

An efficient enhancement in thermal conductivity of water-based **hybrid nanofuid containing MWCNTs‑COOH and Ag nanoparticles: experimental study**

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Abstract

Synergistic efect of MWCNTs-COOH and Ag nanoparticle on improving thermal conductivity of hybrid nanofuid has been explained experimentally in this paper. Diferent concentrations of MWCNTs/water nanofuids (0.004, 0.008, 0.04 and 0.16 vol%) were used and mixed with (0.04 vol%) Ag nanoparticle to prepare hybrid nanofluid. TEM was employed for confrming the size of MWCNTs and Ag nanoparticles in base fuids. Furthermore, SEM and XPS were utilized to characterize the prepared hybrid nanofuid. The hybrid nanofuids' thermal conductivity was measured in varying volume fractions at 20–50 $^{\circ}$ C temperatures. As shown by the results, the ratio of thermal conductivity of hybrid nanofluids is increased in a nonlinear manner as the concentration and temperature increase. It was seen that the hybrid nanofuid's thermal conductivity having 0.04 vol% Ag nanoparticles and 0.16 vol% MWCNTs was synergistically improved by 47.3% in comparison with thermal conductivity the water base fuid. In the end, new thermal conductivity ratio correlation was suggested on the basis of the empirical data. Comparisons of correlation output and experimental thermal conductivity ratio data showed high accuracy and capability in modeling of thermal conductivity ratio data.

Keywords Hybrid nanofuid · Thermal conductivity · MWCNTs-COOH · Ag · New correlation

