PRIMARY RESEARCH PAPER

Phosphatase activities of the aquatic moss Warnstorfia fluitans (Hedw.) Loeske from an acidic stream in North-East England

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Abstract A study was made of the aquatic environment, tissue nutrient composition and surface phosphatase activities of the aquatic moss Warnstorfia fluitans in Brandon Pithouse Stream, a small acidic stream in N-E England. The water, which originates from an underground spring, had been pH 2.6 for at least 30 years, but about 3.9 during the present study. The moss was by far the most abundant phototroph during all this period. Seasonal changes in aqueous nitrogen and phosphorus fractions were measured over a 2-year period near the source. Most of the filtrable N and P were at times organic, but the very high N:P ratio (even if organic N is excluded) suggests that only organic phosphate is likely to be important

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for the moss. There was a high peak in organic phosphate in late spring in both study years. Surface phosphomonoesterase (PMEase) and phosphodiesterase (PDEase) activities were highly correlated in the field and in axenic culture, though there were some differences in response to environmental factors. Axenic material showed higher PMEase and PDEase activities when grown with organic P than with inorganic P. Although the data suggest that internal P content is an important factor influencing phosphatase activities, PDEase activity was especially marked when the moss was grown with the diester, DNA, as P source, indicating that at least one of its surface phosphatases can also respond directly to the environment.

Keywords Moss Warnstorfia Phosphatase Phosphomonoesterase · Phosphodiesterase · Acidic stream

Abbreviations

Introduction

A number of moss species have been shown to form surface phosphatase enzymes in aquatic environments (Turner et al., 2001, 2003). The internal phosphate content appears to be a key factor influencing phosphatase development of these mosses (Press & Lee, 1983; Whitton et al., 2005). In the two widespread northern hemisphere species whose surface phosphatase activities have received the most study, Rhynchostegium riparioides (Christmas & Whitton, 1998a) and Fontinalis antipyretica (Christmas & Whtton, 1998b), there is a low level of surface PMEase (phosphomonoesterase) activity when the moss is phosphate-rich, but this increases markedly with increasing phosphate-limitation (Christmas & Whitton, 1998a). Surface PDEase (phosphodiesterase) activity is apparently absent under phosphate-rich conditions and is induced later than PMEase activity; hence the greater the phosphate-limitation, the greater the relative importance of PDEase. However, not all aquatic phototrophs respond in the same way. For instance, in some cyanobacteria there is no PMEase activity under phosphate-rich conditions and PDEase activity is induced at the same time as PMEase activity (Whitton et al., 2005).

The concentrations of nutrient elements and phosphatase activities of mosses can provide a record of the nutritional status of the environment (Steinman & Boston, 1993; Steinman, 1994; Garcia-Alvaro et al., 2000; Whitton et al., 2002). As the aquatic mosses whose phosphatase activities have been reported in detail mostly come from streams and small rivers where peat is important in the catchment, but with the pH of the water typically near neutral, it was decided to investigate the surface phosphatase activities of a moss from a quite different environment, an acidic stream. Acid mine drainage water is low in soluble (inorganic) phosphate because of the precipitation by aluminium (Gross, 2001) and further removal of soluble inorganic phosphate occurs when the water rises above about pH 3.0 in waters rich in iron, leading to precipitation of iron "oxides". A further problem in some drainage waters is the presence of arsenate at concentrations likely to interfere with inorganic phosphate uptake. The availability of, and ability to use, organic phosphates is therefore likely to be especially important in highly acidic waters derived from mine wastes, though not in some volcanic waters, where inorganic phosphate may be quite high (Brock, 1978). Warnstorfia fluitans (Hedw.) Loeske was chosen for the study reported here, because observations have been made on its occurrence in an acidic stream (Brandon Pithouse Stream) over a long period (Hargreaves et al., 1975; Gimeno et al., 1998).

A preliminary study at this site in 1999 (Ellwood et al., 2002) showed that PMEase and PDEase activities were low from January to July, but high for the rest of the year, when the tissue N:P ratio (by mass) exceeded 10. The aim of the project was to determine the factors influencing the PMEase and PDEase activity of a moss from a moss from a highly acidic stream. Sufficient data were obtained to permit comparison with R. riparioides and F. antipyretica, including seasonal changes. Axenic material was included in the study in order to ensure that key results were not an artefact due to the presence of periphytic bacteria or fungi.

