




# Subtle Chemical Variations with Strong Ecological Significance: Stereoselective Responses of Male Orchid Bees to Stereoisomers of Carvone Epoxide

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## Abstract

Different enantiomers of chiral compounds within floral perfumes usually trigger distinct responses in insects; however, this has frequently been neglected in studies investigating semiochemicals in plant-pollinator interactions. Approximately 1000 neotropical plants produce floral perfumes as the only reward for pollinators, i.e. male euglossine bees. The chiral compound carvone epoxide is a key component of the scent bouquet of many perfume-rewarding plants that are pollinated by males of *Eulaema*. Here, we tested the biological activity of the four carvone epoxide stereoisomers to four *Eulaema* species occurring in the Atlantic Rainforest of NE-Brazil. We determined the stereochemistry of carvone epoxide in the floral scent of several *Catasetum* species, tested whether the antennae of bees respond differentially to these stereoisomers and investigated if there is a behavioural preference for any of the stereoisomers. We found that 1) *Catasetum* species emit only the (–)-*trans*-stereoisomer of carvone epoxide, 2) for *E. atleticana* and *E. niveofasciata* antennal responses to the (–)-*trans*-carvone epoxide were significantly stronger than those to (–)-*cis*-carvone epoxide, 3) the strength and pattern of antennal responses to all 4 stereoisomers (separately tested) did not differ among *Eulaema* species, and 4) there were significant differences in attractiveness of the four stereoisomers to the bees species with the (–)-*trans*-stereoisomer being particularly attractive. We assume (–)-*trans*-carvone epoxide to be the dominant isomer in perfume-rewarding plants pollinated by *Eulaema*. The universal occurrence of carvone epoxide in *Catasetum* species pollinated by *Eulaema*, suggests that this compound has evolved in perfume-rewarding as a specific attractant for *Eulaema* bees as pollinators.

**Keywords** *Catasetum* · Chirality · Electrophysiology · *Eulaema* · Euglossine bees · Perfume-rewarding plants · Pollinators

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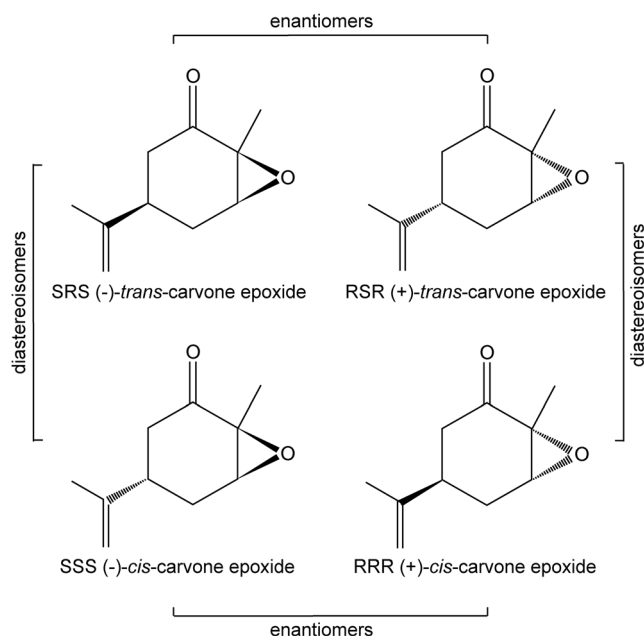
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## Introduction

The biological function of an organic compound is determined by its properties, such as chain length, functional groups, and stereochemistry (König and Hochmuth 2004). Floral scents are often complex blends of volatile organic compounds (VOCs), of which many are chiral (e.g. Williams and Whitten 1983). Chiral compounds have rigid spatial arrangements of atoms that contain no mirror plane, or no centre of inversion (i.e. no rotation-reflection axis), and thus occur in two non-superimposable mirror-image forms (enantiomers; Moss 1996). Stereoisomers are characterized by differences in physicochemical properties and usually trigger distinct biological responses in insects (e.g. Ayasse et al. 2003; Dopman et al. 2010; Erbilgin and Raffka 2000; Linn et al. 2003; Teale et al. 1994). This is especially true for enantiomers, which only differ in physical properties, i.e., their optical rotation.

However, most studies that investigate the chemical communication by olfactory cues between plants and their pollinators do neither elucidate the absolute configuration of chiral floral scent compounds nor study their potentially differential role in attracting pollinators (Dötterl et al. 2006; Gerlach and Schill 1991; Hills et al. 1972; Milet-Pinheiro and Gerlach 2017; Schneider et al. 2013; Williams and Whitten 1983). This is also true for perfume-rewarding plants, which are pollinated by male euglossine bees, even if the significance of chirality in these associations has long been outlined (Eltz et al. 1999; Williams and Whitten 1983).

