



# Pharmacological and insecticidal attributes of common fruit peels: a review

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## Abstract

India has the biggest market for fruits, vegetables, and flowers in Asia and generates enormous amounts of garbage every day. Numerous food sectors throughout the globe generate a substantial amount of waste in the form of peels or by-products. The environmental system is severely harmed by these wastes. Today, numerous methods are employed to come up with other applications for these wastes since the by-products are a great source of diverse bioactive compounds beneficial to human health, such as creatine, carotenoids, flavonoids, polyphenols, and polysaccharides. The current review offers several unique perspectives on using fruit peels to minimise waste generation and its detrimental effects. An innovative bio-refinery strategy would try to turn fruit peel waste into a greater variety of useful compounds. The leftovers from most extraction procedures may also be utilised as sustainable resources to make biofuels. This paper also discusses the use of various techniques employed while analysing substances in plant-based goods, including the latest advanced technology. Moreover, to make certain effective waste management solutions, a careful assessment of the economic and environmental advantages relative to current choices is required.

**Keywords** Fruit peel · Extract · Antioxidant · Extraction · Risk assessment · Antimicrobial

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## 1 Introduction

In developing countries, plants were used traditionally to prepare herbal medicines as they exhibit several pharmacological properties (Ahmad et al., 1998). India is home to around 15% of the 20,000 medicinal plants found worldwide, or 300–5000 plants (Lydia et al., 2016a). Because of their ease of availability and little to no toxicity, previous studies have focused on the development and use of therapeutic plants. The growing trend of using plants as medicine has resulted in about 80% of the world's population relying on alternative medicine sources. One substitute source for this is fruit peels, which are discarded into the environment instead of being employed for their various pharmacological characteristics and are regarded as agricultural waste (Abdollahzadeh et al., 2011; Saleem & Saeed, 2020). The consumption of the fruits, (like production of jams and jellies) produced wastes on exceptionally large scale by fruit processing industries and cause environmental pollution due to improper utilization and disposal. Therefore, using fruits peels to prepare herbal drugs are beneficial to our environment in all aspects, as they can reduce the environmental pollution, can be used to make products from them and, also, hundreds of diseases can be cured by the products prepared from these wastes. The peels of such fruits are a great source of nutrients and function as a rich source of valuable compounds, supplements, food additives and some value-added products can be produced by them (Pathak et al., 2019). It's been noted that fruit peels include phenolics, tannins, flavonoids, triterpenoids, glycosides, carotenoids, ellagitannins, anthocyanins, vitamin C, and essential oils. The amount of total phenolics and flavonoids in the same fruit peels might vary depending on the solvent used, as shown in Table 1. Sugar components can also be extracted from peels of different fruits such as almond fruit, banana, calabash, custard apple, muskmelon, orange, papaya, pineapple, pomegranate (Kumar et al., 2012). Some fruit peels namely *Carica papaya* (papaya), *Ananas comosus* (Pineapple) and *Actinidia deliciosa* (Kiwi) have proteolytic enzyme activity, hence, they can be a suitable alternative to produce enzymes at low cost as they utilize the fruit waste and reduce the pollution (Vel & Stanley, 2015). Due to their potential for diversified bioactivities, such as anti-carcinogenic, digestive, analgesic, antioxidative, antiviral, and antibacterial effects, moreover, these volatile substances are exploited in the pharmaceutical business (Liang et al., 2022). According to Parashar et al., 2014, fruit peels have enormous the ability to function as a source of newer, better, healthier, and more efficient for various purposes. The fruits have specific and identifiable scent, since there is a diverse variety of volatile chemicals. Aroma is a key sensory quality that has a considerable impact on how customers perceive and accept fruit and its products. The applications of common fruit peels are depicted in Fig. 1. Marina and Noriham (2014) studied the mango peel extract possesses more phenolic content compared to guava and papaya peel extract i.e., mango peel extract has more antioxidant effects than guava and papaya peel extract. Antioxidants are frequently employed to increase the shelf life and stabilization of lipids and foods containing lipids additionally to preserve and safeguard foods against microbial spoilage, dis-colouration, or degeneration brought on by auto-oxidation (Sultana et al., 2012). Radical initiated reaction can be prevented by antioxidants. Antioxidant and antibacterial property are reported in different kind of fruit peels such as *Maclura pomifera* (Osage orange), *Punica granatum* (Pomegranate or Anar), *Musa balbisiana* (Banana), *Citrus limetta* (Sweet lime or Mousambi), *Malus domestica* (Apple), *Carica papaya* (Papaya), and *Mangifera indica* (Mango) (Prakash et al., 2013). Table 2 demonstrates the antioxidant capability of various fruit peels as determined by various workers through different tests (Fig. 2).

**Table 1** Total Phenolic and Flavonoid contents present in different fruit peels

Name of fruit peels	Extraction solvent	Total phenolic content	Total flavonoid content	References
Pomegranate peel	Aqueous	161.25 mg/g	–	Kanatt et al., (2010)
<i>Citrus reticulata</i>	Aqueous	40.8 mg GAE/g	7.62 mg CE/g	Ho & Lin, (2008)
<i>Citrus sunki</i>	Ethanol	14.52 mg/g	–	Shin et al., (2011)
Orange peel	Methanol	165.38 mg/g	28.36 µg/g	Hegazy & Ibrahim, (2012)
	Ethanol	169.56 mg/g	29.75 µg/g	
<i>Citrus paradassi</i>	Dichloromethane	98.64 mg/g	17.39 µg/g	Oboh & Ademusun, (2012)
	Acetone	145.79 mg/g	21.87 µg/g	
	Hexane	63.20 mg/g	13.89 µg/g	
	Ethyl acetate	85.27 mg/g	18.36 µg/g	
	Acetone	13.1 mg/g	0.93 mg/g	
	Ethanol	0.318 mg GAE/ mg	0.206 mg RE/ mg	
<i>Citrus hystrix</i>	Ethanol	17.99 mg/g	15.11 µg/g	Wijaya et al., (2017)
<i>Citrus maxima</i>	Ethanol	21.38 mg GAE/g	2.59 mg CE/g	Ani & Abel, (2018)
<i>Citrus sinensis</i>	Ethanol	36.09 mg GAE/g	4.61 mg CE/g	Liew et al., (2018)
<i>C. Limonia Osbeck</i>	Methanol	17.37 mg/g	–	Saleem & Saeed, (2020)
	Ethanol	18.7 mg/g	–	
	Methanol	16.2 mg/g	–	
	Ethyl acetate	29.72 mg/g	–	
Orange peel	Aqueous	12.42 mg/g	–	Saleem & Saeed, (2020)
	Ethanol	15.2 mg/g	–	
	Methanol	12.4 mg/g	–	
	Ethyl acetate	25.21 mg/g	–	
Banana peel	Aqueous	30.0 mg/g	196.1 mg/g	Pathak et al., (2017)

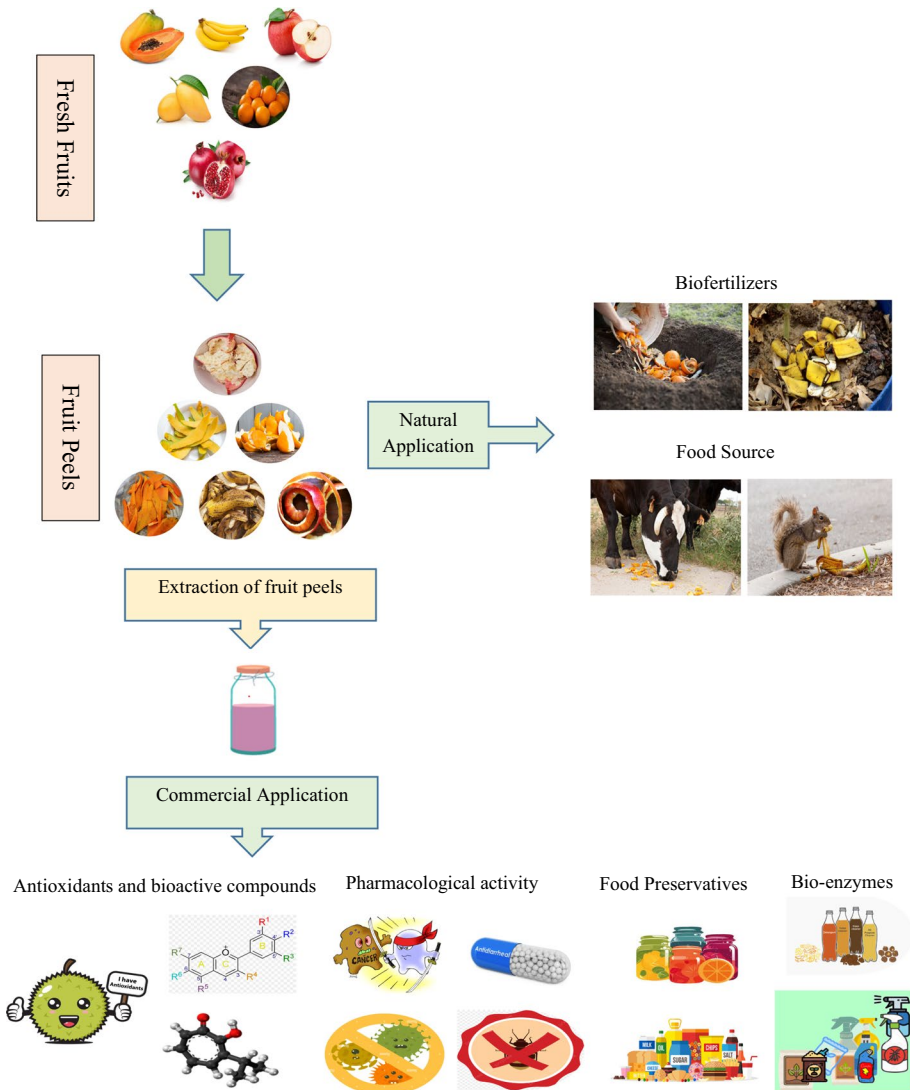
Table 1 (continued)

Name of fruit peels	Extraction solvent	Total phenolic content	Total flavonoid content	References
Banana peel	Ethanol	11.48 mg/g	–	Salern & Saeed, (2020)
	Methanol	8.6 mg/g	–	
	Ethyl acetate	11.48 mg/g	–	
	Aqueous	15.6 mg/g	–	
<i>Mangifera indica</i> peel (Raw)	Acetone	109.70 mg/g	–	Ajila et al., (2007)
<i>Mangifera indica</i> peel (Ripe)	Acetone	100.00 mg/g	–	
<i>Mangifera indica</i> peel (Unripe)	Ethanol	92.6 mg GAE/g	22.2 mg RE/g	Kim et al., (2010)
<i>Mangifera indica</i> peel (Ripe)	Ethanol	70.1 mg GAE/g	21.2 mg RE/g	
Mango peel (Langra)	Methanol	116.8 mg/g	90.89 mg/g	Sultana et al., (2012)
Mango peel (Chonsa)	Methanol	122.6 mg/g	92.55 mg/g	
Papaya peel	n-Hexane	7.32 mg GAE/g	1.10 mg CE/g	Asghar et al., (2016)
	Dichloromethane	21.15 mg GAE/g	1.23 mg CE/g	
	n-Butanol	24.80 mg GAE/g	3.72 mg CE/g	
	Ethyl acetate	27.21 mg GAE/g	8.57 mg CE/g	
	water	32.23 mg GAE/g	12.93 mg CE/g	
	Methanol	35.15 mg GAE/g	13.92 mg CE/g	
Papaya peel	Ethanol	43.79 mg GAE/g	19.81 mg CE/g	
	Acetone	4.22 GAE/ g	247.70 meq/quertin	Lydia et al., 2016b
	Ethanol	3.92 GAE/ g	0.83 meq/quertin	
Apple peels	Aqueous	4.84 GAE/ g	15.48 meq/quertin	
	Acetone	7.68 mg/g	5.34 mg/g	Massini et al., (2013)
	Ethanol	6.35 mg/g	4.76 mg/g	

