

Effect of BN-Added Precursors on Phase Formation and Transport Properties of (Bi,Pb)-2223 HTS

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Abstract (Bi,Pb)-2223 HTSs (high temperature superconductors) were synthesized from nominally pure (reference) and BN-added $\text{Bi}_{1.7}\text{Pb}_{0.3}\text{Ca}_2\text{Sr}_2\text{Cu}_3\text{O}_y(\text{BN})_x$ precursors ($x = 0, 0.10, 0.15,$ and 0.20) by the solid state reaction method using alumina crucibles. The influence of boron nitride addition on the phase formation kinetics and transport properties of (Bi,Pb)-2223 HTSs was studied using X-ray diffraction (XRD), resistivity and critical current density measurements. BN-added compounds reveal a significant enhancement in both the high- T_c 2223 phase formation and critical current density compared to the reference specimen.

Keywords (Bi,Pb)-2223 phase · BN additive · Critical temperature · Critical current density

1 Introduction

The synthesis of a nearly pure single-phase high- T_c Bi-2223 ($T_c \approx 110$ K) is a critical issue in fabrication of Bi-based superconductors. Partial substitution of Bi by Pb is the most widely used method to enhance the formation of the Bi-2223 phase [1]. Since the kinetics of the Bi-2223 phase formation is slow, it requires a long heating (150–200 h) to produce an appropriate fraction of Bi-2223 HTS material in the final product [2]. Moreover, extremely narrow stability range of 2223 results in the formation of intergrowth of the multiple phases coexisting with the 2223 phase [3]. It was also found that the high- T_c 2223 phase is very sensitive to the partial pressure of oxygen and sintering in O_2 as well as in ambient atmosphere suppress the formation of this HTS [4]. Thus, many factors including composition of the precursor powder, synthesis atmosphere, heat treatment conditions, and nature of doped ions significantly influence the final physical properties of the samples [5].

Our previous work shows that the precursors doped with lead borate $\text{Pb}(\text{BO}_2)_2$ and boron oxide B_2O_3 substantially promote the formation of (Bi,Pb)-2223 phase when their thermal processing occurs in alumina crucibles and lead to the enhancement of transport properties compared to the reference sample [6]. The purpose of the present paper is to investigate whether the BN additive has a similar positive effect on the superconducting properties of (Bi,Pb)-2223 HTS ceramics.

2 Experimental

Samples with nominal composition $\text{Bi}_{1.7}\text{Pb}_{0.3}\text{Ca}_2\text{Sr}_2\text{Cu}_3\text{O}_y(\text{BN})_x$ ($x = 0, 0.10, 0.15,$ and 0.20) have been prepared by

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the solid state reaction method from the appropriate mixtures of Bi_2O_3 , PbO , SrCO_3 , CaCO_3 , CuO , and BN . The mixtures were thoroughly ground and heat treated at 850°C for 30 h in alumina crucibles with intermediate grindings. The resulting materials were pressed into pellets of 10 mm diameter and 1.5 mm thickness under hydrostatic pressure at 29 MPa. The pellets were placed in alumina crucibles, then annealed at 840°C for 30 h in air followed by the furnace cooling to room temperature. Synthesized compounds were characterized by powder X-ray diffraction (XRD) analysis using the Dron-3M diffractometer (CuK_α radiation) and the fractional amounts of high- T_c phase were estimated from the XRD data. The resistivity as a function of temperature, $\rho(t)$, and transport critical current density, J_c , were measured by a standard four-probe method for bar-shaped specimens ($\sim 10 \times 0.5 \times 0.5 \text{ mm}^3$) cut from the pellets. The critical current densities were evaluated at the liquid nitrogen temperature in the self-field, with a criterion of $1 \mu\text{V}/\text{cm}$.

