



# Convenient diastereoselective synthesis of annulated 3-substituted-(5*S*\*,6*S*\*,*Z*)-2-(2-(2,4-dinitrophenyl)hydrazono)-5,6-diphenyl-1,3-thiazinan-4-ones

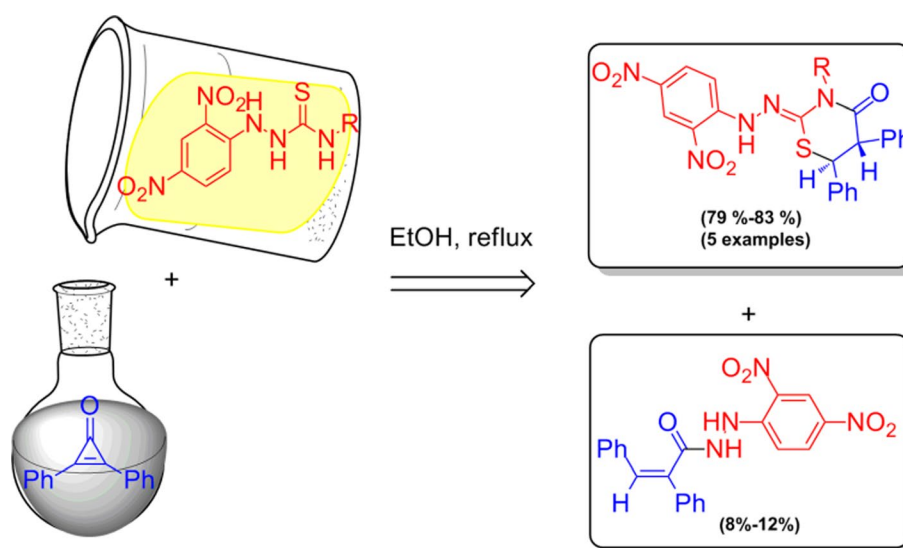
Alaa A. Hassan<sup>1</sup> · Nasr K. Mohamed<sup>1</sup> · Ashraf A. Aly<sup>1</sup> · Hendawy N. Tawfeek<sup>1</sup> · Henning Hopf<sup>2</sup> · Stefan Bräse<sup>3</sup> · Martin Nieger<sup>4</sup>

Received: 7 September 2018 / Accepted: 23 December 2018 / Published online: 3 January 2019  
© Springer Nature Switzerland AG 2019

## Abstract

Racemic 2-(2,4-dinitrophenyl)hydrazono)-5,6-diphenyl-1,3-thiazinan-4-ones and (*Z*)-*N'*-(2,4-dinitrophenyl)-2,3-diphenylacrylohydrazone were formed during the diastereoselective reaction between 4-substituted 1-(2,4-dinitrophenyl) thiosemicarbazides and 2,3-diphenylcycloprop-2-enone under refluxing ethanol. The structures of the synthesized compounds were confirmed by single-crystal X-ray analyses.

## Graphical abstract



**Keywords** 1,3-Thiazinan-4-ones · Substituted 2,3-diphenylacrylohydrazone · (2,4-Dinitrophenyl)-4-substituted thiosemicarbazides · 2,3-Diphenylcyclopropenone · Annulated compounds

**Electronic supplementary material** The online version of this article (<https://doi.org/10.1007/s11030-018-09912-5>) contains supplementary material, which is available to authorized users.

✉ Alaa A. Hassan  
alaahassan2001@mu.edu.eg

Extended author information available on the last page of the article

## Introduction

Thiazinanones, despite being rarely reported, are very interesting compounds due to their important role in medicinal chemistry [1–3]. Substituted thiazinanones exhibited antitumor [4], antifungal activity [5] and

antimalarial activity which evaluated by Kumawat et al. [6], as well as anti-oxidant activity [7]. Reactions containing an amine, carbonyl compounds and a mercapto acid in one-pot three-component condensation or a two-step process afforded thiazinanone derivatives [5]. 3-Alkyl-2-aryl-1,3-thiazinan-4-ones containing a methylsulfonyl pharmacophore were synthesized, and their cyclooxygenase-2-[COX-2] inhibitory activity has been evaluated [8]. 3-Pyridin-2-ylmethyl-1,3-thiazinan-4-ones were synthesized, and their anti-oxidant activities were evaluated [7].

On the other hand, the behavior of 2,3-diphenylcyclopropenone **1** toward compounds containing C=N moieties with the formation of aza-cyclopentanones (pyrrolidinones) have been reported [9–12].

Amidrazones were reacted with **1** in EtOH/Et<sub>3</sub>N with eliminating a molecule of ammonia to give triphenylpyrimidinones [13].

The aza-enamine reactivity is shown by the reaction of alkenylidenehydrazinecarbothioamides **2** with cyclopropenone **1** and the availability of azomethine carbon as well as sulfur atom as nucleophilic sites; thus 3,5-disubstituted 1,3,4-thiadiazolyl-2,3-diphenylpropenones **3** were formed [14] (Scheme 1).

The reaction of **1** with various aldehyde 4-phenylthiosemicarbazones in acetic acid provided pyrrolo[2,1-*b*]oxadiazoles via [2 + 3]cycloaddition and elimination a molecule of H<sub>2</sub>S [15]. Also, thione derivatives such as 2,4-disubstituted thiosemicarbazides (**4**, R=C<sub>6</sub>H<sub>5</sub>; C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub>) reacted with **1** via a nucleophilic attack of terminal-NH<sub>2</sub> of **4** on the carbonyl group of **1** afforded pyridazines **5** (Scheme 1) [16].

The reaction of 2,3-diphenylcyclopropenone **1** with *N*-imidoylthiourea occurs with elimination phenylisothiocyanate and 3-substituted 2,5,6-triphenylpyrimidin-4-ones were formed [17].

El-Sheref subsequently reported that the reaction between pyrazolylthiourea and **1** followed by oxidation with DDQ afforded 5,6-diphenyl-1,3-thiazinones via the formation of pyrazolylimino-3,5,6-triphenyl-1,3-thiazinan-4-ones [18].

To date, no analogous reactions with **1** using hydrazinecarbothioamides **6a–e** have been described. This remarkable versatility in reaction with thiosemicarbazides and bithioureas with **1** warrants further investigation of the reactivity of **6** toward **1** and a comparison of the behavior of **6** with

another different thiosemicarbazides **2** and **4** or bithioureas with **1**.

Optically active sulfur compounds play an important role in the biochemistry of many living organisms and are found in many synthetic drugs and bioactive natural products [19, 20]. Recently, the [3 + 3] cyclization of amides with cyclopropenethiones afforded the formation of 6*H*-1,3-oxazin-6-ones and 6*H*-1,3-thiazin-6-ones [21].