**Table 1** (continued)

Name of fruit peels	Extraction solvent	Total phenolic content	Total flavonoid content	References
Apple peels	methanol/water/acetic acid	32.3 mg GAE/g	-	Agouram et al., (2013)
	ethanol/water	20.1 mg GAE/g	-	
	acetone/water	61.3 mg GAE/g	-	
Apple peels	Ethanol	24.94 mg GAE/g	8.85 mg CE/g	Lee et al., (2016)

GAE/g- Gallic Acid Equivalents/gram; CE/g- Catechin Equivalents/gram; RE- Rutin Equivalent;



**Fig. 1** Fruit peels and their uses in various fields

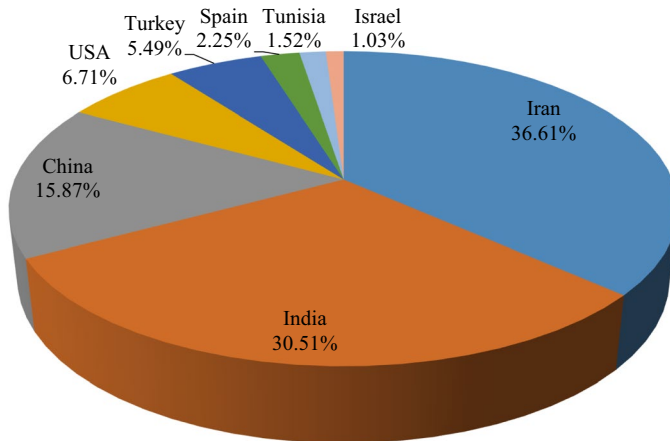
The main goal of this study is to provide a pathway for more effective use of waste from fruit processing that satisfies Sustainable Development Goal No. 12 of the UN: “Sustainable production and consumption”. This objective covers the primary difficulty of effectively managing waste and using natural resources to assist poor countries in transitioning to more sustainable patterns of consumption by 2030. If any action is to be done for a product connected to these fruit peels being utilized in the future as food additives, food preservatives, and toxic-free food grade disinfectants, the food or pharmaceutical industries may find this information to be productive. Given the present issues surrounding the usage of antibiotics, interest in plants possessing

**Table 2** Antioxidant capacity of different fruit peels by using different radical scavenging assay

Name of fruit peels	Scavenging assay	IC <sub>50</sub>	References
Pomegranate	DPPH	4.9 µg/mL	Kanatt et al., (2010)
	Butylated hydroxy toluene	21.2 µg/mL	
Pomegranate	DPPH	0.003 mg/ml	Okonogi et al., (2007)
	ABTS	4.59 mM/mg	
<i>Citrus reticulata</i>	NO	1.98 mg/ml	Ho & Lin, (2008)
	DPPH	371 µg/mL	
<i>Citrus paradissi</i>	DPPH	1.4 mg/ml	Oboh & Ademosun, (2012)
	.OH <sup>-</sup>	5.1 mg/ml	
	Fe <sup>2+</sup>	0.63 mg/ml	
	LPO	151.8 µg/ml	
<i>Citrus maxima</i>	DPPH	1.6 mg/ml	
	.OH <sup>-</sup>	5.6 mg/ml	
	Fe <sup>2+</sup>	0.71 mg/ml	
	LPO	156.1 µg/ml	
<i>Citrus hystrix</i>	DPPH	53.83 mg/ml	Wijaya et al., (2017)
<i>Citrus sinensis</i>	DPPH	13.96 mg TE/g	Liew et al., (2018)
	FRAP	275.62 mmol Fe (II)/ g	
	ORAC	0.73 mol TE/ g	
<i>Citrus clementina</i>	DPPH	63.4 µg/mL	Saratale et al., (2018)
	ABTS	49.6 µg/mL	
Banana peel	DPPH	0.031 mg/ml	Okonogi et al., (2007)
	ABTS	1.80 mM/mg	
Banana peel	DPPH	8.45 µg/mL	Pathak et al., (2017)
Banana peel	Phosphomolybdenum	23.86 mg AAE/g	Hanafy et al., (2021)
<i>Mangifera indica</i> (Raw)	DPPH	1.98 µg of GAE	Ajila et al., (2007)
	LPO	4.59 µg of GAE	
	Soybean Lipoxygenase	5.14 µg of GAE	
<i>Mangifera indica</i> (Ripe)	DPPH	1.83 µg of GAE	
	LPO	3.13 µg of GAE	
	Soybean Lipoxygenase	5.24 µg of GAE	
Apple peel	ABTS	0.71 µM TE/g dry weight	Agourram et al., (2013)
Apple peel	DPPH	12.11 mg/g	Massini et al., (2013)
	FRAP	13.26 mg/g	
Apple peel	DPPH	9.86 mg AAE/g	Lee et al., (2016)
	ABTS	26.30 mM TE/g	

IC<sub>50</sub>-Inhibition concentration; DPPH- 2,2-diphenyl-1-picrylhydrazyl; ABTS-2,2'-azino-bis (3-ethylbenzothiazoline-6-sulfonic acid); NO-Nitric Oxide;.OH<sup>-</sup> Hydroxyl radical; LPO-Lipid peroxidation; FRAP-Ferric Reducing Antioxidant Power; ORAC-Oxygen Radical Absorbance Capacity; Fe<sup>2+</sup> Chelation Assay; TE- Trolox Equivalents; mg AAE/g milligram Ascorbic Acid Equivalent/gram (dry weight); GAE- Gallic Acid Equivalents

antibacterial characteristics has increased. Thus, the aim of this research is to shed light on the biotechnological methods employed to extract the numerous phytochemicals and beneficial compounds from fruit waste. The present review paper provides all



**Fig. 2** Pomegranate fruit production across the world in 2020

the related details of biological activities in a concise manner to the community and readers as well.

## 2 Antioxidant property

Gallic acid and ellagic acid possess strong anti-free radical properties and they reduce the levels of superoxide anion and hydrogen peroxide and detoxify the liver enzymes such as catalase, peroxidase, and superoxide dismutase (Chidambara Murthy et al., 2002).

### 2.1 Pomegranate peel

Due to the antioxidant property of pomegranate, it helps in stabilizing sunflower oil and provides thermal stability (Iqbal et al., 2008). The fruit skins of pomegranates include phenolic compounds such as flavonoids (anthocyanin, catechins) and tannins (punicalin, pedunculagin, punicalagin, ellagic and gallic acid) (Ismail et al., 2012).

### 2.2 Citrus peel

The reports demonstrate that the ethanol-based extract of orange peel exhibits a considerable number of phenolic compounds and there is a close correlation between the total phenolic content and the reducing power (Hegazy & Ibrahim, 2012). A new strategy for the treatment of type 2 diabetes mellitus may be offered by the polyphenolic chemicals derived from *Citrus limetta* peel extract that have both antihyperglycemic and antioxidant effects (Padilla-Camberos et al., 2014). *Citrus maxima* contain many phytochemicals such as flavonoids, alkaloids and phenols and it is a reliable source of nutrients (Ani & Abel, 2018). The antioxidant capacity of phenolic and flavonoid content in plants is the main contributor to the specific biological actions in disease prevention and treatment, according to statistically substantial association between TPC, TFC and the antioxidant assays such



findings support the hypothesis that *C. sinensis* extracts have some antioxidant activity (Liew et al., 2018).

### 2.3 Mango peel

Kim et al. (2010) stated that antioxidant and antiproliferative characteristics of mango peel may be related to the bioactive chemicals that are present there working in concert. The extraction solvent, however, is the most crucial element among the optimal circumstances. The temperature of the extraction process also affects the effectiveness of the process; in particular, temperatures between 50 and 75°C favour the ability of mango peel extracts to suppress lipid peroxidation (Dorta et al., 2012). The results of an HPLC examination showed that the chemical composition of mango peel extract consisted of five primary components: caffeic acid, chlorogenic acid, gallic acid, procyanidin B2, and oleanolic acid, vanillic aldehyde (Bai et al., 2018) (Fig. 3).

### 2.4 Papaya peels

Chukwuka et al. (2013) evaluated the nutritional value of papaya fruit at three distinct stages: unripe, hard ripe and very ripe. Carbohydrate, fat, vitamins, protein, fiber content, calcium, potassium, magnesium, and vitamin-C content decreased as papaya's ripening. However, vitamin A, riboflavin, moisture content was increases as papaya's ripening process. Aqueous extract of papaya peels possesses antioxidant activity (Pawpaw, 2016). Jamal et al. (2017) investigated that *Carica papaya* has the ability to serve as an antioxidant and prevent our body from diseases linked to free radical damage.

### 2.5 Banana peels

Mokbel and Hashinaga (2005) stated that a green peel extract had more significant activity in various solvent extracts than a yellow peel extract. Individual banana peel phenolic compounds, such as dopamine, ferulic acid, and catechins, also have strong antioxidant and antibacterial properties and may be used as food preservatives (Vu et al., 2018).

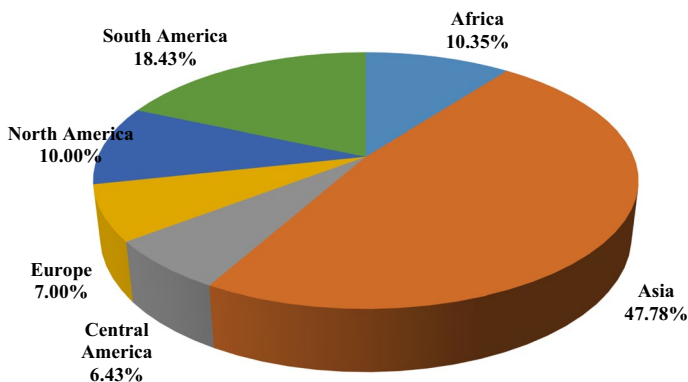


Fig. 3 Citrus fruit production across the world in 2020

## 2.6 Apple peels

The apple peel has the highest concentrations of phenolic chemicals than pulp. Peels possess many phytochemical compounds such as catechins and flavonol glycosides, particularly Rutin (Escarpa & Gonzalez, 1998). Eberhardt et al. (2000) stated that apple fruit has antioxidant activity as 100 gm of fresh apple is equivalent to 1500 mg of vitamin C. In the apple fruit, the compounds, rutin (27.43%), epicatechin (24.93%), dicaffeoylquinic acid (16.14%) and caffeic acid (15.3%) are the most abundant phenolics (Freatianni et al., 2011). The peels had a greater overall concentration of phenolic and flavonoid compounds than the flesh-peel combination or the flesh alone. Apple peels are a valuable source of antioxidants and can provide health advantages when taken due to their high phenolic component concentration and antioxidant activity (Parashar et al., 2014). Apple juice processing generates a huge amount of leftover, known as apple pomace, which consists primarily of peels, seeds, and pulp; and additionally, it supplies bioactive compounds include 114 dietary fibre, protein, biopolymers, and natural antioxidants Cargin and Gnoatto (2017) (Fig. 4).