The production of floral perfumes by certain neotropical plants as reward for male euglossine bee pollinators was first recognized by Vogel (1966). His findings led other contemporaneous scientists to investigate the chemistry of floral perfumes in a myriad of euglossine-pollinated plants, which culminated in the identification of many compounds new to science. In 1985, for example, Lindquist et al. (1985) isolated and described the monoterpenoid (2*S*,5*R*)-(-)-2,3-epoxy-5-isopropenyl-2-methyl-cyclohexanone, also known as (-)-*trans*-carvone epoxide, from the floral fragrance of *Catsetum maculatum*. Later, Whitten et al. (1986) stated that this compound is the major component in the floral fragrance of taxonomically diverse angiosperms (several genera of Orchidaceae and one species of *Dalechampia*, Euphorbiaceae) and is a potent attractant for male euglossine pollinators, mainly of the genus *Eulaema*. Since then, *trans*-carvone epoxide, without information about its absolute configuration, has repeatedly been reported in perfume-rewarding plants of many different genera and families (*Dalechampia*, Euphorbiaceae; Armbruster 1989; several genera of Orchidaceae, Kaiser 1993, 2011; *Gloxinia*, Gesneriaceae; Martel et al. 2019; *Anthurium*, Araceae; Schwerdtfeger et al. 2002; *Unonopsis*, Annonaceae; Teichert et al. 2009). Interestingly, it has never been reported as floral scent constituent in non-perfume-rewarding plants (Knudsen et al. 2006). In perfume-rewarding plants, *trans*-carvone epoxide sometimes co-occurs with its diastereoisomer *cis*-carvone epoxide (Kaiser 1993, 2011; Schwerdtfeger et al. 2002). Carvone epoxide is a complex molecule that has three chiral centers. However, because of the epoxide group, only two diastereoisomers are possible: For each diastereoisomer of carvone epoxide, there are two possible enantiomers, i.e. (1*S*,4*R*,6*S*)-1-methyl-4-(prop-1-en-2-yl)-7-oxabicyclo[4.1.0]heptan-2-one [hereafter (-)-*trans*-carvone epoxide] and (1*R*,4*S*,6*R*)-1-methyl-4-(prop-1-en-2-yl)-7-oxabicyclo[4.1.0]heptan-2-one [(+)-*trans*-carvone epoxide], and (1*S*,4*S*,6*S*)-1-methyl-4-(prop-1-en-2-yl)-7-oxabicyclo[4.1.0]heptan-2-one [(-)-*cis*-carvone epoxide] and (1*R*,4*R*,6*R*)-1-methyl-4-(prop-1-en-2-yl)-7-oxabicyclo[4.1.0]heptan-2-one [(+)-*cis*-carvone epoxide] (Fig. 1).



**Fig. 1** The four different stereoisomers of carvone epoxide

To our best knowledge, there are so far only two studies testing whether stereoisomers of floral compounds play a distinct role in the attraction of male euglossine bees. In the first study, Williams and Whitten (1983) found that male *Eulaema nigrata* bees are attracted to (-)- $\alpha$ -pinene, while (+)- $\alpha$ -pinene has a repellent effect. In the second study, Schorkopf et al. (2011) showed that the (-)-isomer of ipsdienol triggers stronger antennal responses and attracted more males of *Euglossa cyanura* than its optical isomer or a racemic mixture. Common theories suggest that male euglossine bees use perfumes to attract con-specific females and species-specific perfume blends are assumed to act as pre-mating reproductive barrier among sympatric species (Weber et al. 2016; Zimmermann et al. 2009). Chirality, therefore, may act as an effective reproductive barrier, if different species occurring in the same habitat show distinct preference for one stereoisomer over the other(s).

*Catsetum* Rich. Ex Kunth. is the most speciose genus of the subtribe Catasetinae (Orchidaceae), embracing about 170 species (Govaerts et al. 2018). In a recent review on floral scent chemistry and pollinators of *Catsetum*, Milet-Pinheiro and Gerlach (2017) confirmed *trans*-carvone epoxide as a main component of the floral scent of many species pollinated by bees of the genus *Eulaema*. However, the absolute configuration of this compound in most species, as well as the possible stereoselective attraction of different *Eulaema* species, remains unknown. In the present study, we synthesized all four carvone epoxide stereoisomers and tested them in electrophysiological and biological assays with *Eulaema* species. In order to put our findings into an ecological context, we

determined the stereochemistry of carvone epoxide in the floral scent of several *Catasetum* species pollinated by *Eulaema*. Our aim was to explore whether males of different *Eulaema* species can distinguish between different stereoisomers of carvone epoxide in electrophysiological analyses and in the field assays. Specifically, we addressed the following questions: 1) Which stereoisomer of carvone epoxide is emitted by five different *Eulaema*-pollinated species of the genus *Catasetum*? 2) Does strength of antennal responses to the four stereoisomers differ within and among species? 3) Do bees prefer a specific stereoisomer in behavioural field assays?

## Material and Methods

**Plant Species and Collection of Headspace Samples** Floral volatiles of five *Catasetum* species [*C. discolor* Lindl. 1844 ( $N=2$ ), *C. gardneri* Schltr. 1914 ( $N=6$ ), *C. macrocarpum* Rich. Ex Kunth 1822 ( $N=5$ ), *C. osculatum* Lacerda & P. Castro 1995 ( $N=3$ ), and *C. schmidtianum* Miranda & Lacerda 1992 ( $N=1$ )] were collected using dynamic headspace methods (Dötterl et al. 2005). Therefore, inflorescences were enclosed in a polyester oven bag (Toppits®) and the floral scent was trapped for 5 h in an adsorbent tube [glass vials; 15 mg Tenax-TA (mesh 60–80) and 15 mg Carbotrap B (mesh 20–40) fixed using glass wool] using a membrane pump (G12/ 01 EB, Rietschle Thomas, Puchheim, Germany). The flow rate was adjusted to 200 mL/min. Volatiles trapped in these adsorbent tubes were eluted with 200  $\mu$ L *n*-hexane (99.5%, Merck). To detect ambient contaminants, negative controls (empty bags;  $N=1$  for each species) were collected by using the aforementioned methods. All headspace samples were stored in 2 mL screw cap vials at  $-20$  °C until the chemical analyses.

**Synthesis of Carvone Epoxide Stereoisomers** The carvone epoxide stereoisomers were prepared from (*S*)- or (*R*)-carvone according to procedures reported in the literature (Garver et al. 1976; Takita et al. 2011; Wang et al. 2006; Yasuda et al. 1979). (*R*)-(-)-carvone (98%) and (*S*)-(+)-Carvone (90%) were purchased from Sigma-Aldrich but were not pure. The latter explains the lower enantiopurity of the (+)-*trans*- (92%) and the

(-)-*cis*- (85%) epoxides obtained in their synthesis as determined by chiral gas chromatography (Table 1, see also below). Further information on NMR spectra of the synthesized carvone epoxides is included in the supplementary material (see SM1, SM2, SM3, and SM4).

