

The effects of eutrophication and acidification on the ecophysiology of *Ulva pertusa* Kjellman

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Abstract In coastal environments, acidification and eutrophication affect the physiology of marine macroalgae. We investigated the responses of Ulva pertusa Kjellman (Ulvales, Chlorophyta) under such conditions. Samples were cultured at two different pH settings (low, 7.5; high, 8.0) and at three different ammonium levels (low, 4; medium, 60; high, 120 μ M NH₄⁺). Our objective was to analyze the influence that elevated CO₂ and NH₄⁺ might have on pH, oxygen evolution, rates of nutrient uptake, chlorophyll fluorescence, growth, and C/N ratio of that organism. Variability in pH value was enhanced under low pH/high NH4⁺ and was significantly different (p < 0.05) from changes measured when the high pH/low NH4⁺ combination was applied. Rates of NH4⁺ uptake and relative growth rates by U. pertusa were increased under low pH/high NH₄⁺ conditions and that response was significantly different (p < 0.05) from the other treatments. The rate of photosynthetic oxygen evolution and chlorophyll fluorescence were increased under elevated NH4⁺ concentrations (p < 0.05). However, the C/N ratio of U. pertusa was not affected by higher concentrations of CO_2 and NH_4^+ (p > 0.05). Our results indicated that the physiological reactions of this alga were heightened when exposed to either the elevated combination of CO_2/NH_4^+ or even when only the level of NH₄⁺ was raised. Although such excessive growth can lead to bloom formations in coastal areas, this species also has

Ik Kyo Chung ikchung@pusan.ac.kr greater capacity for taking up nutrients and dissolved inorganic carbon.

Keywords Ammonium $(NH_4^+) \cdot Carbon dioxide (CO_2) \cdot Eutrophication \cdot Ocean acidification (OA) \cdot$ *Ulva pertusa* $\cdot Chlorophyceae$

Introduction

Because continuous expansion of industrial activities since the Industrial Revolution has affected coastal environments, many coastal organisms are being challenged by anthropogenic changes such as acidification and eutrophication. The combination of these two environmental phenomena can have individual influences as well as synergistic interactions with marine species (Horta et al. 2012; Reymond et al. 2013). The effects of eutrophication have been studied in various marine organisms, but our current examination focused first on the influence of ocean acidification (OA). The Intergovernmental Panel on Climate Change (IPCC 2014) has projected that atmospheric CO₂ concentrations will increase by 280 to 400 ppm. Caldeira and Wickett (2005) also have predicted decreases in pH values of 0.3 to 0.4 units by the year 2100 and 0.7 to 0.8 units by the year 2300. Ocean acidification presents a severe challenge to the physiology and ecology of marine ecosystems in coastal areas and can have a negative effect on the metabolism of marine organisms (Kroeker et al. 2013; Cornwall and Hurd 2015; Kram et al. 2016). However, some macroalgal species can also show a positive response to acidification (Hurd et al. 2009; Suárez-Álvarez et al. 2012; Sarker et al. 2013) because of their unique carbonconcentrating mechanisms (CCMs) (Raven et al. 2011). Such macroalgae use CCMs that acquire carbon either directly from bicarbonate or via the carbonic anhydrase enzyme

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(Raven 1997; Giordano et al. 2005). Those processes could enhance algal growth because of the energy saved from dissolved inorganic carbon (DIC) transport (Beardall et al. 1998; Beardall and Giordano 2002; Wu et al. 2008).

Cultural eutrophication is another serious environmental issue that is promoted by anthropogenic activities such as urbanization, inflow of waste water, and aquaculture. Excessive input of nutrients can lead to massive algal blooms and induce bacterial respiration. This active respiration can lead to a state of OA because of elevated CO_2 levels in the water (Cai et al. 2011; Sunda and Cai 2012). Eutrophication also damages marine communities and reduces species diversity (Lohman and Priscu 1992; Valiela et al. 1997; de Faveri et al. 2015). Nevertheless, some photosynthetic macroalgae, e.g., *Ulva* spp., have a metabolic advantage under enriched nutrient conditions (Ye et al. 2011; Luo et al. 2012).

