMR spectroscopic methods.

In the IR spectra of the compounds **7a–c** to **13a–c**, NH stretching bands are observed at $3210-3145$ cm⁻¹. Aromatic C–H stretching bands are seen at $3108-3034$ cm⁻¹ and the aliphatic C–H stretching bands are observed at 2960–2905 cm⁻¹. C=O of hydrazone moiety and CH=N stretching bands are observed at $1668-1654$ cm⁻¹ and 1597–1556 cm−1, respectively. C–O and N–H bending bands are seen at 1270–1225 cm−1 and 1089–1067 cm−1, respectively. For compounds **7a–c, 8a–c,** and **13a–c**, stretching bands of ester carbonyl are observed at 1743–1698 cm−1.

In the ¹ H NMR spectra of the compounds **7a–c** to **13a–c**, peaks of NH protons are seen as singlet at δ 10.34–9.97 ppm. Peaks of N=CH protons are observed as singlet at δ 8.25–7.97 ppm. The resonance signals of aromatic protons are observed at δ 7.99–6.60 ppm as singlet, doublet, triplet, and multiplet relative to their chemical environment. Peaks of OCH₃ and Ph–CH₂–N protons are seen as singlet at δ 3.94–3.84 and δ 3.81–3.71 ppm, respectively. Peaks of N–CH₂–C=O protons are seen as singlet at δ 3.39–3.15 ppm. Aliphatic protons of morpholine, piperazine, tetrahydroisoquinoline, and piperidine moieties are observed at δ 3.75–0.82 ppm as singlet, doublet, triplet, and multiplet relative to their chemical environment. For compounds **13a–c**, peaks of OCH_2CH_3 protons are seen as quartet at δ 4.14–4.11 ppm and peaks of OCH_2CH_3 protons are seen as triplet at δ 1.26–1.23 ppm. Chemical shifts, integrations, and splits are fully compatible with the structures.

In the 13 C NMR spectra of target molecules, peaks of HN–C=O and CH=N carbons are seen at δ 166.5–165.6 ppm and δ 147.8–148.4 ppm, respectively. For compounds **7a–c** and **8a–c**, peaks of Ar–C=O carbons are observed at δ 160.0–156.2 ppm. Peaks of aromatic carbons are observed at δ 151.9–109.1 ppm. Peaks of OCH₃ carbons are seen at δ 56.3–56.0 ppm, and peaks of N– CH_2 –C=O carbons are seen at δ 61.5–61.0 ppm. For compounds $13a-c$, peaks of $CH_3CH_2-O-C=O$ carbons are observed at δ 155.2 ppm. Also, for compounds **13a–c**, peaks of $CH_3CH_2-O-C=O$ and $CH_3CH_2-O-C=O$ carbons are observed at δ 43.5–43.4 ppm and δ 14.6 ppm, respectively. Finally, aliphatic carbons of morpholine, piperazine, tetrahydroisoquinoline, and piperidine moieties are observed at δ 66.9–19.3 ppm. Chemical shifts

and the number of the peaks are fully compatible with the structures.

