# Microwave dielectric properties of the low-temperature-fired Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub>–Li<sub>2</sub>TiO<sub>3</sub> ceramics for LTCC applications

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Received: 17 May 2018 / Accepted: 5 July 2018 / Published online: 7 July 2018 © Springer Science+Business Media, LLC, part of Springer Nature 2018

#### Abstract

In this work, 0.73Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub>-0.27Li<sub>2</sub>TiO<sub>3</sub> ceramics were prepared through a traditional solid-state process. The effects of Li<sub>2</sub>O-B<sub>2</sub>O<sub>3</sub>-SiO<sub>2</sub>-CaO-Al<sub>2</sub>O<sub>3</sub> (LBSCA) glass addition on phase formation, microstructure, sintering characteristic and microwave dielectric properties of the ceramics were investigated. A small amount of LBSCA glass addition significantly reduced sintering temperature of the ceramics. X-ray diffraction analysis revealed that Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> and Li<sub>2</sub>TiO<sub>3</sub> phases coexisted without producing any other crystal phases in the sintered ceramics. Dielectric constant and *Qf* values were related to the amount of LBSCA addition and sintering temperatures. The specimens obtained near-zero temperature coefficient ( $\tau_{\rm f}$ ) values through the compensation on the positive  $\tau_{\rm f}$  of Li<sub>2</sub>TiO<sub>3</sub> and the negative  $\tau_{\rm f}$  of Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub>. The 0.73Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub>-0.27Li<sub>2</sub>TiO<sub>3</sub> ceramic with 0.75 wt% LBSCA addition and sintered at 900 °C for 3 h exhibited excellent microwave dielectric properties of  $\varepsilon_r$ =23.907, *Qf*=63050 GHz and  $\tau_f$ =1.2 ppm/°C, which was very suitable for LTCC (low temperature co-fired ceramics) applications.

## 1 Introduction

With the rapid development of wireless communication technology, an increasing number of studies have focused on materials with brilliant microwave dielectric properties. At the same time, to achieve the miniaturisation and integration of microwave devices, many studies have investigated the LTCC technology, which plays an important role in microelectronic applications [1-3]. Ag is widely used as internal-electrode material in LTCC devices to ensure their low cost and high conductivity. Considering that Ag has a melting point of approximately 961 °C, the typical sintering temperature of LTCC materials must be reduced to approximately 900 °C or lower to better co-fire with Ag internal electrodes [4–6]. Low-melting-point glasses were usually used as sintering aids in LTCC materials, which reduced the sintering temperature and achieved optimal microwave dielectric properties. Furthermore, near-zero  $\tau_{f}$  of LTCC

materials are important to obtain stable LTCC microwave components [7–12]. Near-zero  $\tau_f$  LTCC materials can be obtained by properly compounding the materials with negative and positive  $\tau_f$ .

Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> ceramics had excellent microwave dielectric properties with  $\varepsilon_r$  values of ~25.8, a Qf value of ~74,200 GHz and a  $\tau_f$  value of – 13 ppm/°C [13]. However, the  $\tau_{\rm f}$  value of Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> ceramics was negative. On the other hand, Li<sub>2</sub>TiO<sub>3</sub> had attracted considerable attention in microwave dielectric ceramics because of its positive  $\tau_{\rm f}$ value and relatively outstanding microwave dielectric properties [14]. In the present work, we combined  $Li_2ZnTi_3O_8$ with  $Li_2TiO_3$  to obtain materials with near-zero  $\tau_f$  values. And LBSCA glass was adopted as sintering aid to lower the sintering temperature of Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub>-Li<sub>2</sub>TiO<sub>3</sub> compound ceramics because it had very low softening temperature point and efficient help-melting effect [15, 16]. The effects of LBSCA glass addition on the sintering behaviour, microstructure and microwave dielectric properties of the ceramics were investigated and discussed.



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#### 2 Experimental procedure

 $Li_2ZnTi_3O_8$ - $Li_2TiO_3$  ceramic samples were prepared by the conventional solid-state reaction. High-purity Li<sub>2</sub>CO<sub>3</sub> (99%), ZnO (99.5%) and TiO<sub>2</sub> (99%) were used as starting materials. Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> calcined powders were produced by ballmilling Li<sub>2</sub>CO<sub>3</sub>, ZnO and TiO<sub>2</sub> in a 1:1:3 molar ratio. Then the mixed powders were dried and calcined at 900 °C for 3 h. Li<sub>2</sub>TiO<sub>3</sub> calcined powders were produced by ball-milling  $Li_2CO_3$  and  $TiO_2$  in a 1:1 molar ratio. Then the mixed powders were dried and calcined at 820 °C for 3 h. LBSCA glass was prepared by the quenching technology. The oxide raw materials were mixed and melted at 1000 °C for 2 h using an alumina crucible at a Li<sub>2</sub>O:B<sub>2</sub>O<sub>3</sub>:SiO<sub>2</sub>:CaO:Al<sub>2</sub>O<sub>3</sub> molar ratio of 52.45:31.06:11.99:2.25:2.25 [15, 16]. The solution was removed from the furnace and then poured into cold water to obtain the glass. Then, approximately 0-1.5 wt% of LBSCA glass was added into 0.73Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub>-0.27Li<sub>2</sub>TiO<sub>3</sub> compounds respectively (0.73:0.27 was the weight ratio of Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> and Li<sub>2</sub>TiO<sub>3</sub> calcined powders, which was calculated by their respective  $\tau_f$  to obtain near-zero  $\tau_f$  in the compound materials). The mixtures were milled in nylon pots with zirconia balls. After drying and granulation, the powders were pressed into cylinders, which were final sintered at 875-950 °C for 3 h.

