

# Radiological impact assessment of soil and groundwater of Himalayan regions in Uttarakhand, India

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#### Abstract

In this study, we present the results of measurements and radiological impact of natural radioactivity in soil and groundwater of the Himalayan region in the Uttarakhand State of India. The concentrations of primordial radionuclides ( $^{226}$ Ra,  $^{232}$ Th and  $^{40}$ K) in soil samples of the study area were determined using gamma-ray spectrometry by employing the NaI(*Tl*) detector. The concentrations of radon and uranium were also measured in potable groundwater samples using RAD7 and Inductively Coupled Plasma Mass Spectrometry (ICPMS) techniques, respectively. The average specific activities of  $^{226}$ Ra (116 Bq kg<sup>-1</sup>),  $^{232}$ Th (137 Bq kg<sup>-1</sup>) and  $^{40}$ K (735 Bq kg<sup>-1</sup>) in soil were found considerably higher than the corresponding global average values. The average concentrations of radon (35 Bq l<sup>-1</sup>) and uranium (1.3 µg l<sup>-1</sup>) in potable groundwater were found well within the safe limits recommended by the World Health Organization. The effects of natural radioactivity in soil and groundwater are discussed in terms of different risk assessment parameters and dose quantities.

Keywords Sediment radioactivity · Dissolved uranium · Radon · Himalayan region

# Introduction

Human population is continuously exposed to extra-terrestrial and terrestrial sources of radiation. Extra-terrestrial sources of radiation mainly include cosmic rays coming from earth's outer atmosphere while the terrestrial radiation are gamma rays emitted from <sup>40</sup>K and radionuclides of <sup>226</sup>Ra and <sup>232</sup>Th decay series present in soil, rocks and water. According to UNSCEAR (1982), cosmic and terrestrial radiations impart 40% and 50% of the total external radiation dose to humans, respectively [1]. The global mean value of radiation dose from the exposure to cosmic rays at sea level is 31 nGy  $h^{-1}$  [2, 3]. In India, the mean value of dose from the exposure to cosmic rays is  $32 \text{ nGy h}^{-1}$ [3, 4]. The external exposure to gamma rays emitted from <sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K is an important contributor to natural background radiation dose. These radionuclides are distributed in varying concentrations in soil and rocks. Owing to

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their highly unstable nature, these radionuclides spontaneously decay into their daughters with the emission of alpha, beta and gamma radiations. Gamma rays can penetrate the body internally and externally as well. The harmfulness of external radiation exposure depends upon the abundance of naturally occurring gamma emitting radionuclides in soil and rocks which in turn is associated with the geology of a particular area [5-7]. The contributions of  ${}^{40}$ K,  ${}^{238}$ U, and  $^{232}$ Th to global mean value (54 nGy h<sup>-1</sup>) of absorbed gamma dose rate in outdoor environment is 35%, 25% and 40%, respectively [1]. Furthermore, the distribution of <sup>226</sup>Ra and <sup>232</sup>Th in a particular region is directly associated with the levels of <sup>222</sup>Rn, <sup>220</sup>Rn and their progeny which are major contributors to natural background radiation dose received by the human beings [8]. The variation of natural radionuclides is also useful in identifying geo-chemical processes. The natural radionuclides present in the soil also migrate and transfer to biological systems like plants, trees etc. and become a potential source of internal radiation via intake of vegetables, fruits etc. [9]. Thus, a systematic study of natural radionuclides plays a key role in radiation protection, geo scientific studies and in establishing guidelines for the mitigation of these radionuclides. It is, therefore, important to carry out systematic studies to determine the levels of

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<sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K and to estimate associated health risks for human beings.

