



Amphiphilic polymer–drug conjugates based on acid-sensitive 100% hyperbranched polyacetals for cancer therapy

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ABSTRACT

A new type of acid-sensitive 100% hyperbranched polyacetals (HBPA) was synthesized, which could be completely degraded into small molecules under acidic environment and avoid the accumulative toxicity in vivo. The AB₂ monomer was synthesized by 4-carboxybenzaldehyde and 2-bromoethanol. The bulk polycondensation was carried out in vacuum environment to remove water byproduct. The massive terminal aldehyde groups of HBPA were conjugated with mPEG-NH₂ and doxorubicins to form amphiphilic acid-sensitive polymer–drug conjugates (DOX-HBPA-PEG). The stability of the micelles of DOX-HBPA-PEG was evaluated by DLS at different pH value in phosphate buffer saline (PBS). The DOX release in vitro showed that the cumulative release rate was 14.51% in pH 7.4 PBS after 24 h and the cumulative release rate was 48.56% in pH 6.0 PBS after 24 h. The results of cell viability of DOX-HBPA-PEG and HBPA-PEG showed that the polymer–DOX conjugates were effective drug delivery systems. The uptake process of DOX-HBPA-PEG by A549 cells showed that the micelle was totally swallowed in 1 h later. The controllable drug release nature, stability, biocompatibility and completely degradable structures (acid-sensitive) make them to be promising drug delivery systems.

Introduction

Nowadays, numerous drug delivery systems have been reported for delivering anticancer drugs to tumors based on inorganic carriers of mesoporous silica nanocapsules [1, 2], nanobubbles [3], carbon

nanotubes [4, 5] and Au nanoparticles [6, 7] as well as organic carriers of nanoporous polymers [8], hyperbranched polymer [9], dendrimers [10], polymer–drug conjugates [11, 12] and polymer micelles [13]. Most of drug delivery systems are fabricated from polymers or macromolecules. The biocompatibility,

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biodegradability and water solubility of polymers or macromolecules were usually evaluated before use. For drug delivery system, the macromolecules or polymers cannot pass through the endothelium of normal blood vessels [14]; at the same time, the metabolism and excretion are pretty hard for macromolecules or polymers. The rate of renal elimination is inversely correlated with the molecular weight of the polymers [15, 16]. The polymers need to be degraded into small molecules and excreted via kidney, otherwise resulting in accumulative toxicity in vivo. Conversely, the small molecules are delivered systemically and consequently exhibit a non-specific biodistribution, short plasma circulation times and rapid systemic elimination [17]. These properties avoid the accumulative toxicity in specific normal tissues of polymers or macromolecules.

The polymers or macromolecules used for drug delivery should be eliminated from the body, either by excretion of non-biodegradable polymers or by degradation of biodegradable polymers [18]. The backbones of biodegradable polymer usually are linked by ester bonds [19–22], ether bonds [23–27], carbon–carbon bonds [28, 29] and amide bonds [30, 31]. This type of polymer cannot be controllable and rapidly degraded into small molecules in target tissue or in vivo, resulting in long degradation time and accumulative toxicity to normal tissues in vivo. Although many drug delivery systems with partial stimulated bonds or polymer can respond to the change of environment, like temperature [32, 33], pH value [34, 35], light [36, 37], redox-reduction reagent [38, 39] and magnetic field [40, 41], most of polymers or macromolecules for drug delivery systems cannot be completely degraded into small molecules with partial stimulated linker bonds of disulfide bond [38, 42, 43], imine bond [29] and acetal bond [19, 42]. These linker bonds need to be repeat units in backbone of polymers or macromolecules for complete degradation. The backbone with pH or glutathione responsive bonds [44–47] can be completely degraded into small molecule in tumor cells because of the environment of tumor cells with low pH value and high concentration of glutathione.

The perfect polymers or macromolecules in drug delivery system for cancer therapy should be degraded into small molecule in the environment of low pH or high concentration of glutathione after delivering anticancer drugs to tumor cells. The polymers or macromolecules cannot freely pass through the

cell membrane. Only small molecule drugs or monomer can freely pass through the cell membrane followed by blood stream to kidney to excrete via urinary tract. So the completely degradable polymers under low pH or glutathione can avoid the accumulative toxicity of polymers or macromolecules in vivo.

In this paper, we synthesized 100% hyperbranched polyacetals that can completely be degraded into monomers with aldehyde and hydroxyl groups in acidic environment in tumor cells. The hyperbranched polyacetals possess massive terminal aldehyde groups that can react with amino-terminated monopolyethylene glycol and DOX. The micelles of DOX-HBPA-PEG were characterized by NMR, FTIR, SEC and DLS. The properties of drug release and stability were evaluated in vitro, and the cell viability and uptake process were evaluated by A549 cell line in vitro.

Materials and methods

Materials

Doxorubicin hydrochloride (DOX·HCl) was purchased from Aladdin Co. (China). Amino-terminated monopolyethylene glycol (mPEG-NH₂, $M_w = 2000$ - Da) was purchased from Shanghai Yare Biotech, Inc. The 4-carboxybenzaldehyde and 2-bromoethanol, *N,N*-dimethyl formamide (DMF) and tetrahydrofuran (THF) were purchased from J&K Scientific Ltd. (Beijing, China); CCK-8 and Hoechst Staining Kit were purchased from Beyotime company. The pyridinium camphorsulfonate (PCS) catalyst was synthesized using a similar protocol to the literature [46].

