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Computation & theory



Performance of hybrid functional in linear combination of atomic orbitals scheme in predicting electronic response in spinel ferrites ZnFe₂O₄ and CdFe₂O₄

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ABSTRACT

Pure and hybrid density functional theory (DFT) schemes within linear combination of atomic orbitals (LCAO) have been employed to compute Mulliken population (MP), energy bands, density of states (DOS) and electron momentum densities (EMDs) of $TMFe_2O4$ (TM = Zn and Cd). Pure DFT calculations were performed within local density and generalized gradient approximations, while Hartree-Fock exchange contribution is added to DFT for hybrid calculations (B3LYP and PBE0). To validate the performance of hybrid functionals, we have also performed EMD measurements using 661.65 keV γ -rays from ¹³⁷Cs source. Chi-square test predicts an overall better agreement of experimental Compton profile data with LCAO-B3LYP scheme-based momentum densities leading to usefulness of hybrid functionals in predicting electronic and magnetic response of such ferrites. Further, LCAO-B3LYP-based majority- and minority-spin energy bands and DOS for $ZnFe_2O_4$ and $CdFe_2O_4$ predict semiconducting nature in both the compounds. In addition, MP data and equal-valence-electrondensity scaled EMDs show more covalent character of ZnFe2O4 than that of CdFe₂O₄. A reasonable agreement of magnetic moments of both the ferrites with available data unambiguously promotes use of Gaussian-type orbitals in LCAO scheme in exploring magnetic properties of such ferrites.

Introduction

Zinc and cadmium ferrites ($ZnFe_2O_4$ and $CdFe_2O_4$) crystallize in cubic spinel structure and are classified as important functional materials due to their peculiar electronic and magnetic properties [1–4]. Regarding earlier studies, Evans et al. [1] have employed Mössbauer spectroscopy (MS) to calculate nuclear quadrupole coupling constants and isomer shifts. Temperature-dependent electric-field gradients of both the ferrites were studied using time differential perturbed angular correlation technique by Pasquevich and Shitu [2]. Further, density functional theory (DFT) with generalized gradient approximation

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(GGA) and GGA plus one side Coulomb interaction (GGA + U) schemes were applied to highlight Fe–Fe interactions [3]. DFT within local density approximation (LDA), GGA, LDA + U and GGA + U schemes were employed by Cheng and Liu [4] to discuss the cation distribution effect in both the compounds. Electronic and magnetic properties of ZnFe₂O₄ were reported using DFT schemes with different approximations, namely atomic-sphere approximations [5], general potential linearized augmented plane wave method (LAPW) [6], plane wave pseudopotential (PWP) [7] and full potential (FP)-LAPW [8]. On the experimental side, structural and magnetic properties of ZnFe₂O₄ were explored by neutron diffraction, MS, X-ray diffraction (XRD), scanning electron microscopy, energy dispersive spectroscopy, vibrating sample magnetometer and Fourier transform infrared (FTIR) measurements [9–11]. Also, Quintero et al. [12, 13] have studied ZnFe₂O₄ using FP-LAPW method and MS measurements to visualize effect of defects on structural, electronic, hyperfine and magnetic properties. In case of CdFe₂O₄, Mahmood et al. [14] have used DFT with Perdew-Becke-Ernzerhof (PBE) revised for solids (PBESol) with modified Becke-Johnson (mBJ) exchange-correlations potentials to explain optical, magnetic and thermoelectric properties. Measurements on CdFe2O4 which include XRD, electrical conductivity, thermoelectric power, magnetic hysteresis, initial magnetic susceptibility, infrared spectroscopy and MS, inelastic neutron scattering and structural and transport properties have been reported by different groups [15–17]. In addition, density of states (DOS) along with optical and X-ray magnetic circular dichroism properties of CdFe2O4 were discussed by Zaari et al. [18] using FP-LAPW-DFT with mBJ potentials.

It is well known that Compton scattering (CS) measurements can be uniquely applied in testing various exchange–correlation potentials through electron momentum densities (EMDs) [19, 20]. The projection of EMD along the *z*-axis (scattering vector direction) is measured in CS experiment and is defined as Compton profile (CP), $J(p_z)$. Mathematically,

$$J(p_z) = \int_{p_x = -\infty}^{+\infty} \int_{p_{y = -\infty}}^{+\infty} \rho(\overrightarrow{p_x}, \overrightarrow{p_y}, \overrightarrow{p_z}) d\overrightarrow{p_x} d\overrightarrow{p_y}$$
(1)

Here, $\rho(\vec{p_x}, \vec{p_y}, \vec{p_z})$ represents the momentum distribution of the electrons while $\vec{p_x}, \vec{p_y}$ and $\vec{p_z}$ are the

components of electron linear momentum (\vec{p}) along *x*-, *y*- and *z*-directions of the Cartesian coordinate system. Experimentally, we deduce CP from the measured double differential Compton cross section using the relations,

$$J(p_z) = \frac{\text{Double differential Compton cross-section}}{C(E_1, E_2, \theta, p_z)}$$
(2)

Here, $C(E_1, E_2, \theta, p_z)$ depends on the experimental setup with E_1 and E_2 being the incident and scattered energies of photons and θ is the photon scattering angle. The variable *C* in Eq. 2 depends on the experimental geometry also.

