

# Examination of honeys and flowers as soil element indicators

Nikolett Czipa • Gerda Diósi • Clive Phillips • Béla Kovács

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Abstract Detection of soil element deficiencies is time consuming, requiring a major commitment for field work and analysis. Bees concentrate some elements in their honey which could allow soil element concentrations to be predicted without having to take large numbers of soil samples. We measured 14 element concentrations in soil, sunflower, acacia flower and honey samples from two different regions of Hungary. Across sites, the elements with significant correlation coefficients between honey and soil concentrations, in descending order of probability, were Cu > Ba > Sr = Ni> Zn > Mn = Pb >As. Bioconcentration from soil to honey was similar for areas with acacia and sunflower flowers. In the macroelements, it was the greatest for K, S and P and least for Mg and Na, and in the microelements, greatest for B, then Zn, then Cu, then As, Mo and Sr and least for Fe, Ba, Mn and Pb. It is concluded that in acacia and sunflower-growing regions, honey can give an accurate estimate of soil element concentrations for Cu and Ba and provides relevant information for Sr. Ni. Zn, Mn, Pb and As.

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Keywords Bioconcentration  $\cdot$  Honey  $\cdot$  Element  $\cdot$  Flower  $\cdot$  Soil

### Introduction

Honey is a natural substance produced by *Apis mellifera* from flower nectar and/or honeydew. Environmental conditions are favourable for honey production in Hungary, with 18,500 tons produced in 2013, and approximately 15,000 tons per year exported (FAOSTAT 2016). The most important flowers used for honey production in Hungary are acacia, sunflower, linden, silk grass and oilseed rape.

Honey is a complex food and its properties depend on the botanical, environmental and postharvest conditions, including storage and extraction techniques (Pohl 2009). It has a low mineral content (0.1-0.2%) in nectar honeys) that depends on the botanical origin, soil conditions and treatment, rendering it suitable as an environmental indicator (Almeida-Silva et al. 2011). Soil is the main source of both essential and non-essential elements to plants, with uptake depending on soil properties, plant type and farming method. The soils and flowers have a major influence on the mineral composition of honeys, and the mineral profile of honeys can be used to determine the floral and geographical origin of honeys (Pohl 2009; Pohl et al. 2012). Anthropogenic activities, e.g. smelting, mining, burning of fossil fuel and use of fertilizers, pesticides and transport, may also affect soil properties, which change trace element behaviour (Kabata-Pendias and Mukherjee 2007). As bees

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collect pollen from flowers in a large area, about 7 km<sup>2</sup> (Crane 1984), honey potentially gives valuable environmental information from this area. This could obviate the need to take large numbers of soil samples to identify regional element deficiencies or toxicities. Determination of element content of honeys as a bioindicator has been studied by several authors, e.g. Conti and Botrè (2000), Bratu and Georgescu (2005), Rashed et al. (2009), Pohl et al. (2012), Bastías et al. (2013), Al Naggar et al. (2013). In these works, the element content of honeys, pollens or waxes were determined; however, they did not simultaneously examine the element content of soils and flowers from the honey collecting area. Al Naggar et al. (2014) measured the Cu, Zn, Cd, Pb and Fe concentrations in soil and flower samples and determined the transfer rates of these metals from soil to cotton and clover flowers; however, there have been no studies in which the bioconcentration factors have been determined from soil to honey.

The aims of this study were (i) to determine the element content of soil, flower and honey samples; (ii) to calculate the bioconcentration factors between the flower and soil, honey and flower, honey and soil; and (iii) to determine relations between the element content of soil, flower and honey.

