

# Effect of TiO<sub>2</sub> nanoparticles on the microstructure and bonding strengths of Sn0.7Cu composite solder BGA packages with immersion Sn surface finish

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**Abstract** This study investigated the effects of TiO<sub>2</sub> nanoparticles on the interfacial microstructures and bonding strength of Sn0.7Cu (SC) composite solder joints in ball grid array packages with immersion Sn surface finishes. The addition of TiO<sub>2</sub> nanoparticles to SC solders resulted in significant changes in the interfacial microstructure. The nanoparticle addition suppressed the growth of the Cu<sub>6</sub>Sn<sub>5</sub> IMC layer, significantly improving shear strength and reliability of solder joints after multiple reflows. The fracture surfaces of the SC solder exhibited a semi-brittle fracture mode with a relatively smooth surface, while those of the SC composite solder showed typical ductile failures with very refined dimpled surfaces.

## 1 Introduction

Ball grid array (BGA) solder interconnections are currently important in the microelectronics industry. In some BGA packages, solder balls are used to form solder bumps on top of the under bump metallization (UBM) of the BGA. The solder bumps are then soldered onto the metal pads of a

substrate such as a printed circuit board (PCB) for electrical connection. Among the many Pb-free solder alloys, the eutectic Sn–Cu solder (Sn0.7Cu wt%) is considered the most promising candidate alloy to replace the eutectic Sn–Pb solder for wave soldering [1, 2]. This solder is less expensive than Sn–Ag and Sn–Ag–Cu solders, but the melting temperature is slightly higher [3].

Recently, lead-free solders doped with nano-sized, nonreacting, noncoarsening oxide dispersoids have been identified as potential materials to provide higher microstructure stability and better mechanical properties as compared to conventional solders [4–20]. Zhong and Gupta [4] demonstrated that the mechanical properties (microhardness, 0.2% YS, and UTS) increase with the increasing presence of reinforcement, with the best tensile strength, realized in the composite containing 1.5% alumina, far exceeding the strength of the eutectic Sn–Pb solder. Tsao et al. [5–7] reported that the addition of TiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> nanopowders into a Sn–Ag–Cu solder reduced the average size and spacing of Ag<sub>3</sub>Sn particles significantly. Shen et al. [8] investigated the Sn–3.5Ag–ZrO<sub>2</sub> composite solders and found that the size of dendrite β-Sn grains and intermetallic compounds (IMC) was refined.

Actually, the metallurgical reaction between liquid solder and copper, or other metallization, results in an IMC layer at the joint interface. Although the formation of a thin IMC layer is desirable to achieve good metallurgical bonding, it is known that thicker IMC growth affects the mechanical reliability in ball grid array (BGA) and flip-chip solder joints. This problem of IMC overgrowth has a deleterious effect in the case of Pb-free solders [8–11]. However, when a small amount of La (only 0.05 wt%) is added into Sn40Pb solder alloy, the growth of Cu–Sn IMC at the solder joint interface is depressed, and the thermal fatigue life of solder joints is improved [10]. Chuang et al.

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reported that the interfacial IMC layers in Sn58Bi0.5La solder joints were much thinner than those in undoped Sn58Bi [12]. In addition, our findings correspond to similar results presented by Nai et al. [13], who studied the suppression of IMC growth by the addition of carbon nanotubes (CNTs).

In this study, a small amount of TiO<sub>2</sub> nanoparticles was added to the Sn0.7Cu (SC) based solder to produce a new solder alloy. The interfacial intermetallic reactions, bonding strengths, and fracture mode of SC composite solder ball grid array (BGA) solder joints after different reflowing processes were investigated.

