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Effect of Gd and Si co-doping on the band alignment and electrical properties of HfO₂ dielectric films prepared by atomic layer deposition

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ABSTRACT

Gd and Si co-doped HfO₂ gate dielectric thin films were prepared by atomic layer deposition (ALD), while Gd[N(SiMe₃)₂]₃, Hf[NEtMe]₄, and H₂O are chosen to be precursors. The Gd and Si were successfully co-doped into HfO₂ films using only one doping precursor Gd[N(SiMe₃)₂]₃. The doping concentration can be facilely tuned by controlling ALD recipe. The atomic percentages of Si/ (Si + Gd + Hf) and Gd/(Si + Gd + Hf) increase from 11.5 to 28.9% and from 6.8 to 28.4% when changing the ALD cycle ratio of $Gd_rSi_{\nu}O$ to HfO₂ from 1:9 to 1:1. The band gap and band alignment were investigated by X-ray photoelectron spectroscopy. The results imply that the band gap of Gd/Si co-doped HfO₂ films has a positive relation with doping concentration. Moreover, the valence band offset decreases with doping concentration first but then increases, while the change of conduction band offset is opposite. The (1:6)-Hf_xGd_{ν}Si_zO films with 11.6 at.% Gd/(Gd + Hf) exhibit the maximum accumulation capacitance and dielectric constant, which are only slightly smaller than those of the HfO₂ films. Compared to HfO₂ films, the leakage current density of (1:6)-Hf_xGd_ySi_zO films is decreased by at least one order of magnitude. Therefore, Gd and Si co-doping can improve the electrical properties of HfO₂ films.

1 Introduction

Since portable electronic products come into public daily life, the dimension of Si-based metal–oxide– semiconductor field-effect-transistor (MOSFET) devices keeps continuous scaling down following *Moore's law* [1]. In the past decades, SiO₂-based gate dielectric was employed in the mass production of MOSFETs [2, 3]. However, the physical thickness of this insulator layer has been decreased to its physical

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limit, leading to power consumption and reliability issues due to the quantum tunneling effect [4, 5]. To solve the above issues, one of the promising ways is replacing the SiO₂ by a physically thicker layer with a higher dielectric constant dielectric films [4, 6]. Various metal oxides have been applied as high-k gate dielectrics, including ZrO_2 [7], Y_2O_3 [8], HfO_2 [9], Ta₂O₅ [10], Al₂O₃ [11], and MgO [12]. Besides inorganic materials, a kind of organic-inorganic hybrid films as high-k dielectrics can also exhibit excellent properties, such as SiO₂-TiO₂-PVP and BaTiO₃-PMMA-PVP, which show great application potential in flexible electronics [13, 14]. Among various high-k gate dielectric films, HfO2 with high dielectric constant has been demonstrated to be the most capable substitute for SiO₂ to improve device performance [15–17]. Normally, gate dielectric films are expected to remain stay amorphous after a conventional activation annealing (~1000 °C), concerning that grain boundaries may serve as the paths for dopant diffusion and current leakage. Nevertheless, the crystallization temperature of pure HfO₂ is too low $(\sim 500 \text{ °C})$ [18, 19]. Si doping has been reported to suppress the crystallization of HfO₂; however, Si dopant might deteriorate the dielectric constant of HfO_2 films [20, 21]. On the other hand, it has been demonstrated that Gd doping can enhance the permittivity of HfO₂ films [22, 23]. Therefore, co-doping of Gd and Si might be an effective approach to enhance comprehensive electrical properties of HfO₂ films.

