

# **Performance improvement of solar still by using hybrid nanofuids**

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#### **Abstract**

The biggest shortcoming of conventional solar still is its limited productivity. In this work, improvement of solar still performance using two hybrid nanofuids is examined. Use of hot water and two conventional nanofuids were also utilized to study the improvement of the performance of solar still. Hybrid nanofuids, hot water or conventional nanofuids were linked to the solar system with heat exchanger placed at the bottom of the basin of the solar still. Pure desalinated water yield, the first and the second law efficiencies of the solar still with hybrid nanofluid were observed through modeling. Utilization of heat exchanger with hot water improved the solar still performance. Conventional nanofuids showed better performance than hot water. The best performance is obtained with the use of hybrid nanofluids. The yield, thermal efficiency and exergy efficiency with water–Al<sub>2</sub>O<sub>3</sub>–SiO<sub>2</sub> hybrid nanofluid resulted in 4.99 kg m<sup>-2</sup> day<sup>-1</sup>, 37.76% and 0.82%, respectively. The basin water depth, nanofuid fow rate, inlet heat exchanger temperature of the nanofuid and nanoparticle concentration played vital role together with solar radiation and ambient temperature to improve the performance.

Keywords Solar still · Nanofluid · Hybrid nanofluid · Performance · Efficiency

# **Introduction**

The conquest of fnding safe drinking water has been going on since old times. One of the simplest techniques of producing freshwater is solar desalination in solar still. The technological simplicity, use of abundantly available solar energy and low cost make it a popular choice for the people of remote locations specially in less developed countries [[1\]](#page-13-0). Conventional solar still has very low productivity and very poor thermal performance due to convective and radiative heat losses. Use of nanofuid is one of the most notable ways of overcoming these shortcomings. Usage of nanofuid is found in numerous cases in the literature for analyzing free or natural convection, energy and exergy of diferent systems and diferent solar energy applications, etc. [\[2](#page-13-1)[–6](#page-14-0)]. Nanofluids, prepared by adding nanometer-sized additives in base fuid, possess better thermophysical and optical properties to make better heat transfer and reduced heat loss to produce more fresh water. Heat transfer coefficients are increased by using nanofuids which in turn increases

 $\boxtimes$  Ahmet Z. Sahin azsahin@kfupm.edu.sa the production of solar still. That is why nanofuids are being used in solar still nowadays to improve the production rate and the thermal performance together with some other modifcations like adding condenser, heat exchanger, wicks, vacuum system, evacuated tube, etc. [[7–](#page-14-1)[9\]](#page-14-2). Use of Nanofuid in producing fresh water from the salty sea water is very benefcial for the environment too. It was found out that the use of copper oxide–water nanofuid reduces signifcant amount of carbon dioxide [[10](#page-14-3)]. These are the reasons of using nanofuids extensively in various energy related felds such as domestic refrigeration, absorption system, thermal management of diferent electronic equipment, medicine, building heating and cooling, etc. although the preparation cost of nanofluids is quite high  $[11, 12]$  $[11, 12]$  $[11, 12]$ . It was experimentally found that the use of copper oxide  $(Cu<sub>2</sub>O)$  nanofluid with external Thermoelectric glass cover cooling channel enhances the performance of solar still signifcantly. With this modifcation, varying the volumetric amount of nanofuid results in higher amount of pure water from brackish water, better energy efficiency and better exergy efficiency compared to conventional solar stills. The performance with the improved solar still shows that the production of water, energy efficiency and the second law efficiency increases by 81%, 80.6% and 112.5%, respectively [\[13\]](#page-14-6). In another study with  $Cu<sub>2</sub>O$  nanofluid with an external thermoelectric

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condensing channel also resulted in very high efficiency of the solar still. Addition of 0.08% volume fraction of this nanoparticle, the yield of pure water, frst law and second law efficiencies were enhanced by about  $82.4\%$ ,  $81.5\%$  and 92.5%, respectively [\[14](#page-14-7)]. In another study of solar still with external condenser with mass concentration of 0.02–0.03%  $Cu<sub>2</sub>O$  and  $Al<sub>2</sub>O<sub>3</sub>$  nanoparticles resulted in daily efficiency of 84.16% and 73.85%, respectively. The daily efficiency of conventional still was only 34% [\[15](#page-14-8)]. Fresh water production increases by 116% using Aluminum oxide–water nanofuid with external condenser  $[16]$ . A VOF (volume of Fluid) model was utilized to study the heat transfer phenomenon and the performance of a solar still using  $Al_2O_3$ –water nanofuid. The fndings showed that the productivity increases about 25%, Nusselt number increases about 18% when the solid volume fraction of the nanoparticles increased from 0 to 5% [[17\]](#page-14-10).

Arrangement of vacuum creation together with nanofuids also increases the solar still performance signifcantly. Cuprous oxide nanoparticles (concentration of 0.2%) with vacuum and without vacuum increased freshwater production by 133.64% and 93.87%, respectively. 125% and 88.97% increase of distillate production recorded for the case of  $Al_2O_3$  nanofluid [[18](#page-14-11)]. Using  $Al_2O_3$ –Therminol 55 (nHTF), nanofuid with a modifed solar still with a Fresnel lens concentrator with evacuated tube receiver demonstrates remarkable improvement. It was experimentally found out that hourly production of fresh water to be 45–250.27% more than the conventional solar still using 0.1% nHTF nanofluid. Total yield was found 12.19 L m<sup>-2</sup> day<sup>-1</sup> which is much higher than conventional solar still producing only 3.48 L m<sup>-2</sup> day<sup>-1</sup>. The maximum efficiency found was 53.55% with this nanofuid [[19\]](#page-14-12). Hybrid solar still with bottom basin single sloped and upper basin double sloped with  $MgCl<sub>2</sub>$  as regenerator and ZnO nanoparticles was constructed and the performance showed pure water yield of 623.33 mL m<sup>-2</sup> and overall thermal efficiency of 19.20% [\[20\]](#page-14-13). Graphite and Copper oxide micro-fakes were utilized in varying concentrations as nanoparticles in water. The productivity with these nanofuids showed 53.95% and 44.91% more than the conventional simple solar still, respectively. With the use of glass film cooling by water together with these nanofluids improved productivity by 57.60% and 47.80%, respectively. The conventional solar still showed only 30% of productivity [[21\]](#page-14-14). The performance of diferent nanofuids was also investigated in the case of pure water production from a solar still. The solar still with diferent nanofuids is compared to a conventional water solar still. The result shows that the still with  $Al_2O_3$  nanofluid has 29.95%, with ZnO has  $12.67\%$  and with SnO<sub>2</sub> has  $18.63\%$ higher production than conventional solar still [[22\]](#page-14-15).