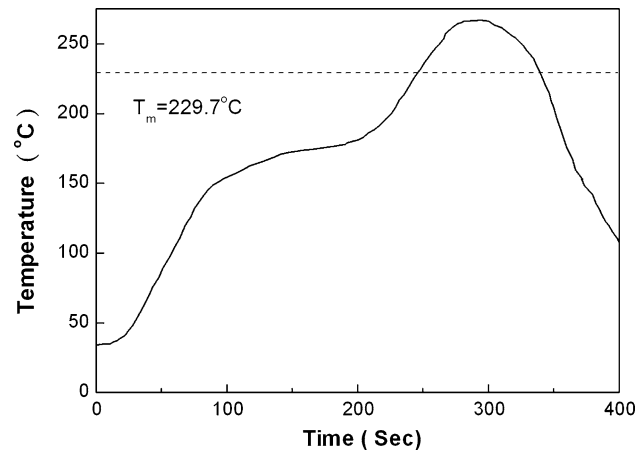
## 2 Experimental

The BGA package used in this study had 49 Cu pads on a FR-4 substrate, and the surfaces of the Cu pads were immersion-plated with 0.2 μm thick Sn film. The solder mask-opening diameter was 550 μm. The SC composite solders were prepared via mechanically dispersing 0.5 and 1.0 wt.% of TiO<sub>2</sub> nanoparticles of about 20 nm in diameter (Nanostructured & Amorphous materials, USA) into SC solder paste. The intermixtures were blended manually for at least 45 min to uniformly distribute the TiO<sub>2</sub> nanoparticles with a water-soluble flux (Qualitek Singapore (PTE) Ltd.). Then the paste intermixture was printed onto alumina substrates using a 150 μm-thick stainless steel solder paste stencil and reflowed in a reflow oven (Reddish Electronic, SM2000CXE) at 250 °C to prepare solder balls of about 0.76 mm in diameter. SC composite solder (wt%) solder balls were dipped in rosin mildly activated (RMA)-type flux, placed on these Sn/Cu pads, and then reflowed in a hot-air furnace (Reddish Electronic, SM2000CXE).

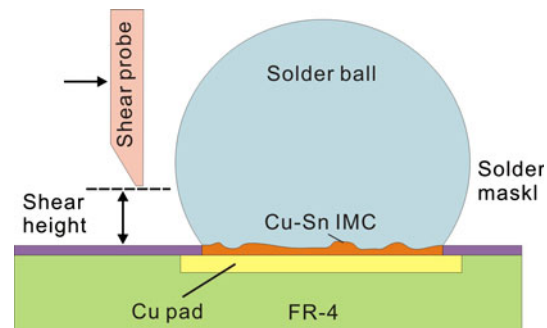
The reflow temperature profile, with a peak temperature at 267 °C for a melting time of 90 s ( $T_m > 229.7$  °C), is shown in Fig. 1. All samples were divided into two groups: one cycle and eight cycles of reflow.

All samples were cross-sectioned through a row of solder balls, ground with 2,000-grit SiC paper, polished with 1 and 0.3 μm Al<sub>2</sub>O<sub>3</sub> powders, and etched with 5% HCl and 95% H<sub>2</sub>O solution. The morphology of the intermetallic compounds was observed by scanning electron microscopy (SEM). The mean thickness of the interfacial IMC was evaluated using image analysis software.

The bonding strengths of the solder joints in the BGA packages after reflowing were tested using Dage 4000 pull test machine, for which the shear rate and shear height were set at 0.1 mm/s and 125 μm (about 1/4 of the reflowed ball height), respectively. Figure 2 shows a schematic illustration of the ball shear test performed in this study. The fractography of the solder joints after ball shear testing was observed by SEM.



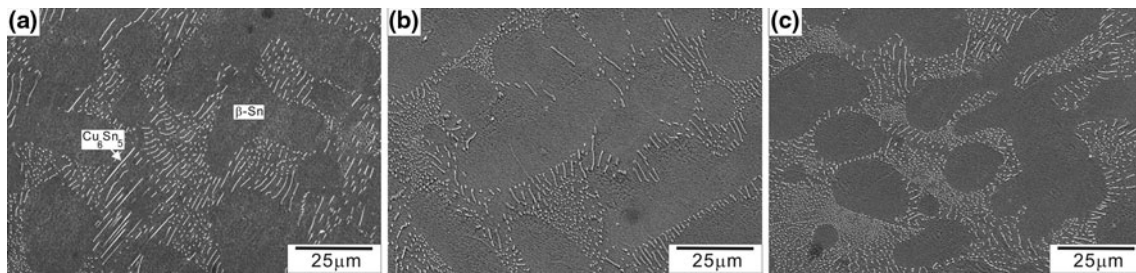
**Fig. 1** Temperature profile for reflowing the SC composite BGA packages in this study



