




# Effect of KNN doping on the dielectric properties of BaTiO<sub>3</sub> lead-free ceramics

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## ABSTRACT

In this study, the phase structure, microstructure, and dielectric properties of BT-xKNN ceramic prepared by traditional solid-phase method were studied. The diffraction peaks of KNN-doped BaTiO<sub>3</sub> ceramics shifted to high angles. Excessive KNN doping inhibited BaTiO<sub>3</sub> grain growth. KNN doping improved the dielectric constant of BaTiO<sub>3</sub> ceramics. When the amount of KNN doped was 0.5 wt%, the dielectric constant can reach 19,359.4. When KNN was doped at 1wt%, the dielectric peak was broadened and the dielectric temperature stability was good. However, the incorporation of KNN increased the dielectric loss of BaTiO<sub>3</sub> ceramics, which may be caused by the increase in conduction loss caused by Nb<sup>5+</sup>. Moreover, the incorporation of KNN did not increase the Curie temperature of BaTiO<sub>3</sub> ceramics, but decreased it, which may be related to the decomposition of KNN during high-temperature sintering.

## 1 Introduction

In recent years, with the rapid growth of the electronic information industry, it has promoted the rapid development of the electronic components industry [1]. As the base material of electronic components, electronic ceramics have increased demand for use and more and more applications [2, 3]. BaTiO<sub>3</sub>-based ceramics are a kind of electronic ceramics with many applications and are typical [4]. BaTiO<sub>3</sub> replaces PbTiO<sub>3</sub>-based ferroelectric materials that pollute the environment in the early days and become an environmentally friendly ferroelectric

ceramic material that meets the requirements of sustainable development [5], due to its excellent electrical properties, such as ferroelectricity, dielectric properties, piezoelectricity, and pyroelectricity. It has a wide range of applications in the fields of capacitors, energy storage devices, transducers, and actuators [6]. However, with the change of temperature, the dielectric constant of pure-phase BaTiO<sub>3</sub> ceramics changes greatly. Therefore, it is necessary to optimize its performance to obtain ceramics with high dielectric constant and high-temperature stability.

The research on BaTiO<sub>3</sub> mainly uses doped solid solution to improve the performance of BaTiO<sub>3</sub>

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ceramics. Wen, Yu, and Liu used the solid-phase method and the sol-gel method to study and analyze the  $(1-x)\text{BaTiO}_3\text{-}x\text{BiAlO}_3$  multiphase ceramic system. The study found that the samples all exhibited relaxation ferroelectric properties and with the increase of the amount of doping, the dielectric peak of the system gradually broadens and the relaxivity increases [7–9]. Khare studied the influence of holding time on the dielectric properties of  $0.9\text{BaTiO}_3\text{-}0.1\text{CaCu}_3\text{Ti}_4\text{O}_{12}$  ceramics and found that the dielectric constant of the sample held for 3 h under the condition of low frequency 1 kHz is the largest and the corresponding loss value is also the smallest [10]. Jiang used the solid-phase method of dual-ion  $\text{Al}^{3+}$  and  $\text{Nb}^{5+}$ -co-doped  $\text{BaTiO}_3$  ceramics. Due to the difference in valence between substituted and substituted ions, more vacancy defects are formed in the system, which affect the final dielectric properties of the system [11]. Zhao and Zhang prepared  $(1-x)\text{BaTiO}_3\text{-}x\text{CuO}$  ceramics by a solid-phase method. The study showed that with the increase of copper oxide doping content, the dielectric constant of the system showed a trend of first increasing and then decreasing. When it is 0.01, the relative dielectric constant at room temperature is 2040 and the dielectric loss value is 0.8% [12, 13].

In recent years, with the promulgation of environmental regulations, high-performance lead-free dielectric materials have been paid more and more attention in the context of overall research [14]. The most researched materials are  $\text{K}_{0.5}\text{Na}_{0.5}\text{NbO}_3$  (KNN) based,  $\text{BaTiO}_3$  (BT) based,  $\text{Bi}_{0.5}\text{Na}_{0.5}\text{TiO}_3$  (BNT) based, bismuth layered structure and tungsten bronze structure, and other ceramics [15–18]. Although BT-based ceramics have excellent dielectric properties, their lower Curie temperature severely limits its application at high temperatures [16]. KNN-based ceramics have a high Curie temperature (420 °C) and in a low-intensity driving electric field, KNN ceramics can obtain high unipolar strain and narrow strain hysteresis [15]. Therefore, this article intends to improve the performance of BT-based ceramics by doping KNN.