## Results and discussion

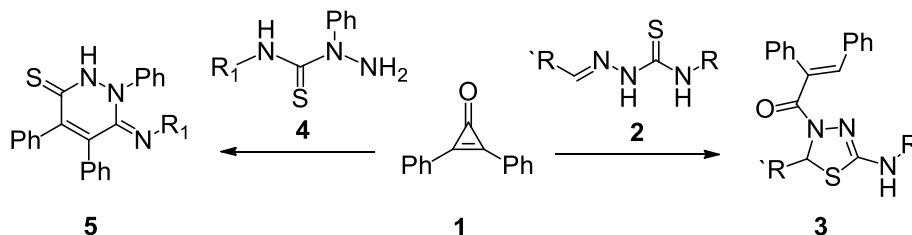
Herein, we report the reaction of 1-(2,4-dinitrophenyl)-4-substituted hydrazinecarbothioamides **6a–e** with 2,3-diphenylcyclopropenone **1** in absolute ethanol under reflux; (5*S*\*,6*S*\*,*Z*)-2-(2-(2,4-dinitrophenyl)-hydrazono)-3-substituted-5,6-diphenyl-1,3-thiazinan-4-ones **7a–e** was precipitated as a major product (79–83%). The filtrate was subjected under chromatographic plates to give only one product namely (*Z*)-*N'*-(2,4-dinitrophenyl)-2,3-diphenylacrylohydrazide **8** as a minor product (8–12%) (Scheme 2).

From the structural investigation, IR spectra of **7a–e** showed the stretching frequency range between 3265 and 3229 cm<sup>-1</sup> due to NH-stretching, 2935–2922 cm<sup>-1</sup> for ali-CH, 1685–1675 cm<sup>-1</sup> for C=O, 1616–1612 cm<sup>-1</sup> for C=N and 1530–1524, 1344–1335 cm<sup>-1</sup> due to nitro groups.

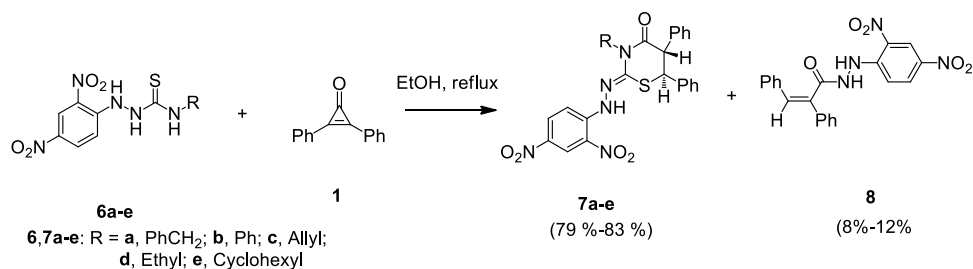
The <sup>1</sup>H NMR spectrum of **7a** (in CDCl<sub>3</sub>) as an example showed a broad singlet at δ = 11.1 ppm due to NH-group, which was confirmed further by D<sub>2</sub>O exchange experiment. A doublet of doublet as AX-system signals at 4.57–4.56 and 5.04–5.03 with coupling constant *J* = 4.0 Hz because of CH-6 and CH-5 of thiazinanones **7a**. The <sup>1</sup>H NMR spectra of **7a–e** showed the absence of any signals due to H-*N*<sup>2</sup> or H-*N*<sup>4</sup> groups of **6a–e** but compound **7a** showed a doublet of doublet signals at 5.40–5.43 and 5.64–5.61 with coupling constant 15.0 Hz for diastereotopic benzyl-CH<sub>2</sub> group.

The <sup>13</sup>C NMR spectrum of **7a** showed signals at δ = 47.15 and 47.90 ppm which were assigned to thiazinanone-CH<sub>6,5</sub>. Another signals at 56.36 ppm are assigned to CH<sub>2</sub>Ph, 168.85 ppm (C=O), 146.67 ppm (C=N) and 144.61 ppm ((NO<sub>2</sub>)<sub>2</sub>-Ar-C-NH). The similarities of <sup>1</sup>H NMR spectra (see experimental part) reveal that the five compounds **7a–e** belong to the same gross structure type namely 3-substituted

**Scheme 1** Previously reported interaction of 2,3-diphenylcyclopropenone **1** with alkenylidene hydrazinecarbothioamides **2** and 2,4-disubstituted thiosemicarbazides **4**



**Scheme 2** Preparation of 2-hydrazothiazinan-4-one derivatives **7a–e** and (*Z*)-*N'*-(2,4-dinitrophenyl)-2,3-diphenylacrylohydrazide **8**



Product	R	Yield (%)	Yield of <b>8</b> (%)
<b>7a</b>	PhCH <sub>2</sub>	81	9
<b>7b</b>	Ph	83	8
<b>7c</b>	Allyl	80	10
<b>7d</b>	Ethyl	79	12
<b>7e</b>	Cyclohexyl	80	8

2-(2-(2,4-dinitrophenyl)hydrazono)-5,6-diphenyl-1,3-thiazinan-4-ones. The elemental analyses and mass spectrometry of **7a–e** clearly showed that the products were formed during the addition of one molecule of **1** to one molecule of **6a–e** without any elimination.

The X-ray crystallographic structure of compound **7a** further supported its relative configurations as (*rac*-5*S*\*,6*S*\*,*Z*)-3-benzyl-2-(2-(2,4-dinitrophenyl)hydrazono)-5,6-diphenyl-1,3-thiazinan-4-one. The molecular of **7a** (Fig. 1 and Tables 1–7, in the crystallographic data) revealed furthermore the formation of 3-benzyl-2-(2-(2,4-dinitrophenyl)hydrazono)-5,6-diphenyl-1,3-thiazinan-4-one in the *cisiod* (*Z*) structure.

