

Comment on "Phase analysis of iron oxides forming the red pigment layer of the ancient earthenwares excavated from the southern Korean Peninsula"

Pranaba K. Nayak1

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In a wonderful attempt, Lee et al. [[1\]](#page-1-0) characterised eight ancient pottery samples procured during the excavation from the southern Korean peninsula. They reported their fnding on various phase analyses by using several complementary techniques such as XRD, Raman and Mössbauer spectroscopy. I felt that the data collected is of rare and excellent quality, which they have subsequently attempted to analyse and interpret. Unfortunately, I would like to bring your kind attention to several errors in their analysis of Mössbauer spectroscopic data, carried out without considering the outcome of the complementary XRD and Raman spectroscopic studies. It resulted in missing mineral phase maghemite, its relative abundance in studied samples and subsequently afecting a signifcant portion of the discussion.

While we agree with the authors that it is always challenging to distinguish between hematite (α -Fe₂O₃) and maghemite (γ -Fe₂O₃) from Mössbauer spectra, most importantly, where they are associated with magnetite (Fe₃O₄), however, it is possible to distinguish them with the help of various complementary techniques such as XRD and Raman spectroscopy, similar to the present case [[2,](#page-1-1) [3\]](#page-1-2). It is evident from Fig. 4 (XRD spectra) that while both magnetite (corresponds to the solid grey line) and maghemite (corresponds to the solid blue line) are present in lower concentration, none of the phases is absent in any sample, and even in some cases, the observed peak is prominent for the latter than the former. Interestingly in Raman spectra (in Fig. 5), maghemite can be identifed with its

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 \boxtimes Pranaba K. Nayak pranaba@hotmail.com broad characteristic bands in the region $341-350$ cm⁻¹, 519–527 cm⁻¹ along with bands in the 738–743 cm⁻¹ range (strongest) [[3\]](#page-1-2). Magnetite presence is evident from its characteristic band at the range of $671-674$ cm⁻¹, also a strong maghemite band. The hematite presence is eminent from the characteristics XRD peaks (Fig. 4) and Raman band in the range of 223–229 cm−1 (strongest) with other bands in 244–252 cm⁻¹, 290–293 cm⁻¹ and 406–411 cm⁻¹ range. Hence, from XRD and Raman spectroscopy, the presence of hematite, maghemite and magnetite is confrmed in all these eight samples [\[4–](#page-1-3)[6\]](#page-1-4).

Accordingly, a ftting model should consist of two sextets corresponding to the hematite and maghemite phases, another two sextets representing tetrahedral and octahedral phases of magnetite along with a doublet attributed to ferrihydrite (total of 26 lines) as observed from Mössbauer spectra and missing from XRD due to their amorphous nature, would suffice to estimate the respective proportions of individual phases [[7–](#page-1-5)[10](#page-1-6)]. However, the authors' Mössbauer analysis ruled out the presence of maghemite, resulting in unreliable and unrelated observations. It is evident from Table 2, the contribution of hematite and magnetite is almost the same (the ratio is 1.15 ± 0.25 ; recalculated for eight samples from Table 2), which is contrary to the obtained information from complementary techniques XRD and Raman study. The missing sextet due to maghemite has signifcantly afected the analysed Mössbauer parameters.

For stoichiometric magnetite, the hyperfne parameters are well-established. At room temperature, isomer shift (relative to α -Fe) at 0.26 and 0.67 mm s⁻¹, quadrupole shift at 0.02 and 0.00 mm s^{-1} with expected hyperfine fields are 490 and 460kOe for tetrahedral and octahedral sites, respectively [[9\]](#page-1-7). For these samples, the authors observed a higher hyperfine field for the tetrahedral site $(502.6 \pm 2.9 \text{kOe})$; recalculated from Table 2) after the ftting, which is the literature reported value for maghemite [[10](#page-1-6)]. Similarly, the isomer shift obtained for the magnetite octahedral site