## 2 Experimental part

The experimental process of preparing  $\text{BaTiO}_3$  (BT) powder and  $\text{K}_{0.5}\text{Na}_{0.5}\text{NbO}_3$  (KNN) powder by solid-phase method is the same, only the pre-sintering

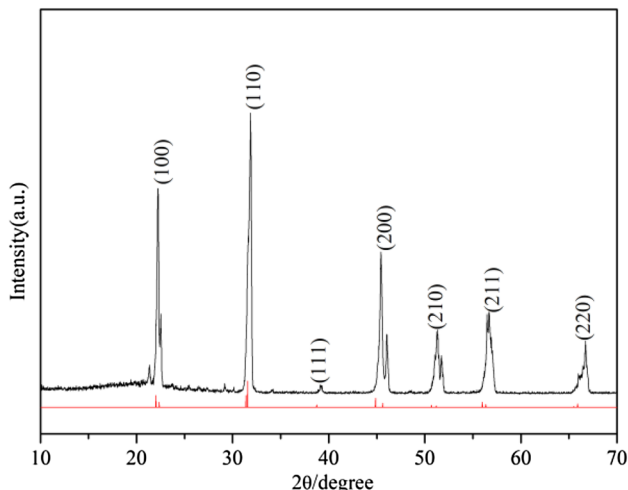
temperature and time are different. Consulting the relevant literature and the related literature [19–22],  $\text{BaTiO}_3$  uses  $\text{BaCO}_3$  and  $\text{TiO}_2$  as raw materials in stoichiometric proportions, wet ball milling for 24 h, drying at 70 °C, and then sintering the mixture obtained after passing through a 100-mesh sieve at 1100 °C for 4 h to obtain  $\text{BaTiO}_3$  powder.  $\text{K}_{0.5}\text{Na}_{0.5}\text{NbO}_3$  (KNN) uses  $\text{K}_2\text{CO}_3$ ,  $\text{Na}_2\text{CO}_3$ , and  $\text{Nb}_2\text{O}_5$  as raw materials and mixes the materials according to the above method. After drying and sieving, they are pre-fired at 850 °C for 2 h to obtain the required powder. Through XRD pattern analysis,  $\text{BaTiO}_3$  powder and KNN powder with good crystallinity were obtained. According to the mass percentage ratio BT-xKNN ( $x = 0.1, 0.3, 0.5, 0.7, 1$  wt%), ball milling for 12 h, after drying and sieving, adding an appropriate amount of 5 wt% PVA to the powder to carry out granulation, then forms a tablet under a pressure of 6–10 MPa, and then sinters it into porcelain at a certain sintering temperature. After polishing and cleaning, the silver paste is coated and fired and the obtained samples are tested for dielectric properties.

The phase structure test of the ceramic sample was carried out with the Rigaku Corporation D/max2500V/pc X-ray diffractometer and the surface of the ceramic sample was tested with the Hitachi S-4800 scanning electron microscope (SEM). Agilent E4980A high-temperature dielectric temperature spectrum test system performs dielectric temperature spectrum test on samples and uses 6500B impedance analyzer to analyze the influence of frequency on dielectric properties.

## 3 Results and discussions

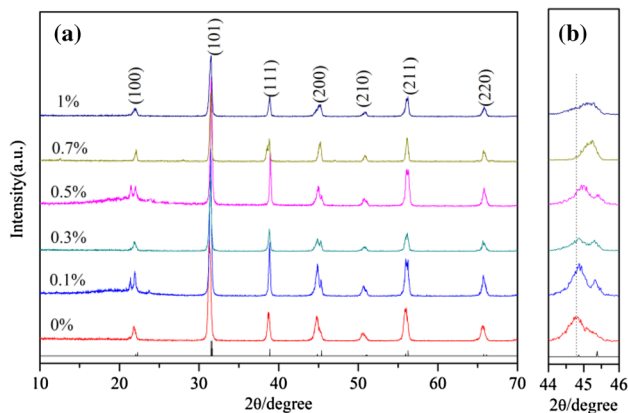
Figure 1 is the XRD pattern of the KNN powder material calcined at 850 °C for 12 h. Through the analysis of Jade software, it is found that the powder sample prepared in the experiment has the same spectrum as the  $\text{NaNbO}_3$  powder PDF card (NO.73-1106), which proves that the pure-phase KNN crystal is synthesized and the main crystal phase diffraction peak is sharp, which proves that the pure-phase KNN powder has good crystallinity.