### Materials and methods

Sampling and sample preparation

Five-five soil, flower and honey samples were collected from five different regions of Hungary in 2015 (Table 1). Two flowers that predominate in these regions are acacia (*Robinia pseudoacacia*) and sunflower (*Helianthus annuus*). Samples of acacia flowers, the soils in which they grew and acacia honeys were collected from one area of Békés County (No. 1) and two areas of Szabolcs-Szatmár-Bereg County (Nos. 2 and 3). Samples of soil in sunflower-growing regions, sunflower flowers and sunflower honeys (Nos. 4 and 5) came from two agricultural areas of Békés County. Soil of Szabolcs-Szatmár-Bereg County (Northern Hungary) is acidic and sandy, and Békés County (East Hungary) has alluvial meadow soil. Every collecting area was free from industrial activity and traffic. The sampling of soil and flower samples was carried out during the bees' collecting time. In the case of soil samples, five samples were collected from every examined area at five randomly selected locations per hectare and from the top 15 cm of the soil. The size of sampling areas was five hectares, so the number of samples was 25 in each area. Samples were homogenized by areas and 1–1 kg of soil was used for element determination. Before the digestion, soil samples were oven-dried at 105 °C for 5 h (Memmert UF 75 Universal Oven, Memmert GmbH + Co. KG, Schwabach, Germany) and then ground.

The sampling of flowers was carried out at the same locations as those used for the sampling of soils. Flower samples were oven-dried at 60 °C for 12 h before digestion. For honey sampling, at each location, five hives were chosen randomly. Honey centrifugation from the hives was conducted separately for each collecting area, so at the end of centrifuging five honey samples were available. The sampling of honey samples (100 g) was carried out immediately after centrifuging from these five plastic barrels. In case of honey samples, the element concentrations were determined in the dry matter.

All samples were stored in sterile glass jars at room temperature before the analysis.

## Determination of the content of elements

All chemicals were analytical grade or better. Ultrapure water (18.2 M $\Omega$  cm) was used to prepare of solutions and dilutions produced by a Milli-Q water purification system (Millipore S.A.S., Molsheim, France). Nitric acid (69% v/v) and hydrogen-peroxide (30% v/v) were from VWR International Ltd. (Radnor, USA). The element standard solutions were prepared from mono-elemental standard solutions (1000 mg L<sup>-1</sup>; Scharlab S.L., Barcelona, Spain).

The digestion of samples for element analysis was carried out according to the method of Kovács et al. (1996). This method has been validated using animal and plant materials in our accredited laboratory (ISO/ IEC 17025:2005). For 3-g plant honey samples and 2-g flower samples 10 ml, and for 3-g soil samples 5 ml, of nitric acid was added, and the samples were allowed to stand overnight. In the pre-digestion phase, the samples were heated at 60 °C for 30 min (plant and honey samples) or 60 min (soil samples). After the samples

Table 1	Sample	type and	floral	and	geographical	origin
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Type of sample	Sample number	Sample name	County	Town
Soil	No. 1S	Acacia soil	Békés	Sarkadkeresztúr
	No. 2S	Acacia soil	Szabolcs-Szatmár-Bereg	Nyírlugos
	No. 3S	Acacia soil	Szabolcs-Szatmár-Bereg	Ömböly
	No. 4S	Sunflower soil	Békés	Sarkadkeresztúr
	No. 5S	Sunflower soil	Békés	Sarkad
Flower	No. 1F	Acacia flower	Békés	Sarkadkeresztúr
	No. 2F	Acacia flower	Szabolcs-Szatmár-Bereg	Nyírlugos
	No. 3F	Acacia flower	Szabolcs-Szatmár-Bereg	Ömböly
	No. 4F	Sunflower flower	Békés	Sarkadkeresztúr
	No. 5F	Sunflower flower	Békés	Sarkad
Honey	No. 1H	Acacia honey	Békés	Sarkadkeresztúr
	No. 2H	Acacia honey	Szabolcs-Szatmár-Bereg	Nyírlugos
	No. 3H	Acacia honey	Szabolcs-Szatmár-Bereg	Ömböly
	No. 4H	Sunflower honey	Békés	Sarkadkeresztúr
	No. 5H	Sunflower honey	Békés	Sarkad