Characterization

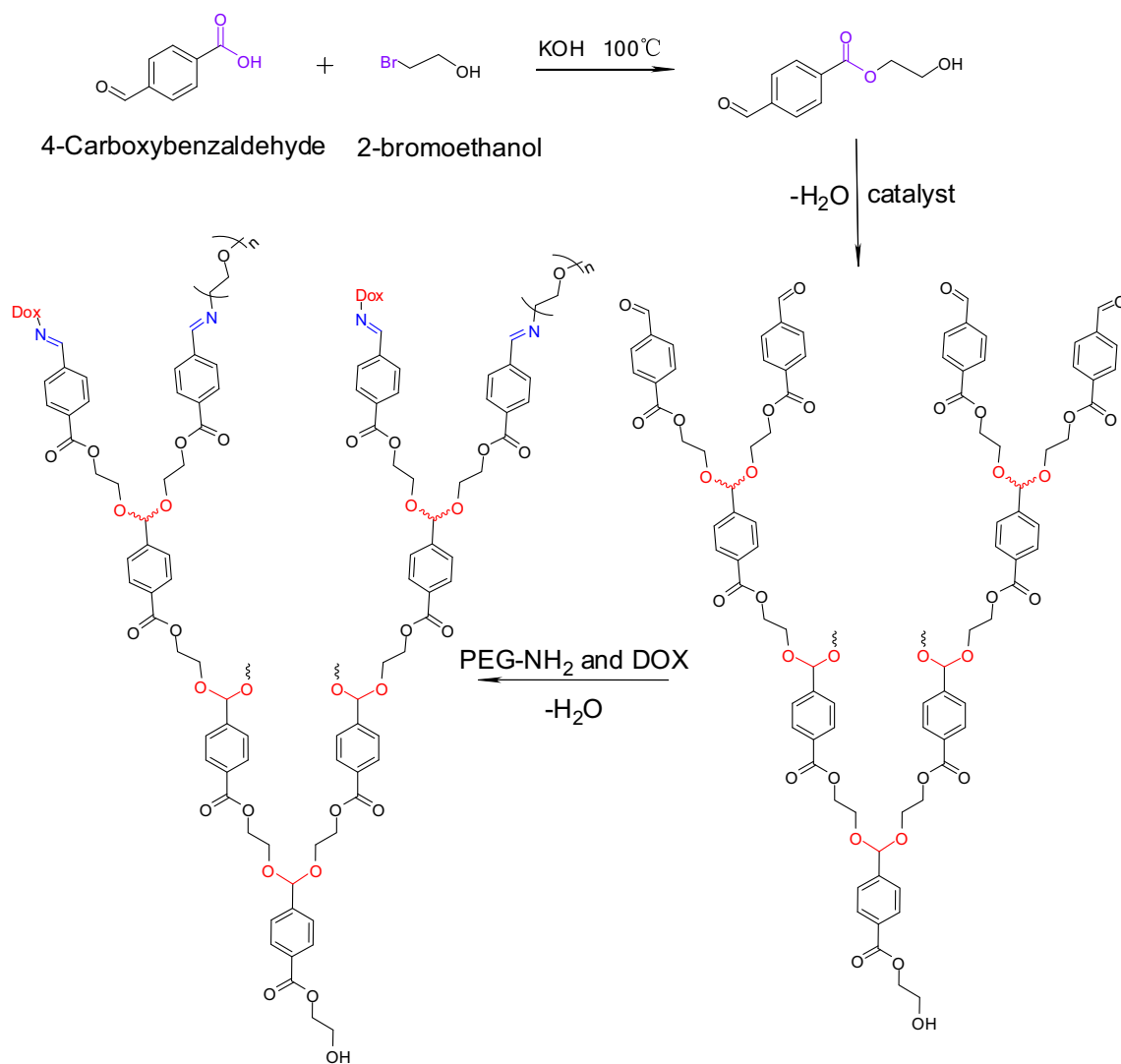
Fourier transform infrared spectroscopy (FTIR) was performed on an MIR-NIR PerkinElmer, 1605 Series spectrophotometer using a diamond attenuated total reflectance accessory (ATR). Nuclear magnetic resonance (NMR) measurement was taken on a Bruker Avance 400 spectrometer (Bruker BioSpin, Switzerland) to collect the ¹H and ¹³C spectra in DMSO-d₆. The average hydrodynamic radius of micelles was measured by using a Zetasizer ZEN 3500 dynamic light scattering (DLS) (Malvern instrument, UK). All DLS measurements were taken with an angle detection of 173° at 25 °C. All data were averaged over two measurements. All samples were filtered through

0.45- μm filters to remove dust prior to use. AUV-2450 UV-Vis spectrophotometer (Shimadzu, Japan) was used to determine the doxorubicin release rate of micelle DOX-HBPA-PEG. Size exclusion chromatography (SEC) measurement was taken on a system equipped a Waters 515 pump with a flow rate of 0.5 mL min^{-1} in THF (HPLC grade) at 25 °C. Detectors were including differential refractometer (OptilabrEX, Wyatt) and multiangle light scattering detector (MALLS) equipped with a 632.8 nm He-Ne laser (DAWN EOS, Wyatt). The refractive index increments of polymers in THF were measured at 25° Cusing an OptilabrEX differential refractometer. ASTRA software (version 5.1.3.0) was utilized for acquisition and analysis of data. Cell viability was detected by M200 Pro NanoQuant (TECAN). The cell

uptake experiment was conducted by inverted fluorescence microscope (Olympus IX73).