## 3 Medicinal values

### 3.1 Pomegranate peel

Pomegranate peels may be a desirable alternative in pharmaceuticals as it exhibits anti-inflammatory, anti-allergic (Mastrogiovanni et al., 2019; Panichayupakaranant et al., 2010), anti-ulcerogenic (Moghaddam et al., 2013), anti-cancer (Asmaa et al., 2015), and anti-diarrheal activities (Zhao et al., 2018). *Punica granatum* peel and seed oil extracts have a hepatoprotective effect against the rat liver injury caused by diethylnitrosamine (DEN) and phenobarbital (PB) (Shaban et al., 2013). Pomegranate (*Punica granatum* L) is valued for its high content of polyphenols including ellagic acid, ursolic acid, oleanolic and gallic acids and other flavonoids. Glucose level in blood, total cholesterol and triglycerides maintained under the effect of peel extract and shows the valuable

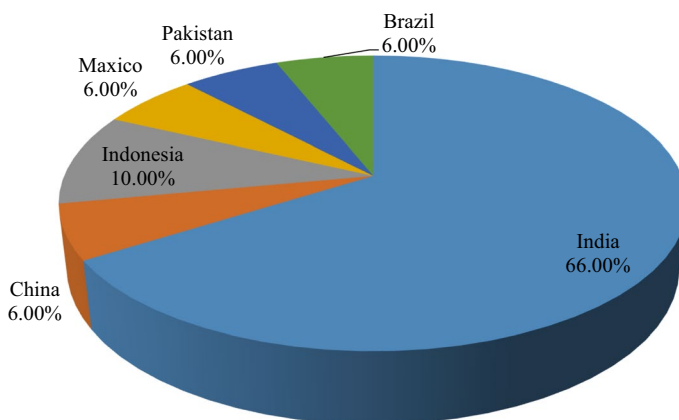


Fig. 4 Mango fruit production across the world in 2020

effect in coronary heart disease (Dos Santos et al., 2016). The action of pomegranate peel extract (PPE) potently suppressed the proliferation of thyroid cancer cell lines and induced apoptosis in cancer cells (Li et al., 2016) (Fig. 5).

PPE reduced the formation of reactive oxygen species (ROS), inflammatory cytokine expression, and cell adhesion molecules in THP-1 monocytic cells (human acute monocytic leukemia cell line) exposed to PM10 as well as avoided inflammatory events brought on by particulate matter (Park et al., 2016). In triple negative breast cancer (TNBC) cells, PPE lowered vimentin, ZEB1, and  $\beta$ -catenin gene expression while increasing E-cadherin expression and caused apoptosis in cells at quite high doses (500, 1000  $\mu\text{g/mL}$ ) (Bagheri et al., 2018). It also helps in maintaining the blood pressure in patients of diabetes mellitus type-2 (Grabež et al., 2020).

### 3.2 Citrus peel

*Citrus microcarpa* (Kalamansi) shows hepato-protective activity against acetaminophen-induced liver damage in male Sprague Dawley rats (Casimiro et al., 2010). Extract of *Citrus limetta* fruit peel using methanol exhibited anticancer efficacy in Swiss albino mice suffering from Ehrlich ascites cancer (Kundu Sen et al., 2012). In male albino mice, *Citrus unshiu* peel extract alleviate hyperglycemia by increasing insulin/glucagon secretion and decreasing hepatic PEPCK gene expression and interaction (Park et al., 2013). Disturbing in insulin secretion in pancreatic cells leading to diabetes, is the most common disease in India. Hydroethanolic extract of *Citrus reticulata* shows anti-diabetic property (Ali et al., 2020). *Citrus sinensis* possesses anti-diabetic and anti-hypercholesteremic activity (Muhtadi et al., 2015). *Citrus* species shows anticancer activity (Nair et al., 2018) and silver nanoparticles of orange peel extract (*Citrus clementina*) shows cytotoxicity against tumour rat glial C6 cells (Saratale et al., 2018).

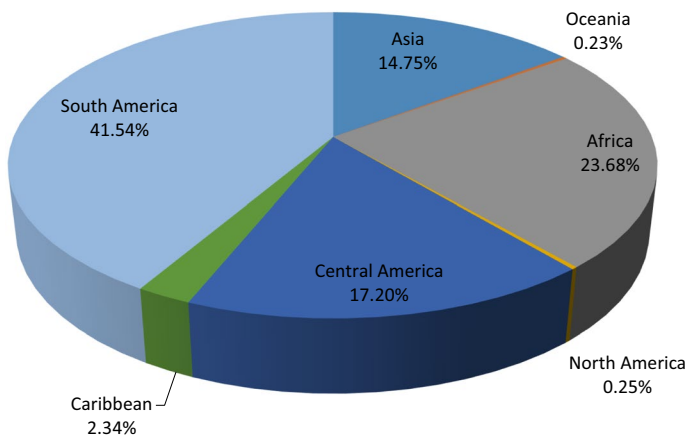


Fig. 5 Papaya fruit production across the world in 2020

### 3.3 Mango peel

Lipid per oxidation, membrane protein degradation, morphological changes and oxidative stress can be protected by using mango peel extract (Ajila & Rao, 2008). Mango peels are also used in fermentation process to produce polygalacturonase using *Aspergillus fumigates* (Garg & Ashfaque, 2010). The extract exhibits various kinds of activities such as anti-proliferative (Ali et al., 2012; Kim et al., 2012) and antidiabetic properties (Gondi & Rao, 2015).

### 3.4 Papaya peels

A live micro-organism that can be consumed as health supplements and provides health benefits is called probiotics. Production of *Lactobacillus acidophilus*, (a probiotic) can be possible with the help of papaya peel extract (Hardia & Iqbal, 2014). John et al. (2021) reported that silver nanoparticles of methanol extract of *Carica papaya* peels possesses antibacterial and anticancer activities (Fig. 6).

### 3.5 Banana peels

Inner peel extract of four species of banana; *Musa sapientum*, *M. paradisiaca*, *M. cavendish* and *M. acuminata*, was evaluated for antioxidant and antihyperglycemic activities. Out of these *M. cavendish* and *M. acuminata* exhibited higher antihyperglycemic activity (Navghare & Dhawale, 2017). Ethanol extract of banana peels lowers the cholesterol level and is helpful in the treatment of obesity (Berawi & Bimandama, 2018). Banana fruit peel extract also possesses anticancer activity, and it may prove beneficial to hematological parameters as there is significant elevation in RBCs, Hb, HCT, MCV, WBCs (Kamal et al., 2019).

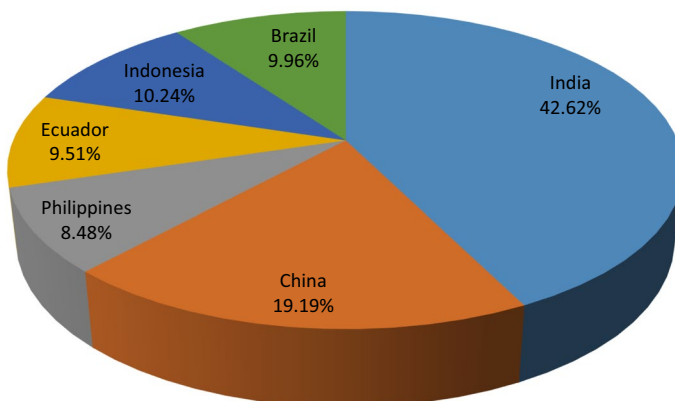


Fig. 6 Banana fruit production across the world in 2020

### 3.6 Apple peels

In the Western world, colorectal cancer is a primary causes of cancer-related fatalities. Colo-rectal cancer is thought to be caused by a combination of variables including diet and lifestyle. Phenolic compounds isolated from apple peel extract are beneficially altered three colo-rectal cancer risk indicators without causing cytotoxicity as it protects DNA degeneration, increase barrier function, and prevent invasion by using HT29, HT115 and CaCo-2 cell lines in vitro (McCann et al., 2007). Against HepG2 human liver cancer cells, MCF-7 human breast cancer cells, and Caco-2 human colon cancer cells, most triterpenoids extracted from apple peels demonstrated highly potent antiproliferative activity (He & Liu, 2008).

Balasuriya and Rupasinghe (2012) analysed that flavonoid rich apple peel inhibits angiotensin converting enzyme (ACE) which is helpful in treatment of high blood pressure. EtOH-AcOEt Extract of apple peel possesses ursolic acid which is responsible for anti-inflammatory activity and used in treatment of arthritis (Pádua et al., 2014). Hashemi and Yousefzadeh (2015) stated that extract of apple peels possesses triterpenoids (Ursolic acid and Oleanoic acid), out of which Oleanoic acid shows powerful anti-cancer activity on human B cell lymphoma. Activated carbon is prepared from apple peels, which is used in wastewater treatment (Enniya et al., 2018). Along with valuable source of antioxidant property, phenolic containing extract of apple peel also shows cyto-protective effect in Insulin-producing RINm5F cells (Yassin et al., 2018). In a hyperlipidemic rat model, the apple peels extract could operate as an antihyperlipidemic agent by lowering LDL and raising HDL levels (Susilowati et al., 2020) (Fig. 7).

## 4 Anti-pathogenicity

### 4.1 Pomegranate peel

Worm-borne illnesses are often persistent and frequently have higher mortality rates. Helminth infections caused by soil-transmitted helminths (STH) affect many people in developing countries, especially in rural areas. Pomegranate peel extract possesses

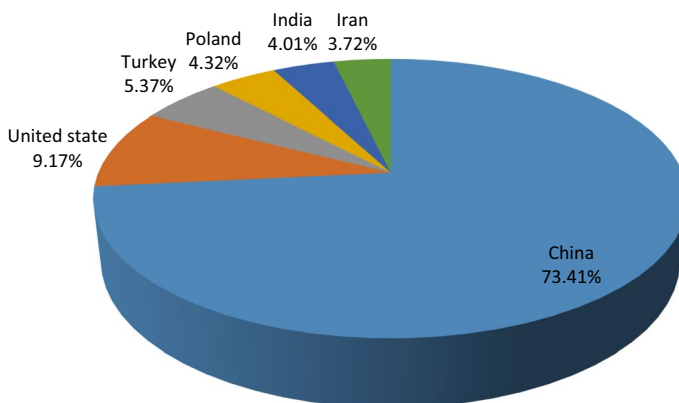


Fig. 7 Apple fruit production across the world in 2020

anthelmintic activity against female *Ascaris suum* (Amelia et al., 2017) under in-vitro conditions, earth worm (*Allolobophora caliginosa*) at the concentration of 300 mg/mL and shows anticoccidial activity against *Eimeria papillata* (Dkhil, 2013). One of the most prevalent gastrointestinal nematode, *Ascaridia galli*, affects a wide spectrum of domestic and wild birds around the world. The infection causes a reduction in body mass. Aqueous extract of pomegranate fruit peels shows anthelmintic property against *Ascaridia galli* (Aziz et al., 2018). Apart from these qualities, pomegranate fruit peel also contains antibacterial and insecticidal capabilities (Tables 3 and 4).