In another analytical study (modeling) of solar still with heat exchanger together with  $SiO_2$ –water and Cu–water nanofluids, freshwater yield, first law and second law efficiencies improved when inlet heat exchanger temperature of nanofuid was more than 60 °C. At lower temperature, solar still with Cu–water nanofuid showed better result than  $SiO<sub>2</sub>$ –water nanofluid. But, with the inlet heat exchanger temperature more than 60 °C,  $SiO_2$ -water nanofluid showed higher yield, higher thermal and exergy efficiency than the Cu–water nanofluid. However, both the nanofluids performed much better than convention solar still at higher temperature. The mass fow rate and concentration of the nanoparticles also have noticeable impact on the perfor-mance of the solar still [[23\]](#page-14-16).

Traditional Pyramid Distiller (TPD) is a solar still in the shape of a pyramid. The performance of TPD increases with the addition of V corrugated absorbers and with the combination of both V corrugated materials and wick materials. The performance of TPD is even further increased with the addition of CuO nanoparticles with the wick materials. This addition of CuO nanoparticles together with the wick materials increased the productivity of fresh water, thermal efficiency and exergy efficiency by  $72.95\%$ ,  $77.9\%$  and  $93\%$ , respectively, compared to TPD [\[24](#page-14-17)]. Thermoelectric heating (TEH) in the basin, TEH together with Silver nanofuid and combination of TEH, Silver nanofuid and external condenser were used separately to check the performance of solar still. The yield of pure water and thermal efficiency of solar still with nanofuid together with condenser showed 100.5% and 26.7% improvement than the solar still without both nanofuid and condenser. Even, the solar still with nanofuid without condenser also showed 50.8% and 30.6% improvement for yield and thermal efficiency, respectively, in comparison with traditional solar still [[25\]](#page-14-18). Environment friendly coffee-based colloid was also used as nanofluid to boost the production of pure desalinated water. The experimental results showed the increase of yield of fresh water, thermal efficiency and exergy efficiency by  $35.14\%$ ,  $35.34\%$ and 46.44%, respectively, compared to traditional solar still [[26\]](#page-14-19). Evacuated tubes together with CuO and Carbon black (CB) nanoparticles are integrated with Conventional Pyramid Solar Still (CPSS) to make the Modifed Pyramid Solar Still (MPSS). Fresh water yield of MPSS with CuO resulted in 54.48% and 27.85% higher production than Conventional Solar Still (CSS) and CPSS, respectively. Again, the production of yield of MPSS with CB was 57.098% and 33.59% higher than the CSS and CPSS, respectively. The efficiency of MPSS reached to 61% and 64% with CuO and CB, respectively [[27\]](#page-14-20).

Thus, it can be said that the performance improvement of the solar still using nanofuid is obvious. However, the selection of the appropriate nanofuid for the highest production and efficiency is not yet decided. The research is being carried out to find out the maximum possible efficiency of the solar still desalination system. One of the areas of study to maximize efficiency of solar still is the utilization of various nanofuids.

In this paper, we aim at fnding out the pure water yield and efficiency of a modified solar still using hybrid nanofuids. Because of possessing better properties than the conventional nanofuids, much better performance is expected from the solar still with hybrid nanofuids.

## **Mathematical modeling**

### **Schematic representation of the solar still**

The solar still considered in this work is equipped with a heat exchanger underneath the feed water (saline or impure water) as shown in Fig. [1](#page-2-0). Hot water, conventional nanofluid or hybrid nanofluid flows through the heat exchanger to exchange heat with the feed water. However, the nanofuid does not mix with the water as the nanofuid fows through the tubes of the heat exchanger. A heater is used to heat the nanofuid before entering the heat exchanger. The schematic view of the heat exchanger inside the solar still is shown schematically in Fig. [2.](#page-2-1) We considered length of solar still,  $L_{ss}=1$  m and width of solar still,  $W_{ss}=0.85$  m for our study. Depth of water  $(d_w)$  of the solar still is varied to observe the change of performance. The input parameters used in the present analysis are the hourly solar radiation intensity, temperature of surrounding atmosphere, speed of wind at

<span id="page-2-0"></span>**Fig. 1** Schematic diagram of solar still system



Following assumptions are considered while analyzing the current solar still system [[8\]](#page-14-21).

- There is no vapor or water leakage from the system
- The temperature of all the layers of basin water is considered same for any fxed period of time
- There is no heat absorption on the outer surface of the glass cover
- The processes in this system are considered to be quasi steady state
- The nanofluid is stable with well dispersed nanoparticles without any sedimentation

#### <span id="page-2-2"></span>**Thermodynamic analysis**

The present model of the modifed solar still consists of frst law and second law analysis. Here, various mathematical equations are presented for energy balance at diferent parts of the system. The main parts of the system are the sloped glass, basin water, heat exchanger and basin liner. Although the evaporation heat transfer coefficient is the dominant factor to produce distilled water; conduction, convection and radiation heat transfer coefficients also play important role.



<span id="page-2-1"></span>



Schematic diagram of the electrical analogy for the heat transfer in and out of the Solar Still is visualized in Fig. [3.](#page-3-0)

#### **First law of thermodynamics analysis**

First law of thermodynamics includes equations of energy balance for specifc components of the solar still system. The equations of energy balance for various components of the system are described in the following paragraphs.

**Glass cover outer surface** The heat reaching the outer surface of the glass cover from the inner surface equals to the amount of heat going from the outer surface to the environment. It is considered that heat is not stored at the outer face of the glass cover

$$
q_1 = q_2 \tag{1}
$$

where

$$
q_1 = h_{t,g,o-a}(T_{g,o} - T_a)
$$
 (2)

$$
q_2 = \frac{k_g}{L_g} (T_{g,i} - T_{g,0})
$$
\n(3)

$$
h_{\text{t,g,o-a}} = 5.7 + 3.8V_{\text{w}} \tag{4}
$$

 $q_1$  reflects heat transfer from the external surface of glass cover to the atmosphere and  $q_2$  denotes heat transfer from inside surface to the outer surface of the glass cover.  $k_g$  is thermal conductivity of glass,  $L_g$  is the thickness of the glass, *T* is temperature.  $h_{t,g,o-a}$  is the total heat transfer coefficient from outer surface of glass to ambient. Subscript i, o, a, t, g



<span id="page-3-0"></span>**Fig. 3** Schematic view of the electrical analogy for the heat transfer in and out of the solar still

means inner, outer, ambient, total, and glass, respectively.  $V_w$  means wind velocity.