¹ Department of High Energy Physics, Tata Institute of Fundamental Research, Homi Bhabha Road, Colaba, Mumbai 400005, India

 $(0.45 \pm 0.03$ mm s⁻¹; recalculated for eight samples from Table 2) is at the lower side but in the range for the reported values of maghemite $[8-10]$ $[8-10]$ $[8-10]$. In this context, it is necessary to include the estimated uncertainties and the line-width of individual sub-spectra in their Mössbauer spectral analysis. It is essential to report the goodness of ft to understand the quality ft and the procedure for obtaining hyperfne parameters. Furthermore, we would like to mention that the Fe–O asymmetric bending band in the $520-240$ cm⁻¹ range and the broad-band near 1310 cm^{-1} is not detected as described by the authors [[1\]](#page-1-0).

In summary, the authors missed the maghemite phase while analysing Mössbauer spectra presented in Fig. 6, resulting in an erroneous dataset in Table 2 and its subsequent interpretation. Their presentation of Mössbauer spectrum analysis lacks the estimated uncertainties of the measurement, and also, they have failed to quantify (or to present) the line-width of individual sub-spectra. The authors must provide a modified result after considering the mineral phases present in the studied materials, which is essential, as reports on these materials are scanty.

References

1. Lee MH, Han MS, Uhm YR et al (2021) Phase analysis of iron oxides forming the red pigment layer of the ancient earthenwares excavated from the southern Korean Peninsula. J Radioanal Nucl Chem. <https://doi.org/10.1007/s10967-021-07866-x>

- 2. Chamritski I, Burns G (2005) Infrared- and Raman-active phonons of magnetite, maghemite, and hematite: a computer simulation and spectroscopic study. J Phys Chem B 109:4965–4968
- 3. Hanesch M (2009) Raman spectroscopy of iron oxides and (oxy) hydroxides at low laser power and possible applications in environmental magnetic studies. Geophys J Int 177:941–948. [https://](https://doi.org/10.1111/j.1365-246X.2009.04122.x) doi.org/10.1111/j.1365-246X.2009.04122.x
- 4. El MY et al (2011) New evidences of in situ laser irradiation effects on γ-Fe₂O₃ nanoparticles: a Raman spectroscopic study. J Raman Spectrosc 42:239–242.<https://doi.org/10.1002/jrs.2762>
- 5. de Faria DLA, Silva SV, de Oliveira MT (1997) Raman microspectroscopy of some iron oxides and oxyhydroxides. J Raman Spectrosc 28:873–878
- 6. Shebanova ON, Lazor P (2003) Raman spectroscopic study of magnetite (FeFe₂O₄): a new assignment for the vibrational spectrum. J Solid State Chem 174:424–430. [https://doi.org/10.1016/](https://doi.org/10.1016/S0022-4596(03)00294-9) [S0022-4596\(03\)00294-9](https://doi.org/10.1016/S0022-4596(03)00294-9)
- 7. Doriguetto AC, Fernandes NG, Persiano AIC, Filho EN, Greneche JM, Fabris JD (2003) Characterization of a natural magnetite. Phys Chem Miner 30:249–255
- 8. de Bakker PMA, de Grave E, Vandenberghe RE, Bowen LH (1990) Mössbauer study of small particle maghemite. Hyperfne Interact 54:493–498
- 9. Greneche JM (2013) The Contribution of 57Fe Mössbauer spectrometry to investigate magnetic nanomaterials. In: Yoshida Y, Langouche G (eds) Mössbauer spectroscopy. Springer, Berlin. https://doi.org/10.1007/978-3-642-32220-4_4
- 10. Shchetinin IV, Seleznev SV, Dorofevich IV (2021) Structure and magnetic properties of nanoparticles of magnetite obtained by mechanochemical synthesis. Metal Sci Heat Treat 63:95–100. <https://doi.org/10.1007/s11041-021-00653-1>

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