## 4.2 Citrus peel

Insect-borne diseases are still a major cause of sickness and death across the world. Ethanolic extract of *Citrus sinensis* exhibits insecticidal properties against the first instar larvae of *Anopheles stephensi*, *Aedes aegypti* and *Culex quinquefasciatus* with LC<sub>50</sub> values 182.4, 204.87 and 244.70 ppm, respectively (Murugan et al., 2012). Silver nanoparticles of *Citrus limetta* peel aqueous extract exhibits antimicrobial properties against five bacteria viz. *Micrococcus luteus*, *Staphylococcus aureus*, *S. epidermidis*, *Streptococcus mutans* and *Escherichia coli* with IC<sub>50</sub> values 4.28, 4.28, 4.28, 4.28 and 4.28 µg/ml, respectively and four yeast fungi, viz. *Candida albicans*, *C. glabrata*, *C. parapsilosis* and *C. tropicalis* with IC<sub>50</sub> values 4.75, 5.94, 4.75 and 5.94 µg/ml, accordingly (Dutta et al., 2020). In addition to these properties, *Citrus* fruit peel extracts have antibacterial properties, as demonstrated in Tables 5.

## 4.3 Mango peel

Dorta et al. (2016) discovered antibiotic action against significant yeast species that cause food spoilage, including *Dekkera anomala*, *Hanseniaspora uvarum*, *Zygosaccharomycetes bailii*, *Z. bisporus*, and *Z. rouxii*, as well as human infections by *Candida bracarensis*, *C. glabrata*, *C. nivariensis*, and *C. parapsilosis*. According to Xing et al. (2021), bactericidal action against *S. aureus* was more effective than against *E. coli* because silver nanoparticles prepared from mango peel extract, can sterilize by destroying cell walls, preventing enzyme respiration, and harming the DNA replication pathway. According to Table 6, mango peel extract has antimicrobial properties against a variety of bacteria, fungi, and viruses.

## 4.4 Papaya peels

Papaya is rich in vitamin A, B, and C as well as proteolytic enzymes like papain and chymopapain, which have antiviral, antifungal, and antibacterial activities (Orhue & Momoh, 2013). Aqueous extract of papaya peel shows insecticidal activity against *Aedes aegypti* (Hayatie et al., 2015). Due to the synergistic action of secondary metabolites like alkaloids, terpenoids, saponins, tannins, and flavonoids, the silver nanoparticles of green papaya peels exhibit an increase in the antibacterial activity (Agarwal et al., 2015). Papaya also has antibacterial and antiviral properties in addition to its insecticidal properties (Table 7).

**Table 3** Antimicrobial properties of pomegranate peels extract prepared by using different solvents

Extraction Solvent	Target organism	Toxicity value	References
Aqueous	<i>Staphylococcus aureus</i> , <i>Escherichia coli</i> and <i>Aspergillus niger</i>	10.7, 340.5, 17.3 µg/mL	Anesini and Perez (1993)
Aqueous	<i>Salmonella typhi</i>	73.4 µg/mL	Perez and Anesini, (1994)
Aqueous, hexane, and alcohol	<i>Bacillus subtilis</i> , <i>Proteus vulgaris</i> , <i>Salmonella typhimurium</i> , <i>Pseudomonas aeruginosa</i> , <i>Escherichia coli</i> and <i>Staphylococcus aureus</i>	–	Ahmad et al. (1998)
Ethanol	<i>Staphylococcus aureus</i>	61.5 µg/mL	Machado et al. (2002)
Acetone <sup>a</sup> , Methanol <sup>b</sup> and Aqueous <sup>c</sup>	<i>Bacillus cereus</i> , <i>Bacillus subtilis</i> , <i>B. coagulans</i> , <i>Staphylococcus aureus</i> , <i>E. coli</i> , and <i>Pseudomonas aeruginosa</i>	200, 150, 187.5, 200, 200 and 200 <sup>a</sup> ppm 250, 250, 250, 350, 500 & 250 <sup>b</sup> ppm 400, 500, 600, 450, 700 & 400 <sup>c</sup> ppm	Negi and Jayaprakasha (2003)
Aqueous and methanol	<i>Escherichia coli</i> , <i>Shigella</i> and <i>Salmonella</i>	–	Alanis et al. (2005)
Carbopol, water and triethanolamine	<i>Streptococcus mutans</i> , <i>Streptococcus sanguis</i> , <i>Streptococcus mitis</i> and <i>Candida albicans</i>	–	Vasconcelos et al. (2006)
Aqueous	<i>S. aureus</i> , <i>B. subtilis</i> , <i>E. coli</i> , <i>Ps. aeruginosa</i> and <i>P. mirabilis</i>	–	McCarrell et al. (2008)
Di ethyl ether, Methanol & Aqueous	<i>Listeria monocytogenes</i> , <i>Staphylococcus aureus</i> , <i>Bacillus subtilis</i> , <i>Escherichia coli</i> , <i>Pseudomonas aeruginosa</i> , <i>Klebsiella pneumoniae</i> , <i>Yersinia enterocolitica</i> , <i>Candida utilis</i> , <i>Saccharomyces cerevisiae</i> , <i>Aspergillus niger</i>	–	Al-Zoreky (2009)
Methanol	<i>Plasmodium</i>	4.2 µg/mL	Dell'Agli et al. (2009)
Ethanol, Methanol, Water and Imazalil	<i>Penicillium digitatum</i>	20.0, 15.0, 22.5 and 27.5 mg/L	Tayel et al. (2009)
Methanol	<i>Aspergillus niger</i> , <i>P. citrinum</i> , <i>R. oryzae</i> , <i>T. reesei</i> and <i>M. indicus</i> , <i>Staphylococcus aureus</i> , <i>Bacillus spp.</i> , <i>E. coli</i> , <i>Klebsiella pneumoniae</i> and <i>Pseudomonas aeruginosa</i>	–	Dahham et al. (2010)

Table 3 (continued)

Extraction Solvent	Target organism	Toxicity value	References
Ethanol	<i>Candida albicans</i> and <i>C. parapsilosis</i>	3.9, 3.9 µg/mL	Endo et al. (2010)
Ethanol	<i>Staphylococcus aureus</i> , <i>Escherichia coli</i> , <i>Aspergillus niger</i> and <i>Saccharomyces cerevisiae</i>	–	Ibrahim (2010)
Aqueous	<i>Staphylococcus aureus</i> , <i>Bacillus cereus</i> , <i>Pseudomonas</i>	–	Kanatt et al. (2010)
Methanol	<i>Propionibacterium acnes</i> , <i>Staphylococcus epidermidis</i> and <i>S. aureus</i>	15.6, 7.81 and 7.81 µg/mL	Panichayupakaranant et al. (2010)
Aqueous, Ethanol & Methanol	<i>Candida albicans</i>	15.0, 10.0 and 7.5 mg/L	Tayel and El-Fras (2010)
Methanol	<i>Streptococcus mutans</i> , <i>Streptococcus sanguinis</i> , <i>Streptococcus salivarius</i> , <i>Staphylococcus aureus</i> , <i>Staphylococcus epidermidis</i> , <i>Actinomyces viscosus</i> , <i>Lactobacillus actidophilus</i> and <i>Candida albicans</i>	–	Abdollahzadeh et al. (2011)
Ethanol	<i>Salmonella</i> strains- <i>S. typhi</i> , <i>S. paratyphi</i> , <i>S. enteritidis</i> , <i>S. typhimurium</i> , <i>S. dubin</i> , <i>S. derby</i> , <i>S. choleraesuis</i> , <i>S. gallinarum</i>	250, 62.5, 1000, 1000, 500, 500, 62.5, 62.5 µg/mL	Choi et al.( 2011)
Aqueous, Methanol and ethanol	<i>Staphylococcus aureus</i> , <i>Escherichia coli</i> , <i>Pseudomonas aeruginosa</i>	–	Khan & Hanee, 2011
Aqueous	<i>Staphylococcus aureus</i> , <i>Escherichia coli</i> , <i>Salmonella</i> Enteritidis, <i>A.parasiticus</i> , and <i>A.parasiticus</i>	–	Oraki et al., 2011
Ethanol and Aqueous	<i>Vibrio cholera</i> , <i>Vibrio parahaemolyticus</i> , species of <i>Shigella</i> , <i>Salmonella</i> , <i>Enteropathogenic Escherichia.coli</i> , <i>Aeromonas hydrophilla</i> , and <i>Candida albicans</i>	–	Pai et al., 2011
Aqueous and methanol	<i>S. aureus</i> , <i>P. aeruginosa</i> , <i>C. albicans</i>	–	Sadeghian et al., 2011



Table 3 (continued)

Extraction Solvent	Target organism	Toxicity value	References
Aqueous <sup>a</sup> and methanol <sup>b</sup>	<i>Penicillium italicum</i> , <i>Rhizopus stolonifer</i> and <i>Botrytis cinerea</i>	17.86, 30.29, 30.92 <sup>a</sup> 28.46, 42.29, 42.77 <sup>b</sup> ppm	Tehraniifar et al., 2011
Aqueous	<i>E. coli</i> , <i>Bacillus subtilis</i> , <i>Bacillus indicus</i> , <i>Enterobacter aerogenes</i> , <i>Serratia marcescens</i> , <i>Brucella</i> spp., <i>Staphylococcus Rhodotorula glutinis</i>	–	Yehia et al., 2011
Ethanol	<i>Trichomonas tenax</i>	–	El-Sherbini & Shoukry, 2012
Aqueous <sup>a</sup> and Methanol <sup>b</sup>	<i>E. coli</i> , <i>K. pneumoniae</i> , <i>B. subtilis</i> and <i>S. aureus</i>	> 12.50, > 12.50, > 12.50 and > 12.50 <sup>a</sup> 0.78, 0.20, 0.39 and 0.39 <sup>b</sup> mg/mL	Fawole et al., 2012
Aqueous	<i>Alternaria alternaria</i> , <i>Penicillium digitatum</i> , <i>P. expansum</i> , <i>Botrytis cineria</i> , <i>Stenophyllum botryosum</i> and <i>Fusarium</i> spp.	–	Glazer et al., 2012
Methanol	<i>Staphylococcus aureus</i> , <i>Enterococcus faecalis</i> , <i>Bacillus cereus</i> , <i>Clostridium perfringens</i> , <i>Listeria monocytogenes</i> , <i>Escherichia coli</i> , <i>Pasteurella multocida</i> , <i>Pseudomonas aeruginosa</i> , <i>Salmonella typhimurium</i> and <i>Yersinia enterocolitica</i>	15.6, 31.3, 31.3, 31.3, 31.3, 83.3, 83.3, > 62.5, 83.3 and 31.3 mg/mL	Naziri et al., 2012
Aqueous <sup>a</sup> , Ethanol <sup>b</sup> & Acetone <sup>c</sup>	<i>B. subtilis</i> , <i>S. aureus</i> , <i>E. coli</i> and <i>S. typhimurium</i> //	< 104, < 104, 207 and < 104 <sup>a</sup> 242, 242, 499 and 242 <sup>b</sup> 444, 444, 3500 and < 222 <sup>c</sup> mg/mL	Nuamsetti et al., 2012
Ethanol, Methanol and Aqueous	<i>Bacillus cereus</i> , <i>S. aureus</i> , <i>B. subtilis</i> , and <i>Escherichia coli</i>	–	Hayrapetyan et al., 2012
Methanol	<i>Candida albicans</i>	250 mg/mL	Shafiqi et al., 2012
Aqueous	<i>Pseudomonas</i>	–	Devatkal et al., 2013