In addition to soil radioactivity, radiation exposure via drinking water route is also an important parameter in radiation protection. In general, drinking water is mainly obtained from groundwater sources, which may be contaminated by several radioactive elements. The concentrations of these elements depend on a physical, chemical and geological feature of the aquifer [10, 11]. Radon and uranium in drinking water are key radioelements from the health risk point of view. The health risk associated with radon in drinking water depends on type of water sources such as groundwater, surface water, tap water etc. If the drinking water is made available from groundwater sources, the radiation risk becomes high due to usually found high radon concentration in groundwater. Further, the levels of radon dissolved in water depend upon rock formation such as alluvial rocks, hard rocks etc. Lower levels of dissolved radon are found in alluvial rock formation as compared to hard rock formation [12]. A positive relation of radon levels in groundwater with the soil depth below ground is reported for hard rock formation of Nalgonda, Andhra Pradesh, India [13]. Radon in drinking and household water is linked with ingestion and inhalation doses to the human population, respectively [14]. Radon ingested through drinking water route is recognized as one of the possible causes of stomach cancer [15, 16]. On the other hand, radon released from household water to indoor air becomes the main contributor to indoor radon and its short lived progeny which contributes more than 50% of the natural background radiation dose to human beings [8]. The radiation exposure to indoor <sup>222</sup>Rn and its daughters on human health are indicated as the most important cause of lung cancer after smoking [17]. Furthermore, the groundwater plays a key role in the migration and redistribution of uranium in the earth's crust. A decreasing trend of uranium concentration in groundwater with soil depth has been reported in a recent study carried out at Nalgonda area of Andhra Pradesh, India [13]. Uranium in drinking water is dangerous to the human population due to its radiological and chemically toxicity. The modes of intake of uranium by the human body are inhalation, ingestion and other (industrial and occupational) exposures. Among these routes of exposure, the ingestion mode is the most important one. Uranium ingested through drinking water route is linked with harmful radiological (carcinogenic) and chemical (non-carcinogenic) effects on human health. The radiological health risk is due to the ionizing radiation of uranium isotopes whereas the chemical risk is due to chemical toxicity of uranium as heavy metal. Some of the diseases associated with the exposure to uranium in drinking water are leukemia, stomach cancer, urinary track cancer, kidney toxicity etc. [18-21]. A reference level of 30  $\mu$ g l<sup>-1</sup> has been suggested by the World Health Organization (WHO) as a limit of maximum concentration for uranium in drinking water [22]. In India, the Atomic Energy Regulatory Board (AERB), Department of Atomic Energy, Govt. of India has suggested a value of 60  $\mu$ g l<sup>-1</sup> as a limit for uranium in drinking water on radiotoxicity [23]. It is therefore, very important to carry out systematic studies on radon and uranium in drinking water sources. In Garhwal Himalayan region, a number of surveys have been carried out in the past on the occurrence of radionuclides in soil [24-27]. However, Mandakini valley has not been covered in these surveys. The radionuclide surveys in this part of Garhwal Himalaya are important due to its different geological structure. The present study has been designed and performed to investigate the levels and radiological effects of natural radioactivity in soil and groundwater of the Himalayan region in the Uttarakhand state of India. For this purpose, activity levels of <sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K were determined in soil samples collected from Mandakini valley using gamma ray spectrometry employing NaI (Tl) detector. The concentrations of radon and uranium were also measured in potable groundwater samples using RAD7 and Inductively Coupled Plasma Mass Spectrometry (ICPMS) techniques, respectively. The radiological impact of natural radioactivity in soil and potable groundwater on human health is estimated in terms of various hazard assessment quantities. The present work will be helpful in providing baseline data of natural radioactivity levels in soil and groundwater for future studies as well as the radiation protection program of the country.

# Geology of the study area

The geological map of the study area showing major formations and sampling locations is shown in Figs. 1 and 2. The investigated region comprises of Mandakini and Bhagirathi valleys of Garhwal Himalaya, India. A total of 38 soil samples were collected from Mandakini valley of Garhwal Himalaya for radionuclide analysis. The results and effects of radon and uranium measurements in water samples collected from Mandakini valley were reported in our recent studies [28, 29]. However, a total of 30 water samples collected from Bhagirathi valley were analyzed for radon and uranium measurements. The major geological formations in Garhwal Himalayan region along Mandakini valley are Berinag, Rautganga, Bhatwari and Munsiari formations. Origin of the Mandakini River is Chorabari glacier at an altitude of 3860 m above sea level. At Rudraprayag (895 m), the Mandakini River joins Alaknanda River. The investigated region is located in Vaikrita Group of Central Crystalline, which is made up of the Munsiari, Joshimath, Pandukeshwar and Pindari fromations [30]. The main rock composition in the Kedarnath region of Munsiari formation is mica shist, calc silicate lances, quartzite and phyllonites etc. [31].



Fig. 1 Geological map [33] of Mandakini valley showing soil sampling locations





The geological formation of Bhagirathi valley comprises of Lesser Himalayan Protozoic sequence and Higher Himalayan Crystalines (Bhatwari Group and Harshil Group). The Bhatwari group is composed of porphyroclastic granite gneiss, garnetiferous mica schist, amphibolite, mylonitized augen gneiss, mica schist, amphibolite, phyllonite and schist rocks [32].

# Materials and methods

#### Soil sample preparation

A total of 38 soil samples were collected from the top surface up to 15 cm below the ground. The collected samples were crushed to fine powder and dried in an oven at about 110 °C for about 24 h after removing organic material, stones, pebbles, roots and vegetation. The samples were then sieved through a sieve of 150 µm size to obtain the fine quality of samples so as to get large surface area. The samples processed in this way were sealed in Marinelli beakers and stored for a period of at least one month for establishment of secular equilibrium among <sup>226</sup>Ra, <sup>232</sup>Th and their decay products taking care to prevent <sup>222</sup>Rn escaping from the beakers. The prepared samples were analyzed for radionuclides measurements using gamma ray spectrometry (Fig. 3).

#### Measurements of ambient dose equivalent rates

The ambient dose equivalent rates  $(\mu Sv h^{-1})$  were measured at 1 m above the ground in all sampling locations using a GM tube based portable gamma survey meter (AT6130A, ATOMTEX, Australia). This instrument is capable of detecting gamma rays in the energy range of 20 keV to 3 MeV and can store 2000 measurement results in its non-volatile memory. The device is calibrated by the company and displays the results on its LCD screen. The geometric mean of

Fi p st gamma dose rate values over all sampling locations may be considered as representative value of gamma dose rate for the region.

