#### **ORIGINAL PAPER**



# Composite nanoarchitectonics of poly(vinylidene fluoride)/graphene for thermal and electrical conductivity enhancement via constructing segregated network structure

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#### Abstract

In this work, a facial approach was proposed to improve the thermal and electrical conductivity of poly(vinylidene fluoride) (PVDF) by constructing a segregated graphene nanoplatelets (GNP) network structure. PVDF/GNP composites with segregated filler network were fabricated by coating GNP on the surface of PVDF powders using ball milling treatment, followed by compression molding technique. PVDF powders of different particle size (e.g., 100 and 400  $\mu$ m) were employed to elucidate the influence of matrix particle size on the properties of PVDF/GNP composites. Results showed that the formation of segregated GNP network contributed to a concurrent enhancement of both thermal and electrical conductivity of corresponding composites. A thermal conductivity as high as 2.23 W/mK coupling with an electrical conductivity of about 0.2 S/m was achieved for segregated PVDF<sub>100</sub>/GNP 6 wt% composite, which demonstrates potential applications in the areas that require both high electric and heat dissipations.

**Keywords** Poly(vinylidene fluoride)  $\cdot$  Graphene nanoplatelet  $\cdot$  Thermal conductivity  $\cdot$  Electrical conductivity  $\cdot$  Segregated network

# Introduction

There is growing interest of metal-replacement using polymer composites in the areas of automotive, aerospace, electronics among others due to their easy processability, tailorable properties, lightweight and excellent resistance to corrosive conditions [1–6]. The intrinsic thermal conductivity of pure polymers is quite low (0.1 ~ 0.5 W/mK) and a certain amount ( $\geq$  30 vol%) of thermal conductive fillers are always required to enhance the thermal conductivity to meet practical application requirements [7–9]. A consensus has been acknowledged that both the formation of three-dimensional thermal conductive pathways and the increase of filler packing density are crucial to the overall enhancement of

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<sup>1</sup> The State Key Laboratory of Polymer Materials Engineering, Polymer Research Institute of Sichuan University, Chengdu 610065, China thermal conductivity for filler-containing polymers composites [10]. It also becomes inevitable to deteriorate both the processability and mechanical properties of polymers due to the formation of filler agglomerates. As a result, it becomes critical to construct thermally conductive pathways within polymer matrices at lower filler concentrations.

A variety of approaches have been proposed to enhance the thermal conductivity of polymer composites [11, 12]. For example, surface modification of fillers is considered as an effective way to reduce the interfacial thermal resistance between host matrix and the added fillers, thereby leading to an increase of thermal conductivity for subsequent composites. Wie et al. [13] reported the thermal conductivity of surface modified boron nitride (BN) loaded epoxy composites reached 11.8 W/mK at a filler concentration of 75 wt%. The enhanced interaction between polymer and BN was considered as the contributing factor that improves the thermal conductivity. A core-shell SiC@GNP structure consisting of silicone carbide (SiC) and graphene (GNP) was proposed by Wang et al. [14], and the thermal conductivity of epoxy composite with 20 wt% core-shell SiC@GNP reached 1.84 W/ mK, which is more than 8 times higher than pure epoxy.

Recently, researchers proposed that constructing novel nanostructures is a more appropriate method which can reduce surface thermal resistance and enhance thermal conductivity of polymer composites [15-22]. The closer distance between fillers can reduce phonon scattering at interfaces between adjacent fillers. Zhang et al. [23] reported that the thermal conductivity of polypropylene/ alumina (PP/Al<sub>2</sub>O<sub>3</sub>) composites reached 1.07 W/mK at 27.5 vol% Al<sub>2</sub>O<sub>3</sub> by constructing continuous segregated filler network structure. The selective localization of Al<sub>2</sub>O<sub>3</sub> to form continuous thermal conductive pathways is believed to be the contributing factor. Li et al. [24] prepared poly(3-hydroxylbutyrate)/BN composites using powder mixing method and the obtained composites exhibited higher thermal conductivity when compared with the samples prepared by solution mixing. The formation of segregated BN network is thought to be crucial for the enhancement of thermal conductivity by reducing thermal boundary resistance between adjacent fillers. Moreover, Wang et al. [25] found that PVDF composites with segregated (BN + CNT) hybrid fillers exhibited a thermal conductivity of 1.8 W/mK at a total filler concentration of 25 vol%, which was 169% higher than that of PVDF/BN/CNT counterparts with random filler distribution state. The above finding was ascribed to the effective utilization of fillers by constructing segregated structure and the reduction of interfacial thermal resistance between adjacent BN particles via the bridging effect of using high aspect ratio CNT.