In the present work, we have employed CRYS-TAL14 code [21] to compute Mullikan population (MP), energy bands, DOS and CPs for TMFe₂O₄ (TM = Zn and Cd). The purpose of present CP measurements is hands-on validation of various types of exchange and correlation potentials and hybrid functionals for reliable electronic properties under the frame work of linear combination of atomic orbitals (LCAO). We have also scaled the experimental and theoretical CP on equal-valence-electrondensity (EVED) to predict a relative nature of bonding in these iso-electronic compounds. Furthermore, trend of bonding in both the ferrites has also been validated by the present MP analysis. Going beyond CPs, energy bands, DOS, band gaps and magnetic moments of both the compounds have been compared with the available data which enables to conclude about applicability of different types of exchange-correlation energy and hybrid schemes in such type of ferrites.

Methodologies

Theory

LCAO calculations have been performed within the hypothesis of DFT and hybridizations of Hartree– Fock (HF) to DFT (so called B3LYP and PBE0) as embodied in the CRYSTAL14 software [21]. We have adopted the LDA and GGA under the DFT scheme. In LCAO calculations, one solves the one electron Schrodinger equation ($\hat{H}\psi = E\psi$) to obtain the crystal wave functions. The Hamiltonian energy operator (\hat{H}) includes kinetic energy, electrostatic potential (arising due to interaction of nuclei with electrons), electrostatic repulsion between electrons and exchange–correlation density functional energy (E_{XC}). First three part of \hat{H} are same in DFT–LDA, DFT–GGA, B3LYP and PBE0 approximations, whereas differences among these approximations occur in terms of E_{XC} . In case of DFT within LDA or GGA, E_{XC} is defined as,

$$E_{XC}^{\text{LCAO-DFT-LDA/GGA}}\{n(\vec{r})\} = \int n(\vec{r})\varepsilon_{xc}\{n(\vec{r})/(n(\vec{r}), |\nabla n(\vec{r})|)\}d\vec{r}$$
(3)

In Eq. 3, ε_{xc} is known as exchange–correlation energy per particle in uniform electron gas and $n(\vec{r})$ is the electron density. In case of LCAO–DFT–LDA scheme, we have adopted exchange and correlation potentials of Dirac-Slater [21] and Perdew and Zunger [22], respectively. While the exchange and correlation energies of PBESol [23] were considered for LCAO–DFT–GGA approximations. In case of hybridized (HF + DFT) approximations (B3LYP and PBE0), $E_{\rm XC}$ is defined as:

$$\begin{split} E_{XC}^{\text{LCAO}-\text{B3LYP}} &= A * E_X^{\text{LDA}} + B * \Delta E_X^{\text{BECKE}} + (1-A) * E_X^{\text{HF}} \\ &+ C * E_C^{\text{LYP}} + (1-C) * E_C^{\text{VWN}} \end{split}$$

$$E_{XC}^{LCAO-PBE0} = D * E_X^{HF} + (1 - D) * E_X^{PBE} + E_C^{PBE}$$
 (5)

The standard values of *A*, *B*, *C* and *D* are 0.80, 0.72, 0.81 and 0.25, respectively [24, 25]. Further E_X^{HF} , E_X^{LDA} , E_X^{BECKE} and E_X^{PBE} are the exchange energies corresponding to HF [21], Dirac-Slater [21], Becke [26] and Perdew et al. [27]. E_C^{LYP} , E_C^{VWN} and E_C^{PBE} are the correlation energies of Lee et al. [28], Vosko et al. [29] and Perdew et al. [27], respectively.

Further, all electron basis sets of Zn, Cd, Fe and O atoms [30] were used after the re-optimization for the lowest energy of both the ferrites using BILLY code [21]. The lattice parameters, position of atoms and space group of spinel ZnFe₂O₄ and CdFe₂O₄ were adopted from Quintero et al. [13] and Zaari et al. [18], respectively. The self-consistent field (SCF) calculations for both the compounds were performed with 29 **k** points in the irreducible Brillouin zone (BZ). The unit cell in cubic spinel structure for ZnFe₂O₄ is plotted using XCrysden visualization software of Kokalj [31] (Fig. 1a) and the corresponding first BZ structure is sketched in Fig. 1b.