In terms of functional form and high rates of growth and nutrient uptake, *Ulva* spp. are opportunistic because of their thin sheet-like thallus (Littler 1980; Cohen and Fong 2006). Worldwide, green tides that result from excessive algal growth and blooms have a negative ecological impact because the decomposition of biomass into the water column decreases oxygen levels and affects the productivity and biodiverse of the local community (Valiela et al. 1997; Hiraoka et al. 2004; Morand and Merceron 2005; Wang et al. 2009). However, these *Ulva* spp. might be beneficial because they can function in bioremediation and as a component of Integrated Multi-Trophic Aquaculture systems (Bolton et al. 2009; Ben-Ari et al. 2014).

Numerous studies have investigated the physiological responses of *Ulva* sp. under different culture conditions that manipulate CO₂ levels, nutrient concentrations, and temperature (Figueroa et al. 2014a, b; Stengel et al. 2014). However, little research has been done using *U. pertusa* and a combination of acidification and eutrophication conditions. Here, we examined the physiological responses of that species to elevated levels of CO₂ and ammonium (NH₄⁺), as well as any possible interactions among different combinations of CO₂ and NH₄⁺ concentrations. We monitored changes in pH levels, oxygen evolution for photosynthesis, rates of NH₄⁺ uptake to show the nutrient removal capacity, fresh weight for growth, chlorophyll fluorescence to show the physiological status during the experiments, and biochemical composition to reveal any change of C/N ratio.

Materials and methods

Experiment preparation

Samples of *Ulva pertusa* were collected at Namhae, Korea (34°56'N, 127°51'E) in April of 2016. At this sampling site, the ambient temperature was approximately 14.7 ± 0.1 °C and

salinity was $28.2 \pm 0.2\%$, as measured with a YSI Pro 2030 (USA) meter. After transport to the laboratory, they were cleaned several times with filtered seawater (0.2 µm) to remove any dirty epiphytes. The samples were then maintained in a culture room in filtered seawater under conditions of 20 °C, 80 μ mol photons m⁻² s⁻¹ (12:12 light/dark), and S = 28%. They were acclimated for 3 days before the experiments began. For each treatment, samples (3 g each) were placed in 600-mL culture beakers, each containing 500 mL of filtered seawater. In all, four replicates were made for each of the treatments. The multi-factorial design comprised two pH levels (low, 7.5; high, 8.0) and three ammonium concentrations (low, 4 µM; medium, 60 µM; high, 120 µM). The lowest pH was selected based on an ocean prediction model (Caldeira and Wickett 2005) while the ambient pH reflected conditions at the sampling site. The minimum NH_4^+ level was based on the ambient concentration $(3.62 \pm 0.31 \ \mu M)$ at the sampling site, while the medium and high NH_4^+ levels had been determined in our preliminary studies (data not shown). These treatment combinations were designated as follows: L_{pH}L_A, low pH/low NH₄⁺; H_{pH}L_A, high pH/low NH₄⁺; L_{pH}M_A, low pH/medium NH₄⁺; H_{pH}M_A, high pH/medium $\rm NH_4^+;~L_{pH}H_A,~low~pH/high~NH_4^+;~and~H_{pH}H_A,~high~pH/$ high NH₄⁺. The cultures were carried out at constant temperature (20 °C) along a constant-temperature table, under LED fixtures that supplied a light intensity checked with a LI-250 light meter (LI-COR, USA). We set the intensity at 80 µmol photons $m^{-2} s^{-1}$ to prevent photochemical stress due to acidification conditions (Liu et al. 2012). Filtered seawater in the low pH culture media was injected with pure CO₂ gas from a tank. To support the low-, medium-, or high-NH₄⁺ treatments, we added ammonium to the seawater to a final concentration of 4, 60, or 120 µM NH₄Cl. The medium in each treatment was replaced every 2 days and nutrients were added daily to avoid any deficiencies. Total alkalinity (TA) of the seawater was determined according to the electrotitration method (Gran 1952), at a precision of $\pm 4 \ \mu mol \ kg^{-1}$. The inorganic carbon concentration and pCO_2 were calculated based on pH, TA, salinity, and temperature, using the CO2SYS software program (Lewis and Wallace 1998). Dissociation constants and KSO₄ values were defined as described by Millero et al. (2006) and Dickson (1990) (Table 1).