Atomic layer deposition (ALD) is one of the promising thin film deposition techniques, which is based on sequential self-limited and complementary surface chemisorption reactions. ALD possesses unique advantages such as large area uniformity, three-dimensional conformality, precise and facile thickness control down to sub-nanometer, low processing temperature, and extremely low damage to the substrate surface [24-26]. Therefore, ALD has attracted great attention in the surface engineering and nanostructure fabrication in recent years [27–29]. Amorphous gate dielectric films are preferred because grain boundaries can act as diffusion path for impurity and leakage current. Furthermore, smoother surfaces and associated improved interface quality can be achieved using amorphous gate dielectric films [5, 30]. ALD has been demonstrated widely to produce amorphous gate dielectric films with excellent interface; hence, ALD is one of the leading

technologies for the high dielectric constant gate insulators deposition [31-33]. The use of Si-containing rare-earth (RE) precursors of tris[bis(trimethylsilyl)amino]RE $\{RE[N(SiMe_3)_2]_3,$ where RE = La[34-36], Pr [37], Lu [38], and Me = CH₃ has been reported to prepare RE silicate by ALD, which is a promising way to intentionally incorporate Si in rareearth oxides. Therefore, Gd and Si co-doped HfO₂ $(Hf_rGd_{\nu}Si_2O)$ films were deposited by ALD, and $Gd[N(SiMe_3)_2]_3$ was the only doping precursor in this work. The doping concentration tuning can be achieved by changing the ALD cycle ratio of Gd_xSi_yO to HfO₂. The effect of doping concentration of Gd and Si on the band alignment and electrical characteristics of the doped HfO₂ films were investigated systematically.

2 Experimental section

Substrates used in this work were B-doped *p*-type Si (100) with resistivity of 1–10 Ω cm. Prior to ALD dielectric films deposition, they were first degreased in ethanol for 5 min by sonication. Then, the Si surface native oxide was removed with 2% diluted hydrofluoric acid solution for 3 min. Finally, the substrates were blown dry with N₂ after rinsing by de-ionized water. The substrates were then transferred into the ALD chamber (Picosun SUNALETM R200 Advanced, Finland) to deposit the $Hf_xGd_ySi_zO$ films at 300 °C. The Hf_xGd_ySi_zO ALD process was performed by combining HfO2 ALD process, using Hf[NEtMe]₄ and H₂O, with the Gd_xSi_yO ALD process, using Gd[N(SiMe₃)₂]₃ and H₂O. Pure N₂ (99.999%) was used as both carrier gas and purge gas. During deposition process, the temperature of $Hf[NEtMe]_4$ and $Gd[N(SiMe_3)_2]_3$ precursors was set as 120 °C and 180 °C to ensure sufficient vapor pressure, while H₂O was kept at room temperature. Pulse time of all precursors was 0.3 s, after which was a 6 s N₂ flow step to blow reaction by-products and redundant precursors away. The composition of $Hf_xGd_ySi_zO$ films was tuned via changing the ALD recipe. Eight kinds of Hf_xGd_ySi_zO samples with various compositions were prepared by depositing the alternate layers of Gd_xSi_yO and HfO_2 with Gd_xSi_yO as the beginning layer. For example, (1:N)-Hf_xGd_ySi_zO films were using Gd_xSi_yO (1 cycle) + HfO₂ (N cycles, *N* = 1, 2, 3, 4, 5, 6, 7, 9) as one loop. The loops were varied to control the thickness of Hf_xGd_ySi_zO films as ~ 10.7 nm. Pure Gd_xSi_yO and HfO_2 films were also prepared for comparison. The related process details are presented in Table 1. The entire sample preparation process was conducted in the clean room.

The chemical composition of Hf_xGd_ySi_zO films were measured by X-ray photoelectron spectroscopy (XPS, Thermo Fisher K-Alpha) with a monochromatic Al K α source (*hv* = 1486.6 eV). The excited photoelectrons were collected at a takeoff angle of 90°. The binding energy scale was calibrated using C1s peak at 284.6 eV. In addition, the valence band spectra and the O1s electron energy loss spectra of the Hf_xGd_{y-1} Si₂O films were also measured by XPS to achieve the band alignment information. Pt top gate electrodes with an area of 1.54×10^{-4} cm² were fabricated on the surface of the Hf_rGd_uSi₂O films using a shadow mask by sputtering. Capacitance–voltage (C-V) and leakage current density-voltage (J-V) characteristics tests were conducted by a Keithley 4200 semiconductor characterization system at room temperature in a closed Cascade Summit 11000B-M probe station.