had cooled, 3 ml hydrogen-peroxide (plant and honey samples) or 5 ml hydrogen-peroxide (soil samples) was added and the main digestion was carried out at 120 °C for 90 min (plant and honey samples) or 4.5 h (soil samples). After digestion, ultrapure water was added to make a final volume of 50 ml. Samples were homogenized and filtered using qualitative filter paper (Sartorius Stedim Biotech S.A., Gottingen, Germany). The concentrations of potassium, magnesium, sodium, phosphorus and sulphur were determined by Inductively Coupled Plasma Optical Emission Spectrometer (ICP-OES) (Thermo Scientific iCAP 6300, Cambridge, UK). The applied wavelengths (nm) were the following: 769.896 nm for K, 279.806 nm for Mg, 818.326 nm for Na, 213.617 nm for P and 182.563 nm for S. The determination of arsenic, barium, boron, cadmium, cobalt, chromium, copper, iron, manganese, molybdenum, nickel, lead, strontium and zinc contents was carried out using Inductively Coupled Plasma Mass Spectrometer (ICP-MS) (Thermo Scientific XSeries 2, Bremen, Germany). The measured isotopes (amu) were 75 for As, 11 for B, 137 for Ba, 111 for Cd, 59 for Co, 52 for Cr, 65 for Cu, 55 for Mn, 95 for Mo, 60 for Ni, 206 for Pb, 80 for Se, 88 for Sr and 66 for Zn. Rhodium was used as internal standard (40  $\mu g L^{-1}$ ).

The operating parameters of ICP-OES and ICP-MS are reported in Table 2. For ICP-OES, the detection limits (DL) were determined for reagent blank samples (n = 10) using the software for ICP-OES (iTEVA) at a confidence level of 99.0%: 0.525 mg kg<sup>-1</sup> for K, 0.104 mg kg<sup>-1</sup> for Mg, 0.488 mg kg<sup>-1</sup> for S. For ICP-MS, the DLs were determined by using the following equation: DL = 3\*SD<sub>reagent blank</sub> (n = 10) / sensitivity. DLs were as follows: 0.0366 µg kg<sup>-1</sup> for As, 2.74 µg kg<sup>-1</sup> for Cd, 0.008 µg kg<sup>-1</sup> for Co, 0.0375 µg kg<sup>-1</sup> for Cr, 0.789 µg kg<sup>-1</sup> for Cu, 0.09 µg kg<sup>-1</sup> for Mn, 0.0187 µg kg<sup>-1</sup> for Mo, 0.0998 µg kg<sup>-1</sup> for Sr and 2.57 µg kg<sup>-1</sup> for Zn.

#### Statistical analysis

Analytical analysis was carried out in triplicate. Data was described by using general terms (mean, standard deviation, minimum and maximum values) and independent samples T test, ANOVA. SPSS for Windows Version 13 (SPSS Inc. Chicago, Illinois, USA) was used for the calculations. Bioconcentration factors (BCF) were determined for flower/soil, honey/flower and honey/soil comparisons by using the following equations:

#### Table 2 Operating parameters of ICP-OES and ICP-MS

Parameters (ICP-OES)		Parameters (ICP-MS)	
Operating power	1350 W	Rf power	1400 W
Plasma gas flow rate	$16  \mathrm{l}  \mathrm{min}^{-1}$	Plasma gas flow rate	$14.0 \ 1 \ \mathrm{min}^{-1}$
Auxiliary gas flow rate	$1.0 \ 1 \ \mathrm{min}^{-1}$	Auxiliary gas flow rate	$1.0 \ \mathrm{l} \ \mathrm{min}^{-1}$
Nebuliser gas flow rate	$1.0 \ 1 \ \mathrm{min}^{-1}$	Nebuliser gas flow rate	$0.9 \ 1 \ \mathrm{min}^{-1}$
Rinsing time	30 s	CCT gas flow rate	$6.0 \text{ ml min}^{-1}$
Rinsing pump speed	75 rpm	Sample uptake rate	$0.5 \text{ ml min}^{-1}$
Stabilization time	5 s	CCT gas	7% H <sub>2</sub> in 93% He
Integration time		Dwell time	100 ms
Low WL range	10 s	Sweeps	9
High WL range	10 s	Main runs	3