Monitoring changes in pH, photosynthetic oxygen evolution, and rates of ammonium uptake

During the first 8 h of treatment, we measured changes in pH of the seawater medium and calculated the rates of oxygen evolution and NH_4^+ uptake. An Orion-250A pH meter (Thermoscientific, USA) was used to record pH levels at 2-h intervals for each treatment. Those values can serve as an indicator of physiological characteristics of algae under culture conditions (Maberly 1990; Murru and Sandgren 2004).

Table 1 Parameters of seawater carbonate system under different culture conditions

Culture conditions	TA	pCO ₂	CO ₂	CO3 ²⁻	HCO ₃ ⁻	DIC
L _{pH} L _A	2012.50 ± 9.57	1404.43 ± 10.61	45.61 ± 0.22	53.48 ± 0.26	1877.81 ± 9.06	1976.90 ± 9.54
L _{pH} M _A	2015.30 ± 7.63	1405.34 ± 11.16	45.64 ± 0.04	53.52 ± 0.40	1879.03 ± 7.54	1978.18 ± 7.63
L _{pH} H _A	2015.50 ± 5.57	1405.70 ± 11.35	45.65 ± 0.04	53.53 ± 0.50	1879.50 ± 7.81	1978.68 ± 7.91
H _{pH} L _A	2019.25 ± 8.30	389.18 ± 10.49	12.64 ± 0.05	148.20 ± 0.63	1645.52 ± 7.03	1806.36 ± 7.72
H _{pH} M _A	2022.05 ± 7.58	389.47 ± 10.52	12.65 ± 0.02	148.31 ± 0.20	1646.72 ± 7.19	1807.68 ± 7.40
H _{pH} H _A	2022.50 ± 8.29	389.57 ± 10.26	12.65 ± 0.01	148.35 ± 0.10	1647.15 ± 7.09	1808.15 ± 7.20

Values are means \pm SD. Both pH and TA (µmol kg⁻¹) were measured directly for each scenario, while *p*CO₂ (µatm), CO₂ (µmol kg⁻¹), CO₃²⁻ (µmol kg⁻¹), HCO₃⁻¹ (µmol kg⁻¹), and DIC (µmol kg⁻¹) were calculated using the CO2SYS program (Lewis and Wallace 1998)

Dark respiration by $L_{pH}H_A$ and $H_{pH}L_A$ samples were measured because we speculated that OA is mediated by photosynthesis. Changes in pH were also investigated in beakers containing culture media without specimens (blanks), under both low- and high-pH conditions.

Photosynthetic oxygen evolution rates (μ mol O₂ g⁻¹ FW h⁻¹) were also analyzed with a Clark-type microelectrode oxygen sensor (Unisense, Denmark). The microelectrode was calibrated with a solution of C₆H₇NaO₆ (sodium ascorbate) and NaOH (sodium hydroxide) that detected response times of less than 1 s.

The rate of NH_4^+ uptake (*V*; μ mol NH_4^+ g⁻¹ FW h⁻¹) was determined based on the average amount that was removed from the culture medium during the incubation period. The measurement method of NH_4^+ uptake rates were described by Parsons et al. (1984). The following equation was used in this calculation:

$$V = (S_i - S_f) \times \text{vol} \times W^{-1} \times T^{-1}$$

where S_i is the initial concentration of NH₄⁺, S_f is the final concentration after *T* hours of culture, vol is the volume of the culture medium, and *W* is the fresh weight of each algal specimen.

