



Assessment of radioactivity levels and radiological-hazard indices in plant fertilizers used in Iraqi Kurdistan Region

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Abstract

The present study is to determine the activity concentration of radioactive nuclide in plant fertilizers used in agriculture in Iraqi Kurdistan region using high-purity germanium detector (HPGe) gamma spectrometer. The results showed that the range of activity concentrations for ²²⁶Ra, ²³²Th, ⁴⁰K and ¹³⁷Cs in chemicals and organic fertilizers are (0.1–134), (0.1–74), (1–12,000) and (0–1) Bq/kg respectively, based on the measured activities that were used to assess the radiological hazards. Radium equivalent activity in some samples exceeds the value (370 Bq/kg) which recommended by the OECD.

Keywords Radioactivity · Organic fertilizer · Chemical fertilizer · Radiological hazards · HPGe detector

Introduction

In latest years, agricultural activities have extended widely, resulting in the applications of the different types of fertilizers. Chemical fertilizers are chemical compounds that provide necessary chemical elements and nutrients to the plants. Fertilizers have become important to the agricultural field all over the world [1]. It has been known that phosphate rocks contain high concentrations of uranium, thorium, radium and their decay products [2]. Phosphate rock is a main raw material used for manufacturing different types of phosphate fertilizers. Therefore, when this rock is processed into phosphates' fertilizers, most of the radioactive nuclides come into the fertilizers. It has been estimated that phosphate fertilizers that applied to the fields could increase the radioactivity level in soils [3].

Fertilizers reorganize naturally occurring radioactive nuclides at trace levels through the environment and become a source of radiation. This phenomenon may result in potential radiological-hazard due to possible migration

of elements of the plant fertilizers to soil and plants, and via the food chain, to human beings where this may lead to internal exposure through ingestion of food grown on fertilized soils [4].

The abnormal content of uranium, thorium and its decay products in ore rocks and fertilizer are the main sources of high radioactivity background parts that have been recognized in several areas of the world [5]. From the point of view of the international atomic energy agency of radioactive nuclides in food and the environment, it is necessary to measure the natural environmental radiation levels provided by ground, air, water, fertilizer etc. and to assess the dose limits of exposure to radiation from these naturally occurring radioactive nuclides [6, 7]. Human activities which involve the presence of natural radiation sources that lead to a significant increase in the exposure of workers, Regulation (EU) No 305/2011 requires information to be made available when products are placed on the market, and the current annual effective dose limits for occupational and public exposure should be maintained [8].

The aims of the present study are to determine the content of radioactivity in fifty different types of plant fertilizers that were commonly used in agriculture in Iraqi Kurdistan Region and to estimate their radiological risk to farmers and workers working in the a fertilizer factories and stores of the long-term exposure due to their application. Comparisons of the presented results were obtained with those of national and the world averages.

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Experimental

To determine the activity concentration of radioactive nuclides, fifty samples of plant fertilizers were collected from markets and companies in Iraqi Kurdistan Region. Properties of these samples were listed in Table 1. Kurdistan is located in the Middle East; it is situated in the north of Iraq, southeast of Turkey, northwest of Iran, and northeast of Syria, as shown in Fig. 1.

The samples were crushed and meshed perfectly to pass through 0.2 mm mesh. One kilogram from each sample was packed in a Marinelli beaker (one-liter size) for gamma spectrometry and sealed for 6 weeks to reach secular equilibrium between the radium contents of the sample and their daughter radionuclides [9, 10].

Each sample from fertilizers was subjected to a gamma-ray spectrometer with HPGe detector setup and the multi-channel analyzer. The high-purity germanium detector is p-type of vertical closed-end coaxial, manufactured by PGT (Princeton Gamma Tec-PGT Company-USA) with the following specifications: Crystal diameter is 70.6 mm, crystal length 70.7 mm, resolution (FWHM) at 122 keV of ^{57}Co is 1.18 keV and at 1332 keV of ^{60}Co is 1.97 keV with relative efficiency 73.8% for this energy. The instrument calibrated for energy by using the standard point gamma-ray sources, such as ^{60}Co (peaks 1173.2 and 1332.5 keV), ^{137}Cs (peak 661.7 keV) and ^{226}Ra (peaks 186.1, 295, 351.9, 609, 665, 1120 and 1764 keV). The efficiency calibration was achieved using the same three standard sources, and the relative efficiency curve was normalized to absolute volume efficiency curve, using two radioactive samples prepared [potassium chloride KCl powder and of Uranylacetate ($\text{UO}_2(\text{OCOCH}_3)_2 \cdot 2\text{H}_2\text{O}$) solution] to satisfy the geometrical conditions of the tested samples.

The detector was placed in a wall of lead with thickness (10 cm) to shield the measuring station against background radioactivity. The samples were placed over the detector for at least 10 h. The spectra were evaluated by Thermo Scientific System 8000 multi-channel analyzer and the computer software program (Quantum Gold 2001 for PGT Company). In order to determine the background distribution in the environment around the detector, an empty sealed beaker was counted in the same manner and in the same geometry as the samples. The measurement time of activity or background was 10 h. (The background spectra were used to correct the net peak area of gamma rays of measured isotopes). After measurement and subtraction of the background, the activity concentration was calculated from following gamma-ray photopeak lines [11, 12].

(a) ^{226}Ra activity concentration was calculated as the weighted average of the activity determined, using the gamma-ray lines 351.9 (35.8%) keV gamma-rays

from ^{214}Pb decay, 609.3 (44.8%), 1120 (14.8%) and 1764.5 keV (15.36%) gamma-rays from ^{214}Bi decay.