WL wavelength

BCF (flower/soil)

element concentration of flower sample element concentration of soil sample

BCF 
$$\left( honey / flower \right)$$

= <u>element concentration of honey sample</u> <u>element concentration of flower sample</u>

= <u>element concentration of honey sample</u> element concentration of soil sample

Differences between elements were analysed by oneway analysis of variance with the statistical package Minitab, using Fisher's Pairwise comparisons test to compare means post hoc. Pearson's correlation coefficients and probabilities were calculated for flower/soil, honey/soil and honey/flower mean measurements at each of the five locations, after ascertaining that data was normally distributed by the Anderson Darling test.

## **Results and discussion**

Macro, micro and trace element content of soil, flower and honey samples

The element concentrations of examined soil, flower and honey samples are presented in Table 3. Analysing the macro element concentrations of soil samples, No. 1S sample showed the highest K. Na and S contents. The highest Mg and P concentrations were determined in No. 4S and No. 5S samples. Examining the mean macro element concentrations, K was present in the highest contents followed by Mg, P, S and Na. Examining the micro element contents, the lowest element concentrations were measured in No. 2S sample, and No. 3S sample showed similar low element contents, except for Mo that was at a high concentration compared to other samples. The highest As, B, Fe and Mn contents were determined in No. 1S sample, and No. 4S sample showed the highest Ba, Cd, Cu, Pb and Sr concentrations. For the other micro elements (Co, Cr, Ni and Zn), the highest contents were measured in the other sunflower soil sample (No. 5S). All of the soil samples contained Fe at the highest concentration and Mo and Cd were measured at the lowest contents. According to the mean micro element contents, Ba was the second most abundant element, followed by Mn, Zn, Cr, Ni, Cu, Sr and Pb. Concentrations of Co, As and B were less than 10,000 µg/kg.

Examining the mean macro, micro and trace elements concentration of soil samples, the sunflower soil samples showed higher element concentrations than acacia soil samples, except for Na, S and Mo; however, 
 Table 3 Results of element contents of examined soil, flower and honey samples. For the county of origin for each sample, see Table 1