Crystalline phase structures were analysed through X-ray diffraction (XRD:DX-2700) using Cu K $\alpha$  radiation. Bulk densities of the sintered samples were measured using the Archimedes method, and the densities were obtained by the ratio of mass and volume. Sample micrographs were examined by scanning electron microscopy (SEM:JEOL JSM6490LV). The microwave dielectric properties of the sintered ceramics in microwave frequency were measured using the Hakki–Coleman method and Agilent N5230A network analyser (300 MHz–20 GHz) in a resonant cavity.  $\tau_{\rm f}$  values were measured at a temperature range of 20–80 °C. The values were calculated from the following formula:

$$\tau_f = \frac{f_T - f_0}{f_0 (T - T_0)} \times 10^6$$

where  $f_T$  and  $f_0$  are the resonant frequencies at 80 and 20 °C, respectively.

### 3 Results and discussion

Figure 1 shows the XRD patterns of  $0.73 \text{Li}_2\text{Zn-Ti}_3\text{O}_8$ -0.27Li<sub>2</sub>TiO<sub>3</sub> ceramics with different LBSCA glasses and sintered at 900 °C for 3 h. All samples contained Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> and Li<sub>2</sub>TiO<sub>3</sub> phases, no other phases were detected. All peaks were indexed in terms of Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub>



**Fig. 1** XRD patterns of  $\text{Li}_2\text{Zn}\text{Ti}_3\text{O}_8$ - $\text{Li}_2\text{Ti}\text{O}_3$  ceramics with x wt% LBSCA additive and sintered at 900 °C for 3 h. (*a*) x=0, (*b*)=0.25, (*c*)=0.5, (*d*)=0.75, (*e*)=1, (*f*)=1.25, and (*g*)=1.5

(PDF #44-1037) and Li<sub>2</sub>TiO<sub>3</sub> (•, #33–0831). Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> is a cubic structure which belongs to the P4332 space group. Zn<sup>2+</sup> and 1/2 of the Li<sup>+</sup> are located in the centre of [ZnO<sub>4</sub>] and [LiO<sub>4</sub>] tetrahedral units [17]. Ti<sup>4+</sup> and the remaining 1/2 Li<sup>+</sup> are located in the [TiO<sub>6</sub>] and [LiO<sub>6</sub>] octahedron centre. Li<sub>2</sub>TiO<sub>3</sub> is a monoclinic structure which belongs to the C2/c(15) space group [18]. Ti<sup>4+</sup> is located in the centre of [TiO<sub>6</sub>] octahedron, and the octahedrons are connected to each other through six-coordinated Li<sup>+</sup>. Due to the large crystal structure difference between Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> and Li<sub>2</sub>TiO<sub>3</sub>, the Li<sub>2</sub>TiO<sub>3</sub> phase beneficially coexisted with the Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> phase in the 0.73Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub>–0.27Li<sub>2</sub>TiO<sub>3</sub> ceramics. LBSCA addition was undetectable, which indicated that LBSCA existed in the amorphous phase.

Figure 2 shows the SEM micrographs of  $0.73Li_2Zn-Ti_3O_8-0.27Li_2TiO_3$  ceramics with different LBSCA glasses. LBSCA addition significantly influenced the densification and average grain size of the compound ceramics. Figure 2a presented a porous microstructure with many intergranular pores. Figure 2b, c showed that the intergranular pores decreased, the grain size increased and the samples became dense with the increase of LBSCA glass content. This phenomenon was due to that LBSCA glass formed a liquid phase during sintering, which accelerated mass transfer and promoted sintering [19]. Figure 2d showed that the sample obtained a dense and uniform microstructure, which was doped with 0.75 wt% LBSCA glass. Figure 2e–g showed that the microstructure of the samples did not change significantly with further increasing LBSCA glass content.

Figure 3 presents the variation in sintered densities of the samples with different LBSCA contents. The sintered densities initially increased and reached their maximum with x = 0.75 wt% and then gradually decreased. The first



increase in densities was due to the liquid phase of LBSCA formed during sintering effectively promoted the densification of the materials [20]. The densities monotonously decreased with further increasing LBSCA content, which could be attributed to two reasons. One reason was that excessive grain boundary amorphous LBSCA glass hindered the further densification of the materials. The other reason was that the content ratio of low-density glass addition contributed to the decrease in densities of the samples.