- (b) The gamma-ray photo peaks used for the determination the activity concentration of the ^{232}Th were 238.6 keV (43%) from ^{212}Pb decay, 583 keV (84.5%) and 2614.5 keV (99.16%) from ^{208}Tl decay and 911.2 keV (26.6%) from ^{228}Ac decay.
- (c) The activity concentration of ^{40}K was directly determined using 1460.8 (10.7%) gamma-rays line.
- (d) The activity concentration of ^{137}Cs was directly determined using 661.7 (85.21%) gamma-rays line.

Theory

Activity concentration

Activity concentration of the isotopes calculated using the following formula [13, 14]:

$$\text{Activity concentration} = \left[\frac{\text{Net count}}{\varepsilon \times I_\gamma \times T \times M} \right] \pm \left[\frac{\text{SD}}{\varepsilon \times I_\gamma \times T \times M} \right] \quad (1)$$

where ε is the absolute gamma peak efficiency of the detector to its particular gamma-ray energy, I_γ is the decay intensity of the specific energy peak (including the decay branching ratio information), T is the counting time of the measurement in second, M is the mass of the sample in kg and SD is the standard deviation of the net count rate per second.

Radium equivalent activity (Ra_{eq})

The total exposure to radiation from these naturally occurring radioactive nuclides was defined in terms of radium equivalent activity (Ra_{eq}) in (Bq/kg) and used to assess the gamma-ray radiation hazards due to specified radionuclides ^{226}Ra , ^{232}Th and ^{40}K . The Ra_{eq} of a sample in (Bq/kg) can be achieved using the following relation [15, 16]:

$$\text{Ra}_{\text{eq}} = (A_{\text{Ra}}) + (A_{\text{Th}} \times 1.43) + (A_{\text{K}} \times 0.077) \quad (2)$$

where A_{Ra} , A_{Th} , A_{K} are the activity concentrations of ^{226}Ra , ^{232}Th and ^{40}K in a unit (Bq/kg) respectively. The published maximal permissible Ra_{eq} is 370 Bq/kg [17, 18].

The absorbed gamma dose rates (D_R)

The absorbed gamma dose rates in air at 1 m above the ground surface for the uniform distribution of naturally occurring radioactive nuclides were calculated based on guidelines provided by [19]:

Table 1 Type, a chemical component and form of plant fertilizer samples under study

No.	Code of fertilizer samples	Type of fertilizer	Chemical components of fertilizer			Form of fertilizer	Made in
			N (%)	P ₂ O ₅ (%)	K ₂ O (%)		
1	K 1	Potassium sulfate	0	0	52	Powder	Belgium
2	K 2	Potassium nitrate	0	0	12	Foliar spray	Turkey
3	N 1	Urea	46	0	0	Granular	Russia
4	N 2	Urea	46	0	0	Granular	Iran
5	N + Ca (NCa 1)	Calcium nitrate	13	0	0	Powder	Belgium
6	N + Ca (NCa 2)	Calcium nitrate	13.5	0	0	Powder	Belgium
7	NP 1	Urea ammonium phosphate	20	20	0	Granular	Russia
8	NP 2	Diammonium phosphate	18	46	0	Granular	Turkey
9	NP 3	Diammonium phosphate	18	46	0	Granular	Jordan
10	NP 4	Diammonium phosphate	18	46	0	Granular	Tunisia
11	NP 5	Urea ammonium phosphate	18	44	0	Powder	Belgium
12	NP 6	Urea ammonium phosphate	18	44	0	Powder	Belgium
13	NP 7	Urea ammonium phosphate	17	44	0	Powder	Netherlands
14	NP 8	Mono ammonium phosphate	17	62	0	Powder	Netherlands
15	NPK 1	N.P.K. complex	12	6	18	Granular	Italy
16	NPK 2	N.P.K. complex	18	18	5	Granular	UAE
17	NPK 3	N.P.K. complex	15	30	10	Powder	Netherlands
18	NPK 4	N.P.K. complex	20	5	5	Powder	Spain
19	NPK 5	N.P.K. complex	13	2	44	Powder	Belgium
20	NPK 6	N.P.K. complex	12	8	40	Powder	Belgium
21	NPK 7	N.P.K. complex	15	8	26	Powder	Belgium
22	NPK 8	N.P.K. complex	20	20	20	Powder	Belgium
23	NPK 9	N.P.K. complex	20	20	20	Powder	Belgium
24	NPK 10	N.P.K. complex	12	8	16	Powder	Germany
25	NPK 11	N.P.K. complex	20	20	20	Powder	Germany
26	NPK 12	N.P.K. complex	12	8	40	Powder	Turkey
27	NPK 13	N.P.K. complex	20	20	20	Powder	Turkey
28	NPK 14	N.P.K. complex	15	30	15	Powder	Turkey
29	NPK 15	N.P.K. complex	10	5	30	Powder	Jordan
30	NPK 16	N.P.K. complex	18	18	18	Powder	Jordan
31	NPK 17	N.P.K. complex	12	47	10	Powder	Israel
32	NPK 18	N.P.K. complex	12	6	40	Powder	Israel
33	NPK 19	N.P.K. complex	20	20	20	Powder	Israel
34	NPK 20	N.P.K. complex	16	8	32	Foliar spray	UAE
35	NPK 21	N.P.K. complex	13	11	11	Foliar spray	UAE
36	NPK 22	N.P.K. complex	10	52	8	Foliar spray	UAE
37	NPK 23	N.P.K. complex	30	10	10	Foliar spray	USA
38	NPK 24	N.P.K. complex	10	20	30	Foliar spray	USA
39	NPK 25	N.P.K. complex	10	52	10	Foliar spray	USA
40	PK (L 1)	Potassium phosphate	0	25	28	Liquid	USA
41	K (L 2)	Potassium nitrate	0	0	26	Liquid	Spain
42	N (L 3)	Urea ammonium nitrate	31	0	0	Liquid	KSA
43	P (L 4)	Super phosphate	0	62	0	Liquid	KSA
44	Organic + NPK (L5)	Organic + complex	7	21	0	Liquid	UAE
45	Organic + NPK (OR1)	Organic + complex	6	12	10	Powder	Spain
46	Organic (OR 2)	Organic	2.03	0.52	0.43	Granular	Germany
47	Organic (OR 3)	Organic	1.12	0.61	0.61	Granular	Turkey