List of symbols

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Introduction

Coolant fuids have many applications in the modern equipment and industries. The heat carrier is commonly used as the heat transfer medium in industrial production that causes heat transfer in a uniform manner. Among a large number of heat carriers, water is mostly employed in the heat recovery system [[1\]](#page-10-0) and solar collector feld [[2\]](#page-10-1). Despite various advantages of water like low viscosity and high heat transfer efficiency, it has a great the biggest flaw, which is its slow heating and thermal inertia; thus, it is not able to satisfy the requirement of modern industry. When the water thermal conductivity is improved, the heating speed will rise. In addition, the cooling time is decreased, reducing the volume of heat transfer device and amount used working fuid. Up to now, the work "nanofuid" was originally invented by Choi [\[3](#page-10-2)] in 1995 in Argonne National Laboratory, USA, as a novel kind of fuid by swinging copper nanoparticles in water to improve their thermo-physical property. After that, various types of nanofuids were developed by diferent researchers, and heat transfer behavior of these nanofuids was determined aiming at using the nanofuids as a new generation of fuids, which show auspicious potential considering cooling and heat transfer [\[4](#page-10-3)[–9](#page-10-4)]. Generally, nanoparticles are aluminum oxide (AI_2O_3) [[10,](#page-10-5) [11\]](#page-10-6), iron (III) oxide (Fe₃O₄) [[12,](#page-10-7) [13](#page-10-8)] and copper oxide (CuO) [\[14](#page-10-9), [15\]](#page-10-10). Out of all the physical characteristics of the nanofuids, thermal conductivity is considered as the main thermo-physical characteristics, with a signifcant impact on developing very efective heat transfer devices like industrial heat exchangers [\[16](#page-10-11)], automobile radiators [[17\]](#page-11-0), solar heat exchanger [\[18](#page-11-1)], energy systems, solar collectors [\[19](#page-11-2), [20\]](#page-11-3) and solar power plants [\[21\]](#page-11-4). Zeng and Xuan $[22]$ investigated the high light absorption efficiency and enhanced heat transfer performance of nanofuids being composed of $SiO₂/Ag$ plasmonic nanoparticles and multi-wall carbon nanotubes (MWCNTs) used for volumetric solar collectors. In this scenario, the functional fuids called nanofuids are a group of heat transfer fuids that contain thermally conductive nanometer-sized particles (range 100 nm). These particles are spread in the fuid medium. Adding very small amounts of nanoparticles in water, oils and other liquids improves optical, magnetic and electronic properties [[23](#page-11-6)]. Unique characteristics are observed for carbon nanotubes among one-dimensional nanostructures. Some of these notable properties include chemical stability, physical strength, high thermal and electrical conductivity, and mechanical resistance. Since early 1990s, many researchers have been attracted by carbon nanotubes through these properties. Since the discovery of carbonaceous materials with high thermal conductivity like graphene, carbon nanotubes (CNT) and graphene oxide (GO), many research works have been allocated to carbonaceous materials-based nanofluids $[24-27]$ $[24-27]$ $[24-27]$. A new class of hybrid nanomaterials was developed by development of carbon nanotubes, which include a combination of carbon nanotubes with metallic, semi-conductive or non-conductive nanoparticles. Single material lacks all demanded properties and characteristics. Hence, trade-off between rheological and thermal characteristics is required. In this case, using hybrid nanofuids is easy since they show improved thermal conductivity due to synergetic impact [[28–](#page-11-9)[31](#page-11-10)]. Carbon nanotubes (CNTs) and multi-walled carbon nanotubes (MWCNTs) are among the widely used nanoparticles $[32-35]$ $[32-35]$. In addition, they are used for constructing hybrid nanofuids. As the advantages of hybrid nanofuids have been recognized, thermal researchers and scientists have started investigating thermal conductivity properties and synthesis of these materials. Turcu et al. [[36\]](#page-11-13) originally prepared a hybrid nanostructure by adding $Fe₃O₄$ nanoparticles on MWCNTs using polypyrrole polymerization. Jha and Ramaprabhu [[37\]](#page-11-14) difused Au-MWNTs, Pd-MWNTs and Ag-MWNTs in de-ionized water and ethylene glycol. Ag-MWNTs showed a better enhancement of thermal conductivity, i.e., 11.30% and 37% and with deionized water and ethylene glycol, at a volume fraction of 0.03% as given by Rostamian et al. [[38\]](#page-11-15); the nanofluid thermal conductivity was enhanced up to 36% using 0.75 vol% CuO-SWCNTs hybrid material. According to Megatif et al. [\[39](#page-11-16)], thermal conductivity of CNT nanofluids was more than $CNT/TiO₂$ hybrid nanofluid. Nine et al. [[40](#page-11-17)] investigated impacts of nanoparticle shape, non-grounded and grounded MWCNTs on $(A₁, O₃$ -MWCNTs/water) hybrid nanofluid's thermal conductivity. Raising ratio of MWCNTs increased thermal conductivity in hybrid nanofuid. The nanoparticles with cylindrical shape had higher enhancement in thermal conductivity in comparison with the particles with spherical shape. It was observed that non-grounded MWCNTs gained higher improvement in thermal conductivity compared to grounded MWCNTs. Shahsavar et al. [[41](#page-11-18)] indicated an impact of sonication on thermal conductivity of $(CNTs + Fe₃O₄/DI$ water) hybrid nanofluid. The maximum concentration of CNTs $(1.535%) + Fe₃O₄ (2.428%)$ with sonication time of 5 min and 25–55 °C temperature range provided the maximum improvement in thermal conductivity of $26.07-34.26\%$. Chen et al. $[42]$ $[42]$ evaluated effective thermal conductivity of $Fe₂O₃$ -MWNT hybrid nanofluid and indicated that the hybrid fllers have a synergistic impact on heat conductive networks. They showed 28% improvement for hybrid nanofuid's thermal conductivity that is higher than thermal conductivity of mono nanofuids. Zadkhast et al. yielded the thermal conductivity of hybrid nanofuids, composed by MWCNT-CuO nanoparticles, and dispersed in the water base fuid at diferent nanoparticle concentration and temperature. According to their results, the maximum thermal conductivity was increased by 30.38% at 50 °C and φ = 0.6% [[43](#page-11-20)]. Farbod and Ahangarpour [\[44\]](#page-11-21) studied the thermal conductivity properties nanofuids with the water base, which contained Ag-decorated MWCNTs. They developed some samples with functionalized and pristine MWC-NTs and varying mass ratios of Ag/MWCNT, refux times, diferent Ag/MWCNTs concentrations and Ag- undecorated and decorated MWCNTs. A considerable enhancement was seen in the thermal conductivity in the MWCNT nanofuids having Ag-decorated MWCNTs in comparison with thermal conductivity of the undecorated samples. Munkhbayar et al. [\[45](#page-11-22)] reported effect of silver nanoparticle loadings $(1 \text{ mass}\%,$ 2 mass%, and 3 mass%) on MWCNTs/water nanofuid's thermal conductivity. The results of measurement indicated that a fuid having 0.05 mass% MWCNTs–3 mass% Ag

