

Functionalize Aramid Fibers with Polydopamine to Possess UV Resistance

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Abstract

The paper designs a novel and green method to improve the surface activity and UV resistance of aramid fbers. The fbers were treated by UV radiation and then functionalized with dopamine, reaching the result that a dense and uniform polydopamine (PDA) was coated on the surface of the fbers. The surface morphology, chemical structure, aggregation structure and wettability, UV resistance and mechanical properties of the aramid fbers were characterized by SEM, AFM, ATR-FTIR, XPS, XRD, UV absorption spectroscopy, contact angle and mechanical properties tests. The results showed that the dopamine was successfully coated on the surface of the fbers, and the chemical structure of the surface of the functionalized fber changed, but the aggregation structure did not change. The surface activity and UV resistance of the functionalized fber signifcantly improved. Compared with untreated aramid fbers, after the fbers were radiated under UV treatment for one week, the retention of strength and elongation at break of the functionalized fber with dopamine increased nearly 20%, respectively. These results demonstrated that PDA layer could signifcantly improve the surface properties of aramid fbers and play a key role in protecting the cortical structure of the fbers. So far, the researches on improving the surface activity and UV resistance of fbers by functionalization with dopamine have rarely been reported, so the method is novel and worth for further studying.

Keywords Aramid fber · Dopamine · Functionalization · Surface activity · UV resistance

1 Introduction

Aramid fber is one of the organic high-performance fbers. Aramid fber combines high strength, thermal stability and chemical stability with low density $[1-4]$ $[1-4]$. Due to its unique excellent properties, it is widely used as the reinforcement material in the advanced polymer composites [[5](#page-13-2)[–8\]](#page-13-3). Furthermore, aramid fber reinforced polymer composites were applied in military applications, aerospace, automotive, electronic products, industry and other felds [\[9](#page-13-4)[–13\]](#page-13-5), and were obtained the wide attention from researchers.

However, the aramid fiber has poor surface activity [[5,](#page-13-2) [14](#page-13-6)[–16\]](#page-13-7) and UV resistance [[17](#page-13-8)[–20\]](#page-13-9), which severely limits the applications of the aramid fber. Solving these two bottlenecks and maintaining the excellent mechanical properties of the aramid fber are a huge challenge, and the researchers have conducted a lot of researches on this. Zhang [[21](#page-13-10)] prepared grafted Kevlar fbers (HSi-g-KF) with improving surface activity and UV resistance by in-situ synthesis of hyperbranched polysiloxanes with double bonds and epoxy groups on the Kevlar fbers. Zhou [\[17](#page-13-8)] improved the surface activity and UV resistance of the fbers with green layer-bylayer (LBL) self-assembly technology by alternately selfassembling $SiO₂$ and MgAlFe layered double hydroxide (LDH) on the surface of the aramid fber. Cai [[22](#page-13-11)] synthesized a UV absorber with strong UV absorption and high surface activity by sequentially forming nano-layered boron nitride (tBN) and PDA layer on the CeO₂ core (PDA@tBN@ $CeO₂$). Zhu [\[23](#page-13-12)] prepared a novel surface-modified aramid fiber with hyperbranched polysiloxane (His)-Ce_{0.8}Ca_{0.2}O_{1.8} hybrid coating by in-situ method, which could improve the UV resistance and the surface activity of fbers. These methods have complicated steps or harsh condition.

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In 2007, Lee [[24](#page-13-13)] was inspired by the viscous protein from the mussels and frst discovered polydopamine polymer and found that polydopamine has strong adhesion and could adhere to almost any surface. Since its inception, polydopamine has been wide studies and applications [\[25,](#page-13-14) [26](#page-13-15)] because of its strong adhesion properties [\[27\]](#page-13-16). In the paper, a simple and green method was used to improve the surface activity and UV resistance of aramid fbers. Firstly, the aramid fber was treated by UV radiation to activate and roughen its surface, and then the fber was functionalized in dopamine solution to obtain a dense and uniform PDA coating on the fber by utilizing the adhesion properties of the PDA.

The paper focuses on the surface activity and UV resistance of the functionalized fbers with dopamine. There is no report on the detailed analysis of the UV resistance of PDA coatings. In this paper, the mechanism and reasons of improving the UV resistance of PDA coating of aramid fbers are clearly elaborated. This paper provides a novel method for improving the UV resistance of aramid fber, which has an important significance for prolonging the service life of aramid fber, and has an important infuence on broadening the applications of aramid fber.