**Chiral Gas Chromatography** Chiral GC-FID (flame ionization detector) was used to elucidate the isomeric composition of carvone epoxide in headspace samples of the five *Catasetum* species and in the synthesized samples (see above). All five plant species are pollinated by male *Eulaema* bees and have the typical caraway note of species that have carvone epoxide as main constituent in their floral scent bouquet (Kaiser 2011). The headspace samples and the synthetic isomers of carvone epoxide were analyzed on an Agilent 7890A gas chromatograph (Santa Clara, USA) equipped with a fused silica capillary column (30 m  $\times$  0.23 mm i.d.), coated with a 0.23  $\mu$ m film of 0,4% heptakis(2,3-di-*O*-methyl-6-*O*-*tert*-butyldimethylsilyl)- $\beta$ -cyclodextrin (DIME- $\beta$ -CD) (30%) in SE-52 (70%), the same as described in (Dötterl et al. 2006). Aliquots of 1  $\mu$ L were injected in splitless mode in a split/splitless injector (Agilent) with temperature set at 250 °C. The temperature of the GC oven was increased from 40 °C to 200 °C at a heating rate of 20 °C per minute. Hydrogen was used as the carrier gas with a constant flow rate of 3 mL/min. The synthetic isomers were injected alone and as a mixture, and the natural samples were injected alone or co-injected with the mixture of the synthetic isomers.

**Bee Species** In this study, we tested four different *Eulaema* species known to occur in the Atlantic Rainforest of Pernambuco State, Northeastern Brazil, i.e. *E. athleticana* Nemésio, 2009, *E. marcii* Nemésio, 2009, *E. nigrita* Lepeletier, 1841, and *E. niveofasciata* Friese, 1899. For all species, 12 antennae of different males were tested. Bees were obtained on scent baits [i.e. filter papers impregnated with 30  $\mu$ L eucalyptol (99%; Merck), benzyl acetate ( $\geq 99\%$ ; Merck), eugenol ( $\geq 98\%$ ; Merck), methyl salicylate ( $\geq 99\%$ ; Merck) or skatole (98%; Merck)], from which they were collected with entomological nets. They were placed in glass vials and kept inside a cooled box until electrophysiological analyses. All bees were collected at the surroundings of the

**Table 1** Isomeric composition of synthesized carvone epoxides according to chiral gas chromatography

Percentage	(-)- <i>trans</i> - (%)	(+)- <i>trans</i> - (%)	(-)- <i>cis</i> - (%)	(+)- <i>cis</i> - (%)
Sample “(-)- <i>trans</i> -”	100	0	0	0
Sample “(+)- <i>trans</i> -”	8	92	0	0
Sample “(-)- <i>cis</i> -”	0	5	85	10
Sample “(+)- <i>cis</i> -”	7	0	0	93

“Mata dos Brennands” (10 m a.s.l.; 8°02′30.5″S, 34°57′54.1″W), municipality of Recife (Pernambuco State - Northeastern Brazil). Sampling of bees was authorized by the Instituto Chico Mendes de Conservação da Biodiversidade (ICMBio) of the Ministério Brasileiro do Meio Ambiente (permit number 53545–1).

**Electroantennographic Measurements** To test if the peripheral olfactory circuitry of the four tested *Eulaema* species differently responds to the carvone epoxide stereoisomers, we used electroantennographic analyses (EAG). For each tested bee individual, one of the antennae was excised at the scape using micro-scissors (Castroviejo, Fine Science tools; 69,121 Heidelberg, Germany). Using a stereomicroscope (Stemi 2000-CS, ZEISS, Oberkochen, Germany) and a razor blade, the excised antenna was cut at the tip (last segment of flagellum) and at the base (first segment of flagellum). The excised antenna was mounted between two glass capillaries that were filled with insect Ringer solution (1 L aqua demin containing 5 g of NaCl, 0.42 g of KCl and 0.19 g of CaCl +2H<sub>2</sub>O) and connected to silver-electrodes. The electrode connected with the base of the antenna was grounded, while the electrode that was connected to the tip transmitted changes of the potential within the antenna to a signal acquisition controller (IDAC-2 Signal acquisition controller; Syntech, Hilversum, the Netherlands). The antennal preparation was placed in front of a glass tube, through which a constant humidified airflow (25 mL/min) was blown, in order to prevent dehydration and to transfer test compounds to the antenna (hereafter scent stimuli).

We applied the scent stimuli to each antennal preparation in constant order [(–)-*trans*-carvone epoxide, (+)-*trans*-carvone epoxide, (–)-*cis*-carvone epoxide and (+)-*cis*-carvone epoxide], each of them in four different ascending concentrations (10<sup>–8</sup>, 10<sup>–6</sup>, 10<sup>–4</sup>, 10<sup>–2</sup>, v/v, in *n*-hexane 99%, Merck), starting and ending with *n*-hexane as negative control.

Because prolonged or repetitive stimulation can lead to decreased antennal responses (Strausfeld and Kaissling 1986), we allowed a resting phase of 90 s between stimuli. For each stimulus, a v-shaped strip of filter paper (0.25 cm<sup>2</sup>; VWR International, Darmstadt, Germany) was inserted in a Pasteur pipette (15 cm, VWR International, Darmstadt, Germany). The filter paper inside the Pasteur pipette was impregnated with 10 µL of the respective testing solution. Before connecting the Pasteur pipettes with the stimuli to a stimulus controller (CS-05; Syntech, Hilversum, the Netherlands), *n*-hexane was allowed to evaporate for 1 min. For measurements, air puffs were delivered through the Pasteur pipettes onto the antenna via this stimulus controller (30 ms, with a pulse flow of 25 mL/min). Antennal responses were recorded by Syntech EAG software (EAGPro, v. 2.1; Hilversum, The Netherlands).