Biological studies and structure–activity relation‑ ship

As shown in Table [1](#page-11-0), 21 compounds exhibited relative more potent inhibitory activity against AR with K_I values ranged from 49.22 ± 3.64 to 1114.00 ± 49.64 nM, and among them, compound **11c**, named *N'*-(4-hydroxy-3-methoxy-5-((3-methylpiperidin-1-yl)methyl) benzylidene)-2-morpholinoaceto hydrazide, displayed the strongest inhibitory effect with an K_I value of 49.22 ± 3.64 nM (Fig. [1\)](#page-11-1). The inhibitor effects of novel acyl hydrazones derived from vanillin compounds against AR were decreased in the following order: $11c > (K_I:$ $49.22 \pm 3.64 \text{ nM}$) $10\text{b} > (K_I: 79.36 \pm 5.77 \text{ nM})$ $7\text{a} > (K_I: 10^{-10}\text{ m})$ $101.00 \pm 8.21 \text{ nM}$) $10a > (K_I: 145.80 \pm 22.30 \text{ nM})$ $8a > (K_I:$ 182.40 ± 14.35 nM) $7c > (K_I: 304.00 \pm 13.36 \text{ nM})$ **9b** > $(K_I:$ $312.80 \pm 53.48 \text{ nM}$ $12b > (K_1: 338.40 \pm 17.18 \text{ nM})$ **9c** > (K_1 : 372.10 ± 62.45 nM) **11b** > (K_1 : $394.00 \pm 22.65 \text{ nM}$ $11a > (K_I: 398.20 \pm 15.20 \text{ nM})$ $12a > (K_1: 437.40 \pm 14.08 \text{ nM})$ $13b > (K_1:$ $444.50 \pm 24.74 \text{ nM}$ $7\text{b} > (K_1: 523.00 \pm 22.54 \text{ nM})$ $12c$ > (K_i: 533.50 ± 26.67 nM) $10c$ > (K_I: 598.70 \pm 19.27 nM) **8c** > (K_I : 746.20 \pm 34.41 nM) **13a** > (K_I : 787.10 ± 39.32 nM) **13c** > (K_I : 854.00 \pm 33.96 nM) **8b** > (K_I : 897.20 \pm 43.63 nM) **9a** > $(K_I: 1114.00 \pm 49.64 \text{ nM}).$

There are diferent types of inhibition, including mixed, non-competitive, competitive, and un-competitive. It would be appropriate to state that the inhibitory potential of the molecules is due to the structural, 3D chemical structure, and conformation features that vary according to the diferent groups on which the backbone structure depends. When compounds **7a** and **7c** are compared, substitution of 4-phenylpiperazin-1-yl structure with 2-morpholino caused a threefold change in the inhibition value. The 4-phenylpiperazin-1-yl group showed a better inhibition efect in the replacement of the phenylfuran-2-carboxylate structure in the structure of compounds **7a** and **7c** with thiophene-2-carboxylate (**8a** and **8c**).

When acetohydrazide compounds were compared (**9c, 10c,** and **11c**), the inhibition efect was observed as follows, respectively: 3-methylpiperidin-1-yl (**11c,** *K*_I: 49.22 \pm 3.64 nM) > 5-morpholinomethyl (9c, *K*_I: $372.10 \pm 62.45 \text{ nM}$ > 4-phenylpiperazin-1-yl (10c, K_1 : 598.70 \pm 19.27 nM). On the contrary, when we look at the inhibition order of 2-(4-phenylpiperazin-1-yl) acetohydrazide compounds, the inhibition effect was observed as follows, respectively: 4-phenylpiperazin-1-yl (**10a)** > 3-methylpiperidin-1-yl (**11a**) > 5-morpholinomethyl (**9a**). Considering the inhibition order

Scheme 1 The synthetic pathway for preparations of aldehydes **2a–b, 3a–e** containing acyl group and Mannich base derived from vanillin

Scheme 2 The synthetic pathway for the preparation of novel acyl hydrazones (**7a–c** and **13a–c**)

of piperazine-1-carboxylate compounds (**13a, 13b** and 13c), compound 13b showed better inhibition effect $(K_I:$ 444.50 ± 24.74 nM).

Potential inhibitory effect of synthesized compounds against AR has been reported in the literature. Yapar et al. [[111\]](#page-20-2) synthesized the novel bis-hydrazone compounds bearing isovanillin moiety and studied inhibition effect of these compounds on AR enzyme activity. They found that the novel bis-hydrazones demonstrated in nanomolar levels as AR inhibitors with K_I values in the range of 13.38–88.21 nM. Maccari et al. [\[112](#page-20-3)] performed inhibition efect of 5-arylidene-2,4-thiazolidinediones on AR enzyme. The authors found that a hydroxyl group on the 5-arylidene moiety led to signifcant inhibitory efect. Alexiou et al. [[113](#page-20-4)] synthesized a series of *N*-(3,5-difuoro-4-hydrozyphenyl)benzenesulfonamide derivatives and studied the

inhibition effect of novel compounds on AR. They enhanced these compounds compared to *N*-benzenesulfonylglycine lead derivative. The most potent inhibitor was found to be compound **66** with the IC_{50} value of 14.1 μ M.