Table 1 (continued)

No.	Code of fertilizer samples	Type of fertilizer	Chemical components of fertilizer			Form of fertilizer	Made in
			N (%)	P ₂ O ₅ (%)	K ₂ O (%)		
48	Organic (OR 4)	Organic	4	0.11	0.55	Powder	Turkey
49	Organic (OR 5)	Organic	3.29	0.5	0.59	Powder	Iraq
50	Organic (OR 6)	Organic	1.82	0.3	0.52	Powder	Iraq

Fig. 1 Map of Iraqi Kurdistan Region

$$D_R(\text{nGy/h}) = (A_{\text{Ra}} \times 0.462) + (A_{\text{Th}} \times 0.604) + (A_{\text{K}} \times 0.0417) \quad (3)$$

Annual effective dose (AED)

In order to estimate the annual effective dose rate in the air, the conversion coefficient from the absorbed dose in the air to the effective dose received by an adult must be considered. Annual estimated average effective dose (AED) received by an individual was calculated using a conversion factor of 0.7 Sv/Gy, which was used to convert the absorbed rate to the human effective dose. The annual effective dose is determined using the following equations [15, 16]:

$$\text{AED} = D_R \times T \times xF \quad (4)$$

where D_R is the calculated dose rate (in nGy/h), T is the occupancy time and F is the conversion factor, this value is published in UNSCEAR 1993 and UNSCEAR 2000

[20, 21], to be 0.7 Sv/Gy for environmental exposure to gamma rays of moderate energy. The outdoor occupancy factor T is about (20% of 8760 h/y). The outdoor annual effective dose is given by the following equation:

$$\text{AED}_{\text{outdoor}}(\mu\text{Sv/y}) = D_R(\text{nGy/h}) \times (0.2 \times 8760 \text{ h/y}) \times 0.7 (\text{Sv/Gy}) \quad (5)$$

The world average annual effective dose (AED) from outdoor or indoor terrestrial gamma radiation only is 70 $\mu\text{Sv/y}$ [21].

External hazard index (H_{ex})

The external hazard index is an evaluation of the hazard caused by gamma-rays emitted from natural radioactive nuclides. It was calculated by assuming that 370 Bq/kg of ²²⁶Ra or 259 Bq/kg of ²³²Th or 4810 Bq/kg of ⁴⁰K produces the same gamma-ray dose rate. The prime objective

of this index is to limit the radiation dose to the admissible permissible dose equivalent limit around 1 mSv/y. In order to evaluate this index, one can use the following relation [15, 18]:

$$H_{\text{ex}} = (A_{\text{Ra}}/370) + (A_{\text{Th}}/259) + (A_{\text{K}}/4810) \leq 1 \quad (6)$$

This model takes into consideration the maximum value of external hazard which is caused by gamma-rays corresponds to a radium equivalent activity of 370 Bq/kg [17, 18].

Excess lifetime cancer risk (ELCR)

Excess lifetime cancer risk measures the additional cancer risk induced by exposure to ionizing radiations. Based on the calculated values of AED, ELCR is calculated using the equation [22, 23]:

$$\text{ELCR}_{\text{outdoor}} = \text{AED}_{\text{outdoor}} (\mu\text{Sv/y}) \times \text{DL} \times \text{RF} \quad (7)$$

where AED is the annual effective dose, DL is the average duration of life which is 70 year, RF is the risk factor given as 0.05 by ICRP 1991 [24].

Results and discussion

The activity concentrations of ^{226}Ra , ^{232}Th , ^{40}K , and ^{137}Cs in different fertilizer samples were measured and listed in Table 2. The results showed that the mean value and a range of activity concentrations of ^{226}Ra , ^{232}Th , ^{40}K and ^{137}Cs in these fertilizers are 7 ± 1 (0.1–134), 4 ± 1 (0.1–74), 2500 ± 100 (1–12,000) and 0.1 ± 0.01 Bq/kg (0–1) Bq/kg respectively. The highest value of activity concentrations of ^{226}Ra and ^{232}Th are found in NPK fertilizer, while the lowest values were found in urea and calcium fertilizers. Except for fertilizer samples (NPK1, NPK2, NPK3, OR3, and OR4), the measured value of activity concentrations of ^{226}Ra and ^{232}Th are lower than the word wide average value of 35 and 30 Bq/kg for ^{226}Ra and ^{232}Th respectively [21]. The variation of radionuclides activities in studying fertilizers may be due to the different origins of raw material and the chemical processes during manufacturing of the fertilizer.