**Inner surface of glass cover** Solar radiation reaches the face of the glass, and some of the incident energy is absorbed in the inner surface. Some heat comes to the interior face of the glass cover from basin water. This phenomenon can be expressed as Eqs.  $(5-7)$  $(5-7)$  $(5-7)$  [\[28\]](#page-14-22)

<span id="page-3-1"></span>
$$
q_2 = q_3 + q_{\text{abs},2} \tag{5}
$$

here,

$$
q_{\text{abs},2} = \left(1 - R_{\text{g}}\right) \alpha_{\text{g}} I_{\text{s}}(t); \tag{6}
$$

<span id="page-3-2"></span>
$$
q_3 = (h_{\rm r, wgi} + h_{\rm ev, wgi} + h_{\rm c, wgi}) (T_{\rm w} - T_{\rm g,i});
$$
\n(7)

The heat absorbed in the inside surface of glass from hourly solar radiation is  $q_{\text{abs},2}$ . The heat reaching the glass inner surface from the water is  $q_3$ . All temperatures here are in kelvin unit. Heat transfer coefficient is denoted by *h*. Subscripts r, ev, c, wgi means radiation, evaporation, conduction and water to inner glass surface, respectively.  $R$ ,  $\alpha$ ,  $I_s$  and  $P$  stands for refectivity, absorptivity, hourly solar radiation intensity and pressure, respectively. Stefan Boltzmann constant and emissivity are marked with  $\sigma$  and  $\varepsilon$ , respectively. The evaporation heat transfer coefficient is estimated using the Dunkle relation stated in Eq. [\(8](#page-3-3)). Radiation and conduction heat transfer coefficients can be found from Eqs.  $(9)$  and  $(10)$  below [[23\]](#page-14-16)

<span id="page-3-3"></span>
$$
h_{\text{ev,wgi}} = 0.016273 \ast \mathbf{h}_{\text{c,w-g,i}} \left( \frac{P_{\text{w}} - P_{\text{g,i}}}{T_{\text{w}} - T_{\text{g,i}}} \right)
$$
(8)

<span id="page-3-4"></span>
$$
h_{\rm r, wgi} = \sigma \left( T_{\rm w}^2 + T_{\rm g,i}^2 \right) \left( T_{\rm w} - T_{\rm g,i} \right) \left[ \frac{1}{\varepsilon_{\rm w}} + \frac{1}{\varepsilon_{\rm g}} - 1 \right]^{-1} \tag{9}
$$

<span id="page-3-5"></span>
$$
h_{\rm c,wgi} = 0.884 \left[ \left( T_{\rm w} - T_{\rm g,i} \right) + \frac{(P_{\rm w} - P_{\rm g,i}) T_{\rm w}}{268,900 - P_{\rm w}} \right]^{1/3} \tag{10}
$$

Here,  $P_w$  and  $P_{gi}$  are the saturated vapor pressure of water at temperature of water and inner surface of glass, respectively, which can be found out using Eqs.  $(11)$  and  $(12)$  $(12)$ 

<span id="page-3-6"></span>
$$
P_{\rm w} = \exp\left(25.317 - \frac{5144}{T_{\rm w}}\right) \tag{11}
$$

<span id="page-3-7"></span>
$$
P_{\rm g,i} = \exp\bigg(25.317 - \frac{5144}{T_{\rm g,i}}\bigg). \tag{12}
$$

**Water of the basin** The water of the solar still absorbs the incident solar radiation, the useful heat gain from the heated nanofuid and the heat from the basin of the still. Thus, the temperature of the water of the still increases

$$
q_3 A_b + Q_{st} = q_4 (A_b - A_{HE}) + q_{abs,3} A_b + Q_u
$$
 (13)

Here,

$$
Q_{\rm st} = M_{\rm w} C_{\rm w} \frac{\mathrm{d}T_{\rm w}}{\mathrm{d}t} \tag{14}
$$

$$
q_{\text{abs},3} = A_{\text{b}} \alpha'_{\text{w}} I_{\text{s}}(t) \tag{15}
$$

$$
q_4 = (A_b - A_{\rm HE})h_{\rm w}(T_b - T_{\rm w})
$$
\n(16)

$$
Q_{\rm u} = n \dot{m}_{\rm riser,HE} C_{\rm p,nf} \left( T_{\rm i,HE} - T_{\rm o,HE} \right) \tag{17}
$$

 $q_{\text{abs,3}}$  is the heat absorbed in the water from hourly solar radiation. The amount of heat reaching the water from the basin is marked with  $q<sub>4</sub>$ . The stored energy due to heat received from nanofuid and solar radiation is identifed by  $Q_{st}$ . The useful energy gain from the heated nanofluid passing through the heat exchanger is  $Q_{\rm u}$ .  $C_{\rm w}$  and  $C_{\rm p, nf}$  are the specifc heat capacity of water and nanofuid, respectively. To increase the water temperature, all the forms of heat transfer namely conduction, radiation and evaporation plays considerable role. Thus, the total heat transfer coefficient in this case equals to the added values of the conduction, radiation and evaporation heat transfer coefficients.

$$
h_{t,\text{wgi}} = h_{c,\text{wgi}} + h_{r,\text{wgi}} + h_{\text{ev,wgi}} \tag{18}
$$

In Eq. [\(18](#page-4-0)), the symbols *t*, *c*, *r*, *ev* and *wgi* represent total, conduction, radiation, evaporation and water to glass inside, respectively. The total heat transfer coefficient from water to inner glass surface is denoted by  $h_{t,wei}$ . The mass of water in the basin is  $M_w$  which can be obtained using Eq. [\(19](#page-4-1))

$$
M_{\rm w} = \rho_{\rm w} \left( A_{\rm b} d_{\rm w} - V_{\rm HE} \right) \tag{19}
$$

$$
\alpha'_{\rm w} = \left(1 - \alpha_{\rm g}\right)\left(1 - R_{\rm g}\right)\left(1 - R_{\rm w}\right)\alpha_{\rm w} \tag{20}
$$

In Eqs. [\(19](#page-4-1)) and [\(20](#page-4-2)),  $A<sub>b</sub>$ ,  $A<sub>HE</sub>$  are the surface area of basin and heat Exchanger, respectively.  $d_w$ ,  $\rho_w$  are the depth of water and density of water, respectively. The volume of heat exchanger is  $V_{\text{HF}}$ . The number of risers of heat exchanger is denoted by '*n*'. Absorptivity of water and glass are symbolized with  $\alpha_w$ ,  $\alpha_g$ , respectively. Reflectivity of water and glass are denoted by  $R_w$  and  $R_g$ , respectively.