The mechanism for the formation of products **7a–e** is presented in scheme 3. The sulfur atom attacks the conjugate

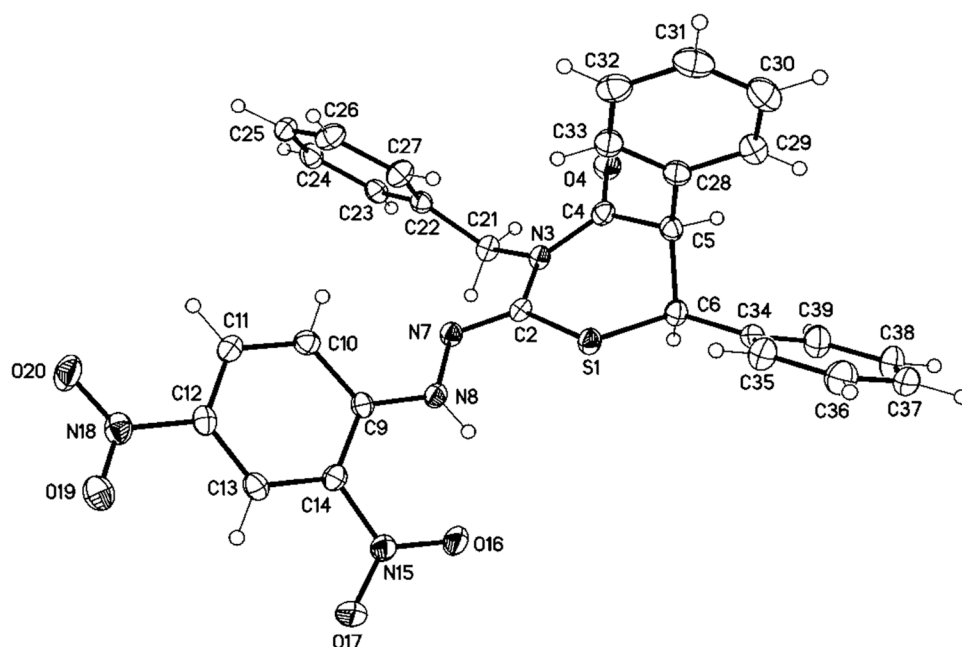
double bond of **1** forming the intermediate **9**. Intramolecular nucleophilic attack of *N*<sup>4</sup>-H on C=O afforded the intermediate **11** which rearranged to give **7a–e** (Scheme 3). On the other hand, *N*<sup>4</sup>-H attacks the carbonyl group of **1** with the formation of **7a–e** via intermediate **12** (Scheme 3).

The ring opening of cyclopropanones has been reported earlier by Gomaa [22] during the reaction of *N*<sup>1</sup>,*N*<sup>2</sup>-diarylformamidines with diphenylcyclopropanone to give 3-aryl-(*N*-4-arylformamidoyl)amino-2,3-diphenylpropionic acids.

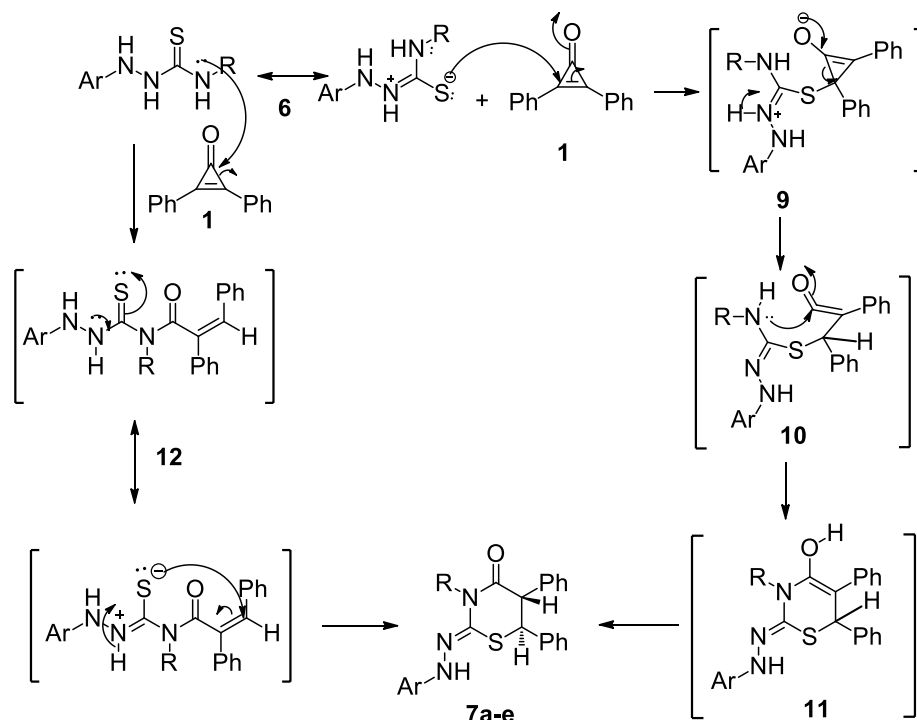
Recently, Wu et al. reported the ring-opening acylation of cyclopropanones with organoboronic acids afforded  $\alpha,\beta$ -diaryl unsaturated ketones [23].

In our study, (*Z*)-*N'*-(2,4-dinitrophenyl)-2,3-diphenylacrylohydrazide **8** was formed as a minor product (8–12%) from the reaction of **1** with **6a–e**.

**Fig. 1** Molecular structure of compound **7a** (displacement parameters are drawn at 50% probability level). The crystallographic numbering does not reflect the IUPAC numbering