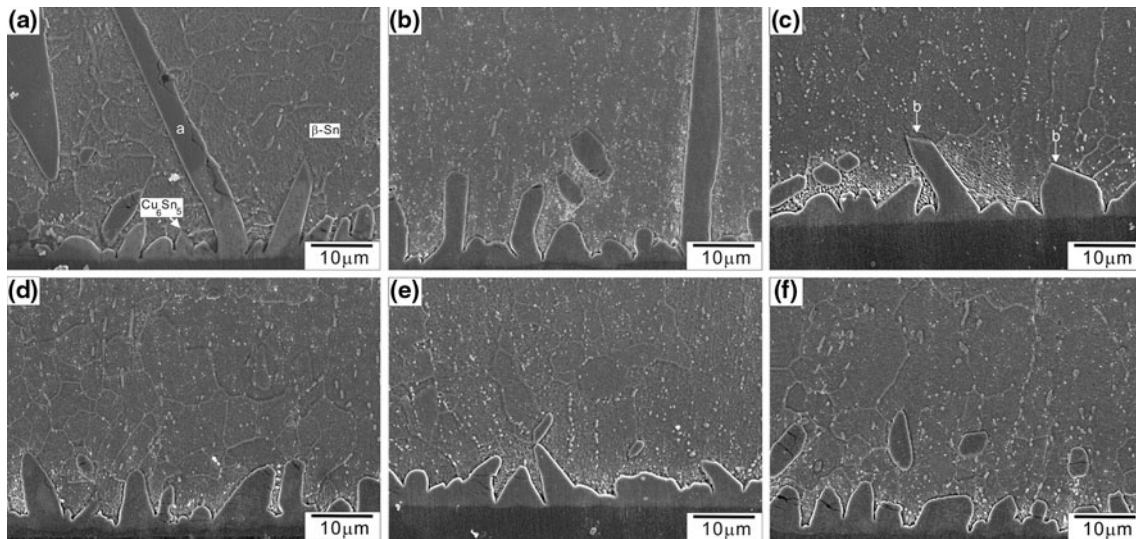
**Fig. 2** Schematic illustrations of the ball shear test

## 3 Results and discussion

The microstructures of the SC composite solder balls before reflow are shown in Fig. 3. The microstructure of SC solder balls consists of β-Sn phases, needle-like Cu<sub>6</sub>Sn<sub>5</sub> IMCs, and eutectic areas. As shown in Fig. 3b and c, the SC composite solder balls had a finer microstructure than that of SC solder balls. Notably, the sizes of the Cu<sub>6</sub>Sn<sub>5</sub> IMCs decreased with increases in the amount of TiO<sub>2</sub> nanoparticles. As is well known, a notable effect of nano-sized, nonreactive, noncourse nanoparticles on solder alloys is to refine the microstructure [14, 15]. The theory of heterogeneous nucleation can also be used to explain the refinement of Cu<sub>6</sub>Sn<sub>5</sub> particles in the prepared SC composite solder. Hao et al. [16] reported that ErSn<sub>3</sub> IMCs formed from the molten solder provided heterogeneous nucleation sites for the IMCs in the soldering. Subsequently, the Cu–Sn IMCs produced during soldering have a uniform size and are evenly distributed in the solder matrix. A similar result was also reported by Wang et al. [17]. This phenomenon indicates that the TiO<sub>2</sub> nanoparticles could improve the microstructure and the reliability of the SC composite solder joints in service [18].



**Fig. 3** SEM micrographs of SC composite solders: **a** SC, **b** SC-0.5TiO<sub>2</sub>, and **c** SC-1TiO<sub>2</sub>



**Fig. 4** Interfacial microstructures of the SC composite solder joints after one (**a**, **c**, **e**) and eight (**b**, **d**, **f**) cycles of reflowing: **a**, **b** SC, **c**, **d** SC-0.5TiO<sub>2</sub>, **e**, **f** SC-1TiO<sub>2</sub>