Synthesis of AB₂ monomer 2-hydroxyethyl-4-formylbenzoate

The synthesis route of AB₂ is described in Scheme 1. The 4-carboxybenzaldehyde (5.00 g, 33.3 mmol), KOH (1.87 g, 33.3 mmol) and 2-bromoethanol (4.16 g, 33.3 mmol) were stirring overnight in 30 mL DMF under 100 °C with reflux condensation in round-bottom flask. The DMF was concentrated by evaporation under a reduced pressure; then, 100 mL ethyl acetate was added in DMF concentrated solution. The ethyl acetate organic solution was washed three times with NaHCO₃ and then washed three times with



Scheme 1 The synthesis route of 100% hyperbranched polyacetals conjugated with PEG and DOX.

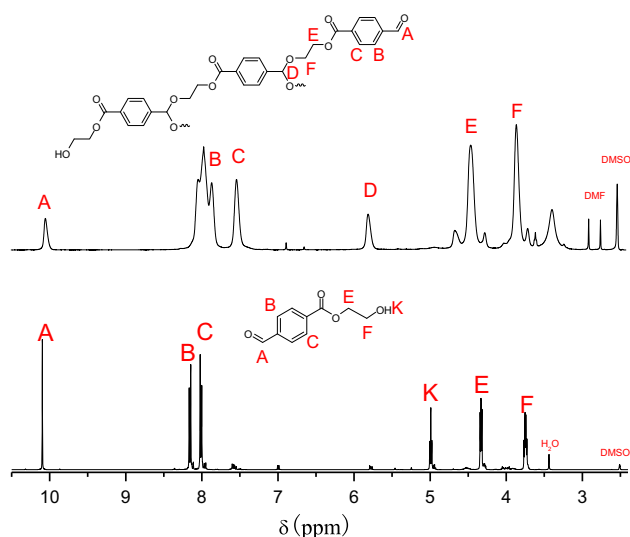


Figure 1 The ^1H NMR spectrum of hyperbranched polyacetals in DMSO-d_6 .

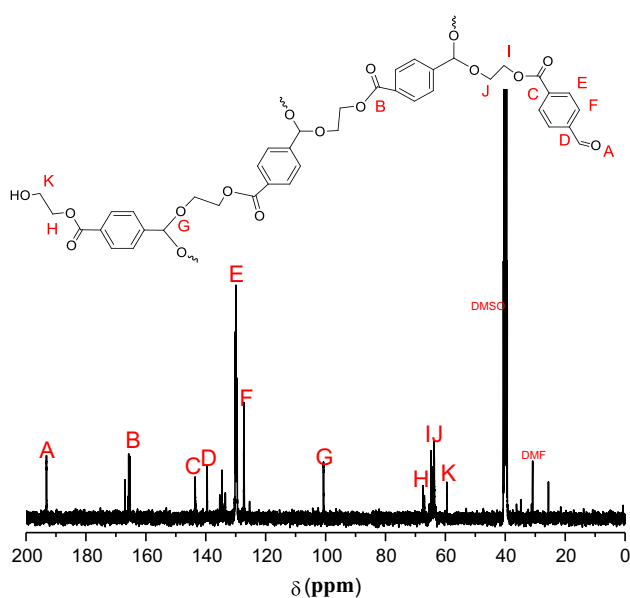


Figure 2 The ^{13}C NMR spectrum of hyperbranched polyacetals in DMSO-d_6 .

pure water. The organic solution of ethyl acetate was concentrated under reduced pressure to yield the yellowish liquid. The yellowish liquid was purified by flash chromatography (1:1 hexane/EtOAc) to supply 3.7 g AB_2 monomer as clear oil (57% yield). IR (KBr): $\nu = 3449, 2960, 2837, 1707 \text{ cm}^{-1}$. ^1H NMR (400 MHz, DMSO-d_6 , δ): 3.78 (t, 2H, $-\text{CH}_2-$), 4.42 (t, 2H, $-\text{CH}_2-$), 5.00 (s, 1H, $-\text{OH}$), 8.00 (d, 1H, $-\text{CH}$), 8.12 (d, 1H, $-\text{CH}$), 10.12 (s, 1H, $-\text{CHO}$).

Polymerization of AB_2 monomer to obtain 100% hyperbranched polyacetals

The polymerization reaction is shown in Scheme 1. AB_2 monomer 2-hydroxyethyl-4-formylbenzoate (3 g, 15.5 mmol) with 2% molpyridinium camphorsulfonate was added to Schlenk tube. The reaction was vacuum degassing three times and carried out in vacuum status under 60°C . The polymerization reaction was carrying out under vacuum pumping interval 12 h to remove the water byproduct. Five days later, polyacetals were dissolved to 2 mL THF and then precipitated twice in methanol (78% yield, white solid). IR (KBr): $\nu = 3546, 2952, 2870, 1707 \text{ cm}^{-1}$. ^1H NMR (400 MHz, DMSO-d_6 , δ): 3.86 (2H, $-\text{CH}_2-$), 4.45 (2H, $-\text{CH}_2-$), 5.80 (1H, $-\text{O}-\text{CH}-\text{O}$), 7.56 (1H, $-\text{CH}$), 8.00 (1H, $-\text{CH}$), 10.02 (1H, $-\text{CHO}$); ^{13}C NMR (400 MHz, DMSO-d_6 , δ): 63.75, 65.28 ($-\text{CH}_2-$), 100.88 ($-\text{O}-\text{CH}-\text{O}$), 127.00, 130.14 (ArCH), 165.18 ($\text{C}=\text{O}$), 192.80 ($-\text{CHO}$).