Figure 2 shows the XRD pattern of BT-KNN-based ceramics sintered at 1350 °C for 4 h. In order to show the effect of KNN doping on the phase structure of  $\text{BaTiO}_3$  ceramics more clearly, the diffraction peaks



**Fig. 1** XRD pattern of KNN powder calcined at 850 °C for 3 h

near the (200) crystal plane are magnified. It can be seen from Fig. 2a that the BaTiO<sub>3</sub> doped with KNN still maintains the perovskite structure, which is similar to the PDF card (NO.05-0626) spectrum of BaTiO<sub>3</sub> powder and there is no miscellaneous phase. The (101) crystal plane diffraction peak is sharp and the crystallinity is very good, but its peak height is different. Doping KNN less than 0.5% can enhance the main peak height and increase the crystallinity. From Fig. 2b, it can be seen that the BaTiO<sub>3</sub> ceramic (200) doped with KNN is offset in the high-angle direction. This may be because KNN is incorporated into the BaTiO<sub>3</sub> lattice in the form of K<sup>+</sup>, Na<sup>+</sup>, and Nb<sup>5+</sup> ions. Their ionic radius are K<sup>+</sup>(0.138 nm), Na<sup>+</sup>(0.102 nm), Nb<sup>5+</sup>(0.064 nm), and Ba<sup>2+</sup>(0.135 nm). According to Bragg Eq. ( $2d\sin\theta = n$ ), the crystal plane spacing  $d$  decreases and then the diffraction angle  $\theta$



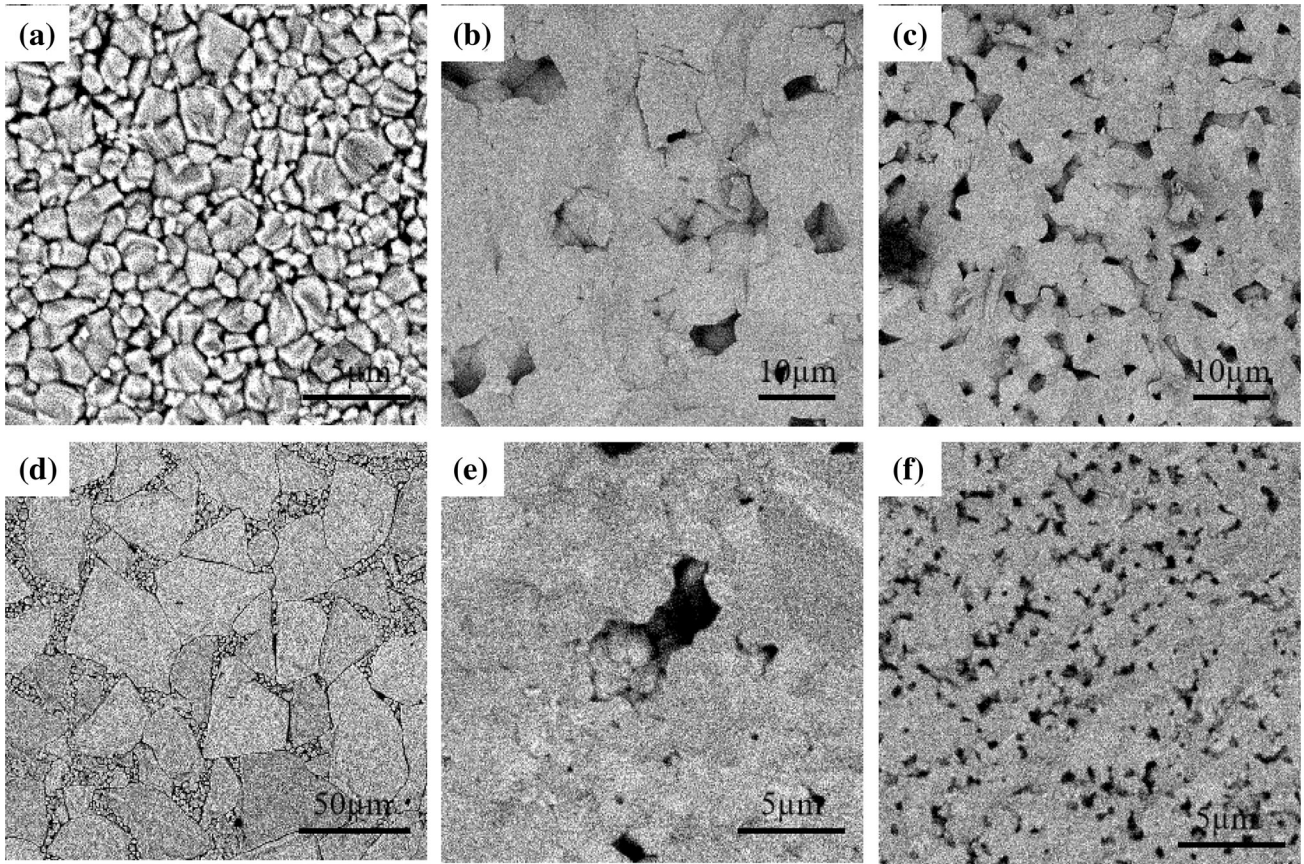
**Fig. 2** **a** XRD pattern of BT-xKNN-based ceramics; **b** Enlarged XRD patterns in the range of 2θ from 44° to 46°