The convection heat transfer coefficient  $h_w$  between the liner of basin and water is derived from Eq. ([21\)](#page-4-3) as follows [\[29\]](#page-14-23)

<span id="page-4-3"></span>
$$
h_{\rm w} = \frac{\overline{\rm Nu}_{\rm L} k_{\rm w}}{L_{\rm c}}\tag{21}
$$

Here,  $k_w$  is thermal conductivity of water and  $\overline{\text{Nu}}_L$  is the average Nusselt number.  $L_c$  is the ratio of surface area to perimeter of the basin

$$
\overline{Nu}_{L} = 0.54 \text{Ra}_{L}^{1/4} \quad \text{when} \quad 10^{4} \le \text{Ra}_{L} \le 10^{7} \tag{22}
$$

$$
\overline{Nu}_{L} = 0.15Ra_{L}^{1/3} \quad \text{when} \quad 10^{7} \le Ra_{L} \le 10^{11} \tag{23}
$$

where

$$
Ra_{L} = \frac{g\beta_{w}(T_b - T_w)L_c^3}{\alpha_{1w}v_w}
$$
\n(24)

here Ra<sub>L</sub>,  $\alpha_{1w}$ ,  $\beta_w$ ,  $v_w$  are Raleigh number, thermal diffusivity, thermal expansion coefficient and kinematic viscosity of the basin water.

**Basin liner** The basin liner absorbs incident solar radiation. The liner also accepts heat from or rejects heat to the atmosphere

$$
q_4 = q_5 + q_{\text{abs},4} \tag{25}
$$

Here

<span id="page-4-4"></span>
$$
q_{\text{abs},4} = \alpha'_{\text{b}}I_{\text{s}}(t) \tag{26}
$$

<span id="page-4-5"></span><span id="page-4-0"></span>
$$
q_5 = h_b (T_a - T_b) \tag{27}
$$

The heat reaching the basin from atmosphere is denoted by  $q_5$  and  $q_{\text{abs},4}$  is the absorbed heat because of solar radiation in the basin liner. From Eqs.  $(26)$  $(26)$  and  $(27)$  $(27)$  $(27)$ , the parameters  $\alpha'_{b}$  and  $h_{b}$  can be computed as in Eqs. [\(28](#page-4-6)) and ([29\)](#page-4-7)

<span id="page-4-6"></span><span id="page-4-1"></span>
$$
\alpha'_{\mathbf{b}} = (1 - \alpha_{\mathbf{g}}) (1 - \alpha_{\mathbf{w}}) (1 - R_{\mathbf{g}}) (1 - R_{\mathbf{w}}) \alpha_{\mathbf{b}} \tag{28}
$$

<span id="page-4-7"></span><span id="page-4-2"></span>
$$
h_{\rm b} = 1 / \left( \frac{L_{\rm ins}}{k_{\rm ins}} + \frac{1}{h_{t, \rm b-a}} \right)
$$
 (29)

Here,  $h<sub>b</sub>$  is the heat transfer coefficient between the basin liner and atmosphere. Insulation thickness and thermal conductivity are represented by  $L_{ins}$  and  $k_{ins}$ , respectively. Absorptivity of the basin is identified with  $\alpha_{\rm b}$ . The total heat transfer coefficient from basin to atmosphere  $(h<sub>t,b-a</sub>)$ is another important parameter which depends on the wind velocity  $(V_w)$ 

$$
h_{t,b-a} = 5.7 + 3.8V_{w}
$$
\n(30)

**Heat exchanger** The energy equation for heat exchanger can be stated as Eq.  $(31)$  $(31)$ 

$$
\frac{\mathrm{d}T_{\rm f}}{\mathrm{d}x} + C_1 T_{\rm f}(x) = C_2 \tag{31}
$$

Here, the constants  $C_1$  and  $C_2$  can be found using Eqs. ([32\)](#page-5-1) and [\(33\)](#page-5-2) [[23\]](#page-14-16)

$$
C_1 = \frac{\pi d_{\text{riser,HE}} h_{\text{nf,HE}}}{\dot{m}_{\text{riser,HE}} C_{\text{p,nf}}}
$$
(32)

$$
C_2 = \frac{\pi d_{\text{riser,HE}} \left( h_{\text{nf,HE}} T_{\text{w}} + \frac{\alpha'_{\text{HE}} I_s(t)}{2} \right)}{\dot{m}_{\text{riser,HE}} C_{\text{p,nf}}}
$$
(33)

Here,  $\dot{m}_{\text{riser,HE}}$  and  $d_{\text{riser,HE}}$  are the nanofluid mass flow rate and heat exchanger riser diameter, respectively.  $C_{p,nf}$ is the specifc heat capacity of nanofuid. The heated nanofuid exchanges heat with the water and the basin liner of the solar still. Temperature of the nanofuid exiting the heat exchanger,  $T_{\text{o HE}}$ , can be determined from Eq. [\(34\)](#page-5-3)

$$
T_{\text{o,HE}} = \left(T_{\text{i,HE}} - \frac{C_2}{C_1}\right) e^{-AL_{\text{riser,HE}}} + \frac{C_2}{C_1}
$$
 (34)