As shown in Fig. 2, the activity concentration of ^{226}Ra in most studied fertilizer samples are higher than the activity concentration of ^{232}Th in these samples, which may be due to the accumulation of dissolved uranium and its products, in the form of urinal complex in the seawater during geological formation of the phosphate rocks [25].

Figures 3 and 4 shows the relationships between activity concentrations of ^{226}Ra and ^{232}Th with concentration of P_2O_5 in fertilizer samples. It is clear that there is no linear relationship between activity concentrations of ^{226}Ra and

^{232}Th and concentration of P_2O_5 in fertilizer samples, and the variation of radionuclides activities in studying fertilizers due to different origins of raw material and the chemical processes during the manufacturing of the fertilizer.

The highest value of activity concentration ^{40}K was found in potassium sulfate fertilizer, while the lowest value was found in urea and calcium fertilizers. This is due to high concentration of K_2O in potassium sulfate fertilizer, because there is a good relationship between concentration of K_2O and ^{40}K radioactive isotope content, and the presence of radioactive ^{40}K , whose natural abundance in natural potassium K is 0.0118%, and this relation was found in Fig. 5. The activity concentration of ^{40}K in potassium sulfate and NPKs fertilizer samples was higher than the word wide average value for ^{40}K which is 400 Bq/kg [21]. While the activity concentration of ^{40}K in Urea, Calcium, NP, and organic samples, was lower than the word wide average value. The activity concentration of ^{137}Cs was found only in organic fertilizer samples, this may be due to the production process of the organic fertilizers. The results for the activity concentration of ^{226}Ra , ^{232}Th , ^{40}K , and ^{137}Cs , of the present work, compared with other studies are presented in Table 3.

In order to compare the activity concentrations of ^{226}Ra , ^{232}Th and ^{40}K in fertilizer samples, the radium equivalent activity (Ra_{eq}) as a common index was used to obtain the sum of activities. To estimate the radiological risk to farmers and workers working in a fertilizer factory of the long-term exposure due to their application, the external absorbed dose rate (D), annual effective dose (AED), external hazard index (H_{ex}) and excess lifetime cancer risk (ELCR) for different types of fertilizer samples were calculated and listed in Table 4.

The value of radium equivalent activity (Ra_{eq}) in the fertilizer samples varied from (0.4 to 900) Bq/kg, the variation of radium equivalent activity (Ra_{eq}) in studying fertilizers, due to the different amount content of ^{226}Ra , ^{232}Th and ^{40}K in fertilizers. The calculated value of radium equivalent (Ra_{eq}), except for fertilizer samples (K1, NPK1, NPK5, NPK6, NPK7, NPK12, NPK13, NPK15, NPK18, NPK20, NPK24 and L1), were lower than the recommended level of 370 Bq/kg [17, 18, 31].

The calculated value of the external absorbed dose rate (D_{R}) of the fertilizer samples varied from (0.2 to 480) nGy/h. The value of absorbed dose rate (D_{R}) in Potassium Sulfate, Potassium Nitrate and NPK complex fertilizers were higher than the worldwide average value, while the value of absorbed dose rate (D_{R}) in urea, diammonium phosphate, urea ammonium phosphate, calcium nitrate and organic were lower than the worldwide average value of 59 nGy/h [21].

Table 2 The activity concentrations of ^{226}Ra , ^{232}Th , ^{40}K and ^{137}Cs in different types of fertilizer samples