will increase large. So the ions with a smaller radius are incorporated into the BaTiO<sub>3</sub> lattice and the diffraction peaks shift to the high-angle direction [23, 24].

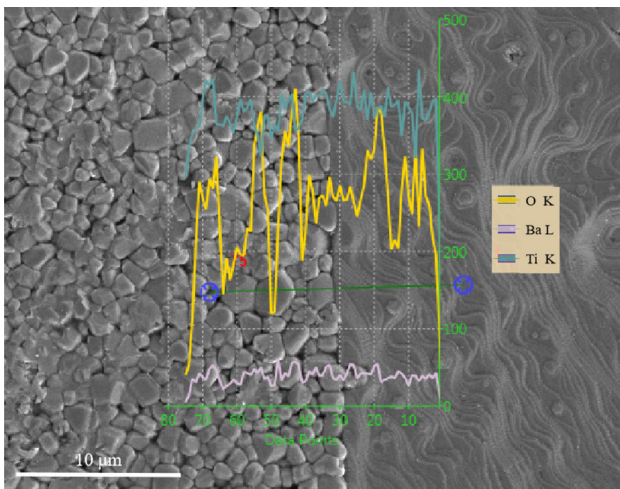
Figure 3a–f is the SEM spectra of BT-KNN-based ceramic samples, where a–f is the amount of KNN doped, which is 0%, 0.1%, 0.3%, 0.5%, 0.7%, and 1%. Figure 4 is the EDS energy spectrum analysis diagram of the BT-0.5% KNN-based ceramic sample. It can be seen from Fig. 3 that with the increase of KNN doping, the grain size of BaTiO<sub>3</sub> ceramics first increases and then decreases. Pure-phase BaTiO<sub>3</sub> ceramics have a grain size of about 3 μm. When 0.1% KNN is incorporated, the grain size is about 10 μm. When 0.3% KNN is incorporated, the grain size is about 5 μm. When 0.5% KNN is doped, combined with the analysis of EDS energy spectrum when 0.5% KNN is doped in Fig. 4, two crystal grains appear. The small crystal grain size is similar to that of pure-phase BaTiO<sub>3</sub>, and the crystal grain size is about 3 μm. The large grain size is about 50 ~ 70 μm. When doped with 0.7% KNN, the grain size is about 3 μm. When doped with 1% KNN, it can be clearly seen that the grain size is already less than 1 μm.

Table 1 shows the density change of BaTiO<sub>3</sub> ceramics with different amounts of KNN, and Fig. 5 shows the density and shrinkage changes of BT-xKNN ceramics. It can be seen from Fig. 5 that as the amount of KNN added increases, the density of BaTiO<sub>3</sub> ceramics first increases and then decreases. When the doping amount of KNN is 0.1%, the density is 5.8028 g/cm<sup>3</sup>, which is the highest, and the incorporation of a large amount of KNN inhibits the crystal growth and reduces the density. This corresponds to the appearance of the ceramic in Fig. 3.