PEGylation of hyperbranched polyacetals and synthesis of DOX-HBPA-PEG

HBPA (200 mg) and mPEG-NH₂ (100 mg) were dissolved in DCM (10 mL) in 25-mL round-bottom flask with Molecular Sieves. A few drops of acetic acid dropped in flask. The reaction was carried out at room temperature for 24 h. The reaction solution was put into dialysis bag ($M_w = 3500$) and then placed in ethanol. 12 h later, put the dialysis bag into water and change pure water three times. The solution of HBPA-PEG was freeze-dried powder (60% yield). IR (KBr): $\nu = 2874, 1707, 1640$ ($-\text{CH}=\text{N}-$) cm^{-1} . ^1H NMR (400 MHz, DMSO-d_6 , δ): 3.50 (4H, $-\text{CH}_2\text{CH}_2\text{O}-$), 3.86 (2H, $-\text{CH}_2-$), 4.45 (2H, $-\text{CH}_2-$), 5.80 (1H, $-\text{O}-\text{CH}-\text{O}$), 7.56 (1H, $-\text{CH}$), 8.00 (1H, $-\text{CH}$), 8.36 (1H, $-\text{CH}=\text{N}-$), 10.02 (1H, $-\text{CH}$); ^{13}C NMR (400 MHz, DMSO-d_6 , δ): 63.75, 65.28 ($-\text{CH}_2-$), 100.88 ($-\text{O}-\text{CH}-\text{O}$), 127.00, 130.14 (ArCH), 165.18 ($\text{C}=\text{O}$), 192.80 ($-\text{CHO}$).

HBPA-PEG (130 mg) and DOX-HCl (25 mg) were dissolved in DCM (10 mL) in 25-mL round-bottom flask with Molecular Sieves. The reaction was carried out at room temperature for 24 h. The reaction solution was put into dialysis bag ($M_w = 3500$) and then placed in DMF and change three times DMF solvent. 24 h later, put the dialysis bag into water and change pure water three times. The micelle solution of DOX-HBPA-PEG was reassembled by centrifugalization; then, the supernatant was freeze-dried red powder.

Figure 3 The ^{13}C - ^1H COSY spectrum of hyperbranched polyacetals in DMSO-d_6 .

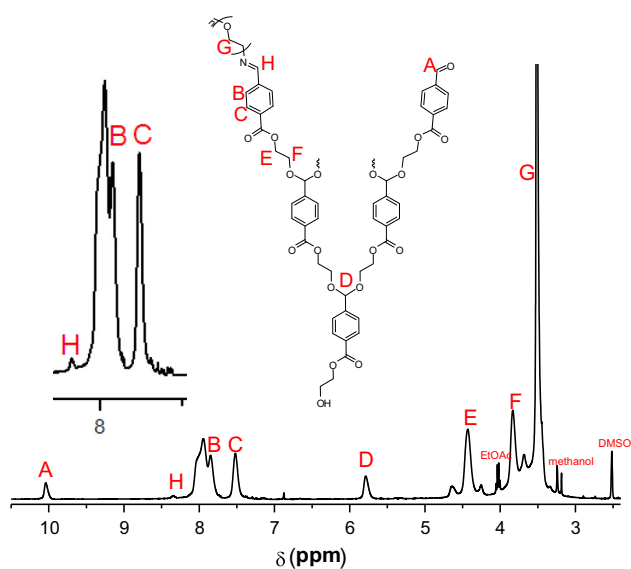
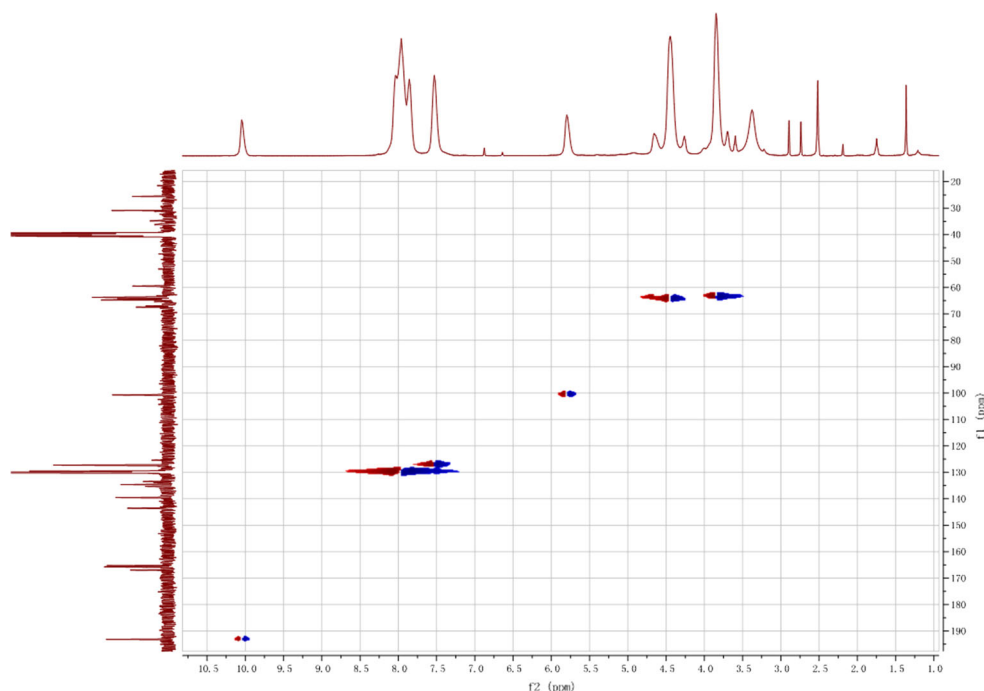


Figure 4 The ^1H NMR spectrum of HBPA-PEG in DMSO-d_6 .