**Scheme 3** Mechanism for the formation of 2-hydrazothiazinan-4-one derivatives **7a-e**

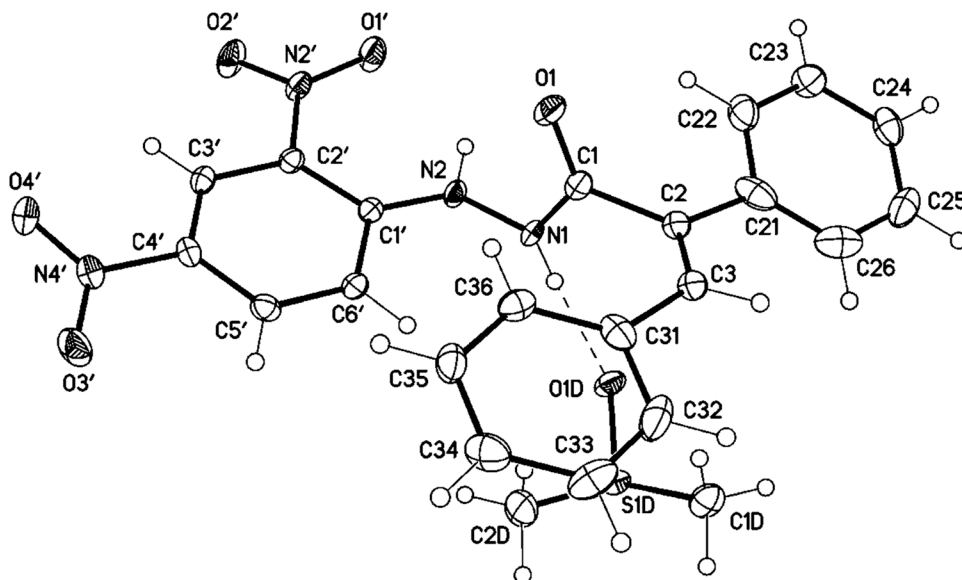


The compound **8** shows IR absorption at 3320–3247  $\text{cm}^{-1}$  due to the NH groups, strong band at 1673  $\text{cm}^{-1}$  corresponding to carbonyl group and bands at 1528, 1344  $\text{cm}^{-1}$  attributed to nitro groups. The  $^1\text{H}$  NMR spectrum of **8** showed multiplet signals at 6.42 due to trisubstituted acryloyl hydrazide-CH, 9.43 (NH), in addition to the aromatic protons. In the  $^{13}\text{C}$  NMR spectrum of **8**, the signal at  $\delta = 165.18$  was assigned to amide-CO, 145.10 due to  $((\text{NO}_2)_2\text{-Ar-C-NH}$ , 133.68 and 139.75 was attributed to acryloyl hydrazide C2 and C3.

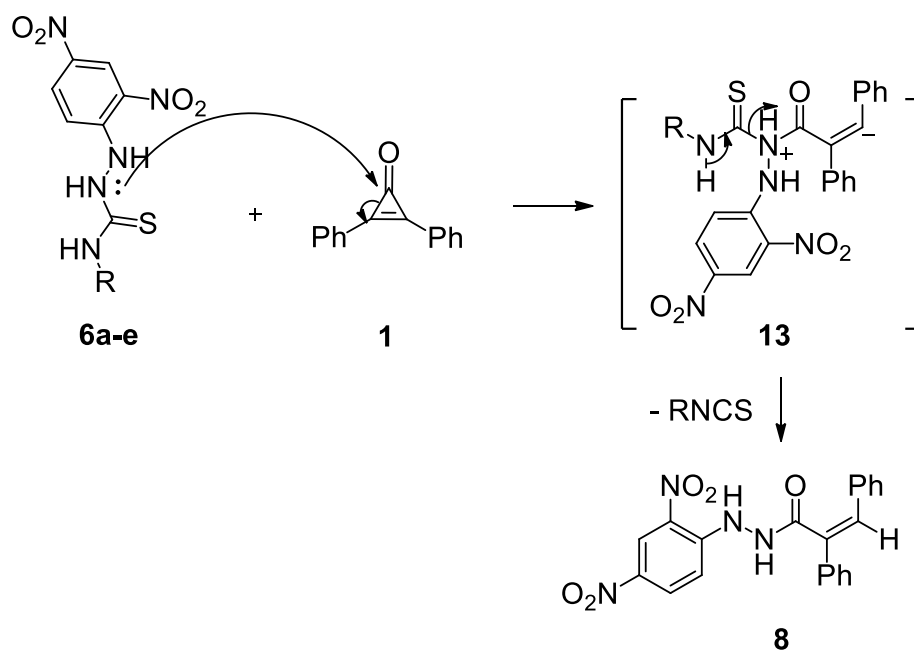
The structure of (*Z*)-*N'*-(2,4-dinitrophenyl)-2,3-diphenylacryloyl hydrazide **8** was determined by X-ray analysis (Fig. 2, Tables 8–15 in supplementary data). The X-ray structure confirms the *trans* (*E*) geometry of the two phenyl groups with respect to the C2–C3 double bond (note that the crystallographic numbering does not correspond to the systematic IUPAC numbering rules).

The hydrazide **8** was formed via the nucleophilic addition of *N*<sup>2</sup>-H on the C=O of **1** with the formation of intermediate **13**. Elimination of RNCS from **13** afforded the formation of **8** (Scheme 4).

**Fig. 2** Molecular structure of **8** (minor disordered part and solvent omitted for clarity, displacement parameters are drawn at 50% probability level)



**Scheme 4** The plausible mechanism for the formation of diazenyl-1,2,4-triazolthione **3a–e**



In order to optimize the reaction conditions, we change the solvent of the reaction to  $\text{CH}_3\text{CN}$  or  $\text{CH}_2\text{Cl}_2$ ,  $\text{CH}_3\text{OH}$ , ethyl acetate and tetrahydrofuran. However, the yields of **7a–e** decreased and in some cases such as ethyl acetate and tetrahydrofuran only traces of **7a–e** were observed detectable by TLC. The excess of one of the reaction partners, namely diphenylcyclopropanone **1** or thiosemicarbazides **6a–e**, led to a significant decrease in the yields.

## Conclusion

Nucleophilic attack of dinitrophenyl-4-substituted thiosemicarbazides on 2,3-diphenylcyclopropanone afforded the formation of racemic 2-(2,4-dinitrophenyl)hydrazono)-5,6-diphenyl-1,3-thiazinan-4-ones as major products and (*Z*)-*N'*-(2,4-dinitrophenyl)-2,3-diphenylacrylohydrazide as minor product.

## Experimental

Melting points were measured with Gallenkamp melting point apparatus. Infrared spectrum (IR) was recorded with Alpha, Bruker FT-IR instruments taken as KBr disks:  $^1\text{H}$  NMR at 400 MHz and  $^{13}\text{C}$  NMR at 100 MHz on a Bruker AM 400 spectrometry with TMS as internal standard ( $\delta=0$ ), and data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, br = broad). For  $^{13}\text{C}$  NMR, TMS ( $\delta=0$ ) was used as internal standard and spectra were obtained with

complete proton decoupling. Mass spectra were obtained using Finnigan MAT instrument (70 eV, EI-mode). Elemental analyses for C, H, N, and S were carried out using an Elmyer 306. Preparative layer chromatography (plc) was carried out on glass plates covered with a 1.0 mm thick layer of slurry-applied silica gel (Merck  $\text{P}_{254}$ ).

## Starting materials

The start materials **6a–e** 2-(2,4-dinitrophenyl)-*N*-substituted hydrazinecarbothioamides were prepared from the reaction between 1-(3,5-dinitrophenyl) hydrazine and the corresponding isothiocyanates in absolute ethanol under refluxing temperature according to reported literature [24, 25]. 2,3-Diphenylcycloprop-2-enone **1** was purchased from Fluka.

## General procedure

An equimolar amounts of 2,3-diphenylcycloprop-2-enone **1** and the appropriate 1,4-disubstituted thiosemicarbazides **6a–e** were mixed in absolute ethanol and refluxed for about 4–6 h, furnished reddish orange precipitates of (5*S*\*,6*S*\*,*Z*)-2-(2-(2,4-dinitrophenyl)hydrazono)-3-substituted-5,6-diphenyl-1,3-thiazinan-4-one derivatives **7a–e**, and the residue were subjected to chromatographic separation using plc and toluene/ethyl acetate (10:3) as eluent to give (*Z*)-*N'*-(2,4-dinitrophenyl)-2,3-diphenylacrylohydrazide **8** as a separated zone.