Table 3 (continued)

Extraction Solvent	Target organism	Toxicity value	References
Ethanol	( <i>Escherichia coli</i> , <i>Pseudomonas aeruginosa</i> , <i>Staphylococcus aureus</i> <i>Streptococcus faecalis</i> , <i>Erwinia cartovora</i> and <i>Xanthomonas campestris</i> )	–	Hassan et al., 2013
Ethanol	<i>Escherichia coli</i> , <i>Pseudomonas aeruginosa</i> , <i>Morganella</i> sp. <i>Staphylococcus aureus</i> , and <i>Micrococcus</i> sp., <i>Candida albicans</i> , <i>Aspergillus niger</i> , <i>Alternaria alternata</i> and <i>Fusarium</i> sp., <i>Penicillium</i> sp.	–	Eldiasty et al., 2014
Hydro-alcoholic	<i>Staphylococcus aureus</i> , <i>Bacillus subtilis</i> , <i>Pseudomonas aeruginosa</i> , <i>Escherichia coli</i> , <i>Trichophyton mentagrophytes</i> , <i>T. rubrum</i> , <i>Microsporium canis</i> , and <i>M. gypseum</i>	62.5, 62.5, 250, >1000, 125, 125, 250 and 250 µg/mL	Foss et al., 2014
Ethanol, Methanol and Aqueous	<i>Staphylococcus aureus</i> , <i>Enterobacter aerogenes</i> , <i>Salmonella typhi</i> and <i>Klebsiella pneumoniae</i>	–	Malviya et al., 2014
Methanol	<i>Staphylococcus aureus</i> and <i>Escherichia coli</i>	125 and 250 ppm	Emam-Djomeh et al. (2015)
Ethanol, Methanol and Aqueous	<i>Botrytis cinerea</i> , <i>Fusarium sambucinum</i> , <i>Penicillium digitatum</i> and <i>A. niger</i>	–	Hamunda et al. (2015)
Ethanol and Aqueous	<i>Staphylococcus aureus</i>	650 and 650 µg/mL	Marchi et al., 2015
Aqueous	<i>A. Niger</i> and <i>A. flavus</i> , <i>E. coli</i> , <i>P. aeruginosa</i> , <i>S. aureus</i> and <i>S. typhi</i>	–	Bharani and Namasivayam (2016)
Methanol	<i>Fusarium sambucinum</i>	20 mg/mL	Elsherbiny et al., (2016)
Acetone <sup>a</sup> , Aqueous <sup>b</sup> , Ethanol <sup>c</sup> and Methanol <sup>d</sup>	<i>B. subtilis</i> , <i>E. coli</i> , <i>S. aureus</i> , <i>P. aeruginosa</i> , <i>S. typhimurium</i> and <i>A. niger</i>	0.25, 0.46, 0.36, 0.41, 0.66 and 0.55 <sup>a</sup> 0.43, 0.84, 0.66, 0.72, 1.30 and 1.19 <sup>b</sup> 0.35, 0.52, 0.45, 0.57, 0.83 and 0.89 <sup>c</sup> 0.31, 0.54, 0.42, 0.48, 0.78 and 0.69 <sup>d</sup> mg/mL	Ismail et al., 2016

Table 3 (continued)

Extraction Solvent	Target organism	Toxicity value	References
Hexane	<i>Penicillium digitatum</i> and <i>Penicillium italicum</i> , <i>Penicillium expansum</i> and <i>Botrytis cinerea</i>	–	Nicosia et al., 2016
Methanol	<i>Giardia lamblia</i>	–	Al-Margin (2017)
Aqueous	<i>Fusarium oxysporum</i>	–	Rongai et al., 2017
Methanol	<i>Salmonella enterica</i> , <i>Escherichia coli</i> , <i>Shigella sonnei</i> , <i>Enterococcus faecalis</i> , <i>Staphylococcus aureus</i> , <i>Bacillus subtilis</i>	171, 255, 171, 255, 182 and 171 µg/mL	Rosas-Burgos et al. (2017)
Methanol and Aqueous	<i>Staphylococcus aureus</i> , <i>Bacillus sp</i> and <i>Streptococcus pneumoniae</i> , <i>Escherichia coli</i> , <i>Klebsiella sp.</i> , <i>Shigella sp.</i> , and <i>Pseudomonas sp</i>	–	Hassan et al., 2018
Aqueous <sup>a</sup> and Methanol <sup>b</sup>	<i>Pseudomonas putida</i> , <i>Saccharomyces cerevisiae</i> and <i>Penicillium digitatum</i> ,	0.072, –; 0.180 <sup>a</sup> 0.061, 0.152 and 0.061 <sup>b</sup> g/mL	Kharchoufi et al., 2018
Aqueous	<i>Bacillus cereus</i> , <i>Staphylococcus aureus</i> , <i>Listeria monocytogenes</i> , <i>Salmonella enteritidis</i> , <i>Pseudomonas aeruginosa</i> and <i>Escherichia coli</i>	15.63, 15.63, 31.25, 62.5, 62.5 and 62.5 mg/mL	Alexandre et al., 2019
Starch film)	<i>S. aureus</i> , and <i>Salmonella</i>	–	Ali et al., 2019
Ethanol	<i>Influenza virus</i>	6.4 µg/mL	Moradi et al., 2019
Methanol	<i>Cronobacter sakazakii</i>	6.67 mg/mL	Polat Yemis et al., 2019
Aqueous, Ethanol, hydro-ethanol and Methanol	<i>F. oxysporum</i>	–	Rongai et al., 2019
Aqueous	<i>Monilinia fructigena</i> and <i>Monilinia laxa</i>	9.93 and 12.84 mg/mL	El Khetabi et al., 2020

a, b, c, d are toxicity values of respective solvents

**Table 4** Insecticidal properties of pomegranate peels extract prepared by using different solvents

Solvent	Target organism	Toxicity value	References
Ethanol	<i>Culex pipiens</i>	1253.9 ppm	Eldiasty et al., 2014
Aqueous, Ethanol & Methanol	<i>Tribolium castaneum</i>	–	Hamunda et al. 2014
Aqueous, Ethanol & Methanol	<i>Myzus persicae</i> and <i>Pthorimaea operculella</i>	–	Hamunda et al. (2015)

## 4.5 Banana peels

Chabuck et al. (2013), stated that aqueous extracts of fresh yellow banana peels could be utilized in place of synthetic medications in the treatment of bacterial illnesses. These extracts are effective against both Gram positive and negative bacteria. Bagul et al. (2017) extracted 3-carboxycoumarins compound from the waste of banana peels and used as biological agent for antibacterial, anticancer and cosmetics properties. According to Behiry et al. (2019), the methanolic extract from *Musa paradisiaca* peels demonstrated potential as a wood-biofungicide against the development of *Fusarium culmorum* and *Rhizoctonia solani*, moreover a bactericide against *Agrobacterium tumefaciens*, which might be regarded as a natural preservative for wood. Table 8 below discusses the antibacterial properties of banana peels.

## 4.6 Apple peels

Du et al. (2011) observed antibacterial activity of apple fruit skin against *Listeria monocytogenes* and *Escherichia coli*. According to Prakash et al. (2013), aqueous extract of apple peel shows the antibacterial activity against *Bacillus*, *Pseudomonas*, *Escherichia coli*, and *Klebsiella*. Lee et al. (2016) stated that organic solvent extracts of apple peel possess antimicrobial activity against *Salmonella enterica*, *Shigella flexneri*, *Listeria monocytogenes* and *Staphylococcus aureus*. Table 9 below discusses the antibacterial properties of apple peels.

# 5 Food preservative

## 5.1 Pomegranate peel

Pomegranate peel exhibits many bioactive compounds; hence, its by-products are used as preservative in food industry, it helps in preservation of many fruits such as strawberry (Zhang et al., 2010), citrus fruits (Pangallo et al., 2021) and it can also increase the shelf life of edible oils (El-Hadary and Taha, 2020). It also improves the characteristic of juices (Trigo et al., 2020) and pomegranate peel extract (PPE) added to popular chicken meat products to extend their shelf life in cold storage by 2–3 weeks, PPE also worked well in preventing oxidative degradation in these chicken products (Kanatt et al., 2010).

**Table 5** Antimicrobial properties of Citrus peels Extracts

Species of citrus	Solvent	Target organism	Toxicity value	References
<i>Citrus sinensis</i>	Aqueous	<i>Aspergillus niger</i>	15.6 µg/mL	Anesini and Perez (1993)
<i>C. sinensis</i>	Aqueous	<i>Salmonella typhi</i>	10.3 µg/mL	Perez and Anesini (1994)
<i>Citrus limon</i>	Acetone, ethanol and methanol	<i>Pseudomonas aeruginosa</i> , <i>Salmonella typhimurium</i> , and <i>Micrococcus aureus</i>	-	Dhanavade et al. (2011)
<i>Citrus sinensis</i> , <i>C. maxima</i> and <i>C. reticulata</i>	Aqueous, ethanolic and chloroform	<i>Aspergillus niger</i> , <i>Candida albicans</i> , <i>S. aureus</i> , and <i>S. pyogenes</i>	-	Mathur et al. (2011)
<i>C. sinensis</i>	Methanol	<i>Escherichia coli</i>	0.78 µg/mL	Dhiman et al. (2012)
<i>Citrus limon</i> and <i>C. limetta</i>	Aqueous	<i>Staphylococcus aureus</i> , <i>S. epidermidis</i> , <i>Streptococcus pneumoniae</i> , <i>S. pyogenes</i> , <i>S. agalactiae</i> , <i>Enterococcus faecalis</i> , <i>Escherichia coli</i> , <i>Enterobacter aerogenes</i> , <i>Klebsiella pneumoniae</i> , <i>Proteus spp.</i> , <i>Salmonella Typhi</i> , <i>Acinetobacter spp.</i> , <i>Moraxella catarrhalis</i> , <i>Pseudomonas aeruginosa</i> and <i>Candida albicans</i> (yeast)	-	Hindi and Chabuck (2013)
<i>Citrus unshiu</i>	Ethanol	<i>B. cereus</i> , <i>S. aureus</i> , and <i>L. monocytogenes</i>	20, 10 and 10 mg/mL	Min et al. (2014)