2 Experimental

2.1 Experimental Materials

Aramid fber (Kevlar-29): Model 956, specifcation 1500D, linear density 1670dtex, DuPont, USA. 3-Hydroxytyramine Hydrochloride: Grade Reagent (RG), purity 99%+, molecular weight 189.64, Adamas. Tris (Tris reagent): grade ACS, purity 99%, molecular weight 121.14, Adamas. Ethanol (ethyl alcohol): density (20 $^{\circ}$ C) is 0.789–0.791 g/ml, Analytical Grade (AR), content≥99.7%, Tianjin Fuyu Chemical.

2.2 Cleaning the Aramid Fiber

The aramid fber was cut into short fbers about 3–5 cm, ultrasonically cleaned in absolute ethanol for 3 h, then washed fve times with deionized water, fnally dried in a blast oven at 60 \degree C for 10 h, and the aramid fiber was taken out for later use.

2.3 UV Radiation Treatment of Aramid Fiber

The aramid fber was placed in an ultraviolet aging test chamber (model: GM-UVA-2130, Shenzhen Guangmai Instrument Equipment Co., Ltd.) for 0 h, 10 h, 20 h, 30 h and 40 h, respectively, to obtain treated aramid fber with ultraviolet radiation. UV-0 h, UV-10 h, UV-20 h, UV-30 h and UV-40 h were marked separately.

2.4 Preparation of KF‑UV‑PDA

The dopamine hydrochloride was dissolved in deionized water to prepare a dopamine solution with a concentration of 2.0 g/L. The Tris–HCl reagent was added to the dopamine solution, and the pH of the solution was adjusted to 8.5 [[24\]](#page-13-13) with a pH meter (Metteler Toledo FE-20). In general, a 2.0 g/L dopamine solution and a 1.2 g/L Tris solution can bring the solution to a predetermined pH [\[28](#page-13-17), [29](#page-13-18)]. The ultraviolet radiation-treated aramid fber was placed in prepared dopamine solution, allowed to stand at room temperature for 24 h, and then placed in a 40 °C blast oven for 24 h. Finally, the obtained KF-UV-PDA fber was used for the subsequent test.

3 Characterization

3.1 Scanning Electron Microscopy Test (SEM)

The surface morphology of the aramid fber was observed with the scanning electron microscope (Quanta FEG 250, FEI Instruments, USA).

3.2 Atomic Force Microscopy Test (AFM)

The surface of the aramid fber was scanned in a tapping mode using the German Bruker's Dimension Icon. All data were passed through the NanoScope Analysis software. A second-order leveling was performed to ensure the images at the same height, thereby eliminating errors caused by changes in angle.

3.3 Transmission Electron Microscope Test (TEM)

The aramid fiber was tested by a transmission electron microscope of JEM-1230 by JEOL to obtain the appearance and the thickness of the PDA coating.

3.4 Infrared Analysis (ATR‑FTIR)

A bundle of fbers was placed on a Fourier infrared spectrometer (Nicolet iS50, American Thermo Fisher Scientifc) testing platform, and the aramid fber was tested in attenuated total refection infrared (ATR-FTIR) mode with a spectral resolution of 2 cm⁻¹, the wave number range is $600 \sim 4000 \text{ cm}^{-1}$, the number of scans is 32 times.

3.5 X‑ray Photoelectron Spectroscopy (XPS)

The surface elements and their contents of the aramid fber were analyzed by X-ray photoelectron spectroscopy

(K-Alpha+, Thermo Fisher Scientifc, USA). The X-ray source used in the experiment was an Al K α (Mono Al K α) with an energy of 1486.6 eV, 6 mA \times 12 kV, a beam spot size of 400 μ m, an operating vacuum of 2×10^{-7} mba in the analysis chamber, and a CAE mode in the scanning mode. When the fber was scanned in full spectrum, the energy was 100 eV and the step size was 1 eV. When the fber was performed in narrow spectrum, the energy was 30 eV and the step size was 0.1 eV. And each sample was tested for five times.