No.	Code of fertilizer samples	Activity concentrations of ^{226}Ra (Bq/kg)	Activity concentrations of ^{232}Th (Bq/kg)	Activity concentrations of ^{40}K (Bq/kg)	Activity concentrations of ^{137}Cs (Bq/kg)
1	K 1	1 ± 0.2	0.2 ± 0.02	12,000 ± 600	BDA
2	K 2	16 ± 1	16 ± 1	1200 ± 60	BDA
3	N 1	0.2 ± 0.04	0.2 ± 0.03	0.4 ± 0.03	BDA
4	N 2	0.1 ± 0.03	0.1 ± 0.02	1 ± 0.3	BDA
5	N + Ca (NCa 1)	2 ± 0.1	0.2 ± 0.03	2 ± 0.3	BDA
6	N + Ca (NCa 2)	2 ± 0.2	0.2 ± 0.05	1 ± 0.2	BDA
7	NP 1	0.6 ± 0.1	4 ± 0.2	53 ± 2	BDA
8	NP 2	13 ± 1	3 ± 0.2	60 ± 3	BDA
9	NP 3	8 ± 1	1.5 ± 0.1	14 ± 1	BDA
10	NP 4	6 ± 1	12 ± 1	20 ± 1	BDA
11	NP 5	0.2 ± 0.06	5 ± 0.2	2 ± 0.5	BDA
12	NP 6	0.4 ± 0.1	10 ± 1	4 ± 0.5	BDA
13	NP 7	0.2 ± 0.04	0.2 ± 0.03	1 ± 0.2	BDA
14	NP 8	0.4 ± 0.1	0.2 ± 0.04	13 ± 2	BDA
15	NPK 1	134 ± 6	1 ± 0.2	4000 ± 200	BDA
16	NPK 2	13 ± 1	74 ± 3	1000 ± 50	BDA
17	NPK 3	0.6 ± 0.2	0.2 ± 0.03	2200 ± 100	BDA
18	NPK 4	0.3 ± 0.1	1 ± 0.3	1000 ± 50	BDA
19	NPK 5	0.1 ± 0.01	0.8 ± 0.2	10,000 ± 500	BDA
20	NPK 6	1 ± 0.2	1 ± 0.3	7000 ± 500	BDA
21	NPK 7	1 ± 0.04	0.2 ± 0.04	5000 ± 250	BDA
22	NPK 8	0.5 ± 0.2	1 ± 0.3	2400 ± 120	BDA
23	NPK 9	0.3 ± 0.03	0.1 ± 0.02	3800 ± 200	BDA
24	NPK 10	16 ± 1	15 ± 1	2400 ± 130	BDA
25	NPK 11	0.6 ± 0.1	0.1 ± 0.02	3500 ± 180	BDA
26	NPK 12	1 ± 0.3	0.2 ± 0.03	7000 ± 400	BDA
27	NPK 13	1 ± 0.3	0.1 ± 0.02	6000 ± 300	BDA
28	NPK 14	1 ± 0.01	0.3 ± 0.03	3400 ± 200	BDA
29	NPK 15	1 ± 0.2	1 ± 0.4	5300 ± 300	BDA
30	NPK 16	1 ± 0.2	0.2 ± 0.03	3200 ± 200	BDA
31	NPK 17	0.6 ± 0.1	0.4 ± 0.1	2400 ± 100	BDA
32	NPK 18	1 ± 0.2	0.3 ± 0.08	7000 ± 400	BDA
33	NPK 19	1 ± 0.3	0.2 ± 0.04	4000 ± 200	BDA
34	NPK 20	0.2 ± 0.03	0.1 ± 0.01	5000 ± 300	BDA
35	NPK 21	0.3 ± 0.09	0.3 ± 0.08	2000 ± 100	BDA
36	NPK 22	0.6 ± 0.1	0.1 ± 0.03	1800 ± 100	BDA
37	NPK 23	1 ± 0.1	0.2 ± 0.05	2000 ± 100	BDA
38	NPK 24	1 ± 0.2	0.3 ± 0.02	6200 ± 300	BDA
39	NPK 25	1 ± 0.3	0.5 ± 0.1	2000 ± 30	BDA
40	PK (L 1)	0.6 ± 0.2	0.2 ± 0.03	6400 ± 300	BDA
41	K (L 2)	0.5 ± 0.1	0.1 ± 0.03	2100 ± 100	BDA
42	N (L 3)	1 ± 0.1	0.4 ± 0.06	21 ± 2	BDA
43	P (L 4)	1 ± 0.1	3 ± 0.1	8 ± 1	BDA
44	Organic + NPK (L5)	0.4 ± 0.1	0.2 ± 0.03	9 ± 1	0.2 ± 0.03
45	Organic + NPK(OR1)	7 ± 1	6 ± 0.2	1000 ± 100	0.3 ± 0.05
46	Organic (OR 2)	12 ± 1	6 ± 0.3	100 ± 5	BDA
47	Organic (OR 3)	42 ± 1	11 ± 1	600 ± 30	0.1 ± 0.04
48	Organic (OR 4)	38 ± 1	25 ± 1	1700 ± 100	BDA

Table 2 (continued)

No.	Code of fertilizer samples	Activity concentrations of ^{226}Ra (Bq/kg)	Activity concentrations of ^{232}Th (Bq/kg)	Activity concentrations of ^{40}K (Bq/kg)	Activity concentrations of ^{137}Cs (Bq/kg)
49	Organic (OR 5)	2 ± 0.3	0.5 ± 0.1	300 ± 30	BDA
50	Organic (OR 6)	4 ± 0.4	2 ± 0.3	250 ± 30	1 ± 0.2
Over all	Range	0.1–134	0.1–74	1–12,000	0–1
	Average	7	4	2500	0.1

*BDA below minimum detectable activity

Fig. 2 Activity concentrations of ^{226}Ra and ^{232}Th in fertilizer samples

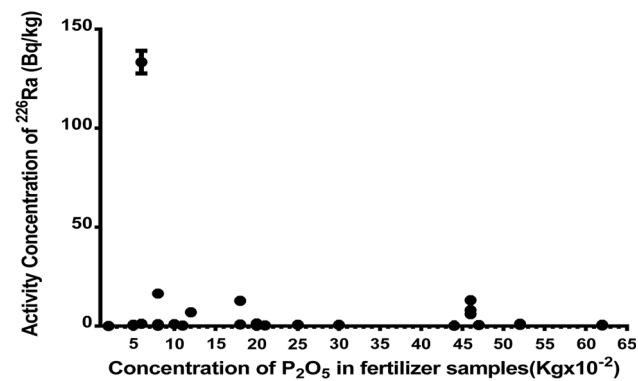
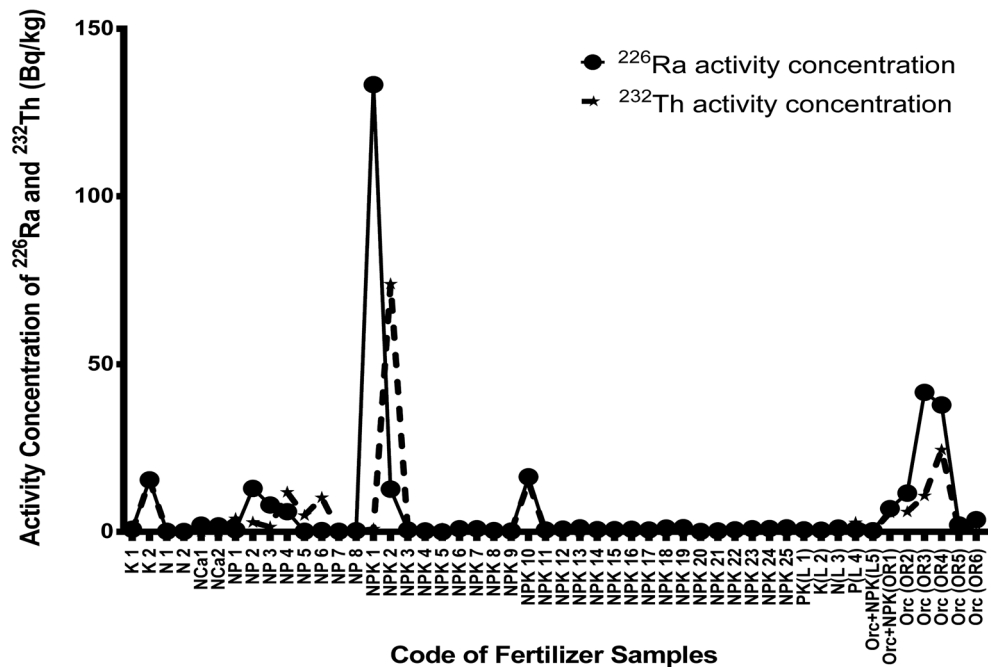


