Microwave dielectric properties of the low-temperature-fired Li₂ZnTi₃O₈–Li₂TiO₃ ceramics for LTCC applications

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Abstract

In this work, $0.73Li₂ZnTi₃O₈ - 0.27Li₂TiO₃$ ceramics were prepared through a traditional solid-state process. The effects of Li₂O–B₂O₃–SiO₂–CaO–Al₂O₃ (LBSCA) glass addition on phase formation, microstructure, sintering characteristic and microwave dielectric properties of the ceramics were investigated. A small amount of LBSCA glass addition significantly reduced sintering temperature of the ceramics. X-ray diffraction analysis revealed that $Li_2ZnTi_3O_8$ and $Li_2Ti_3O_8$ phases coexisted without producing any other crystal phases in the sintered ceramics. Dielectric constant and *Qf* values were related to the amount of LBSCA addition and sintering temperatures. The specimens obtained near-zero temperature coefficient (τ_f) values through the compensation on the positive τ_f of Li₂TiO₃ and the negative τ_f of Li₂ZnTi₃O₈. The 0.73Li₂ZnTi₃O₈– 0.27Li₂TiO₃ ceramic with 0.75 wt% LBSCA addition and sintered at 900 °C for 3 h exhibited excellent microwave dielectric properties of ε_r = 23.907, Qf = 63050 GHz and τ_f = 1.2 ppm/°C, which was very suitable for LTCC (low temperature co-fired ceramics) applications.

1 Introduction

With the rapid development of wireless communication technology, an increasing number of studies have focused on materials with brilliant microwave dielectric properties. At the same time, to achieve the miniaturisation and integration of microwave devices, many studies have investigated the LTCC technology, which plays an important role in microelectronic applications $[1-3]$ $[1-3]$. Ag is widely used as internal-electrode material in LTCC devices to ensure their low cost and high conductivity. Considering that Ag has a melting point of approximately 961 °C, the typical sintering temperature of LTCC materials must be reduced to approximately 900 °C or lower to better co-fire with Ag internal electrodes [\[4](#page-4-2)–[6\]](#page-4-3). Low-melting-point glasses were usually used as sintering aids in LTCC materials, which reduced the sintering temperature and achieved optimal microwave dielectric properties. Furthermore, near-zero τ_f of LTCC materials are important to obtain stable LTCC microwave components [[7–](#page-4-4)[12](#page-4-5)]. Near-zero τ_f LTCC materials can be obtained by properly compounding the materials with negative and positive τ_f .

 $Li₂ZnTi₃O₈$ ceramics had excellent microwave dielectric properties with ε_r values of ~25.8, a Qf value of ~74,200 GHz and a τ_f value of -13 ppm/ \degree C [\[13](#page-4-6)]. However, the τ_f value of $Li_2ZnTi_3O_8$ ceramics was negative. On the other hand, $Li₂TiO₃$ had attracted considerable attention in microwave dielectric ceramics because of its positive τ_f value and relatively outstanding microwave dielectric prop-erties [[14](#page-4-7)]. In the present work, we combined $Li_2ZnTi_3O_8$ with Li_2TiO_3 to obtain materials with near-zero τ_f values. And LBSCA glass was adopted as sintering aid to lower the sintering temperature of $Li₂ZnTi₃O₈–Li₂TiO₃ compound$ ceramics because it had very low softening temperature point and efficient help-melting effect [\[15](#page-4-8), [16\]](#page-4-9). The effects of LBSCA glass addition on the sintering behaviour, microstructure and microwave dielectric properties of the ceramics $\overline{\otimes}$ Xiaoli Tang
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2 Experimental procedure

 $Li₂ZnTi₃O₈–Li₂TiO₃$ ceramic samples were prepared by the conventional solid-state reaction. High-purity $Li₂CO₃$ (99%), ZnO (99.5%) and TiO₂ (99%) were used as starting materials. Li₂ZnTi₃O₈ calcined powders were produced by ballmilling Li_2CO_3 , ZnO and TiO₂ in a 1:1:3 molar ratio. Then the mixed powders were dried and calcined at 900 °C for 3 h. $Li₂TiO₃$ calcined powders were produced by ball-milling $Li₂CO₃$ and TiO₂ in a 1:1 molar ratio. Then the mixed powders were dried and calcined at 820 °C for 3 h. LBSCA glass was prepared by the quenching technology. The oxide raw materials were mixed and melted at 1000 °C for 2 h using an alumina crucible at a $Li₂O:B₂O₃:SiO₂:CaO:AI₂O₃ molar$ ratio of 52.45:31.06:11.99:2.25:2.25 [\[15](#page-4-8), [16\]](#page-4-9). The solution was removed from the furnace and then poured into cold water to obtain the glass. Then, approximately 0–1.5 wt% of LBSCA glass was added into $0.73Li₂ZnTi₃O₈ - 0.27Li₂TiO₃$ compounds respectively (0.73:0.27 was the weight ratio of $Li₂ZnTi₃O₈$ and $Li₂TiO₃$ calcined powders, which was calculated by their respective τ_f to obtain near-zero τ_f in the compound materials). The mixtures were milled in nylon pots with zirconia balls. After drying and granulation, the powders were pressed into cylinders, which were final sintered at 875–950 °C for 3 h.