3 Results and discussion

The Hf_xGd_ySi_zO ALD process is illustrated in Fig. 1a, where the composition of Hf_xGd_ySi_zO films can be facilely tuned by controlling the ratio of Gd_xSi_yO to HfO₂ ALD cycles. XPS was applied to determine the composition of various Hf_xGd_ySi_zO films, and the typical XPS survey spectrum for (1:6)-Hf_xGd_ySi_zO film is presented in Fig. 1b. The C1s, Hf 4f, Gd 4d, Si 2p, and O1s peaks located at 284.6 eV, 17.5 eV, 143.2 eV, 102.2 eV, and 530.6 eV can be easily discerned, confirming that both Gd and Si from Gd[N(SiMe₃)₂]₃ source can be doped into HfO₂ films. Furthermore, the chemical composition of various Hf_xGd_ySi_zO films gained from XPS are summarized in Table 1. Figure 1c plots the relationship between the doping concentration (Gd/(Gd + Si + Hf) at.% and Si/(Gd + Si + Hf) at.%) and ALD cycle ratio of $Gd_xSi_yO/(Gd_xSi_yO + HfO_2)$. It can be also seen that the Gd and Si co-doping concentrations can be elaborately tuned over a wide range by the ALD cycle ratio of $Gd_rSi_{12}O$ and HfO_2 . As a result, the Hf_rGd_{12} Si_zO films with Si/(Si + Gd + Hf) at.% between 11.5 and 28.9% and Gd/(Si + Gd + Hf) at.% between 6.8 and 28.4% can be achieved via changing the ALD cycle ratio of Gd_xSi_vO/HfO₂ from 1:9 to 1:1. To avoid the effects from the silicon substrate during XPS measurement, the composition of pristine Gd_xSi_yO film on Ge substrate was also detected by XPS, as shown in the inset of Fig. 1c. Si 2p signal can still be detected, and the Si/(Si + Gd) at.% is around 40%, which is in accord with the results of pristine Gd_{x} - $Si_{\nu}O$ on Si substrate. Therefore, it can be demonstrated that the Si XPS signals of Hf_xGd_ySi_zO samples attribute to Si dopant in $Hf_xGd_ySi_zO$ films rather than Si substrates.

The band offsets at Hf_xGd₁Si₂O/Si interface were also explored by XPS. The valence band offset (VBO, ΔE_{v}) can be determined by Kraut's method, assuming that the energy difference between the core level and the valence band (VB) edge of the substrate keeps unchanged after dielectrics films deposition [39, 40]. Thus, valence band maximum (VBM) of Si (E_{VBM} (Si)) was chosen as the reference to determine the VBO between the $Hf_xGd_ySi_zO$ films and the Si substrate. Figure 2a shows the XPS valence band spectra of the Si substrate and the Hf_xGd_ySi_zO films deposited on Si substrates. The VBM of the clean Si substrate $(E_{\text{VBM}}(\text{Si}))$ has been determined to be 0.52 eV by using the linear extrapolation as shown in Fig. 2a [41, 42]. The VBMs of the (1:N)-Hf_xGd_ySi_zO/Si samples $(E_{VBM}(Hf_xGd_ySi_zO))$ are determined to be 3.12 eV, 3.09 eV, 2.72 eV, 2.71 eV, 2.77 eV, 2.77 eV, 3.08 eV, and 3.19 eV, respectively. Therefore, the

Table 1 Process details and
composition of (1:N)-
Hf_xGd_ySi₂O, Gd_xSi_yO, and
HfO2 samples

Gd/Hf cycle ratio	1:0	1:1	1:2	1:3	1:4	1:5	1:6	1:7	1:9	0:1
Loops	153	63	40	29	23	19	16	14	11	107
Total cycles	153	126	120	116	115	114	112	112	110	107
Si at.%	38.4	28.9	21.3	19.6	17.9	16.3	14.9	13.7	11.5	0
Gd at.%	61.6	28.4	21.3	16.0	13.4	11.8	9.9	8.8	6.8	0
Gd/(Gd + Hf) at.%	100	39.9	27.1	19.9	16.3	14.1	11.6	10.2	7.7	0