Evaluation of stability, drug release of DOX-HBPA-PEG and drug loading

The freeze-dried red powder of DOX-HBAP-PEG (9 mg) was solubilized in pH 6.0 and pH 7.4 PBS, respectively. The micelle of DOX-HBAP-PEG (3 mg mL^{-1}) was centrifuged, and the supernatant was evaluated with the time passing at the condition of pH 6.0 and pH 7.4 PBS. The drug release rate of DOX from the micelle of DOX-HBAP-PEG was

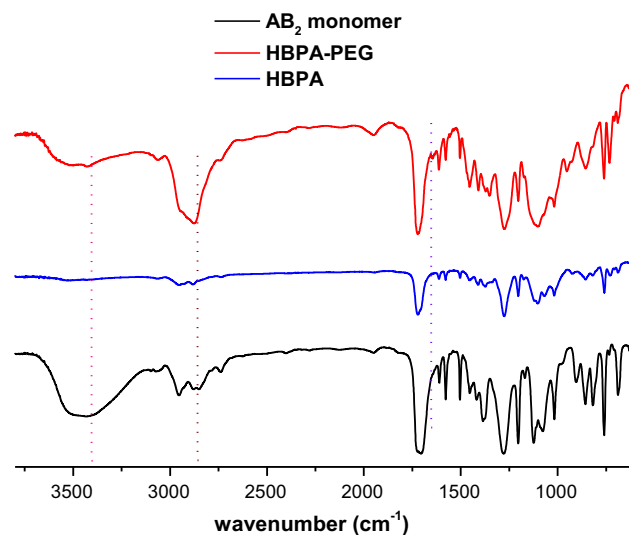
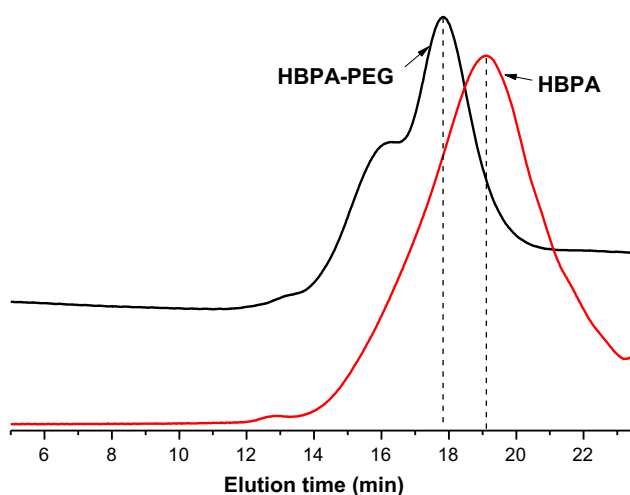


Figure 5 The FTIR spectra of HBPA-PEG, HBPA and AB_2 monomer.

measured by UV-Vis spectrophotometer at 496 nm. The four parts of 3 mL micelle solution with pH 7.4 PBS were put into dialysis bag ($M_w = 1000$); then, the dialysis bags were put into pH 6.0 and pH 7.4 PBS, respectively. 3 mL dialysis solution outside dialysis bag was taken out at every interval time; then, add 3 mL fresh PBS to dialysis solution. The eight solution samples which were taken out from solution outside dialysis bag at eight points-in-time were prepared to be measured. The experiments

Table 1 The molecular weight of HBPA and HBPA-PEG

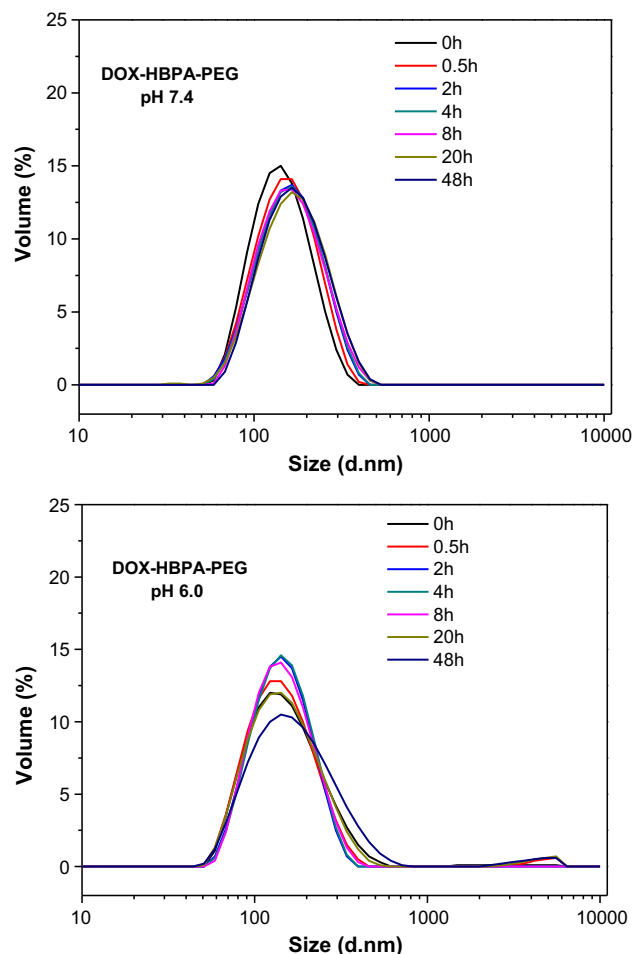
Sample	M_w (g mol ⁻¹)	M_n (g mol ⁻¹)	PDI
HBPA	9100	6300	1.46
HBPA-PEG	13900	10400	1.34

**Figure 6** The elution time (t)-dependent SEC-RI curves of polymer HBPA and HBPA-PEG with a flow rate of 0.5 mL min⁻¹ in THF.

were carried out twice in parallel. The drug loading is 12% measured by UV-Vis spectrometer at 496 nm.