Fig. 3 Relationship between activity concentration of ^{226}Ra and concentration of P_2O_5 in fertilizer samples

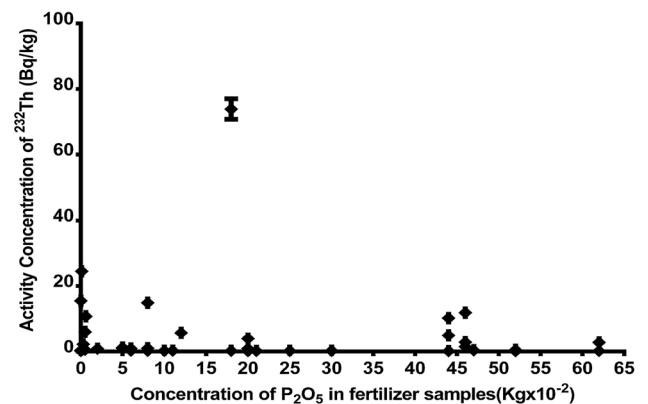


Fig. 4 Relationship between activity concentration of ^{232}Th and concentration of P_2O_5 in fertilizer samples

The calculated values of outdoor annual effective dose (AED) in the fertilizer samples varied from (0.2 to 590) $\mu\text{Sv/y}$. The calculated value of annual effective dose in

potassium sulfate, potassium nitrate and NPK complex fertilizers were higher than the worldwide average value, while the measured value of in urea, diammonium

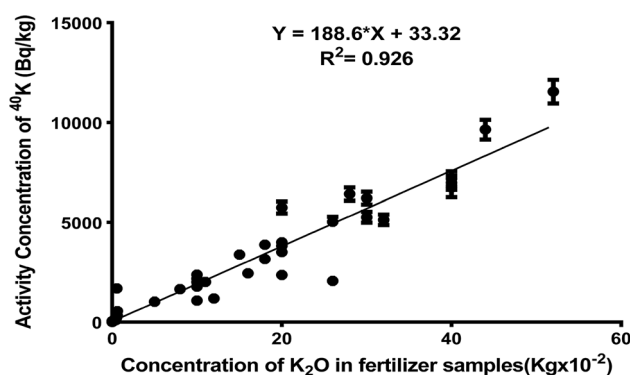


Fig. 5 Relationship between activity concentration of ^{40}K and concentration of K_2O in different fertilizer samples

phosphate, urea ammonium phosphate, calcium nitrate and organic were lower than the worldwide average value of 70 $\mu\text{Sv/y}$ [21].

The calculated value of external hazard index (H_{ex}) in the fertilizer samples varied from (0 to 2.4) mSv/y , the data indicated that the value of external hazard index in fertilizer samples (K1, NPK1, NPK5, NPK6, NPK12, NPK13, NPK18, NPK24, L1) were higher than the safety limits (1 mSv/y) [14, 15, 17]. While the value of external hazard index in other fertilizer samples within the safety limit. Thus, the use of chemical fertilizers (K1, NPK1, NPK5, NPK6, NPK12, NPK13, NPK18, NPK24, L1) in large extent can be harmful to the health of farmers and workers working in a fertilizer factory.

The calculated value of excess lifetime cancer risk (ELCR) in the fertilizer samples varied from (0 to 20×10^{-4})/person. The results of the present study

indicated that the use of plant fertilizers to enhance crop yield enhances the concentration of natural radioactive nuclides in soil and hence the exposure of farmers working in the fields.

No linear relationships have been found between external hazard index (H_{ex}) with ^{226}Ra and ^{232}Th activity concentrations, but present result in Fig. 6 shows a good correlation between external hazard index (H_{ex}) and ^{40}K activity concentration with a correlation coefficient of ($R^2 = 0.98$), this means that most of the hazard caused by nuclear radiation from fertilizer samples were coming from ^{40}K radioactive nuclide (or by the high concentration of ^{40}K radioactive nuclei in the fertilizer samples).

Conclusions

The activity concentration of radioactive nuclide and related radiation hazards in plant fertilizer samples being used in Iraqi Kurdistan Region were determined by using a HPGe gamma spectrometer.