Table 5 (continued)

Species of citrus	Solvent	Target organism	Toxicity value	References
<i>C. Limon</i> and <i>C. reticulata</i>	Hexane, Ethyl Acetate, Acetone, Methanol and Water	<i>Bacillus megaterium</i> , <i>B. subtilis</i> , <i>Corynebacterium rubrum</i> , <i>Staphylococcus aureus</i> , <i>S. epidermidis</i> , <i>Citrobacter freundii</i> , <i>Enterobacter aerogenes</i> , <i>Klebsiella pneumoniae</i> , <i>Proteus mirabilis</i> , <i>Salmonella typhimurium</i> , <i>Candida albicans</i> , <i>C. glabrata</i> , <i>C. neoformans</i> and <i>C. epicola</i>	–	Rakholiya et al. (2014)
<i>C. limon</i> , <i>C. reticulata</i> and <i>C. sinensis</i>	Ethanol	<i>Bacillus subtilis</i> , <i>Enterococcus faecalis</i> , <i>Staphylococcus aureus</i> , <i>Listeria innocua</i> , <i>L. monocytogenes</i> , <i>Clostridium sporogenes</i> , <i>C. difficile</i> , <i>Acinetobacter lwoffii</i> , <i>A. junii</i> , <i>A. pittii</i> , <i>A. baumannii</i> , <i>Klebsiella spp.</i> , <i>E. coli</i> , <i>Proteus vulgaris</i> , <i>Salmonella typhimurium</i> , <i>Pseudomonas spp.</i> and <i>Cronobacter sakazakii</i>	–	Casquete et al. (2015)
<i>Citrus limetta</i>	Aqueous	<i>E. coli</i> , <i>Staphylococcus aureus</i> , <i>Sterptococcus pyogenes</i> , <i>Salmonella typhi</i> , <i>Pseudomonas aeruginosa</i> , <i>Aspergillus niger</i> , <i>Fusarium</i> and <i>Cladosporium</i>	–	Saramanda and Kaparapu (2017)
<i>Citrus clementina</i>	Aqueous	<i>Escherichia coli</i> , <i>Bacillus cereus</i> and <i>Staphylococcus aureus</i>	40.0, 40.0 and 50.0 µg/mL	Saratatale et al., 2018

Table 5 (continued)

Species of citrus	Solvent	Target organism	Toxicity value	References
<i>C. sinensis</i>	Methanol <sup>a</sup> and Aqueous <sup>b</sup>	<i>Pseudomonas aeruginosa</i> , <i>Klebsiella pneumoniae</i> , <i>Serratia marcescens</i> , <i>Escherichia coli</i> , <i>Proteus vulgaris</i> , <i>Salmonella typhi</i> , <i>Staphylococcus aureus</i> , <i>Enterococcus faecalis</i> , <i>Aeromonas hydrophila</i> , <i>Streptococcus pyogenes</i> , <i>Listeria monocytogenes</i> , <i>Lactobacillus casei</i> , <i>Aspergillus niger</i> and <i>Penicillium citrinum</i> , <i>Candida albicans</i> and <i>Saccharomyces cerevisiae</i>	350, 310, 280, 320, 290, 280, 420, 400, 410, 370, 510, 520, > 1000, > 1000, > 1000 and > 1000 <sup>a</sup> And 210, 200, 250, 270, 210, 220, 340, 260, 400, 340, 470, 510, 950, > 1000, > 1000 and 930 <sup>b</sup> µg/mL	Saleem and Saeed (2020)
<i>C. Limonia Osbeck</i>	Methanol <sup>a</sup> and Aqueous <sup>b</sup>	<i>Pseudomonas aeruginosa</i> , <i>Klebsiella pneumoniae</i> , <i>Serratia marcescens</i> , <i>Escherichia coli</i> , <i>Proteus vulgaris</i> , <i>Salmonella typhi</i> , <i>Staphylococcus aureus</i> , <i>Enterococcus faecalis</i> , <i>Aeromonas hydrophila</i> , <i>Streptococcus pyogenes</i> , <i>Listeria monocytogenes</i> , <i>Lactobacillus casei</i> , <i>Aspergillus niger</i> and <i>Penicillium citrinum</i> , <i>Candida albicans</i> and <i>Saccharomyces cerevisiae</i>	210, 140, 200, 220, 220, 250, 320, 260, 280, 210, 340, 270, 960, > 1000, > 1000 and 660 <sup>a</sup> And 160, 130, 150, 140, 130, 210, 270, 140, 280, 250, 300, 320, 900, 960, 920 and 960 <sup>b</sup> µg/mL	Saleem and Saeed (2020)

a, b are toxicity values of respective solvents

**Table 6** Antimicrobial properties of Mango peels Extracts prepared by using different solvents

Extraction solvent	Target organism	Toxicity value	References
Ethanol	<i>S. aureus</i> Strains	–	Oliveira et al. (2016)
Aqueous	<i>B. cereus</i> , <i>B. megaterium</i> , <i>B. subtilis</i> , <i>Corynebacterium rubrum</i> , <i>L. monocytogenes</i> , <i>Micrococcus flavus</i> , <i>S. albus</i> , <i>E. coli</i> , <i>Klebsiella aerogenes</i> , <i>Enterobacter aerogenes</i> , <i>Proteus mirabilis</i> , <i>Pseudomonas aeruginosa</i> , <i>Candida albicans</i> , and <i>C. neoformans</i>	–	Rakholiya et al. (2014)
Aqueous, Ethanol and Acetone	<i>E. coli</i> , <i>Salmonella typhi</i> , <i>Enterobacter Shigella</i> species and <i>Aspergillus Niger</i>	–	Thambi et al. (2016)
Methanol	<i>S. Aureus</i> , <i>E. coli</i> , <i>E. faecalis</i> , <i>B. subtilis</i> and <i>B. cereus</i>	–	Falusi et al. (2017)
Ethanol	<i>Staphylococcus aureus</i> , <i>Escherichia coli</i> , <i>Pseudomonas aerogenes</i> ,, <i>Bacillus cereus</i> and <i>Candida albicans</i>	–	El-Desoukey et al. (2020)
Ethanol	<i>Colletotrichum gloeosporioides</i> , <i>Mucor</i> , <i>Sclerotinia sclerotiorum</i> and <i>Fusarium oxysporum</i>	702.239, 757.178, 1275.7 and 930.07 ppm	Rojas et al. (2020)
Aqueous	<i>S. Aureus</i> and <i>E. coli</i>	1.557 and 2.335 mg/mL	Xing et al. (2021)



**Table 7** Antimicrobial properties of Papaya peels Extracts prepared by using different solvents

Extraction solvent	Target organism	Toxicity value	References
Ethanol	<i>Staphylococcus aureus</i> , <i>Escherichia coli</i> , <i>Pseudomonas aeruginosa</i> , <i>Proteus species</i> and <i>Enterococcus faecalis</i>	–	Akujobi et al. (2010)
Aqueous, 70% ethanol, 80% methanol and Acetone	<i>Escherichia coli</i> , <i>Staphylococcus aureus</i> , <i>Pseudomonas aeruginosa</i> , <i>Aspergillus niger</i> , <i>Candida albicans</i> and <i>Trichophyton rubrum</i>	–	Khan et al. (2012)
Water <sup>a</sup> , Ethanol <sup>b</sup> , 1% HCl <sup>c</sup> , Acetone <sup>d</sup> and Petroleum Ether <sup>e</sup>	<i>S. aureus</i> , <i>E. coli</i> and <i>Pseudomonas aeruginosa</i>	30, 30, 0.0 <sup>a</sup> 28, 28, 28 <sup>b</sup> 9.6, 9.6, 15.6 <sup>c</sup> 24, 30, 30 <sup>d</sup> 2.0, 2.0, 2.0 <sup>e</sup> mg/mL	Orhue and Momoh (2013)
Hexane, Ethyl Acetate, Acetone, Methanol and Water	<i>Bacillus megaterium</i> , <i>B. subtilis</i> , <i>Corynebacterium rubrum</i> , <i>Staphylococcus aureus</i> , <i>S. epidermidis</i> , <i>Citrobacter freundii</i> , <i>Enterobacter aerogenes</i> , <i>Klebsiella pneumoniae</i> , <i>Proteus mirabilis</i> , <i>Salmonella typhimurium</i> , <i>Candida albicans</i> , <i>C. glabrata</i> , <i>C. neoformans</i> and <i>C. epicola</i>	–	Rakholiya et al. (2014)
Methanol, Ethanol, Ethyl Acetate, Dichloromethane, n-Butanol and n-hexane	<i>Staphylococcus aureus</i> , and <i>Bacillus cereus</i> , <i>Escherichia coli</i> and <i>Pasteurellamultocida</i>	–	Asghar et al. (2016)
Aqueous and Ethanol	<i>Escherichia coli</i> , <i>Staphylococcus aureus</i> and <i>Pseudomonas aeruginosa</i>	–	Egbunu et al. (2016)
Aqueous <sup>a</sup> and Acetone <sup>b</sup>	<i>Bacillus subtilis</i> , <i>E. coli</i> , <i>Klebsiella pneumoniae</i> , <i>Staphylococcus aureus</i> and <i>Aspergillus flavus</i>	10, 10, 1.25, 1.25 and 500 mg/μL <sup>a</sup> 1.25, 1.25, 2.5, 1.25 and 500 mg/μL <sup>b</sup>	Lydia et al. (2016b)
Ethanol	<i>S. aureus</i> , <i>B. subtilis</i> , <i>P. aeruginosa</i> , and <i>E. coli</i>	–	Siddique et al. (2018)
Methanol	<i>E. coli</i> , <i>S. aureus</i> , <i>Klebsiella pneumoniae</i> and <i>Pseudomonas aeruginosa</i>	19.90, 51.94, 3.489 and 5.805 μL/mL	John et al. (2021)