3.6 X‑ray Difraction Analysis (XRD)

The structural of fbers were tested with using an X-ray diffractometer (X PertPower, Panaco X-Ray Analytical Instruments, Netherlands). The experiment used a Cu target with the range of 5°–90° and the scanning speed was 10°/min.

3.7 Contact Angle Test

The aramid woven fabric was functioned with PDA with the same method as the method of the functioned fiber. The contact angle test was carried out on the aramid woven fabric by the contact angle measuring instrument (Model DSA25, KRÜSS, Germany) to test the wetting ability of the fber. The liquid used in the experiment was deionized water.

3.8 UV Absorption Spectroscopy

UV–visible spectrophotometer (Lambda 950 double-beam type, PerkinElmer, USA) was used to test the absorption capacity of the ultraviolet light of the aramid fber. The testing wavelength was 200~800 nm.

3.9 Mechanical Properties Analysis

XQ-1C fiber strength extensometer (Shanghai Xinmao Instrument Co., Ltd.) was used to test the tensile properties of aramid fber, the tensile rate was 10 mm/min, and the clamping distance was 20 mm. 30 valid data were conducted for each sample.

3.10 KF‑UV (40 h)‑Ultrasonic Treatment of PDA Fibers

Ultrasonic treatment (JP-020, ultrasonic power 120 W, ultrasonic frequency 40KHz, Shenzhen Jiemeng Cleaning Equipment Co., Ltd.) was used to get ultrasonic KF-UV(40 h)- PDA fber. And the adhesion property of the dopamine coating was then evaluated by observing the SEM image of the surface of the fber.

4 Results and Discussion

Figure [1](#page-2-0) showed the schematic diagram that the aramid fber was modifed under UV radiation and functionalized by dopamine. After the aramid fber was under UV radiation, the surface of the fber became rough and could destroy the amide bond in molecular chain of the surface, so a small amount of carboxyl functional group was obtained by the reaction of oxidation [\[30\]](#page-13-19). Dopamine can self-polymerize on the surface of the fber and dopamine was rich in phenolic hydroxyl functional groups. The carboxyl functional group on the surface could be esterifed with the phenolic hydroxyl group and the interaction between $\pi-\pi$ occurred. Under the

Fig. 1 Schematic diagram of aramid fber undergoing UV radiation (**a**) and dopamine functionalization (**b**)

dual interaction, the PDA coating could tightly adhere on the aramid fber and increase the coverage of the coating.

4.1 Scanning Electron Microscopy (SEM) Test

Figure [2](#page-3-0) represented scanning electron micrographs of the untreated fber and functionalized fber with PDA. It could be clearly seen from the fgure that the surface of KF was very smooth and clean, and there were no cracks and deposits on the surface. However, the surface of KF-UV-PDA had dopamine coating and small particles, which made the surface uneven and increased the roughness of the surface. This is benefcial to improve the surface activity and interfacial adhesion of the aramid fber. What's more, as the time of UV irradiation increased, the PDA coating became dense. Especially when the UV radiation was 40 h, the coating of the KF-UV(40 h)-PDA was the most dense. This indicated that the modifed aramid fber with UV irradiation was more benefcial to the adhesion of dopamine on its surface. From the pictures of the KF and KF-UV(40 h)-PDA samples, it

Fig. 2 SEM of the untreated fber and functionalized fber with dopamine

was apparent that the fber was signifcantly darkened by the functionalization with dopamine, which also indicated that the PDA was successfully coated on the aramid fber.

4.2 Atomic Force Microscopy (AFM) Test

Figure [3](#page-6-0) showed the surface topography of the aramid fiber using a tapping mode to strike the surface with a probe tip in a vibrating state to obtain an AFM image of the aramid fber morphology. Figure [3](#page-6-0)a showed the atomic force microscopy of the untreated aramid fber. It can be clearly seen from the fgure that the surface of the aramid fber was very smooth and clean, which is confrmed by the results of scanning electron microscopy. It can be clearly seen from the phase diagram of Fig. [3a](#page-6-0)–f that compared with the aramid fber, the surface of the dopamine functionalized fber has a diferent crystal phase, which is caused by the dopamine plating. And as the UV irradiation time increased, the thickness of the dopamine coating increased and became denser. Especially for the KF-UV(40)-PDA sample, the dense polydopamine coating on the surface of the aramid fber can be clearly seen, which fully demonstrated that the dopamine was successfully plated on the fber. It can be seen from Table [1](#page-6-1) that the root mean square roughness (Rq) and the arithmetic mean roughness (Ra) of the untreated aramid fber were small, 10.7 nm and 7.6 nm, respectively. But as the UV radiation time increased, the value of (Rq) and (Ra) of the functioned fber with dopamine functionalization tended to increase overall. Based on the results of AFM and SEM, the coating density of KF-UV(40 h)-PDA samples was the best. The next parts selected KF-UV(40 h)-PDA to represent the performance of KF-UV-PDA fbers.