We further tested solutions of eugenol and benzyl alcohol in a concentration of 10 µL/mL (10<sup>–2</sup>) as positive control. Both substances are well known attractants to euglossine bees (Ramírez et al. 2002). Absolute antennal responses to these substances did not differ among the four tested bee species (PERMANOVA: pseudo-F<sub>12, 239</sub> = 0.35, *P* = 1).

**Field Assays** The attractiveness of the four carvone epoxides (10<sup>–2</sup> in *n*-hexane, v/v) to the different *Eulaema* species (*E. atleticana*, *E. marcii*, *E. nigrita*, *E. niveofasciata*) was tested in field assays at the same localities where we caught bees for electrophysiological measurements. We tested the compounds individually against *n*-hexane from 06:00 to 10:00 h within the time interval in which *Eulaema* bees are most active at our study site (Milet-Pinheiro, pers. obs.). Field assays were performed on 12 different non-consecutive days (*N* = 24 for each test compound). Each of the four carvone epoxides and the solvent control *n*-hexane were tested twice a day for periods of 30 min. To exclude possible influence of temporal variation in foraging activity of bees, we tested the four compounds in different orders. Overall, the four compounds were tested equally among different time periods.

In the field assays, we attached filter papers (size 5 × 5 cm) to a tree using a pin at a height of approximately 180 cm. We used a pipette (LabMate Soft, 10–100 µL; PZ HTL S.A, Warszawa, Poland) to apply 30 µL of each test solution on a filter paper at the beginning of the assay and again 15 min later. Each compound was tested on different trees to avoid a possible learning effect by the bee individuals. The scent emission of 30 µL of the tested isomers corresponded to the mean scent emission of inflorescences of *Eulaema*-pollinated plant species (*Catsetum discolor*, *C. gardneri*, *C. macrocarpum*, *C. osculatum*, *C. schmidtianum*, Brandt et al., unpublished). If several approaches / landings were obviously from a single bee individual, it was counted as the same response. We did not collect bees during assays to avoid interference in responses by other bees. Thus, it is possible that one single bee individual has responded more than once if it has left the range of vision to come back later during assays. However, considering the frequency of visits and the size of the study field, as well as the large sample size, we believe that eventual repetitive responses by any bee individual might not have biased our results.

## Statistical Analyses

**Absolute EAG Responses** To test for differences in EAG responses among the tested bee species, we first normalized the EAG responses against *n*-hexane using the option provided in the Syntech software. Thereby, we corrected for a change in antennal sensitivity during measurements. These normalized

EAG responses are in the following referred to as “absolute EAG responses”. Absolute EAG responses were log-transformed (to achieve normality of residuals and reduce heteroscedasticity). Using these responses, we calculated univariate Euclidean distance matrices (for each compound individually) and performed multifactorial general linear mixed model analyses (LMM) [random factors: individual (nested in bee species), fixed factors: bee species and stereoisomer, covariate: stimulus concentration]. We tested for differences among the four different concentrations and for differences in comparison to *n*-hexane within each species (within-species comparison for a concentration effect). With this analysis strategy on data derived from repeated measurements, any effects on signal amplitudes due to individual antenna length could be corrected statistically. Based on these LMM models, further posterior comparisons were calculated (i.e. pairwise tests).

**Relative EAG Responses** We standardized the absolute EAG responses in relation to responses to the reference substances (i.e. eugenol and benzyl alcohol). The relative response (%) to a carvone epoxide stereoisomer in four different concentrations was calculated as the absolute EAG response of an individual divided by the mean response of the same individual to the two reference substances. This standardization was necessary since absolute EAG responses are influenced by the length of the antenna (Nagai 1981; Roelofs 1984), which differed between the four tested bee species.

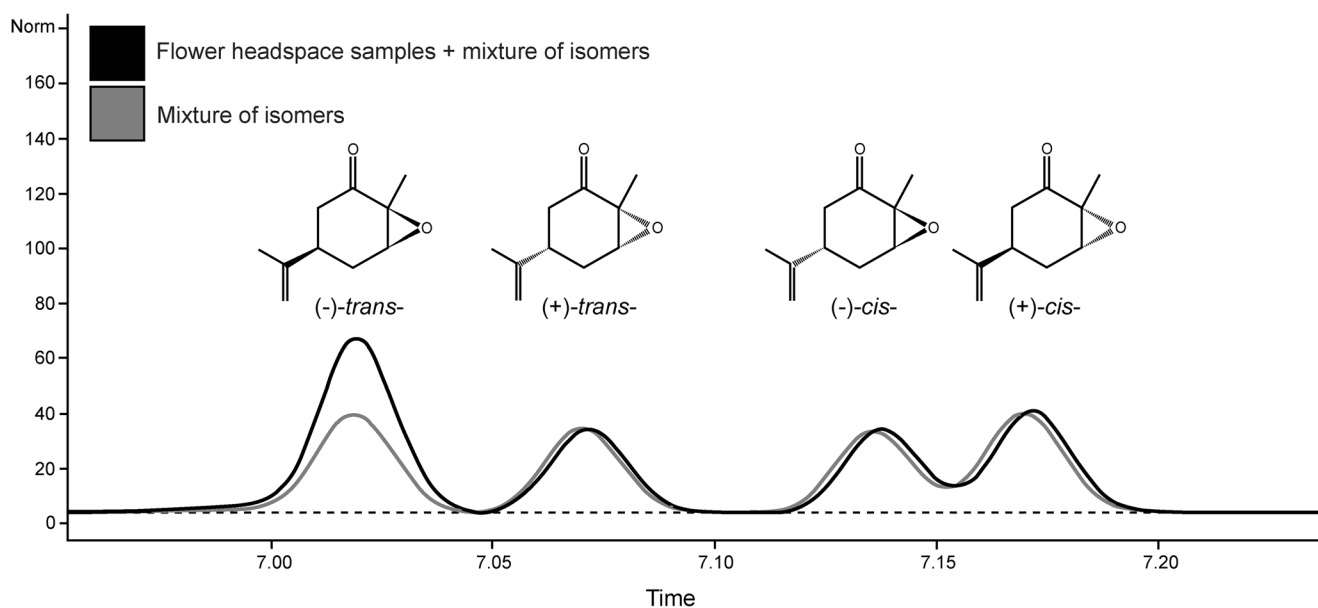
To test whether (1) a single bee species responds differently to the four stereoisomers and (2) different bee species respond