In silico studies

Table [2](#page-12-0) summarizes the results of the determination of ADME-related parameters for novel acyl hydrazones. New acyl hydrazones were identifed as hit-agents with drug-like efects based on ADME properties calculations. According to this information, the molecular weights (MWs, 392.45–542.68) and dipole moments (Dipole, in the 2.65 to 9.64) of the novel acyl hydrazones derived from vanillin compounds (**7a–c** and **13a–c**) have reported being in the permissible values. Volume (in range 1227.48 to 1752.91), which is the total solvent-accessible volume descriptor, was determined to be in the permissible ranges for these hydrazones (**7a–c** and **13a–c**), compared with reference values. The logP values, such as QPlogPoct, QPlogPw, QPlogPo/w, QPlogS, QPlogBB, QPlogKp, and QPlogKhs, are in ranging from 20.27 to 29.35, 12.63 to 16.76, 0.69 to 4.61, −6.46 to -1.26 , -1.14 to -0.24 , -7.17 to -3.28 , and -0.55 to 0.93, respectively, and indicates of target derivatives (**7a–c** and **13a–c**) have the high capacity. The values of human oral absorption (HOAs) were higher than 30%, and van der Waals surface area of polar nitrogen and oxygen atoms (PSA, in the range 86.82 to 134.05) indicate that all analogs (**7a–c** and **13a–c**) had at the acceptable values. All the acyl hydrazones have displayed normal Caco-2 cell permeability rates (except for compounds **13b** and **13c**; QPPCaco,

Fig. 1 The Lineweaver–Burk plots of novel acyl hydrazone derivative **11c**

in the 23.47 to 245.90), and MDCK cell permeability values (except for compound **13a**, **13b**, and **13c**; QPPMDCK, in range 10.49 to 214.30). Indeed, all newly synthesized acyl hydrazones derived from vanillin compounds (**7a–c** and **13a–c**) displayed good drug-like properties with zero violation of Lipinski's rule (except for compounds **10a–b**, **12a**, and **13a–c**) and zero or one violation of the Jorgensen's rule (except for compounds **12a**) (Table [2](#page-12-0)). Moreover, the ADME-Tox values calculated for *N*′-(4-hydroxy-3 methoxy-5-((3-methylpiperidin-1-yl)methyl)benzylidene)- 2-morpholinoaceto hydrazide **11c** might explain why, being a potent AR inhibitor, this ligand has the most AR inhibitory activity in biological experiments.

Molecule	K_{I} (nM) ^a	R^2	Inhibition type	Molecule	K_{I} (nM) ^a	R^2	Inhibition type
7a	101.00 ± 8.21	0.9924	Competitive	10c	598.70 ± 19.27	0.9937	Noncompetitive
7b	523.00 ± 22.54	0.9911	Noncompetitive	11a	398.20 ± 15.20	0.9917	Noncompetitive
7c	$304.00 + 13.36$	0.9975	Competitive	11 _b	394.00 ± 22.65	0.9835	Noncompetitive
8a	182.40 ± 14.35	0.9924	Competitive	11c	49.22 ± 3.64	0.9938	Competitive
8b	897.20 ± 43.63	0.9890	Noncompetitive	12a	437.40 ± 14.08	0.9963	Noncompetitive
8c	$746.20 + 34.41$	0.9872	Noncompetitive	12 _b	338.40 ± 17.18	0.9883	Noncompetitive
9a	$1114.00 + 49.64$	0.9887	Noncompetitive	12c	533.50 ± 26.67	0.9877	Noncompetitive
9b	312.80 ± 53.48	0.9915	Mixed	13a	787.10 ± 39.32	0.9900	Noncompetitive
9c	372.10 ± 62.45	0.9921	Mixed	13 _b	444.50 ± 24.74	0.9849	Uncompetitive
10a	145.80 ± 22.30	0.9935	Mixed	13c	854.00 ± 33.96	0.9922	Noncompetitive
10 _b	79.36 ± 5.77	0.9940	Competitive	Epalrestat ^b	855.50 ± 61.46	0.9853	Noncompetitive