Table 2 shows the Curie temperature and dielectric constant of BT-xKNN ceramics. It can be seen from Table 2 that the Curie temperature of pure-phase BaTiO<sub>3</sub> is 123.8 °C and the dielectric constant is 2441.5. With the incorporation of KNN, the Curie temperature decreases while the dielectric constant increases. The reason for the increase of the dielectric constant is that the donor element is doped to release the charge and the dipole increases, which leads to the increase of the dielectric constant [25]. The actual Curie temperature of the KNN ceramic material is 420 °C. In theory, the high Curie temperature should increase the Curie temperature of BaTiO<sub>3</sub>, but actually lowers the Curie temperature. This may be because K<sub>2</sub>O and Na<sub>2</sub>O volatilize at around 900 °C



**Fig. 3** SEM spectrum of BT-xKNN-based ceramics with **a**  $x = 0\%$ ; **b**  $x = 0.1\%$ ; **c**  $x = 0.3\%$ ; **d**  $x = 0.5\%$ ; **e**  $x = 0.7\%$ ; **f**  $x = 1\%$



**Fig. 4** EDS energy spectrum analysis of BT-0.5% KNN-based ceramic samples

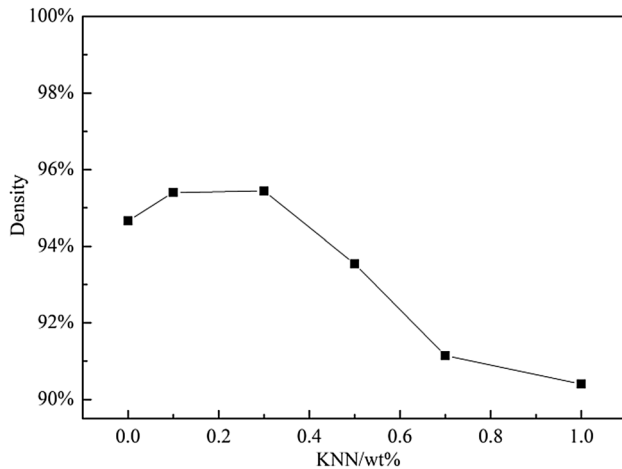
and the temperature is too high, which leads to the imbalance of the KNN stoichiometric ratio. Figure 6 shows the KNN phase diagram. From the KNN phase diagram, it can be seen that the phase structure is very complex and the stoichiometric ratio is very

**Table 1** Density changes of BaTiO<sub>3</sub> ceramics with different KNN contents

KNN/(wt%)	Density/(g·cm <sup>-3</sup> )	Density (%)
0 wt%	5.7553	94.66
0.1 wt%	5.8028	95.44
0.3 wt%	5.8002	95.40
0.5 wt%	5.6870	93.54
0.7 wt%	5.5412	91.14
1 wt%	5.4962	90.40

complicated. The imbalance caused by cooling and crystallization may not be the material with high Curie point. The decrease in Curie temperature may also be due to the fact that KNN is incorporated into BaTiO<sub>3</sub> in the form of K<sup>+</sup>, Na<sup>+</sup>, and Nb<sup>5+</sup> into the BaTiO<sub>3</sub> lattice. The ionic radius of these elements is smaller than that of Ba<sup>2+</sup>, and the decrease in  $c/a$  reduces the Curie temperature.

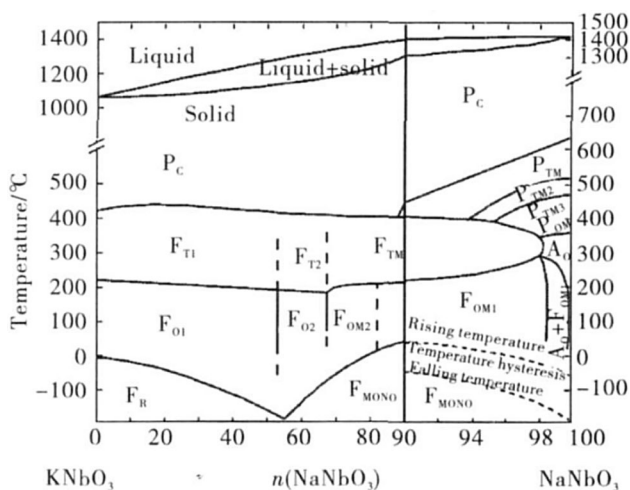
Figure 7 is a graph showing the change in dielectric constant versus temperature and the change in