a, b, c, d are toxicity values of respective solvents

**Table 8** Antimicrobial properties of Banana peels extracts prepared in different solvent

Extraction Solvent	Target organism	Toxicity value	References
Acetone	<i>Bacillus cereus</i> , <i>Salmonella enteritidis</i> , <i>Escherichia coli</i> , <i>Bacillus subtilis</i> and <i>Staphylococcus aureus</i>	250, 350, 300, 270 and 330 ppm	Mokbel and Hashimaga (2005)
Aqueous	<i>S. aureus</i> , <i>S. pyogenes</i> , <i>E. aerogenes</i> , <i>K. pneumoniae</i> , <i>E. coli</i> , <i>M. catarrhalis</i> and <i>Candida albicans</i>	–	Chabuck et al. (2013)
Ethanol <sup>a</sup> and Aqueous <sup>b</sup>	<i>Bacillus subtilis</i> , <i>Staphylococcus aureus</i> , <i>Micrococcus leutus</i> , <i>Klebsiella Pneumoniae</i> , <i>Pseudomonas aeruginosa</i> , <i>Escherichia coli</i> and <i>Salmonella typhi</i>	512.5, 512.5, 256, 32, 128, 64 and 16 <sup>a</sup> and 1025, ml, ml, 128, 512, 64 and ml <sup>b</sup> mg/mL	Ehiowemwenguan et al. (2014)
Hexane, Ethyl Acetate, Acetone, Methanol and Water	<i>Bacillus megaterium</i> , <i>B. subtilis</i> , <i>Corynebacterium rubrum</i> , <i>Staphylococcus aureus</i> , <i>S. epidermidis</i> , <i>Citrobacter freundii</i> , <i>Enterobacter aerogenes</i> , <i>Klebsiella pneumoniae</i> , <i>Proteus mirabilis</i> , <i>Salmonella typhimurium</i> , <i>Candida albicans</i> , <i>C. glabrata</i> , <i>C. neoformans</i> and <i>C. epicola</i>	–	Rakholiya et al. (2014)
Isopropyl alcohol	<i>Porphyromonas gingivalis</i> and <i>Aggregatibacter actinomycetemcomitans</i>	–	Kapadia et al. (2015)
Aqueous	<i>S. aureus</i> , <i>B. subtilis</i> , <i>P. aeruginosa</i> , <i>E. coli</i>	–	Siddique et al. (2018)
Methanol	<i>Agrobacterium tumefaciens</i> , <i>Dickeya solani</i> , <i>Erwinia amylovora</i> , <i>Pseudomonas cichorii</i> , <i>Serratia plymuthica</i> , <i>Fusarium culmorum</i> and <i>Rhizoctonia solani</i>	–	Behiry et al. (2019)
Ethyl acetate and Aqueous	<i>Pseudomonas aeruginosa</i> , <i>Klebsiella pneumoniae</i> , <i>Serratia marcescens</i> , <i>Escherichia coli</i> , <i>Proteus vulgaris</i> , <i>Salmonella typhi</i> , <i>Staphylococcus aureus</i> , <i>Enterococcus faecalis</i> , <i>Aeromonas hydrophila</i> , <i>Streptococcus pyogenes</i> , <i>Listeria monocytogenes</i> , <i>Lactobacillus casei</i> , <i>Aspergillus niger</i> and <i>Penicillium citrinum</i> , <i>Candida albicans</i> and <i>Saccharomyces cerevisiae</i>	410, 370, 500, 520, 300, 310, 790, > 1000, 670, 720, 780, 600, 980, > 1000, 540 and 670 <sup>a</sup> And 380, 350, 290, 490, 280, > 1000, 650, > 1000, 630, 590, 670, 570, 970, > 1000, 510 and 530 <sup>b</sup> µg/mL	Saleem and Saeed (2020)

a, b are toxicity values of respective solvents

**Table 9** Antimicrobial properties of Apple peels extracts prepared from different solvent

Extraction solvent	Target organism	Toxicity value	References
Acetone: Water: Acetic acid; Ethyl acetate: Methanol: Water and Ethanol: Water	<i>Escherichia coli</i> , <i>Staphylococcus aureus</i> , <i>Pseudomonas aeruginosa</i> , <i>Enterococcus faecalis</i> and <i>Listeria monocytogenes</i>	–	Alberto et al. (2006)
Styrene–divinylbenzene	<i>Helicobacter pylori</i>	31.4 µg/mL	Pastene et al. (2009)
Aqueous	<i>Listeria monocytogenes</i> and <i>Escherichia coli</i>	–	Du et al. (2011)
Ethanol	<i>Bacillus cereus</i> , <i>Escherichia coli</i>	2.50 µg/g	Freatianni et al. (2011)
Aqueous	<i>Bacillus</i> , <i>Pseudomonas</i> , <i>Escherichia coli</i> , and <i>Klebsiella</i>	–	Prakash et al. (2013)
Acetic acid, Tartaric acid, Citric acid, Propionic acid and Lactic acid	<i>Salmonella enterica</i> , <i>Shigella flexneri</i> , <i>Listeria monocytogenes</i> and <i>Staphylococcus aureus</i>	0.80, 0.80, 6.25 and 6.25 <sup>a</sup> 1.07, 1.07, 6.25 and 6.25 <sup>b</sup> 12.50 <sup>c</sup> 0.67, 0.80, 6.25 and 6.25 <sup>d</sup> 1.33, 1.33, 6.25 and 8.33 <sup>e</sup>	Lee et al. (2016)

a, b, c, d, e are toxicity values of respective solvents

## 5.2 Citrus peel

The pectin from citrus peels may be used as a natural and inexpensive binder in dosage forms. It may prove to be superior to commercially available synthetic binders (Khule et al., 2012). Pectin is mostly used in food as a gelling, thickening and stabilizing agent. The most common use is to give jams, jellies, and marmalades a jelly-like consistency when they would otherwise be sweet liquids. The presence and extraction of pectin from the sweet lime peel was confirmed by Devi et al. (2014).

## 5.3 Mango peel

Mango peel extract-containing PVA-cyclodextrin-gelatin film is a feasible solution for the creation of active food packaging that may be utilised as the main packaging material (Kanatt & Chawla, 2018). Mango waste product components are being proposed as novel sources of sustainable packaging materials, and they can also be used as active ingredients (Oliver-Simancas et al., 2021).

# 6 Production of bio-enzymes

Fruit wastes may be readily used in homes to create a crude solution of the hydrolytic enzymes, often known as bioenzymes or trash enzymes. Bio-composites are inexpensive materials that have the ability to increase plant production, improve soil texture, enrich the soil's nutritional quality, and prevent the formation of diseases.

## 6.1 Citrus peel

The production of pectinases, cellulases, and hemicellulases from *Citrus limetta* peels and endopolygalacturonase is made from *Citrus limon* peels demonstrated to be a superior substrate (Pathak et al., 2017). The eco-enzyme solutions are inexpensive since they are made from fruit waste, which is plentiful in supply. They offer significant promise for treating home wastewater as well as decreasing the burden on managing organic solid waste. Additionally, the eco-enzyme solutions created in this manner have demonstrated utility for cleaning, various residential, and agricultural uses (Patel et al., 2021).

## 6.2 Banana peels

According to Kokab et al. (2003), with the help of banana peel and solid-state fermentation (SSF), *Bacillus subtilis* synthesized alpha amylase. Many beneficial substances were isolated from banana peels (pectin, beta-cryptoxanthin, phenolic antioxidants, trans  $\alpha$  carotene, trans  $\beta$  carotene, beta-tocopherol, vitamin E, phenolic compounds, gallic acid, ketones sterols, lipids, esters, pigments, glucose, fructose, amines, and ethylene) (Pathak

et al., 2016). Banana Peels used as a substrate to produce enzymes such as  $\alpha$ -amylase, cellulase, xylanase, laccase, manganese peroxidase and lipase (Pathak et al., 2017).

## 7 Other uses

### 7.1 Pomegranate peel

Pomegranate peel extracts possess many kinds of activities such as it inhibits the melanosis formation in white shrimp (Fang et al., 2013), inhibits mild steel corrosion in hydrochloric acid conditions. The efficiency of inhibition increased as the extract concentration was raised and the effect was attributed to the chemical components found in the extract solution adsorbing on the mild steel surface (Ashassi-Sorkhabi et al., 2015). The ZnO nanoparticles prepared from fruit peel extract of pomegranates were used as reducing and stabilizing agents in synthesis process, which also exhibits antibacterial and anticancer properties (Sukri et al., 2019).

### 7.2 Citrus peel

Several bacteria (*Actinobacillus succinogenes*, *Anaerobiospirillum succiniciproducens*, *Mannheimia succiniciproducens*), yeast (*S. cerevisiae*), and fungi (*Byssoschlamys nivea*, *Aspergillus* etc.) were employed to produce succinic acid from Orange, Mausami, and Lemon Peels. Orange peel is also used to prepare citric acid and lactic acid as natural sources (Pathak et al., 2017).

### 7.3 Papaya peels

Biogas is renewable fuel which is produced by organic breakdown of agricultural, animal and food wastes. It is a mixture of gases, as methane and carbon dioxide are the major constituents. It can be used in different fields such as heating and electricity generation and used for vehicle fuel. Dahunsi et al., (2017) used the extract of papaya peel waste in successful production of biogas. Chaubey et al., (2018) reported the hydrochloric acid (HCl) extract of *Carica papaya* peel to inhibit the corrosion effect in aluminum alloy.

### 7.4 Banana peels

Banana peels are helpful to remove cadmium ions from industrial wastewater, thus they may be employed in wastewater treatment (Memon et al., 2008). The release of liquid petroleum hydrocarbon into the marine ecosystem due to human activity, causes pollution to the environment, and is called oil spill. Banana peels can be used as oil spill cleanup due to its satisfactory results on sorption capacity for gas oil (El-Din et al., 2018). In water and wastewater treatment turbidity arises due to colloidal and suspended particles, and it can be removed by coagulation–flocculation method. The aqueous extract of banana peel extract acts as natural coagulants to remove turbidity from stimulated turbid water (Deverey et al., 2019).

### 7.4.1 Extraction techniques

A useful resource to produce dietary supplements, nutritional additives, and therapeutic foods is bioactive chemicals. It has been shown that the majority of these bioactive compounds might have health-promoting properties, including anti-cancer, cardioprotective, antimicrobial, antiseptic, and anti-obesity properties. The bioactive compounds being extracted may need different extraction techniques. The separation processes can be impacted by several factors, including heat, plant components, pressure, and solvent type (Milovanovic et al., 2022). Flow-chart of extraction techniques listed in Fig. 8.

Conventional processes are referred to as usual ways since they have been in use for an extended period. The three conventional extraction techniques are found to be hydro-distillation, Soxhlet extraction, and maceration (Ali et al., 2022). Even nevertheless, these methods were initially devised just for lipid extraction, significant bioactive chemicals have long been extracted from a number of plant sections using Soxhlet extraction. As a standard technique for isolating essential oil, hydro-distillation is an established process. A common method for obtaining medicinal plants is maceration, which is typically used to herbal remedies (Khoddami et al., 2013).