Sample site

Brandon Pithouse Stream (NZ 222405, 210 m a.s.l.) is 8 km south-west of Durham City, N-E England. (The name of the stream relates to a deep coal mine several hundred meters from the site during the first half of the 20th century, but this probably had no influence on the stream.) The stream rises from a piped drain in a field and forms part of the catchment of Red Burn, a sec-

ond-order tributary of the River Wear. The pH of the water at source was 2.6 and remained at this value for several hundred metres when first measured by B.A.W. in the late 1960s (Hargreaves et al., 1975). (Records for W. fluitans were listed under its synonym, Drepanocladus fluitans.) Physiological studies were reported on several algae from this site (Hargreaves & Whitton, 1976), but not on phosphate. At this period the first few metres of the stream were dominated by large growths protonema, almost certainly the protonema of W. fluitans, though further downstream this moss occurred as leafy shoots (B.A. Whitton, unpublished data). Towards the end of the 1990s the pH of the water started to rise due to activities of an opencast mining company in the catchment; it was about pH 3.5 in 1999 (Ellwood et al., 2002). Although the pH had risen further by 2002, Warnstorfia remained the most abundant phototroph in this stream ever since it was first studied.

Soils in the catchment are a mixture of Cambic Stagno Gleys and disturbed soils. Until largely removed by the mining company, the catchment contained seams of a high-grade coal with low sulphur content. It seems probable that the acidity of the stream results from oxidation of iron pyrite rather than sulphur in the coal.

Materials and methods

Field

A range of physical and chemical variables was recorded on each sampling, which was conducted monthly from March to December 1999 and then twice a month until February 2001. Although sampling was 2 months out of phase with the calendar year, for simplicity the data are reported as 1999 or 2000, unless there is a need to highlight information for the months which do not fall within that calendar year. At the source the entire flow of the spring is piped, which allows measurement by conversion of the time taken to fill a 5-dm³ plastic measuring cylinder $(m^{-3} s^{-1})$. Temperature and conductivity were measured with a WTW (Wissenschaftliche Technische Werkstätten, Weilheim, Germany) model LF 320 meter with a TetraCon 325 probe. Calibration with BDH (BDH Chemicals Ltd, Poole, UK) standard solutions (36 and 1,413 μ S cm⁻¹) was carried out monthly. Dissolved oxygen was measured with a WTW model OXI 196 fitted with an EOT 196 probe. Calibration was done before each set of measurements using a WTW Oxical-S sleeve, with checks to ensure no membrane damage or air bubbles. The pH was measured in the field with a WTW meter (model pH 196 with a WTW SenTix electrode type 41–3). Calibration was performed prior to measurements using the standard BDH pH buffers: 4.0, 7.0, and 9.22. Values for mean pH were determined after antilogarithmic conversion.

Plastics and glassware were soaked in 2% (v/v) Decon[®] (Decon Laboratories Ltd., Hove, UK) solution for at least 24 h and rinsed four times in deionised water. All water samples for analysis of dissolved nutrients were passed at the site through $0.45 \mu m$ cellulose-acetate membranes, (previously soaked in 10% HCl and rinsed in deionised water) held in a Millipore Swinnex filter housing. All water samples for total nutrient analysis were collected in bottles primed with stream water. All moss shoots were taken from submerged plants in positions with obvious flow. The shoots were washed by repeat rinses in a plastic bag with water from the stream, then drained of excess water and placed in a sterile plastic bag, stored in a cool box with ice and transported to the laboratory.

Analysis

Water for P analysis was passed through a cationexchange column (Amberlite resin, IR-120 (H), 14–52 mesh, particle size 0.30–1.18 mm) to remove iron and possibly other metals, which precipitate during digestion. Anions were not separated, since tests with standard additions indicated no interference.

Ammonium, nitrate and phosphate fractions were analysed using a Skalar San^{plus} segmented flow analyser (Skalar Analytical, Breda, The Netherlands; 1995). The use of 3 and 5-cm path length cuvettes gave limits of detection of approximately 2 μ g l⁻¹ for N and P, respectively. TN, FON, TP, FOP were estimated following alkaline persulphate digestion of filtered and unfiltered samples (Langer & Hendrix, 1982; Lambert & Maher, 1994).