Crystalline phase structures were analysed through X-ray diffraction (XRD:DX-2700) using Cu Kα radiation. Bulk densities of the sintered samples were measured using the Archimedes method, and the densities were obtained by the ratio of mass and volume. Sample micrographs were examined by scanning electron microscopy (SEM:JEOL JSM6490LV). The microwave dielectric properties of the sintered ceramics in microwave frequency were measured using the Hakki–Coleman method and Agilent N5230A network analyser (300 MHz–20 GHz) in a resonant cavity. τ_f values were measured at a temperature range of 20–80 °C. The values were calculated from the following formula:

$$
\tau_f = \frac{f_T - f_0}{f_0(T - T_0)} \times 10^6
$$

where f_T and f_0 are the resonant frequencies at 80 and 20 °C, respectively.

3 Results and discussion

Figure [1](#page-1-0) shows the XRD patterns of $0.73Li₂Zn Ti_3O_8 - 0.27Li_2Ti_3$ ceramics with different LBSCA glasses and sintered at 900 °C for 3 h. All samples contained $Li₂ZnTi₃O₈$ and $Li₂TiO₃$ phases, no other phases were detected. All peaks were indexed in terms of $Li₂ZnTi₃O₈$

Fig. 1 XRD patterns of $Li_2ZnTi_3O_8-Li_2TiO_3$ ceramics with x wt% LBSCA additive and sintered at 900 °C for 3 h. (*a*) $x=0$, (*b*)=0.25, $(c) = 0.5$, $(d) = 0.75$, $(e) = 1$, $(f) = 1.25$, and $(g) = 1.5$

(PDF #44-1037) and Li_2TiO_3 (\bullet , #33–0831). $Li_2ZnTi_3O_8$ is a cubic structure which belongs to the P4332 space group. Zn^{2+} and 1/2 of the Li⁺ are located in the centre of ZnO_4] and $[LiO₄]$ tetrahedral units $[17]$. Ti⁴⁺ and the remaining $1/2$ Li⁺ are located in the [TiO₆] and [LiO₆] octahedron centre. $Li₂TiO₃$ is a monoclinic structure which belongs to the C2/c(15) space group [[18](#page-4-11)]. Ti⁴⁺ is located in the centre of $[TiO₆]$ octahedron, and the octahedrons are connected to each other through six-coordinated Li⁺. Due to the large crystal structure difference between $Li₂ZnTi₃O₈$ and $Li₂TiO₃$, the $Li₂TiO₃$ phase beneficially coexisted with the Li₂ZnTi₃O₈ phase in the 0.73Li₂ZnTi₃O₈–0.27Li₂TiO₃ ceramics. LBSCA addition was undetectable, which indicated that LBSCA existed in the amorphous phase.

Figure [2](#page-2-0) shows the SEM micrographs of $0.73Li₂Zn Ti_3O_8 - 0.27Li_2TiO_3$ ceramics with different LBSCA glasses. LBSCA addition significantly influenced the densification and average grain size of the compound ceramics. Figure [2a](#page-2-0) presented a porous microstructure with many intergranular pores. Figure [2b](#page-2-0), c showed that the intergranular pores decreased, the grain size increased and the samples became dense with the increase of LBSCA glass content. This phenomenon was due to that LBSCA glass formed a liquid phase during sintering, which accelerated mass transfer and promoted sintering [\[19](#page-4-12)]. Figure [2d](#page-2-0) showed that the sample obtained a dense and uniform microstructure, which was doped with 0.75 wt% LBSCA glass. Figure [2](#page-2-0)e–g showed that the microstructure of the samples did not change significantly with further increasing LBSCA glass content.

Figure [3](#page-3-0) presents the variation in sintered densities of the samples with different LBSCA contents. The sintered densities initially increased and reached their maximum with $x=0.75$ wt% and then gradually decreased. The first

increase in densities was due to the liquid phase of LBSCA formed during sintering effectively promoted the densification of the materials [[20](#page-4-13)]. The densities monotonously decreased with further increasing LBSCA content, which could be attributed to two reasons. One reason was that

excessive grain boundary amorphous LBSCA glass hindered the further densification of the materials. The other reason was that the content ratio of low-density glass addition contributed to the decrease in densities of the samples.