Fig.3 Sintered densities of  $Li_2ZnTi_3O_8$ - $Li_2TiO_3$  ceramics with different LBSCA glass and sintered from 875 to 950 °C



Fig.4 Permittivity of  $Li_2ZnTi_3O_8$ - $Li_2TiO_3$  ceramics with x wt% LBSCA glass and sintered from 875 to 950 °C

Figure 4 shows the variation in  $\varepsilon_r$  values of the ceramics with different LBSCA glass contents and sintered under different temperatures. Obviously, the tendency of  $\varepsilon_r$  variation was consistent with that of density. As LBSCA content increased,  $\varepsilon_r$  gradually increased and reached the maximum with 0.75 wt% LBSCA. After that,  $\varepsilon_r$  slightly decreased with further increasing LBSCA content. Similar to the variation of density, this trend remained the same under different sintering temperatures. Therefore, sintered density was the main factor determining the  $\varepsilon_r$  value of the ceramics.

Figure 5 shows the Qf values of  $0.73 \text{Li}_2\text{Zn}$ -Ti<sub>3</sub>O-0.27Li<sub>2</sub>TiO<sub>3</sub> ceramics with different LBSCA contents and sintered at 875–950 °C. The Qf values initially increased and then decreased with increasing LBSCA content. The maximal Qf values were obtained with 0.75 wt% LBSCA



Fig.5 Qf values of Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub>-Li<sub>2</sub>TiO<sub>3</sub> ceramics with different LBSCA glass and sintered from 875 to 950 °C



Fig. 6  $\tau_f$  values of Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub>-Li<sub>2</sub>TiO<sub>3</sub> ceramics with different LBSCA glass and sintered at 900 °C for 3 h

content, in spite of different sintering temperatures. After that, Qf values gradually decreased with further increasing LBSCA content. This phenomenon was very close to the variation of density. The microwave dielectric losses can be divided into internal and external losses [21]. Internal losses are related to the internal crystal structure of the dielectric material and are mainly caused by the lattice vibration modes, whereas extrinsic losses are associated with many factors, such as second phases, oxygen vacancies, grain size and densification. Therefore, the initial increase in Qf values might be attributed to the increase in densification and average grain size. The subsequent decrease in Qf values was mainly due to the reduced density. Furthermore, the relatively high loss of glassy phase was also responsible for the decrease of Qf values.

Figure 6 shows the  $\tau_f$  values of  $0.73 Li_2 Zn$ -Ti<sub>3</sub>O<sub>8</sub>–0.27Li<sub>2</sub>TiO<sub>3</sub> ceramics sintered at 900 °C for 3 h.  $\tau_{\rm f}$ can be tuned by the mixtures of dielectrics with opposite  $\tau_{\rm f}$  values. A previous report showed that Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> has a negative  $\tau_f$  value of -13.75 ppm/°C, whereas Li<sub>2</sub>TiO<sub>3</sub> had a positive  $\tau_f$  value of 35.78 ppm/°C. Therefore, the expected near-zero  $\tau_f$  values could be achieved by the compensation of the positive  $\tau_f$  of Li<sub>2</sub>TiO<sub>3</sub> and the negative  $\tau_f$  of Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> in  $0.73 \text{Li}_2 \text{ZnTi}_3 \text{O}_8 - 0.27 \text{Li}_2 \text{TiO}_3$ . Furthermore,  $\tau_f$  altered from -6 to 7 ppm/°C with increasing LBSCA content in this study. This fact might due to the influence of the glass's  $\tau_{\rm f}$  and the variation of sintered densities. The specimen with 0.75 wt% LBSCA and sintered at 900 °C could obtain very high *Qf* value (which was 63,050 GHz) and near-zero  $\tau_{\rm f}$ of 1.2 ppm/°C, which was considered suitable for LTCC applications.

# 4 Conclusion

In this study, the effects of LBSCA addition on the phase formation, sintering characteristic, microstructure and microwave dielectric properties of the  $0.73Li_2Zn-Ti_3O_8-0.27Li_2TiO_3$  ceramics were investigated. Proper LBSCA glass addition could effectively densify the samples and improve the microwave dielectric properties. No chemical reaction occurred between the Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> and Li<sub>2</sub>TiO<sub>3</sub> ceramics. A near-zero  $\tau_f$  value was obtained through the compensation of the positive  $\tau_f$  of Li<sub>2</sub>ZnTi<sub>3</sub>O<sub>8</sub> and Li<sub>2</sub>TiO<sub>3</sub> sample with 0.75 wt% LBSCA glass addition and sintered at 900 °C presented excellent dielectric properties with  $\varepsilon_r$  = 23.907, Qf=63,050 GHz and  $\tau_f$ =1.2 pm/°C. The proposed ceramic was a perfect candidate material for LTCC applications.

Acknowledgements This work was supported by the National Natural Science Foundation of China under Grant Nos. 61471096 and 61771104 and Sichuan science and technology program. Special Projects on Science and Technology of Guizhou Province [2016]3011. And Dongguan entrepreneurial talent program.

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