Elements	No. 1S	No. 2S	No. 3S	No. 4S	No. 5S	$Mean \pm SD$
$K (mg kg^{-1})$	$3451\pm11$	$825\pm12$	$1009 \pm 5$	$2410\pm17$	$3322\pm29$	$2203\pm1243$
$Mg (mg kg^{-1})$	$2729\pm70$	$653\pm9$	$792\pm9$	$3039\pm18$	$2891\pm32$	$2021\pm1191$
Na (mg $kg^{-1}$ )	$215\pm4$	$27.7 \pm 1.1$	$42.4\pm0.3$	$89.7\pm3.3$	$103\pm2$	$94.9\pm74.0$
$P (mg kg^{-1})$	$772\pm19$	$225\pm5$	$233\pm14$	$600\pm20$	$834\pm14$	$533\pm290$
S (mg kg <sup>-1</sup> )	$731\pm9$	$161 \pm 10$	$140\pm4$	$209\pm2$	$229\pm5$	$294\pm247$
As $(\mu g k g^{-1})$	$10,869 \pm 323$	$1473\pm5$	$1423\pm76$	$7014\pm46$	$8003\pm53$	$5756\pm4180$
B ( $\mu g k g^{-1}$ )	$3216\pm23$	$2359\pm15$	$2304\pm18$	$2611 \pm 6$	$3152\pm 8$	$2728\pm432$
Ba ( $\mu g \ kg^{-1}$ )	$73,142 \pm 144$	$15,334 \pm 102$	$17,029 \pm 180$	$300,\!105\pm559$	$297,000 \pm 694$	140,521 ± 146,129
$Cd (\mu g kg^{-1})$	$176\pm4$	$63.9\pm1.0$	$34.5\pm2.9$	$594\pm12$	$518\pm5$	$277\pm261$
Co ( $\mu g k g^{-1}$ )	$8608\pm58$	$1498 \pm 1$	$1984\pm32$	$9479\pm52$	$10,706 \pm 58$	$6455\pm4371$
$Cr (\mu g kg^{-1})$	39,269 ± 158	$5376\pm84$	$6734\pm84$	$66,163 \pm 176$	$72,447 \pm 506$	37,998 ± 31,714
$Cu (\mu g kg^{-1})$	$22,444 \pm 174$	$6556 \pm 153$	$4874 \pm 102$	$46,070 \pm 149$	$45,\!278 \pm 428$	$25,044 \pm 20,043$
$Fe (mg kg^{-1})$	25,426 ± 3	$4505\pm2$	$5561 \pm 6$	24,343 ± 2	22,638 ± 2	$16,\!495 \pm 10,\!516$
Mn ( $\mu g k g^{-1}$ )	$161,\!472\pm452$	94,823 ± 554	$137,\!426\pm 614$	$139,722 \pm 319$	$155,627 \pm 716$	137,814 ± 26,120
Mo ( $\mu g k g^{-1}$ )	$79.4 \pm 1.8$	$110 \pm 1$	$152 \pm 6$	$45.8\pm4.3$	$69.3 \pm 4.3$	$91.2\pm40.9$
Ni ( $\mu g k g^{-1}$ )	$39,170 \pm 379$	$4218\pm23$	$5471 \pm 16$	56,797 ± 217	$63,504 \pm 447$	33,832 ± 27,918
Pb ( $\mu g k g^{-1}$ )	$18,201 \pm 143$	$7689 \pm 10$	$5426 \pm 63$	30,834 ± 151	$27,523 \pm 164$	$17,935 \pm 11,400$
Sr ( $\mu g k g^{-1}$ )	$22,071 \pm 441$	$5625\pm75$	$4206\pm90$	38,416 ± 232	33,558 ± 252	20,775 ± 15,655
$Zn (\mu g kg^{-1})$	$64,310 \pm 201$	$14,204 \pm 137$	$15,137 \pm 114$	$105,993 \pm 2123$	$109,102 \pm 833$	61,749 ± 46,475
Elements	No. 1F	No. 2F	No. 3F	No. 4F	No. 5F	Mean $\pm$ SD
$K (mg kg^{-1})$	$17,036 \pm 157$	$18,636 \pm 242$	$16,344 \pm 354$	$17,458 \pm 122$	$16,447 \pm 21$	$17,184 \pm 929$
$Mg (mg kg^{-1})$	941 ± 23	838 ± 13	907 ± 22	$1320 \pm 11$	$1663 \pm 18$	$1134 \pm 350$