Except for fertilizer samples (NPK1, NPK2, NPK3, OR3, and OR4), the measured value of activity concentrations of ^{226}Ra and ^{232}Th are lower than the world wide average value. The activity concentration of ^{40}K in potassium sulphate and NPKs fertilizer samples was higher than the world average value, while its activity in urea, calcium, NP and organic samples was lower than the world average value. No relations have been found between activity concentrations of ^{226}Ra and ^{232}Th with the concentration of P_2O_5 in fertilizer samples. The use of potassium sulphate and NPKs fertilizer made the calculated value of external

Table 3 Comparison of the obtained values of activity concentrations of ^{226}Ra , ^{232}Th , ^{40}K and ^{137}Cs in fertilizer samples with data from similar studies in the different country of the world

Country	^{226}Ra		^{232}Th		^{40}K		^{137}Cs		References
	Min.	Max.	Min.	Max.	Min.	Max.	Min.	Max.	
Activity concentrations (Bq/kg) of ^{226}Ra , ^{232}Th , ^{40}K and ^{137}Cs in chemical fertilizer samples									
Algeria	134 ± 24	190 ± 30	117 ± 10	131 ± 16	5312 ± 249	$11,645 \pm 550$	7.5 ± 0.3	11.7 ± 0.6	[26]
Nigeria	10 ± 7	451 ± 14	BDA	15 ± 3	3972 ± 417	5089 ± 111			[27]
Egypt	3	283	2	74	9	6501			[25]
KSA	29	121	3	57	227	4227			[28]
Iraq	13 ± 1	89 ± 3	1 ± 0.2	27 ± 1	12 ± 1	2276 ± 18			[10]
Iraq	0.1 ± 0.01	134 ± 6	0.1 ± 0.02	74 ± 3	1 ± 0.2	$12,000 \pm 600$	0	0.3 ± 0.05	Present study
Activity concentrations (Bq/kg) of ^{226}Ra , ^{232}Th , ^{40}K and ^{137}Cs in organic fertilizer samples									
Nigeria	13 ± 0.01	66 ± 1	15 ± 0.01	60 ± 1	120 ± 1	908 ± 1			[29]
KSA	21	70	3	21	115	523			[28]
Iraq	12 ± 5	69 ± 7	11 ± 2	52 ± 5	165 ± 7	3593 ± 34			[30]
Iraq	2 ± 0.3	42 ± 1	0.5 ± 0.1	25 ± 1	100 ± 5	1700 ± 100	0	1 ± 0.2	Present study

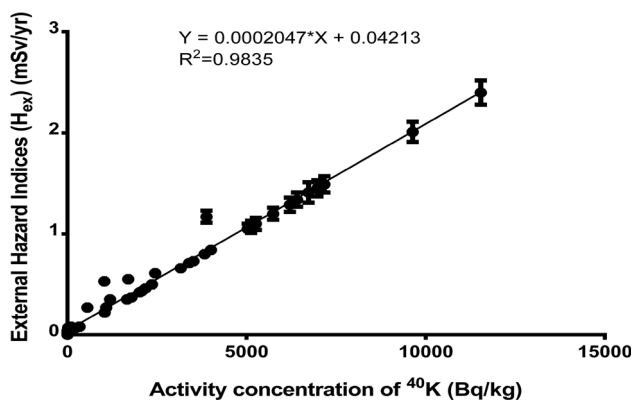
*BDA below minimum detectable activity

Table 4 Radium equivalent (Ra_{eq}), External Hazard Index (H_{ex}), External Doses Rate (D_R), Annual Effective Dose (AED) and Excess lifetime cancer risk (ELCR) for different types of fertilizer samples