composite presented the maximum enhancement of thermal conductivity (14.5% at 40 $^{\circ}$ C). However, there are a number of studies concerning the nanofuids with Ag nanoparticles added to CNTs. Also, the obtained results still not satisfy as expected. One of the common procedures in the last decade was modeling by curve ftting of thermal conductivity data. Thus, new correlations for thermal conductivity prediction need to be developed. In this context, for the frst time we present the results of the efects of MWCNTs-COOH loadings (0.004 vol%, 0.008 vol%, 0.04 vol% and 0.16 vol%) on thermal conductivity of Ag/water-based nanofuid and developed correlations for predicting thermal conductivity of MWCNTs-COOH/Ag/water hybrid nanofuid.

Experimental

Hybrid nanofuid preparation

MWCNTs-COOH and silver nanoparticles have been used for the preparation of hybrid nanofuid as the following reasons:

- (a) Dispersity and stability of MWCNTs-COOH decorated with silver nanoparticles are higher than unmodifed MWCNTs, and the mentioned phenomenon is very important for increasing the thermal conductivity and application as heat transfer nanofuid.
- (b) The thermal conductivity of MWCNTs-COOH decorated with silver nanoparticles is higher than MWC-NTs-COOH, and the obtained data from experiment can prove the mentioned claim. The possible reason is that silver itself is a good thermal conductor (k Ag = 419 W m⁻¹ K⁻¹), the boundary scattering losses [[46](#page-11-23)], and interfacial resistance to heat flow [[47](#page-11-24)] between Ag/MWCNTs-COOH and base fluid can be reduced. Therefore, the decoration of MWC-NTs-COOH with AgNPs offers an excellent means to recover the potential reduction in thermal conductivity arising from functionalization of MWCNTs.

The MWCNTs-COOH and Ag nanoparticles were prepared by the US Research Nanomaterials, Inc. Table [1](#page-2-0) gives the specifcations and detailed information about the nanomaterials. Difusing CNTs in a base fuid is one of the main steps when the carbon nanofuids are prepared. Because of strong van der Waals forces between carbon surfaces as well as the high aspect ratio, the CNTs dispersion in aqueous media is a challenging task. CNTs have a hydrophobic nature. Therefore, it is not possible to disperse them in water under normal conditions [\[48](#page-11-25), [49](#page-11-26)]. In two ways, it is possible to have stable nanofuids with carbon nanotubes. The frst way is using a surfactant like Arabic gum. In the second way, the carbon nanotubes can be functionalized. Undesired impacts may be observed on the thermal conductivity of the suspensions by adding a surfactant. Thus, it seems that use of functionalized carbon nanotubes is more suitable. Functionalizing MWCNTs by the use of COOH functional groups provide hydrophilic carbon nanotubes. Therefore, the stability of the prepared nanofuids is enhanced [\[50\]](#page-12-0).

In the current study, a two-step approach, which is extensively used by the researchers [\[6](#page-10-12), [51–](#page-12-1)[54\]](#page-12-2), is utilized for preparation of the MWCNTs-COOH/Ag/water hybrid nanofuid in varying concentrations using water as the base fuid. In this approach, the nanoparticles, which are individually developed, are difused in the base fuid using appropriate dispersion methods. In the frst step, a DI-water-based Ag nanofuid with a nanoparticle concentration of 0.04 vol% was made by a two-step approach because of its low cost and simplicity compared to the one-step approach. Finally, the MWCNTs-COOH/Ag/water hybrid nanofluids were developed with a fxed concentration of Ag (0.04 vol%) and MWCNT with four varying concentrations (0.004 vol%, 0.008 vol%, 0.04 vol% and 0.16 vol%). The distribution and size of nanoparticle in the base fuid have an important role in specifying the hybrid nanofuid's cooling and thermal behavior. Thus, the main problem with the two-step approach is preventing from adherence and agglomeration of nanoparticles. For this purpose, the hybrid nanofuid was mixed for 15 min by the use of Homogenizer (MICCRA D-9) and a mechanical stirrer (IKA RW20). Afterward, ultrasonic waves were presented to the nanofuid for 5 min using a power of 400 W and a frequency of 24 kHz by Ultrasonic Homogenizer (UP400S, Hielscher GmbH) in order to dismantle agglomerated particles, allowing these particles to be resolved in the hybrid nanofuid. Ice water was added to the ultrasonic bath repeatedly during ultrasonication so that the suspension temperature is not increased. A schematic preparation of water-based MWCNTs-COOH/Ag hybrid nanofuid and thermal network of nanomaterials is shown in

