

Labeling bombesin-like peptide with ^{99m}Tc via hydrazinonicotinamide: description of optimized radiolabeling conditions

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Abstract Bombesin (BNN)-like peptides have very high binding affinity for the gastrin-releasing peptide (GRP) receptor. The goal of the current study was to optimize the labeling conditions of a new ^{99m}Tc -radiolabeled BNN-like peptide based on the bifunctional chelating ligand HYNIC using different co-ligands (EDDA and tricine). The radiolabeling conditions (pH, amount of co-ligand, amount of stannous chloride, temperature and reaction time) for newly-formed ^{99m}Tc -tricine-HYNIC-Q-Litorin and ^{99m}Tc -EDDA-HYNIC-Q-Litorin were optimized and evaluated by RHPLC and RTLC. Radiochemical yields for ^{99m}Tc -tricine-HYNIC-Q-Litorin and ^{99m}Tc -EDDA-HYNIC-Q-Litorin were 98.0 ± 1.7 and $97.5 \pm 2.5\%$, respectively. When EDDA was used as co-ligand, the labeling of ^{99m}Tc -EDDA-HYNIC-Q-Litorin was optimal in the following reaction mixture: HYNIC-peptide: EDDA: 10 μg /5 mg, pH 3, SnCl_2 concentration: 12 μg /0.1 mL, reaction temperature: 100 °C, reaction time: 15 min. Besides, the optimum conditions were HYNIC-peptide:tricine: 10 μg /50 mg, pH 5, SnCl_2 concentration: 12 μg /0.1 mL, reaction temperature: 100 °C, reaction time: 15 min for preparing ^{99m}Tc -tricine-HYNIC-Q-Litorin. The manufactured ^{99m}Tc -HYNIC-Q-Litorin conjugates may offer new possibilities for imaging cancer cells expressing bombesin receptors.

Keywords HYNIC-BNN-like peptide · Litorin · Gastrin-releasing peptide · ^{99m}Tc · Co-ligand

Introduction

Small radiolabeled peptides have become an important class of radiopharmaceuticals for diagnostic tumor imaging and other diseases in nuclear medicine. Technetium-99m (^{99m}Tc) is one of these highly used isotopes with sufficiently long half life and hence wider commercial availability [1–8]. To effectively deliver ^{99m}Tc to the targeted cells, the isotope is attached to a peptide specific to a particular receptor expressed in the cells. For example, gastrin releasing peptide receptor (GRPr) is expressed in several human tissues such as breast, prostate, lung and pancreatic cancer. As a promising class of ligands to GRPr, bombesin (BNN) or BNN-like peptide such as litorin function as growth stimulant and therefore plays an important role in carcinogenesis [1, 9–11]. In our preliminary study, we have shown that the compound obtained by simply labeling litorin with ^{99m}Tc (^{99m}Tc -litorin) has been highly uptaken by the pancreas in normal rats [11]. This preliminary study demonstrated potential of litorin in radiopharmaceutical studies and justified further work.

Another way of labeling a BNN-analogue peptide with ^{99m}Tc is to use a bifunctional chelating agent (BFCA) conjugated to a peptide. This configuration provides a high specific activity at the binding site for Technetium. A variety of BFCAs have been developed and currently available [6–8, 12–21]. One of the attractive BFCAs is 6-hydrazino nicotinic acid (HYNIC). HYNIC is attached to the amino group at the N-terminus of the BNN-like peptide via solid phase peptide synthesis method. HYNIC makes high efficiency labeling with ^{99m}Tc possible, and the final product very often requires no purification and exhibits high specific activity even with very low concentrations of the peptide [19, 22]. Chelating HYNIC with

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Preparations of ^{99m}Tc -tricine and ^{99m}Tc -EDDA

To test whether ^{99m}Tc was properly attached to tricine-HYNIC-Q-Litorin or EDDA-HYNIC-Q-Litorin during the procedures described above or bonded to tricine or EDDA alone, we separately prepared ^{99m}Tc -tricine and ^{99m}Tc -EDDA batches for quality control purposes. For preparing ^{99m}Tc -tricine, fifty mg of tricine was dissolved in water (150 μL) in a tube and 50 $\mu\text{g}/100 \mu\text{L}$ stannous chloride solution along with 111 MBq $\text{Na}^{99m}\text{TcO}_4$ were added to the solution. The pH of the reaction was 4.6. The tube was kept at room temperature for 30 min. For preparing ^{99m}Tc -EDDA, 50 mg of EDDA was dissolved in pH 7 tampon (500 μL) in a tube. 50 $\mu\text{g}/100 \mu\text{L}$ stannous chloride solution and about 111 MBq $\text{Na}^{99m}\text{TcO}_4$ were added to the tube. The tube was stored at room temperature for 30 min. The quality control of the resulting samples was carried out using RHPLC and RTLC.