Table 1 Inhibition data of AR with the novel acyl hydrazones derived from vanillin compounds and standard inhibitor epalrestat

^aThe test results were expressed as means of triplicate assays \pm SEM

b Epalrestat was used as a control for the AR enzyme

 $\lfloor \frac{1}{2} \rfloor$ 725.00), computed dipole moment of the compound (Dipole; 1.00-12.50), total solvent-accessible volume in cubic angstroms using a probe with a 1.4 Å Radius (Volume; 500.00-2000.00), octanol/gas partition coefficient (QPlogPoct; 8.00–35.00), water/gas partition coefficient (QPlogPw; 4.00–45.00), octanol/water partition coefficient (QPlogPo/w; - 2.00 to 6.50), aqueous solubility (QPlogS; - 6.50 to 0.50) Narious computational pharmacodynamic and pharmacokinetic parameters of synthesized compounds in this research were predicted such as molecular weight of the compound (MW; 130.00–
725.00), computed dipole moment of the com bility (QPlogS; −6.50 to 0.50), apparent Caco-2 cell permeability in nm/sec (QPPCaco;<25 poor,>500 great), brain/blood partition coefcient (QPlogBB; −3.00 to 1.20), apparent MDCK cell permeability in mm/s (QPPMDCK; <25 poor,>500 great), skin permeability (QPlogKp; -8.00 to -1.00), prediction of binding to human serum albumin (QPlogKhsa; -1.50 to 1.50),
human oral absorption (HOA; <25 poor, >500 gre cell permeability in nm/s (QPPMDCK;<25 poor,>500 great), skin permeability (QPlogKp; −8.00 to −1.00), prediction of binding to human serum albumin (QPlogKhsa; −1.50 to 1.50), human oral absorption (HOA;<25 poor,>500 great), van der Waals surface area of polar nitrogen and oxygen atoms (PSA; 7.00 to 200.00), number of violations of Lipinski's rule of fve (max. octanol/gas partition coefficient (QPlogPoct; 8.00–35.00), water/gas partition coefficient (QPlogPw; 4.000–45.00), octanol/water partition coefficient (QPlogPo/w; −2.00 to 6.50), aqueous solu-4), and number of violations of Jorgensen's rule of three (max. 3) 4), and number of violations of Jorgensen's rule of three (max. 3)

^bEpalrestat was used as a control for the AR enzyme Epalrestat was used as a control for the AR enzyme

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Fig. 2 Molecular docking of aldode reductase (AR; PDB code: 4JIR) with native ligand EPR ((5-((2*E*)-2-methyl-3-phenylprop-2 en-1-ylidene)-4-oxo-2-thioxo-1,3-thiazolidin-3-yl)acetic acid). **A** 3D ligand interaction diagram of 4JIR with native ligand EPR. **B** 2D docking pose of native ligand EPR with the key amino acids within the binding pocket of 4JIR

Fig. 3 Molecular docking of aldode reductase (AR; PDB code: 4JIR) with compound **11c** (*N'*-(4-hydroxy-3-methoxy-5-((3-methylpiperidin-1-yl)methyl)benzylidene)-2-morpholinoaceto hydrazide). **A** 3D ligand interaction diagram of 4JIR with compound **11c**. **B** 2D docking pose of compound **11c** with the key amino acids within the binding pocket of 4JIR