Cross-sectional microstructures of the three types of solder joints are shown in Fig. 4, in which one cycle and eight cycles of reflow are compared. After reflowing, the immersion Sn thin film on the Sn/Cu pads has dissolved into the solder matrix, and a Cu<sub>6</sub>Sn<sub>5</sub> IMC layer has formed at the SC composite solder/Cu interface. It can be seen in Fig. 4a that the Cu<sub>6</sub>Sn<sub>5</sub> IMC layer has grown into an elongated shape, and wicker-Cu<sub>6</sub>Sn<sub>5</sub> IMCs (mark a) have grown into the solder matrix. The wicker-Cu<sub>6</sub>Sn<sub>5</sub> IMCs in these images were several hundreds of micrometers in length. Kim et al. [8] reported that large Ag<sub>3</sub>Sn and long Cu<sub>6</sub>Sn<sub>5</sub>-whiskers are formed in all SAC/Cu joints and in high Cu-content solder joints. These large or long IMCs induce brittle fracturing at an interface and provide crack initiation sites. However, how the complex growth mechanisms govern the formation and morphological evolution of the interfacial layers is still unclear [19–21]. These authors had reported that wicker-Cu<sub>6</sub>Sn<sub>5</sub> IMCs were also observed in solidified tin and noted to emanate from the tops of the Cu<sub>6</sub>Sn<sub>5</sub> scallops. It is suggested that the formation of the wicker-Cu<sub>6</sub>Sn<sub>5</sub> IMCs emerging from the

scallops is related to a local concentration increase in copper, and that they grow into the solder matrix by the reaction of tin (Sn) with this copper (Cu). Nucleation is not required because they grow from the tops of the scallops, and in the same crystal orientation as the scallops. This could be attributed to the following [20–22]: (1) The soldering process provides access for diffusion of Cu and Sn atoms and thus causes solidification from preferred growth, with the crystallographic c-axis pointing along the concentration gradient; and (2) the formation of wicker-Cu<sub>6</sub>Sn<sub>5</sub> IMCs could be due to a reduction of the copper concentration in the liquid. However, wicker-Cu<sub>6</sub>Sn<sub>5</sub> IMCs would be adverse to the reliability of the solder joints.

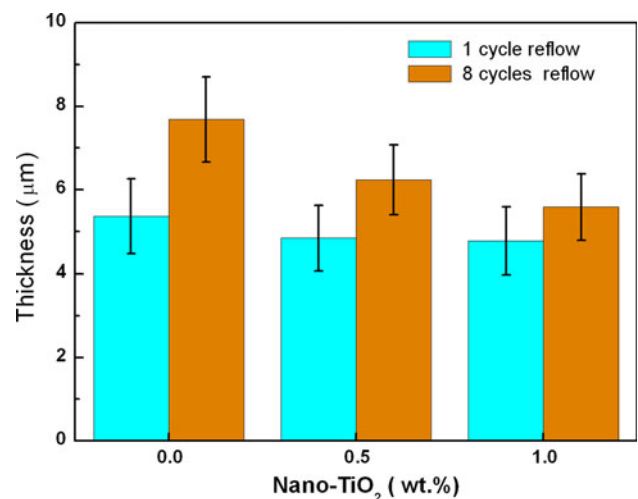
The addition of a small percentage of TiO<sub>2</sub> nanoparticles alters the lead-free SC composite solder/pad interface morphology after one time of reflow, as shown in the SEM micrographs in Fig. 4c and e. It is clear that the long wicker-Cu<sub>6</sub>Sn<sub>5</sub> IMCs disappeared at the interface with an increase in the content of TiO<sub>2</sub> nanoparticles to 0.5–1 wt%. After eight cycles of reflow, the morphology of the Cu<sub>6</sub>Sn<sub>5</sub> IMC layer changed from regular prism-type (mark b) to