differently to the four stereoisomers, we followed the procedure of Brandt et al. (2017). After standardization of responses, we used the multiple measurements of an individual (antenna) to the different concentrations of a specific compound and calculated a linear regression line using R (v 3.5.1). The intercept of this line was extracted and represents the position where the regression line crosses the Y-axis. This value is a measure for the overall strength of responses. Using the intercept instead of absolute responses, allows us to avoid repetitive statistical analyses for each of the four different concentrations. These individual intercepts were analyzed with a linear mixed model using species, stereoisomers and their interaction as fixed factors and individuals as random factor.

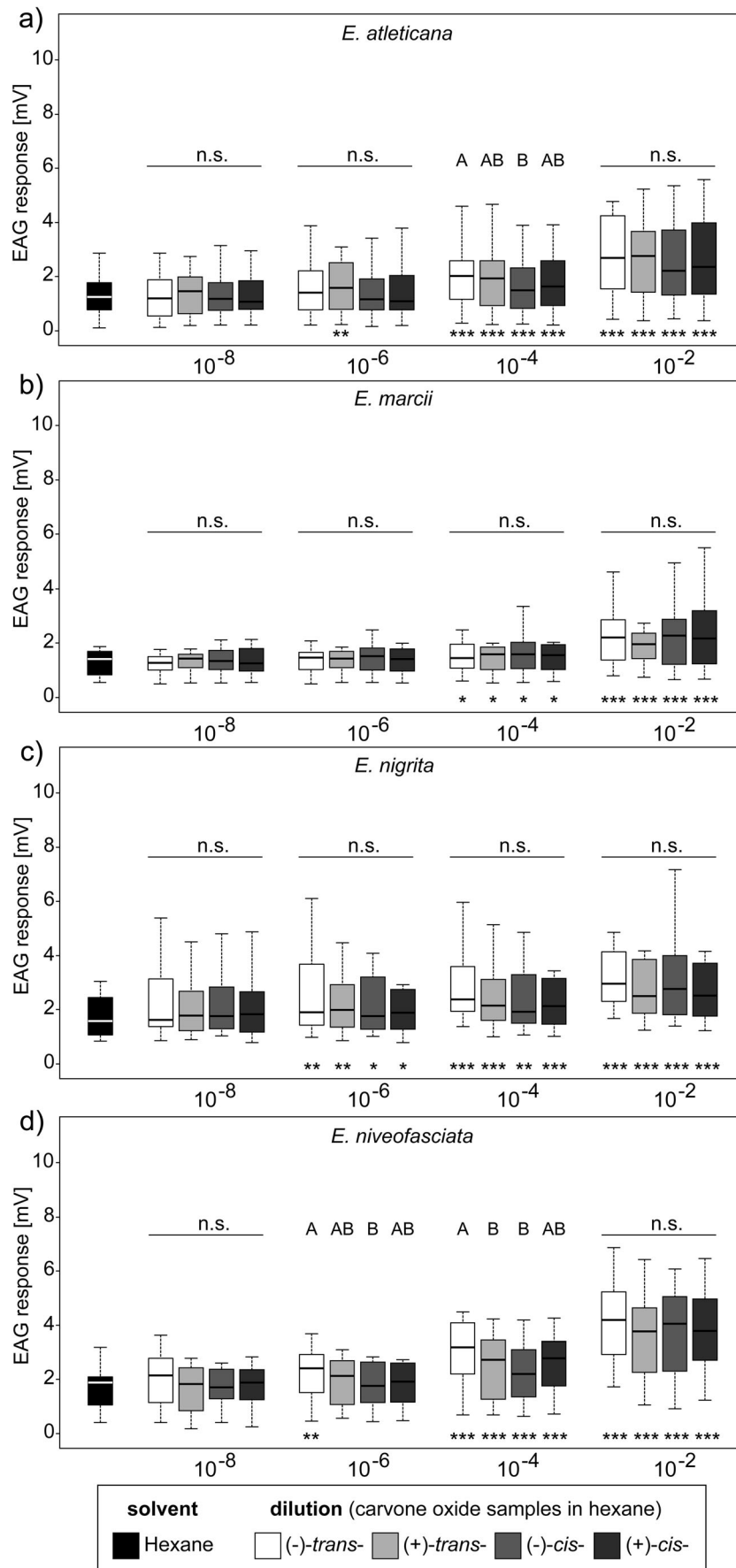
All procedures concerning linear mixed models (LMM) were performed with R (v3.5.1; R Core Team 2018) and some of its additional packages. Here ‘lme4’ (v. 1.1.18; Bates et al. 2014) and ‘lmerTest’ (v. 3.0.1; Kuznetsova et al. 2017) were used for designing, calculating and testing overall effects of linear mixed effect models, ‘emmeans’ (v. 1.2.4; Lenth 2018), and ‘multcomp’ (v. 1.4.8; Hothorn et al. 2008) for further post-hoc tests and pairwise comparisons.