3 Results and Discussion

Our earlier investigations show that the rate of high- T_c phase formation, and hence, the value of zero resistivity temperature (T_c^{off}) are extremely sensitive to the thickness (height) of the nominally pure (un-doped) precursors when their thermal processing occurs for ~ 50 – 60 h in alumina crucibles or on alumina plates. Figure 1 illustrates the effect of thickness (d) of undoped (Bi,Pb)-2223 precursors on the superconducting transition. It is evident that the T_c^{off} value decreases with increasing d . Based on the experimental results of our group, $T_c^{\text{off}}(d)$ dependence can be subdivided in three

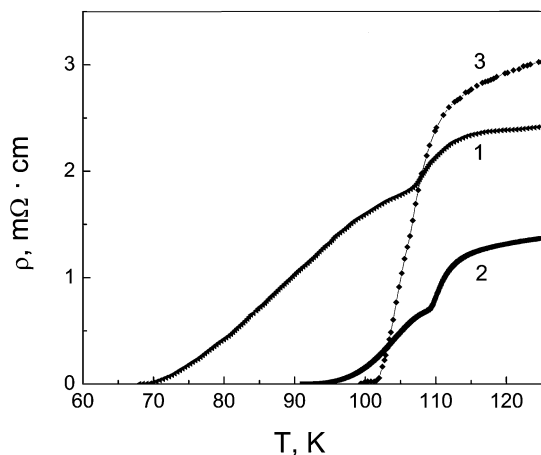


Fig. 1 Superconducting transition as a function of thickness of undoped $\text{Bi}_{1.7}\text{Pb}_{0.3}\text{Ca}_2\text{Sr}_2\text{Cu}_3\text{O}_y$ sintered at 840 – 850°C . 1—heat treatment of powder ($d \approx 12$ mm, 30 h) and pellet (~ 1.5 mm thick, 30 h) in alumina crucible; 2—heat treatment of powder ($d \approx 7$ mm, 30 h) and pellet (~ 1.5 mm thick, 30 h) in alumina crucible; 3—heat treatment of powder ($d \approx 2$ mm, 30 h) and pellet (~ 1.5 mm thick, 30 h) on alumina plate

main areas: (i) T_c^{off} is about 65 – 75 K for $d \approx 12$ – 15 mm, (ii) $T_c^{\text{off}} \approx 85$ – 95 K for $d \approx 6$ – 8 mm, and (iii) $T_c^{\text{off}} > 100$ K for $d \leq 3$ mm. We also found that in contrast to the undoped (Bi,Pb)-2223 samples, the superconducting transition of boron-doped (Bi,Pb)-2223 material does not correlate noticeably with the d values ranging from 2 to 15 mm. The T_c^{off} higher than 100 K were obtained in (Bi,Pb)-2223 HTSs doped with lead borate $\text{Pb}(\text{BO}_2)_2$ and boron oxide B_2O_3 [6]. In the present study, BN-added (Bi,Pb)-2223 precursors were ~ 12 mm in thickness. XRD patterns of the reference and BN-added fully processed specimens are presented in Fig. 2. The dominance of the low- T_c 2212 phase over the high- T_c 2223 phase was observed in the reference and low-content BN doped ($x = 0.10$) samples. The lattice parameters of (Bi,Pb)-2212 phase derived from XRD data are $a = 5.428 \text{ \AA}$, $b = 5.430 \text{ \AA}$, $c = 31.040 \text{ \AA}$ for undoped specimen and $a = 5.428 \text{ \AA}$, $b = 5.407 \text{ \AA}$, $c = 30.936 \text{ \AA}$ for BN-doped sample with $x = 0.10$. Obviously, both samples have an orthorhombic structure. The lattice parameter “ a ” remains unchanged, but the “ b ” and “ c ” parameters decrease with BN addition. This corresponds to the contraction of about 0.8 % of the unit cell volume. This result suggests that due to their extremely small ionic radius B^{3+} ions may occupy substitutional or interstitial positions in the lattice. With an increase of the BN content up to $x = 0.15$, the 2223 phase is remarkably enhanced and its increase is associated with the decrease of 2212 phase; although, a very low- T_c 2201 phase appears for $x = 0.10$ and is intensified with in-

