

# Impact of city effluents on water quality of Indus River: assessment of temporal and spatial variations in the southern region of Khyber Pakhtunkhwa, Pakistan

Ilham Khan • Azim Khan • Muhammad Sohail Khan • Shabnam Zafar • Asma Hameed • Shakeel Badshah • Shafiq Ur Rehman • Hidayat Ullah • Ghazala Yasmeen

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Abstract The impact of city effluents on water quality of Indus River was assessed in the southern region of Khyber Pakhtunkhwa, Pakistan. Water samples were collected in dry (*DS*) and wet (*WS*) seasons from seven sampling zones along Indus River and the physical, bacteriological, and chemical parameters determining water quality were quantified. There were marked temporal and spatial variations in the water quality of Indus River. The magnitude of pollution was high in *WS* compared with *DS*. The quality of water varied across the sampling zones, and it greatly depended upon the nature of effluents entering the river. Water samples exceeded the WHO permissible limits for pH, EC, TDS, TS, TSS, TH, DO, BOD, COD, total coliforms, *Escherichia coli*, Ca<sup>2+</sup>, Mg<sup>2+</sup>, NO<sub>3</sub><sup>-</sup>, and PO<sub>4</sub><sup>2-</sup>. Piper

I. Khan · A. Khan · S. Zafar · A. Hameed · H. Ullah Institute of Chemical Sciences, Gomal University, Dera Ismail Khan, Pakistan

M. S. Khan (⊠) Faculty of Agriculture, Gomal University, Dera Ismail Khan, Pakistan e-mail: sohail.wur@gmail.com

#### S. Badshah

National Water Quality Laboratory, PCRWR, Islamabad, Pakistan

# S. U. Rehman

National Water Quality Laboratory, PCRWR, Dera Ismail Khan, Pakistan

#### G. Yasmeen

Institute of Chemical Sciences, Bahauddin Zakariya University, Multan, Pakistan

analysis indicated that water across the seven sampling zones along Indus River was alkaline in nature. Correlation analyses indicated that EC, TDS, TS, TH, DO, BOD, and COD may be considered as key physical parameters, while Na<sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, Cl<sup>-</sup>, F<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, PO<sub>4</sub><sup>2-</sup>, and SO<sub>4</sub><sup>2-</sup> as key chemical parameters determining water quality, because they were strongly correlated (r > 0.70) with most of the parameters studied. Cluster analysis indicated that discharge point at Shami Road is the major source of pollution impairing water quality of Indus River. Wastewater treatment plants must be installed at all discharge points along Indus River for protecting the quality of water of this rich freshwater resource in Pakistan.

**Keywords** City effluents · Freshwater · Indus River · Pakistan · Pollution · Spatio-temporal variations · Water quality · Water chemistry

#### Introduction

River and streams, key sources of freshwater, are continuously being polluted by three major sources of water pollution—agriculture, human settlements, and industries (FAO 2017). This situation has been further aggravated as a result of rapid population growth, increased urbanization and the release of the municipal, domestic, and industrial wastewater into freshwater sources (Iqbal 2013; Khatri and Tyagi 2015). Globally, 80% of untreated wastewater from municipal and industry sources finds its way into freshwater bodies every year (WWAP 2017). According to the United Nations (UN), about 1500 km<sup>3</sup> wastewater is produced annually, which is six times more water than exists in all the rivers of the world (WWAP 2003). Moreover, the untreated wastewater discharges entering into water bodies from various toxic sources have been shown to affect aquatic life depending on their chemical specificity, toxicity, bioavailability, and uptake by organisms (Bukola et al. 2015). Furthermore, key water resources are depleting at a threatening rate in many parts of the world mainly due to increase in global temperatures and uncertainties in weather projections (Foster et al. 1999).

Pakistan is a prime example of a country where freshwater resources are sharply diminishing, in quantity and quality, due to low precipitation/higher evaporation caused by climate change and the heavy influx of pollutants (Ahmed and Ali 2000). According to various studies, the water quality of major cities in Pakistan is deteriorating at an alarming rate because of uncontrolled disposal of municipal wastewater into freshwater resources (Bhutta et al. 2002). A study by Shafiq et al. (2011) indicated that large influx of wastewater in the form of human excreta has been added to Indus River contaminating it and leading to spread of various fatal diseases. Similarly, a study by Iqbal et al. (2006) showed that water of Kallar Kahar Lake in Punjab Province, Pakistan, was seriously deteriorating on account of accumulation of different toxic elements above the threshold limits from continuous influxes of wastewater. A study by Shah et al. (2012) revealed existence of high content of cation and trace metals in the surface water (i.e., rivers and streams) of the southern region of the Khyber Pakhtunkhwa (KPK) province. Mohsin et al. (2013) assessed the quality of ground water in Bahawalpur City of Pakistan. Their results indicated significant contamination in ground water with high values of physical-chemical characteristics than the permissible limits. In a more recent study, Zafar et al. (2017) indicated that irrigation water in the southern regions of KPK was not fit for agricultural production on account of accumulation of toxic effluents in the water system.

Water pollution in Pakistan is a serious issue that needs proper monitoring and assessment studies. A significant fraction of the population living in rural areas use natural water sources for their daily water needs both for human and livestock. Direct or indirect usage of polluted water is a main cause of the spread of skin diseases, eye infections, hair loss, etc. (Westcot 1997). Every year, the number of patients with water-borne diseases is increasing at disturbing rates; this may be linked to unsafe drinking water as well as unhealthy water environments due to improper domestic waste disposal (Hashmi and Shahab 1999; Jurate and Sillanpä 2006). According to World Health Organization (WHO 2000), more than 80% of diseases are water-borne in humans. About 40% of deaths are linked to water-borne diseases in Pakistan as reported by Pakistan National Conservation Strategy (SDPI 1993).

The Indus River is a major river with a total length of 3180 km making it one of longest rivers in Asia. It *flows through* China (western Tibet), India (Jammu and Kashmir), and Pakistan. It can be rightly called the "life line" of Pakistan as it accounts for > 80% of Pakistan's agricultural production (Meadows 1999; Rasheed 2013). After *originating* in the highlands of Kailash *Mountains* of Tibetan Plateau, it *runs* from *north* to south *through* the entire length of the Pakistan and finally empties into the Arabian Sea. The Indus River and its tributaries are flanked by one of the world's most fertile and populous regions of Pakistan. Therefore, ensuring the bio-safety of this crucial freshwater resource require regular monitoring of water quality.

The physical, bacteriological, and chemical parameters of water play a significant role in classifying and assessing water quality. However, studies aimed at monitoring the water quality as well as assessing the extent of pollution of freshwater sources in Pakistan particularly the Indus River are scarce (Shafiq et al. 2011; Shah et al. 2012). The present study is, therefore, a part of a comprehensive effort conducted to assess the hazardous effects of city effluents entering into Indus River and to quantify the important physical, chemical, and bacteriological parameters which affect water quality of Indus River. Our approach will help determining the extent of pollution impairing water quality of Indus River. The results of this study will help in devising sustainable strategies for protecting the quality of water and aquatic biota of this crucial freshwater resource.

#### Materials and methods

#### Study area

The study was conducted in Dera Ismail Khan (D.I.Khan) areas adjoining Indus River (Fig. 1). D.I.Khan is the southernmost district of province KPK bounded by the Indus River to the east and the Suleman Mountains in the west. D.I.Khan has extreme conditions being semi-arid to subtropical with an annual rainfall ranging from 180 to



Fig. 1 Map of Dera Ismail Khan (D. I. Khan), a southern region of Khyber Pakhtunkhwa, Pakistan (indicating seven sampling zones along the Indus River. Red marks with numbers 1 through 7 indicate

300 mm (Khan et al. 2010). The total population of D.I.Khan is approximately 1.01 million (Mehdi et al. 2009) and about 80% of the population live in rural areas. The communities stretched along the Indus River extensively use its water for drinking, irrigation and other domestic purposes without any treatment. The study area

the seven sampling zones: Himmat, Kokaar, Thoyal Faazal, Shami Road, Army Camp, Ghafar Tee, and Darya Khan, respectively. This figure has been reproduced from Mehdi et al. (2009)

of the river started where the major discharges of domestic sewage and industrial effluents from sugar, soap, textile, and oil milling are the main sources of pollutants in the Indus River. In addition, there is no proper sewerage system in the whole city (UNICEF 2010) and large fluxes of pollutants in the form of municipal wastewater are