**(5S\*,6S\*,Z)-3-benzyl-2-(2-(2,4-dinitrophenyl)hydrazono)-5,6-diphenyl-1,3-thiazinan-4-one (7a)**

Reddish orange crystals (acetonitrile), yield 470 mg (81%), mp. 240–242°C; IR (KBr)  $\nu$ : 3265 (NH), 3072 (Ar–CH), 2928 (ali–CH), 1681 (C=O), 1612 (C=N), 1579 (Ar–C=C), 1530 and 1333  $\text{cm}^{-1}$  ( $\text{NO}_2$ );  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$ : 4.56–4.57 (dd, 1H,  $J=4.0$  Hz, thiazinanone-H6), 5.03–5.04 (dd, 1H,  $J=4.0$  Hz, thiazinanone-H5), 5.40–5.43 (d, 1H,  $J=15.0$  Hz,  $\text{CH}_2$ -benzyl), 5.61–5.64 (d, 1H,  $J=15.0$  Hz,  $\text{CH}_2$ -benzyl), 6.81–6.88 (m, 3H, Ar–H), 7.13–7.56 (m, 13H, Ar–H), 8.27–8.30 (m, 1H, Ar–H), 9.10 (m, 1H, Ar–H), 11.02 (br, 1H, hydrazo-NH);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 47.15 (thiazinanone-CH-6), 47.90 (thiazinanone-CH-5), 56.36 ( $\text{CH}_2\text{Ph}$ ), 116.05, 123.51, 127.47, 128.01, 128.16, 128.32, 128.44, 128.67, 128.76, 129.43, 130.08, 130.25 (Ar–CH), 129.21, 132.51, 133.77, 137.05, 137.87 (Ar–C), 144.61 (Ar–C–NH), 146.67 (C=N), 168.85 (C=O); MS ( $m/z$ ): 553 ( $\text{M}^+$ , 47), 462 (13), 347 (10), 207 (51), 182 (58), 149 (56), 91 (100), 77 (15); Anal. Calcd. for  $\text{C}_{29}\text{H}_{23}\text{N}_5\text{O}_5\text{S}$  (553.59): C, 62.92, H, 4.19, N, 12.65, S, 5.79. Found: C, 62.79, H, 4.06, N, 12.47, S, 5.65.

**(5S\*,6S\*,Z)-2-(2-(2,4-dinitrophenyl)hydrazono)-3,5,6-triphenyl-1,3-thiazinan-4-one (7b)**

Reddish orange crystals (acetonitrile), yield 447 mg (83%), mp. 230–232 °C; IR (KBr)  $\nu$ : 3262 (NH), 3076 (Ar–CH), 2932–2922 (ali–CH), 1685 (C=O), 1615 (C=N), 1577 (Ar–C=C), 1532 and 1330  $\text{cm}^{-1}$  ( $\text{NO}_2$ );  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$ : 4.54–4.55 (d, 1H,  $J=4.12$  Hz, thiazinanone-H6), 5.18–5.19 (d, 1H,  $J=4.12$  Hz, thiazinanone-H5), 6.80–7.00 (m, 4H, Ar–H), 7.10–7.32 (m, 9H, Ar–H), 7.35–7.52 (m, 3H, Ar–H), 8.04 (m, 1H, Ar–H), 8.98 (m, 1H, Ar–H), 10.98 (br, hydrazo-NH);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 47.60 (thiazinanone-CH5), 47.95 (thiazinanone-CH6), 115.97, 123.62, 127.71, 127.99, 128.39, 128.87, 129.35, 129.62, 130.08, 130.66, 131.03, 131.21 (Ar–CH), 129.35, 132.32, 133.77, 136.88, 137.86 (Ar–C), 144.57 (Ar–C–NH), 147.31 (C=N), 168.70 (C=O); MS ( $m/z$ ): 539 ( $\text{M}^+$ , 62), 462 (23), 357 (48), 207 (61), 135 (37), 77 (100); Anal. Calcd. for  $\text{C}_{28}\text{H}_{21}\text{N}_5\text{O}_5\text{S}$  (539.56): C, 62.33; H, 3.92; N, 12.98; S, 5.94. Found: C, 62.18; H, 3.77; N, 12.87; S, 5.86.

**(5S\*,6S\*,Z)-3-allyl-2-(2-(2,4-dinitrophenyl)hydrazono)-5,6-diphenyl-1,3-thiazinan-4-one (7c)**

Reddish orange crystals (acetonitrile), yield 402 mg (80%), mp. 208–210°C; IR (KBr)  $\nu$ : 3242 (NH), 3088 (Ar–CH), 2959–2925 (ali–CH), 1682 (C=O), 1614 (C=N), 1587 (Ar–C=C), 1529 and 1342  $\text{cm}^{-1}$  ( $\text{NO}_2$ );  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$ : 4.42–4.43 (d, 1H,

$J=4.08$  Hz, thiazinanone-H6), 4.70–4.92 (m, 2H, allyl- $\text{CH}_2\text{N}$ ), 4.96–4.97 (d, 1H,  $J=4.08$  Hz, thiazinanone-H5), 5.20–5.45 (m, 2H, allyl- $\text{CH}_2=$ ), 5.90–6.10 (m, 1H, allyl- $\text{CH}=\text{}$ ), 6.60–6.70 (m, 2H, Ar–H), 6.75–6.82 (m, 2H, Ar–H), 7.00–7.30 (m, 6H, Ar–H), 7.65 (m, 1H, Ar–H), 8.24 (m, 1H, Ar–H), 9.05 (m, 1H, Ar–H), 11.02 (br, 1H, hydrazo-NH);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 47.05 (allyl- $\text{CH}_2\text{N}$ ), 47.15 (thiazinanone-CH6), 47.92 (thiazinanone-CH5), 118.87 (allyl- $\text{CH}_2=$ ), 115.99, 123.60, 128.15, 128.31, 128.41, 129.24, 130.15, 130.22, 130.68 (Ar–CH), 133.75 (allyl- $\text{CH}=\text{}$ ), 128.79, 131.89, 133.69, 137.84 (Ar–C), 144.66 (Ar–C–NH), 146.32 (C=N), 168.43 (C=O); MS ( $m/z$ ): 503 ( $\text{M}^+$ , 72), 457 (18), 404 (26), 207 (55), 182 (100), 99 (66), 77 (87); Anal. Calcd. for  $\text{C}_{25}\text{H}_{21}\text{N}_5\text{O}_5\text{S}$  (503.53): C, 59.63; H, 4.20; N, 13.91; S, 6.37. Found: C, 59.45; H, 4.07; N, 13.74; S, 6.24.