Experiment

To ratify the choice of exchange and correlation potentials through reconciliation of theoretical and experimental CPs, we have employed 20 Ci ¹³⁷Cs Compton spectrometer [32] to measure CPs of $TMFe_2O_4$ (TM = Zn and Cd). In the present measurements, an absolute momentum resolution (Gaussian full width at half maximum) of the experimental setup was 0.34 a.u. Due to difficulties in growing large size single crystals (15 mm diameter and 2 mm thickness) and to discuss the relative nature of bonding on EVED scale of ZnFe₂O₄ and CdFe₂O₄, we have taken pallets of high purity (> 99%) polycrystalline powder of both the ferrites. Further, γ-radiations of energy 661.65 keV were allowed to incident on pallets of individual ferrite and the scattered radiations $(160 \pm 0.6^{\circ})$ scattering angle) were energy analyzed by a high purity Ge detector (GL0510P, Canberra made). During the exposure time of 162.4 (273.8) h for ZnFe₂O₄ (CdFe₂ O₄), the integrated Compton intensity was found to be 3.17×10^7 (4.44 $\times 10^7$) counts. The stability of the acquisition system was monitored from time-to-time by two weak radio-isotopes, namely ⁵⁷Co and ¹³³Ba. To obtain absolute CP, the raw Compton spectra of both the ferrites were corrected for systematic corrections (background, detector efficiency, sample absorption, stripping-off the low energy tail and CS cross section) with computer code of Warwick group [33]. To obtain true singly scattered photon profile, we have also corrected the data for the effect up to triple scattering using Monte Carlo method [34]. Each Compton line was normalized to corresponding free atom (FA) CP area using the tabulated values of Biggs et al. [35]. The value of FA CP area for ZnFe₂O₄ $(CdFe_2O_4)$ was 51.62 (57.77) e^- in the momentum range 0–7 a.u.

Results and discussion

MP analysis

MP data of charge transfer for $TMFe_2O_4$ (TM = Zn and Cd) using LCAO–DFT–LDA, LCAO–DFT–GGA, LCAO–B3LYP and LCAO–PBE0 schemes have been presented in Table 1. Here, the transition metal atoms (Zn/Cd and Fe) in TMFe₂O₄ (TM = Zn/Cd) donate charge to the oxygen atoms and total charge from







donor is equally distributed among O atoms for both the compounds. The total numerical values of charge transfer in ZnFe₂O₄ (CdFe₂O₄) are 4.12 (4.32), 4.16 (4.40), 4.44 (4.60) and 4.52 (4.72) e^- using LCAO– DFT–LDA, LCAO–DFT–GGA, LCAO–B3LYP and LCAO–PBE0 schemes, respectively. It is observed that the charge transfer in CdFe₂O₄ is higher than that in ZnFe₂O₄ for each approximation (LCAO–DFT– LDA/LCAO–DFT–GGA/LCAO–B3LYP/LCAO–PBE0). Such trend of MP data indicates more ionic (or less

Table 1 Mulliken's population (MP) data for $TMFe_2O_4$ (TM = Zn and Cd) using LCAO–DFT–LDA, LCAO–DFT–GGA, LCAO–B3LYP and LCAO–PBE0 schemes as mentioned in the text

Scheme	Amount of charge transfer (e^-)				
	Donor atom	Acceptor atoms			
	Zn/Cd (2)	Fe (4)	O (8)		
(a) $ZnFe_2O_4$					
LCAO-DFT-LDA	1.04	1.54	1.03		
LCAO-DFT-GGA	1.06	1.55	1.04		
LCAO-B3LYP	1.16	1.64	1.11		
LCAO-PBE0	1.18	1.67	1.13		
(b) $CdFe_2O_4$					
LCAO-DFT-LDA	1.18	1.57	1.08		
LCAO-DFT-GGA	1.22	1.59	1.10		
LCAO-B3LYP	1.30	1.65	1.15		
LCAO-PBE0	1.32	1.70	1.18		

Here, TM (Zn and Cd) and Fe atoms are the donor atoms while O atoms are the acceptor atoms. The numbers of equivalent atoms are shown in the brackets

covalent) character in $CdFe_2O_4$ than that in $ZnFe_2O_{4/4}$ as larger value of charge transfer in any compound reflects more ionicity. Also, in all the adopted approximations, contribution of Fe atom in total charge transfer data is found to be approximately similar for both the compounds. However, difference in total charge transfer exists due to the contribution of Zn/Cd atom in TMFe₂O₄ environment. In addition to MP data for charge transfer, we have also calculated overlap population (OP) between the nearest neighbor atoms in both the compounds. The OP value of identical Fe–O in TMFe₂O₄ (TM = Zn/Cd) are found to be 0.051, 0.051, 0.049 and 0.048 e^- using LCAO-DFT-LDA, LCAO-DFT-GGA, LCAO-B3LYP and LCAO-PBE0, respectively, for both the compounds. While these values for Zn-O (Cd-O) in case of $ZnFe_2O_4$ (CdFe_2O_4) are observed as 0.127 (0.047), 0.126 (0.045), 0.124 (0.039) and 0.120 (0.035) e^- using LCAO-DFT-LDA, LCAO-DFT-GGA, LCAO-B3LYP and LCAO-PBE0, respectively. The OP values of Zn-O in case of ZnFe₂O₄ are higher than that for Cd–O in CdFe₂O₄ using all the LCAO approximations. It indicates more covalent (or less ionic) character of ZnFe₂O₄ than that in CdFe₂O₄ because large value of OP indicates more covalent character of the compound. It is worth mentioning that MP analysis of charge transfer is bound to be quite reliable as we have included large number of atomic orbitals in each LCAO computation using DFT-LDA, DFT-GGA, B3LYP and PBE0 prescriptions. Such MP analysis have successfully been reported in various other oxides, viz. TMWO₄ (TM = Co, Ni, Cu, Zn and Cd)



[36, 37], Ag_2TMO_4 (TM = Cr and Mo) [38] and $BaTiO_3$ [39].