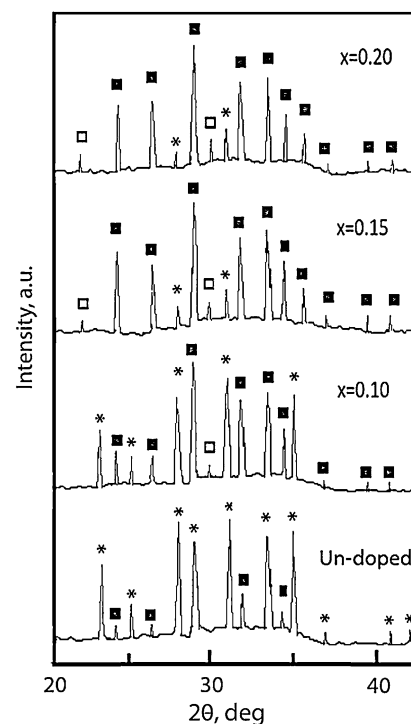


Fig. 2 XRD patterns of $\text{Bi}_{1.7}\text{Pb}_{0.3}\text{Ca}_2\text{Sr}_2\text{Cu}_3\text{O}_y(\text{BN})_x$ samples with various BN amounts. ■—2223 phase, *—2212 phase, □—2201 phase

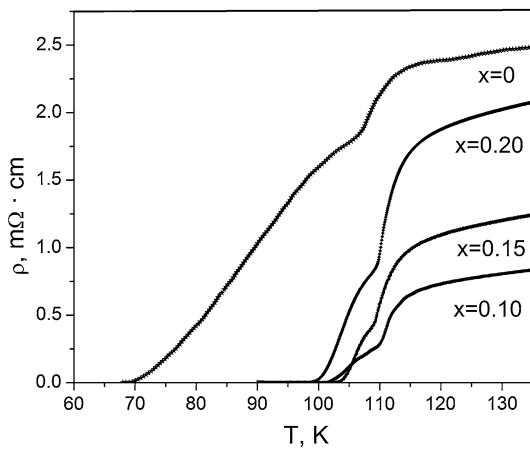


Fig. 3 Resistivity versus temperature for $\text{Bi}_{1.7}\text{Pb}_{0.3}\text{Ca}_2\text{Sr}_2\text{Cu}_3\text{O}_y$ $(\text{BN})_x$ samples

creasing BN concentration. The peaks corresponding to the BN could not be observed in the XRD patterns due to the minute amounts (less than 0.5 wt%) of this additive. It is observed for $x = 0.15$ composition that all dominant peaks belong to the orthorhombic (Bi,Pb)-2223 phase with the lattice parameters of $a = 5.428 \text{ \AA}$, $b = 5.482 \text{ \AA}$, $c = 37.330 \text{ \AA}$. A further addition of BN up to $x = 0.20$ shows no systematic variation in the lattice parameters with respect to the $x = 0.15$ sample. This might be attributed to the distortion of crystal lattice. The volume fractions (V) of the (Bi,Pb)-2223 phase were estimated from XRD intensity ratios of the (Bi,Pb)-2223 and Bi-2212 phases using the following equation [7]: $V = \{I_{H(115)} / [I_{H(115)} + I_{L(115)}]\} \times 100$ [%], where $I_{H(115)}$ and $I_{L(115)}$ denote the intensities of the (115) peaks of the (Bi,Pb)-2223 and (Bi,Pb)-2212 phases, respectively. In this case, the very small amount of Bi-2201 phase detected on the XRD patterns was ignored. The calculated volume fraction of (Bi,Pb)-2223 phase increases from ~ 10 % for undoped specimen to ~ 25 % for $x = 0.1$ and reaches the maximum value of ~ 75 % for $x = 0.15$ – 0.20 in a short sintering time (60 h), which indicates that BN addition accelerates the solid state reaction rate and hence the (Bi,Pb)-2223 formation. Figure 3 represents the temperature dependence of resistivity $\rho(T)$ for the reference and BN-added (Bi,Pb)-2223 samples. The $\rho(T)$ dependences clearly exhibit a two-step transition, reflecting the coexistence of high- T_c 2223 and low- T_c 2212 phases. Splitting of the resistive transition (double superconducting transition typical of a two-phase sample) might also be due to the presence of oxygen content inhomogeneities in the HTS [8]. Onset temperature T_c^{on} of the superconducting transition is about 115 K for all the samples. For the reference specimen zero resistivity is reached at $T_c^{\text{off}} = 70$ K. T_c^{off} increases up to 103 K with increasing x from zero to 0.15, then drops to 98 K at $x = 0.20$. These curves show that the sample with $x = 0.15$ is the best in the present study. It is also seen that the nor-