Sr. no	Sampling zone	Latitude	Longitude	Altitude (m)	Code
1	Himmat	31° 55′ 51.45″ N	70° 55′ 56.29″ E	179	Zone A
2	Kokaar	31° 54′ 22.28″ N	70° 55′ 45.78″ E	182	Zone B
3	Thoyal Faazal	31° 50′ 43.44″ N	70° 55′ 16.35″ E	170	Zone C
4	Shami Road	31° 50′ 30.91″ N	70° 55′ 27.16″ E	170	Zone D
5	Army Camp	31° 49′ 16.71″ N	70° 55′ 34.68″ E	176	Zone E
6	Ghafar Tee	31° 47′ 10.42″ N	70° 55′ 18.30″ E	168	Zone F
7	Darya Khan	31° 45′ 42.96″ N	70° 55′ 47.73″ E	169	Zone G

Table 1Description of sampling zones identified for collection of water samples from Indus River, Dera Ismail Khan (southern region of<br/>Khyber Pakhtunkhwa, Pakistan)

thrown directly into Indus River through seven main discharge points stretched along the river area in D.I.Khan (Shafiq et al. 2011). The study area was divided into seven sampling zones representing the whole system and covering the total area of 32.4 km shown in Fig. 1. Each sampling zone was assigned a code (i.e., zones A through G). Further geographic details about the sampling zones are presented in Table 1. The average distance between any two zones was 5.23 km.

# Sample collection

The weather profile of the study location can be classified into two distinct seasons—(1) winter season locally called as (*Rabi*) from October–March with mean maximum air temperature ( $24 \pm 1.9 \text{ °C}$ ) and (2) summer season locally named as (*Kharif*) from April– September with mean maximum temperatures ( $36 \pm$ 1.4 °C) (Fig. 2). Of the two seasons, *Rabi* season has a total rainfall (327.8 mm) of about 40% of the annual total with a mean rainfall of  $18.6 \pm 8.5$  mm, whereas *Kharif* season has the maximum (60%) share in the total



rainfall with a mean rainfall of  $36.1 \pm 14.7$  mm. Hence, the two seasons can be referred to as dry and wet seasons (i.e., DS and WS), respectively. The rate of inflow of toxic effluents from city municipalities and industries becomes higher in WS as compared to DS due to maximum pressure in the discharge channels directed into Indus River (Iqbal 2013). Water samples were collected in the first week of January and July 2015 representing DS and WS, respectively. Representative water samples were taken in the morning (between 07:00 to 10:00 AM) from each of the seven sampling zones in two seasons (DS and WS). Single grab samples of water were obtained from the river at distance 5 m from the point of discharge and at a depth of half a meter from the surface at each sampling zone in a ten-replicate form resulting in a total of 70 samples (n = 70) per each season (DS and WS) and a total of 140 ( $70 \times 2$ ) samples. All samples were collected in 1.5-l sterilized polypropylene bottles first washed with 5% HNO<sub>3</sub> and then rinsed with deionized water. After sample collection, bottles were tightly closed, stored at 4 °C and transported to the laboratory within 3 h for the analysis.



 Table 2
 List of physical parameters and their test methods

Parameters	Unit	Apparatus (accuracy)	Test methods
рН	_	JENWAY-3020 portable pH meter (± 0.002)	The probe of pH meter was calibrated with deionized water before taking the measurement
Electrical conductivity (EC)	$\mu S \ cm^{-1}$	inoLab Cond Level 1 portable EC meter $(\pm 0.5\%)$	The probe of EC meter was calibrated with deionized water before taking the measurement
Total dissolved solids (TDS)	mg $L^{-1}$	inoLab Cond Level 1 portable EC meter ( $\pm 0.5\%$ )	This was determined by the portable EC meter in TDS mode. The probe of EC meter was calibrated with deionized water before taking the measurement
Total solids (TS)	mg $L^{-1}$	-	APHA (1998) method 2540-B
Total suspended solids (TSS)	mg $L^{-1}$	-	This was determined by subtracting TDS from TS
Total hardness (TH)	mg $L^{-1}$	-	APHA (1998) method 2340-C
Dissolved oxygen (DO)	mg $L^{-1}$	Hanna-I98193 portable DO and BOD meter $(\pm 1.0)$	The probe of EC meter was calibrated with deionized water before taking the measurement
Biological oxygen demand (BOD)	mg $L^{-1}$	Hanna-I98193 portable DO and BOD meter $(\pm 1.0)$	The probe of EC meter was calibrated with deionized water before taking the measurement
Chemical oxygen demand (COD)	mg $L^{-1}$	-	USEPA (1999) method 410.4

Recommended standard protocols and guidelines of WHO (2011a) were used for sample collection to avoid any kind of contamination. Water samples collected were used to analyze the range of physicochemical and bacteriological parameters determining water quality.

#### Analytical procedures

Physical and chemical parameters (i.e., pH, electrical conductivity (EC), total dissolved solids (TDS), dissolved oxygen (DO), and biological dissolved oxygen

(BOD)) were determined directly at the sampling site using portable measuring devices (cf. Table 2).

Values for remaining parameters (i.e., total solids (TS), total suspended solids (TSS), total hardness (TH), chemical dissolved oxygen (COD), cations (sodium (Na<sup>+</sup>), potassium (K<sup>+</sup>), lithium (Li<sup>+</sup>), calcium (Ca<sup>2+</sup>), and magnesium (Mg<sup>2+</sup>), anions (chloride (Cl<sup>-</sup>), fluoride (F<sup>-</sup>), bicarbonate (HCO<sub>3</sub><sup>-</sup>), nitrate (NO<sub>3</sub><sup>-</sup>), phosphate (PO<sub>4</sub><sup>2-</sup>), and sulfate (SO<sub>4</sub><sup>2-</sup>), were determined in the laboratory following standard methods (APHA 1998; ASTM 1998; AOAC 2000, 2002).

Table 3 List of chemical parameters and their test methods

Parameters	Unit	Apparatus (accuracy)	Test methods
Cations			
Sodium (Na <sup>+</sup> )	${ m mg}~{ m L}^{-1}$	Corning Flamephotometer-410 $(\pm 0.5\%)$	AOAC (2000) method 969.23
Potassium (K <sup>+</sup> )	${ m mg}~{ m L}^{-1}$	Corning Flamephotometer-410 ( $\pm 0.5\%$ )	AOAC (2000) method 969.23
Lithium (Li <sup>+</sup> )	${ m mg}~{ m L}^{-1}$	Corning Flamephotometer-410 ( $\pm 0.5\%$ )	AOAC (2000) method 969.23
Calcium (Ca2+)	${ m mg}~{ m L}^{-1}$	_	EDTA titration following ASTM (1998) method D511-93A
Magnesium (Mg <sup>2+</sup> )	${ m mg}~{ m L}^{-1}$	_	APHA (1998) method 3500-Mg E
Anions			
Chloride (Cl <sup>-</sup> )	${ m mg}~{ m L}^{-1}$	_	AOAC (2002) method 4110
Fluoride (F <sup>-</sup> )	${ m mg}~{ m L}^{-1}$	_	SPADNS method of APHA (1998)
Bicarbonate (HCO3 <sup>-)</sup>	${ m mg}~{ m L}^{-1}$	_	APHA (1998) method 2320
Nitrate $(NO_3^{-})$	${ m mg}~{ m L}^{-1}$	_	APHA (1998) method 4130
Phosphate (PO <sub>4</sub> <sup>2-</sup> )	${ m mg}~{ m L}^{-1}$	_	APHA (1998) method 4500-P
Sulfate (SO <sub>4</sub> <sup>2–</sup> )	$mg \; L^{-1}$	-	APHA (1998) method 4500-E

Details of the apparatus used and standard methods followed are summarized in Tables 2 and 3.

Water samples were assessed for total coliforms and *Escherichia coli* (*E. coli*) within 24 h of sample collection. The samples were processed in a laminar flow hood using sterilized culture media (USEPA 1999). The bacterial load in 100 mL water samples was estimated by most probable number (MPN) technique (Eaton et al. 1995; Benson 1998; APHA 2005).