Fig. [1.](#page-3-0) The stabilities of the hybrid nanofuids were studied (with the naked eye) approximately 3 days after ultrasonication, and no sedimentation was observed in the samples.

Experimental sets and thermal conductivity measurement

A KD2 Pro (Decagon Devices, Inc., USA) thermal properties analyzer with a maximum error of almost 5% was employed in the current work so that the thermal conductivity of functionalized MWCNTs/Ag–water nanofuids is measured. More detailed information about KD2 Pro thermal analyzer device is listed in Table [2](#page-3-1). This analyzer provides measurement of the fuids' thermal conductivity on the basis of the transient hot wire approach [[55,](#page-12-3) [56\]](#page-12-4) using a KS-1 (stainless steel single needle) sensor. Having a diameter of 1.27 mm and a length of 60 mm, this sensor is inserted vertically into the hybrid nanofuid located in a stable temperature bath (WNE 7, Memmert GmbH+ Co.KG, Germany) with 2 Pt100 sensors Class A used to keep a constant temperature during the test with an accuracy of 0.1 °C. The transient hot wire approach is a reliable and quick method where a heat pulse is applied to a needle. In this method, the temperature response with time is monitored during and after the heat pulse at besides needle both. The temperature response nature is a followup of the thermal characteristics of the material. Prior to measuring the nanofuids' thermal conductivity, DI water as base fuid was used for calibration of the device. The measurements of the thermal conductivity were reiterated for three times. The samples' thermal conductivity was measured in the temperature range of 20–50 °C. For having an acceptable accuracy in the experiments, the needle

Table 2 Specifcations of KD2 Pro thermal analyzer device measurement (certifcate of quality assurance)

Measure- ment speed	Accuracy	Operating environment	Sensor	Serial number
1 min	$+5\%$ thermal conductiv- ity	$0-50$ °C	KS-1-00086 KP2620	

of the device should be totally immersed in the fuid, and its orientation in the sample vials should be vertical. In addition, measurement of thermal conductivity should be carried out in a completely relaxed condition with no vibration and movement in fuid environment. In order to ensure results of measurement, each sample was tested repeatedly several times at diferent temperatures, and the average values were regarded as the empirical data. It was seen that the thermal conductivity of the base fuid, which was measured experimentally, showed a highly low diference with pure water in ASHRAE Handbook [\[57](#page-12-5)] and in good agreement. Figure [2](#page-4-0) indicates the results. According to the experimental data, the thermal conductivity enhancement and the thermal conductivity ratio are expressed as follows:

Thermal conductivity ratio =
$$
\frac{k_{\text{hnf}}}{k_{\text{bf}}}
$$
 (1)

(2) Thermal conductivity enhancement (%) = $\frac{k_{\text{hnf}} - k_{\text{bf}}}{l}$ $\frac{k_{\text{bf}}}{k_{\text{bf}}} \times 100$

where k_{bnf} and k_{bf} denote the thermal conductivity of hybrid nanofuid and thermal conductivity of base fuid.

Fig. 1 Illustration of AgNPs@MWCNTs-COOH water-based nanofuid

Fig. 2 Comparison between experimental and ASHRAE handbook data for DI water at diferent temperatures

Results and discussion

Material characterization

SEM and TEM evaluation

Morphological investigation of the prepared nanohybrid material was carried out to evaluate the successful procurement of modifed MWCNTs-COOH with Ag nanoparticles (AgNPs@MWCNTs-COOH). For the aim, scanning-EM images of each nanomaterials were evaluated carefully. According to the SEM image of MWCNTs-COOH (Fig. [3a](#page-4-1)), a spaghetti-like porous reticular structure of multi-walled carbon nanotubes could be detected. Figure [3](#page-4-1)b illustrates the SEM image of modifed MWCNTs with Ag nanoparticles. As it is obvious in this fgure, the silver nanoparticles in nanometer size have been well distributed on the surface of MWCNTs-COOH. The mentioned observations from SEM image of AgNPs@MWCNTs-COOH can prove the successful decoration of multi-walled carbon nanotubes-COOH