**Field Assays** To test for differences in the number of visits by a specific bee species to the four stereoisomers, we calculated a pairwise Euclidean distance matrix and performed a two-factorial PERMANOVA analysis (factors: stereoisomer and bee species) followed by post-hoc analyses.

PERMANOVA is a technique for testing the simultaneous response of one or more variables to one or more factors in an



**Fig. 2** Chromatograms of a synthetic mixture of the four carvone epoxide isomers ((-)-*trans*-, (+)-*trans*-, (-)-*cis*-, and (+)-*cis*-; grey curve) and the co-injection of this mixture with a flower headspace sample of *Catasetum macrocarpum* (black curve)



◀ **Fig. 3** Absolute EAG responses of male bees of a) *E. atleticana*, b) *E. marcii*, c) *E. nigrita* and d) *E. niveofasciata* ( $N=12$  each) to different concentrations of the four carvone epoxides and to the control *n*-hexane (%; Median, 25% - 75%, confidence interval of 95%). Asterisks indicate significant differences when compared to *n*-hexane (linear mixed model; \* / \*\* / \*\*\* =  $P \leq 0.05$  / 0.01 / 0.001). Different letters indicate significant differences between tested stereoisomers within a concentration

ANOVA experimental design based on a similarity/distance matrix with permutation methods (Anderson et al. 2008). The PERMANOVA and the Euclidean resemblance matrices were calculated in PRIMER 6 (version 6.1.15; PRIMER-E Ltd. 2012) in combination with the add-on PERMANOVA + (version 1.0.5; PRIMER-E Ltd. 2012).

## Results

**Chemical Configuration of Carvone Epoxide in *Catasetum* Spp.** Co-injections of the synthetic stereoisomers of carvone epoxide with flower headspace samples revealed that all five *Catasetum* species, i.e., *C. discolor*, *C. gardneri*, *C. macrocarpum*, *C. osculatum*, and *C. schmidtianum* emit only the (–)-*trans*- stereoisomer of carvone epoxide (Fig. 2).

## Electroantennographic Measurements

**Absolute EAG Responses** All four stereoisomers of carvone epoxide elicited positive antennal responses in the four species (i.e. *E. nigrita*, *E. niveofasciata*, *E. atleticana*, and *E. marcii*) with the highest responses found towards (–)-*trans*-carvone epoxide (LMM substance differences:  $F_{3,657} = 8.98$ ,  $P < 0.001$ ). Depending on species and stereoisomer, positive responses were detected either at the highest concentrations of the stereoisomers tested ( $10^{-4}$  and  $10^{-2}$ ) or additionally at a lower concentration of  $10^{-6}$ . There were no positive responses to concentrations lower than  $10^{-6}$  in any tested species (Fig. 3). Within *E. atleticana* and *E. niveofasciata*, but not within the other species, there were differences among the four stereoisomers at a given concentration (LMM overall differences species x substance:  $F_{9,657} = 4.47$ ,  $P < 0.001$ ); see also Fig. 3). In particular, for these two species antennal responses to the (–)-*trans*-carvone epoxide were significantly stronger than those to (–)-*cis*-carvone epoxide.

**Relative EAG Responses** The overall strength (intercept of regression lines) of antennal responses to the different stereoisomers was similar among all bee species (main test LMM species x substance:  $F_{9,132} = 1.86$ ,  $P = 0.057$ ). However, antennal sensitivity of *Eulaema* species to each stereoisomer of carvone epoxide differed significantly (main test LMM substance:  $F_{3,132} = 7.38$ ,  $P < 0.001$ ; Fig. 4). In *E. nigrita* only, the

responses to (–)-*trans*-carvone epoxide were stronger than to the other tested stereoisomers.

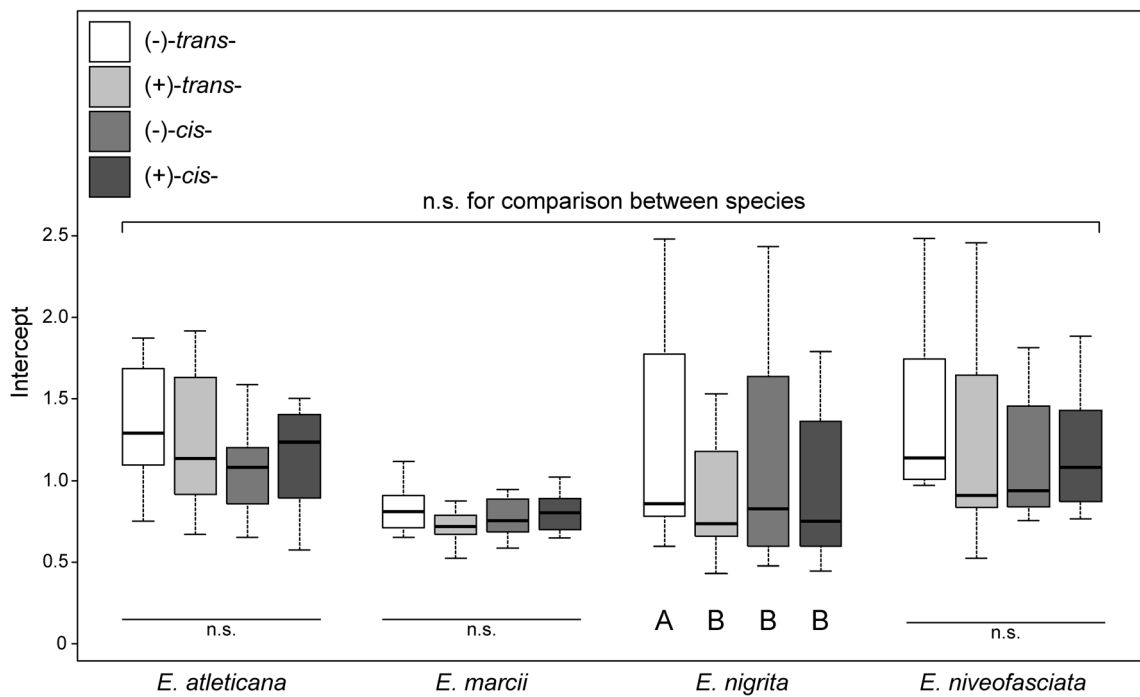
**Field Assays** In total 235 euglossine males were attracted by carvone epoxides and no bee responded to the solvent control *n*-hexane. Beside euglossine bees, no other insect responded to the compounds. Among the four *Eulaema* species occurring in the study area, only *E. marcii* was not attracted by any of the stereoisomers (Fig. 5). For the other three tested bee species, the attractiveness differed significantly among the stereoisomers (PERMANOVA: pseudo- $F_{12,479} = 11.68$ ,  $P < 0.001$ ). Both *trans*-carvone epoxide enantiomers always attracted more bees than the solvent control, with the (–)-*trans*-stereoisomer being more attractive than the (+)-*trans*-stereoisomer in two of the species (*E. atleticana*, *E. nigrita*). With the exception of *E. nigrita*, none of the *cis*-carvone epoxide enantiomers attracted more bees than the solvent control (Fig. 5).