**(5S\*,6S\*,Z)-2-(2-(2,4-dinitrophenyl)hydrazono)-3-ethyl-5,6-diphenyl-1,3-thiazinan-4-one (7d)**

Reddish orange crystals (acetonitrile), yield 387 mg (79%), mp. 200–202°C; IR (KBr)  $\nu$ : 3238 (NH), 3093 (Ar–CH), 2935–2923 (ali–CH), 1678 (C=O), 1616 (C=N), 1585 (Ar–C=C), 1526 and 1330  $\text{cm}^{-1}$  ( $\text{NO}_2$ );  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$ : 1.35 (t, 3H,  $J=7.77$  Hz,  $\text{CH}_3$ ), 4.15–4.18 (q, 2H,  $J=7.77$  Hz,  $\text{CH}_2$ ), 4.45–4.46 (d, 1H,  $J=4.10$  Hz, thiazinanone-H6), 4.96–4.97 (d, 1H,  $J=4.10$  Hz, thiazinanone-H5), 6.64–6.70 (m, 2H, Ar–H), 6.76–6.81 (m, 2H, Ar–H), 7.06–7.09 (m, 6H, Ar–H), 7.68 (m, 1H, Ar–H), 8.30 (m, 1H, Ar–H), 9.03 (m, 1H, Ar–H), 11.04 (br, 1H, hydrazo-NH);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 12.86 ( $\text{CH}_3$ ), 29.06 ( $\text{CH}_2$ ), 47.20, 47.70 (thiazinanone-CH6,5), 116.11, 123.63, 128.00, 128.30, 128.79, 129.22, 129.78, 130.10, 130.47 (Ar–CH), 129.12, 133.43, 133.80, 134.00, 137.87 (Ar–C), 144.62 (Ar–C–NH), 146.63 (C=N), 168.60 (C=O); MS ( $m/z$ ): 491 ( $\text{M}^+$ , 48), 462 (27), 445 (35), 402 (19), 182 (100), 87 (70), 77 (91); Anal. Calcd. for  $\text{C}_{24}\text{H}_{21}\text{N}_5\text{O}_5\text{S}$  (491.52): C, 58.65; H, 4.31; N, 14.25; S, 6.52. Found: C, 58.51; H, 4.20; N, 14.07; S, 6.38.

**(5S\*,6S\*,Z)-3-cyclohexyl-2-(2-(2,4-dinitrophenyl)hydrazono)-5,6-diphenyl-1,3-thiazinan-4-one (7e)**

Reddish orange crystals (acetonitrile), yield 436 mg (80%), mp. 236–238°C; IR (KBr)  $\nu$ : 3229 (NH), 3090 (Ar–CH), 2938–2923 (ali–CH), 1675 (C=O), 1613 (C=N), 1586 (Ar–C=C), 1527 and 1331  $\text{cm}^{-1}$  ( $\text{NO}_2$ );  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$ : 1.08–1.98 (m, 10H, cyclohexyl- $\text{CH}_2$ ), 2.02–2.46 (m, 1H, cyclohexyl-CH), 4.35–4.36 (d, 1H,  $J=4.11$  Hz, thiazinanone-H6), 4.80–4.81 (d, 1H,  $J=4.11$  Hz, thiazinanone-H5), 6.62–6.70 (m, 2H, Ar–H), 6.84–6.92 (m, 2H, Ar–H), 7.04–7.46 (m, 6H, Ar–H), 7.71 (m, 1H, Ar–H), 8.32 (m, 1H, Ar–H), 9.10 (m, 1H, Ar–H),

11.00 (br, hydrazo-NH);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 25.74, 26.53, 29.78 (cyclohexyl- $\text{CH}_2$ ), 47.63, 47.91 (thiazinanone- $\text{CH}_6,5$ ), 57.08 (cyclohexyl-CH), 115.93, 123.64, 128.04, 128.22, 128.71, 129.57, 130.25, 130.37, 130.77 (Ar-CH), 129.11, 133.38, 133.69, 137.96 (Ar-C), 144.59 (Ar-C-NH), 146.51 (C=N), 168.87 (C=O); MS ( $m/z$ ): 545 ( $\text{M}^+$ , 51), 462 (43), 499 (28), 455 (39), 182 (100), 141 (68), 77 (91); Anal. Calcd. For  $\text{C}_{28}\text{H}_{27}\text{N}_5\text{O}_5\text{S}$  (545.61): C, 61.64; H, 4.99; N, 12.84; S, 5.88. Found: C, 61.48; H, 4.83; N, 12.74; S, 5.75.

#### (Z)-N'-(2,4-dinitrophenyl)-2,3-diphenylacrylohydrazide (**8**)

Yellow crystals (acetonitrile), yield 8–12%, mp. 168–169°C; IR (KBr)  $\nu$ : 3247 (NH), 3130 (Ar-H), 2930 (ali-H), 1693 (C=O), 1591 (Ar-C=C), 1542 and 1334  $\text{cm}^{-1}$  ( $\text{NO}_2$ );  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$ : 6.42 (s, 1H, acryl-CH), 7.10–7.55 (m, 10H, Ar-H), 7.94 (m, 1H, Ar-H), 8.48 (br, 1H, NH), 9.04 (m, 1H, Ar-H), 9.43 (br, 1H, amide-NH);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 117.65, 123.18, 126.17, 127.52, 128.15, 128.26, 129.19, 129.38, 129.85 (Ar-CH), 133.68, 139.75 (C1 and C2-acrylohydrazide); 130.00, 134.34, 135.60, 137.80 (Ar-C); 145.10 (Ar-C-NH), 165.18 (C=O); MS ( $m/z$ ): 404 ( $\text{M}^+$ , 100), 356 (23), 221 (26), 205 (13), 195 (18), 181 (32), 138 (20), 77 (41); Anal. Calcd. for  $\text{C}_{21}\text{H}_{16}\text{N}_4\text{O}_5$  (404.38): C, 62.37; H, 3.99; N, 13.86. Found: C, 62.25; H, 3.86; N, 13.77.