Oven-dried $(105^{\circ}C)$ and pre-weighed moss shoots (about 15 mg) were prepared for digestion by immersion overnight in a mixture of sulphuric acid, selenium, potassium sulphate and salicylic acid prior to digestion in order to prevent any loss of $NO₃–N$ during the digestion step (Novazmsky et al., 1983; Houba et al., 1989). Digestion procedures broadly followed the methods of Walinga et al. (1989) using a Skalar 5620/40 digestion block (5600 controller). The clear digest was diluted to 250 ml in deionised water and N (as ammoniacyl-N) and P (as phosphate-P) were then measured. In order to confirm that digestion was complete, standard reference material (Platyhypnidium riparioides = Rhynchostegium riaparioides: No. 61, individual No. 272; Community Bureau of Reference, ECC) of known N and P contents was also digested and analysed.

Culture

Standard sterile culture procedures were adopted for isolation of an axenic, clonal culture and subsequent experiments with this. Solutions unsuitable for autoclaving were passed through pre-sterilised membrane filters (Whatman, 0.2 μ m). All vessels were soaked (24+ h) in a 2% v/v deionised water/Deacon 90 solution, stoppered (non-absorbent cotton wool), capped with foil and then autoclaved for 45 min at 1×10^5 Pa.

An axenic culture was obtained by adopting the following procedure for a number of capsules of the moss. A capsule, together with 1 cm of seta for ease of handling, was taken from field material. On return to the laboratory the capsule (with calyptra still in place) was surface sterilised by immersing it in a small volume of 70% v/v ethanol for 30 s in a sterile petri dish. It was then transferred to 20% v/v sodium hyperchlorite solution with 0.05% Tween-20 (polyoxyethylenesorbitan monolaurate) for 15 min. The capsule was next washed five times in autoclaved deionised water, followed by suspension in a drop of water and dissection to release the spores. The suspension was pipetted into several flasks of sterile liquid medium and incubated for a period of about a

month whilst the spores developed through a prolific protonemal stage into young shoots. Individual shoots from each flask were then tested for the presence of contaminants with four different bacterial and fungal growth media. These plates were incubated at 20° C for at least 1 week. Those flasks which appeared to be contaminantfree were also checked carefully by light microscopy. Only shoots taken from flasks, which tests indicate were axenic, were used for the subsequent experiments.

The growth medium was buffered (pH 3.6) Chu 10 D, a version of the Chu No. 10 medium (Chu, 1942) modified by Grainger et al. (1989). Standard growth conditions were 20° C, with constant illumination from above $(50 \pm 3 \text{ }\mu\text{mol})$ photon m^{-2} s⁻¹). Each flask was shaken daily by hand to maintain the shoots submerged. Material for experiments was grown in Chu 10 D medium containing 4 mg l^{-1} N and 20 µg l^{-1} P for 7 days to standardise the intracellular P concentrations. The shoots were then cut to a 3-cm length and resuspended in 250-ml Erlenmeyer flasks containing 100 ml of Chu 10 D medium.

The ability to use organic phosphate as a P source for growth was tested with glucose 1-phosphate representing PME, or DNA representing PDE (herring sperm, degraded free acid type IV, 400 Kbp fragments). These tests were compared with shoots incubated in medium with inorganic phosphate at the same P concentration. Three replicates for each treatment were used. The medium included 4 mg l^{-1} N (supplied as CaNO₃) and 0.25, 0.5 or 1.0 mg l^{-1} P. The possibility of autodegradation of the organic phosphates was ruled out by FRP analysis of medium incubated under similar conditions, but without shoots.

Phosphatase assays

Assays were conducted using the methods of Turner et al. (2001) with one of the analogue substrates, pNPP, bis-pNPP or MUP. Assays on field material were always made within 24 h of collection (usually quicker). Each assay consisted of five 2-cm shoot apices, with four replicates for each measurement. The shoots were placed in glass vials, with 2.9 ml medium containing the

appropriate buffer: pH 3.0–6.5; DMG; pH 7.0–8.0, HEPES; pH 8.5–10.0, glycine. The vials were placed in a shaking water bath (ca. 100 strokes min $^{-1}$) at 20°C and about 20 µmol photon m $^{-1}$ s $^{-1}$ and left for 20 min to equilibrate. The assays were then made for 20 min using 0.1 ml substrate to give a final concentration of $100 \mu M$. Assays with axenic material were performed on the entire excised shoot (approximately 3 cm).