4.3 Transmission Electron Microscope (TEM) Test

Figure [4](#page-6-2) represented the transmission electron micrograph and the thickness of the KF-UV(40 h)-PDA sample. It was evident from Fig. [4](#page-6-2)a and b that there was a uniform and dense PDA coating on the surface of the aramid fber. The thickness of the PDA coating was measured by the Nano Measurer test software. The thickness was tested at 50 various points, and the data distribution of the thickness of the PDA layer was as shown in Fig. [4c](#page-6-2). From the date, we got the result that the average thickness of the PDA layer was about 107 nm.

4.4 Infrared Analysis (ATR‑FTIR)

Figure [5](#page-7-0) represented the infrared characteristic absorption peaks of aramid fber, modifed fber under UV radiation and functionalized aramid fber with dopamine. Among them, Fig. [5](#page-7-0)a showed the infrared spectrum of the KF-UV fber obtained under UV treatment. It can be seen from Fig. [5a](#page-7-0) that the absorption peak at 1715 cm⁻¹ represented the cleavage of the amide bond and the production of the carboxylic acid group. At 1637 cm^{-1} , there was a strong carbonyl group C=O absorption peak, (belonging to the class I absorption peak of amide), and the peak intensity of the C=O peak increased with the increase of UV irradiation time; the NH peak at 1538 cm−1 also increased with the increase of UV radiation, which indicated that the C=O and NH group between adjacent molecular chains was broken [[30](#page-13-19)]. The results can be attributed to the breakage of the fber molecular chain and the oxidation of the edge groups during UV irradiation [[31\]](#page-13-20). In order to confrm the chemical structure of the coating, the functionalized aramid fber with dopamine (KF-UV-PDA) was tested by infrared test, and the results were shown in Fig. [5](#page-7-0)b. It can be seen from Fig. [5](#page-7-0)b that a bending vibration absorption peak of hydroxyl group (–OH) appeared at 1223 cm^{-1} , which indicated that dopamine was successfully plated on the surface of the aramid fber, and the intensity of this peak was in accordance with the time of UV radiation. That was to say, as the UV radiation time increased, the dopamine was more likely to adhere to the aramid fiber. At 1611 cm^{-1} , a characteristic ring vibration absorption peak attributed to the aromatic ester [\[32\]](#page-13-21), which was most likely due to the reaction of the C=O group produced under UV radiation on the fbers with the phenolic hydroxyl in dopamine, which further demonstrated that UV radiation could produce carboxyl functional groups on the aramid fbers.

4.5 X‑ray Photoelectron Spectroscopy (XPS)

Figure [6](#page-8-0) showed the XPS results of untreated aramid fbers, modifed fber under UV radiation and functionalized aramid fber with dopamine. Figure [6a](#page-8-0)–f showed the XPS diagram of untreated aramid fber and UV radiated aramid fber. It can be concluded from Fig. [6](#page-8-0)a and Table [2](#page-9-0) that: the ratio of N/C and the ratio of O/C of the KF was 0.10, 0.17, respectively; while the ratio of N/C and the ratio of O/C of the KF-UV(40 h) fber was 0.05, 0.21, respectively. These results indicated that compared with KF, the ratio of N/C of KF-UV(40 h) fber decreased, but the ratio of O/C increased. The reason was very likely that UV radiation destroyed the amide bond in the molecular chain of the aramid fber, and during the UV irradiation, the oxidation reaction on aramid fber occurred, so that the content of N element decreased and the content of the containing oxygen group on the surface of the fber increased [\[31,](#page-13-20) [33,](#page-14-0) [34\]](#page-14-1). From the C1s spectrum 6(b–f), it can be found that: compared with untreated KF, the fber after UV irradiation increased the peak of the $-COO-(288.5 \text{ eV})$ group because the surface of fiber was oxidized to produce carboxy after UV irradiation. And the content of the –COO– group increased as the UV irradiation time increased, which indicated that the content of the