# Radionuclide analysis using gamma ray spectrometry

The levels of <sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K in the soil samples were determined by using a gamma-ray spectrometric system manufactured by ATOMTEX (Belarus) installed at Nuclear Research Laboratory, H.N.B. Garhwal University, Tehri Garhwal. A NaI(Tl) scintillation detector of size 63 mm × 63 mm is employed in the spectrometric system for the detection of natural radionuclides. The details of measurement procedure, calibration of detector etc. are given elsewhere [24-27, 34].

### Measurements of uranium concentration in potable groundwater

The measurements of uranium in 30 potable groundwater samples were carried out using inductively coupled plasma mass spectrometry (ICPMS) technique (Perkin Elmer, model-ELAN DRC-e). ICPMS is a an analytical technique for accurate and reliable measurements of various elements. In this technique, the water samples to be analyzed are ionized with inductively coupled plasma source and then mass spectrometry is utilized for the detectection and quantification of various elements. The detailed description of sampling procedure and measurement technique is given elsewhere [28].

# Measurements of radon concentrations in potable groundwater

Radon concentration in potable groundwater was measured using a semiconductor detecor based RAD7 monitor with

RAD  $H_2O$  accessory. The semiconductor detector used in this monitor is silicon detector which converts the alpha radiation into an electrical signal. The detailed methodology for the measurement of radon in water using RAD7 monitor is given Prasad et al. [29].

#### **Risk assessment**

The radiological effects of natural radionuclides present in soil are expressed in terms of following risk assessment quantitites.

#### **Radium equivalent activity**

The distribution of <sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K is not uniform in soil or rocks. The effects of these radionuclides on human body also vary from one radionuclide to other. The combined effect of these three radionuclides on human body can be expressed in terms of various hazard assessment quantities. Radium equivalent activity (Ra<sub>eq</sub>) is one of these quantities which is expressed as follows [35]:

$$Ra_{eq} = \left(\frac{A_{Ra}}{370} + \frac{A_{Th}}{259} + \frac{A_K}{4810}\right) \times 370$$
(1)

where  $A_{Ra}$ ,  $A_{Th}$  and  $A_K$  are activities of <sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K in Bq kg<sup>-1</sup>, respectively.

### External hazard index (H<sub>ex</sub>)

The external exposure to gamma radiations emitted from  $^{226}$ Ra series elements,  $^{232}$ Th series elements and  $^{40}$ K is expressed in terms of external hazard index (H<sub>ex</sub>). The external hazard index (H<sub>ex</sub>) is widely used in radiological protection and is defined as follows [35]:

$$H_{ex} = \frac{A_{Ra}}{370} + \frac{A_{Th}}{259} + \frac{A_K}{4810}$$
(2)

where  $A_{\text{Ra}}$ ,  $A_{\text{Th}}$  and  $A_{\text{K}}$  are the activities of <sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K in Bq kg<sup>-1</sup>, respectively. The external hazard index must be less than unity for the safety purpose [35]:

# Internal hazard index (H<sub>in</sub>)

<sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K may be potential source of internal exposure to the households if the local soils and/or rocks are used as building material. The internal exposure is mainly due to inhalation of alpha radiation emmited from indoor <sup>222</sup>Rn, <sup>220</sup>Rn and their progeny. In radiological protection, the widely used unit of internal exposure is internal hazard index (H<sub>in</sub>) which is defined as follows [35]:

$$H_{ex} = \frac{A_{Ra}}{185} + \frac{A_{Th}}{259} + \frac{A_K}{4810}$$
(3)

#### Gamma index ( $I_{\gamma}$ )

The gamma index  $(I_{\gamma})$  is the measure of external exposure gamma rays emitted from building materials. It takes into account the quantities of soil and/or rocks and the ways in which they are used in construction of a building. It is defined by European Commission as follows [35]:

$$I_Y = \frac{A_{Ra}}{300} + \frac{A_{Th}}{200} + \frac{A_K}{3000}$$
(4)

The values of  $I_{\gamma} \ge 0.5$  and  $I_{\gamma} \ge 0.5$  correspond to a dose rate criteria of 0.3 mSv y<sup>-1</sup> and 1 mSv y<sup>-1</sup>, respectively. The building materials with  $I_{\gamma} \ge 1$  should be avoided for safety purpose.

#### Alpha index

The radiation exposure to inhaled  $^{222}$ Rn originated from  $^{226}$ Ra present in the building materials can be estimated in terms of alpha index (I $\alpha$ ), which is defined as follows [35]:

$$I_{\alpha} = \frac{A_{Ra}}{200}$$
(5)

The recommended maximum value of <sup>226</sup>Ra activity is 200 Bq kg<sup>-1</sup>, which gives I $\alpha$ =1. If activity concentration of <sup>226</sup>Ra in building materials exceeds 200 Bq kg<sup>-1</sup>, the radon exhaled from building materials to indoor environment can be equal to higher than 200 Bq m<sup>-3</sup> which may cause lung cancer in households. Therefore, <sup>226</sup>Ra content in building materials should not exceed 200 Bq kg<sup>-1</sup>. In other words, I $\alpha$ must lie within unity for safe use of soil or rocks as building materials.