Fig. 3 SEM images of MWCNTs-COOH (**a**) and AgNPs@MWC-NTs-COOH (**b**)

with Ag nanoparticles. TEM technique was used for further morphological investigation of the prepared nanocomposite. Figure [4a](#page-5-0), b illustrates the TEM image of multi-walled carbon nanotubes-COOH and Ag nanoparticles, respectively. Figure [4c](#page-5-0) shows the TME image of AgNPs@MWCNTs-COOH. As it is illustrated in this fgure (Fig. [4c](#page-5-0)), the surface of MWCNTs-COOH is appropriately modifed with spherical-shaped silver nanoparticles.

XPS investigation

The X-ray photoelectron spectroscopy data of decorated MWCNTs-COOH with silver NPs are demonstrated in Fig. [5](#page-5-1)a. As it is demonstrated in Fig. [5](#page-5-1)b, the observed peaks at 284.8 and 286.4 eV are assigned to C–C (carbon) of graphite and C–O, respectively. The observed additional peak at 288.9 eV is related to the C atom in COOH functional groups on the surface of MWCNTs. The presence of COOH groups on the surface of multi-walled carbon nanotubes made the material more hydrophilic, and therefore, the material has more appropriate sites for coupling with Ag nanoparticles. The observed peaks at 368.11 eV and 374.13 eV are related to the 3d5/2 and 3d3/2 of silver, respectively. The obtained 3d doublet spin-orbit splitting of silver is approximately 6 eV, and the observation can prove that silver is in the zero valence state (Ag^0) and it is in the metallic form. In the spectrum of the material, there is not observed any peaks for 3d of silver in oxide states, and the observation can prove that silver nanoparticles are stable in the atmosphere. The peak at 532.7 is related to the surface oxygen atoms of multi-walled carbon nanotubes-COOH.

Thermal conductivity study

Figure [6](#page-6-0) indicates the thermal conductivity of the resulting hybrid nanofuids with varying concentrations as a function of diferent temperatures. As shown by the results, the thermal conductivity of the hybrid nanofuid is dependent on the measured temperature and the MWCNTs/Ag material concentration. The hybrid nanofluid with a higher concentration of MWCNTs/Ag material for each measured temperature from 20 to 60 °C has higher thermal conductivity, and increasing the temperature raises the thermal conductivity for each hybrid nanofuid. It is possible to attribute an increase in thermal conductivity, based on temperature, to an increase in improvement of nanofuid molecules level in higher temperature and kinetic energy of them. Such increase shows the essential role that the temperature plays in improvement of the nanofuid's thermal conductivity. As the stability of fuid layer around the nanoparticle is greater than the overall fuid, the former shows higher thermal conductivity. Integration of this layer around the particle is such that it can be regarded as a part of nanoparticle. As the

Fig. 4 TEM images of MWCNTs-COOH (**a**), AgNPs (**b**) and AgNPs@MWCNTs-COOH (**c**)

temperature increases, movement of nanoparticle movement is increased. As the collisions between surface atom fuid molecules and nanofuid are greater, thermal conductivity is increased. Because of an increase in the temperature, thermal conductivity increases, and volume fractions can be assigned to detachment of intermolecular bonds available in the fuid layers, increased Brownian motion, and increased collisions between nanoparticles. When aggregation and volume fraction of nanoparticles are increased, clusters of nanoparticle are formed. The formation of clusters provides the chance for faster heat transfer from these solid regions compared to liquids leading to more thermal conductivity. A simultaneous rise in volume fraction and temperature, the impacts of particle clustering and Brownian motion lead to a considerable increase in thermal conductivity. As increasing temperature increases Brownian motion, increased volume fraction at higher temperatures has a more impact on increasing thermal conductivity. When the volume fraction is low, the temperature has a lower impact on the thermal conductivity. When the volume fraction is increased, the impact of an increase in temperature is intensifed because the increase in random motion amplifes the Brownian motion impact [[58,](#page-12-6) [59](#page-12-7)]. The thermal conductivity shows a nonlinear increase with volume fraction and with temperature, and it shows consistency with another literature works [[60,](#page-12-8) [61](#page-12-9)]. Figure [7](#page-6-1) indicates the amount of relative thermal conductivity of nanofuid in comparison with the base fuid. The hybrid nanofuid's thermal conductivity was enhanced at varying temperatures and solid volume fraction about 19–47%. The