Evaluation of radiochemical purity

The radiological purity of the specimens obtained by using the above procedures was evaluated with the help of TLC on silica gel sheets (TLC-SG, Merck, Germany) and different solvent systems [(A = 1% NaCl/acetone/acetonitrile (2/1/1), citrate-dextrose buffer solution (ACD), serum physiologic (SF) and 50% acetonitrile (ACN)]. The thin sheets were scanned with BioScan TLC-scanner (Bioscan AR-2000, Washington, DC). Relative front (R_f) values and labeling efficiencies were measured from the chromatograms.

RHPLC analysis

The product ^{99m}Tc -HYNIC-tricine/EDDA-Q-Litorin was also characterized by low pressure gradient HPLC system. HPLC analysis was performed on LC-10 ATvp quaternary pump, UV detector (Shimadzu SPD-10ATvp, Macherey-Nagel, EC 250/4.6 Nucleodur 100-5 C18 column) and

20 μL loop and settled with a Cd(Te) detector equipped with a RAD-501 single channel analyzer. HPLC solvents consisted of 0.1% TFA in H_2O (solvent A) and 0.1% TFA in CH_3CN (solvent B) at a flow rate of 1 mL/min. The HPLC gradient system begins with a solvent composition of 100% A:0% B from 0 to 3 min, 50% A:50% B from 3 to 23 min, 30% A:70% B from 23 to 26 min and 100% A:0% B from 26 to 30 min. The UV detector was settled at 215 nm.

Results and discussion

Results of RTLC and RHPLC analysis

Table 1 lists the R_f values measured from RTLC analysis performed on radiolabeled compounds produced in this study using four different solvent systems. According to the results in Table 1, $^{99m}\text{TcO}_4^-$, ^{99m}Tc -tricine and ^{99m}Tc -EDDA have shown strong reaction to all four solvents, as indicated by the high R_f values. Increase in R_f value indicated migration of the samples from the origin to the front of the TLC sheet. The compound reduced ^{99m}Tc had reaction in only ACD solution. Both ^{99m}Tc -tricine-HYNIC-Q-Litorin and ^{99m}Tc -EDDA-HYNIC-Q-Litorin had lower readings for two solvent systems ACD and SF, indicating that their samples remained at the origin of the TLC sheet. When other solvents (A and 50% acetonitrile) were used, these samples migrated from origin to the front of the sheet and attained higher R_f values. Close analysis of the results in Table 1 also suggested that only ACD can differentiate successfully ^{99m}Tc labeling in both ^{99m}Tc -tricine-HYNIC-Q-Litorin and ^{99m}Tc -EDDA-HYNIC-Q-Litorin, as indicated by the lower R_f values therein.

The RHPLCs of ^{99m}Tc -tricine and ^{99m}Tc -tricine-HYNIC-Q-Litorin obtained in this study were over plotted in Fig. 2. The chromatograms of ^{99m}Tc -tricine and ^{99m}Tc -tricine-HYNIC-Q-Litorin depicted single peak for each compound with retention times (R_t) of 3.84 and 13.6 min, respectively. Similarly, when the co-ligand EDDA was used, one peak was again observed in chromatogram of

Table 1 R_f values measured from RTLC analysis performed on radiolabeled compounds using different solvent systems

	[1% NaCl/acetone/ acetonitrile (2/1/1)] = A	Citrate-dextrose buffer solution (ACD)	Serum physiologic (SF)	50% Acetonitrile
$^{99m}\text{TcO}_4^-$	0.96	0.90	0.90	0.96
Reduced ^{99m}Tc	0.04	0.92	0.05	0.06
^{99m}Tc -Tricine	0.87	0.80	0.62	0.88
^{99m}Tc -Tricine-HYNIC-Q-Litorin	0.60	0.08	0.04	0.66
^{99m}Tc -EDDA	0.90	0.88	0.85	0.88
^{99m}Tc -EDDA-HYNIC-Q-Litorin	0.69	0.10	0.07	0.71