### Air absorbed dose rate

The air absorbed dose rate {D (nGy h<sup>-1</sup>)} from the exposure to  $^{226}$ Ra,  $^{232}$ Th and  $^{40}$ K at 1 m height from the earth's surface can be calculated from activity concentrations of  $^{226}$ Ra,  $^{232}$ Th and  $^{40}$ K in soil as follows [35]:

$$D(nGy h^{-1}) = 0.462A_{Ra} + 0.604A_{Th} + 0.0417A_{K}$$
(6)

It can be considered that <sup>238</sup>U is in secular equilibrium with <sup>226</sup>Ra in the given sample of soil. However, if any disequilibrium exists between <sup>226</sup>Ra and <sup>238</sup>U, it does not affect the dose assessment significantly if one calculates the dose from <sup>226</sup>Ra instead of <sup>238</sup>U. This is due to the fact that about 98% of the dose from <sup>238</sup>U decay series is contributed by the elements of <sup>226</sup>Ra sub-series [8].

#### Annual effective dose

The annual average effective dose rates can be calculated from air absorbed dose rate using following relation given by UNSCEAR [8].

where  $0.7 \text{ Sv Gy}^{-1}$  is the factor which is used to convert absorbed dose rate (nGy h<sup>-1</sup>) to equivalent effective dose rate (mSv  $y^{-1}$ ), 0.8 and 0.2 are the indoor and outdoor occupancy factors, respectively. The multiplication factor of  $10^{-6}$  is used to convert the unit of effective dose from nSv to mSv.



**Fig. 4** Frequency distribution of radium equivalent activity (Bq  $kg^{-1}$ ) in Mandakini valley of Uttarakhand state of India

# Health risks associated with uranium in drinking water

#### **Radiological risk assessment**

The health risk associated with the ionizing radiation arising from radioactive isotopes of uranium in drinking water is known as radiological risk. It can be defined in terms of excess cancer (ECR) as follows [36]:

$$ECR = A_U \times R \tag{9}$$

where  $A_{II}$  represents activity concentration (Bq l<sup>-1</sup>) of uranium and *R* is the risk factor. Further, *R* is defined as follows:

Risk factor (R) = 
$$r \times I$$
 (10)

where r is cancer risk constant  $(1.19 \times 10^{-9})$  of uranium for mortality and I is per capita activity intake of uranium via drinking water. The value of I can be calculated by multiplying water consumption rate  $(4.05 \ 1 \ day^{-1})$  with average life (23250 days) of a person [37].

#### **Chemical risk assessment**

The health risk due to the chemical toxicity of uranium as a heavy metal is termed as chemical risk associated with uranium in drinking water. It can be defined in terms of lifetime average daily dose (LADD) as follows [38-41]:

$$LADD(\mu g kg^{-1} d^{-1}) = \frac{A_U \times IR \times ED \times EF}{AT \times BW \times 365}$$
(11)

where  $A_{II}$  is uranium activity in  $\mu g l^{-1}$ , IR is ingestion rate  $(4.05 \ 1 \ d^{-1})$ , ED is exposure duration (63.7 years i.e. 23,250 days), EF is exposure frequency (365 d  $y^{-1}$ ), BW is average body weight (70 kg) and AT is the averaging time i.e. life expectancy (63.7 years i.e. 23,250 days).

### Hazard quotient (HQ)

The extent of harm produced by the consumption of uranium via drinking water can be expressed in terms of hazard

Table 1 Descriptive statistics of activity levels primordial radionuclides in soil samples of Garhwal Himalaya, India and associated hazard assessment quantities

	Activity level Equivalent Ac	s of <sup>226</sup> Ra, <sup>232</sup> Th ctivity (Ra <sub>eq</sub> ) (B	n & <sup>40</sup> K and Radi q kg <sup>-1</sup> )	um	Hazard	lassessme	ent quantit	ies		
	<sup>228</sup> Ra	<sup>232</sup> Th	<sup>40</sup> K	Ra <sub>eq</sub>	H <sub>ex</sub>	H <sub>in</sub>	$I_{\Upsilon}$	$I_{\alpha}$		
Min	$34.2 \pm 4.43$	$28.7 \pm 9.3$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$			0.17				
Max	$229 \pm 23.4$	$295 \pm 29.5$	$1360 \pm 194$	599	1.62	2.06	2.08	1.15		
AM	97	129	541	322	0.87	1.13	1.15	0.48		
Median	92	115	478	307	0.83	1.10	1.09	0.46		
GM	87	119	365	306	0.83	1.07	1.09	0.43		
SD	45	51	389	109	0.29	0.39	0.38	0.22		



Fig. 5 Variation of health hazard indices associated with the levels of natural radionuclides in Mandakini valley of Uttarakhand state of India

quotient (HQ) which is defined as a ratio of LADD to reference dose i.e.

$$HQ = \frac{LADD}{R_f D}$$
(12)

where  $R_fD$  is the reference dose [28, 41]. The reference doses ( $R_fD$ ) recommended by the World Health Organization (WHO), the Health Canada, the United States Environmental Protection Agency (USEPA) and the Atomic Energy Regulatory Board (AERB) are given in Table 4. In this study, the value of  $R_fD$  (4.53 µg kg<sup>-1</sup> d<sup>-1</sup>) recommended by AERB was used for HQ calculation.

# Dose assessment from the exposure to uranium via drinking water

The ingestion doses associated with the consumption of uranium via drinking water can be calculated using following expression given by IAEA [28, 42, 43].