**Fig. 5** BT-xKNN ceramic density change graph

**Table 2** BT-xKNN ceramic Curie temperature and dielectric constant value

KNN/(g·cm <sup>-3</sup> )	$T_c/^\circ\text{C}$	$\epsilon$
0 wt%	123.8	2441.5
0.1 wt%	119.3	3578.34
0.3 wt%	114.3	3620.72
0.5 wt%	110.2	19,359.4
0.7 wt%	114.3	3637.6
1 wt%	111.2	2365.6



**Fig. 6** KNN phase diagram

dielectric loss versus temperature of BaTiO<sub>3</sub> ceramics with different KNN doping levels. It can be seen from Fig. 7a that the dielectric constant of pure-phase BaTiO<sub>3</sub> will increase stepwise near the Curie temperature and a dielectric peak appears. This is

because BaTiO<sub>3</sub> undergoes a ferroelectric–paraelectric transition near the Curie temperature. The BaTiO<sub>3</sub> crystal form changes from a tetragonal phase to a square phase and the polarization direction changes and spontaneous polarization increases [25]. When the amount of KNN incorporated is less than 0.5%, it can be found that the dielectric peak is enhanced relative to the pure-phase BaTiO<sub>3</sub>, which may be beneficial to the increase in the lift-to-drag ratio of PTC ceramics. As KNN continues to increase, the Curie peak appears to have a voltage peak, the dielectric peak is broadened, and the stability of the dielectric constant is improved. It can be seen from Fig. 7b that the dielectric loss after doping KNN is greater than the dielectric loss of pure-phase BaTiO<sub>3</sub> ceramics, which may be caused by the increase of conductivity loss when KNN is doped.

Figure 8 is the change of dielectric constant with frequency and the change of dielectric loss with frequency of BaTiO<sub>3</sub> ceramics with different KNN doping levels. It can be seen from Fig. 8a that when the amount of KNN incorporated is 0.5%, the dielectric constant is much larger than that of other groups and the dielectric constant decreases with the increase of frequency. This may be because the samples with 0.5% KNN are mainly space charge polarization. This corresponds to Fig. 8b, which is dominated by conductance loss and large dielectric loss. The other component samples are mainly relaxed polarization.

## 4 Conclusion

BT-xKNN ( $x = 0, 0.1, 0.3, 0.5, 0.7, 1\text{wt}\%$ ) ceramics were prepared by solid-phase method. The results show that the diffraction peaks of KNN-doped BaTiO<sub>3</sub> ceramics shifted to high angles and the grain size first increased and then decreased with the increase of KNN doping. The Curie temperature of the BaTiO<sub>3</sub> ceramic doped with KNN decreases and the dielectric constant increases. But the dielectric loss also increases. When the amount of KNN incorporated is 0.5wt%, the minimum Curie temperature is 110.2 °C and the maximum dielectric constant is 19359.4. With the increase of KNN doping amount, a pressure peak phenomenon appears, the dielectric peak is broadened, and the temperature stability of the dielectric constant is improved.