In the present study, a series of poly(vinylidene fluoride) (PVDF) composites were prepared with graphene nanoplatelets (GNP) being adopted as the thermal conductive fillers. Ball milling treatment method was employed to prepare GNP@PVDF hybrids which were further adopted to prepare PVDF-based composites with segregated filler structure. The effect of constructing segregated GNP network on the properties such as morphology, thermal and electrical conductivity, rheological behavior and thermal stability of corresponding composites were studied in detail.

## **Experimental section**

### Materials

Poly(vinylidene fluoride) (PVDF) powders of different particle size (e.g., 100 and 400  $\mu$ m) under the trademark of FR906 were provided by Shanghai Huayi 3F Co., Ltd (China). Graphene nanoplatelets (GNPs) with an average size of 7 ~ 12  $\mu$ m were purchased from Shenzhen Tulingjinhua Technology Co., Ltd (China). All materials were used as-received without further treatment.

#### Sample preparation

PVDF composites with segregated GNP network were prepared as per the following procedure: firstly, powder form PVDF and a predetermined amount of GNP (0, 1, 2, 4 and 6 wt%) were treated using an omnidirectional planetary ball mill. The weighed PVDF and GNP powders were placed in an agate tank with the presence of 5 mm agate balls (the mass ratio of grinding ball to powder is 100:1). The duration of ball milling process lasted 1 h and the rotational speed was set at 500 rpm. Then, the obtained GNP@PVDF hybrids were hot pressed at 200 °C and 10 MPa for 10 min. Samples prepared with 100 µm PVDF particles were termed at s-PVDF<sub>100</sub>/GNP-x, where s- indicated samples with segregated GNP network and x referred to the weight fraction of GNP in respective samples. The same nomenclature rules applied to samples prepared using 400 µm PVDF particles. The schematic diagram illustrating the preparation of s-PVDF/GNP composites is displayed in Scheme 1.

In addition, melt blending of PVDF particles with GNP was conducted to prepare PVDF/GNP composites with discretely distributed GNP, which were termed as m-PVDF/GNP. The concentration of GNP in melt blended PVDF/GNP composites was the same to that of s-PVDF/GNP counterparts.

## Characterization

The morphology of PVDF (particle size was about 400  $\mu$ m) before and after coating GNP was observed using scanning electron microscopy (SEM, JSM-7800F, JEOL, Japan). The morphology of GNP was observed using transmission electron microscopy (TEM, JEM-2100, JEOL, Japan) at 200.0 kV. The segregated structure was observed and confirmed using a polarized optical microscope (POM, DM2700P, Leica, Germany). The thermal conductivity of all samples was measured by a thermal constant analyzer (TPS2200, Hot-Disk, Sweden). Electrical conductivity of was tested by a Keithley multimeter system (DAQ6510, Keithley, USA). The thermal stability of all samples was measured by a simultaneous thermogravimetric analyzer (TGA/DSC3 + , Mettler Toledo, Switzerland) at a heating rate of 10 °C/min in N<sub>2</sub> atmosphere.

## **Results and discussion**

Figure 1a, b demonstrated the morphology of PVDF particles (about 400  $\mu$ m) and GNP, respectively. After ball milling, the surface of PVDF became rougher due to the coating of GNP, as displayed in Fig. 1c, d. The



Scheme 1 Preparation of PVDF-based composites with segregated GNP network

morphology of s-PVDF/GNP composites prepared with different PVDF particle sizes was investigated by optical microscope. Figure 1e, f showed that GNP particles which were selectively localized at the interfaces were observed for composites prepared with both PVDF particle sizes. Under such circumstances, the formed segregated structure facilitates the construction of intact thermal conductive pathways, which is crucial to improving the thermal conductivity for subsequent composites.

The thermal conductivity of PVDF/GNP composites prepared by different methods was shown in Fig. 2. Figure 2 showed that pure PVDF exhibited inferior thermal conductivity (0.24 W/mK) at room temperature. There is obvious increase of thermal conductivity with an incremental loading content of GNP. The thermal conductivity of PVDF/ GNP 6 wt% composites that prepared by melting mixing reached 1.46 W/mK, which is 6.1 times higher than that of neat PVDF. The above improvement was related to the overlapping of GNP within the host matrix, which is beneficial to the formation of intact thermal conductive pathways and therefore improve the thermal conductivity of corresponding composites.