Reduce the amount of organic solvent used, speed up the extraction process, and are easy to use in innovative methods of extraction. Methods for extracting bioactive ingredients that are primarily focussed on enzyme-assisted extraction, solvent extraction (SE), solid-liquid extraction, pulsed electric field (PEF), microwave-assisted extraction (MAE),

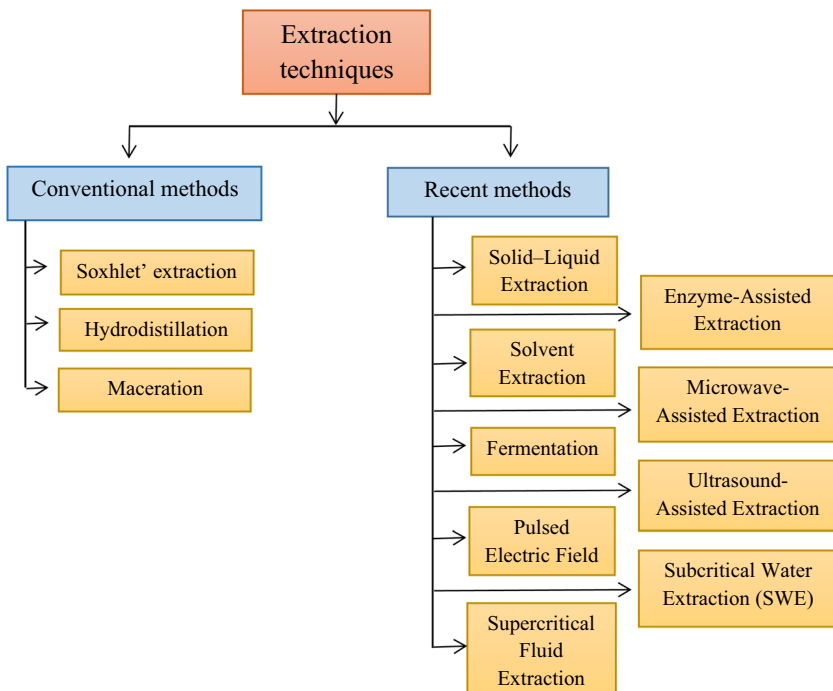


Fig. 8 Schematic representation of different techniques of extraction

subcritical water extraction (SCW), ultrasound-assisted extraction (UAE), and supercritical fluid extraction (SFE) (Ali et al., 2022). Solid–liquid extraction is one of the most widely applied techniques for phenolic chemical extraction from agricultural and culinary waste. It does, however, come with a cost, restricted output, lengthy processing periods, and the use of organic solvents, which are excellent for extracting phenolic substances but have several drawbacks of their own, including toxicity, heating, and non-biodegradability (Welton, 2015). In solvent extraction, solvents need less energy and are recyclable, non-volatile, biodegradable, and non-toxic. The selection of a solvent affects the extraction process. The best option would be a solvent with a low boiling point, rapid mass transfer, and little toxicity. Solvent extraction is a better approach than other comparable ones because of its low operating costs and ease of usage (Ali et al., 2022). Frequently, enzymes are employed in enzyme-assisted extraction to extract beneficial compounds from food waste. Because it uses water (solvent) instead of organic solvents, enzyme-aided extraction is thought to be a more environmentally benign method of extracting bioactive chemicals and oil. The foundation is in the capacity of enzymes to facilitate interactions in aqueous environments with mild processing conditions (Khoddami et al., 2013; Nadar et al., 2018). Using microorganisms to transform food waste products into useful products, fermentation is one of the first product-specific processes. Solid-state fermentation is the method of fermentation wherein microbes develop on solid substrates without encountering liquid. The process of liquefying or fermenting a medium in an environment of water is known as submerged fermentation (Sadh et al., 2018). Fruit wastes and remains may presently be transformed into valuable chemicals using a new and creative method called pulsed electric field (PEF). To achieve this remarkable separation method, a material is "sandwiched" between two electrodes and exposed to high-voltage microsecond pulses. PEF is a wonderful way to separate compounds that are heat-sensitive (Barba et al., 2015). Heat is produced inside the substance during due to the fact that microwave-assisted extraction (MAE) conductivity of ions and dipolar rotation of the interior molecules. Requiring fewer organic solvents, requiring shorter time for extraction (usually less than 30 min), and producing higher extraction yields are some benefits of using MAE procedures (Khoddami et al., 2013). Liquid solvents are used to extract chemical and inorganic components derived from solid matrices more easily by using ultrasonic radiation, which has frequencies higher than 20 kHz. Like MAE, ultrasound-aided extraction (UAE) reduces the quantity of solvent and time needed for successfully extracting phenolic chemicals from agri-food residues (Ali et al., 2022). A supercritical fluid extraction is a kind of solvent that develops when the fluid's temperature and pressure rise over its critical values. Moreover, compared to other procedures, there is a much-reduced chance of contaminating the sample with solvent contaminants and the deterioration of extracted chemicals may be prevented in the absence of light and air (Khoddami et al., 2013). Subcritical water extraction (SCWE) is additional eco-friendly extraction techniques that has been effectively used to separate phenolic chemicals. It has been shown that SCWE treatment is strong enough to remove a variety of polar to low-polar molecules. For the extraction of significant volumes of phenolic chemicals without leaving harmful organic solvent residues, SCWE may be a useful substitute industrial technique (Khoddami et al., 2013).

#### 7.4.2 Risk assessment study

Another aspect of this review is the risk assessment study of these fruit peels. Pomegranate fruit peels do not significantly alter the body mass, haematological, cytotoxic, or

genotoxic characteristics of Swiss mice (Vale et al., 2020). The levels of albumin, total protein, alkaline phosphatase (ALK-P), aspartate amino transferase (AST), and alanine amino transferase (ALT) remained largely unchanged when Wistar rats were compared to the normal control group using aqueous extracts of banana fruit peels (Edenta et al., 2017). Like this, Ehigiator et al. (2018) conducted the study demonstrating the negligible impact of banana peel extracts on the liver, kidney, or serum lipids in Sprague Dawley rats. The parameters included in this study were organ weight, serum alkaline phosphatase, aminotransferases, total cholesterol, conjugated bilirubin, total bilirubin, urea, uric acid, creatinine, glucose, low density lipoprotein, high density lipoprotein, and triglycerides total protein, albumin, and serum electrolytes. According to Organisation for Economic Co-operation and Development (OECD) Guidelines 423, Reddy et al. (2020) conducted an oral acute toxicity analysis in Alzheimer's Wistar albino rats and observed no significant changes in behavioural parameters or unfavourable effects from a single dosage of 2000 mg/kg papaya peel extract. Thus, turning fruit leftovers (peel) into necessary materials or goods might not only open new research avenues but also assist mitigate existing environmental problems.

### 7.4.3 Uncertainties affecting the scalability and commercialization of fruit peels

The scalability of common fruit peels waste management using their pharmaceutical and insecticidal attributes can be significantly affected by various uncertainties. Here is a breakdown of the key uncertainties and their potential impacts:

#### a) Resource availability and variability

Fluctuations in fruit production, seasonality, and pest infestations can impact the consistent availability of fruit peels in enormous quantities. Variations in peel quality (e.g., ripeness, pesticide residues) may influence how well bioactive chemicals are extracted and how effective they are. Extraction of bioactive substances in large quantities might require dedicated land for peel drying or composting, potentially competing with agricultural or other land uses. Water scarcity in certain regions could limit the feasibility of washing and processing copious quantities of peels.

#### b) Technical uncertainties

Efficient and scalable techniques used to extract bioactive substances from fruit peels are crucial for economic viability. Scaling up lab-scale processes to industrial levels might encounter unforeseen challenges, leading to reduced yields or increased processing costs. Formulating bioactive extracts into stable and effective pharmaceuticals or insecticides can be challenging. Uncertainties regarding shelf life, compatibility with other ingredients, and potential deterioration in different environmental circumstances can affect product marketability and long-term efficacy.

#### c) Regulatory and market uncertainties

Extensive safety and efficacy testing is required for commercializing pharmaceuticals and insecticides derived from fruit peels. Regulatory hurdles and lengthy approval processes can significantly delay or even prevent market entry.

Public perception and acceptance of natural products generated from waste items such as fruit peels can be uncertain. The market for these goods might be in need of be lower than anticipated, affecting the economic viability of scaling up waste management efforts.

#### d) Environmental and sustainability uncertainties

The environmental impact of large-scale peel processing needs careful consideration. Increased water and energy consumption, potential generation of hazardous



waste during extraction, and transportation and storage-related greenhouse gas emissions are all aspects that can undermine the sustainability of the waste management approach. Fruit peel waste management using their bioactive properties might compete with existing practices like composting or anaerobic digestion. Careful evaluation of the environmental and economic benefits compared to existing options is crucial to ensure optimal waste management strategies.

Overall, addressing these uncertainties through comprehensive research, pilot projects, and robust economic and environmental assessments is essential for realizing the full potential of scaling up common fruit peels waste management using their valuable pharmaceutical and insecticidal attributes. By proactively addressing these uncertainties, researchers, entrepreneurs, and policymakers can contribute to developing sustainable and scalable solutions for fruit peel waste management, while simultaneously unlocking the potential of these natural resources for pharmaceutical and insecticidal applications.

## 8 Conclusion

India generates a staggering quantity of garbage every day, with fruit peel trash accounting for twenty to fifty percent of the waste generated by the fruit processing industry. These days, it's rare to find opportunities to recover useful components from fruit peel waste that able to be employed as a feedstock. The initial components and all processing stages possess an effect on the waste's composition, amount, and quality. Increased sustainability in the use of all resources (such as water, fertilisers, preserving soil health, and effective land management) would result from more thorough and efficient use of biomass related to food production. The findings of this review show that peel extracts have various therapeutic effects in addition to being high in flavonoids and polyphenols. These fruit peel extracts are quite effective in treating ailments, especially considering their plant origin and minimal environmental impact. Therefore, it makes sense to suggest turning fruit peels—which are now underutilized—into more commercially viable raw materials in order to make greater use of them. A byproduct called peel extract may be added to dietary supplements and pharmaceutical products. A recent report found that extracts from pomegranate peels have the strongest antibacterial properties of any frequently consumed fruit. The present review, therefore, demonstrated that fruit peels have a great deal of promise to act as a prototype for future antibacterial therapies that are safer, more effective, and better. Moreover, these natural fruit-based medicines can be disposed of safely than conventional drugs. It also increases the food items' shelf life as well as is a highly nutritious supplement. Apple pomace has a great deal of potential for use as a source of novel, strong, safe, and superior antioxidant and antibacterial chemicals, according to the findings of numerous fruit peel studies. Furthermore, there is still a lot of uncharted territory to be covered and the majority of action mechanisms that need to be explained, despite the fact that many of study has been done in previously, few years regarding the possible health advantages of chemicals that have been reported. Considering the current review, fruit peels have a considerable therapeutic potential and provide a number of health advantages for both humans and animals.

Therefore, the potential uses of fruit peel extracts as an inexpensive and abundant source of phenolic chemicals depend on clearly defined goals for future research: (a) the creation of inexpensive, effective techniques for phenolic compound recovery; (b) the potential application of phenolic compounds as beneficial ingredients in food or

medicine; (c) the production of valuable goods; and (d) a decrease in environmental pollution. Fruit peels are suggested for use in a number of disciplines. An overview of sophisticated extraction methods for isolating and purifying phenolics from plant-based sources was given in this paper, along with outlines on certain sophisticated approaches for plant phenolic separation and identification. Subcritical water extraction has a number of benefits over conventional extraction techniques, including less expensive extracting agents, quicker extraction times, higher-quality extract, and a more environmental friendly process. The administration of fruit peel extracts in this particular scenario did not result in significant organ damage; nevertheless, the observed changes might be attributed to the extracts' low toxicity when used over an extended period of time. In addition to helping to develop scalable and sustainable methods for managing fruit peel waste, researchers, entrepreneurs, and legislators can unlock the potential of these natural resources for use in pharmaceutical and insecticidal applications by taking proactive steps to address these uncertainties.

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**Data availability** The data will be made available on the request.

## Declarations

**Conflict of interest** There is no conflict of interest to the authors.

**Involvement of human/animal in the article** In this review article, there is no participation or involvement of any human/animal.

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