Fig. 3 Atomic force microscopy of untreated fber and functional-◂ ized fber with dopamine **a** KF, **b** KF-UV(0 h)-PDA, **c** KF-UV(10 h)- PDA, **d** KF-UV(20 h)-PDA, **e** KF-UV(30 h)-PDA, **f** KF-UV(40 h)- PDA)

Table 1 Surface roughness of untreated KF and KF-UV-PDA fbers

Sample	Ra (root mean square roughness)/nm	R_a (arithmrtic mean roughness)/nm	
КF	10.7	7.6	
$KF-UV(0 h) - PDA$	19.9	14.9	
KF-UV(10 h)-PDA	23.9	18.9	
$KF-UV(20 h)-PDA$	28.3	22.5	
KF-UV(30 h)-PDA	43.2	33.5	
$KF-UV(40 h)-PDA$	33.9	29.5	

carboxyl functional group on the fber increased. According to the results of the XPS Fig. [6](#page-8-0)g, h, compared with KF-UV fber, KF-UV(40 h)-PDA fber increased a peak of –OH(286.1 eV) group, which was a characteristic absorption peak of dopamine indicating that dopamine was successfully coated on the surface of the fiber, which was mutually confrmed by the results of SEM and FTIR.

4.6 X‑ray Difraction Analysis (XRD)

It was known from the results of FTIR and XPS that the chemical structure of the surface of the aramid fber changed after UV radiation and functionalization with dopamine. Then, whether the UV radiation and functionalization with

Fig. 5 Infrared spectrum of untreated aramid fber, modifed fber under UV radiation and functionalized aramid fber with dopamine

dopamine also changed the aggregation structure of the aramid fber? So, the X-ray difraction pattern of the untreated aramid fber, the UV radiated fber and the functional fber with dopamine were carried out, and the results were shown in Fig. [7.](#page-9-1) As can be seen from the fgure, KF-UV-PDA fber, KF-UV fber and KF had similar XRD difraction patterns, which were mainly two sharp difraction peaks: 2*θ* were 20.5 and 22.8 corresponding to the characteristics (110) and the (200) crystal plane of aramid fber, respectively [[35](#page-14-2), [36](#page-14-3)]. The results indicated that UV radiation and functionalization with dopamine did not change the aggregation structure of the aramid fber. However, the UV radiated fber, the intensity of the characteristic difraction peak was lower than that of the untreated aramid fber, indicating that the crystallinity of the UV radiated fber was lowered. Through the testing, the corresponding crystallinities of KF, KF-UV fiber and KF-UV-PDA fiber were 81.11%, 58.60% and 65.46%, respectively. Because the aramid fber underwent UV radiation and damaged the surface structure of the fber, resulting in a decrease in the overall crystallinity. However, the crystallization degree of the functional fber with dopamine improved, which may be due to the fact that dopamine reacted with the functional groups on the surface of the fber, improving the regularity of the fber to a certain extent, so the crystallinity was improved. In summary, the treatment method used in this paper was mild and did not damage to the aggregation structure of aramid fber.

4.7 Surface Infltration Performance Analysis

Common contact angles were used to characterize the surface wetting of fbers [[36\]](#page-14-3). Figure [8](#page-9-2) showed the contact angle of untreated KF and KF-UV-PDA fbers with deionized water. It can be seen from the fgure that KF had the largest contact angle and exhibited hydrophobicity, while as the UV radiation time increased, the contact angle of the fber became smaller, which indicated that the hydrophilicity of the KF-UV-PDA fbers increased, proving that the surface wettability of KF-UV-PDA fbers was better than KF, especially the KF-UV(40 h)-PDA fber had the best surface wettability. The reason was more likely that the improvement of the surface wettability was related to the change of the structure of the aramid fber. After the fber was functionalized of with dopamine, the hydrophilic active group (hydroxyl group) on the fber increased, which was confrmed by the FTIR and XPS results.