scallop-type (Fig. 4d, f). Suh et al. [23] studied the size distribution and morphology of  $\text{Cu}_6\text{Sn}_5$  IMC layers in wetting reactions between molten solder and copper. The morphology of the  $\text{Cu}_6\text{Sn}_5$  IMCs was dependent on the composition of the solder. When the Pb content was further decreased <30 wt%, faceted scallops were observed along with some round scallops. Finally, when the solder became pure tin, only faceted scallops were observed. Wang et al. [24] reported that the growth of scallop-type grains is mainly caused by the ripening flux, while the growth of prism-type grains is caused by the interfacial reaction flux. Bian et al. [25] have found that there exists not only short-range order (SRO) but also medium-range order (MRO) in liquid metals. Also, Zhao et al. [26] found no MRO and a liquid structure in molten Sn0.7Cu solder similar to that of pure Sn, while for Sn<sub>2</sub>Cu solder, the MRO  $\text{Cu}_6\text{Sn}_5$  clusters at 260 °C have a great contribution to the liquid structure. The MRO in liquid solder quickens the formation of interfacial IMCs in the very early stage of soldering and also accelerates the ripening process of IMC grains at the interface. In our experiment, the Cu concentration of SC composite solder is similar to pure Sn under one cycle of reflow, so there may also have been SRO  $\text{Cu}_6\text{Sn}_5$  in the SC composite solder. However, the Cu concentration in the SC composite solder ball matrix increased under eight cycles of reflow. There may have been both SRO  $\text{Cu}_6\text{Sn}_5$  clusters and MRO  $\text{Cu}_6\text{Sn}_5$  clusters in the liquid solder. Thus, since the Cu concentration of the SC composite solder ball matrix was enhanced, there may have been, in addition to SRO  $\text{Cu}_6\text{Sn}_5$  clusters, MRO  $\text{Cu}_6\text{Sn}_5$  clusters in the liquid solder. This indicates that the MRO  $\text{Cu}_6\text{Sn}_5$  in liquid solder quickens formation of the scallop  $\text{Cu}_6\text{Sn}_5$  grains in the ripening process at the interface. As a result, the morphology of the IMC layer formed in the SC composite solder/Cu joints changed from prism-type to scallop-type after multiple cycles of reflowing.

The thicknesses of the  $\text{Cu}_6\text{Sn}_5$  IMC layers at the solder/pad interfaces of the SC composite solder packages versus added wt% of  $\text{TiO}_2$  nanoparticles are also summarized in Table 1 and shown in Fig. 5. The average thickness of the IMC layer was calculated using the following equation:  $T = S/L$ , where L is the length measured, and S is the

**Table 1** The overall thickness of Cu–Sn IMC layer of the Sn0.7Cu composite solder BGA packages with Ag/Cu after different numbers of reflow cycles

Sample #	Addition (wt%)	Cu–Sn IMC layer ( $\mu\text{m}$ )	
		Reflow 1	Reflow 8
SC	Nil	$5.37 \pm 0.89$	$7.69 \pm 1.02$
1	0.5	$4.85 \pm 0.78$	$6.24 \pm 0.84$
2	1	$4.78 \pm 0.81$	$5.59 \pm 0.79$



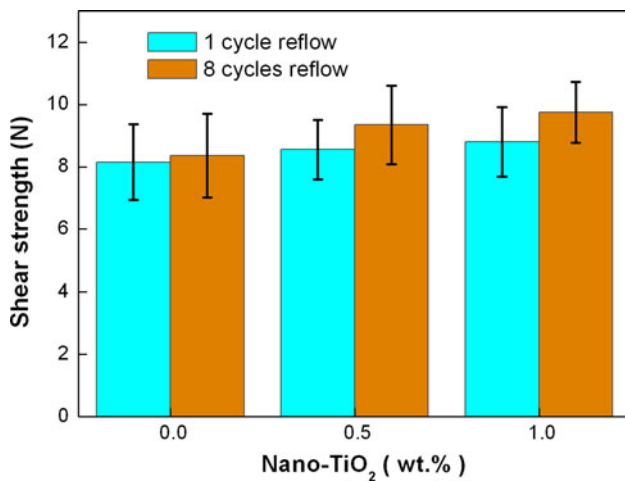
**Fig. 5** IMC layer thickness of SC composite solder joints

integral contour area of the IMC layer at the interface. With multiple reflows, the thickness of the IMC layer increased with increasing number of reflow cycles. In addition, it was observed that the IMC layer thickness of SC composite solder joints decreased with an increase in the addition of  $\text{TiO}_2$  nanoparticles into a SC solder mixture. Nai et al. [27–32] studied the suppression of CNTs on the growth of IMC layers between Sn3.5Ag0.7Cu composite solders and Cu pads after aging. Liu et al. [29] reported that the dispersion of  $\text{Y}_2\text{O}_3$  degrades the grain boundary migration activation energy, and the size of the  $\text{Cu}_6\text{Sn}_5$  particles and the thickness of  $\text{Cu}_6\text{Sn}_5$  IMC layer decrease. Shen and Chen [30] added nano-sized  $\text{ZrO}_2$  particles into a Sn9Zn solder to suppress the growth of Ni–Zn IMC in Sn9Zn/Ni solder joints. This suggests that the  $\text{TiO}_2$  nanoparticles suppressed the growth of the  $\text{Cu}_6\text{Sn}_5$  IMC layer.