#### Single-crystal X-ray structure determination of **7a** and **8**

Suitable crystals were obtained by recrystallization from acetonitrile. The single-crystal X-ray diffraction study was carried out on a Bruker D8 Venture diffractometer with Photon100 detector at 123(2) K using Cu- $K\alpha$  radiation ( $\lambda = 1.54178 \text{ \AA}$ ). Direct Methods for **7a** (SHELXS-97) [26] and dual space methods for **8** (SHELXT) [27] were used for structure solution and refinement was carried out using SHELXL-2014 (full-matrix least-squares on  $F^2$ ) [28]. Hydrogen atoms were localized by difference electron density determination and refined using a riding model (H(N) free). Semi-empirical absorption corrections were applied. For **8** an extinction correction was applied. In **8** the 2,3-diphenylacrylo substituent is disordered (see cif-files for details).

**Compound 7a** Red crystals,  $\text{C}_{29}\text{H}_{23}\text{N}_5\text{O}_5\text{S}$ ,  $M_r = 553.58$ , crystal size  $0.24 \times 0.06 \times 0.02 \text{ mm}$ , monoclinic, space group  $P2_1/c$  (No. 14),  $a = 19.9611(6) \text{ \AA}$ ,  $b = 11.4148(4) \text{ \AA}$ ,  $c = 10.9667(4) \text{ \AA}$ ,  $\beta = 93.844(1)^\circ$ ,  $V = 2493.16(15) \text{ \AA}^3$ ,  $Z = 4$ ,  $\rho = 1.475 \text{ Mg/m}^{-3}$ ,  $\mu(\text{Cu-K}\alpha) = 1.601 \text{ mm}^{-1}$ ,  $F(000) = 1152$ ,  $2_{\text{max}} = 144.2^\circ$ , 21195 reflections, of which 4907 were independent ( $R_{\text{int}} = 0.030$ ), 364 parameters, 1 restraint,  $R_1 = 0.046$

(for  $4478 I > 2\sigma(I)$ ),  $wR_2 = 0.118$  (all data),  $S = 1.08$ , largest diff. peak/hole =  $0.723/-0.490 \text{ e \AA}^{-3}$ .

**Compound 8** Yellow crystals,  $\text{C}_{21}\text{H}_{16}\text{N}_4\text{O}_5 \cdot \text{C}_2\text{H}_6\text{OS}$ ,  $M_r = 482.50$ , crystal size  $0.32 \times 0.16 \times 0.12 \text{ mm}$ , triclinic, space group  $P-1$  (No. 2),  $a = 9.0191(3) \text{ \AA}$ ,  $b = 11.3050(3) \text{ \AA}$ ,  $c = 11.8112(3) \text{ \AA}$ ,  $\alpha = 87.986(1)^\circ$ ,  $\beta = 69.079(1)^\circ$ ,  $\gamma = 79.272(1)^\circ$ ,  $V = 1104.58(6) \text{ \AA}^3$ ,  $Z = 2$ ,  $\rho = 1.451 \text{ Mg/m}^{-3}$ ,  $\mu(\text{Cu-K}\alpha) = 1.732 \text{ mm}^{-1}$ ,  $F(000) = 504$ ,  $2\theta_{\text{max}} = 144.0^\circ$ , 15,897 reflections, of which 4309 were independent ( $R_{\text{int}} = 0.022$ ), 313 parameters, 10 restraints,  $R_1 = 0.038$  (for  $4234 I > 2\sigma(I)$ ),  $wR_2 = 0.093$  (all data),  $S = 1.06$ , largest diff. peak/hole =  $0.584/-0.369 \text{ e \AA}^{-3}$ .

**Acknowledgements** Alaa A. Hassan is indebted to the AvH-Foundation for the donation of a Shimadzu 408 IR instrument.


#### References

- Rawal RK, Tripathi R, Katti SB, Pannecouque C, De Clercq E (2007) Synthesis and evaluation of 2-(2,6-dihalophenyl)-3-pyrimidinyl-1,3-thiazolidin-4-one analogues as anti-HIV-1 agents. *Bioorg Med Chem* 15:3134–3142. <https://doi.org/10.1016/j.bmc.2007.02.044>
- Rawal RK, Tripathi R, Katti SB, Pannecouque C, De Clercq E (2007) Design, synthesis, and evaluation of 2-aryl-3-heteroaryl-1,3-thiazolidin-4-ones as anti-HIV agents. *Bioorg Med Chem* 15:1725–1731. <https://doi.org/10.1016/j.bmc.2006.12.003>
- Verma A, Saraf SK (2008) 4-Thiazolidinone-A biologically active scaffold. *Eur J Med Chem* 43:897–905. <https://doi.org/10.1016/j.ejmech.2007.07.017>
- Kamel MM, Ali HI, Anwar MM, Mohamed NA, Soliman AM (2010) Synthesis, antitumor activity and molecular docking study of novel Sulfonamide-Schiff's bases, thiazolidinones, benzothiazinones and their C-nucleoside derivatives. *Eur J Med Chem* 45:572–580. <https://doi.org/10.1016/j.ejmech.2009.10.044>
- Verma A, Verma SS, Saraf SK (2010) A DIC mediated expeditious small library synthesis and biological activity of thiazolidin-4-one and 1,3-thiazinan-4-one derivatives. *J Heterocycl Chem* 47:1084–1089. <https://doi.org/10.1002/jhet.429>
- Kumawat MK, Singh UP, Singh B, Prakash A, Chetia D (2016) Synthesis and antimalarial activity evaluation of 3-(3-(7-chloroquinolin-4-ylamino)propyl)-1,3-thiazinan-4-one derivatives. *Arab J Chem* 9:S643–S647. <https://doi.org/10.1016/j.arabj.2011.07.007>
- Bosenbecker J, Bareño VDO, Difabio R, Vasconcellos FA, Dutra FSP, Oliveira PS, Barschak AG, Stefanello FM, Cunico WJ (2014) Synthesis and antioxidant activity of 3-(Pyridin-2-ylmethyl)-1,3-thiazinan(thiazolidin)-4-ones. *Biochem Mol Toxicol* 28:425–432. <https://doi.org/10.1002/jbt.21581>
- Zebardast T, Zarghi A, Daraie B, Hédayati M, Dadrass OG (2009) Design and synthesis of 3-alkyl-2-aryl-1,3-thiazinan-4-one derivatives as selective cyclooxygenase (COX-2) inhibitors. *Bioorg Med Chem Lett* 19:3162–3165. <https://doi.org/10.1016/j.bmcl.2009.04.125>
- Eicher T, Franke G (1981) On the reaction of diphenylcyclopropanone with guanidines. *Chem, Liebigs Ann.* <https://doi.org/10.1002/jlac.198119810802>