Na (mg kg <sup><math>-1</math></sup> )	$65.8 \pm 1.8$	33.3 ± 1.5	$35.4 \pm 0.9$	31.8 ± 1.1	$34.4\pm0.8$	$40.1 \pm 14.4$
$P (mg kg^{-1})$	$2292 \pm 52$	$3796 \pm 182$	$3726 \pm 104$	$2150\pm54$	$2918\pm98$	$2976\pm773$
$S (mg kg^{-1})$	$2849 \pm 107$	$1735 \pm 26$	$1647 \pm 23$	$1751 \pm 25$	$1809 \pm 1$	$1958\pm501$
As $(\mu g k g^{-1})$	$119.5 \pm 0.6$	$135 \pm 4$	$104 \pm 3$	$61.4 \pm 1.3$	$89.2 \pm 2.3$	$102 \pm 28$
$B (\mu g kg^{-1})$	$11,670 \pm 192$	13,990 ± 132	$14,711 \pm 681$	$51,793 \pm 411$	$78,061 \pm 48$	34,045 ± 29,703
Ba ( $\mu g k g^{-1}$ )	$4218\pm45$	2011 ± 14	3175 ± 12	$7094 \pm 45$	$7456 \pm 14$	$4791 \pm 2402$
Cd ( $\mu$ g kg <sup>-1</sup> )	$6.28\pm0.01$	$3.18\pm0.09$	$2.22 \pm 0.06$	$393\pm11$	$328 \pm 2$	$147\pm197$
Co ( $\mu$ g kg <sup>-1</sup> )	$120 \pm 1$	$59.5 \pm 1.2$	$61.3 \pm 0.1$	$161 \pm 1$	$191 \pm 2$	$119\pm59$
$Cr(\mu g kg^{-1})$	$86.6 \pm 1.7$	$251 \pm 19$	$300 \pm 1$	$89.8 \pm 4$	$86.2 \pm 0.1$	$163\pm104$
Cu ( $\mu$ g kg <sup>-1</sup> )	$9888 \pm 13$	$9537\pm58$	$6679 \pm 62$	$16,084 \pm 15$	$16,203 \pm 101$	$11,678 \pm 4262$
$Fe (mg kg^{-1})$	$108 \pm 1$	$128 \pm 1$	$93.8\pm0.6$	$49.6\pm0.0$	$67.9\pm0.0$	$89.4 \pm 31.2$
Mn ( $\mu g k g^{-1}$ )	$25,885 \pm 184$	25,608 ± 245	$29,970 \pm 389$	$11,771 \pm 26$	$17,394 \pm 57$	23,326 ± 5521
Mo ( $\mu g k g^{-1}$ )	$261 \pm 24$	$413 \pm 2.24$	$481\pm21$	$207 \pm 1$	211 ± 1	$315\pm125$
Ni ( $\mu g k g^{-1}$ )	$5933\pm84$	$10,518 \pm 60$	$6079\pm84$	$3671\pm 5$	$3749\pm86$	$5990\pm2780$
Pb ( $\mu g k g^{-1}$ )	$65.2 \pm 1.9$	$45.7 \pm 2.6$	$41.0 \pm 8.2$	$138\pm4$	$127\pm0.13$	$83.3 \pm 45.9$
Sr ( $\mu g k g^{-1}$ )	$4360\pm17$	$2575\pm13$	$2159 \pm 21$	$17,504 \pm 41$	$16,132 \pm 64$	$8546\pm7612$
$Zn (\mu g kg^{-1})$	33,023 ± 111	32,151 ± 182	30,631 ± 249	41,366 ± 58	$44,059 \pm 109$	$36,246 \pm 6040$
Elements	No. 1H	No. 2H	No. 3H	No. 4H	No. 5H	Mean $\pm$ SD
$K (mg kg^{-1})$	$285\pm2$	$209 \pm 1$	$228 \pm 1$	$431\pm9$	$492\pm4$	$329\pm126$
$Mg (mg kg^{-1})$	$2.82 \pm 0.15$	$1.22\pm0.01$	$1.37\pm0.22$	$15.1\pm0.1$	$13.8\pm0.0$	$6.88 \pm 6.99$
Na (mg kg <sup><math>-1</math></sup> )	$2.75 \pm 0.11$	$2.31 \pm 0.42$	$2.36\pm0.49$	$3.61\pm0.0$	$4.97\pm0.12$	$3.20 \pm 1.12$
$P (mg kg^{-1})$	$28.9 \pm 1.5$	$31.2\pm0.8$	$30.7 \pm 1.2$	$52.9\pm0.8$	$61.8 \pm 1.6$	41.1 ± 15.2
$S (mg kg^{-1})$	$18.7\pm1.5$	$17.3\pm0.6$	$12.7\pm2.4$	$20.3\pm0.1$	$23.6\pm0.3$	$18.5\pm4.0$