No.	Code of samples	Ra_{eq} (Bq/kg)	D_R (nGy./h)	AED (μ Sv./y)	H_{ex} (mSv/y)	ELCR $\times 10^{-4}$ (1/person)
1	K 1	900 \pm 50	480 \pm 25	590 \pm 30	2.4 \pm 0.1	20 \pm 1
2	K 2	130 \pm 10	70 \pm 5	80 \pm 5	0.4 \pm 0.02	3 \pm 0.1
3	N 1	0.4 \pm 0.1	0.2 \pm 0.04	0.2 \pm 0.05	0	0
4	N 2	0.4 \pm 0.08	0.2 \pm 0.04	0.2 \pm 0.05	0	0
5	N + Ca (NCa1)	2.4 \pm 0.2	1 \pm 0.1	1 \pm 0.1	0	0
6	N + Ca (NCa2)	2 \pm 0.4	1 \pm 0.2	1 \pm 0.2	0	0
7	NP 1	10 \pm 1	5 \pm 0.2	6 \pm 0.3	0	0.2 \pm 0.01
8	NP 2	22 \pm 1	10 \pm 1	13 \pm 1	0.06 \pm 0.01	0.4 \pm 0.02
9	NP 3	11 \pm 1	5 \pm 0.3	1 \pm 0.4	0	0.2 \pm 0.01
10	NP 4	25 \pm 1	11 \pm 1	13 \pm 1	0.07 \pm 0.01	0.4 \pm 0.02
11	NP 5	7 \pm 1	3 \pm 0.2	4 \pm 0.2	0	0.1 \pm 0.01
12	NP 6	15 \pm 1	7 \pm 0.2	8 \pm 0.2	0.04 \pm 0.01	0.3 \pm 0.08
13	NP 7	0.4 \pm 0.1	0.2 \pm 0.05	0.3 \pm 0.06	0	0
14	NP 8	2 \pm 0.3	1 \pm 0.2	1 \pm 0.2	0	0
15	NPK 1	440 \pm 20	220 \pm 11	270 \pm 10	1.2 \pm 0.06	9 \pm 0.4
16	NPK 2	200 \pm 10	95 \pm 5	110 \pm 5	0.5 \pm 0.03	4 \pm 0.2
17	NPK 3	170 \pm 10	90 \pm 5	110 \pm 5	0.5 \pm 0.02	4 \pm 0.2
18	NPK 4	81 \pm 5	44 \pm 2	54 \pm 5	0.2 \pm 0.01	2 \pm 0.1
19	NPK 5	750 \pm 40	400 \pm 20	490 \pm 30	2 \pm 0.1	20 \pm 1
20	NPK 6	520 \pm 40	280 \pm 20	340 \pm 30	1.4 \pm 0.1	10 \pm 1
21	NPK 7	400 \pm 20	210 \pm 11	260 \pm 10	1 \pm 0.05	9 \pm 1
22	NPK 8	180 \pm 10	100 \pm 5	120 \pm 20	0.5 \pm 0.03	4 \pm 0.2
23	NPK 9	300 \pm 15	160 \pm 10	200 \pm 10	0.8 \pm 0.04	7 \pm 0.4
24	NPK 10	220 \pm 12	120 \pm 10	150 \pm 10	0.6 \pm 0.03	5 \pm 0.2
25	NPK 11	270 \pm 14	150 \pm 10	180 \pm 10	0.7 \pm 0.04	6 \pm 0.3
26	NPK 12	550 \pm 30	300 \pm 20	370 \pm 20	1.5 \pm 0.1	10 \pm 1
27	NPK 13	440 \pm 25	240 \pm 10	290 \pm 20	1.2 \pm 0.06	10 \pm 1
28	NPK 14	260 \pm 15	140 \pm 10	170 \pm 10	0.7 \pm 0.04	6 \pm 0.3
29	NPK 15	410 \pm 20	220 \pm 10	270 \pm 10	1.1 \pm 0.06	9 \pm 1
30	NPK 16	250 \pm 13	130 \pm 7	160 \pm 10	0.7 \pm 0.03	5 \pm 0.3
31	NPK 17	180 \pm 10	100 \pm 5	120 \pm 10	0.5 \pm 0.03	4 \pm 0.2
32	NPK 18	540 \pm 30	290 \pm 20	360 \pm 20	1.5 \pm 0.1	10 \pm 1
33	NPK 19	310 \pm 20	170 \pm 10	210 \pm 10	0.8 \pm 0.04	6 \pm 0.4
34	NPK 20	400 \pm 20	210 \pm 10	260 \pm 10	1.1 \pm 0.06	9 \pm 0.5
35	NPK 21	160 \pm 10	80 \pm 5	100 \pm 5	0.4 \pm 0.02	3 \pm 0.2
36	NPK 22	130 \pm 10	70 \pm 5	90 \pm 5	0.4 \pm 0.02	3 \pm 0.2
37	NPK 23	140 \pm 10	80 \pm 5	90 \pm 5	0.4 \pm 0.02	3 \pm 0.2
38	NPK 24	480 \pm 25	260 \pm 10	320 \pm 20	1.3 \pm 0.07	10 \pm 1
39	NPK 25	160 \pm 3	85 \pm 1	100 \pm 2	0.4 \pm 0.018	3 \pm 0.6
40	PK (L 1)	500 \pm 30	270 \pm 15	330 \pm 20	1.3 \pm 0.07	10 \pm 1
41	K (L 2)	160 \pm 10	90 \pm 5	110 \pm 5	0.4 \pm 0.02	4 \pm 0.2
42	N (L 3)	3 \pm 0.3	2 \pm 0.1	2 \pm 0.2	0	0
43	P (L 4)	5 \pm 0.4	2 \pm 0.2	3 \pm 0.2	0	0
44	Organic + NPK (L 5)	1 \pm 0.2	0.7 \pm 0.1	1 \pm 0.1	0	0
45	Organic + NPK(OR 1)	100 \pm 5	50 \pm 3	60 \pm 5	0.3 \pm 0.01	2 \pm 0.1
46	Organic (OR 2)	30 \pm 1	13 \pm 1	16 \pm 1	0.1 \pm 0.01	0.5 \pm 0.02
47	Organic (OR 3)	100 \pm 4	50 \pm 2	60 \pm 5	0.3 \pm 0.01	2 \pm 0.1
48	Organic (OR 4)	200 \pm 10	100 \pm 5	130 \pm 10	0.6 \pm 0.03	4 \pm 0.2

Table 4 (continued)

No.	Code of samples	Ra _{eq} (Bq/kg)	D _R (nGy./h)	AED (μSv./y)	H _{ex} (mSv/y)	ELCR × 10 ⁻⁴ (1/person)
49	Organic (OR 5)	30 ± 3	15 ± 1	20 ± 2	0.1 ± 0.01	0.6 ± 0.1
50	Organic (OR 6)	25 ± 3	13 ± 1	16 ± 2	0.1 ± 0.01	0.5 ± 0.06
Over all	Range	0.4–900	0.2–480	0.2–590	0–2.4	0–20
	Average	200	110	134	0.6	4

**Fig. 6** Relationship between activity concentration of ⁴⁰K and External Hazard Indices (H_{ex}) in fertilizer samples

hazard index (H_{ex}) higher than the recommended level and harmful for the health of farmers and workers working in a fertilizer factories and stores. On the other hand, fertilizer samples of urea, calcium nitrate, urea ammonium phosphate, di-ammonium phosphate and organic were safer for agriculture and they have a low effect on human health.

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