The average shearing forces of the SC composite solder joints are summarized in Table 2 and plotted in Fig. 6. The average shear strength of the SC composite solder joints after eight cycles of reflow was higher than that of the SC composite solder joints after 1 cycle of reflow by about 2.6% in SC solder, 9.4% in SC-0.5 $\text{TiO}_2$  solder, and 10.8% in SC-1 $\text{TiO}_2$  solder. In addition, the changes in shearing forces with larger additions of  $\text{TiO}_2$  nanoparticles into the solder matrix are clear. When the  $\text{TiO}_2$  nanoparticle content is at 1 wt%, after one cycle and eight cycles of reflow,

**Table 2** Ball Shear strength of the Sn3.5Ag0.5Cu composite solders

Sample #	Addition (Wt%)	Ball Shear strength (N)	
		Reflow 1	Reflow 8
SC	Nil	$8.15 \pm 1.21$	$8.76 \pm 1.34$
1	0.5	$8.55 \pm 0.96$	$9.35 \pm 1.26$
2	1	$8.8 \pm 1.12$	$9.75 \pm 0.98$



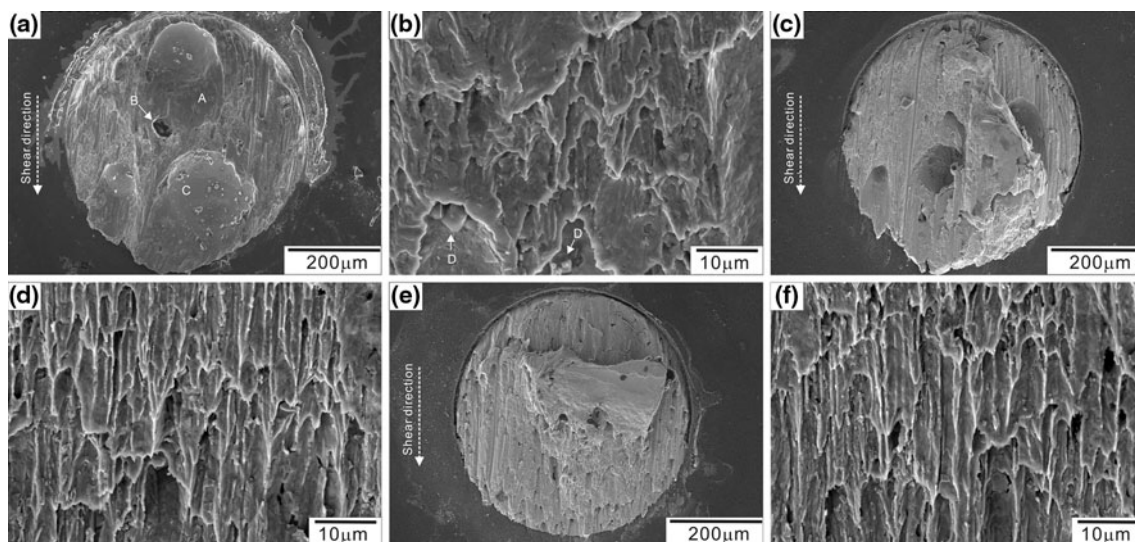
**Fig. 6** Variation of average shear strengths of SC composite solder joints

shear strengths show increases of 8.0 and 11.3%, respectively. It is suggested that repeated reflow cycles enhance the larger shearing area due to enlargement of the contact area and reduction in the height of the solder joints. Thus, the shear strength is raised. Moreover, multiple cycles of reflowing may enhance Cu interdiffusion, allowing it to react rapidly with Sn and precipitate  $\text{Cu}_6\text{Sn}_5$  IMC in the solder ball matrix. Another important point is that addition of  $\text{TiO}_2$  nanoparticles could promote a high nucleation density of stable second-phase dispersoid particles ( $\text{TiO}_2$  and  $\text{Cu}_6\text{Sn}_5$ ) in solder ball joints. As a result, both  $\text{TiO}_2$  nanoparticles and  $\text{Cu}_6\text{Sn}_5$  IMC act as potential sites that (1) pin grain boundaries, thereby impeding sliding of the grain boundaries, and (2) increase the number of dislocation densities and obstacles, thereby restricting the motion of