To terminate the assay and develop the product colour 2.5 ml assay mixture was immediately removed and added to 0.25 ml terminator solution (Fedde & Whyte, 1990) in a test tube and mixed. This step was precisely timed to gain accurate representation of activity. Following analysis, the material was dried at 105° C for 24+ h and then weighed to 0.00001 g on a microbalance.

In order to establish the pH optimum, assays were conducted at 0.5 and 1.0 intervals within the pH range of $3.0-10.0$ for 20 min, at 20° C using 100μ M substrate concentration. A study was also made of the short-term response to temperature over the range $0-50^{\circ}$ C, using 10° C intervals. Each population was assayed for 20 min at 100 μ M pNPP/bis-pNPP concentration, to pH 5.5 and 10° C intervals from 0 to 50 $^{\circ}$ C.

Comparisons were made between the phosphatase activities of whole shoots (axenic) and 2 cm apical tips (field) in order to establish whether results for field material might be influenced by residual bacterial activity in the field material.

Results

Stream water

There were obvious changes in nutrient concentrations during the 2-year period (Fig. 1), with a peak in FOP occurring in late spring in both years, but otherwise no obvious similarity between the 2 years. Mean annual concentration of NH₄–N was significantly higher ($P < 0.01$) in 2000 (303 \pm 76 µg N l⁻¹) than in 1999 (176 \pm 60 µg N l⁻¹). Overall, inorganic N was more important than organic N, whereas the converse was true for phosphate. The ratio of filtrable N:P (by mass) was almost always very high, whether or not organic N was included (Table 1).

Morphology

Axenic material was quite similar morphologically to the field population, but axenic shoots

Fig. 1 Concentrations of filtrable combined nitrogen (ammonium, nitrate, organic) and filtrable phosphorus (reactive, organic) in Brandon Pithouse Stream. Samples taken from March 1999 to February 2001

99/00) and March 2000 to February 2001(sample year 00/01); values for N:P ratio (by mass) are included										
Variable	Sample year 99/00		Sample year 00/01							
	Mean	SD	Mean	SD						
Flow $(dm^{-3} s^{-1})$	0.5	(0.4)	0.7	0.5						
Temperature $(^{\circ}C)$		0.5	8.9	0.6						
Conductivity (μ S cm ⁻¹)	1,173.6	397.1	1,273.4	319						

Table 1 Physical and chemical variables of Brandon Pithouse stream from March 1999 to February 2000 (sample year

pH 3.7 0.4 3.9 0.5 20.8 Dissolved oxygen (%) 31.1 30.8 13.1 20.8
FRP (μ g $]^{-1}$) 31.1 22.2 7.7 5.6 FRP (μ g l⁻¹) 19 22.2 7.7 5.6 FOP (μ g l⁻¹) 34.1 46.7 37.4 43.6 TIN:FRP 694 694 66.6 84.3 FON:FOP $-$ 86.3 97.6

were less robust and there was rare branching. Further observations on the field population showed that branching was more frequent during periods of low flow. From the basal part of the apical tips of the axenic shoots used in the investigation a small number of rhizoids where also observed, however this was noted for a few shoots only. In the field the shoots occur in very dense beds (shoot number to unit area), which act as traps for detritus, a potential source of N and P. deriving from the stream water or the shoots themselves. This clumping of shoots, which was not present in the cultures, may also have an effect on morphology. General observations of cultured and field shoots, although there are no available data on the latter, suggest that growth (elongation) is faster in the culture shoots.

Tissue nutrients

Tissue N concentration was almost always higher in 2000 than 1999, with the mean value significantly higher $(P < 0.01)$. In contrast, tissue P concentration did not show a consistent pattern through the year, though the average was not much less than in 1999 (Fig. 2). There was no relationship between tissue N and tissue P; the N:P ratio was mostly much higher in 2000 than 1999. Seasonal trends were not evident and the N and P contents were not significantly correlated $(P > 0.05; Fig. 2)$.