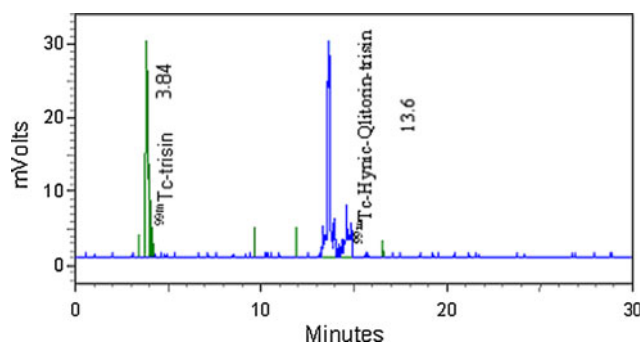


Fig. 2 RHPLC chromatograms of ^{99m}Tc -tricine and ^{99m}Tc -tricine-HYNIC-Q-Litorin

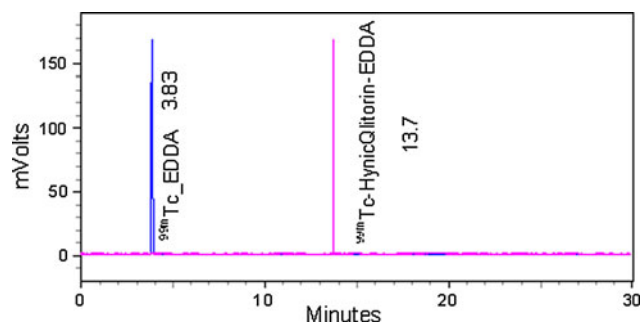


Fig. 3 RHPLC chromatograms of ^{99m}Tc -EDDA and ^{99m}Tc -EDDA-HYNIC-Q-Litorin

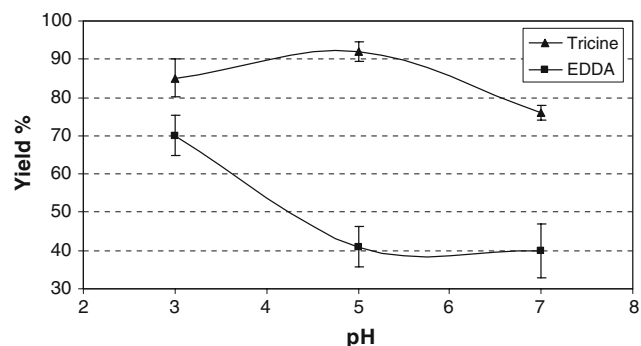


Fig. 4 Effect of pH on radiolabeling yield of ^{99m}Tc -HYNIC-Q-Litorin conjugates. Error bars denote standard deviation. Yields for tricine were obtained with tricine = 50 mg, reaction temperature = 100 °C, reaction time = 15 min, and stannous chloride = 100 μg . Yields for EDDA were similarly obtained with EDDA = 5 mg, reaction temperature = 100 °C, reaction time = 15 min, and stannous chloride = 50 μg

each compound (Fig. 3). The corresponding R_t s were 3.83 and 13.7 min, respectively. Combining the RTLC results in Table 1 with the RHPLC results in Figs. 2 and 3 together indicates a fairly high labeling yield for both ^{99m}Tc -tricine-HYNIC-Q-Litorin and ^{99m}Tc -EDDA-HYNIC-Q-Litorin.

In order to optimize the radiochemical yield, different amounts of co-ligands and stannous ion concentrations and pH values were employed, and the temperature and reaction time were manipulated during the radiolabeling