As expected, the thermal conductivity of PVDF/GNP composites with segregated filler network exhibited much higher values when compared with the melt blended PVDF/ GNP counterparts. For example, the thermal conductivity of s-PVDF<sub>400</sub>/GNP 6 wt% composites reached 1.76 W/mK which is 20.5% higher than melt blended PVDF/GNP counterparts. The above observation is undoubtedly related to the formation of thermal conductive network arising from the formation of segregated GNP network. However, samples prepared using melt blending method exhibited random filler distribution, and intact thermal conductive network pathways were hardly formed at lower GNP concentrations. In this scenario, the



Fig. 1 (a) SEM image of PVDF400 particles; (b) TEM image of GNP; (c) and (d) were images of GNP@PVDF hybrid which was prepared by ball milling; (e) and (f)

thermal conductivity of m-PVDF/GNP composites was lower than that of s-PVDF/GNP counterparts. In addition, samples with segregated filler network that prepared from different particle sizes of PVDF showed different increasing trend of thermal conductivity, especially when the content of GNP is above 2 wt% (as given in Fig. 2). Figure 1c, d showed that the GNP could be coated on the surface of PVDF particles with the aid of ball milling treatment, and the PVDF particles with 100  $\mu$ m are believed to exhibit higher specific surface area which is beneficial to the attachment of GNP during powder mixing. This led to the formation of denser heat conductive pathways in s-PVDF<sub>100</sub>/GNP composite. According to literature



Fig. 2 Thermal conductivity of PVDF/GNP composites as a function of filler content and preparation method

[2], both the formation of thermal conductive pathways and increasing filler packing density were favorable for improving thermal conductivity of filler-containing polymer composites. Thus, a denser and more perfect thermal conductive network is likely constructed in terms of samples prepared with small size PVDF particles, thereby leading to a significant improvement of thermal conductivity for s-PVDF<sub>100</sub>/GNP composites when compared with s-PVDF<sub>400</sub>/GNP counterparts with increasing filler concentrations.

The electrical conductivity of PVDF/GNP composites as a function of filler content was given in Fig. 3. The neat PVDF is electrically insulative [9, 26] with an electrical



Fig. 3 Electrical conductivity of PVDF/GNP composites as a function of filler content and preparation method

conductivity of about  $1.5 \times 10^{-14}$  S/m. In spite of the preparation method, the electrical conductivity of PVDF/GNP composites increased monotonically with increasing GNP concentration. For example, the electrical conductivity of melt blended PVDF/GNP 6 wt% composites reached 0.01 S/m, which is nearly 12 orders of magnitude higher than that of pure PVDF.

It is known [27–29] that insulative polymers will become electrically conductive once the added fillers attain an intact physical network within the host matrix. Figure 3 clearly exhibited that the electrical conductivity of s-PVDF/GNP composites is invariably higher than the melt blended counterparts. In addition, the electrical conductivity of s-PVDF<sub>400</sub>/GNP composites is higher than that of s-PVDF<sub>100</sub>/GNP composites when the content of GNP is lower than 4 wt%. For example, the electrical conductivity of s-PVDF<sub>100</sub>/GNP 1 wt% is  $5.1 \times 10^{-12}$  S/m, which is about 4 orders of magnitude lower than that of s-PVDF<sub>400</sub>/GNP 1 wt% counterpart. This indicated that the formation of more intact electrically conductive network is favored when larger size PVDF particles were used to prepare s-PVDF/ GNP composites. According to the percolation theory [30], the electrical conductivity of polymer composites would experience a distinct increase when the added fillers attain a three-dimensional network structure. More intact conductive pathways are expected to be formed with further increasing GNP content. As a result, the difference of electrical conductivity between s-PVDF100/GNP and s-PVDF400/GNP composites is minimized when the concentration of GNP is no less than 4 wt%.

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 Table 1
 Characteristic thermal analysis data of pure PVDF and PVDF/GNP composites that abstracted from TGA analysis

Sample	T <sub>5</sub> (°C)	T <sub>30</sub> (°C)	HRI (°C)	$T_{p}(^{o}C)$
PVDF	424.6	461.3	218.8	467.0
m-PVDF/GNP 6 wt%	462.5	480.5	231.9	483.0
s-PVDF <sub>100</sub> /GNP 6 wt%	455.9	474.7	228.9	478.6
s-PVDF <sub>400</sub> /GNP 6 wt%	444.0	468.9	224.9	472.5

HRI= $0.49*[T_5+0.6*(T_{30}-T_5)]$  was calculated as per [33, 34]