Cell viability in vitro

The cell viability of DOX-HBPA-PEG and HBPA-PEG was evaluated using the Cell Counting Kit (CCK-8) assay. The concentrations of DOX in micelle of DOX-HBPA-PEG were 1, 5, 10 $\mu\text{g mL}^{-1}$, and the concentrations of HBPA-PEG were 56, 112, 280 $\mu\text{g mL}^{-1}$, respectively. The Lung A549 cells were seeded in a 96-well culture plate at a density of 10^4 cells per well and cultured in DMEM medium supplemented with 10% fetal bovine serum at 37 °C in a humidified environment of 5% CO₂ for 1 day. Thereafter, the cells were incubated with DOX-HBPA-PEG and HBPA-PEG for 24 and 48 h, respectively. 10 μL of CCK-8 solution was added to each well and incubated for further 1 h at 37 °C. The cell viability was obtained by scanning with a microplate reader at 430 nm. The relative cell viability (%) was expressed as a percentage of that of the control culture. The experiments were carried out six times in parallel. The results presented are the average data.

**Figure 7** The stability with time passing of DOX-HBPA-PEG in PBS at different pH.**Table 2** The size of DOX-HBPA-PEG measured by dynamic light scattering in PBS

Time (h)	DOX-HBPA-PEG			
	pH 6.0		pH 7.4	
	Size (d.nm)	PDI	Size (d.nm)	PDI
4	133	0.134	148	0.136
20	136	0.187	154	0.121
48	152	0.228	156	0.152

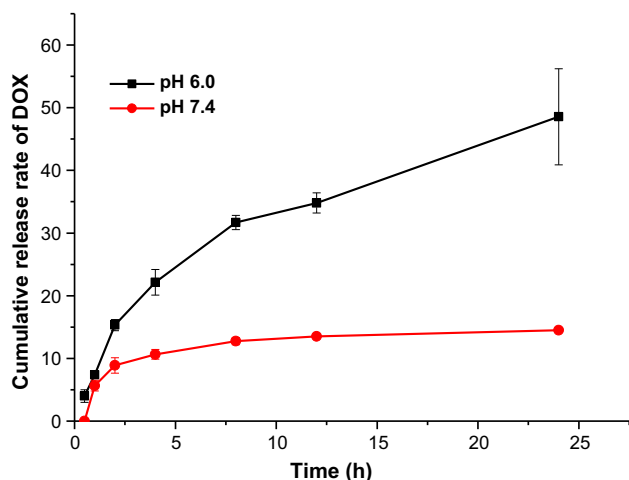


Figure 8 The cumulative release rate of DOX from the micelle of DOX-HBPA-PEG in PBS at the conditions of pH 7.4 and pH 6.0 (the experiments were carried out twice in parallel).

Cellular uptake assay

Lung A549 cells were seeded with a density of 5×10^4 per dish in 35 mm plastic microscopy dishes and incubated overnight at 37 °C. Then, the A549 cells were treated with 0.25 mg mL^{-1} DOX-HBPA-PEG micelles for 0 min, 15 min, 1 h, 4 h, respectively. Then, the cells were fixed and stained by Hoechst Staining Kit for 10 min, respectively, then gently rinsed with PBS three times and observed with fluorescence microscope.

Results and discussion

Synthesis of AB₂ monomer 2-hydroxyethyl-4-formylbenzoate and 100% hyperbranched polyacetals

The synthesis route of AB₂ and 100% hyperbranched polyacetals is described in Scheme 1. The characteristic peak of –CHO– is showed at 10.12 ppm by ¹H NMR (Fig. 1) and the characteristic peak of –OH at the wavenumber 3449 cm^{-1} by IR spectrum. The crude material of 4-carboxybenzaldehyde can be washed out by NaHCO₃ water solution; then, the product was purified by flash chromatography (1:1 hexane/EtOAc) to obtain pure AB₂ monomer for polycondensation reaction. The reaction of AB₂ monomers was catalyzed by PCS in bulk polymerization. The molar masses of polyacetalization are