The basic chemistry of water samples across all the seven sampling zones were assessed by constructing a Piper diagram (Piper 1944). A Piper diagram consists of a combination of triangles and a diamond-shaped quadrilateral. The relative abundance of cations in (% mg L<sup>-1</sup>) is plotted on left cations' triangle, while the relative concentration of anions, also in % mg L<sup>-1</sup>, is simultaneously plotted on right anions' triangle and the resultant points from the cation and anion triangular plots are placed over the inner diamond-like quadrilateral structure. Water types are designated according to the zones in which these points fall on the middle quadrilateral plot. Each point represents one water type.

# Statistical analyses

Descriptive statistics were performed to study the extent of variation between two sampling seasons for various parameters. Combined analysis of variance across the seasons was performed to test the significance and extent of differences among the sampling zones. Pearson correlations were estimated across seven sampling zones and two seasons to assess the inter-relationships among the physicochemical parameters of water. Finally, a cluster analysis based on Euclidean distance matrix was performed to identify and determine spatial patterns among the seven sources of effluents (i.e., sampling zones) stretched along Indus River. All statistical analyses were performed in Genstat (Payne et al. 2009).

# Results

Assessing temporal variations in water quality of Indus River

Different seasons had a significant (P < 0.01) effect on physical, bacteriological, and chemical parameters of water. The variation (mean, standard error, minimum,

maximum values, and ranges) of different parameters across seven sampling zones for two sampling seasons (i.e., *DS* and *WS*) are summarized in Tables 4 and 5. Both seasons showed wide ranges of variation in the physicochemical characteristics and bacteriological assessment of water across the seven sampling zones. However, the extent of variation among zones was comparatively higher in the case of *WS* than *DS*. This was evident from higher ranges of values in *WS* for most of the parameters studied. The results are further elaborately discussed in following subsections.

#### Temporal variations in the physical parameters of water

The results indicated that the water samples were slightly acidic to moderately alkaline (6.38-8.72) and (6.88-8.90) in both the seasons DS and WS, respectively (Table 4). Mean pH values exceeded the permissible limit (6.5) of drinking water standards of WHO ( 2011a) in both seasons. Data further illustrated that mean value of pH in WS (8.01  $\pm$  0.06) was higher than  $DS(7.56\pm0.17)$  and indicated moderately alkaline nature of water in the wet season. Mean EC values exceeded the permissible limit (250  $\mu$ S cm<sup>-1</sup>) of drinking water standards of WHO (2011a) in both the seasons (Table 4). EC values were comparatively high in WS  $(575.00 \pm 49.14 \ \mu \text{S cm}^{-1})$  than DS  $(556.60 \pm$ 54.88  $\mu$ S cm<sup>-1</sup>). Mean values of TDS, TS, TSS, and TH in DS ranged from 171.00-660.50, 304.0-841.0, 85.50-180.60, and  $142.50-325.50 \text{ mg L}^{-1}$ , respectively (Table 4). In WS, values of these parameters ranged from 220.00-710.50, 351.00-895.00, 128.00-225.00, and 202.00–392.00 mg  $L^{-1}$ , respectively (Table 4). Values of these parameters surpassed the WHO (2011a) permissible limits in both the seasons. However, mean values were much higher in WS. Furthermore, mean values of BOD and COD exceeded the WHO (2011a) permissible limits (5 and 10 mg  $L^{-1}$ , respectively) in both the seasons (Table 4). Data further indicated increasing trends in mean DO and COD values with lower BOD values in DS. Almost opposite trends were observed in WS (Table 4).

# Temporal variations in the bacteriological parameters of water

Bacteriological analyses result revealed that water samples collected during both seasons were contaminated with coliform and *E. coli* exceeding the WHO (2011a) Environ Monit Assess (2018) 190: 267

**Table 4**Mean, minimum (Min), maximum (Max), standard error(SE), and range of physical and bacteriological parameters (acrossseven sampling zones (n = 70)) in water of Indus River, Dera

Ismail Khan (southern region of Khyber Pakhtunkhwa, Pakistan) in two sampling seasons

Parameter	Unit	Mean	$\pm$ SE	Min	Max	Range	WHO (2011a) limit
Dry season (D	S)						
pН	_	7.56	0.17	6.38	8.72	2.34	6.5
EC	$\mu S \text{ cm}^{-1}$	556.60	54.88	299.20	1163.00	864.20	250
TDS	${ m mg}~{ m L}^{-1}$	314.90	31.32	171.00	660.50	489.50	500
TS	${ m mg}~{ m L}^{-1}$	448.90	34.29	304.00	841.00	537.00	500
TSS	${ m mg}~{ m L}^{-1}$	134.00	5.40	85.50	180.60	95.10	150
TH	${ m mg}~{ m L}^{-1}$	197.10	11.35	142.50	325.50	183.00	300
DO	${ m mg}~{ m L}^{-1}$	2.492	0.044	1.998	2.687	0.689	NE
BOD	${ m mg}~{ m L}^{-1}$	33.38	2.26	23.38	58.69	35.31	5
COD	${ m mg}~{ m L}^{-1}$	72.21	2.28	45.08	82.00	36.92	10
Coliform	$(MPN \ 100 \ mL^{-1})$	53.53	19.07	2.00	255.00	253.00	2.0
E. coli	(MPN 100 mL <sup>-1</sup> )	23.13	7.33	2.00	100.00	98.00	0
Wet season (W	/S)						
pН	-	8.01	0.06	6.88	8.90	2.02	6.5
EC	$\mu S \text{ cm}^{-1}$	575.00	49.14	310.40	1182.00	871.60	250
TDS	${ m mg}~{ m L}^{-1}$	362.50	29.94	220.00	710.50	490.50	500
TS	${ m mg}~{ m L}^{-1}$	498.50	26.88	351.00	895.00	544.00	500
TSS	${ m mg}~{ m L}^{-1}$	178.30	4.13	128.00	225.00	97.00	150
TH	${ m mg}~{ m L}^{-1}$	257.80	13.24	202.00	392.00	190.00	300
DO	${ m mg}~{ m L}^{-1}$	2.130	0.034	1.900	2.416	0.516	NE
BOD	${ m mg}~{ m L}^{-1}$	36.68	3.09	26.87	62.25	35.39	5
COD	${ m mg}~{ m L}^{-1}$	58.91	3.14	48.57	86.06	37.49	10
Coliform	(MPN 100 mL <sup>-1</sup> )	74.78	27.090	2.20	399.00	396.80	2.0
E. coli	$(MPN \ 100 \ mL^{-1})$	45.94	16.340	2.00	240.00	238.00	0

permissible limits (Table 4). Comparison of both seasons indicated that mean values of total coliforms  $(74.78 \pm 27.09 \text{ MPN } 100 \text{ mL}^{-1})$  and *E. coli*  $(45.94 \pm 16.34 \text{ MPN } 100 \text{ mL}^{-1})$  were high in *WS* and relatively low in *DS* for total coliforms  $(53.53 \pm 19.07 \text{ MPN } 100 \text{ mL}^{-1})$  and *E. coli*  $(23.13 \pm 7.33 \text{ MPN } 100 \text{ mL}^{-1})$ .

# Temporal variations in the chemical parameters of water

Mean concentrations of cations (Na<sup>+</sup>, K<sup>+</sup>, Li<sup>+</sup>, Ca<sup>2+</sup>, and Mg<sup>2+</sup>) in *DS* were 38.00±6.57, 7.00±1.07,  $0.989\pm0.047$ , 46.86±2.37, and 19.74± 1.50 mg L<sup>-1</sup>, respectively (Table 5). In *WS*, values of these cations were 42.80±4.37, 7.93±1.09, 1.489 ±0.056, 50.66±2.43, and 22.74±0.93 mg L<sup>-1</sup>, respectively (Table 5). Among the cations, mean values

of Ca<sup>2+</sup> and Mg<sup>2+</sup> exceeded the WHO (2011a) permissible limits of 60 and 35 mg  $L^{-1}$ , respectively in both the seasons. Data of anions (Cl<sup>-</sup>, F<sup>-</sup>, HCO<sub>3</sub><sup>-</sup>,  $NO_3^{-}$ ,  $PO_4^{2-}$ , and  $SO_4^{2-}$ ) indicated that their mean concentrations were  $39.06 \pm 5.78$ ,  $0.241 \pm 0.043$ ,  $132.90 \pm 8.27$ ,  $3.229 \pm 0.730$ ,  $0.130 \pm 0.009$ , and  $86.29 \pm 10.70 \text{ mg L}^{-1}$ , respectively, in DS (Table 5). In WS, concentrations of these anions were  $43.43 \pm$ 5.84,  $0.275 \pm 0.051$ ,  $146.60 \pm 8.69$ ,  $4.24 \pm 0.57$ ,  $0.134 \pm 0.005$ , and  $91.12 \pm 10.91$  mg L<sup>-1</sup>, respectively (Table 5). Concentrations of anions in both seasons were below the WHO (2011a) permissible limits in both seasons, except for  $PO_4^{2-}$ , which exceeded the permissible limit of 0.1 mg L<sup>-1</sup>. It was further noted that maximum values of NO<sub>3</sub><sup>-</sup> exceeded the permissible limit (10 mg  $L^{-1}$ ) set by PSQC (2008) in both seasons (Table 5). The distribution of the