## Discussion

In our study, we found that 1) all tested *Catasetum* species emit only the (–)-*trans*-stereoisomer of carvone epoxide, 2) within *E. atleticana* and *E. niveofasciata* antennal responses to the (–)-*trans*-carvone epoxide were significantly stronger than those to (–)-*cis*-carvone epoxide, 3) the strength and pattern of antennal responses to all 4 stereoisomers (separately tested) did not differ among *Eulaema* species, and 4) there were significant differences in attractiveness of the four stereoisomers of carvone epoxide to the bee species with the (–)-*trans*-stereoisomer being particularly attractive.

Our enantioselective analyses confirm previous studies (Lindquist et al. 1985, Whitten et al. (1986) by showing that the (–)-*trans*-stereoisomer of carvone epoxide is a key component in the scent profile of perfume-rewarding plants. The absence of relevant variation in the isomeric pattern of carvone epoxide among such plants suggests that reproductive isolation by selective attraction of euglossine bees is not mediated by different isomeric patterns of this compound. Instead, plants with carvone epoxide in their profile also seem to selectively attract their specific euglossine pollinators by complex blends of volatiles consisting of a few major compounds and a myriad of minor constituents, as suggested for other plant species (Dodson et al. 1969; Whitten et al. 1986). Indeed, plants producing carvone epoxide also release a high number of other components, including chiral ones (see e.g. Kaiser 1993; Milet-Pinheiro et al. 2015b; Whitten et al. 1988; Williams and Whitten 1983).

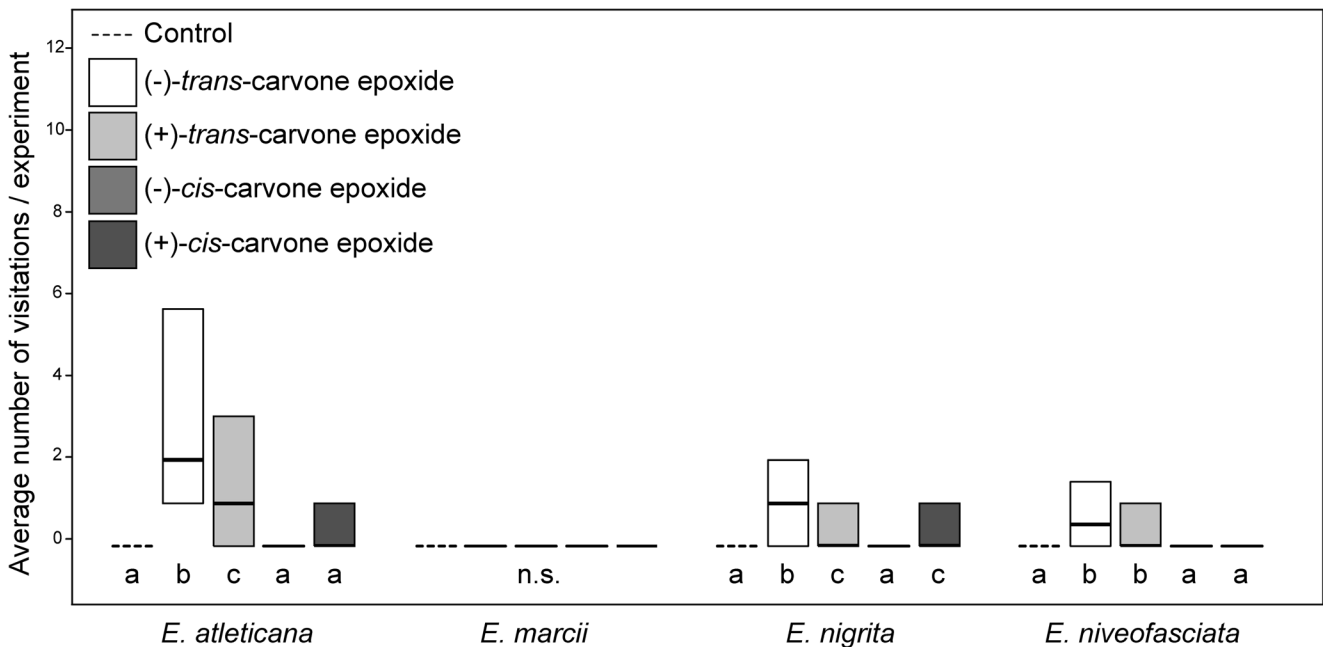
The emission of (–)-*trans*-carvone epoxide is commonly found within several genera (including *Catasetum*) of different plant families and is considered to be a result of evolutionary convergence in floral reward of perfume-rewarding plants