Fig. 4 The HOMO–LUMO plot of the most potent AR inhibitory **11c**. The red color-coding area specifes the most negative potential region, while the blue color-coding area defnes the most positive potential region of the compound

Molecular docking experiments were used to obtain substantial insight into the origins of the structure–activity connections examined for the new acyl hydrazones. Initially, the native ligand EPR ((5-[(2E)-2-methyl-3-phenylprop-2-en-1-ylidene]-4-oxo-2-thioxo-1,3-thiazolidin-3-yl)acetic acid) in the AR receptor's binding site (PDB code 4JIR) [\[92](#page-19-8)] was employed in the redocking computation. At a rootmean-square deviation (RMSD) of 0.10 (docking score of−7.04 kcal/mol and MM-GBSA value of−41.07 kcal/ mol), the docked pose of EPR overlapped with the pose in the X-ray crystal structure of the AR (Fig. [2\)](#page-13-0). This redocking experiment was crucial in determining which model structure would best accommodate all of the newly synthesized AR ligands. Then, using the Glide Ligand Docking tool in this series, the generated binding model was used to perform docking calculations of the most potent AR inhibitor **11c**. A docking score of−8.07 kcal/mol and MM-GBSA value of−69.69 kcal/mol indicated compound **11c** are a tight binder for AR compared to EPR. The carboxy moiety formed an H-bond with residue Trp111 (distance 2.16 Å), while the –NH group displayed π -cation interaction with Phe122. Furthermore, compound **11c** monitored hydrophobic interactions with residues Trp20, Val47, Tyr48, Trp79, Phe121, Tyr209, Trp219, Ile260, Cys298, Leu300, Leu301, and Cys303 played signifcant roles in the binding of the ligand with 4JIR (Fig. [3](#page-14-0)).

To explain the structural parameters, the DFT calculation was performed for compound **11c**, which has the most potent AR inhibitory activity and was optimized at the level of B3LYP/6-31G∗∗. In chemical reactivity, derivative **11c** is sparkling, and the HOMO (highest occupied molecular orbitals)-LUMO (lowest unoccupied molecular orbitals) gap increases the charge transfer of the compounds. The electron density is indicated by the intensity of the color that refects the distinctive feature of the molecule. Because electrons can move quickly between energy levels in the HOMO and LUMO, energy gap levels reveal the delicate nature of reactivity. The energy gap of the compound **11c** in the HOMO–LUMO analysis is 0,160,511 eV, and the HOMO–LUMO plot of 11c is shown in Fig. [4.](#page-15-0) From this plot, it is seen that the value of ΔE decreases in the case of complex, which further supports the binding framework and that compound **11c** has signifcant chemical reactivity and polarizability.

Conclusion

A series of acyl hydrazones derived from vanillin were synthesized and their effects on the AR were investigated. K_I values in the range of 49.22 ± 3.64 to 897.20 ± 43.63 nM. Compounds **11c** and **10b** against AR enzyme activity were identifed as the highly potent inhibitors than epalrestat. AR is novel molecular target involved in diferent pathways related to the development of type II diabetes mellitus and related comorbidities. The design of efective bioavailable inhibitors for AR enzyme is still an urgent need. We expect that our fndings will lead to the development of novel AR inhibitors based on inhibition and molecular docking investigations. We also hope that our compounds will be good therapeutic candidates with further investigation.

Supplementary Information The online version contains supplementary material available at<https://doi.org/10.1007/s11030-022-10526-1>.

Acknowledgements This work was supported by the Research Fund of Ardahan University (Grant Number 2019-008), the Research Fund of Erzincan Binali Yıldırım University (Grant Number FBA-2017- 501), and the Research Fund of Anadolu University (Grant Number 2102S003).

Declarations

Competing interests The authors declare no confict of interest.

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