Growth rates, chlorophyll fluorescence, and C/N ratio

Algal growth rates, chlorophyll fluorescence, and C/N ratio were measured after 14 days of incubation. Changes in biomass were determined at the end of the experimental period, and the relative growth rate (RGR; $\% \text{ day}^{-1}$) was calculated as follows (Yong et al. 2013):

 $RGR = (\ln W_2 - \ln W_1) \times 100 \times T^{-1}$

where W_1 is the initial fresh weight and W_2 is the final fresh weight after *T* days of incubation.

Chlorophyll fluorescence was checked with a pulse amplitude modulation fluorometer (DIVING-PAM, Walz, Germany). The maximum quantum yield of Photosystem II was calculated as (Cosgrove and Borowitzka 2011):

$$F_{\rm v}/F_{\rm m} = (F_{\rm m} - F_{\rm o})/F_{\rm m}$$

where F_v is variable fluorescence, F_m is maximum fluorescence after dark-adaptation, F_o is minimum fluorescence after dark-adaptation, and F_v/F_m is photosynthetic efficiency, as measured using a saturating pulse under dark-adaptation. Samples were kept in the dark for 15 min before chlorophyll fluorescence measurement.

Tissue carbon (C) and nitrogen (N) in the alga were analyzed after the experiment. Samples were dried at 60 °C for 48 h and then ground to powder with a Tissuelyser LT (Qiagen, Germany). Approximately 2 to 3 mg of ground tissues was used with an elemental analyzer (Flash 2000 Series; Thermo Fisher Scientific, USA). The C/N ratio was calculated on a molar basis.

Statistical analysis

Two-way analysis of variance (ANOVA) was conducted with all experimental data. The distributions of values obtained for pH, rate of photosynthetic oxygen evolution, NH_4^+ uptake rate, relative growth rate, chlorophyll fluorescence, and C/N ratio were tested for normality and homogeneity before the statistical analysis began. Tukey's tests were used to compare among treatments. Differences were considered statistically significant at *p* <0.05. All analyses were performed with the SPSS Program (version 23.0).

Results

Change in pH, photosynthetic oxygen evolution, and ammonium uptake rates

Initial pH values in the culture beakers were recorded before the experiments began. After 8 h of treatment, the maximum increase in pH was 2.76 ± 0.07 units, which occurred under the L_{pH}H_A treatment (Fig. 1), while the smallest rise, 2.12 ± 0.17 units, resulted from the $H_{pH}L_A$ combination. For pH, responses to treatment, from highest to lowest, followed the order of $L_{pH}H_A > L_{pH}M_A > L_{pH}L_A > H_{pH}H_A > H_{pH}M_A > H_{pH}L_A$. When the same NH4⁺ concentration was examined across treatments, ΔpH in low-pH treatment beakers was significantly higher than that in the high-pH beakers (Table 2). At each pH level, the change in pH under high NH₄⁺ was greater, but not significantly, than levels calculated under low or medium NH₄⁺ conditions (Table 2). After 8 h of treatment, pH values were not significantly different among culture conditions but all were above 10.00 units (Fig. 1, Table 3). The pH did not change over time in cultures that had no specimen (i.e., blank treatment) and values ranged from 7.50 to 7.51 under low pH conditions and from 8.00 to 8.01 for high pH conditions.

Photosynthetic oxygen evolution rates during the 8-h experiments ranged from 42.39 ± 2.49 to $69.75 \pm 4.25 \ \mu mol O_2$ g⁻¹ FW h⁻¹ (Fig. 2). When pH remained constant, photosynthetic oxygen evolution rates increased as the NH₄⁺ concentration rose, and differences among treatments were significant. However, at individual NH₄⁺ levels, evolution rates were not significantly different (Table 2).