Energy bands and density of states

Majority (spin-up) and minority (spin-down) energy bands of LCAO-B3LYP scheme are presented in Figs. 2 and 3 for ZnFe₂O₄ and CdFe₂O₄, respectively. Except some fine structures, the energy bands of LCAO-B3LYP are in consonance with LCAO-DFT-LDA, LCAO-DFT-GGA and LCAO-PBE0 schemes; hence, energy bands are shown only for B3LYP scheme (which performed well in reproducing EMDs, as discussed later). Also, our energy bands of ZnFe₂O₄ and CdFe₂O₄ reasonably resemble with the available data [6, 7, 14]. From Figs. 2 and 3, a significant energy gap between the valence band maximum (VBM) and conduction band minimum (CBM) is observed in both the majority- and minority-spin channels of both the compounds. Present band structures indicate a direct band gap (E_{σ}) semiconducting nature of ZnFe₂O₄ and CdFe₂O₄, as VBM and CBM are found at the same Γ point. The band gap values for both the compounds are lower in case of minority-spin bands channels than that in majorityspin channel (Figs. 2, 3). In Table 2, we have collated band gap values derived using various approximation (LCAO-DFT-LDA, LCAO-DFT-GGA, LCAO-B3LYP and LCAO-PBE0) along with available theoretical and experimental data [7, 12-14, 18, 40-42] for TMFe₂O₄. Our band gap values using LCAO-DFT-

LDA and LCAO-DFT-GGA schemes underestimate the experimental band gap [40-42] for both the compounds, while a reverse trend is found for LCAO-B3LYP and LCAO-PBE0 schemes. It is seen that B3LYP approach shows a closer agreement with the experimental band gap than other schemes considered within the periphery of LCAO prescription. It is worthwhile to mention that most of the theoretical prescriptions using pure LDA and GGA schemes have underestimated the band gap values than the experimental band gaps, as evident from Table 2. This trend is consistent with the general observation that the HF calculations overestimate the band gap (because of lack of screening in the exchange term leading to an over stabilization of occupied states) and the DFT-based LDA and GGA theories underestimate the band gap. It is quite satisfying that in this situation the hybrid method (mixing of HF to DFT) like B3LYP minimizes the deviation of band gap from the experimental values and leads to reasonable electronic response for such compounds.

In Fig. 4a–d, we have shown DOS for spin-up (\uparrow) and spin-down (\downarrow) states using LCAO–B3LYP scheme for 3*d*, 4*s* states of Zn; 3*d*, 4*s* states of Fe; 2*s*, 2*p* states of O and total DOS for ZnFe₂O₄. Similarly, spin-projected DOS for CdFe₂O₄ environment are plotted in Fig. 5a–d. The Fermi energy (E_F) is shifted to 0 eV. The majority-spin and minority-spin DOS for both the compounds unambiguously reconfirm the semiconducting nature of both the compounds. In Fig. 4 (Fig. 5), the DOS in energy range – 8.34

Figure 2 a Majority- and **b** minority-spin energy bands of ZnFe₂O₄ using LCAO– B3LYP scheme along the high symmetry directions of BZ. The positions of Γ , X, W and L vertices are (0, 0, 0), (1/2, 0, 1/2), (1/2, 1/4, 3/4) and (1/2, 1/2, 1/2), respectively.





Figure 3 a Majority- and **b** minority-spin energy bands of CdFe₂O₄ using LCAO–B3LYP scheme along the high symmetry directions of BZ.