The mean values for tissue N and P were lower in laboratory culture than in the field (Table 2). However, during growth in batch culture, the tissue P concentration was higher on day 3 than at the beginning of the experiment or at any time subsequently (Fig. 3). The effect was apparent even with $0.25 \text{ mg } l^{-1}$ P in the medium, though most obvious with 1 mg l^{-1} . As a consequence, the lowest N:P ratio occurred on day 3 (Table 2). However, by day 30 the tissue N:P was lower in

Fig. 2 Tissue N and P concentration and N:P ratio of 2-cm tips of Warnstorfia from Brandon Pithouse stream between March 1999 and February 2001

Material	Tissue N (mg N g^{-1} d.wt)				Tissue P (mg P g^{-1} d.wt)				Tissue N:P			
	Year 1		Year 2		Year 1		Year 2		Year 1		Year 2	
	Mean	SD.	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD
Field shoots Axenic shoots	25.8	10.6	30	6.64	2	0.3	2.1	0.5	11.4	2.8	14.8	2
Inorganic phosphate	16.8	4.4			1.5	0.9	\equiv	$\overline{}$	14.1	6.9		
Glucose-1-phosphate	13.5	3.1			$1.1\,$	0.5		-	14.7	5.6		
DNA	15	3.1	\equiv			0.3	$\qquad \qquad =$	$-$	16.5	6		

Table 2 Nitrogen and Phosphorus concentrations of 2-cm tips of Warnstorfia taken from Brandon Pithouse Stream from March 1999 to February 2000 (sample year 1: $n = 13$)

and from March 2000 to February 2001 (sample year 2: $n = 22$) and the concentrations of N and P in the axenic shoots from the laboratory study

(–) Data not available

culture than the mean value for field material (Table 2).

Use of organic substrates

The moss grew on glucose-1-phosphate and DNA almost as well as Pi (Fig. 3), with some indication of the response being more rapid in the sequence $Pi \ge G-1-P > DNA$ e.g. relative concentrations of tissue P on day 3. There was an obvious lag in the increase of shoot length with DNA at all three P concentrations tested, though the increase per unit time was similar later.

Phosphatase activities

PMEase and PDEase activities were detectable throughout the sampling and experimental period (Fig. 4); there were only a couple of measurements which approached the detection limit in the field. The majority of field measurements gave quantitatively similar PMEase results with pNPP and MUP, except during the latter part of 1999 and the first two months of 2000, which showed higher values using MUP (Fig. 4). PMEase hydrolysis of pNPP ranged from 11.4 to 101.5 µmol pNP g⁻¹ d.wt h⁻¹, whilst hydrolysis of MUP ranged from 10.1 to 174.5 μ mol MU g⁻ 1 d.wt h⁻¹. PDEase activity ranged from 2.0 to 97.8 µmol pNP g⁻¹ d.wt h⁻¹. In 1999, PMEase and PDEase activity showed low values in spring and early summer and high values from late summer to early winter. No pattern of seasonal change was evident in 2000. PMEase (pNPP and MUP) and PDEase activities were strongly correlated ($pNPP$ and bis- $pNPP$, $r = 0.67$, MUP and bis-pNPP, $r = 0.73$; $P < 0.01$ in both cases).

The axenic material showed higher PMEase and PDEase activities when grown with organic P than with Pi (Fig. 3). PMEase activity showed a slight increase when PDE (i.e. DNA) was the source and the converse was also true. However, the increase in PDEase activity was greater when shoots were cultured with DNA than with glucose-1-phosphate (Fig. 3).

The influence of environmental factors was tested on field material during a standard 20-min assay. There was a linear increase in PMEase and PDEase activities between 0 and 50° C (*r* values; PMEase 0.96, PDEase 0.94: data not shown). Above 60° C there was visible cell damage at the end of the assay, while activity at 70° C was less than at 60 \degree C. The Q_{10} value (ratio of activity at a given temperature t and the activity at $t + 10$ peaked between 10 and 20° C and then decreased as temperature increased. The range of Q_{10} values for PMEase (1.0–3.0) was larger than PDEase (1.4–2.11), showing PMEase to be more influenced by temperature.