Here  $T_{\text{i,HE}}$  is the temperature of heated nanofluid entering the heat exchanger. The length of each riser of the heat exchanger is marked by  $L_{\text{riser,HE}}$ . The coefficient of heat transfer of nanofluid  $(h_{nf,HE})$  flowing to the heat exchanger and the Nusselt number for this fowing fuid can be computed from Eqs. [\(35\)](#page-5-4) and ([36\)](#page-5-5) [[23](#page-14-16), [29](#page-14-23)]

$$
h_{\text{nf,HE}} = \frac{\text{Nu}_{\text{nf,HE}} k_{\text{nf}}}{d_{\text{rise,r,HE}}}
$$
(35)

$$
Nu_{nf,HE} = 1.86 \left(Re_{nf,HE}Pr_{nf,HE} \frac{d_{riser,HE}}{L_{riser,HE}}\right)^{1/3} \left(\frac{\mu_{nf}}{\mu_{w}}\right)^{0.14} \tag{36}
$$

Reynolds and Prandtl numbers are two important parameters to compute the heat transfer coefficient and the Nusselt number. These two parameters can be found out by Eqs.  $(37)$  $(37)$  $(37)$  and  $(38)$  $(38)$  as shown below

$$
\text{Re}_{\text{nf,HE}} = \frac{4\dot{m}_{\text{riser,HE}}}{\pi d_{\text{riser,HE}} \mu_{\text{nf}}} \tag{37}
$$

$$
Pr_{nf,HE} = \frac{\mu_{nf} C_{p,nf}}{k_{nf}}
$$
\n(38)

Here, Re, Pr, Cp,  $K$  and  $\mu$  represent the Reynolds number, Prandtl number, heat capacity, thermal conductivity and viscosity of the nanofuid fowing through the heat exchanger. Subscripts 'nf' and 'HE' denote Nanofuid and Heat Exchanger, respectively.

<span id="page-5-0"></span>**Production of distilled water** The hourly productivity of pure water  $m_{\text{pw}}$  in kg m<sup>-2</sup> h<sup>-1</sup> unit can be expressed as Eq. [\(39\)](#page-5-8) [[23](#page-14-16)]

<span id="page-5-8"></span>
$$
m_{\rm pw} = \frac{3600h_{\rm ev,w-gi}(T_{\rm w} - T_{\rm gi})}{h_{\rm fg}}\tag{39}
$$

<span id="page-5-1"></span>Here,  $h_{\text{fg}}$  is the latent heat of vaporization of water at temperature of  $T_w$  which can be estimated using Eq. ([40](#page-5-9)). Total amount of pure water for a day is nothing but the sum of the hourly amounts of pure water

<span id="page-5-2"></span>
$$
h_{\rm fg} = 3161.5 - (2.40741T_{\rm w})\tag{40}
$$

$$
M_{\rm pw} = \sum m_{\rm pw} \tag{41}
$$

<span id="page-5-9"></span>Here  $M_{\text{pw}}$  is the total amount of distilled water for a day.

<span id="page-5-3"></span>**First law efficiency** The thermal efficiency  $(\eta_{\text{th,ss}})$  of solar still is nothing but the ratio of output energy and input energy. In this case, output is the amount of fresh water. So, the output energy is obviously the latent heat given off to produce the fresh water. The thermal efficiency can be estimated either by Eq.  $(42)$  or  $(43)$  $(43)$   $[28]$ 

<span id="page-5-10"></span>
$$
\eta_{\text{th,ss}} = \frac{\sum m_{\text{pw}} x h_{\text{fg}}}{\sum I_{\text{s}}(t) x A_{\text{b}} x \Delta t + Q_{\text{hs}} x \Delta t} \tag{42}
$$

<span id="page-5-4"></span>Here,  $Q_{\text{hs}}$  is the power of heat source to heat the nanofuid before entering the heat exchanger. This heat source can be solar energy too. In that case, the heat source  $Q_{\text{hs}}$ would be replaced by solar collectors to heat the nanofuid. Then, Eq.  $(42)$  can be rewritten as Eq.  $(43)$  to get the thermal efficiency

<span id="page-5-11"></span><span id="page-5-5"></span>
$$
\eta_{\text{th,ss}} = \frac{\sum m_{\text{pw}} x h_{\text{fg}}}{\sum I_{\text{s}}(t) x A_{\text{b}} x \Delta t + \sum I_{\text{sc}}(t) x A_{\text{c}} x \Delta t}
$$
(43)

 $I_s$  and  $I_{sc}$  means the irradiation on the solar still and solar collector, respectively.  $A_b$  and  $A_c$  are the areas of the basin and solar collector, respectively.

#### <span id="page-5-6"></span>**Second law analysis**

<span id="page-5-7"></span>Exergy efficiency  $(\eta_{Ex,ss})$  is the ratio of output exergy to input exergy as depicted in Eq. ([44\)](#page-6-0) [\[30\]](#page-14-24). Temperatures of the ambient, the water, the sun and the heat source must be known to compute the exergy efficiency

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$$
\eta_{\text{Ex,ss}} = \frac{\dot{E} x_{\text{output}}}{\dot{E} x_{\text{input}}}
$$
(44)

The output exergy is nothing but the evaporative exergy that can be found from Eq.  $(45)$ 

$$
\dot{E} x_{\text{evap}} = \dot{E} x_{\text{output}} = A_{\text{b}} h_{\text{ev,w-g,i}} (T_{\text{w}} - T_{\text{g,i}}) \left( 1 - \frac{T_{\text{a}}}{T_{\text{w}}} \right) (45)
$$

The input exergy  $(\dot{E} x_{input})$  is formed of the combination of solar still and heat source as shown in Eq. ([46](#page-6-2)). The exergy input for solar still is dependent on the incident solar radiation. The power of heat source constitutes the exergy input for heat source

$$
\dot{E}x_{input} = \dot{E}x_{ss} + \dot{E}x_{hs}
$$
 (46)

$$
\dot{E} x_{ss} = A_b I_s(t) \left[ 1 - \frac{4T_a}{3T_{sun}} + \frac{1}{3} x \left( \frac{T_a}{T_{sun}} \right)^4 \right]
$$
(47)

$$
\dot{E} x_{hs} = Q_{hs} \left[ 1 - \frac{4T_a}{3T_{hs}} + \frac{1}{3} x \left( \frac{T_a}{T_{hs}} \right)^4 \right]
$$
 (48)

'Ex' stands for exergy. The subscripts 'ss' and 'hs' mean solar still and heat source, respectively.  $T_a$ ,  $T_{\text{sun}}$  and  $T_{\text{hs}}$  are the temperatures of the atmosphere, the sun and heat source, respectively. If solar collector is used to heat the nanofuid, then  $Q_{\text{hs}}$  would be replaced by  $A_c$  I<sub>s</sub>(*t*) and  $T_{\text{hs}}$  is changed to  $T_{\text{sun}}$  as written in Eq. ([47\)](#page-6-3) to get the input exergy from the heat source.