Rates of NH_4^+ uptake by *Ulva pertusa* ranged from 0.08 ± 0.01 to $3.33 \pm 0.10 \mu mol NH_4^+ g^{-1}$ FW h⁻¹ (Fig. 3).



Fig. 1 Maximum values and extent of variation in pH in response to different treatment combinations over 8 h. Data are means \pm SD (n = 4)

Table 2 Results of two-way ANOVA derived from physiologicalactivities (extent of variation in pH, photosynthetic oxygen evolutionrate, ammonium uptake rate, relative growth rate, photosyntheticefficiency, and C/N ratio) of Ulva pertusa

Source	df	MS	F value	p value
Extent of variation in pH				
pH level	1	1.62	226.29	< 0.05
Ammonium concentration	2	< 0.05	3.50	0.05
pH level × ammonium concentration	2	< 0.05	0.59	0.57
Photosynthetic oxygen evolution rate (µ	μ mol O ₂ g ⁻¹ FW h ⁻¹)			
pH level	1	134.24	4.10	0.05
Ammonium concentration	2	891.71	27.23	< 0.05
pH level × ammonium concentration	2	40.66	1.24	0.31
Ammonium uptake rate (μ mol NH ₄ ⁺ g ⁻¹ FW h ⁻¹)				
pH level	1	0.13	58.61	< 0.05
Ammonium concentration	2	18.81	8389.01	< 0.05
pH level × ammonium concentration	2	0.12	51.49	< 0.05
Relative growth rate (% day ⁻¹)				
pH level	1	1.04	36.15	< 0.05
Ammonium concentration	2	4.37	151.74	< 0.05
pH level × ammonium concentration	2	0.29	10.02	< 0.05
Photosynthetic efficiency (F_v/F_m)				
pH level	1	< 0.05	0.04	0.84
Ammonium concentration	2	< 0.05	7.48	< 0.05
pH level \times ammonium concentration	2	< 0.05	1.31	0.30
C/N ratio (%)				
pH level	1	< 0.05	0.21	0.65
Ammonium concentration	2	0.37	2.27	0.21
pH level \times ammonium concentration	2	< 0.05	0.15	0.87

MS means mean square

Values were highest for the $L_{pH}H_A$ samples and differences were significant among treatment combinations (p < 0.05). The lowest rate was $0.08 \pm 0.01 \mu mol NH_4^+ g^{-1} FW h^{-1}$ for $H_{pH}L_A$ samples and was similar to that calculated for the $L_{pH}L_A$ combination (p > 0.05). When rates were compared among all treatments, values were higher under low pH conditions than under high pH treatment for all samples exposed to the H_A status (p < 0.05). However, at the medium NH₄⁺ concentration, differences in responses were not significant (p > 0.05). When NH₄⁺ levels were elevated, uptake rates were increased at each pH value (Table 2). In particular, the rate was approximately 41 times faster for $L_{pH}H_A$ than for $H_{pH}L_A$.

Growth rates, chlorophyll fluorescence, and C/N ratio

During the 14 days of observation, relative growth rates for *U. pertusa* ranged from 2.23 \pm 0.13 to 4.27 \pm 0.09% day⁻¹ (Fig. 4), with values being lowest under H_{pH}L_A conditions. The highest rate was associated with the L_{pH}H_A treatment and was significantly different from rates calculated for the other

Table 3Maximum values andextent of variation in pH inresponse to different culturetreatment combinations over 8 h

	Maximum pH		Extent of variation	n
	L _{pH} (low pH)	H _{pH} (high pH)	L _{pH} (low pH)	H _{pH} (high pH)
$L_A (low NH_4^+)$	$10.11\pm0.01a$	$10.12\pm0.17a$	$2.61\pm0.01a$	$2.12\pm0.17b$
M _A (medium NH ₄ ⁺)	$10.14\pm0.06a$	$10.14\pm0.06a$	$2.64\pm0.06a$	$2.14\pm0.06b$
H_A (high NH_4^+)	$10.26\pm0.07a$	$10.19\pm0.01a$	$2.76\pm0.07a$	$2.19\pm0.01b$