(-8.65) eV to the $E_{\rm F}$ level are mostly contributed by the 3d (4d) electrons of Zn (Cd) and 3d electrons of Fe atom along with a small contribution of 2p electrons of O atoms. In Fig. 4d, the majority-spin states dominate in the formation of the DOS in the energy range -8.34 to -6.77 eV, while the DOS in the energy range + 2.40 to + 5.23 eV are majorly contributed by the minority-spin states of ZnFe₂O₄. In case of $CdFe_2O_4$ (Fig. 5d), contribution of minority-spin (majority-spin) DOS are found to be absent in the energy range -7.36 to -6.46 (+2.10 to +4.52) eV. In Fig. 4a, the majority-spin DOS of Zn-3d states are dominated in the energy range - 8.34 to - 6.60 eV while the trend becomes reverse and dominancy of minority-spin DOS of 3d states of Zn prevails between -6.60 and -5.53 eV. Similarly, major role of majority-spin states of Fe is found between the energy range -8.34 to -3.30 eV (Fig. 4b). In the conduction band region between +2.40 and + 5.23 eV, the DOS are majorly governed by 3d minority-spin states of Fe atom along with a small contribution of minority-spin states of 2p electrons of O atom (Fig. 4b, c). In Fig. 5a, major contribution of majority-spin states of 4d electrons of Cd is seen between - 8.65 and - 7.89 eV and a reverse trend is observed between - 7.89 and - 7.38 eV. For 3d electrons of Fe (Fig. 5b), the majority-spin contribution dominates in the valence band region (- 8.65 to - 0.36 eV) and minority-spin contribution dominates in conduction band region (+ 2.10 to + 4.52 eV) of CdFe₂O₄. From Fig. 4c for ZnFe₂O₄ and Fig. 5c for

CdFe₂O₄, a small contribution below the VBM is found from majority- and minority-spin states of O, while the contribution of only minority-spin states of O atoms in CBM is witnessed. It is observed that the VBM is formed by the majority-spin states of 3*d* (4*d*) of Zn (Cd) and 3*d* states of Fe along with the a small contribution of majority-spin O-2*p* states for ZnFe₂O₄ (CdFe₂O₄). Also, the CBM is built mainly by minority-spin of 3*d* (4*d*) states of Fe along with a small contribution of minority-spin of 2*p* states of O atom in ZnFe₂O₄ (CdFe₂O₄).

Compton profiles

In Fig. 6a, b, the anisotropies between unconvoluted theoretical CPs ($J_{110}-J_{100}$, $J_{111}-J_{100}$ and $J_{111}-J_{110}$) using LCAO-DFT-LDA, LCAO-DFT-GGA, LCAO-B3LYP and LCAO-PBE0 schemes are presented for ZnFe2O4 and CdFe₂O₄, respectively. An overall trend of oscillations in the anisotropies (J_{110} – J_{100} , J_{111} – J_{100} and $J_{111}-J_{110}$) for TMFe₂O₄ (TM = Zn and Cd) using LCAO computations (DFT-LDA, DFT-GGA, B3LYP and PBE0) are found to be similar. In the higher momentum region ($p_z \ge 4.0$ a.u.), negligible anisotropic effects in momentum densities of both the compounds have been observed for all the considered schemes (LCAO-DFT-LDA, LCAO-DFT-GGA, LCAO-B3LYP and LCAO-PBE0). In high momentum region, such observations are quite expected as core electrons which contribute to formation of CPs in this region remain unaffected in directional momentum densities. Anisotropic effects are observed in the low



Table 2 Band gap (E_g) for ZnFe₂O₄ and CdFe₂O₄ using various combinations of exchange–correlation potentials within LCAO schemes as mentioned in the text along with the available data

Approach	Band gap (E_g) in eV			
	ZnFe ₂ O ₄	CdFe ₂ O ₄		
(i) Present computations				
LCAO–DFT–LDA	1.06	0.52		
LCAO-DFT-GGA	1.37	0.40		
LCAO-B3LYP	2.40	2.10		
LCAO-PBE0	3.22	3.26		
(ii) Available theories				
(a) Plane wave pseudopotential (PWP) [7] with				
LDA-CA-PZ	0.87	_		
GGA-PBE	0.84	_		
GGA–RPBS	0.93	_		
GGA–PW91	0.79	_		
GGA-WC	0.87	_		
GGA-PBESOL	0.80	_		
(b) $FP-LAPW + lo [12]$	2.20	_		
(c) FP-LAPW + lo with $GGA + U$ [13]				
Non inverted	2.00	_		
Inverted	2.10	_		
(d) Projector-augmented wave (PAW) [40] with		_		
GGA + U (spinel)	1.68	_		
PBE0 (spinel)	3.68	_		
GGA + U (inverse)	1.91	_		
PBE0 (inverse)	3.37	_		
(e) FP-LAPW with PBESOL $+ mBJ [14]$	_	1.86		
(f) FP-LAPW–DFT [18] with				
GGA + U	_	1.56		
TB-mBJ	-	1.88		
(iii) Available experiment				
Yao et al. [7]	1.90	_		
Valeznuela et al. [41]	1.92	_		
Akamatsu et al. [42]	_	1.46		