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**Table 5** Mean, minimum (Min), maximum (Max), standard error (SE), and range of different chemical parameters, i.e., cations (across seven sampling zones (n = 70)) in water of Indus River,

Dera Ismail Khan (southern region of Khyber Pakhtunkhwa, Pakistan) in two sampling seasons

Parameter	Unit	Mean	± SE	Min	Max	Range	WHO (2011a) limit
Dry season (D	S)						
Na <sup>+</sup>	${ m mg}~{ m L}^{-1}$	38.00	6.57	10.45	109.20	98.75	200
$K^+$	mg $L^{-1}$	7.00	1.05	2.85	16.80	13.95	NE
Li <sup>+</sup>	mg $L^{-1}$	0.989	0.047	0.570	1.292	0.722	2.5
Ca <sup>2+</sup>	mg $L^{-1}$	46.86	2.37	34.20	71.40	37.20	60
Mg <sup>2+</sup>	mg $L^{-1}$	19.74	1.50	11.54	35.72	24.18	35
Cl	mg $L^{-1}$	39.06	5.78	17.10	105.00	87.90	250
$F^{-}$	mg $L^{-1}$	0.241	0.043	0.010	0.693	0.683	1.5
$HCO_3^-$	mg $L^{-1}$	132.90	8.27	95.00	231.00	136.00	NE
$NO_3^{-}$	mg $L^{-1}$	3.229	0.730	0.921	11.280	10.360	10 <sup>a</sup> , 50
$PO_4^{2-}$	mg $L^{-1}$	0.130	0.009	0.088	0.235	0.147	0.1
$\mathrm{SO_4}^{2-}$	${ m mg}~{ m L}^{-1}$	86.29	10.70	32.30	178.50	146.20	250
Wet season (W	VS)						
Na <sup>+</sup>	mg $L^{-1}$	42.80	4.37	14.45	115.00	100.55	200
$K^+$	mg $L^{-1}$	7.93	1.09	3.65	18.10	14.45	NE
Li <sup>+</sup>	${ m mg}~{ m L}^{-1}$	1.489	0.056	0.870	1.930	1.060	2.5
Ca <sup>2+</sup>	${ m mg}~{ m L}^{-1}$	50.66	2.43	37.60	75.30	37.70	60
Mg <sup>2+</sup>	mg $L^{-1}$	22.74	0.93	14.00	38.82	24.82	35
Cl	mg $L^{-1}$	43.43	5.84	21.20	109.30	88.10	250
$F^{-}$	mg $L^{-1}$	0.275	0.051	0.030	0.744	0.715	1.5
$HCO_3^-$	mg $L^{-1}$	146.60	8.69	107.70	246.20	138.50	NE
$NO_3^-$	mg $L^{-1}$	4.24	0.57	1.60	12.02	10.42	10 <sup>a</sup> , 50
$PO_4^{2-}$	${ m mg}~{ m L}^{-1}$	0.134	0.005	0.095	0.246	0.151	0.1
$\mathrm{SO_4}^{2-}$	mg $L^{-1}$	91.12	10.91	36.00	184.32	148.32	250

NE not established

<sup>a</sup> Permissible limit set by PSQCA (2008)

concentration of cations and anions in Indus River was observed in following increasing order:  $Ca^{2+} > Na^+ > Mg^{2+} > K^+ > Li^+$  and  $HCO_3^- > SO_4^{2-} > C\Gamma > NO_3^- > F^- > PO_4^{2-}$ , respectively, during both seasons.

Assessing spatial variations in water quality of Indus River

A preliminary analysis of the data indicated that trends in water quality determining parameters across sampling zones did not change between the two seasons. Therefore, combined analysis of variance across the seasons was performed to test the significance and extent of differences among the sampling zones. Results indicated that different sampling zones had a significant (P < 0.01) effect on the mean values of physical, bacteriological and chemical parameters studied. The results are elaborately discussed in following subsections.

# Spatial variations in the physical parameters of water

Mean pH was high (8.69) in sampling zone G in Darya Khan area followed by pH value of 8.63 recorded in zone F (Ghafar Tee) (Table 6). Minimum pH value (6.98) was recorded in sampling zone E (Army Camp). Data further illustrated that water samples collected from all the seven zones (A to G) exceeded the permissible pH limit (6.5–8.5) of WHO (2011a) (Table 6). The pH in most of the sampling zones was above 7 (i.e., alkaline). EC was

Table 6EstimatIsmail Khan (sou	ed mean (across 1 thern region of k	two samplin Chyber Pakt	ig seasons (n = itunkhwa, Pak	= 20)) values cistan)	of different p	ohysical and	bacteriologic	al parameter	s of water fo	r each indivi	dual sampling zone c	of Indus River, Dera
Sampling area	Area code	Hq	EC (µS cm <sup>-1</sup> )	$TDS \pmod{1}$	$\mathop{\rm TS}_{({\rm mg} \ L^{-1})}$	$TSS \ (mg \ L^{-1})$	$\begin{array}{c} TH \\ (\text{mg } L^{-1}) \end{array}$	$\begin{array}{c} DO \\ (mg \ L^{-1}) \end{array}$	$\begin{array}{c} BOD \\ (mg \ L^{-1}) \end{array}$	$\begin{array}{c} \text{COD} \\ (\text{mg } L^{-1}) \end{array}$	Coliform (MPN 100 mL <sup>-1</sup> )	E. coli (MPN 100 mL <sup>-1</sup> )
Himmat	Zone A	7.39 d	579.2 b	347.2 b	464.8 c	142.2 e	240.3 b	2.47 c	34.99 b	58.05 b	31.17 c	13.50 c
Kokaar	Zone B	7.72 b	394.0 d	238.8 c	373.8 e	156.2 d	180.3 g	2.59 b	27.93 c	50.16 c	3.20 e	2.10 d
Thoya Faazal	Zone C	7.54 cd	559.2 c	343.8 b	494.8 b	172.2 b	230.3 c	2.46 c	35.78 b	57.23 b	8.42 d	12.23 c
Shami Road	Zone D	7.55 cd	1117.4 a	652.8 a	825.8 a	194.2 a	340.3 a	2.01 d	58.36 a	80.60 a	305.67 a	157.67 a
Army Camp	Zone E	6.98 e	407.1 d	243.8 c	376.8 e	154.2 d	200.3 e	2.61 b	28.25 c	50.49 c	89.50b	51.00 b
Ghafar Tee	Zone F	8.63 a	324.0 e	203.8 d	344.8 f	162.2 c	190.3 f	2.64 a	25.67 d	47.91 d	2.60 e	2.20 d
Darya Khan	Zone G	8.69 a	590.0 b	346.0 b	434.8 d	112.2 f	210.3 d	2.45 c	35.54 b	58.14 b	8.78 d	3.07 d
LSD ( $P < 0.05$ )		0.15	15.07	4.60	10.17	2.94	4.82	0.015	0.793	0.821	1.168	7.03
WHO (2011a) pe	rmissible limit	6.5	250	500	500	150	300	NE	5	10	2.0	0
Means within a r	ow followed by	different lett	ters are signifi	cantly differe	ent accordin	g to Fisher's	least signific	cant differen	ce (LSD) Te	st $(P < 0.05)$		
NE not establishe	q											