Data are means \pm SD (n = 4). Within a column, values not followed by same letter are significantly different at p < 0.05

treatments (p < 0.05). At a high NH₄⁺ level, growth was more rapid under low pH conditions than under high pH conditions (p < 0.05). By contrast, low and medium levels of NH₄⁺ did not have a significant effect on growth rates (p > 0.05). At each pH level, relative growth rates increased as the NH₄⁺ concentration rose (Table 2).

Photosynthetic efficiencies, as measured by chlorophyll fluorescence, ranged from 0.54 \pm 0.04 (H_{pH}L_A) to 0.67 \pm 0.06 (L_{pH}H_A) after 14 days (Fig. 5). Significant differences were detected between the L_{pH}L_A and L_{pH}H_A conditions (Fig. 5, Table 2). Although F_v/F_m values were greater under high NH₄⁺ than under other conditions, they were similar when the low NH₄⁺ and medium NH₄⁺ treatments were compared.

Neither the pH conditions generated by the addition of CO₂ nor NH₄⁺ concentrations affected the C/N ratios of *U. pertusa* (Table 2). Levels of C and N were 25.65 \pm 0.65 to 27.26 \pm 0.73% and 3.09 \pm 0.04 to 3.15 \pm 0.06%, respectively, and did not differ significantly among treatments (p > 0.05). The C/N ratio in tissues ranged from 8.24 \pm 0.15 to 8.74 \pm 0.22, but did not differ significantly among culture conditions (Table 4).

H_{pHLA} L_{pHLA} H_{pHMA} L_{pHMA} H_{pHHA} L_{pHHA} Culture conditions

Fig. 2 Photosynthetic oxygen evolution rates (μ mol O₂ g⁻¹ FW h⁻¹) in response to different media culture conditions. Significant differences among conditions are indicated by different *letters* (p < 0.05). Data are means \pm SD (n = 4)

Discussion

The physiological responses of Ulva pertusa were influenced by pH and NH₄⁺ status. This was demonstrated by the changes in pH, oxygen evolution, ammonium uptake, algal growth, chlorophyll fluorescence, and C/N ratio. The addition of CO_2 gas to the culture medium, which was used to mimic OA, altered pH values because of chemical reactions in the solution. When that gas was dissolved, carbonic acid formed and then partitioned into bicarbonate and hydrogen ions (Falkowski and Raven 2007). Through this process, the dissolved hydrogen ions caused pH to decrease while the DIC concentration increased in that solution. We noted that changes in pH values were greater under low pH conditions than under the ambient pH treatment due to photosynthetic activity. This was a result of greater DIC available for photosynthesis, which enhances pH levels (Zhang et al. 2012). In response to higher CO₂, many species down-regulate their CCM activity to save energy, thereby improving their rates of growth (Sarker et al. 2013).

Although our data showed that pH values under all of our culture conditions rose to almost 10.20 after 8 h, those differences were not significant. Zou (2014) has demonstrated that the pH compensation point of *Ulva prolifera* remains near



Fig. 3 Ammonium uptake rates (μ mol NH₄⁺ g⁻¹ FW h⁻¹) by *Ulva* pertusa under different culture conditions. Significant differences among treatments are indicated by different *letters* (p < 0.05). Data are means \pm SD (n = 4)