Fig. 6 Correlation between ambient dose equivalent rate measured by portable survey meter and air absorbed gamma dose rate estimated from soil radioactivity

Ingestion Dose(mSv y<sup>-1</sup>) = A<sub>U</sub> × I × 365 × F (13)

where  $A_U$ , I and F is the uranium activity in water (Bq l<sup>-1</sup>), daily water intake (l d<sup>-1</sup>) and dose conversion factor ( $4.5 \times 10^{-8} \,\mu\text{Sv} \,\text{y}^{-1}/\text{Bq} \,\text{l}^{-1}$  for <sup>238</sup>U), respectively. The daily water intake (I) for the people of different age groups are given in Table 5.

# Dose assessment from the exposure to <sup>222</sup>Rn in water

Radon in water affects the human body in two ways. First, the radon ingested through drinking water route imparts radiation dose to stomach. Second, the radon escaped from household water to indoor air can be a potential contributor to inhalation dose to lungs. Thus, stomach and lungs both are affected by high levels of radon in water.

The annual ingestion dose  $(mSv y^{-1})$  due to radon in drinking water for the people of different age groups can be estimated as follows [29]:

 
 Table 2
 Descriptive statistics of dose rate quantities measured using portable survey meter and estimated from soil radioactivity

	Ambient dose equivalent rate $(nGy h^{-1})$ measured using survey	Air absorbed gamma dose rate $(nGy h^{-1})$ estimated from soil radio-	Annual dose (m	effective Sv y <sup>-1</sup> )
	meter	activity	Indoor	Outdoor
Min	80	69	0.3	0.1
Max	260	261	1.3	0.3
AM	150	145	0.7	0.2
Median	150	139	0.7	0.2
GM	143	137	0.7	0.2
GSD	47	49	0.2	0.1

	Uranium activity (µg l <sup>-1</sup> )	Excess cancer risk	$\begin{array}{c} LADD \\ (\mu g \ kg^{-1} \ d^{-1}) \end{array}$	HQ
Min	$0.02 \pm 0$	$6.16 \times 10^{-8}$	0.001	0.0003
Max	$6.2 \pm 0.15$	$1.74 \times 10^{-5}$	0.36	0.08
AM	1.3	$3.54 \times 10^{-6}$	0.07	0.02
SD	1.6	$4.38 \times 10^{-6}$	0.09	0.02
GM	0.5	$1.36 \times 10^{-6}$	0.03	0.01

Ingestion dose 
$$(mSv y^{-1}) = A_{Rn} \times DWI \times DCF \times t$$
 (14)

where  $A_{Rn}$  is activity concentration of radon in water (Bq 1<sup>-1</sup>), DWI is daily water intake (1 d<sup>-1</sup>), DCF is dose conversion factor (10<sup>-8</sup> Sv Bq<sup>-1</sup>) and t is time of exposure (365 days y<sup>-1</sup>) [29].

The inhalation dose associated with the exposure to <sup>222</sup>Rn escaped from household water to indoor environment can be calculated as follows [29]:

Inhalation dose 
$$(mSv y^{-1}) = A_{Rn} \times \frac{R_a}{R_w} \times F \times T_{Indoor} \times 8760 \times DCF$$
(15)

where  $A_{Rn}$  is <sup>222</sup>Rn concentration (Bq m<sup>-3</sup>) in water,  $R_a/R_w = 10^{-4}$ ,  $R_a$  is <sup>222</sup>Rn concentration in air,  $R_w$  is <sup>222</sup>Rn concentrations water, F is equilibrium factor (0.4) between <sup>222</sup>Rn and progeny,  $T_{Indoor}$  is indoor occupancy factor (0.8) and DCF is dose conversion factor (9 nSv Bq<sup>-1</sup> h<sup>-1</sup> m<sup>3</sup>).

# **Results and discussion**

# Radionuclides distribution in soil and its radiological implications

The measured values of  $^{226}$ Ra,  $^{232}$ Th and  $^{40}$ K and estimated hazard assessment quantities are given in Table 1.

The measured values of <sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K concentrations ranged from  $34.2 \pm 4.43$  to  $229 \pm 23.4$  Bq kg<sup>-1</sup> with an average of 97 Bq kg<sup>-1</sup>,  $28.7 \pm 9.3$  to  $295 \pm 29.5$  Bq kg<sup>-1</sup> with an average of 129 Bq kg<sup>-1</sup> and  $15.2 \pm 3$  to  $1360 \pm 194$  Bq kg<sup>-1</sup> with an average of 541 Bq kg<sup>-1</sup>, respectively. The average values of measured <sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K concentrations were found considerably greater than the corresponding values at global level given by UNSCEAR [8]. According to UNSCEAR, global average values of <sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K in soil are 35, 30 and 400 Bq kg<sup>-1</sup>, respectively [8]. A wide variation was observed in the distribution of natural radionuclides in the area. The radiological effects of natural radionuclides



Fig. 7 Frequency distribution of uranium activity in drinking water samples of Bhagirathi valley in Garhwal Himalaya, India