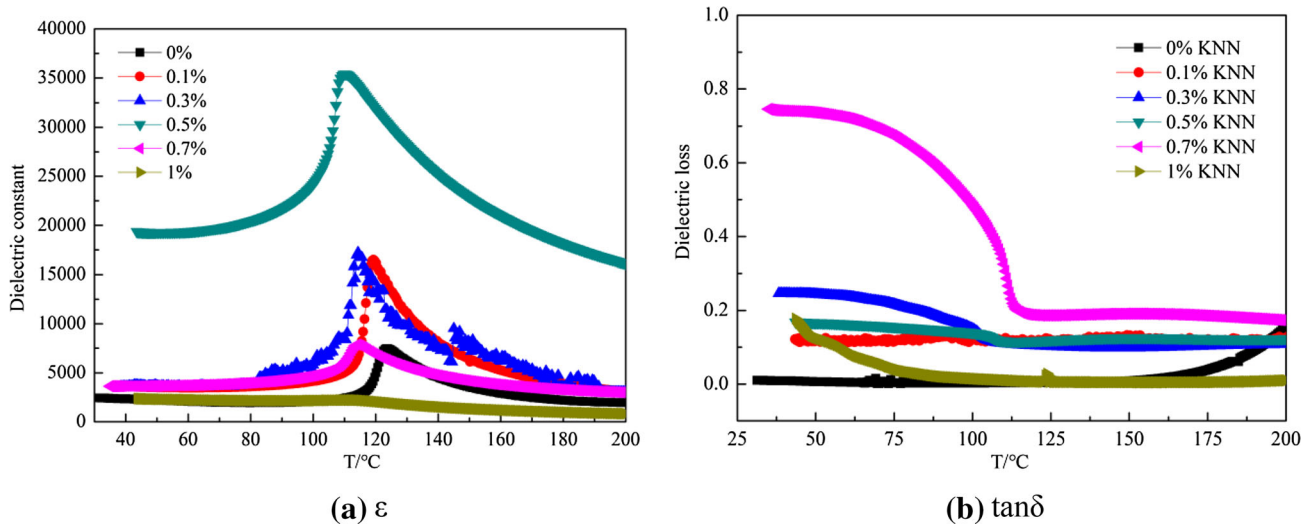


Fig. 7 Dielectric thermogram of BT-xKNN ceramic sample with a  $\epsilon$ ; b  $\tan \delta$

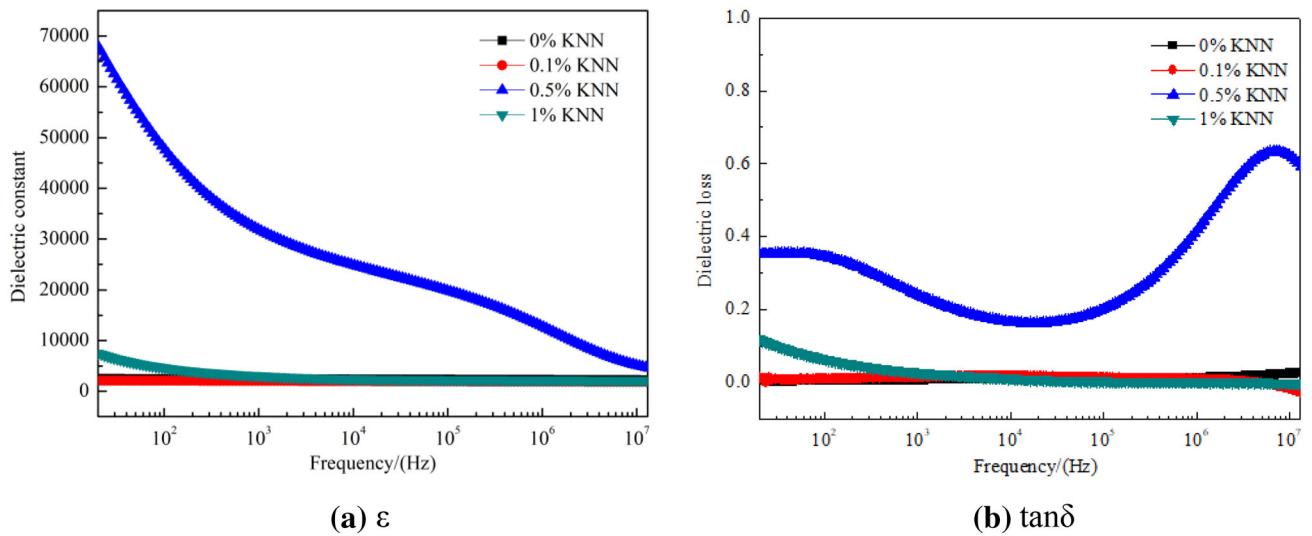


Fig. 8 Dielectric spectrum of BT-xKNN ceramic sample with a  $\epsilon$ ; b  $\tan \delta$

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### Declarations

**Conflict of interest** The authors declare that they have no known competing financial interest or personal relationships that could have appeared to influence the work reported in this paper.

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