momentum region ($p_z \leq 4.0$ a.u.), which are dictated by dispersive nature of energy bands and their degeneracy in different branches of BZ. In low momentum side, the trend of anisotropies in hybrid schemes (LCAO-B3LYP and LCAO-PBE0) is almost similar, while it differs from the overall trend seen for LCAO-DFT-LDA and LCAO-DFT-GGA schemes. Such a trend in hybrid calculations may be due to the incorporation of HF exchange energy component in B3LYP and PBE0 schemes, which was absent in DFT with LDA and GGA. As expected, the oscillations in CP anisotropies in Fig. 6a–b are also in consonance with the cross-overs and degenerate states of majority- and minority-spin energy bands (Figs. 2, 3) for both the compounds. The positive values of anisotropies $(J_{111}-J_{100} \text{ and } J_{111}-J_{110})$ in Fig. 6a, b near $p_z = 0.0$ a.u. are due to the large degenerate states along [111] direction as compared to [100] and [110] directions. Decreasing order of $J(p_z = 0.0 \text{ a.u.})$ values for the three crystallographic directions is found to be $[111] \rightarrow [110] \rightarrow [100]$ for both the compounds which shows signatures of atomic density effects in different principal orientations. Also, the negative amplitude near $p_z = 1.5$ a.u. along $J_{111}-J_{100}$ in Fig. 6a, b arises due to zone boundary of $\Gamma-X$ (0.39 a.u.; n = 4 for ZnFe_2O_4 and 0.38 a.u.; n = 4 for CdFe_2O_4) branch. It may be noted that some fine structures may be invisible in the anisotropies (Fig. 6) because of cancellation effect on taking the differences of CPs. Anisotropic measurements of CPs for ZnFe_2O_4 and CdFe_2O_4 are necessitated to validate the theoretical anisotropic effects in momentum densities (Fig. 6).

The difference in CPs between convoluted theoretical (LCAO–DFT–LDA, LCAO–DFT–GGA, **Figure 4** Majority-spin (\uparrow) and minority-spin (\downarrow) density of states (DOS) for **a** 3*d* and 4*s* states of Zn, **b** 3*d* and 4*s* states of Fe, **c** 2*s* and 2*p* states of O and **d** total ZnFe₂O₄ using LCAO– B3LYP approximation.







Figure 6 Anisotropies in the unconvoluted theoretical Compton profiles of **a** $ZnFe_2O_4$ and **b** $CdFe_2O_4$ calculated using LCAO–DFT–LDA, LCAO–DFT–GGA, LCAO–B3LYP and LCAO–PBE0 approximations for the directional pairs (i) J_{110} – J_{100} , (ii) J_{111} – J_{100} and (iii) J_{111} – J_{100} . The solid lines are drawn for a quick view of trend.



(b) 0.9 LCAO-DFT-LDA (i) J₁₁₀-J₁₀₀ 0.6 LCAO-DFT-GGA -0-LCAO-B3LYP 0.3 LCAO-PBE0 0.0 -0.3 -0.6 -0.9 (ii) J₁₁₁-J₁₀₀ 0.9 (e/a.u.) 0.6 0.3 0.0 2 -0.3 -0.6 -0.9 0.6 (iii) J₁₁₁-J₁₁₀ 0.3 0.0 -0.3 ō 3 5 6 p_z (a.u.)

LCAO-B3LYP and LCAO-PBE0) and experimental data have been plotted in Fig. 7a, b for ZnFe₂O₄ and CdFe₂O₄, respectively. The numerical values of unconvoluted theoretical (LCAO-DFT-LDA, LCAO-DFT-GGA, LCAO-B3LYP and LCAO-PBE0) and experimental CPs for ZnFe₂O₄ and CdFe₂O₄ (with statistical errors) have been collated in Tables 3 and 4, respectively. In Tables 3 and 4, total CPs for each approximation is calculated by adding the FA core CP contribution [35] to the respective LCAO-based CP data of valence electrons. Also, to mimic experimental resolution, each theoretical data is convoluted with fwhm of 0.34 a.u. (momentum resolution of present experimental setup) before taking the difference between theoretical and experimental CPs. From Tables 3, 4 and Fig. 7a, b, it is observed that experimental CP data of both the compounds underestimate the momentum densities in the low momentum region ($p_z \leq 1.0$. a.u.). This trend reverses in the region $1.0 \le p_z \le 4.0$ a.u. Also, the difference in the higher momentum region ($p_z \ge 4.0.$ a.u.) is very small (within the experimental error) for CdFe₂O₄. Such trend in high momentum region is expected because of contribution of core electrons whose wave functions are well defined by HF approximation. For quantitative conclusions related to goodness of agreement between theoretical CP with the experimental data for both the compounds, we have



Figure 7 Different profiles deduced from isotropic convoluted theoretical (LCAO–DFT–LDA, LCAO–DFT–GGA, LCAO–B3LYP and LCAO–PBE0 approximations) and experimental Compton profiles for **a** $ZnFe_2O_4$ and **b** CdFe₂O₄. The solid lines are drawn for a quick view of trend.

undertaken χ^2 fitting. For this purpose, we have used the following formula,

due to the different contribution of hybridization in exchange and correlation potentials and the per-

$$\chi^{2} = \sum_{p_{z}=0}^{7} \left[\frac{J^{\text{LCAO}-\text{DFT}-\text{LDA}/\text{LCAO}-\text{DFT}-\text{GGA}/\text{LCAO}-\text{B3LYP}/\text{LCAO}-\text{PBE0}(p_{z}) - J^{\text{Experiment}}(p_{z})}{\sigma(p_{z})} \right]^{2}$$
(6)