 Table 4 Recommendations of various health agencies on uranium in drinking water

Uranium activity (µg l <sup>-1</sup> )	ECR	Refer- ence Dose (µg kg <sup>-1</sup> l <sup>-1</sup> )	Agency	References
30	$8.4 \times 10^{-5}$	2.26	WHO	[22]
20	$5.6 \times 10^{-5}$	1.51	Health Canada	[40]
30	$8.4 \times 10^{-5}$	2.26	USEPA	[45]
60	$1.68 \times 10^{-4}$	4.53	AERB	[23]

present in the soil of study area were determined in terms of radium equivalent activity ( $Ra_{eq}$ ), external hazard index ( $H_{ex}$ ), internal hazard index ( $H_{in}$ ), gamma index ( $\Upsilon$ ) and alpha index ( $\alpha$ ).  $Ra_{eq}$  was found to vary from 146 to 599 Bq kg<sup>-1</sup> with an average of 322 Bq kg<sup>-1</sup>. The mean value of estimated  $Ra_{eq}$  is well within the safe limit of 370 Bq kg<sup>-1</sup> [4]. Frequency distributions for radium equivalent activity is shown in Fig. 4. It is worth noticing that 10 out of 38 analyzed samples have  $Ra_{eq}$  values higher that the safe limit of 370 Bq kg<sup>-1</sup>.

The calculated values of external hazard index  $(H_{ex})$ , internal hazard index  $(H_{in})$ , gamma index  $(I_{\gamma})$  and alpha index  $(I_{\alpha})$  were found to range from 0.40 to 1.62 with an average of 0.87, 0.49 to 2.06 with an average of 1.13, 0.54 to 2.08 with an average of 1.15 and 0.17 to 1.15 with an average of 0.48, respectively. The variation of these hazard indices is shown in Fig. 5. It is clear from the figure that at significant number of locations the values of hazard indices are greater than unity which indicates the possibility of radiation hazard if the local soil and/or rocks is brought in use for construction of buildings. However, in majority of locations the values of hazard indices are well within unity (the safe limits of radionuclides exposure).

# Radiation dose rates due to natural radionuclides in soil

The descriptive statistics for measured air absorbed gamma dose rates, estimated air absorbed gamma dose rates and annual effective dose rates (indoor and outdoor) due to natural radionuclides in soil are shown in Table 2. The estimated air absorbed gamma dose rate from the exposure to <sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K in soil was found to be in the range of 69 to 261 nGy  $h^{-1}$  with a mean value of 145 nGy  $h^{-1}$ . The contributions of <sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K to average air absorbed gamma dose rate were found to be 45 nGy  $h^{-1}$ , 78 nGy h<sup>-1</sup> and 22 nGy h<sup>-1</sup>, respectively. The air absorbed gamma dose rate measured using a radiation survey meter directly at sampling locations was found to vary from 80 to 260 nGy  $h^{-1}$  with an average of 150 nGy  $h^{-1}$ . The annual effective dose rates in the indoor and outdoor environments were found to vary from 0.3 to 1.3 mSv  $y^{-1}$  with an average of 0.7 mSv  $y^{-1}$  and 0.1 to 0.3 mSv  $y^{-1}$  with an average of 0.2 mSv  $y^{-1}$ , respectively. The estimated annual effective dose rates at all locations are far less than the global average value of 2.4 mSv  $y^{-1}$  received from all natural background radiation sources [8]. According to European Commission (EC), building materials should be avoided if the excess gamma radiation originating from them exceeds the annual effective dose of an individual by  $0.3 \text{ mSv y}^{-1}$  [44].

The values of air absorbed gamma dose rate estimated using soil radioactivity was found in good agreement with the ambient dose equivalent rate values measured using portable survey meter. A correlation between the air absorbed gamma dose rates estimated from soil activity and ambient dose equivalent rates directly measured using pocket survey meter (GM detector) is shown in Fig. 6. A strong positive correlation (R = 0.99, slope = 0.97) between estimated and measured data shows a good agreement between two methods. This validates the accuracy and reliability of results.

#### Levels and effects of uranium in drinking water

The measurements of uranium concentration in drinking water samples collected from 30 groundwater sources (one sample from each source) located in Bhagirathi valley of Garhwal Himalaya, India were carried out using ICMPS. The descriptive statistics for uranium activity in drinking water samples of investigated region along with associated radiological (carcinogenic) and chemical (non-carcinogenic) health risks are shown in Table 3. The measured values of uranium concentration were found to vary from  $0.02 \pm 0$  to  $6.2 \pm 0.15 \ \mu g \ l^{-1}$  with an average of 1.3  $\mu g \ l^{-1}$ . The concentration of uranium in all drinking water samples were found well within the safe limit of 30  $\mu$ g l<sup>-1</sup> recommended by World Health Organization (WHO) and United States Environmental Protection Agency (USEPA) [22, 45]. In India, the Atomic Energy Regulatory Board (AERB), Department of Atomic Energy, Govt. of India has prescribed a limit of 60  $\mu$ g l<sup>-1</sup> for uranium in drinking water on radiotoxicity [23]. Frequency distribution of uranium concentration in the analyzed drinking water samples is shown in Fig. 7. The concentration of uranium in 22 out of 30 analyzed samples was found in the range of 0.02 to 2  $\mu$ g l<sup>-1</sup>. The calculated values of excess cancer risk (ECR) linked with the exposure to uranium in all drinking