<span id="page-6-0"></span>
$$
k_{\rm hnf} = k_{\rm bf} \left[ \frac{\frac{(\varphi_{\rm np1} k_{\rm np1} + \varphi_{\rm np2} k_{\rm np2})}{\varphi_{\rm tot}} + 2k_{\rm bf} + 2(\varphi_{\rm np1} k_{\rm np1} + \varphi_{\rm np2} k_{\rm np2}) - 2\varphi_{\rm tot} k_{\rm bf}}{\frac{(\varphi_{\rm np1} k_{\rm np1} + \varphi_{\rm np2} k_{\rm np2})}{\varphi_{\rm tot}} + 2k_{\rm bf} - (\varphi_{\rm np1} k_{\rm np1} + \varphi_{\rm np2} k_{\rm np2}) + \varphi_{\rm tot} k_{\rm bf}} \right] \tag{50}
$$

<span id="page-6-6"></span><span id="page-6-5"></span>
$$
\varphi_{\text{tot}} = \varphi_{\text{np1}} + \varphi_{\text{np2}} \tag{51}
$$

<span id="page-6-1"></span>The Viscosity of any conventional and hybrid nanofuid can be found with the help of Brinkman model as illustrated in Eq. ([52](#page-6-7)) and [\(53](#page-6-8)) below [\[31\]](#page-14-25)

<span id="page-6-7"></span>
$$
\mu_{\rm nf} = \frac{\mu_{\rm bf}}{(1 - \varphi)^{2.5}}\tag{52}
$$

<span id="page-6-8"></span>
$$
\mu_{\rm hnf} = \frac{\mu_{\rm bf}}{\left[1 - \left(\varphi_{\rm np1} + \varphi_{\rm np2}\right)\right]^{2.5}}
$$
(53)

<span id="page-6-3"></span><span id="page-6-2"></span>The specific heat capacity  $(C_p)$  and density  $(\rho)$  of conventional nanofuid depend on the concentration of the nanoparticle and on the base fluid. The determination of  $C_p$  and  $\rho$  can be accomplished according to Eqs. ([54\)](#page-6-9) and [\(55](#page-6-10)) as depicted below [\[32](#page-14-26)]

<span id="page-6-9"></span>
$$
\rho_{\rm nf} = \varphi_{\rm np} \rho_{\rm np} + \left(1 - \varphi_{\rm np}\right) \rho_{\rm bf} \tag{54}
$$

<span id="page-6-10"></span>
$$
\rho_{\rm nf} C_{\rm p, nf} = \varphi_{\rm np} \rho_{\rm np} C_{\rm p, np} + \left(1 - \varphi_{\rm np}\right) \rho_{\rm bf} C_{\rm p, bf} \tag{55}
$$

The specific heat capacity  $(C_p)$  and density  $(\rho)$  of any hybrid nanofuid depend on the concentration of individual nanoparticles. Analytical expression of  $C_p$  and  $\rho$  can be represented as in Eqs.  $(56)$  $(56)$  $(56)$  and  $(57)$  $(57)$  shown below  $[33]$  $[33]$  $[33]$ 

<span id="page-6-12"></span><span id="page-6-11"></span>
$$
\rho_{\rm hnf} = \varphi_{\rm np1} \rho_{\rm np1} + \varphi_{\rm np2} \rho_{\rm np2} + \left(1 - \varphi_{\rm np1} - \varphi_{\rm np2}\right) \rho_{\rm bf} \tag{56}
$$

$$
\rho_{\rm hnf} C_{\rm p, hnf} = \varphi_{\rm np1} \rho_{\rm np1} C_{\rm p, np1} + \varphi_{\rm np2} \rho_{\rm np2} C_{\rm p, np2} + \left(1 - \varphi_{\rm np1} - \varphi_{\rm np2}\right) \rho_{\rm bf} C_{\rm p, bf} \tag{57}
$$

## **Numerical solution of the model**

The solution for the present model has been carried out numerically by using EES software. In this regard, the nanofuid properties were evaluated frst. For a conventional nanofuid with a single nanoparticle in a single base fuid, thermal conductivity can be found following Eq. [\(49](#page-6-4)). Thermal conductivity of a hybrid nanofuid with two nanoparticles in a single base fuid can be determined according to the Maxwell model as shown in Eq.  $(50)$  $(50)$  and  $(51)$  $(51)$  below  $[31]$ 

$$
k_{\rm nf} = k_{\rm bf} \frac{k_{\rm p} + 2k_{\rm bf} - 2\varphi (k_{\rm bf} - k_{\rm p})}{k_{\rm p} + 2k_{\rm bf} + \varphi (k_{\rm bf} - k_{\rm p})}
$$
(49)

<span id="page-6-4"></span>Then, the useful heat gain from the nanofuid to the solar still was evaluated by using the temperatures of nanofuid flowing in and out of heat exchanger. Thus, the temperature of the basin water was determined for each time step. At the same time, the relevant heat transfer coefficients were evaluated using the empirical equations given above in Sect. [2.2.](#page-2-2) Then, the amount of the distillate production was calculated using the heat transfer and thermodynamics relations. Consequently, the first law and the second law efficiencies were evaluated. The fowchart for the numerical solution is shown in Fig. [4.](#page-7-0)

<span id="page-7-0"></span>



<span id="page-7-1"></span>**Fig. 5** Solar radiation and ambient temperature of a typical summer day in Dhahran, Kingdom of Saudi Arabia [\[34\]](#page-15-0)

<span id="page-7-2"></span>



# **Experimental data**

In the present work, we used the solar radiation, temperature of surrounding atmosphere and speed of wind of a typical summer day at Dhahran, Saudi Arabia as input data for the solar still performance evaluation. Dhahran is situated at 17 meters above the mean sea level in the East of Saudi Arabia. The geographical coordinated of the site are 26.2° N and 50.0° E, respectively. The average

<span id="page-7-3"></span>**Table 2** Geometrical dimensions of the solar still

Parameters	Description	Value
$L_{ss}$	Length of solar still	1 <sub>m</sub>
$W_{ss}$	Width of solar still	$0.85$ m
$d_{\text{head}}$	Diameter of header of heat exchanger	3.4 cm
dt	Diameter of riser tube of heat exchanger	$1.7 \text{ cm}$
$L_{\text{riser}}$	Length of a riser tube	$0.77 \text{ m}$

hourly data from 9 am to 3 pm were used in the simulations. Average wind speed is taken as  $8 \text{ m s}^{-1}$ . The solar radiation and the ambient temperature for a typical summer day at Dhahran are shown in Fig. [5](#page-7-1). Table [1](#page-7-2) expresses the hourly average solar radiation and temperature of surrounding atmosphere of Dhahran.