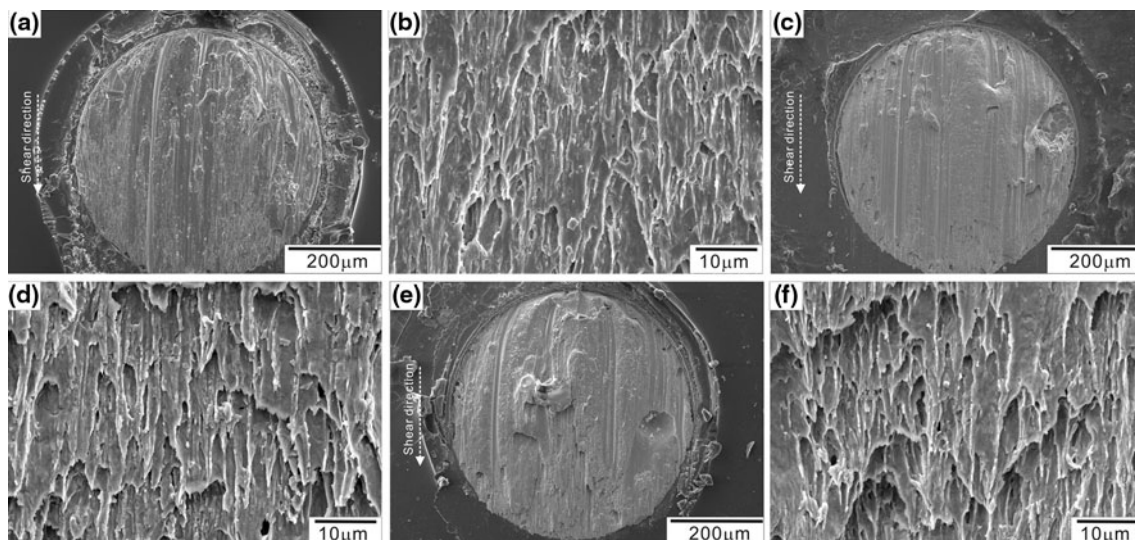
dislocation [5, 31]. It should also be noted that the shear strength increased in the SC composite solder joints. Zhong, et al. [32] has reported that the shearing forces of the Sn37Pb and Sn3Ag0.5Cu solder joints tend to increase slightly with an increase in the number of reflow cycles due to diffusion-induced solid solution strengthening of the bulk solder and augmentation of the shearing area.

Figures 7 and 8 show typical fractography of the SC composite solder joints in BGA packages after 1 and 8 cycles of reflow, respectively. Figure 7a and b shows the typical semi-brittle fracture mode with a relatively smooth surface (mark A), coarse IMC (mark B), and very rough dimples (mark C) occurring in the SC solder after one cycle of reflow. In addition, the  $\text{Cu}_6\text{Sn}_5$  IMC particles can be seen clearly at the center of each dimple. It can be speculated that the crack initiates and propagates at the brittle  $\text{Cu}_6\text{Sn}_5$  IMC in an early stage of the shearing. As a result, the typical fracture is seen as a combination of solder shear and solder/ $\text{Cu}_6\text{Sn}_5$  IMC separation (mark D). Hence, the average shearing force is lower than only that of SC composite solder joints [33].

However, after the addition of  $\text{TiO}_2$  nanoparticles, the fracture mode of the SC composite solder was dramatically changed. When 0.5 wt%  $\text{TiO}_2$  nanoparticles was added, the fracture mode clearly appeared to have a rough surface, small dimples, and a small partial ductility fracture in the edge area of the joint; i.e., the ductile fractured dimples throughout the matrix, as shown in Fig. 7c and d. From Fig. 7e and f, it is confirmed that the fracture surface of the SC-1TiO<sub>2</sub> composite solder joints exhibited a larger ductility fracture with a smaller dimple surface. The fracture occurred in all the solder joints as cracks propagated through the SC composite solder balls, which ruptured



**Fig. 7** Typical fracture surfaces of the SC composite solder joints after one reflow cycle: **a, b** SC, **c, d** SC-0.5TiO<sub>2</sub>, and **e, f** SC-1TiO<sub>2</sub>