Investigated location	State	Uranium con (µg l <sup>-1</sup> )	centration	References
		Range	Average	
Bathinda district	Punjab	0.48–571.7	84.70	[46]
Amritsar & Gurdaspur districts	Punjab	1.24-45.42	14.91	[47]
Kangra, Mandi, Kullu & Shimla districts	Himachal Pradesh	0.56-10.11	2.17	[47]
Jaduguda	Jharkhand	0.03-11.6	NA	[48]
Tummalapalle	Andhra Pradesh	0.38-79.70	15.65	[49]
Khalilabad, Gorakhpur, Maharajganj, and Kushinagar districts	Uttar Pradesh	0.20-64.0	11.1	[50]
Jaipur & Ajmer districts	Rajasthan	0.3-61.9	NA	[51]
Fatehbad district	Haryana	1.1–113	22.3	[52]
Mohali	Punjab	0.63-24.20	NA	[53]
Mandakini valley (Garhwal Himalaya)	Uttarakhand	0.02-63.3	7	[28]
Bhagirathi valley (Garhwal Himalaya)	Uttarakhand	0.02-6.2	1.3	Present Study

Table 5The comparison ofranges and average valuesof uranium concentration indrinking water from differentparts of India

Table	5 Descriptive stati	istics for estimated	annual inge	estion dose	(µSv y <sup>-1</sup> ) fre	om the expc	sure to uranium i	n drinking w	vater			
S.No.	Annual ingestion	1 dose (µSv y <sup>-1</sup> )										
	Infants		Children		Males			Females			Pregnancy	Lactation
	0-6 m (I=0.7)	7-12  m (I=0.8)	$\frac{1-3}{(I=1.3)}$ year	4–8 year (I=1.7)	9–13 year (I=2.4)	14–18 year (I=3.3)	Adults $(I=3.7)$	9-13 year (I=2.1)	14–18 year (I=2.3)	Adults $(I=2.7)$	14–50 year (I = 3.0)	14-50  year  (I=3.8)
Min	0.01	0.01	0.01	0.02	0.02	0.03	0.03	0.02	0.02	0.02	0.03	0.03
Max	1.8	2.0	3.3	4.3	6.1	8.4	9.5	5.4	5.9	6.9	<i>T.T</i>	9.7
AM	0.4	0.4	0.7	0.9	1.2	1.7	1.9	1.1	1.2	1.4	1.6	2.0
SD	0.4	0.5	0.8	1.1	1.5	2.1	2.4	1.3	1.5	1.7	1.9	2.4
GM	0.1	0.2	0.3	0.3	0.5	0.7	0.7	0.4	0.5	0.5	0.6	0.8
I=Coi	sumption of water	r in litre per day [3	8									

Table 7	Descript	tive statistics fo	r estimated annu	al effective o	lose (μSv y <sup>-1</sup>	) due to inges	stion and inhals	ation of <sup>22</sup>	<sup>2</sup> Rn water					
	(Bq 1 <sup>-1</sup> )	Age-depender	nt ingestion dose	e (µSv y <sup>-1</sup> )										Inhala-
		Infants		Children		Males			Females			Pregnancy	Lactation	tion Dose (μSv y <sup>-1</sup> )
		0–6 Months	7–12 Months	1–3 Year	4–12 Year	9-13 Year	14-18 Year	Adults	9-13 Year	14-18 Year	Adults	14-50 Year	14-50 Year	
Min	22	56	64	104	137	193	265	297	169	185	217	241	305	56
Max	58	148	169	275	360	508	669	783	445	487	572	635	804	146
AM	35	90	103	168	220	310	426	478	271	297	349	387	491	89
SD	10	27	30	50	65	91	126	141	80	88	103	114	145	26
GM	34	87	66	161	211	298	410	459	261	285	335	372	472	86

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Fig. 8 Frequency distribution of radon concentration  $(Bq l^{-1})$  in potable groundwater of Bhagirathi valley

water samples were found well below the safe limits recommended by WHO, Health Canada, USEPA and AERB. The recommendations of these health agencies on uranium concentration and its radiological risk (ECR) in drinking water are given in Table 4. The hazard quotient (HQ) which is the measure of chemical toxicity of uranium in drinking water was found to range from 0.0003 to 0.08. No samples were found to have HQ<sup>></sup> 1, indicating that there is not risk due to chemical toxicity of due to ingestion of uranium drinking water.

The ranges and mean values of uranium concentrations in drinking water samples from different parts of India are shown in Table 5. In the Himalayan region of Himachal Pradesh (Kangra, Mandi, Kullu & Shimla areas), the uranium concentration is reported to be in the range of 0.56 to 10.11  $\mu$ g l<sup>-1</sup> with an average of 2.17  $\mu$ g l<sup>-1</sup> [47]. The uranium concentration in potable groundwater (ranging between  $0.02 \pm 0$  to  $63.7 \pm 4.3 \ \mu g \ l^{-1}$  with a mean of 7  $\mu g \ l^{-1}$ ) in Mandakini valley of Garhwal Himalaya is recently reported in our recent study [28]. It has been observed that the concentrations of uranium in potable groundwater of Mandakini valley are considerable higher than those observed in Bhagirathi valley. The higher values of uranium concentrations in Mandakini valley of Garhwal Himalaya are attributed to localised uranium mineralization in the form of granitic rocks in the upper part of Mandakini valley [28]. The estimated values of age dependent ingestion dose rates from the exposure to uranium through drinking water is presented in Table 6. The calculated dose rates were observed well within 100  $\mu$ Sv y<sup>-1</sup>, a reference value recommended by WHO [41].