The overall geometric dimensions of the solar still are  $1 \text{ m} \times 0.85 \text{ m}$ . Table [2](#page-7-3) below describes the detail geometrical dimensions of the solar still.

# **Result analysis and discussion**

The mathematical model described above is used to assess the characteristics of the solar still equipped with heat exchanger. The performance characteristics presented in this work include the distilled water yield, thermal and exergy efficiencies. Mass flow rate and temperature of nanofuid at inlet of the heat exchanger and depth of basin water were some of the key parameters considered in the current study. The primary objective of the present study is to illustrate the impacts of hybrid nanofuid as heat transfer fuid on the performance characteristics of the solar still. In this regard, the importance of the heat exchanger is illustrated first. Then, the effect of using conventional nanofuid fowing through the heat exchanger is explored. Finally, replacing the conventional nanofuid with the hybrid nanofuid is investigated to show the possible impact of the hybrid nanofuid on the performance characteristics of solar still.

<span id="page-8-0"></span>

Supplemental heating of the basin water using a heat exchanger raises the water temperature, thus enhances the evaporation rate and thereby increases the yield. Figure [6](#page-8-0) provides a comparison of the features of solar still with and without a heat exchanger. Performance characteristics of the conventional solar still (CSS) were compared with the solar still equipped with heat exchanger. Three diferent inlet temperatures of hot water as heat transfer fuid were considered. The heat exchanger inlet temperatures of the heat transfer fuid were set to be 50, 60, and 70 °C, respectively, as depicted in Fig. [6](#page-8-0). The depth of the basin water was fixed at 6 cm. Mass flow rate of 0.01 kg  $s^{-1}$ 

and 0.016 kg s<sup> $-1$ </sup> were considered. Figure [6a](#page-8-0), b shows the daily yield of distilled water for the conventional solar still and the still equipped with heat exchanger. Incorporating heat exchanger enhances the daily yield except in the case of low inlet temperature of 50  $\degree$ C and low flow rate of 0.01 kg s<sup> $-1$ </sup> for which the yield is below the CSS. However, the daily yield improves with the rise in fow rate of hot water as seen in Fig. [6](#page-8-0)b. The daily yield is found to be nearly four times higher than that of CSS with inlet temperature of hot water ( $T_{\text{iHE}}$ ) of 70 °C and high flow rate  $(m)$  of 0.016 kg s<sup>-1</sup>. Thermal efficiency (Fig. [6c](#page-8-0), d) and the exergy efficiency (Fig.  $6e$  $6e$ , f) also show similar trend



<span id="page-9-0"></span>**Fig. 7** Comparison of solar still performance between hot water and conventional nanofuid fowing through the heat exchanger

as the daily yield. Both the energy and exergy efficiencies increase considerably when the heat transfer fuid enters into the heat exchanger with a very high temperature and flow rate. The increase in the exergy efficiency is almost sevenfold. Since high inlet fuid temperature increases the performance of the solar still, the inlet temperature

<span id="page-10-0"></span>**Table 3** Thermophysical properties of nanoparticles used in the analysis

$\text{Al}_2\text{O}_3$	CuO	SiO <sub>2</sub>
3970	6400	2220
765	531	1000
40	76.5	1.38

of heat transfer fuid was kept constant at 70 °C in further analysis.

Use of nanofuids improves the heat transfer characteristics in thermal systems. Compared to hot water, the infuence of water– $A1_2O_3$  nanofluid on the performance of the system is visualized in Fig. [7.](#page-9-0) The production of pure distilled water increases when water– $Al_2O_3$  nanofluid is used (Fig. [7b](#page-9-0)) instead of hot water (Fig. [7a](#page-9-0)) as heat transfer fuid. The yield increases with the rise of mass fow rate of the nanofuid entering the heat exchanger. It should be noted that the yield remains always high when the depth of water is low (i.e. 6 cm). It is also interesting to note that the increase of the daily pure water yield is considerably higher when the mass fow rate of the fuid (water or nanofuid) inside the heat exchanger increases. This is because of additional heat transfer and further elevation of basin water temperature that results in more evaporation. The energy and exergy efficien-cies (Fig. [7c](#page-9-0)–f) increase when using water– $Al_2O_3$  nanofluid in the heat exchanger, especially with higher flow rate and lower basin water depth. It should be noted that the enhancement in the exergy efficiency is more pronounced.

The enhancement in heat transfer characteristics that result in the increase of daily pure desalinated water by using nanofuids is due to the thermophysical properties of the nanoparticles such as the specifc heat capacity and the thermal conductivity (see Table [3](#page-10-0)). A comparison of performance parameters when using two diferent nanofuids, namely water– $Al_2O_3$  and water–CuO, is presented in Fig. [8.](#page-11-0) The base case (i.e. the case of hot water as the heat transfer fuid) is also added for comparison. Here, nanofuids water– $Al_2O_3$  and water–CuO are depicted as NF1 and NF2, respectively. The concentrations for both nanofuids are taken as 1%.