Road). Minimum EC value (324.0  $\mu$ S cm<sup>-1</sup>) was recorded in water samples taken from sampling zone F (Ghafar Tee). Results further indicated that EC among all the seven sampling zones was above WHO (2011a) permissible limit (250  $\mu$ S cm<sup>-1</sup>, Table 6), which may be considered alarming. TDS were maximum (652.8 mg  $L^{-1}$ ) in the water samples collected from sampling zone D (Shami Road) which was well above the permissible limit (500 mg  $L^{-1}$ ) set by WHO (2011a) (Table 6). Minimum and statistically nonsignificant TDS (238.8 and 243.8 mg  $L^{-1}$ ) were observed in water samples taken from sampling zones B (Kokaar) and E (Army Camp), respectively. TS were below WHO (2011a) permissible limit of 500 mg  $L^{-1}$  among all the seven sampling zones except zone D (Shami Road) where maximum TS  $(825.8 \text{ mg L}^{-1})$  were recorded (Table 6). Minimum TS (344.8 mg  $L^{-1}$ ) were recorded in water samples taken from sampling zone F (Ghafar Tee). TSS were maximum (194.2 mg  $L^{-1}$ ) in sampling zone D (Shami Road) among all the zones (Table 6). TSS were minimum (112.2 mg  $L^{-1}$ ) in water samples obtained from sampling zone G (Darya Khan). TSS levels surpassed the WHO (2011a) permissible limit of 150 mg  $L^{-1}$  in most sampling zones (A to F), except zone G (Darya Khan) (Table 6). TH values were high  $(340.3 \text{ mg L}^{-1})$ in sampling zone D (Shami Road), whereas lowest  $(190.3 \text{ mg L}^{-1})$  in the sampling zone F (Ghafar Tee). Water samples collected at zone D (Shami Road) exceeded the permissible TH limit (300 mg  $L^{-1}$ ) of WHO (2011a) among all the sampling zones (Table 6). Concentration of DO was high  $(2.64 \text{ mg L}^{-1})$  in sampling zone F (Ghafar Tee) (Table 6). Minimum DO concentration (2.01 mg  $L^{-1}$ ) was recorded in water samples taken from sampling zone D (Shami Road). BOD concentration was high  $(58.36 \text{ mg L}^{-1})$  in sampling zone D (Shami Road) (Table 6). Minimum BOD (25.67 mg  $L^{-1}$ ) was observed in water samples taken from sampling zone F (Ghafar Tee). Trends of COD across the sampling zones were almost similar as observed for BOD (Table 6). The maximum COD concentration (80.60 mg  $L^{-1}$ ) was recorded in sampling zone-D (Shami Road), while the lowest concentration (47.91 mg  $L^{-1}$ ) was recorded in the sampling zone F (Ghafar Tee). Mean concentrations of BOD and COD exceeded the WHO (2011a) permissible limits (5 and 10 mg  $L^{-1}$ , respectively) across the sampling zones.

high (1117.4  $\mu$ S cm<sup>-1</sup>) in sampling zone D (Shami

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# Spatial variations in the bacteriological parameters of water

The bacterial population was found to be higher than the WHO (2011a) safe limits across the sampling zones (Table 6). The population of total coliforms in water samples ranged from 2.60 to 305.67 MPN 100 mL<sup>-1</sup> across seven sampling zones (Table 6). The population of total coliforms was extremely high (305.67 MPN 100 mL<sup>-1</sup>) in water samples of zone D (Shami Road) followed by 89.50 MPN 100 mL<sup>-1</sup> recorded at zone E (Army Camp). The population of total coliforms was lowest (3.20 and 2.60 MPN 100 mL<sup>-1</sup>) in water samples of zone B (Kokaar) and zone F (Ghafar Tee), respectively. The population of E. coli varied from 2.10 to 157.67 MPN 100 mL<sup>-1</sup> across the sampling zones (Table 6). Water samples collected from zone D (Shami Road) indicated the highest population of E. coli  $(157.67 \text{ MPN } 100 \text{ mL}^{-1})$  followed by 51.00 MPN 100 mL<sup>-1</sup> recorded in water samples of zone E (Army Camp). The lowest population of E. coli was recorded in water samples collected from zone G (Darva Khan), zone F (Ghafar Tee), and zone B (Kokaar) having statistically nonsignificant values of 3.07, 2.20, and 2.10 MPN 100 mL<sup>-1</sup>, respectively. Overall, sampling zones were ranked based on their level of contamination with total coliforms as zone D > zone E > zone A > zone G > zone C > zone B > zone F. On the basis of contamination with E. coli, sampling zones were ranked as zone D > zone E > zone A > zone C > zone G > zone F > zone B.

# Spatial variations in the chemical parameters of water

Analysis of cations showed that mean concentration of Na<sup>+</sup> was maximum (106.40 mg  $L^{-1}$ ) in sampling zone D (Shami Road) and minimum (13.27 mg  $L^{-1}$ ) in the sampling zone F (Ghafar Tee). However, Na<sup>+</sup> concentrations were below the permissible limit (200 mg  $L^{-1}$ ) of WHO ( 2011a) among all the sampling zones (Table 7). K<sup>+</sup> concentration was high (16.47 mg  $L^{-1}$ ) in sampling zone D (Shami Road) and low (3.82 and 3.46 mg  $L^{-1}$ ) in sampling zone E (Army Camp) and zone B (Kokaar), respectively (Table 7). As WHO has not established the permissible limit for K<sup>+</sup> in freshwater, therefore, comparison with the permissible limit was not made. Li<sup>+</sup> content ranged from 0.83 to 1.48 mg  $L^{-1}$  (Table 7). However,  $Li^{1+}$  concentrations were below the permissible limit (2.5 mg  $L^{-1}$ ) of WHO (2011a) among the seven sampling zones (Table 7). Li<sup>+</sup> concentrations were comparatively high  $(1.48 \text{ and } 1.45 \text{ mg } \text{L}^{-1})$  in sampling zone A (Himmat) and zone E (Army Camp), while lower concentration of  $Li^+$  (0.83 mg  $L^{-1}$ ) was recorded in the sampling zone D (Shami Road). Ca<sup>2+</sup> concentration was high (69.93 mg  $L^{-1}$ ) in water samples collected from zone D (Shami Road) which exceeded the WHO (2011a) permissible limit of 60 mg  $L^{-1}$ . Ca<sup>2+</sup> concentration was minimum  $(37.34 \text{ mg L}^{-1})$  in the sampling zone F (Ghafar Tee) among all the sampling zones (Table 7). Mg<sup>2+</sup> concentration was maximum (35.52 mg  $L^{-1}$ ) in sampling zone D (Shami Road), whereas minimum  $(18.40-18.51 \text{ mg L}^{-1})$ in sampling zones A (Himmat), E (Army Camp), F (Ghafar Tee), and G (Darya Khan) (Table 7). Mg<sup>2+</sup>

Table 7	Estimated	mean	values	(across	two	sampling	seasons	(n = 20)	)) of	different	chemical	parameters	(cations)	of	water	for	each
individua	l sampling	zone o	of Indus	River, l	Dera	Ismail Kh	an (soutl	hern reg	ion o	f Khyber	Pakhtunk	hwa, Pakista	an)				

Sampling area	Area code	$\operatorname{Na}^{+}(\operatorname{mg} \operatorname{L}^{-1})$	$\mathrm{K}^{+}(\mathrm{mg}\;\mathrm{L}^{-1})$	$Li^+$ (mg $L^{-1}$ )	$\operatorname{Ca}^{2+}(\operatorname{mg} L^{-1})$	$Mg^{2+}$ (mg L <sup>-1</sup> )
Himmat	Zone A	33.25 d	6.44 c	1.48 a	57.00 b	18.45 c
Kokaar	Zone B	17.21 f	3.46 f	1.05 d	41.88 d	13.65 d
Thoya Faazal	Zone C	40.11 c	5.45 d	1.25 c	45.82 c	25.50 b
Shami Road	Zone D	106.40 a	16.47 a	0.83 e	69.93 a	35.52 a
Army Camp	Zone E	23.13 e	3.82 f	1.45 ab	42.49 d	18.51 c
Ghafar Tee	Zone F	13.27 g	4.43 e	1.26 c	37.34 e	18.49 c
Darya Khan	Zone G	48.35 b	12.47 b	1.33 bc	45.90 c	18.40 c
LSD ( $P < 0.05$ )		1.81	0.35	0.12	0.68	0.47
WHO (2011a) per	missible limit	200	NE	2.5	60	35

Means within a row followed by different letters are significantly different according to Fisher's least significant difference (LSD) test (P < 0.05)

NE not established

concentrations were below the permissible limit (35 mg  $L^{-1}$ ) of WHO (2011a) among most of the sampling zones except zone D (Shami Road) where Mg<sup>2+</sup> concentration exceeded the permissible limit (Table 7).