#### Levels and effects of radon in drinking water

As stated earlier, the measurements of radon concentrations were carried out in potable groundwater samples collected from 30 locations in Bhagirathi valley of Garhwal Himalaya. The descriptive statistics for radon concentrations in potable groundwater samples and associated age dependent doses are shown in Table 7. Radon concentration in analyzed groundwater samples was found to vary from 22 to 58 Bq  $l^{-1}$  with an average of 35 Bq  $l^{-1}$ . It has been observed that radon concentration in all the analyzed samples were found to exceed the safe limit of 11 Bq 1<sup>-1</sup> recommended by USEPA [46]. However, all water samples were found to have radon concentration well withing the maximum concetration limit of 100 Bq 1<sup>-1</sup> recommended by WHO for its safe use as drinking purpose [41]. The frequency distribution of radon concentration in water samples is shown in Fig. 8. The high levels of radon in analyzed water samples may due to reason that samples

Radon Concentration (Bq  $l^{-1}$ ) Investigated Location State References Range Average Amritsar City Punjab 0.53-11.20 NA [54] Jodhpur & Nagaur districts Rajasthan 0.5 - 15NA [55] Jalandhar Punjab 0.34-3.84 1.46 (AM) [56] Fatehbad district Haryana 1.4-22.6 NA [57] Bhilangana valley (Garhwal Himalaya) Uttarakhand 1 - 16828 (GM) [58] 5-336 Kumaun Himalaya Uttarakhand 61 (GM) [58] New Tehri area (Garhwal Himalaya) Uttarakhand 2 - 5831 (AM) [29] Karnprayag area (Pindar valley, Garhwal Uttarakhand 2-47 15 (AM) [29] Himalaya) Mandakini valley (Garhwal Himalaya) Uttarakhand 1.7 - 40045 (AM) [29] Bhagirathi valley (Garhwal Himalaya) Uttarakhand 22 - 5835 (AM) Present Study

AM arithmetic mean, GM geometric mean, NA not available

Table 8The comparison ofranges and average values ofradon concentration in drinkingwater from different parts ofIndia

were directly taken from groundwater sources. The measurements were performed on the sampling spots immediately after taking samples from groundwater (springs and bore wells) sources so as to avoid the release of radon from water samples. The comparison of ranges and average values of radon concentrations from diffeent parts of India are shown in the Table 8. The average value of radon concentration (35 Bq  $l^{-1}$ ) in groundwater of Bhagirathi valley is lower than the recently reported [29] values of radon concentration in groundwater (45 Bg  $1^{-1}$ ) in Mandakini valley but higher than those in Karnprayag area (15 Bq  $l^{-1}$ ) and New Tehri area (31 Bq l<sup>-1</sup>) of Garhwal Himalaya. The higher values of radon concentration in Mandakini valley may be attributed to the fault fault-lineament (FL) type and fractured joint (FJ) type springs in this valley. These types of springs present an increased rock surface area to water volume ratio; thereby increasing the radon emission rate [29]. However, the springs in Karnparayag and New Tehri areas are colluvial (CL) and fluvial deposit (FD) type springs which possess high water carrying capability, high permeability and high porosity. The high porosity of these springs results in the migration of radon gas into atmosphere; thereby decreasing the radon concentration in water [29].

The measured values of radon cocentrations in potable groundwater samples of Bhagirathi valley were used to calculate annual effective dose due to ingestion for different age groups and inhalation of radon released from household water. The observed values of age dependent ingestion dose and inhalation dose are presented in Table 7.

# Conclusions

The activity levels of natural radionuclides (<sup>226</sup>Ra, <sup>232</sup>Th & <sup>40</sup>K) in soil samples at majority of locations in Mandakini valley of Garhwal Himalaya are higher than the corresponding average values in the world. The high values of natural radionuclides in soil samples show the uranium and thorium mineralization in the region. The observed high values hazard assessment quantities calculated for the soil of the region indicate that use of local soil and/or rocks as building materials should be avoided in the investigated region in order to protect the dwellers from possibility of radiation hazards. The levels of uranium in all drinking water samples of Bhagirathi valley of Garhwal Himalaya are well within the safe limits recommended by WHO and USEPA. The activity levels of radon in potable groundwater at majority of locations are higher than the reference value of 11 Bq  $l^{-1}$ recommended by USEPA. However, radon levels in potable groundwater are well within the safe limit of 100 Bq l<sup>-1</sup> recommended by WHO. The results of measurements of natural radioactivity levels in soil and groundwater will be useful in radiation protection, geo scientific research, exploration of radionuclide minerals and in establishing guidelines for radiation protection program of the country. In addition, the results of radioactivity in soil and groundwater will be helpful in understanding the geochemical processes in the Himalayan regions.

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