procedures. The effect of pH variation on the yield of the final product ^{99m}Tc -HYNIC-Q-Litorin was demonstrated in Fig. 4, when tricine or EDDA was used as co-ligands. The labeling yields for both showed dependence on the reaction pH when three pH values 3, 5 and 7 were employed. In the presence of tricine, the mean radiolabeling yield had parabolic behavior, meaning it measured at $84.1 \pm 5.0\%$ when pH was low at 3, then reached to a peak value of $92.0 \pm 2.7\%$ at pH of 5 and dropped back to $76.0 \pm 2.0\%$ level when pH was 7. The labeling with the EDDA conjugate, on the other hand, exhibited a monotonically declining trend with the increased pH levels. The yield was maximum at $70.1 \pm 5.2\%$ at pH = 3, but dropped to $41.0 \pm 5.0\%$ at pH = 5 and remained at $40.0 \pm 7.0\%$ when pH = 7. The literature relevant to the different construct of the ^{99m}Tc -HYNIC-peptides has reported variations in the pH value measured from the reported reaction products. In few studies carried out with HYNIC-bombesin (7,14) NH_2 was labeled with ^{99m}Tc using co-ligand tricine, pH value was reported as 7 [1]. In other studies, the pH of reaction solution was indicated as 5 [2, 15, 21]. Because of these differences in the reported pH values in the literature, we opted to examine the influence of pH on the labeling yield in the current study, and performed experiments at three different pH values 3, 5 and 7, as described above. From this extensive analysis, we determined the optimum pH as 5 for the new peptide conjugate ^{99m}Tc -tricine-HYNIC-Q-Litorin when tricine was used as the co-ligand. When EDDA was used, we found out that 3 was the optimum pH value. However, few publications reported pH value of 7 when the experiments were performed with EDDA [1, 16, 21]. From Fig. 4, the yield was $70.1 \pm 5.2\%$ at pH = 3 but the yields at pH = 5 or 7 were only about 40%. This clearly demonstrates that the previously published procedures are not consistent with our results and produces substantially less yield. However, our peptide is a new design and structurally different from the ones used in the previous publications. Therefore, this point must carefully be taken into considerations when interpreting our findings and comparing our new results with those published by others.

As we stated before, we used tricine and EDDA as co-ligands in this study. Tricine is one of the most used co-ligands in producing ^{99m}Tc -HYNIC-peptide conjugates. But the selection of the amount of tricine has been a crucial matter, affecting the radiochemical purity of the ^{99m}Tc -HYNIC-peptide. The different concentrations of tricine were used in previous publications [1, 2, 12, 13, 18, 19, 21]. The existing literature has also reported that the absence of excess co-ligand causes instability of the ^{99m}Tc -peptide conjugate and lower tricine concentration (<10 mg/mL) induces formation of ^{99m}Tc -colloid ($^{99m}\text{TcO}_2$) [1]. In radiolabeling HYNIC-BN conjugate,

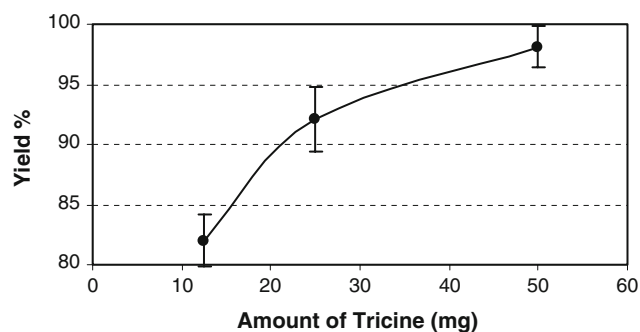


Fig. 5 Effect of the amount of tricine on the radiolabeling yield of ^{99m}Tc -HYNIC-Q-Litorin conjugates. *Error bars* denote standard deviation. Yields for tricine were obtained with pH = 5, reaction temperature = 100 °C, reaction time = 15 min, and stannous chloride = 100 μg

optimum tricine amount was determined as 20 mg and the yield was about 98% [23]. To determine if we can reach the same level of yield with lesser quantity of tricine, we investigated the radiolabeling yield by using 50 mg and also at reduced amounts of 25 and 12 mg of tricine in our experiments. The results obtained with these tricine amounts are presented in Fig. 5. From the curve in the figure, the yield can be seen as increasing monotonically with the amount of tricine and reads 82.2 ± 2.2 , 92.1 ± 2.7 and $98.1 \pm 2.7\%$ for the amounts tested. We attained maximum yield at 50 mg of tricine, which is consistent with the previous report by King et al. [18], also produced for our conjugate as well.