 $NF1 = > water - Al<sub>2</sub>O<sub>3</sub> (1\% Al<sub>2</sub>O<sub>3</sub>)$  $NF2 \Rightarrow water-CuO (1\% CuO)$ 

In Fig. [8,](#page-11-0) the mass flow rate of the fluid is fixed at 0.016 kg s<sup>-1</sup>, the depth of the basin water is kept at 6 cm and two cases of heat exchanger inlet temperatures (50 °C and 70 °C) are considered. The pure desalinated water yield is considerably higher when the heat exchanger inlet temperature is higher for all cases of heat transfer fuids (Fig. [8a](#page-11-0), b). Both nanofuids produced higher daily yield of pure desalinated water than hot water at  $T_{\text{HF}}$  of both 50 °C and 70 °C. The daily pure desalinated water yield with heat exchanger inlet temperature of 50 °C is 1.315 kg m<sup>-2</sup> day<sup>-1</sup> for both NF1 and NF2 nanofuid. When the heat exchanger inlet temperature is 70 °C, NF1 yields slightly more desalinated water than NF2. In this case, the daily yield of pure water increases to 4.89 and 4.87 kg m<sup>-2</sup> day<sup>-1</sup> for NF1 and NF2, respectively. However, NF2 results in slightly higher thermal efficiency than NF1 (Fig.  $8c$ , d). The thermal efficiency for the case of heat exchanger inlet temperature of 70 °C is 37.3% and 37.5% for NF1 and NF2, respectively. The exergy efficiency shows a similar trend as the daily yield (Fig. [8](#page-11-0)e, f). The exergy efficiency at  $T_{\text{iHE}}$  of 70 °C is about 0.8% for both NF1 and NF2.

Hybrid nanofluids offer the possibility of tradeoff between advantages and disadvantages of diferent nanoparticles. Both the specifc heat capacity and the thermal conductivity of nanoparticles are desired to be high for enhancing heat transfer in solar still. Accordingly, in addition to the alumina  $(Al_2O_3)$  nanoparticles, two CuO and SiO<sub>2</sub> nanoparticles are considered. As compared with the alumina nanoparticles, CuO has a higher thermal conductivity, and  $SiO<sub>2</sub>$  has higher specific heat capacity (Table [3](#page-10-0)). Thus, we design two hybrid nanofluids by adding CuO and  $SiO<sub>2</sub>$  nanoparticles, respectively, to the water– $Al_2O_3$  nanofluid as follows:

 $HNF1$  => Water- $Al_2O_3$ -CuO $(0.1\% Al_2O_3$  and  $0.1\%$ CuO) HNF2 => Water-Al<sub>2</sub>O<sub>3</sub> -SiO<sub>2</sub>(0.1% Al<sub>2</sub>O<sub>3</sub> and 0.1% SiO<sub>2</sub>)

Performance characteristics of the solar still with the hybrid nanofuids HNF1 and HNF2 are shown in Fig. [9.](#page-12-0) The daily pure desalinated water yield (Fig. [9](#page-12-0)a, b), the thermal efficiency (Fig.  $9c$ , d) and the exergy efficiency (Fig.  $9e$ , f) improve signifcantly as the hybrid nanofuid fow rate increases. Improvement of these performance characteristics are more pronounced for basin water depth of 6 cm. To see the infuence of hybrid nanofuids on the performance, a comparison of the performance characteristics is given in Fig. [10](#page-13-2). In this particular scenario, inlet temperature and the rate of the nanofluid flow were fixed at 70  $^{\circ}$ C, 0.016 kg s−1, respectively. Basin water depth was kept at 6 cm. The performance comparison of solar still with hot water, NF1, NF2, HNF1, and HNF2 is presented in Fig. [10.](#page-13-2) The daily pure desalinated water yield is maximum with the hybrid nanofuid HNF2 as seen in Fig. [10](#page-13-2)a. Pure desalinated water production in this case is 4.991 kg m<sup>-2</sup> day<sup>-1</sup>. Thus, the enhancement in the daily yield of pure desalinated water by using the hybrid nanofluid HNF2 is 298.3% compared to CSS. Hybrid nanofuid HNF2 also provides better performance compared to hot water and other conventional nanofluids (NF1 and NF2). The thermal efficiency is maximum (37.76%) when using hybrid nanofuid HNF2 as shown in Fig. [10](#page-13-2)b. Hybrid nanofluid HNF1 yields slightly



<span id="page-11-0"></span>**Fig. 8** Comparison of solar still performance with water and conventional nanofuids

higher thermal efficiency than the conventional nanofluids NF1 and NF2. NF2 performed better than NF1 in this case. The superiority of the hybrid nanofuids over the other conventional nanofuids can be clearly seen in Fig. [10](#page-13-2)c when comparing the exergy efficiencies. Again, the hybrid nanofluid HNF2 showed the highest exergy efficiency  $(0.82\%)$ .



<span id="page-12-0"></span>**Fig. 9** Performance of solar still with hybrid nanofuids fowing through the heat exchanger

This is followed by the hybrid nanofuid HNF1. The two conventional nanofuids NF1 and NF2 exhibit lower exergy efficiency than hybrid nanofluids but much larger exergy efficiency than hot water.



<span id="page-13-2"></span>**Fig. 10** Performance comparison of solar still with water, conventional nanofuids and hybrid nanofuids fowing through heat exchanger

## **Conclusions**

The performance of solar still ftted with heat exchanger has been analyzed in this study. Hot water and conventional nanofluids were flown through heat exchanger besides hybrid nanofuid to make performance comparison. The mathematical modeling of the modifed solar still system is performed using the thermodynamic and heat transfer principles. The performance characteristics such as daily yield of pure distilled water, thermal and the exergy efficiencies were analyzed. First, the performance improvement using a heat exchanger with the solar still is illustrated. Then, the enhancement of the performance using conventional nanofuids in the heat exchanger is studied. Finally, hybrid nanofuids were considered in place of conventional nanofuids and the results for the performance characteristics were compared. The following conclusions can be derived from the present work.

- Heat exchanger enhances the performance of the solar still by improving the pure desalinated water production and efficiency. The higher the temperature of the heat transfer fuid entering the heat exchanger, the better the performance of the solar still.
- Use of nanofuid fuid improves the performance signifcantly. Daily yield of pure desalinated water using NF1 at 70 °C heat exchanger inlet temperature is as high as 4. 89 kg m<sup>-2</sup> day<sup>-1</sup>.
- The daily pure desalinated water yield reaches its maximum of 4.991 kg m<sup>-2</sup> day<sup>-1</sup> using water–Al<sub>2</sub>O<sub>3</sub>–  $SiO<sub>2</sub>$  (HNF2) hybrid nanofluid. This corresponds to an enhancement of 298.3% compared to CSS.
- Hybrid nanofluid, HNF2 yields thermal and exergy efficiencies of 37.76% and 0.82%, respectively, making it superior to other nanofuids in this analysis.

In conclusion, it is observed that hybrid nanofluid enhances the performance of the solar still signifcantly.

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