Fig. 4 Relative growth rates of *Ulva pertusa* under different CO₂ and ammonium levels. Significant differences among culture conditions are indicated by different *letters* (p < 0.05). Data are means \pm SD (n = 4)

pH 10.40 under experimental conditions and pH is not affected by different culturing treatments because U. prolifera has a similar extract HCO_3^{-} capacity under a range of conditions. Maberly (1990) has reported that photosynthesis is limited at pH >9.0, which is contrary to our results. Under natural conditions, *Ulva* spp. utilize two means for taking up HCO_3^{-1} when pH levels are high. The first mechanism involves external carbonic anhydrase, which converts, through dehydration, HCO_3^{-1} to CO_2 that is then transported into the cells (Beer and Israel 1990; Axelsson et al. 1995; Larsson et al. 1997; Björk et al. 2004). The second possible mechanism is direct uptake of HCO₃⁻ via anion exchange protein under experimental conditions (Axelsson et al. 1995; Larsson et al. 1997; Björk et al. 2004). We suspect that U. pertusa uses both mechanisms for taking up HCO_3^{-} under either low- or high-pH conditions in the laboratory (i.e., beaker cultivation).

We found that photosynthetic oxygen evolution rates for *U. pertusa* were influenced by elevated nutrient concentrations. Those rates improved under both low and high pH when NH_4^+ levels increased. At higher NH_4^+ , the difference in relative rates was significant between low NH_4^+ and medium NH_4^+ conditions but not between the $L_{pH}H_A$ and $H_{pH}H_A$



Fig. 5 Photosynthetic efficiency (F_v/F_m) of *Ulva pertusa* under different culture combinations. Significant differences among treatments are indicated by different *letters* (p < 0.05). Data are means \pm SD (n = 4)

treatments. These results demonstrated that oxygen evolution by *U. pertusa* is more strongly modulated by a rise in NH_4^+ concentrations than by pH levels.

The greater rates of NH_4^+ uptake by U. pertusa in response to elevated CO₂ and NH₄⁺ levels or NH₄⁺ alone noted here are similar to those reported for Porphyra leucosticte, Ulva rigida, Sargassum fusiforme, Gracilaria lemaneiformis, U. prolifera, Ulva linza, and Pyropia haitanensis (Mercado et al. 1999; Gordillo et al. 2001; Zou 2005; Xu et al. 2010; Luo et al. 2012; Xu and Gao 2012; Chen et al. 2016). For example, higher N concentrations lead to faster uptake by Ulva spp. (Luo et al. 2012). Elevated CO₂ levels also enhance the uptake of N (Gordillo et al. 2001; Zou 2005; Liu and Zou 2015). When more nutrients are available, Ulva spp. can take up N more efficiently (Björnsäter and Wheeler 1990; Runcie et al. 2003; Pérez-Mayorga et al. 2011). The distinct morphological characteristics of *Ulva* spp. facilitate the opportunistic uptake of nutrients, which can then increase growth rates when NH₄⁺ concentrations are higher (Duke et al. 1987; Arévalo et al. 2007). We also found that uptake rates were greatest in response to the interactive effect of L_{pH}H_A treatment. Therefore, future investigations should focus on aspects of

 Table 4
 Tissue C, N (% in DW),

 and C/N molar ratio of Ulva
 pertusa

 pertusa
 under different culture

 conditions
 conditions

	C (%)	N (%)	C/N molar ratio
L _{pH} L _A (low pH/low NH ₄ ⁺)	$27.26 \pm 0.73a$	3.15 ± 0.06a	$8.65 \pm 0.33a$
H _{pH} L _A (high pH/low NH ₄ ⁺)	$26.98 \pm 1.02a$	$3.09\pm0.04a$	$8.74\pm0.22a$
L _{pH} M _A (low pH/medium NH ₄ ⁺)	$26.44\pm0.59a$	$3.15\pm0.04a$	$8.40\pm0.21a$
H _{pH} M _A (high pH/medium NH ₄ ⁺)	$26.23\pm0.98a$	$3.13\pm0.05a$	$8.37\pm0.29a$
$L_{pH}H_A$ (low pH/high NH_4^+)	$25.65\pm0.65a$	$3.11\pm0.04a$	$8.24\pm0.15a$
$H_{pH}H_A$ (high pH/high NH_4^+)	$25.87\pm0.78a$	$3.11\pm0.02a$	$8.32\pm0.20a$