An experiment to determine any possible effect of light showed that there was a marked effect on PMEase but not on PDEase activity. In the dark (0 μ mol photons m⁻² s⁻¹) PMEase activity was nearly 2-fold higher than in natural light (13 µmol photons $m^{-2} s^{-1}$). There was no significant change in PDEase activity ($P < 0.05$) under the light conditions tested. However, a repeat experiment using another sample of field material and comparing the effect dark and low

Fig. 3 Influence of P source (Pi, glucose-1-phosphate, DNA) and concentration $(0.25, 0.5, 1.0 \text{ mg } l^{-1} \text{ P})$ on N and P concentrations, N:P ratio (by mass) and growth (shoot

length) of shoots of axenic Warnstorfia. Incubation conditions 20 $^{\circ}$ C and 50 µmol photon m⁻² s⁻¹. Shoots sampled on days 0, 3, 5, 10, 17, 24, 31

level natural light showed the opposite response for PMEase activity, with activity in the dark about 60% of that in the light.

The pH profile for PMEase and PDEase activities was similar for field and axenic material, including the optimum pH; however, the optimum was apparently slightly higher for PDEase (5.5) than PMEase (5.0) (Fig. 5). There was no visible cell damage at either of the pH limits tested (3.0, 10.0). A repeat experiment using 1 µM MUP showed no difference in the form of the response curve from that of $100 \mu M$ MUP over most of the pH range, but there were marked differences at pH 3.0, especially for PDEase activity, where there was a second peak at pH 3.0 using $1 \mu M$ substrate.

PMEase and PDEase activities showed a hyperbolic relationship with substrate activity (Fig. 6). Linear transformations of the data gave mean K_{m} values of 321 and 138 μ M, and V_{max} values of 318 and 117 µmol pNP g⁻¹ d.wt h⁻¹ for PMEase and PDEase, respectively. These values indicate that potential reaction rate of PMEase is higher, but with a lower substrate affinity, than PDEase. The K_m and V_{max} values for PMEase of the axenic population were lower than the field

Fig. 4 Changes in PMEase and PDEase activities of 2-cm tips of Warnstorfia from March 1999 to February 2001. Values are mean \pm SE; $n = 4$

material (87.2 μ M and 35.8 μ mol pNP g⁻¹ d.wt h⁻¹, respectively).

There was no significant correlation between phosphatase activities and stream water nutrient status ($P > 0.05$; data not shown). However, there were significant inverse relationships between tissue N and stream water P and between tissue P and stream water N (Table 3). Tissue N was also significantly positively related to stream water $NO₃–N$. In the seasonal study the PMEase and PDEase activities were inversely related to tissue P concentration and PDEase was inversely related to tissue N concentration. The only significant correlation with tissue N:P was PDEase activity of axenic shoots cultured in Pi (Table 4). Activity of the axenic shoots showed inverse relationships with cellular N and P and positive relation with cellular N:P.

Filtrable organic phosphate concentration was higher than that of filtrable inorganic phosphate in all 24 months. The large peaks of organic phosphate in the spring of both years are similar to the peaks reported for other streams in northern England at this time of year (e.g. Livingstone & Whitton, 1984). The study showed not only that W. fluitans can produce surface enzymes hydrolysing organic phosphates, but that these permit the organism to obtain phosphate from these sources. It seems less likely that filtrable organic N is important, because the aqueous N:P ratio of the water was usually very high even when only inorganic N is considered. The differing hydrolytic rates of the two phosphomonoesters substrates $(pNPP$ and MUP) suggest that there may be a range of specific PMEases or a broadly specific PMEase with various substrate affinities. Differences in response to pNPP and MUP have been reported in some, but not all, studies of surface phosphatase activity of aquatic phototrophs (Whitton et al., 2005).

The response of phosphatase activities to environmental factors was quite similar in field and axenic material, suggesting that any influence of periphytic bacteria was negligible compared with activity of the moss. Marked changes were observed in PMEase and PDEase activities of the field population during the 2-year period. PDEase activity was high from late summer to early winter 1999 at a time when the tissue N:P ratio was especially high, suggesting that the increased activity was a response to P limitation. PMEase assayed with MUP was also high at this time, but not that assayed with $pNPP$, raising the possibility that a further PMEase becomes active under some conditions. However, the greater activity shown with MUP was apparent during a period when N:P was relatively low (Fig. 3), rather than the converse as might be expected. Both field and laboratory results indicate that PDEase activity is often as high as PMEase activity assayed with pNPP, though less in the stream material during the period when PMEase activity assayed with MUP was very high. Upland populations of the aquatic moss F. antipyretica sometimes also show higher PDEase than PMEase activity (Christmas