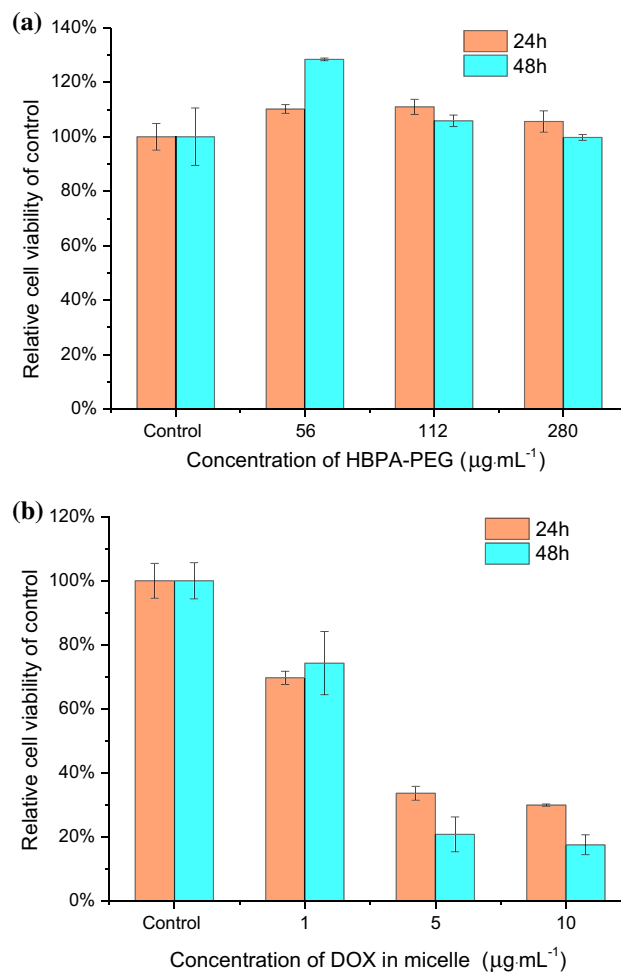


Figure 9 The cell viability of **a** HBPA-PEG (56, 112, 280 $\mu\text{g mL}^{-1}$) and **b** DOX-HBPA-PEG micelle (8.3, 41.7, 83.3 $\mu\text{g mL}^{-1}$) with different concentration at 24 and 48 h later.

increased as the removing the water byproduct. The product of polyacetals was characterized by ¹H, ¹³C NMR spectra. The polyacetals characteristic peak of OH obviously decreased compared to AB₂ monomer, meaning the acetal bond is continuously forming. The acetal bonds are particular structure for the polymer of polyacetals. The characteristic peak of –O–CHO– is 5.80 ppm in ¹H NMR, and the characteristic peak of –O–CHO– is 100.88 ppm in ¹³C NMR (Figs. 1, 2). The structure of polyacetals further confirmed by using 2D ¹³C–¹H COSY spectra. The cross dot of 100.88 ppm in ¹³C NMR and 5.80 ppm in ¹H NMR in Fig. 3 means the C–H correlation peak of acetals (–O–C–O–).

PEGylation of hyperbranched polyacetals

The hyperbranched polyacetals were characterized by NMR and FTIR. The signal of aldehyde proton (10.02 ppm) is decreased, and the signal of imine bond proton (8.36 ppm) clearly appeared, meaning the PEG was conjugated with HBPA (Fig. 4). The FTIR also demonstrated the imine bond (1640 cm^{-1}) is formed. The characteristic peak of $-\text{OH}$ in HBPA is disappeared, and the characteristic peak of imine bond in HBPA-PEG is appeared (Fig. 5). The molecular weight of HBPA-PEG and HBPA in Table 1 showed that two mPEG-NH₂ molecules at least were

conjugated with HBPA. The shoulder peak further confirms that more molecules of mPEG-NH₂ were conjugated with HBPA (Fig. 6 and Table 1).

The stability of DOX-HBPA-PEG

The amphiphilic polymer of DOX-HBPA-PEG was evaluated by DLS in pH 7.4 and pH 6.0 PBS at different time. The size of DOX-HBPA-PEG did not remarkable change over time at pH 7.4 PBS. To simulate the tumor extracellular environment, we evaluated the stability of micelle of DOX-HBPA-PEG in pH6.0 PBS. Figure 7 showed the micelle display