The thermal stability of pure PVDF and PVDF/GNP 6 wt% composites that prepared by different methods was evaluated using TGA and corresponding results for respective samples were displayed in Fig. 4 and Table 1. Results indicated that the main pyrolysis process occurred in the temperature range of 400~500 °C, which is believed related to the degradation of chain backbone of PVDF [31, 32]. The thermal decomposition temperatures (i.e.,  $T_5$  and  $T_{30}$ ) are the temperatures at which 5 and 30 wt% mass loss were recorded), heat-resistance index (HRI) [33, 34] and peak decomposition temperatures (T<sub>n</sub>, the maximum mass loss temperature that determined from DTG curves) of all samples were listed in Table 1. Results showed that the  $T_5$ , HRI and  $T_p$  of pure PVDF are 424.6, 218.8 and 467.0 °C. Figure 4 and Table 1 revealed that the thermal stability of PVDF improved with the incorporation of GNP. Additionally, the melt blended PVDF/GNP composites exhibited the highest  $T_5$ , HRI and  $T_p$ , suggesting that the randomly distributed GNP particles were favorable for the enhancement



Fig. 4 Thermogravimetric analysis (TGA) and differential thermogravimetric (DTG) analysis of pure PVDF and PVDF/GNP composites that prepared by different methods



Fig. 5 The storage modulus (G') and loss modulus (G'') of PVDF and PVDF/GNP composites that prepared by different methods

of thermal stability of PVDF. This could be explained by the following aspects: (1) the presence of planar GNP particles would, to some extent, barrier the volatilization of decomposed fragments due to the "tortuous effect", which improved the thermal stability of polymer composites [35, 36]; (2) the GNP particles acted as radical scavengers which might suppress the chain transfer reaction and thus prevent the degradation of polymer chains [37].

Figure 4 also showed that the thermal stability of PVDF/ GNP composites followed a sequence of m-PVDF/GNP 6 wt% > s-PVDF<sub>100</sub>/GNP-6 > s-PVDF<sub>400</sub>/GNP-6. The elevated thermal stability of melt blended PVDF/GNP composites is ascribed to the random distribution of GNP within the host matrix, as described previously. However, it should be noticed that either the "tortuous effect" or "scavenge effect for free radicals" would be somewhat compromised in terms of PVDF/GNP samples with segregated filler structure, which leads to a reduced thermal stability for s-PVDF/ GNP composites. Moreover, smaller size PVDF particles were more likely to be fully covered with GNP particles during the ball milling process. Under such circumstances, s-PVDF<sub>100</sub>/GNP-6 composites exhibited relatively higher thermal stability as compared with that of s-PVDF<sub>400</sub>/GNP-6 counterparts.

Rheological analysis is considered as a reliable way to characterize the relationship between the internal microstructure and properties of polymer composites. Figure 5 demonstrated the storage modulus (G') and loss modulus (G") of PVDF/GNP composites with different filler content and preparation method. It is clear that the values of G' and G" increased with increasing GNP content, especially in the low frequency region. The presence of GNP particles restricted free movement of polymer chains, which resulted in an enhancement of subsequent composites. It is worth mentioning that both s-PVDF<sub>100</sub>/GNP and s-PVDF<sub>400</sub>/GNP composites exhibited lower G' when compared with that of melt blended PVDF/GNP counterparts, which is likely associated with the state of distribution of GNP in corresponding samples. It is believed that GNP particles were randomly distributed within melt-blended PVDF composites, which could limit free mobility of polymer chains. However, the selective localization of GNP between adjacent PVDF particles had milder chain confinement effect on the polymer chains. As a result, a relatively lower reinforcement effect was observed for s-PVDF/GNP composites.

# Conclusion

In this study, we proposed a feasible method to fabricate thermally and electrically conductive poly(vinylidene fluoride)/graphene (PVDF/GNP) composites with segregated filler network structure. PVDF particles of different size (i.e., 100 and 400 µm) were employed to prepare GNP@ PVDF hybrids using ball milling treatment, followed by hot pressing. Regardless of the size of PVDF particles, composites prepared with segregated filler structure exhibited much higher thermal and electrical conductivity than those prepared by melt mixing. However, the size of PVDF particles seemed to play a crucial role in building thermally and electrically conductive pathways. The thermal conductivity of samples prepared with smaller PVDF particles was higher when compared with the other composites, which was related to the construction of a much denser and more perfect thermal conductive network. For example, the thermal conductivity of s-PVDF<sub>100</sub>/GNP 6 wt% composite reached 2.23 W/mK, which is nearly 829% and 52.7% higher than pure PVDF and m-PVDF/GNP 6 wt% counterparts, respectively.

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## Declarations

**Conflict of interest** The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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