Data are means \pm SD (n = 4). Values not followed by same letter are significantly different at p < 0.05

metabolism, e.g., algal growth and nutrient uptake, in combination with CO₂, NH₄⁺, and other environmental factors such as light intensity and temperature (Suárez-Álvarez et al. 2012). Our results indicated that relative growth rates of U. pertusa were faster when either nutrient levels alone or in combination with CO2 were increased. Both factors enhance seaweed growth and biomass accumulations in U. prolifera, U. linza, Hypnea spinella, Chondrus crispus, and P. haitanensis (Luo et al. 2012; Suárez-Álvarez et al. 2012; Xu and Gao 2012; Sarker et al. 2013; Chen et al. 2016). In our experiments, algal growth was maximized under L_{pH}H_A conditions and performance was significantly different when compared with results from the other treatments. Young and Gobler (2016) have also reported that the growth of Ulva spp. is promoted by elevated CO₂ and nutrient concentrations. Because U. pertusa utilizes both CO2 and NH4⁺ for photosynthesis, that process can possibly be improved under OA and eutrophication conditions. In fact, such a scenario has been described with G. lemaneiformis and P. haitanensis (Xu et al. 2010; Chen et al. 2016). Therefore, we conclude that the interaction between CO₂ and nutrient levels can have a dramatic impact on the growth of U. pertusa.

In addition to some changes in pH values, ammonium uptake rates, and relative growth rates, the photosynthetic efficiency (F_v/F_m) of *U. pertusa* was significantly affected by NH₄⁺ concentrations. A previous study has found that photosynthetic efficiency is improved when nitrogen is more readily available (Zou and Gao 2014). Dawes and Koch (1990) have also shown that increased N concentrations can enhance chlorophyll fluorescence in algae. This is in contrast to a previous report that chlorophyll fluorescence is greater under higher CO₂ concentrations (Chen et al. 2015). However, Liu et al. (2012) have found that elevated CO₂ induces photochemical stress in *U. prolifera* under high-light conditions. Our results indicated that *U. pertusa* was not subjected to such stress because the experiments involved less intense lighting.

Regardless of treatment, the C/N ratio of U. pertusa was not significantly affected by elevated CO_2 or NH₄⁺ concentrations. Similar findings have been described for *Porphyra leucosticta* and *Hypnea spinella* (Mercado et al. 1999; Suárez-Álvarez et al. 2012). However, we calculated a C/N ratio of approximately 8:1, which is lower than the ratio of 10:1 that generally occurs when nitrogen supplies are abundant (Gómez-Pinchetti et al. 1998; Suárez-Álvarez et al. 2012). Chen and Johns (1991) have suggested that the C/N ratio is a good tool for assessing the physiological status of macroalgae. In fact, under our culture conditions, samples were able to maintain a good status throughout the experimental period.

Although *U. pertusa* is currently regarded as a harmful species in several coastal areas, our findings indicated that it can also provide a mitigating solution to problems associated with acidification and eutrophication because of its capacity to

take up nutrients and DIC. Nevertheless, this study focused on short-term responses by *U. pertusa* and was conducted only on a small experimental scale. Therefore, further investigations should include long-term and large-scale examinations, such as mesocosm and field studies, so that researchers can obtain more practical information as they develop new strategies to deal with OA and eutrophication.

In summary, our data demonstrated that the physiological activities of *U. pertusa* are positively affected by OA and eutrophication. Increases in either CO_2 levels or NH_4^+ concentrations had a strong impact on variations in pH, photosynthetic oxygen evolution rates, and chlorophyll fluorescence. In particular, the combination of elevated CO_2 and NH_4^+ had the greatest influence on NH_4^+ uptake and algal growth. By comparison, rises in levels of CO_2 and NH_4^+ did not affect the C/N ratio of this species.

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