In the previous studies, the maximum radiolabeling yield was obtained with 5 mg of EDDA [1, 16, 18]. To test if this amount is also optimal for producing our conjugate, we have selected two more data points around this optimal value and used 3, 5 and 7 mg EDDA. Figure 6 depicts the percentage yield values associated with these amounts. The mean yield values obtained with 3 and 5 mg were close to

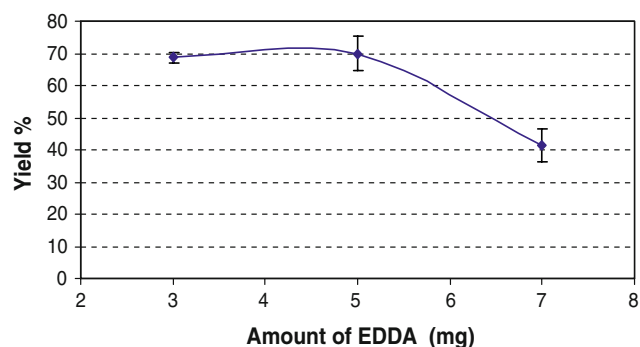


Fig. 6 Effect of the amount of EDDA on the radiolabeling yield of ^{99m}Tc -HYNIC-Q-Litorin conjugates. *Error bars* denote standard deviation. Yields for EDTA were obtained with pH = 3, reaction temperature = 100 °C, reaction time = 15 min, and stannous chloride = 50 μg

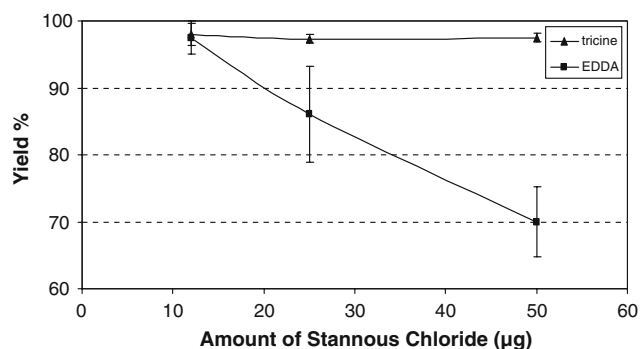


Fig. 7 Effect of stannous chloride on the radiolabeling yield of ^{99m}Tc -HYNIC-Q-Litorin conjugates. *Error bars* denote standard deviation. Yields for tricine were obtained with tricine = 50 mg, reaction temperature = 100 °C, reaction time = 15 min, and pH = 5. Yields for EDDA were similarly obtained with EDDA = 5 mg, reaction temperature = 100 °C, reaction time = 15 min, and pH = 3

each other at 68.7 ± 1.5 and $70.0 \pm 5.2\%$ (maximum yield), respectively, but it was $41.4 \pm 5.2\%$ for 7 mg of EDDA. The standard deviation for the measurement with 5 mg was lower than that of 3 mg. When this difference is considered, the results at 3 and 5 mg may not be different at a statistically significant level. The consequence of this is that the optimal EDDA amount may lie between 3 and 5 mg for our conjugates.

The effect of the amount of stannous chloride on the radiolabeling yield is summarized in Fig. 7. Initially, we used 12 μg of stannous chloride, same as the amount reported by others [1]. At this amount, we observed both tricine and EDDA produced nearly maximum yield of about 98% for both conjugates. Because of attaining nearly perfect yield at 12 μg and our determination to define the behavior of the yield with larger amounts of stannous chloride, we tested the yield efficiency at two more data points at 25 and 50 μg of stannous chloride. The data from these experiments showed that the tricine did not have significant influence on the yield of its conjugate, but the yield for the EDDA containing conjugate reduced with increase in stannous chloride amount in the medium. The reduction was monotonic and noticeable at $86.1 \pm 7.2\%$ and $70.0 \pm 5.2\%$ for 25 and 50 μg of stannous chloride, respectively.

The labeling yields for both tricine and EDDA conjugates were demonstrated for three different reaction temperatures of 25, 50 and 100 °C in Fig. 8. The corresponding yields were very close at 98.0 ± 1.7 and $97.5 \pm 2.5\%$ and maximized when the temperature was 100 °C, respectively. The effects of reaction time on the yields are shown in Fig. 9. The highest labeling yields were obtained when the time was 15 min for both conjugates. Our optimal temperature and reaction time findings are in agreement with the results from other studies carried out