Here, $\sigma(p_z)$ is the statistical error at each p_z value. The χ^2 value for ZnFe₂O₄ (CdFe₂O₄) using LCAO– DFT–LDA, LCAO–DFT–GGA, LCAO–B3LYP and LCAO–PBE0 approximations are computed as 4540.14 (2847.40), 4484.60 (2776.84), 3951.46 (2356.82) and 4189.12 (2534.73), respectively. Hence, the lowest value of χ^2 indicates an overall better agreement by LCAO–B3LYP scheme for both the compounds. Further similar type of performance of both the hybridized approximations (LCAO–B3LYP and LCAO–PBE0) is due to an incorporation of HF exchange effects in such computations. The difference in LCAO–B3LYP and LCAO–PBE0 data may be

centage of mixing of HF with DFT (being 20 and 25% in B3LYP and PBE0 schemes). In the low momentum side differences between theoretical and experimental CPs may be due to non-inclusion of relativistic effects and Lam–Platzman (LP) correlation [19] in LCAO calculations and further possibility for improvement in the quality of Gaussian basis sets used in the present computations.

EVED

To highlight relative nature of bonding in the present iso-electronic compounds, we have rescaled the

Table 3 Unconvoluted theoretical (LCAO–DFT–LDA, LCAO–DFT–GGA, LCAO–B3LYP and LCAO–PBE0) and experimental Compton profiles along with statistical errors ($\pm \sigma$) for ZnFe₂O₄

<i>p</i> _z (a.u.)	$J(p_z)$ (e/a.u.)					
	Theory	Expt.				
	LCAO–DFT–LDA	LCAO-DFT-GGA	LCAO–B3LYP	LCAO-PBE0		
0.0	26.136	26.117	25.996	26.029	24.659 ± 0.058	
0.1	26.098	26.080	25.960	25.993	24.571 ± 0.058	
0.2	25.831	25.816	25.700	25.734	24.341 ± 0.057	
0.3	25.394	25.382	25.280	25.314	23.954 ± 0.057	
0.4	24.728	24.719	24.643	24.676	23.407 ± 0.056	
0.5	23.955	23.950	23.900	23.932	22.714 ± 0.054	
0.6	22.990	22.988	22.953	22.982	21.905 ± 0.053	
0.7	21.878	21.882	21.858	21.884	21.003 ± 0.052	
0.8	20.621	20.630	20.631	20.653	20.014 ± 0.050	
1.0	17.952	17.968	18.018	18.030	17.772 ± 0.046	
1.2	15.341	15.357	15.395	15.392	15.540 ± 0.042	
1.4	13.143	13.155	13.183	13.170	13.484 ± 0.038	
1.6	11.248	11.254	11.281	11.265	11.678 ± 0.035	
1.8	9.718	9.720	9.751	9.735	10.134 ± 0.031	
2.0	8.468	8.467	8.495	8.480	8.920 ± 0.029	
3.0	4.794	4.791	4.793	4.785	5.002 ± 0.018	
4.0	3.006	3.006	3.010	3.007	3.099 ± 0.012	
5.0	2.092	2.091	2.093	2.092	2.193 ± 0.010	
6.0	1.542	1.542	1.543	1.543	1.651 ± 0.007	
7.0	1.167	1.167	1.167	1.167	1.183 ± 0.005	



<i>p</i> _z (a.u.)	$J(p_z)$ (e/a.u.)					
	Theory	Expt.				
	LCAO–DFT–LDA	LCAO-DFT-GGA	LCAO–B3LYP	LCAO-PBE0		
0.0	28.354	28.329	28.211	28.243	27.237 ± 0.059	
0.1	28.308	28.285	28.162	28.194	27.089 ± 0.059	
0.2	28.042	28.021	27.897	27.929	26.792 ± 0.058	
0.3	27.586	27.569	27.460	27.492	26.347 ± 0.058	
0.4	26.874	26.861	26.784	26.816	25.750 ± 0.056	
0.5	26.051	26.042	25.997	26.027	25.008 ± 0.055	
0.6	25.039	25.035	25.006	25.034	24.128 ± 0.054	
0.7	23.859	23.861	23.844	23.869	23.123 ± 0.052	
0.8	22.534	22.542	22.550	22.570	22.009 ± 0.050	
1.0	19.726	19.742	19.793	19.802	19.599 ± 0.046	
1.2	16.961	16.978	17.019	17.015	17.192 ± 0.042	
1.4	14.589	14.602	14.637	14.625	14.918 ± 0.038	
1.6	12.502	12.510	12.546	12.532	12.898 ± 0.034	
1.8	10.789	10.793	10.825	10.812	11.221 ± 0.031	
2.0	9.386	9.387	9.412	9.400	9.854 ± 0.028	
3.0	5.317	5.315	5.319	5.312	5.543 ± 0.017	
4.0	3.555	3.554	3.554	3.552	3.601 ± 0.012	
5.0	2.646	2.646	2.647	2.645	2.653 ± 0.009	
6.0	2.042	2.042	2.042	2.041	2.079 ± 0.007	
7.0	1.587	1.587	1.587	1.587	1.609 ± 0.006	