Fig. 5 XPS spectra of AgNPs@MWCNTs-COOH (**a**); and high-resolution of C1*s*, O1*s* and Ag3*d* spectrum for prepared nanocomposite (**b**)

Fig. 6 Effects of φ on the thermal conductivity for different temperatures

T = 50 °C *T* = 45 °C 0.95 *T* = 40 °C *T* = 35 °C *T* = 30 °C Thermal conductivity/w m^{-1} k^{-1} 0.9 $= 25 °C$ *T* = 20 °C *T* Thermal conductivity/w 0.85 0.8 0.75 0.7 The country 0.05 0.1 0.15 0.2 Volume fractions of MWCNTs

Fig. 8 Effects of *T* on the thermal conductivity for different solid volume fraction

Fig. 9 Efects of *T* on the thermal conductivity ratio for diferent solid volume fraction

highest enhancement of hybrid nanofuids' thermal conductivity was observed at 50 °C. The hybrid nanofuids' thermal conductivity based on a volume fraction at the experimental temperature is indicated in Fig. [8.](#page-6-2) As observed in data, increasing volume fraction and temperature increased thermal conductivity. Figure [9](#page-6-3) indicates the measured improved thermal conductivity ratio of COOH-functionalized MWC-NTs/Ag–water hybrid nanofuid as a function of the solid volume fraction in varying temperatures. Increasing the volume fraction has a larger impact when the two variables (solid volume fraction and temperatures) are compared. As the volume fraction is decreased, the impact of temperature reduces. Moreover, when the volume fraction is increased,

peratures

the impact of a temperature increase on thermal conductivity of hybrid nanofuid is augmented because the Brownian motion impact amplifes with increasing temperature. It is because of higher occurrence of random motion. As the temperature rises, movement of nanoparticle increases. Thus, because of higher occurrence of collisions between fuid molecules and surface atoms, thermal conductivity is improved [[62](#page-12-10)]. Furthermore, the temperature does not have a signifcant impact at a low concentration of hybrid nanofuid, while its impact is greater at higher concentrations. For clarifcation, consider a certain volume of the hybrid nanofuid: as the solid volume fraction is increased, the amount of collisions is increased because of Brownian motion of particles. On the other hand, as the temperature is increased, the MWCNTs kinetic energy increases in the hybrid nanofuid.

Concentration/vol%	$T = 20 °C\frac{9}{6}$	$T=25^{\circ}C/%$		$T = 30^{\circ}C\frac{1}{6}$ $T = 35^{\circ}C\frac{1}{6}$ $T = 40^{\circ}C\frac{1}{6}$		$T = 45^{\circ}C\frac{1}{6}$ $T = 50^{\circ}C\frac{1}{6}$	
$Ag:0.04\% + MWCNTs-COOH:0.004\%$	18.96	19.74	20.38	20.92	21.39	21.80	22.17
$Ag:0.04\% + MWCNTs-COOH:0.008\%$	22.04	23.23	24.21	25.04	25.77	26.41	26.98
$Ag:0.04\% + MWCNTs-COOH:0.040\%$	26.51	27.95	29.14	30.14	31.02	31.80	32.50
$Ag:0.04\% + MWCNTs-COOH:0.160\%$	33.21	36.53	39.29	41.66	43.75	45.61	47.29

Table 3 Enhancement of thermal conductivity of hybrid nanofuid at diferent concentrations and temperatures