Results regarding the analysis of anions indicated that Cl<sup>-</sup> concentration was maximum (102.18 mg L<sup>-1</sup>) in sampling zone D (Shami Road), while minimum  $(20.18 \text{ mg L}^{-1})$  in the sampling zone F (Ghafar Tee) (Table 8). Cl<sup>-</sup> concentrations were below the permissible limit (250 mg  $L^{-1}$ ) of WHO (2011a) among the seven sampling zones. Concentration of F<sup>-</sup> was maximum in the sampling zone D (Shami Road), whereas minimum in the sampling zone A (Himmat). None of the samples exceeded the F<sup>-</sup> permissible limit  $(1.5 \text{ mg L}^{-1})$  of WHO (2011a) among the seven sampling zones (Table 8). HCO<sub>3</sub><sup>-</sup> concentration was maximum (226.10 mg  $L^{-1}$ ) in sampling zone G (Darya Khan) and minimum (106.58 mg  $L^{-1}$ ) in the sampling zone B (Kokaar) (Table 8). As WHO has not established the permissible limit for HCO<sub>3</sub><sup>-</sup> in freshwater, therefore, comparison with the permissible limit could not be made. The concentration of NO3<sup>-</sup> was maximum  $(10.99 \text{ mg } \text{L}^{-1})$  in sampling zone D (Shami Road), whereas minimum concentration of NO3<sup>-</sup> was recorded in sampling zones B (Kokaar), E (Army Camp), F (Ghafar Tee), and G (Darya Khan) which appeared statistically at par with each other with 1.52, 1.83, 1.81, and 1.62 mg  $L^{-1}$  NO<sub>3</sub><sup>-</sup>, respectively (Table 8). Among the sampling zones, NO<sub>3</sub><sup>-</sup> concentration exceeded the permissible limit of 10 mg L<sup>-1</sup> set by PSQCA (2008) in sampling zone D (Shami Road). PO<sub>4</sub><sup>2-</sup> concentration was maximum (0.228 mg L<sup>-1</sup>) in sampling zone D (Shami Road), while the minimum concentration was recorded in the sampling zone F (Ghafar Tee). Results further indicated that PO<sub>4</sub><sup>2-</sup> concentration exceeded the WHO (2011a) permissible limit (0.1 mg L<sup>-1</sup>) in almost all the seven sampling zones (Table 8). The concentration of SO<sub>4</sub><sup>2-</sup> was maximum (172.42 mg L<sup>-1</sup>) in sampling zone D (Shami Road), while minimum (40.35 mg L<sup>-1</sup>) in sampling zone F (Ghafar Tee). SO<sub>4</sub><sup>2-</sup> concentrations were below the permissible limit (250 mg L<sup>-1</sup>) of WHO (2011a) among the seven sampling zones.

#### Determination of water type of Indus River

The basic chemistry of water samples across all the seven sampling zones with respect to the presence of some key cations like Na<sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup> and Mg<sup>2+</sup> and some chief anions like HCO<sub>3</sub><sup>-</sup>, CO<sub>3</sub><sup>2-</sup>, Cl<sup>-</sup>, and SO<sub>4</sub><sup>2-</sup> was determined by constructing a Piper trilinear diagram (Fig. 3). Results indicated that the water in the majority of seven sampling zones along Indus River in D.I.Khan is alkaline in nature. Results further revealed no major change in water types temporally, indicating the impact of rainfall pattern (i.e., *DS* and *WS*) on major ion chemistry of water was negligible.

**Table 8** Estimated mean values (across two sampling seasons (n = 20)) of different chemical parameters (anions) of water for each individual sampling zone of Indus River, Dera Ismail Khan (southern region of Khyber Pakhtunkhwa, Pakistan)

Sampling area	Area code	$Cl^{-}(mg\;l^{-1})$	$F^{-}(mg\;L^{-1})$	$\mathrm{HCO_3}^{-}(\mathrm{mg}\;\mathrm{L}^{-1})$	$\mathrm{NO_3}^-(\mathrm{mg}\ \mathrm{L}^{-1})$	$PO_4^{2-} (mg L^{-1})$	$SO_4^{2-} (mg L^{-1})$
Himmat	Zone A	30.54 d	0.024 f	126.87 c	4.53 b	0.131 c	125.24 b
Kokaar	Zone B	30.55 d	0.187 d	106.58 f	1.52 d	0.102 d	54.33 e
Thoya Faazal	Zone C	34.09 c	0.344 b	136.44 b	3.85 c	0.135 b	120.13 c
Shami Road	Zone D	102.18 a	0.676 a	124.23 d	10.99 a	0.228 a	172.42 a
Army Camp	Zone E	27.00 e	0.198 d	126.09 d	1.83 d	0.105 d	71.44 d
Ghafar Tee	Zone F	20.18 f	0.143 e	129.20 c	1.81 d	0.093 e	40.35 f
Darya Khan	Zone G	44.18 b	0.225 c	226.10 a	1.62 d	0.128 c	36.40 g
LSD ( $P < 0.05$ )		1.60	0.015	2.39	0.44	0.004	2.94
WHO (2011a) per	missible limit	250	1.5	NE	10 <sup>a</sup> , 50	0.1	250

Means within a row followed by different letters are significantly different according to Fisher's least significant difference (LSD) test (P < 0.05)

NE not established

<sup>a</sup> Permissible limit set by PSQCA (2008)

Fig. 3 Piper diagram illustrating hydrochemical characterization of Indus River water in dry season (*DS*) and wet season (*WS*). Each circle and triangle indicates the mean value of cations and anions in *DS* and *WS*, respectively, of the seven sampling zones in Dera Ismail Khan (southern region of Khyber Pakhtunkhwa, Pakistan)



Assessing relationships among physicochemical characteristics of Indus River water

 $Li^+$  (r = 0.10); Na<sup>+</sup> and HCO<sub>3</sub><sup>-</sup> (r = 0.18); Li<sup>+</sup> and HCO<sub>3</sub><sup>-</sup> (r = 0.30); DO and Li<sup>+</sup> (r = 0.28).

Analysis of correlation coefficients as presented in Table 9 indicated very strong positive correlations (r > 0.70) of some physical parameters (EC, TDS, TS, TH, BOD, and COD) with the rest of physicochemical parameters of water with r values ranging from 0.71 to 0.99. Among the chemical parameters, Na<sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, Cl<sup>-</sup>, F<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, PO<sub>4</sub><sup>2-</sup>, and SO<sub>4</sub><sup>2-</sup> showed very strong positive correlations (r > 0.70) with most of parameters (r = 0.73-0.99).

The results further indicated moderately strong positive correlations (0.40 < r < 0.70) between pH and HCO<sub>3</sub><sup>-</sup> (r = 0.61); TDS and TSS (r = 0.46); TS and TSS (r = 0.54); TSS and Ca<sup>2+</sup>, Mg<sup>2+</sup>, Cl<sup>-</sup>, F<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, PO<sub>4</sub><sup>2-</sup>, SO<sub>4</sub><sup>2-</sup>, BOD, and COD (r = 0.42, 0.64, 0.42, 0.55, 0.59, 0.43, 0.52, 0.53, and 0.46, respectively); K<sup>+</sup> and HCO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup> (r = 0.48); Ca<sup>2+</sup> and F<sup>-</sup> (r = 0.62); SO<sub>4</sub><sup>2-</sup> and Cl<sup>-</sup>, F<sup>-</sup> (r = 0.69, 62, respectively). There were weak positive correlations (r < 0.40) between pH and K<sup>+</sup>, Li<sup>+</sup>, DO (r = 0.24, 0.22, 0.11, respectively); EC and TSS (r = 0.37); TDS and HCO<sub>3</sub><sup>-</sup> (r = 0.10); TSS and Na<sup>+</sup>, K<sup>+</sup>, Li<sup>+</sup> (r = 0.38, 0.16, 0.23, respectively); TH and

The results indicated very strong negative correlations (r > 0.70) between DO and most of the parameters (r = 0.81-0.99). There were moderately strong negative correlations (0.40 < r < 0.70) between pH and SO<sub>4</sub><sup>2-</sup> (r = 0.52); Li<sup>+</sup> and Cl<sup>-</sup>, F<sup>-</sup> (r = 0.40, 0.44, respectively). There were weak negative correlations (r < 0.40) between pH and most of the physicochemical parameters with r values ranging from 0.04 to 0.16; Li<sup>+</sup> and most of the parameters (r = 0.14-0.36); HCO<sub>3</sub><sup>-</sup> and TS, TSS, Ca<sup>2+</sup>, Mg<sup>2+</sup>, F<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, DO with r values ranging from 0.01 to 0.38.