Table 4 Unconvoluted theoretical (LCAO–DFT–LDA, LCAO–DFT–GGA, LCAO–B3LYP and LCAO–PBE0) and experimental Compton profiles along with statistical errors ($\pm \sigma$) for CdFe₂O₄

LCAO-B3LYP-based CP and experiment profile on EVED (p_z/p_F) parameters (Fig. 8). Values of Fermi momentum (p_F) for ZnFe₂O₄ and CdFe₂O₄ were taken as 1.48 and 1.42 a.u., respectively. The outer electrons configurations of Zn (Cd), Fe and O atoms in $TMFe_2O_4$ (TM = Zn and Cd) environment are considered as $3d^{10}4s^2$ ($4d^{10}5s^2$), $3d^64s^2$ and $2s^22p^4$, respectively. Since the total valence electrons in both the iso-electronic compounds are 52, therefore each EVED profile has been normalized to 26 e^- in the major p_z range of 0–2 a.u. From Fig. 8, EVED ($p_z/$ $p_{\rm F} = 0$) value of CdFe₂O₄ using LCAO-B3LYP (experiment) is found to be 2.61 (0.76) % lesser than that of ZnFe₂O₄. Such a trend indicates more ionic (or less covalent) character of CdFe₂O₄ than ZnFe₂O₄. This is admittedly due to a fact that the lower value of $J(p_z/$ $p_{\rm F} = 0.0$) indicates the lesser sharing of the electrons along the bond directions which further leads to lesser covalent (or higher ionic) character in the compound. Trend shown by present theoretical (B3LYP) and experimental EVED profiles is also in accordance with earlier discussed MP and OP analysis. Such type of confirmation of charge transfer and

overlap populations (as deduced from MP analysis) was also quite successful in other compounds like TMWO₄ (TM = Zn and Cd) [37], Ag₂TMO₄ (TM = Cr and Mo) [38] and TMTiO₃ (TM = Ba, Sr and Ca) [39].

LCAO-based magnetization

Local magnetic moments for Fe ions in ZnFe₂O₄ (CdFe₂O₄) have been calculated using LCAO-DFT-LDA, LCAO-DFT-GGA, LCAO-B3LYP and LCAO-PBE0 schemes and the values are found to be 3.47 (3.87), 3.99 (4.07), 4.24 (4.24) and 4.31 (4.32) $\mu_{\rm B}$ per formula unit, respectively. Here, the unit cell magnetic moment for ZnFe₂O₄ (CdFe₂O₄) are also calculated as 7.87 (9.23), 9.57 (9.93), 9.99 (9.99) and 10.01 (10.01) μ_B from LCAO–DFT–LDA, LCAO–DFT–GGA, LCAO-B3LYP and LCAO-PBE0 schemes, respectively. Our LCAO-B3LYP formulation-based magnetic moments of Fe ions for both the ferrites are in accordance with the available data [3, 8, 9, 12–14]. In case of LCAO-B3LYP, the magnetic moment of Fe ions is approximately equal to the experimental value of 4.22 $\mu_{\rm B}$. Also, our unit cell magnetic moment using



Figure 8 Equal-valence-electron-density (EVED) profiles of isoelectronic ZnFe₂O₄ and CdFe₂O₄ for **a** LCAO–B3LYP and **b** experimental. In the inset, the values of $J(p_z/p_F)$ near $p_z/p_F = 0$ are shown.

LCAO–B3LYP scheme for ZnFe₂O₄ is found to be same as reported by Soliman et al. [8] using FP-LAPW approach.

Conclusions

MP analysis, energy bands, DOS, band gaps and CPs computed using DFT–LDA, DFT–GGA, B3LYP and PBE0 within LCAO approximations along with Compton profile measurements for $ZnFe_2O_4$ and $CdFe_2O_4$ are presented. Further, MP analysis shows the charge transfer from zinc/cadmium and iron to oxygen atoms. On the basis of EVED CPs and MP data, more ionic character in CdFe₂O₄ than that in ZnFe₂O₄ was observed. The magnetic moments are well explored by the LCAO–B3LYP approximation. A closer agreement of LCAO–B3LYP-based Compton profiles with the experimental CPs, reproducibility of

band gaps, confirmation of trend of B3LYP-based MP and OP data together with electronic response and magnetization data unambiguously warrants use of hybrid functionals for exchange and correlation potentials in the spinel ferrites, as reported here.

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Compliance with ethical standards

Conflict of interest The authors declare that they have no conflict of interest.

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