Fig. 10 Comparison of thermal conductivity for various mono and hybrid nanofuid

Thus, the number of collisions among the nanoparticles is raised. Hence, impacts of increasing the solid volume fraction are boosted by increasing the temperature [\[50](#page-12-0)]. Table [3](#page-7-0) also shows enhancement of thermal conductivity of hybrid nanofuid at diferent concentrations and temperature. The synergistically improved impact of thermal conductivity of water-based MWCNTs/Ag hybrid nanofuid between Ag/ water and MWCNTs/water nanofluids is represented in Fig. [10.](#page-7-1) This figure confirms high efficiency of our material for improving the thermal conductivity of nanofuid in comparison with MWCNTs nanofuid or Ag nanoparticles. It is because of the synergistic impact of high thermally conducting individual components like Ag nanoparticles and MWCNTs. In addition, if MWCNTs and Ag nanoparticles are added, the overall surface area is increased in the hybrid nanofuid. Therefore, the thermal conductivity of the hybrid nanofuids is improved. Hence, the hybrid materials have synergistic characteristics in compression with the single nanoparticles. Also, Table [4](#page-7-2) shows that comparison of enhancement thermal conductivity data for water-based COOH-MWCNTs/Ag hybrid nanofuid with Ag/water and MWCNTs-COOH/water nanofuid.

Proposing new correlation for calculation of the thermal conductivity ratio

Various theoretical approaches can be used for evaluating the thermal conductivity of nanofuids for anticipating and comparing with the empirical fndings. The classical models like Hamilton–Crosser and Maxwell are popular for anticipating the thermal conductivity of well-dispersed fuids having solid particles. For highly limited works of authors in the hybrid nanofuids feld, a few numbers of limited models have been proposed up to now for estimation of thermophysical characteristics of these nanofuids. On the other hand, many of the classical models do not address the hybrid nanofuids, and their feature alterations are not accurately estimated because these models do not consider the impact of the thermal interface resistance between the base fuid and hybrid material, and they are just used for spherical particles [[63](#page-12-11)]. Thus, a simple correlation with the proper precision is also suggested in the current work for predicting the results of the hybrid nanofuids thermal conductivity ratio.

Table 4 Comparison of enhancement thermal conductivity data for water-based MWCNTs-COOH/Ag hybrid nanofuid with Ag/water and MWCNTs-COOH/water nanofuid

Concentration /vol%	$T = 20^{\circ}C/\%$	$T = 25^{\circ}C/\%$	$T = 30^{\circ}C/\%$ $T = 35^{\circ}C/\%$		$T=40^{\circ}$ C/%	$T=45^{\circ}C/%$	$T = 50^{\circ}C/%$
MWCNTs-COOH:0.004%	8.58	10.41	11.93	13.23	14.37	15.38	16.30
Ag: 0.04%	17.18	17.48	17.72	17.92	18.10	18.25	18.39
$Ag: 0.04\% + MWCNTs-COOH: 0.004\%$	18.96	19.74	20.38	20.92	21.39	21.80	22.17
MWCNTs-COOH:0.160%	27.43	28.09	28.64	29.10	29.50	29.85	30.17
$Ag:0.04\% + MWCNTs-COOH:0.16\%$	33.21	36.53	39.29	41.66	43.75	45.61	47.29

Fig. 11 Comparisons between experimental data and formula outputs for thermal conductivity

As shown by the empirical data on the thermal conductivity of COOH-functionalized MWCNTs/Ag–water hybrid nanofuid with changes in the nanocomposite temperature and concentrations (two signifcant factors that infuence on thermal conductivity ratio $[64, 65]$ $[64, 65]$ $[64, 65]$ $[64, 65]$ $[64, 65]$, a new correlation was conducted for calculating thermal conductivity ratio using curve ftting approach based on Marquardt–Levenberg algo-rithm [\[66](#page-12-14)]. It provides a high accuracy with R^2 = 0.992. This correlation uses the rule of the mixture model [[67\]](#page-12-15) with the equation as follows:

$$
\frac{k_{\text{hnf}}}{k_{\text{bf}}}=aT^{\text{b}}+c\varphi^{\text{d}}+e(T\varphi)^{\text{f}}-g\tag{3}
$$

In this correlation, φ denotes the nanocomposite concentration, *T* denotes the temperature, and $k_{\text{bnf}}/k_{\text{bf}}$ denotes the thermal conductivity ratio. Additionally, the hnf and bf subscripts represent hybrid nanofuid and base fuid, respectively. In this equation, the constant values are:

$$
a = 125/44366318; b = 0/00039598240056; c = 342/1123212
$$

\n
$$
d = 0/00010161334154; e = 0/0039849271853
$$

\n
$$
f = 1/6256254647; g = 466/32447742
$$
\n(4)

Figure [11](#page-8-0) indicates the comparison between the values taken from proposed formula and the empirical data with an absolute average deviation of 0.59% and standard deviation of 0.81%. The two dashed lines shown in this fgure are the 99% prediction bounds. This comparison is used for the solid concentrations and all temperatures. It is observed that the correlation has presented adequate results, and it can be used for engineering applications with a coefficient of determination (R^2) of 0.992.