Fig. 5 Effect of pH on the PMEase and PDEase activities (µmol MU g^{-1} d.wt h⁻¹) at 1 and 100 µM and MUP substrate concentration of 2-cm tips of field and axenic Warnstorfia shoots

& Whitton, 1998b). This contrasts with the shoots of terrestrial mosses, where (surface) PDEase activity is usually less and sometimes (Palustriella commutata var. commutata and Polytrichum commune) absent (Turner et al., 2001). As some terrestrial mosses possess rhizoids penetrating the substratum, perhaps these provide a site for PDEase activity, which seems especially likely in a species such as P. commune, which possesses primitive conducting tissue (Proctor, 2000). Rhizoid development on Warnstorfia was so sparse it seems unlikely these contributed much to the observed PDEase activity. The effect of light on phosphatase activity indicated the importance of light for PMEase activities of Warnstorfia, but more detailed study is required on the influence of other factors on its response to light. On the other hand, it seems that PDEase activity is not light dependant.

There appears to be a threshold value for tissue P concentration, below which activity starts to increase markedly. This was about 2 mg P g^{-1} d.wt in field material and 1.5 mg P g^{-1} d.wt in culture, perhaps reflecting the differences in average tissue P concentration between field and cultured material (Table 2). Further study is needed to clarify what factors influence this. Possibilities include the differing ionic environment, since this may influence nutrient uptake (Glime & Vitt, 1984) and the presence of a current passing water over the moss in the field. The higher tissue N of the field population during the second year may be influenced by the changes in the aquatic environment that occurred as a result of mining activity within the catchment area. Notably, there was a greater contribution of NH4–N to TIN, which rose from 13% to 37%, and also an increase in water pH (see Fig. 1 and Table 1). As the apical tips of aquatic bryophytes generally reflect their chemical environment (Brown, 1984) then the increase in tissue N could be expected given the change in environmental conditions. Warnstorfia appears to be slightly less responsive to P limitation than F. antipyretica and R. riparioides (Christmas & Whitton 1998a), where PMEase activity increased when the tissue P concentration fell below 3 mg P g^{-1} d.wt.

As already reported (for 1999: Ellwood et al., 2002), there were no significant relationships between phosphatase activity of the field population and any of the measured ambient variables. If some feature of internal P content is the factor

Fig. 6 Effect of substrate concentration on PMEase and PDEase activities of 2-cm tips of Warnstorfia shoots. Values are mean \pm SE; $n = 3$

Tissue nutrient	Aqueous phosphorus fraction					Aqueous nitrogen fraction					
	NO2	$\rm NH_{4}$	TIN	FON	TN	FRP	FOP	TP	TOP	FTP	
N	$-0.79**$	-0.22	$0.69**$	0.02	$0.60**$	0.3	$-0.47*$	$-0.43*$	$-0.42*$	$-0.49*$	
P	$-0.45**$	$0.42*$	$-0.37*$	0.14	0.04	-0.33	0.19	-0.04	0.01	0.06	
N: P	-0.18	0.06	-0.17	0.02	0.05	0.04	0.03	0.02	0.02	0.04	

Table 3 Relationship between stream water nitrogen and phosphorus concentrations and the nitrogen and phosphorus content, and the N:P of the 2-cm tips of Warnstorfia sampled between March 1999 and February 2001

The results are shown as correlation coefficients (r values), with significant values shown in bold: *P < 0.05; **P < 0.01

involved in inducing synthesis of PMEase and PDEase, a close relationship might not be expected because activity would reflect aqueous P concentration over a period representing the response time of the moss to changing P status. A lag in the response would be expected, as with PMEase of Rhynchostegium riparioides (Christmas & Whitton, 1998a). The results contrast with the situation for the stream alga Stigeoclonium, where PMEase activity was inversely related to tissue P measured on the same day (Gibson & Whitton, 1987). However, Stigeoclonium grows much more rapidly than Warnstorfia, so the prior period of days influencing activity is much less. Another difference might be internal storage and translocation of nutrients, which is reported for some mosses, though mostly with respect to N (Bates, 1994; 2000; Wells & Brown, 1996; Eckstein & Karlsson, 1999). However, this has not yet been investigated in Warnstorfia.