4.8 Ultraviolet Absorption Spectrum Analysis

Figure [9](#page-9-3) showed the UV absorption spectra of KF and KF-UV(40 h)-PDA fibers. It can be clearly seen from Fig. [9](#page-9-3) that compared with KF, the absorption intensity of the ultraviolet absorption peak near 217 nm and 305 nm of KF-UV(40 h)-PDA fber was signifcantly improved by 106.83%, 57.29%, respectively. What's more, the width of the absorption peak near 305 nm was increased. The reasons for the improvement of the absorption intensity near 217 nm was that the surface of the KF-UV(40 h)-PDA fiber had hydroxyl group (–OH) chromophore in the dopamine, which had a lone pair of electrons, and itself could not absorb light larger than 200 nm. However, when it was connected to the chromophore (double bond in the benzene ring), it could interact with the π electron in the benzene ring, so that the $n \rightarrow \sigma^*$ transition energy was reduced, thereby enhancing its absorption intensity. The reason of the improvement of absorption peak near 305 nm was that after the UV radiation, the hydrogen bond and the amide bond between the molecular chains of the aramid fiber were broken, and

Fig. 6 XPS spectrum of untreated aramid fber, UV radiated fber and functionalized fber with dopamine

Table 2 Chemical composition on the surface of untreated KF, KF-UV(40 h) and KF-UV(40 h)-PDA fber

Fiber	Chemical composites (atomic%)			
	\subset	N		
KF	78.36	7.96	13.67	
$KF-UV(40 h)$	79.25	4.29	16.46	
$KF-UV(40 h)-PDA$	79.51	5.72	14.77	

Fig. 7 XRD pattern of KF, KF-UV(40 h) and KF-UV(40 h)-PDA fber

Fig. 8 Contact angle test chart of untreated KF and KF-UV-PDA fbers

then the oxidation of the UV radiation produced carboxyl functional group, which could generate an ester with the hydroxyl group in dopamine. The carbonyl group in the ester and the benzene ring of dopamine could produce conjugated reaction [\[37](#page-14-4)], which reduced the transition energy of $n \rightarrow \pi^*$, thereby signifcantly increasing the absorption intensity near

Fig. 9 Ultraviolet absorption spectrum of KF and KF-UV(40 h)-PDA fiber

305 nm. In summary, the functional fber with dopamine had excellent UV resistance, and the change of the absorption intensity of the absorption peak was closely related to the change of chemical structure on the fber. Therefore, the ultraviolet absorption spectrum had a great signifcance for studying the chemical structure of the aramid fber.

4.9 Analysis of Mechanical Properties

The mechanical properties of KF and KF-UV-PDA fbers were shown in Fig. [10.](#page-10-0) It can be seen from the figure that as the UV radiation time increased, the strength and elongation at break of the fber gradually decreased. Compared with KF, the strength and elongation at break of KF-UV(40 h) fber decreased about 19.15% and 11.26%, respectively. This was because that after the fber was radiated by UV, defects such as gullies appeared on the surface, where stress concentration tended to occur during the stretching process, which was prone to break, thereby the mechanical properties were lower than the untreated fber. The strength and elongation at break of KF-UV(40 h)-PDA fber were reduced by 16.71% and 9.58%, respectively. It was clear that the strength and elongation at break of fber were improved after functionalization with dopamine. Moreover, compared with KF-UV(40 h) fiber, the strength and elongation at break of KF-UV(40 h)-PDA fbers increased by 2.99% and 3.05%, respectively. This might be the contribution of the dopamine coating on the fber surface. That was to say, dopamine reacted with the carboxylic acid functional groups on the surface of the fber to compensate for some of the defects on the fber. As can be seen from the mechanical data from Fig. [10,](#page-10-0) the dopamine layer was advantageous for improving the mechanical properties of the fber.

 \overline{a}

Ó

 0_h

 $10h$

Fig. 10 Mechanical properties of KF and KF-UV-PDA fbers

4.10 Research on the Properties of Fiber after 1 Week of UV Radiation

4.10.1 Scanning Electron Microscopy Analysis

The common testing characterization method against ultraviolet radiation was to observe the surface structure and tensile properties of the fber. Therefore, KF and KF-UV-PDA fbers after 168 h of UV radiation with SEM and tensile properties were carried out. Figure [11](#page-11-0) showed a scanning electron micrograph of aramid fber after 168 h UV irradiation. It can be clearly seen from the KF diagram that the surface of the fber became rough, the deep gully appeared in the axial direction of the fber. The fber had obvious fracture, the skin layer of the fber was upwarpped, and deposits were formed on the surface. After the UV radiation of the aramid fber, the molecular chain was broken, and the terminal group was oxidized. Compared with untreated KF, the surface of KF-UV-PDA fber was less damaged, and the surface of PDA particles and coating had a small amount of shedding, but the fber had no gully on the surface. It showed that dopamine could be frmly coated on the surface of the aramid fber after long time of the UV radiation. The results of the SEM were a good proof that the functionalized fber with dopamine had an excellent performance of UV resistance.