Determination of transfer factor(s) of pollutants impairing water quality of Indus River

Cluster analysis was used to identify and determine the key source of pollutants impairing water quality of Indus River in D.I.Khan region. Results showed that seven sampling zones can be classified into three distinct groups on the basis of their extent of variability for physical, bacteriological, and chemical characteristics of water (Fig. 4). Results indicated that sampling zones A

Iable 9 Pea Indus River,	rson's correlation Dera Ismail Khar	matrix indicating 1 Khan (southern 1	the association will region of Khyber ]	thin physical and cl Pakhtunkhwa, Pak	nemical paramete istan	rs (pool data of sev	en sampling zones	and two sampling	seasons ( $n = 140$	)) of water in
	Нd	EC	TDS	TS	TSS	HT	$\mathrm{Na}^+$	$\mathrm{K}^{+}$	Li <sup>+</sup>	Ca <sup>2+</sup>
Hq	I									
EC	-0.16									
TS	-0.11	0.99**	- 000 -							
ST	CI.0 -	0.37	0.46*	- 54*						
TH	-0.03	0.85**	0.91**	0.91**	-0.71 **	I				
Na <sup>+</sup>	-0.07	0.99**	0.99**	0.97**	0.38	0.85 **	I			
$\mathrm{K}^{+}$	0.24	0.89**	$0.88^{**}$	0.83 **	0.16	$0.74^{**}$	0.92**	I		
Li <sup>+</sup>	0.22	-0.36	-0.26	-0.30	0.23	0.10	-0.33	-0.25	I	
$Ca^{2+}$	-0.23*	$0.93^{**}$	$0.93^{**}$	$0.92^{**}$	0.42*	$0.89^{**}$	$0.88^{**}$	$0.77^{**}$	-0.14	I
${ m Mg}^{2+}$	-0.10*	$0.88^{**}$	$0.91^{**}$	$0.94^{**}$	$0.64^{**}$	$0.89^{**}$	$0.90^{**}$	$0.73^{**}$	-0.19	$0.78^{**}$
Cl	-0.07	$0.97^{**}$	$0.96^{**}$	$0.96^{**}$	0.42*	$0.83^{**}$	$0.98^{**}$	$0.89^{**}$	-0.40	$0.85^{**}$
Ч	-0.08	$0.84^{**}$	$0.84^{**}$	$0.88^{**}$	0.55*	$0.73^{**}$	$0.88^{**}$	$0.72^{**}$	-0.44	$0.62^{**}$
$HCO_{3}^{-}$	$0.61^{**}$	0.08	0.10	-0.01	-0.38*	0.05	0.18	0.48*	0.30	-0.03
$NO_{3}^{-}$	-0.22	$0.94^{**}$	$0.95^{**}$	$0.98^{**}$	0.59*	$0.91^{**}$	$0.91^{**}$	$0.74^{**}$	-0.27	$0.94^{**}$
$PO_4^{2-}$	-0.17	$0.99^{**}$	$0.99^{**}$	$0.99^{**}$	0.43	$0.87^{**}$	$0.99^{**}$	$0.86^{**}$	-0.36	$0.92^{**}$
$\mathrm{SO}_4^{2-}$	$-0.52^{**}$	$0.81^{**}$	$0.85^{**}$	$0.88^{**}$	0.52*	$0.78^{**}$	$0.74^{**}$	0.48*	-0.20	$0.87^{**}$
DO	0.11	-0.99**	$-0.99^{**}$	$-0.99^{**}$	-0.47	$-0.91^{**}$	$-0.99^{**}$	$-0.88^{**}$	0.28	-0.93 **
BOD	-0.16	$0.99^{**}$	0.99**	$0.99^{**}$	0.53*	$0.88^{**}$	$0.99^{**}$	$0.90^{**}$	-0.32	$0.99^{**}$
COD	-0.12	$0.99^{**}$	0.99**	0.99**	$0.46^{**}$	$0.90^{**}$	$0.96^{**}$	$0.83^{**}$	- 0.23	$0.90^{**}$
	$Mg^{2+}$	CI	μ	$HCO_3^-$	$NO_3^-$	$\mathrm{PO_4}^{2-}$	$\mathrm{SO_4}^{2-}$	DO	BOD	COD
11 <sup></sup>										
рн FC										
SUT										
TS										
TSS										
HT										
$Na^+$										
$\mathbf{K}^{+}$										
Li <sup>+</sup>										
Ca <sup>2+</sup>										
Mg	0									
CI	$0.86^{**}$	I								
F	0.90**	$0.90^{**}$	I							

	M_2+	Ļ	Ľ	- 0011		PO 2-	cO 2-			
	Mg	CI	Г	ncO <sub>3</sub>	NU <sub>3</sub>	$rO_4$	504	DO	BUD	n
$HCO_3^-$	-0.01	0.08	-0.03	1						
$NO_3^-$	$0.92^{**}$	$0.91^{**}$	$0.80^{**}$	-0.17	Ι					
$\mathrm{PO_4^{2^-}}$	$0.91^{**}$	$0.97^{**}$	$0.86^{**}$	0.03	$0.96^{**}$	Ι				
$\mathrm{SO_4}^{2-}$	$0.81^{**}$	$0.69^{**}$	0.62*	-0.33*	$0.90^{**}$	$0.83^{**}$	Ι			
DO	-0.91 **	$-0.97^{**}$	$-0.85^{**}$	-0.08	$-0.96^{**}$	$-0.99^{**}$	$-0.81^{**}$	Ι		
BOD	$0.98^{**}$	$0.94^{**}$	$0.89^{**}$	0.13	$0.98^{**}$	$0.99^{**}$	$0.83^{**}$	$-0.99^{**}$	I	
COD	$0.94^{**}$	0.99**	$0.80^{**}$	0.02	$0.91^{**}$	$0.99^{**}$	$0.86^{**}$	$-0.99^{**}$	$0.99^{**}$	I
*Significant	at $P \le 0.05$ , **sign	nificant at $P \leq 0.01$								

 Table 9 (continued)

(Himmat), B (Kokaar), C (Thoya Faazal), E (Army Camp), and F (Ghafar Tee) were clustered together forming "group 1." This suggests that these environments were highly correlated and cause little variation in the concentration of various physicochemical parameters determining water quality. On the other hand, sampling zones G (Darya Khan) and D (Shami Road) were not linked to either of the other zone(s) and created two separate groups viz. "group 2" and "group 3," respectively. This could mean that both these water environments are distinct in their physicochemical makeup.

#### Discussion

Results indicated marked temporal variations in physicochemical concentrations and bacteriological parameters. This was evident from the higher values of mean and ranges for physico-chemical parameters like pH, EC, TDS, TS, TSS, TH, BOD, COD, total coliforms, E. coli, cations (Na<sup>+</sup>, K<sup>+</sup>, Li<sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>), and anions  $(Cl^-, F^-, HCO_3^-, NO_3^-, PO_4^{2-}, SO_4^{2-})$  in the wet season. Results revealed that variations in these parameters may vary with fluctuations in weather conditions determining pollution load. Overall, the magnitude of pollution was high in wet season compared with the dry season. This could be due to the high influx of pollutants in the water system (Carr and Neary 2008; Memon et al. 2011) mainly due to excessive total rainfall (216.5 mm) compared with lower (111.3 mm) rainfall in dry season (Fig. 2). Data also indicated decreasing trends in mean DO and COD values with higher BOD values in wet season. This suggested that significant amounts of organic substances were being released into the water with a high oxygen demand mainly because of increased pollution load during the wet season. During the dry season, the situation improved slightly, perhaps because of low effluent discharge conditions in the water system. The deficiency of the oxygen in the water is a shelter for bacteria and other pathogens, which are anaerobic and injurious to human health (Radha et al. 2007). Almost similar results were reported by Pathak et al. (2012). The bacteriological analyses also revealed similar facts as mean values of total coliforms and E. coli were high in wet season than dry season. This could perhaps be due to the increased accumulation of municipal waste in wet season offering favorable conditions for the organisms to sustain and multiply (Shar et al. 2008). Results concluded that river water quality was highly deteriorated and contaminated



by the bacterial population which may pose a potential public health hazard (Usharani et al. 2010). Furthermore, the majority of the parameters exceeded the WHO (2011a) safe limits, which clearly indicated the poor suitability of water for drinking and/or irrigation purpose (Shah et al. 2012; Zafar et al. 2017).