Si at.% and Gd at.% refer to the atomic percent of Si/(Si + Gd + Hf) and Gd/(Si + Gd + Hf), respectively

Hf 4f

50

-5



VBOs of (1:N)-Hf_xGd_ySi_zO/Si samples (ΔE_v (Hf_xGd_y- Si_zO/Si) are calculated as 2.60 eV, 2.57 eV, 2.20 eV, 2.19 eV, 2.25 eV, 2.25 eV, 2.56 eV, and 2.67 eV using the following formula: $\Delta E_{\rm v}({\rm Hf}_{\rm r}{\rm Gd}_{\rm u}{\rm Si}_{\rm z}{\rm O}/{\rm Si}_{\rm Z}$

Si) = $E_{\text{VBM}}(\text{Hf}_x\text{Gd}_y\text{Si}_z\text{O})$ - $E_{\text{VBM}}(\text{Si})$. The conduction band offset (CBO, ΔE_c) can be achieved by $\Delta E_c(Hf_xGd_ySi_zO/Si) = E_g(Hf_xGd_ySi_zO)$ - $E_{g}(\text{Si})-\Delta E_{v}(\text{Hf}_{x}\text{Gd}_{v}\text{Si}_{z}\text{O}/\text{Si})$, subtracting the band gap of the Si substrate and the VBO from the band gap of $Hf_xGd_ySi_zO$. The band gaps of $Hf_xGd_ySi_zO$ films can

be determined by the O1s electron energy loss spectra by using the linear extrapolation method [43, 44], as shown in Fig. 2b. Thus, the band gaps of the (1:N)- $Hf_xGd_ySi_zO$ films ($E_g(Hf_xGd_ySi_zO)$) can be determined to be 6.13 eV, 6.09 eV, 5.96 eV, 5.92 eV, 5.86 eV, 5.82 eV, 5.73 eV, and 5.61 eV, respectively. Therefore, the CBOs of (1:N)-Hf_xGd_{ν}Si_zO/Si samples $(\Delta E_c(Hf_rGd_{\nu}Si_zO/Si))$ are estimated to be 2.41 eV, 2.40 eV, 2.64 eV, 2.61 eV, 2.49 eV, 2.45 eV, 2.05 eV, and 1.82 eV, respectively.

According to the band gaps, VBOs, and CBOs results, the schematic band alignment diagram of the (1:N)-Hf_xGd_ySi_zO/Si heterostructures is constructed, as shown in Fig. 3. The band structures indicate that the band gap increases with doping concentration, which may attribute to the existence of Gd-O-Hf bonding and Si–O bonding [23]. In addition, the VBO decreases first and then increases, and the CBO increases first and then decreases along with the Gd/ Hf atomic ratio rising. Therefore, the band offsets symmetry of Hf_xGd_ySi_zO/Si heterostructures can be improved by tuning the composition of $Hf_{r}Gd_{\mu}Si_{z}O$. For example, (1:6)-Hf_xGd_uSi_zO films exhibit both large VBO (2.25 eV) and CBO (2.45 eV) values, beneficial for inhibiting the leakage current by huge barrier heights.

The MOS capacitors with the same physical thickness of (1:N)-Hf_xGd_ySi_zO films on Si substrates were fabricated. The high-frequency (1 MHz) *C*–*V* characteristics of these capacitors were measured. The accumulation capacitances of these Pt/(1:N)- Hf_x-Gd_ySi_zO/Si capacitors are 0.93, 1.01, 1.08, 1.04, 0.98, 0.83, 0.82, and 0.76 μ F cm⁻², respectively. Considering the Gd/(Gd + Hf) atomic percent of the (1:N)-Hf_xGd_ySi_zO films in Table 1, the accumulation capacitance as a function of the Gd/(Gd + Hf) atomic percentage is plotted in Fig. 4. The results indicate that the accumulation capacitance increases first and then decreases along with doping concentration. Thus, the Hf_xGd_ySi_zO films with Gd/(Gd +



Fig. 3 The schematic band structure diagram of the (1:N)- $Hf_xGd_ySi_zO/Si$ heterostructures

Hf) of 11.6 at.% exhibit the maximum accumulation capacitance and dielectric constant. Our results are consistent with previous literature by Adelmann et al. [45], which reported that Gd-doped HfO₂ films with the doping concentration of 11.1 at.% exhibit the maximum dielectric constant.