Evidence from the present study suggests that internal phosphate concentration is involved in regulating phosphatase activity of Warnstorfia. Tissue N and P concentrations of field shoots were inversely related to the aqueous P and N, respectively (Table 4). In the batch culture experiment on the influence of initial P concentration, the lowest aqueous P concentration led to the lowest tissue P concentration and the highest PMEase and PDEase activities. However, the external P source also has an effect, because culturing with glucose-1-phosphate led to slightly higher PMEase and culture with DNA to markedly higher PDEase. The enzymes are essential for the moss to make use of the organic phosphate and, at least in the case of PDEase, the high level of activity cannot simply be a reflection of reduced internal tissue P concentration, because the effect was already pronounced by day 3.

It is difficult to assess the extent to which PDEase activity in Warnstorfia is a response to tissue P limitation or to the presence of suitable substrate in the environment, because both factors appear to be important. The monthly sampling in 1999 showed PDEase activity to increase during periods of low FOP, but this may simply be due to insufficient sampling as within-month variation in 2000 showed large differences in activity. In the

Substrate	% N	% P	N:P
pNPP	-0.218	$-0.551**$	0.245
MUP	-0.295	$-0.522**$	0.110
$bis-pNPP$	$-0.439**$	$-0.663**$	0.124
PMEase	$-0.585*$	$-0.640*$	0.568
PDEase	$-0.686*$	$-0.784**$	$0.776**$
PMEase	-0.480	-0.444	0.424
PDEase	$-0.698*$	-0.290	0.029
PMEase	-0.326	-0.392	0.315
PDEase	$-0.885**$	$-0.636*$	0.475

Table 4 Relationship between rates of PMEase and PDEase activity and tissue N and P concentrations of 2-cm tips of Warnstorfia

The field population was sampled from March 1999 to February 2001. Results shown as correlation coefficient (r value), with significant values shown in bold: $*P < 0.05$; $*P < 0.01$

culture experiment PDEase activity doubled within the first three days of addition of a PDE substrate, but subsequently decreased again as the tissue P concentration increased.

The high temperature optimum in short-term assays of phosphatase activities of Warnstorfia is similar to that found in other studies of surface phosphatase (Whitton et al., 2005). The pH optimum of PMEase activity of Warnstorfia (5.0) is also similar to that of other aquatic mosses, such as *F. antipyretica* (5.5: Christmas & Whitton, 1998b) and R. riparioides (5.5: Ellwood, 2002). The value of pH 5.0 is within the range sometimes found in the stream water during the present study, but much higher than the value (2.6) occurring from at least the mid-1960s (probably much earlier) and the early 1990s. There was no indication that a substrate concentration $(1 \mu M)$ more like that occurring in the field leads to an optimum for PMEase activity more like field conditions, as occurs with some other organisms (Whitton et al., 2005). There appeared to be a second peak in PDEase activity at pH 3.0 with 1 μ M, but not 100 μ M bis-MUP, so perhaps Warnstorfia is particularly efficient at using diesters at low pH. Protonema occurred only sparsely during the present study, but this growth stage dominated the uppermost part of the stream in the 1960s (B.A. Whitton, unpublished data). It would of interest to know whether its phosphatase activities respond differently to pH.

It was possible to apply linear transformations of the Michaelis–Menton equation to the data as for pure enzyme systems. Rates of PMEase and PDEase activity in the natural population were similar, ranging from 303.9 to 101.5 and 30.8 to 97.8 µmol pNP g⁻¹ d.wt h⁻¹, respectively. However, affinity for PME was 3-fold lower than for PDE. It was suggested by Christmas and Whitton (1998b) that the low affinity of PDEase of F. antipyretica sampled from an upland stream could be important for utilising pulses of high concentrations of PDE, rather than continuous low concentrations. If so, the high affinity of PDEase recorded here may be important for continuously low PDE concentrations. The axenic population had much lower PMEase K_m values than the field population. As there was no organic P source in the standard medium, a reduction in

 K_m would fit with the theory of affinity of a species adjusting to the type and concentration of the substrate.

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