Table 3	(continued)
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As (µg kg <sup>-1</sup> )	$15.4 \pm 0.1$	$11.2 \pm 0.1$	$11.7\pm0.2$	$15.6 \pm 0.1$	$16.9\pm0.1$	$14.1 \pm 2.5$
B ( $\mu g k g^{-1}$ )	$2971 \pm 44$	$2797\pm60$	$2851\pm90$	$3244\pm90$	$4775\pm18$	$3328\pm827$
Ba ( $\mu g k g^{-1}$ )	$18.9\pm0.2$	$10.7\pm0.5$	$15.1 \pm 1.8$	$39.9 \pm 1.6$	$34.6 \pm 1.2$	$23.8 \pm 12.7$
$Cd (\mu g kg^{-1})$	<dl< td=""><td><dl< td=""><td><dl< td=""><td><math display="block">1.18\pm0.19</math></td><td><math display="block">0.750\pm0.012</math></td><td><math display="block">0,967\pm0.306</math></td></dl<></td></dl<></td></dl<>	<dl< td=""><td><dl< td=""><td><math display="block">1.18\pm0.19</math></td><td><math display="block">0.750\pm0.012</math></td><td><math display="block">0,967\pm0.306</math></td></dl<></td></dl<>	<dl< td=""><td><math display="block">1.18\pm0.19</math></td><td><math display="block">0.750\pm0.012</math></td><td><math display="block">0,967\pm0.306</math></td></dl<>	$1.18\pm0.19$	$0.750\pm0.012$	$0,967\pm0.306$
$Co (\mu g kg^{-1})$	<dl< td=""><td><dl< td=""><td><dl< td=""><td><math display="block">0.871\pm0.050</math></td><td><math>1.71 \pm 0.15</math></td><td><math display="block">1.29\pm0.60</math></td></dl<></td></dl<></td></dl<>	<dl< td=""><td><dl< td=""><td><math display="block">0.871\pm0.050</math></td><td><math>1.71 \pm 0.15</math></td><td><math display="block">1.29\pm0.60</math></td></dl<></td></dl<>	<dl< td=""><td><math display="block">0.871\pm0.050</math></td><td><math>1.71 \pm 0.15</math></td><td><math display="block">1.29\pm0.60</math></td></dl<>	$0.871\pm0.050$	$1.71 \pm 0.15$	$1.29\pm0.60$
$Cr (\mu g kg^{-1})$	<dl< td=""><td><dl< td=""><td><dl< td=""><td><dl< td=""><td><dl< td=""><td><dl< td=""></dl<></td></dl<></td></dl<></td></dl<></td></dl<></td></dl<>	<dl< td=""><td><dl< td=""><td><dl< td=""><td><dl< td=""><td><dl< td=""></dl<></td></dl<></td></dl<></td></dl<></td></dl<>	<dl< td=""><td><dl< td=""><td><dl< td=""><td><dl< td=""></dl<></td></dl<></td></dl<></td></dl<>	<dl< td=""><td><dl< td=""><td><dl< td=""></dl<></td></dl<></td></dl<>	<dl< td=""><td><dl< td=""></dl<></td></dl<>	<dl< td=""></dl<>
$Cu (\mu g kg^{-1})$	$76.3\pm3.0$	$57.5 \pm 1.2$	$51.9\pm2.0$	$129 \pm 3$	$131 \pm 4$	$89.3\pm38.6$
$Fe (mg kg^{-1})$	$0.437\pm0.026$	$0.287\pm0.004$	$0.421\pm0.008$	$0.612\pm0.011$	$0.553 \pm 0.009$	$0.462 \pm 0.126$
$Mn (\mu g kg^{-1})$	$191 \pm 4$	$123 \pm 3$	$140 \pm 3$	$146 \pm 2$	$172 \pm 4$	$154 \pm 27$
Mo ( $\mu g k g^{-1}$ )	$0.514\pm0.560$	$0.871 \pm 0.026$	$0.831\pm0.048$	<dl< td=""><td><math display="block">0.326\pm0.02</math></td><td><math>0.636 \pm 0.261</math></td></dl<>	$0.326\pm0.02$	$0.636 \pm 0.261$
Ni ( $\mu g k g^{-1}$ )	$14.8\pm0.9$	$20.9\pm1.9$	$25.9\pm0.3$	$3.90\pm0.19$	$6.38 \pm 1.26$	$14.3\pm9.3$
Pb ( $\mu g k g^{-1}$ )	$2.45\pm0.0$	$0.516\pm0.014$	$0.451 \pm 0.041$	$7.46\pm0.30$	$6.77\pm0.12$	$3.53\pm3.38$
$Sr(\mu g kg^{-1})$	$44.5\pm2.9$	$28.4\pm0.3$	$31.9\pm1.9$	$98.2\pm0.6$	$93.2 \pm 2.6$	$59.2\pm33.9$
$Zn (\mu g kg^{-1})$	$967\pm46$	$420\pm4$	$575\pm5$	$2866\pm130$	$3233\pm21$	$1612\pm1333$