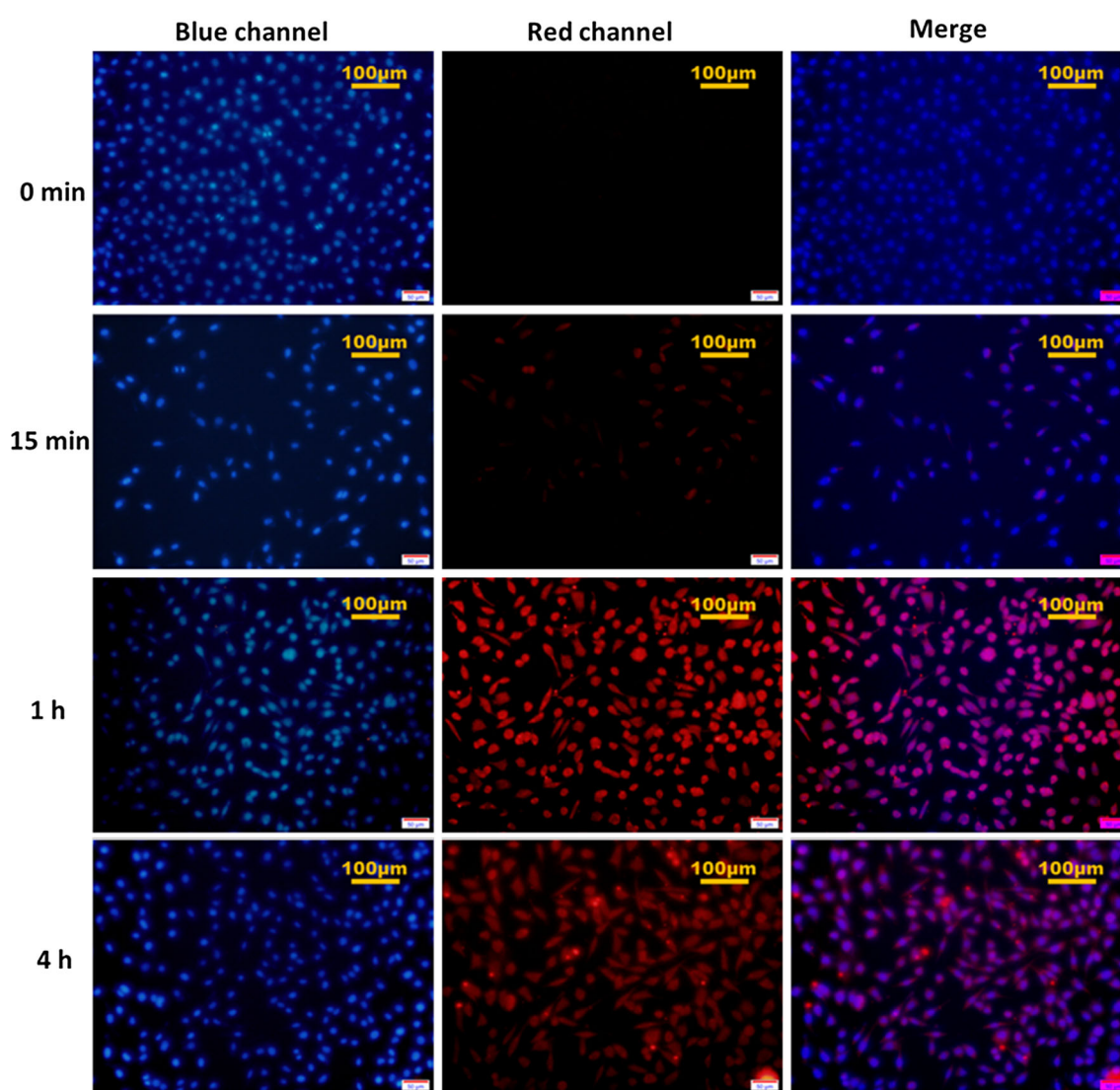


Figure 10 Investigation of DOX-HBPA-PEG endocytosis by A549 cell at different time (*blue channel*: nucleuses were dyed by Hoechst 33258, *red channel*: DOX-HBPA-PEG).

the double peak and the distribution is broader in pH 6.0 PBS. The results of DOX-HBPA-PEG measured by dynamic light scattering (Table 2) further confirmed that the micelle of DOX-HBPA-PEG was stable in pH 7.4 PBS. The data in Table 2 showed that the polydispersity index (PDI) and size of the micelle of DOX-HBPA-PEG obviously increased with the time passing in pH 6.0 PBS and the PDI and size of the micelle of DOX-HBPA-PEG did not clearly change with the time passing in pH 7.4 PBS. It means the micelle of DOX-HBPA-PEG was disrupted in acidic environment and stable in physiological environment *in vivo*.

Drug release of DOX-HBPA-PEG

The drug release property of DOX-HBPA-PEG was evaluated at the condition of pH 7.4 and pH 6.0 in PBS (Fig. 8). The micelle of DOX-HBPA-PEG was stable in pH 7.4 PBS, and the drug release rate is 14% 24 h later. The micelle of DOX-HBPA-PEG was unstable in pH 6.0 PBS, and the drug release rate is 48% 24 h later. The results of accumulated drug release rate in pH 7.4 and pH 6.0 PBS showed the drug of DOX was controllable released in acidic environment.

Cell viability of HBPA-PEG and DOX-HBPA-PEG *in vitro*

The cell viability of HBPA-PEG and DOX-HBPA-PEG was evaluated by A549 cells *in vitro* at 24 and 48 h later. The cell viability of HBPA-PEG with different concentration did not change. The results of Fig. 9a illustrated that the biocompatibility of HBPA-PEG was pretty good, even the highest concentration of HBPA-PEG is 280 $\mu\text{g mL}^{-1}$. The decreased cell viability of DOX-HBPA-PEG was typically responsive to the increased DOX concentrations in micelle (Fig. 9b). The results of cell viability in Fig. 9 illustrated the polymer of HBPA-PEG have no toxicity compared to polymer–drug conjugate of DOX-HBPA-PEG at the same concentration. It means the cell apoptosis was triggered by DOX.

Cellular uptake assay

To investigate the distribution of micelle of DOX-HBPA-PEG in A549 cells, the nucleus were dyed by Hoechst 33258 at 0 min, 15 min, 1 h, 4 h, respectively,

after being treated with 0.25 mg mL^{-1} DOX-HBPA-PEG. The micelle of DOX-HBPA-PEG was swallowed slightly by A549 cells at 15 min later in red channel and merge channel. One hour later, the micelle of DOX-HBPA-PEG totally overlapped with nucleus in merge channel. These results in Fig. 10 further confirmed that the cell apoptosis was triggered by DOX.

Conclusions

Hyperbranched polyacetals is an acidic-sensitive polymer which can completely degrade into small molecule in acidic environment. We adopted a simple method to synthesize the 100% hyperbranched polyacetals. The AB_2 monomer 2-hydroxyethyl-4-formylbenzoate was catalyzed by PCS with bulk polymerization. The polymerization products of polyacetals were modified by PEG-NH₂ to form amphiphilic polymer for drug delivery system. The synthesis route was characterized by ¹H, ¹³C NMR, ¹³C–¹H COSY spectra and FTIR spectrum. The terminal aldehyde groups of polyacetals were conjugated with DOX to form acidic-sensitive imine bonds. The micelle of DOX-HBPA-PEG was very stable in pH 7.4 PBS, and the DOX can be controllable released at pH 6.0 PBS. The results of cell viability and uptake confirm that the micelle of DOX-HBPA-PEG is an effective drug delivery system for cancer therapy. The controllable drug release nature, stability, biocompatibility and completely degradable structures (acid-sensitive) make them to be promising drug delivery systems.

Acknowledgements

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Compliance with ethical standards

Conflict of interest The authors declare that they have no conflict of interest.

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