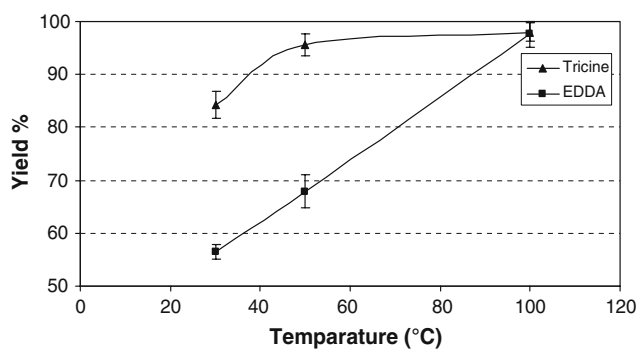


Fig. 8 Effect of temperature on the radiolabeling yield of ^{99m}Tc -HYNIC-Q-Litorin conjugates. Error bars denote standard deviation. Yields for tricine were obtained with tricine = 50 mg, stannous chloride = 12 μg , reaction time = 15 min, and pH = 5. Yields for EDDA were similarly obtained with EDDA = 5 mg, stannous chloride = 12 μg , reaction time = 15 min, and pH = 3

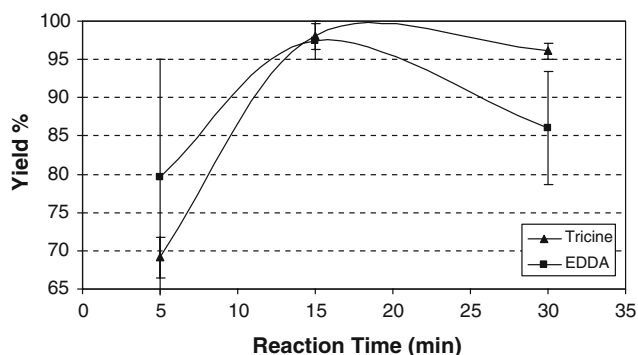


Fig. 9 Effect of reaction time on the radiolabeling yield of ^{99m}Tc -HYNIC-Q-Litorin conjugates. Error bars denote standard deviation. Yields for tricine were obtained with tricine = 50 mg, stannous chloride = 12 μg , reaction temperature = 100 $^{\circ}\text{C}$, and pH = 5. Yields for EDDA were similarly obtained with EDDA = 5 mg, stannous chloride = 12 μg , reaction temperature = 100 $^{\circ}\text{C}$, and pH = 3

with BNN-like other peptides conjugated to ^{99m}Tc with co-ligand tricine or EDDA [1, 2, 15, 16, 18]. It is important to note that the optimal temperature and reaction time depends on underlying peptide and in some cases, maximum yield can be obtained not necessarily at 100 $^{\circ}\text{C}$, but at room temperature [16].

In above sections, we showed that the radiolabeling efficiencies of ^{99m}Tc -tricine-HYNIC-Q-Litorin and ^{99m}Tc -EDDA-HYNIC-Q-Litorin were 98.0 ± 1.7 and $97.5 \pm 2.5\%$, respectively. Liu et al. suggested that EDDA is a potentially tetradentate ligand and therefore is expected to be more stable ^{99m}Tc -conjugate than that of tricine [21]. The higher symmetry associated with EDDA conjugate is suggested as the reason for stability because it results in fewer coordination isomers than those obtained with tricine. However, our findings indicated no remarkable differences in the radiolabeling yields. This is not a surprising

finding because similar observations were made with different peptide conjugates including tricine and EDDA in the past [1].

Conclusion

In this study, we demonstrated that it is possible to produce litorin based radiolabeling agents ^{99m}Tc -tricine-HYNIC-Q-Litorin and ^{99m}Tc -EDDA-HYNIC-Q-Litorin at high yields. The optimal conditions for ^{99m}Tc -tricine-HYNIC-Q-Litorin are: HYNIC-peptide:tricine: 10 $\mu\text{g}/50$ mg, pH 5, SnCl_2 concentration: 12 $\mu\text{g}/0.1$ mL, reaction temperature: 100 $^{\circ}\text{C}$, reaction time: 15 min. The corresponding values for ^{99m}Tc -EDDA-HYNIC-Q-Litorin are: HYNIC-peptide:EDDA: 10 $\mu\text{g}/5$ mg, pH 3, SnCl_2 concentration: 12 $\mu\text{g}/0.1$ mL, reaction temperature: 100 $^{\circ}\text{C}$, reaction time: 15 min. It remains to the future work to investigate the biological activity profiles of these conjugates for imaging GRPr expressing tumor cells.

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