**Fig. 4** Intercept of regression lines (strengths of responses) of standardized EAG responses for tested bee species (Median, 25% - 75%, confidence interval of 95%). Different letters indicate significant differences among tested stereoisomers within a species

(Whitten et al. 1986). This underlines the function of (–)-trans-carvone epoxide as main attractive signal for a selective subset of euglossine visitors (Whitten et al. 1986) and its possible role as a private communication channel for euglossine bees (Raguso 2008; Whitten et al. 1986). This is supported by our finding that only euglossine males are attracted by carvone

epoxides. Although we observed a few *Euglossa* individuals on scent baits with carvone epoxide [mainly on the (–)-trans] in the field assays, the universal presence of this compound in floral perfumes of *Catasetum* species pollinated by *Eulaema*, as well as its absence in floral perfumes of *Euglossa*-pollinated species (Milet-Pinheiro and Gerlach



**Fig. 5** Number of visits to the four carvone epoxides and the solvent control (Median, 25% - 75%). Different letters indicate significant differences in visits within bee species to the four different stereoisomers and to

the solvent control. “n.s.” indicates that *E. marcii* responded similarly to the different stereoisomer of carvone epoxide and the solvent control



2017), suggest that this compound evolved as a specific attractant of *Eulaema* bees as pollinators.

In comparison with other specialist bees, e.g. oligoleges, that are able to perceive components involved in host finding at concentrations as low as  $10^{-8}$  (see e.g. Brandt et al. 2017), the antennae of *Eulaema* bees to carvone epoxides do not seem to be that sensitive. Though comparisons of sensitivity among bees to different compounds in EAG measurements have to be considered with caution, it is interesting that the volatile host recognition cues of oligolectic bees are only produced in trace amounts (Milet-Pinheiro et al. 2015a), whereas carvone epoxide is typically produced in high amounts by *Eulaema*-pollinated perfume-producing plants (Armbruster 1989; Milet-Pinheiro and Gerlach 2017; Schwerdtfeger et al. 2002; Whitten et al. 1986). Therefore, a high sensitivity to carvone epoxide might not be required for effective communication between pollinators of the genus *Eulaema* and their perfume sources.

One interesting aspect in our study was the fact that male bees of the species *E. marcii* were not attracted by any of the tested carvone epoxide stereoisomers in the field assays, although the electroantennographic measurements revealed that they are able to perceive these compounds starting at a concentration of  $10^{-4}$ . These findings illustrate that antennal sensitivity is not always correlated with the attractiveness of a compound (see e.g. Gemeno et al. 2003; Jhumur et al. 2007). In contrast to what we observed for *E. marcii*, the results for the other three tested bee species (*E. atleticana*, *E. nigrita* and *E. niveofasciata*) demonstrate a relationship between antennal detection of the different carvone epoxide stereoisomers and their attractiveness to bees. Both the antennal responses and the results of the behavioral field assays point to a preference for (–)-*trans*-carvone epoxide over the other tested stereoisomers. This preference might be even more pronounced, because the purity of the different carvone epoxides we tested was different. If bees were indeed more sensitive to (–)-*trans*-carvone epoxide, than responses to (+)-*trans* and (+)-*cis* isomers, which contained about 7–8% of the (–)-*trans* isomer (see Table 1), might generally have been amplified. In the case of the (–)-*cis*- isomer, which did not contain (–)-*trans*-carvone epoxide, significant antennal and behavioral differences within a tested species were always found in relation to the (–)-*trans*-carvone epoxide. Irrespective of this potential bias, our findings of the antennal responses and the field assays are in agreement with the study of Schorkopf et al. (2011), who demonstrated that antennal sensitivity to ipsdienol enantiomers correlate with its attractiveness. More precisely, male *Euglossa cyanura* did not only exhibit a behavioral preference for (–)-ipsdienol, but the antennae also showed stronger electroantennographic responses to the (–)- than to the (+)-isomer. Furthermore, the findings of our electroantennographic measurements and the results of our field assays suggest that *Eulaema atleticana*, *E. nigrita*,

and *E. niveofasciata* possess different olfactory receptors for detecting distinct stereoisomers. If bees would have just one receptor for detecting the various carvone epoxides, we would have expected that their attractiveness is the same. In other plant-insect interactions where chiral compounds are involved, there are examples for both cases, i.e. different stereoisomers are detected by the same or different receptor(s) (e.g. Schneider et al. 2013).

Altogether our findings suggest that (–)-*trans*-carvone epoxide is the key stereoisomer in *Eulaema*-pollinated plant species releasing carvone epoxide. It is very likely that this stereoisomer is responsible for the attraction of various, but not all *Eulaema* species. Stereoisomeric patterns of this compound are very similar within *Catasetum* meaning that plants do not split niches by emitting different enantiomers of carvone epoxide. Therefore, reproductive isolation among species of this genus (and other perfume-rewarding plants) might rather be acquired through other isolating mechanism, such as different blooming times, geographical distribution and specific placement of pollinarium at pollinators' body (Hills et al. 1972). Future investigations elucidating the absolute configuration of other chiral compounds and testing the selective attractiveness of different stereoisomers/stereoisomeric patterns to euglossine bees might reveal whether stereochemistry is involved in reproductive isolation of perfume-rewarding plants.

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