10. Takahashi M, Funaki T, Honda H, Yokoyama Y, Takimoto H (1982) Formation of 1,2,3,5-tetrasubstituted 2-pyrrolin-4-ones and 1,2,3-trisubstituted pyrroles from diphenylcyclopropenone and 1,4-diazabutadienes. *Heterocycles* 19:1921–1924. <https://doi.org/10.3987/R-1982-10-1921>
11. Gomaa MAM, Döpp D (2003) Reaction of *N*<sup>1</sup>,*N*<sup>2</sup>-diarylamidines with (2,3-diphenylcyclopropen-1-ylidene)propanedinitrile: synthesis of [2-arylamino-4(1H)-pyridinylidene]-propanedinitriles. *Synthesis* 10:1545–1548. <https://doi.org/10.1055/s-2003-40522>
12. Gomaa MAM (2002) Reaction of diimines and azines with diphenylcyclopropenone. *J Chem Soc: Perkin Trans* 1:341–344. <https://doi.org/10.1039/b109711n>
13. Aly AA, Ramadan M, Abd Al-Aziz M, Fathy HM, Bräse S, Brawn AB, Niger M (2016) Reaction of amidrazones with 2,3-diphenylcyclopropenone: synthesis of 3-(aryl)-2,5,6-triphenylpyrimidin-4(3H)-ones. *J Chem Res* 40:637–639. <https://doi.org/10.3184/174751916x14743924874916>
14. Hassan AA, Abdel-Latif FF, Nour El-Din AM, Mostafa SM, Nieger M, Bräse S (2012) Synthesis of (*E*)-2,5-disubstituted 1,3,4-thiadiazolyl-2,3-diphenylpropenones from alkenylidenehydrazinecarbothioamides. *Tetrahedron* 68:8487–8492. <https://doi.org/10.1016/j.tet.2012.07.063>
15. Aly AA, Hassan AA, Ameen MA, Brown AB (2008) Chemistry of cyclopropenones: synthesis of new pyrrolo[2,1-*b*]-1,3,4-oxadiazoles. *Tetrahedron Lett* 49:4060–4062. <https://doi.org/10.1016/j.tetlet.2008.04.066>
16. Aly AA, Hassan AA, Gomaa MAM, El-Sheref EM (2007) Unusual reactivity of thiosemicarbazides towards 2,3-diphenylcyclopropenone: synthesis of new pyridazinethiones and 1,2,4-triazolo[4,3-*b*]pyridazinethiones. *Arkivoc* xiv:1–11. <https://doi.org/10.3998/ark.5550190.0008.e01>
17. Aly AA, Nour El-Din AM, Gomaa MAM, Brawn AB, Fahmi MS (2007) Unusual reactivity of 2,3-diphenylcyclopropenone towards *N*-imidoylthioureas; facile synthesis of 3-aryl-2,5,6-triphenylpyrimidin-4(3H)-one (Part III). *J Chem Res* 2007:439–441. <https://doi.org/10.3184/030823407x234563>
18. El-Sheref EM (2017) One-pot synthesis of 3-substituted-2-[(3-methyl-1H-pyrazol-5-yl)imino]-5,6-diphenyl-1,3-thiazin-4-ones. *J Sulfur Chem* 39:625–634. <https://doi.org/10.1080/17415993.2017.1337121>
19. Clayden J, MacLellan P (2011) Asymmetric synthesis of tertiary thiols and thioethers. *Beilstein J Org Chem* 7:582–595. <https://doi.org/10.3762/bjoc.7.68>
20. Damani LA (1989) Sulphur-containing drugs and related organic compounds. Wiley, New York
21. Niu B, Jiang B, Yua LZ, Shi M (2018) Base-promoted [3 + 3] cyclization of cyclopropenones and cyclopropenethiones with amides for the synthesis of 6H-1,3-oxazin-6-ones and 6H-1,3-thiazin-6-ones. *Org Chem Front* 5:1267–1271. <https://doi.org/10.1039/c8qo00091c>
22. Gomaa MAM (2004) Reaction of *N*<sup>1</sup>,*N*<sup>2</sup>-Diarylamidines with 2,3-Diphenylcyclopropenone. *Z Naturforsch B* 59:597–600. <https://doi.org/10.1515/znB-2004-0520>
23. Shan L, Wu G, Liu M, Gao W, Ding J, Huang X, Wu H (2018)  $\alpha$ ,  $\beta$ -Diaryl unsaturated ketones via palladiumcatalyzed ring-opening of cyclopropenones with organoboronic acids. *Org Chem Front* 5:1651–1654. <https://doi.org/10.1039/c8qo00241j>
24. Parashar B, Jain A, Bharadwaj S, Sharma VK (2012) Synthesis and pharmacological properties of some novel pyrazolidine and pyrazole derivatives. *Med Chem Res* 21:1692–1699. <https://doi.org/10.1007/s00044-011-9687-0>
25. Begum S, Choudhary MI, Khan KM (2009) Synthesis, phytotoxic, cytotoxic, acetylcholinesterase and butyrylcholinesterase activities of *N,N'*-diaryl unsymmetrically substituted thioureas. *Nat Prod Res* 23:1719–1730. <https://doi.org/10.1080/14786410802223778>
26. Sheldrick GM (2008) A short history of SHELX. *Acta Cryst A* 64:112–122. <https://doi.org/10.1107/S0108767307043930>
27. Sheldrick GM (2015) SHELXT-Integrated space-group and crystal-structure determination. *Acta Cryst A* 71:3–8. <https://doi.org/10.1107/S2053273314026370>
28. Sheldrick GM (2015) Crystal structure refinement with SHELXL. *Acta Cryst C* 71:3–8. <https://doi.org/10.1107/S2053229614024218>

**Publisher's Note** Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

## Affiliations

Alaa A. Hassan<sup>1</sup>  · Nasr K. Mohamed<sup>1</sup> · Ashraf A. Aly<sup>1</sup> · Hendawy N. Tawfeek<sup>1</sup> · Henning Hopf<sup>2</sup> · Stefan Bräse<sup>3</sup> · Martin Nieger<sup>4</sup>

<sup>1</sup> Chemistry Department, Faculty of Science, Minia University, El-Minia 61519, Egypt

<sup>2</sup> Institut für Organische Chemie, Technische Universität Braunschweig, 38092 Brunswick, Germany

<sup>3</sup> Institute of Organic Chemistry, Karlsruhe Institute of Technology, Fritz-Haber-Weg 6, 76131 Karlsruhe, Germany

<sup>4</sup> Department of Chemistry, University of Helsinki, A. I. Virtasen aukio I, P.O. Box 55, 00014 Helsinki, Finland