There were marked spatial variations in physical, bacteriological, and chemical characteristics of water (Popa et al. 2012). Moreover, values of most of the physical, bacteriological, and chemical parameters revealed an increasing trend in the majority of the sampling zones. One particular location, zone D (Shami Road), excelled in the majority of parameters (i.e., EC, TDS, TS, TSS, TH, BOD, COD, total coliforms, E. coli, Na<sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>,  $Mg^{2+}$ , Cl<sup>-</sup>, F<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, PO<sub>4</sub><sup>2-</sup>, SO<sub>4</sub><sup>2-</sup>) which may be considered alarming. For instance, EC is directly related to the concentration of various ionized substances (anions and cations) in the water (Laluraj and Gopinath 2006). The excess of such substances in the water may cause serious health problems in the living population (Borgmann et al. 2005; Javaid et al. 2008). Estimation of TDS provides a qualitative measure of the amount of dissolved ions in water. Elevated levels of TDS may indicate a high concentration of inorganic salts principally  $Ca^{2+}$ ,  $Mg^{2+}$ ,  $K^+$ ,  $Na^+$ ,  $HCO_3^-$ ,  $Cl^-$ , and  $SO_4^{2-}$  in water (WHO 2003). Furthermore, increased TDS can result in water having a bitter or salty taste, result in incrustations and corrosion, and reduced efficiency of water filters and equipment (Oram 2012). High values of TS and TSS may indicate high extent of organic particles from decomposing materials (Murphy 2007). High TH values may indicate high concentration of dissolved Ca<sup>2+</sup> and Mg<sup>2+</sup> ions in water (Wurts and Masser 2004). Increased TH can cause lime buildup in pipes and water heaters, may decrease soap's cleaning ability and turn white laundry gray (Mechenich and Andrews 2004). Excessive concentrations of NO<sub>3</sub><sup>-</sup> in water may result from sewage disposal into freshwater sources and can be hazardous to health, especially for infants and pregnant women (WHO 2011b; Oram 2012). High concentrations of Cl<sup>-</sup>, F<sup>-</sup>,  $PO_4^{2-}$ , and  $SO_4^{2-}$  may indicate the presence of anthropogenic pollutants in wastewater discharges into Indus River (Iscen et al. 2008; Khatri and Tyagi 2015). The increased presence of these ions in natural water may cause serious environmental problems because of its contribution to the eutrophication process (Egemen 2000; Khan et al. 2007).

Our results also indicated that direct discharge of untreated domestic, municipality, and industrial wastewater into the river was responsible for the high organic pollution that led to high BOD, COD, and low DO values across the sampling zones. Low concentrations of DO may induce a negative impact on the aquatic biota (Helmer and Hespanhol 1997). Results revealed that quality of Indus River water was deteriorated by the presence of bacterial population at multiple locations. Moreover, water samples collected from all the seven zones exceeded the permissible limits of WHO (2011a) for the majority of the parameters. This is an indication of potential degradation of the water reflecting the health hazards for human use (Amin et al. 2010). Assessing the basic chemistry of water samples via Piper trilinear diagram indicated an alkaline nature of Indus River in D.I.Khan. This was also evident from the fact that pH in most of the sampling zones was above 7 (i.e., alkaline), which indicates that water is probably hard and contains  $Ca^{2+}$  and  $Mg^{2+}$  (David 2004). Increased  $HCO_3^-$  ions in water may cause high pH values as are the major contributors in alkalinity development (Mechenich and Andrews 2004). Almost similar results were found by Shah et al. (2012) and Zafar et al. (2017).

The correlation analysis among physicochemical characteristics of Indus River clearly showed that high values of pH might be caused by high concentrations of  $HCO_3^{-}$  as indicated by their strong positive correlation with each other. Furthermore, it was clearly evident that fluctuations in EC correlated positively with the majority of parameters studied (Sunitha et al. 2005). This suggested that water quality can be assessed with sufficient accuracy just by the measurement of EC alone. This could provide a means for easier and faster monitoring of water quality (Kalyanaraman and Geetha 2005). Another objective of this study was to determine the key parameters determining water quality in the river. Among the large number of parameters studied, EC, TDS, TS, TH, DO, BOD, and COD may be considered as key physical parameters, while Na<sup>+</sup>, K<sup>+</sup>,  $Ca^{2+}$ ,  $Mg^{2+}$ ,  $Cl^{-}$ ,  $F^{-}$ ,  $NO_{3}^{-}$ ,  $PO_{4}^{2-}$ , and  $SO_{4}^{2-}$  as key chemical parameters determining water quality, because they were correlated with most of the parameters (Trivedi et al. 2009). Therefore, such an analysis could be a very useful tool for predicting water quality across the environments. Overall, it was concluded that the extent of pollution in the Indus River water is strongly related to the degree of effluents entering the aquatic environment. Mahajan et al. (2005) and Zafar et al. (2017) also concluded that most of physicochemical parameters of water are correlated indicating their interdependence with each other (Chaubey and Patil 2015). The results proved to be a useful mean for rapid monitoring of water quality with the help of systematic calculations of the correlation coefficient between water quality parameters (Heydari et al. (2013).

We used cluster analysis to determine the transfer factor(s) of pollutants impairing water quality of Indus River. Results indicated that sampling zones- D (Shami Road) and G (Darya Khan) were not related in terms of their water quality. This was also evident from the fact that majority of the physical, bacteriological, and chemical parameters attained their maximum values at sampling zone D (Shami Road) followed by zone E (Army Camp), while sampling area at Darya Khan (zone G) was observed as the less polluted one. This could be related to the nature and extent of municipal wastewater accumulating into these two locations at Indus River, D.I.Khan. Zone D (Shami Road) is located in a densely populated area of the D.I.Khan city and contains the main discharge point receiving contaminants from whole city's municipal sewage. Generally, levels of measured parameters were low in sampling zone G (Darya Khan) in comparison to other zones. This could be related to the dilution effect on account of its furthest location from the rest of the effluent discharge points. This suggests that extent of hazardous accumulation of certain elements in Indus River water is controlled by both the type and continuous inflow of toxic effluents at various discharge points. These results were in agreement with the study of Tahir and Bhatti (1994) who concluded that direct discharge from untreated wastes is the main cause of high levels of dissolved and suspended solids, hardness, and other chemical and biological properties of water. There are no sewage effluents treatment plants in the whole D.I.Khan region and untreated wastewater are disposed directly into Indus River. This may pose adverse effects on the aquatic biota, besides making the water unfit for domestic and agricultural use. Cluster analysis thus determined the extent of spatial variation on the basis of variations in physical and chemical water characteristics of each sampling zone and indicated a strong association between spatial effects and water quality of Indus River.

#### Conclusions

In this study, we assessed the hazardous effects of contaminants entering into Indus River and quantified the important physical, bacteriological, and chemical parameters impacting the water quality of the river. There were marked temporal and spatial variations in the water quality of Indus River. However, the interactive impacts of spatio-temporal factors on water quality were conservative. Our results clearly indicated the accumulation of high levels of bacterial population, certain cations as well as anions in freshwaters of Indus River at most of the sampling sites which is impacting the physical characteristics of water as well. The magnitude of pollution was high in wet season compared with the dry season. Overall results suggested that the uncontrolled and continuous discharge of wastewater into the Indus River at seven discharge points has a considerable negative impact on the water quality of the river. Moreover, sewage drains at Shami Road (sampling zone D) are the major source of polluting Indus River in D.I.Khan, a southern region of Khyber Pakhtunkhwa, Pakistan. It is therefore highly recommended that the injudicious disposal of municipal wastewater into the river should be discouraged. Wastewater treatment plants must be installed at all the discharge points along Indus River in order to reduce future hazards in reference to aquatic and human life.

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