Fig. 3 Sintered densities of $Li_2ZnTi_3O_8-Li_2TiO_3$ ceramics with different LBSCA glass and sintered from 875 to 950 °C

Fig. 4 Permittivity of $Li₂ZnTi₃O₈ - Li₂TiO₃$ ceramics with x wt% LBSCA glass and sintered from 875 to 950 °C

Figure [4](#page-3-1) shows the variation in ε_r values of the ceramics with different LBSCA glass contents and sintered under different temperatures. Obviously, the tendency of ε_r variation was consistent with that of density. As LBSCA content increased, ε_r gradually increased and reached the maximum with 0.75 wt% LBSCA. After that, ε_r slightly decreased with further increasing LBSCA content. Similar to the variation of density, this trend remained the same under different sintering temperatures. Therefore, sintered density was the main factor determining the ε_r value of the ceramics.

Figure [5](#page-3-2) shows the *Qf* values of $0.73Li₂Zn Ti₃O-0.27Li₂TiO₃$ ceramics with different LBSCA contents and sintered at 875–950 °C. The *Qf* values initially increased and then decreased with increasing LBSCA content. The maximal *Qf* values were obtained with 0.75 wt% LBSCA

Fig. 5 Qf values of $Li₂ZnTi₃O₈ - Li₂TiO₃$ ceramics with different LBSCA glass and sintered from 875 to 950 °C

Fig. 6 τ_f values of $Li_2ZnTi_3O_8-Li_2TiO_3$ ceramics with different LBSCA glass and sintered at 900 °C for 3 h

content, in spite of different sintering temperatures. After that, *Qf* values gradually decreased with further increasing LBSCA content. This phenomenon was very close to the variation of density. The microwave dielectric losses can be divided into internal and external losses [[21\]](#page-4-14). Internal losses are related to the internal crystal structure of the dielectric material and are mainly caused by the lattice vibration modes, whereas extrinsic losses are associated with many factors, such as second phases, oxygen vacancies, grain size and densification. Therefore, the initial increase in *Qf* values might be attributed to the increase in densification and average grain size. The subsequent decrease in *Qf* values was mainly due to the reduced density. Furthermore, the relatively high loss of glassy phase was also responsible for the decrease of *Qf* values.

Figure [6](#page-3-3) shows the τ_f values of 0.73Li₂Zn- $Ti_3O_8 - 0.27Li_2TiO_3$ ceramics sintered at 900 °C for 3 h. τ_f can be tuned by the mixtures of dielectrics with opposite τ_f values. A previous report showed that $Li_2ZnTi_3O_8$ has a negative τ_f value of -13.75 ppm/°C, whereas Li_2TiO_3 had a positive τ_f value of 35.78 ppm/ $\rm ^{\circ}C$. Therefore, the expected near-zero τ_f values could be achieved by the compensation of the positive τ_f of Li_2TiO_3 and the negative τ_f of $Li_2ZnTi_3O_8$ in 0.73Li₂ZnTi₃O₈–0.27Li₂TiO₃. Furthermore, τ_f altered from −6 to 7 ppm/°C with increasing LBSCA content in this study. This fact might due to the influence of the glass's τ_f and the variation of sintered densities. The specimen with 0.75 wt% LBSCA and sintered at 900 °C could obtain very high *Qf* value (which was 63,050 GHz) and near-zero τ_f of 1.2 ppm/°C, which was considered suitable for LTCC applications.

4 Conclusion

In this study, the effects of LBSCA addition on the phase formation, sintering characteristic, microstructure and microwave dielectric properties of the $0.73Li₂Zn Ti₃O₈ - 0.27Li₂TiO₃$ ceramics were investigated. Proper LBSCA glass addition could effectively densify the samples and improve the microwave dielectric properties. No chemical reaction occurred between the $Li₂ZnTi₃O₈$ and $Li₂TiO₃$ ceramics. A near-zero τ_f value was obtained through the compensation of the positive τ_f of $\text{Li}_2 \text{TiO}_3$ and the negative τ_f of Li₂ZnTi₃O₈. The 0.73Li₂ZnTi₃O₈–0.27Li₂TiO₃ sample with 0.75 wt% LBSCA glass addition and sintered at 900 °C presented excellent dielectric properties with $\varepsilon_r = 23.907$, Qf =63,050 GHz and τ_f =1.2 ppm/°C. The proposed ceramic was a perfect candidate material for LTCC applications.

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