4.10.2 Analysis of Mechanical Properties

Figure [12](#page-12-0) showed the mechanical properties of KF and KF-UV-PDA fibers after 168 h of UV radiation. It can be seen from the Fig. [12](#page-12-0) that as the UV irradiation time increased, the strength and elongation at break of the fber were gradually increased. When the time of UV irradiation was 40 h, the mechanical properties of the sample reached

 $20h$ Sample

4.11 Ultrasonic Treatment of KF‑UV(40 h)‑PDA Fiber

The PDA layer was coated on the fber evenly and densely, but what about its adhesion? With regards to this, ultrasonic instruments (JP-020, ultrasonic power 120 W, ultrasonic frequency 40KHz, Shenzhen Jiemeng Cleaning Equipment Co., Ltd.) was used for KF-UV(40 h)-PDA fber for 5 min, 20 min, 40 min and 60 min sonication, and the samples were numbered US-5, US-20, US-40, and US-60, respectively. The SEM images of the obtained fber were shown in Fig. [13](#page-12-1). The adhesion of dopamine was evaluated by observing the degree of shedding of the PDA coating on the surface. It can be clearly seen from the figure that when the KF-UV(40 h)-PDA fber was treated for 60 min, the PDA coating on the surface had slightly shedding, which fully showed that PDA layer had strong adhesion on the fiber, and had a potential in practical applications.

AKF-UV

 $30h$

KF-UV-PDA

40h

Fig. 11 SEM of KF and KF-UV-PDA fbers after 168 h UV radiation

5 Conclusion

(1) It can be seen from the SEM and AFM images that the surface of the aramid fber was very smooth. When the aramid fber was functionalized with dopamine, the coating and particles of dopamine can be clearly seen on the surface of the fiber. Especially for the KF-UV(40 h)-PDA fber, there was a dense and uniform PDA coating on the fber, which made the fber rough. According to the contact angle test, the surface

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wettability of the functionalized fber with dopamine was improved, which indicated that the PDA coating increased the surface activity of the fber. According to the TEM test, the PDA layer was about 107 nm.

(2) It can be seen from the results of the FTIR that after the UV radiation, the chemical structure of the surface changed, the carboxyl group was oxidized to increase the reactive groups on the fber, which was more advantageous to coat dopamine on the fber. It was found by XPS results that as the UV irradiation time increased,

Fig. 12 Mechanical properties of KF and KF-UV-PDA fbers after 168 h UV radiation

the amount of amide bond cleavage increased, and the presence of carboxyl functional groups was further confirmed, which verified that dopamine was successfully coated on the fber surface. According to the XRD test, the aggregation structure of the fber did not change after the aramid fber was functionalized by UV radiation and dopamine.

- (3) The UV absorption spectrum was closely related to the structure of the surface of aramid fber. Corresponding to the absorption peaks around 217 nm and 305 nm, compared with KF, the UV absorption intensity of the functional fber with dopamine was distinctly enhanced, so the PDA coating had good UV resistance.
- (4) KF and KF-UV-PDA fbers were under UV radiation for 168 h, SEM results showed that the PDA coating could protect the cortical structure of aramid fber very

Fig. 13 SEM images of KF-UV(40 h)-PDA fber after ultrasonic treatment at diferent times

well. Compared with KF, the mechanical properties test indicated that KF-UV-PDA fber had high mechanical property retention rate after 168 h UV radiation. Especially for KF-UV (40 h)-PDA fber, the retention of strength and elongation at break was 23.12% and 19.42% higher than results of the untreated fber, respectively. This strongly proves that PDA coating has good UV resistance and could increase the service life of aramid fbers. Moreover, the paper provides a novel approach to improve the UV resistance of aramid fbers.

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