Figure 5a, b compares the high-frequency (1 MHz) C-V and J-V characteristics of (1:6)-Hf_xGd_ySi_zO films and HfO₂ films, respectively. The accumulation capacitance densities of (1:6)-Hf_xGd_ySi_zO films and HfO₂ films are 1.08 μ F cm⁻² and 1.17 μ F cm⁻², which can be extracted from Fig. 5a. The calculated relative dielectric constant of the (1:6)-Hf_xGd_ySi_zO films is 13.3, which is only slightly smaller than that of the HfO₂ films (14.4). According to Fig. 5b, it can be observed that the leakage current density of (1:6)-Hf_xGd_uSi₂O films and HfO₂ films at gate voltage of -1 V are 1.22×10^{-4} A cm⁻² and 2.62×10^{-3} A cm⁻², respectively. Compared to HfO₂ films, the leakage current density of (1:6)-Hf_xGd_ySi_zO films is lower with more than an order of magnitude. Therefore, comprehensive electrical properties of HfO₂ films can be improved by Gd and Si co-doping, reducing the leakage current density effectively without deteriorating the dielectric constant obviously. Furthermore, the current density of Hf_xGd_{ν} -Si_zO/Si heterostructures as a function of Gd/ (Gd + Hf) atomic percentage was also provided, as shown in the inset of Fig. 5b. It can be seen that (1:3), (1:5), and (1:6)-Hf_xGd_ySi_zO films all can exhibit a decreased current density due to their both large VBO and CBO. However, the (1:9)-Hf_xGd_uSi_zO film



Fig. 4 Accumulation capacitance of the $Hf_xGd_ySi_zO$ films on Si substrates as a function of the Gd/(Gd + Hf) atomic percentage



Fig. 5 a C-V and b J-V characteristics of (1:6)-Hf_xGd_ySi_zO films and HfO₂ films on Si substrates. The inset of b is the leakage current density of the Hf_xGd_ySi_zO films on Si substrates as a function of the Gd/(Gd + Hf) atomic percentage

still shows a large current density, which may be ascribed to its small CBO.

4 Conclusion

Gd and Si co-doped HfO₂ films (Hf_xGd_uSi_zO) were successfully deposited by ALD in this work. XPS results demonstrate that Gd and Si can be co-doped into HfO₂ films using the only doping precursor $Gd[N(SiMe_3)_2]_3$. By tuning the ALD cycle ratio, the atomic percentages of Si/(Si + Gd + Hf), Gd/(Si + Gd + Hf) and Gd/(Gd + Hf) can be tuned between 11.5 and 28.9, 6.8 and 28.4, and 7.7 and 39.9%, respectively. The band alignment and electrical properties of ALD Hf_xGd_ySi_zO films on Si substrates with different doping concentrations were investigated. With increasing the Gd/Si doping concentration, the band gap increases, the VBO value decreases first and then increases, while the CBO value increases first and then decreases. Thus, (1:6)-Hf_rGd₁Si₂O films achieve the best band offset symmetry with both large VBO (2.25 eV) and CBO (2.45 eV). The (1:6)-Hf_xGd_ySi_zO films with 11.6 at.% Gd/(Gd + Hf) exhibit the maximum accumulation capacitance and dielectric constant which are only slightly smaller than those of the HfO₂ films. Compared to HfO₂ films, the leakage current density of (1:6)-Hf_xGd_ySi_zO films is decreased by at least one order of magnitude. Above results indicate that codoping of Si and Gd maybe a promising way to improve comprehensive electrical properties of HfO₂ films, effectively reducing the leakage current density while maintaining high dielectric constant.

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Compliance with ethical standards

Conflict of interest The authors declare that there is no conflict of interest.

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