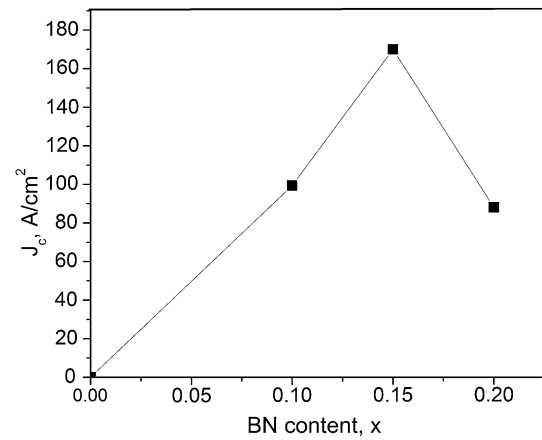


Fig. 4 Relationship between critical current density and BN content

mal state resistivity decreases with the introduction of BN and then increases with increasing doping level. This behavior may be related to the appearance and increase of very low- T_c Bi-2201 phase upon increasing the BN concentration. Presence of 2201 phase in the doped compositions deteriorates a coupling between superconducting grain boundaries which results in the reduction of T_c^{off} at $x > 0.15$. Figure 4 reveals the relationship between the transport J_c values (77 K, zero field) and BN content. The enhancement of critical current density by increasing BN content seems to result from the increase of (Bi,Pb)-2223 phase fraction with increasing x from zero to 0.15. After passing the maximum value at $x = 0.15$ (170 A/cm²), J_c decreases for higher doping level. In agreement with the XRD and resistivity data shown in Figs. 2 and 3, lower J_c value for $x = 0.20$ (88 A/cm²) imply the deterioration of intergranular coupling due to slight increase of 2201 phase with increasing BN content. Maximum value of J_c for BN-added (Bi,Pb)-2223 is lower as compared to the $\text{Pb}(\text{BO}_2)_2$ and B_2O_3 -doped specimens (215 A/cm² and 190 A/cm², respectively [6]). Incorporation of low-melting $\text{Pb}(\text{BO}_2)_2$ and B_2O_3 ($T_m = 500$ °C and 450 °C, respectively) into (Bi,Pb)-2223 precursors accelerates the (Bi,Pb)-2223 phase growth [6]. In the present study, similar positive effect of BN additive on the phase formation and superconducting properties of (Bi,Pb)-2223 HTS was observed. BN starts to oxidize at 700 °C, but the rate is low up to 1000 °C, after which oxidation becomes a major problem [9]. Oxidation reaction leads to the decomposition of BN to B_2O_3 and N_2 [10, 11]. Based on the obtained results, we assume that the appearance of boron-containing liquid phase during the annealing of BN-doped $\text{Bi}_{1.7}\text{Pb}_{0.3}\text{Ca}_2\text{Sr}_2\text{Cu}_3\text{O}_y$ $(\text{BN})_x$ precursor powders at 840–850 °C allows the enhancement of high- T_c phase formation due to a higher diffusion rate of elements required to form the (Bi,Pb)-2223 phase.

4 Conclusion

Phase evolution and transport properties of BN-added (Bi,Pb)-2223 HTSs prepared by the heat treatment of $\text{Bi}_{1.7}\text{Pb}_{0.3}\text{Ca}_2\text{Sr}_2\text{Cu}_3\text{O}_y(\text{BN})_x$ ($x = 0\text{--}0.2$) precursors in alumina crucibles were investigated. Addition of boron nitride accelerates the formation of high- T_c (Bi,Pb)-2223 phase and leads to the increase in the critical current density compared to the reference sample. We could conclude that choosing the optimum doping level it could be possible to prepare BN-added (Bi,Pb)-2223 HTS materials with even higher fraction of the high- T_c phase as well as with much better critical current density in a short heating time of $\sim 50\text{--}60$ h.

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