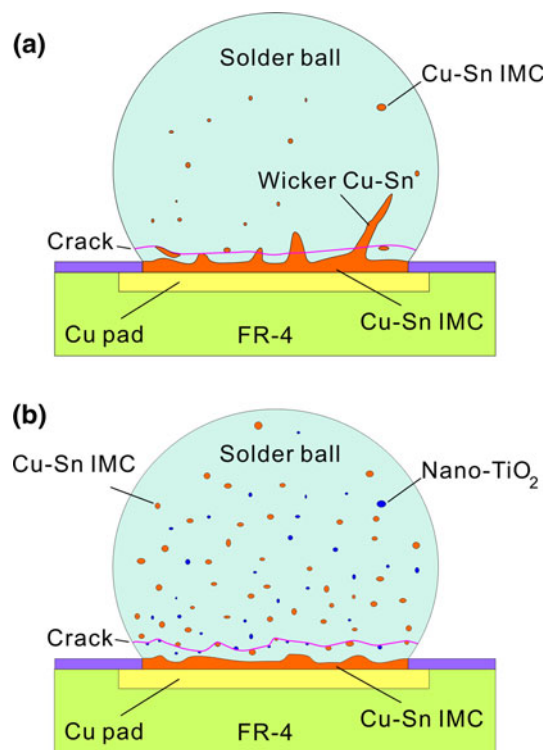
**Fig. 8** Typical fracture surfaces of the SC composite solder joints after eight reflow cycles: **a, b** SC, **c, d** SC-0.5TiO<sub>2</sub>, and **e, f** SC-1TiO<sub>2</sub>

mostly along the submicro Cu<sub>6</sub>Sn<sub>5</sub> IMCs and solder matrix. This indicates that the solder/pad bond is much stronger than the shear strength of the bulk solders.

On the other hand, the fracture mode of the SC composite solder joints changed obviously after eight cycles of reflow, as shown in Fig. 8. The fracture surfaces of SC solder joints after eight cycles of reflow exhibited a brittle fracture with a smooth surface, as shown in Fig. 8a and b. However, with the addition of TiO<sub>2</sub> nanoparticles, the fracture mode exhibited a typical ductile characteristic and a finer dimple surface, as shown in Fig. 8c and f. The schematic diagrams shown in Fig. 9 summarize the two fracture modes: A fracture surfaces of the solder joints revealed that the crack propagation occurred above the IMC/solder interface and revealed the dimple formation mechanism. Crack propagation along the intermetallic interface in the SC composite solders does not take place. This is likely a result of stresses within the solder being more homogeneously distributed and not concentrated at the IMC/solder interface. Thus, in SC composite solder, the fracture surfaces appeared to be ductile, with very fine dimpled surfaces, due to the homogeneous distribution of TiO<sub>2</sub> particles. Such enhanced ductility has profound implications for improving the mechanical shock resistance (such as drop failure) of electronic packages containing SC composite solders.

#### 4 Conclusions

In the present work, the interfacial reaction, morphology of IMCs, and shear strength of SC composite solder and BGA substrates with different numbers of reflow cycles were investigated. In addition, a more uniform microstructure of



**Fig. 9** Schematic drawing of shear fracture modes of BGA solder joints: **a** SC solder, **b** SC composite solder

solder alloys has been obtained in SC composite solder. A continuous layer of elongated-shaped Cu<sub>6</sub>Sn<sub>5</sub> IMCs was found at the interfaces in the SC solder joints after different numbers of reflow cycles, and wicker-Cu<sub>6</sub>Sn<sub>5</sub> IMCs grew into the solder matrix. It was found, however, that the addition of small amounts of TiO<sub>2</sub> nanoparticles to SC solders changes the morphology of the IMC layer by eliminating the wicker-Cu<sub>6</sub>Sn<sub>5</sub> IMCs and suppressing the

growth of the Cu<sub>6</sub>Sn<sub>5</sub> IMC layer. After multiple cycles of reflowing, the morphology of the IMC layer formed in SC composite solder/Cu joints changes from prism-type to scallop-type, and the average shear force is enhanced by about 2.6% in SC solder, 9.4% in SC-0.5TiO<sub>2</sub> solder, and 10.8% in SC-1TiO<sub>2</sub> solder. As a result, the shear strength of SC composite solder joints is improved. Despite the multiple reflow cycles, the fracture surface of SC solder exhibited a semi-brittle fracture mode with a relatively smooth surface, while SC composite solder showed typical ductile failures with very refined dimpled surfaces.

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