Fig. 12 Calculated margin of deviation for all data

Moreover, a parameter known as marginal deviation is developed for verifying the suggested correlation accuracy, which is given as:

Deviation margin =

\n
$$
\left[\frac{\left(\frac{k_{\text{Inf}}}{k_{\text{bf}} \right)} \right]_{\text{exp}} - \left(\frac{k_{\text{hnf}}}{k_{\text{bf}} \right)} \right] \times 100 \quad (5)
$$

where exp and corr subscripts denote the empirical results and the results predicted by the proposed formula, respectively. The accuracy of the proposing formula between experimental values and predicted data is also evaluated in Fig. [12.](#page-8-1) As seen, most points have the deviation less than 1%, which is appropriate for an empirical correlation. Figure [13](#page-9-0) displays the more accurate comparison of the experimental data with the proposed formula as a function of hybrid nanofuid concentration at varying temperatures. As their advantages, these fgures show the diference between the experimental results and model outputs in each experiment.

Uncertainty calculation

The uncertainty of the obtained data was calculated from the bias error (equipment) and precision error obtained from the deviation in the experimental results. The bias error in monitoring of thermal conductivity by KD2 Pro was \pm 5%. The bias error for the weighing scale (Adventurer, Ohaus, Nanikon, Switzerland) and temperature (Pt100 sensors Class A) was ± 0.001 g and ± 0.1 °C. The uncertainty of the experiment is obtained by the following [[68](#page-12-16)[–70](#page-12-17)]:

$$
u = \pm \sqrt{u_b^2 + u_p^2}
$$
 (6)

$$
u_{\rm b} = \pm \sqrt{\left(\frac{\Delta k}{k}\right)^2 + \left(\frac{\Delta w}{w}\right)^2 + \left(\frac{\Delta T}{T}\right)^2} \tag{7}
$$

$$
u_{\rm p} = \pm t_{\rm v,p} \rm SD \tag{8}
$$

where $U_{\rm B}$, $U_{\rm p}$, $t_{\rm v,p}$ and SD are bias error, precision or random error in measurement with $p\%$ probability, weighing function for *v* degree of freedom and *p*% probability (95%) and sample standard deviation, respectively. According to the presented defnitions, the obtained uncertainty percent for the obtained data was between 6.44% and 8.95%.

Conclusions

In conclusion, Ag-decorated MWCNTs/water hybrid nanofuid was successfully prepared via two-step method. Improvement of thermal conductivity of the hybrid nanofuid shows a synergistic impact between MWCNTs and Ag nanoparticles based on following factors: (1) highly thermal conductive MWCNTs receive and transport heat from the Ag, facilitating further heat dissipation; (2) an adequate twodimensional well-interconnected MWCNTs/Ag nanostructure establishing more efficient heat transfer networks; (3) Ag nanoparticles and MWCNTs increase the overall surface area in the hybrid nanofuid. The diagrammatic depiction for synergistic impact on heat conductive network established by MWCNTs and Ag nanoparticles is also proposed. Using experimental data, the thermal conductivity ratio showed that the highest improvement of thermal conductivity of hybrid nanofuid was 47.3%. This was observed at 50 °C for hybrid nanofuid having 0.04 vol% Ag nanoparticles and 0.16 vol% MWCNT-COOH materials. According to the analysis, it is found that nanocomposite-based hybrid nanofuids (MWCNTs/Ag) have better the thermal characteristics compared to the single-nanoparticle-based nanofuids (MWCNTs or Ag). Therefore, these hybrid nanofuids serve as the future nanofuids for heat exchange devices. As the last point, a new correlation was suggested using empirical data for anticipating the thermal conductivity ratio of MWC-NTs-COOH/Ag/water hybrid nanofuid. The highest value of deviation margin was 1%. According to the comparisons, there was a very good consistency between the empirical data and correlation outputs.

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