statistically verified differences were determined only in the case of Ba, Cd, Cr, Cu, Pb, Sr and Zn contents (Table 4). Comparing the results of two different counties, the determined element concentrations were higher in samples from Békés County than in samples from Szabolcs-Szatmár-Bereg County, except for Mo. Significant differences (P value < 0.05) were found in K, Mg, P, As, Co, Cr, Cu, Fe, Mo, Ni, Pb, Sr and Zn concentrations (P values = 0.01, < 0.001, 0.01, 0.02, 0.002, 0.03, 0.05, <0.001, 0.05, 0.01, 0.03, 0.02 and 0.02, respectively). Note, we also found that, comparing the soils used for the growing of acacia and sunflowers, there were significant increases in the following elements in the sunflower soils, compared with the acacia soils: Cd (P = 0.005), Cr (P = 0.04), Cu (P = 0.02), Pb (P = 0.04), Sr (P = 0.04) and Zn (P = 0.04).

In Table 3, the element concentrations of flower samples are shown. Examining the macro element concentrations No. 1F sample showed the highest Na and S contents, and the highest K and P concentrations were determined in No. 2F sample. No. 5F sample showed the highest Mg content. In every sample, K was present in the highest concentration and based on mean element contents P was the second most abundant element followed by S, Mg and Na. The highest As, Fe and Ni contents were measured in No. 2F sample and No. 3F sample showed the highest Cr, Mn and Mo concentrations. Sunflower flower samples showed the highest Cd, Pb and Sr (No.4F) as well as the highest B, Ba, Co, Cu and Zn (No. 5F) contents. Examining the sunflower flower samples more than 90 times Cd concentration was determined in these flower samples compared to acacia flower samples. Based on mean micro element concentrations, the most abundant element was the Fe followed by Zn, B, Mn, Cu, Sr, Ni and Ba. The concentrations of Mo, Cd, Cr, Co, As and Pb were under 1000  $\mu$ g/kg. Comparing the two sunflower samples, the concentrations of Cr, Cu, Mo and Pb were very similar; however, No. 4F sample showed higher Cd and Sr concentrations than No. 5F sample. Comparing acacia flower samples, No. 1F samples showed higher Ba, Cd, Pb and Sr concentrations than the other two acacia flower samples. Higher B, Ba, Cd, Co, Cu, Pb, Sr and Zn contents were measured in sunflower flowers than acacia flowers (Table 4); however, significant differences (P value < 0.005) between the acacia and sunflower flowers existed for B, Ba, Co, Cu, Fe, Mn, Pb, Sr and Zn contents. Examining the flower samples from two different soil types showed significant differences for P, Co, Cr and Mo (*P* = 0.02, 0.04, 0.002 and 0.007, respectively).

Examining the macro element contents of honey samples, K was present in the highest concentrations followed by P and S in all of the honey samples (Table 3). In the case of acacia honey samples, No. 1H sample showed higher K, Mg, Na and S contents than the other two acacia honey samples. No. 2H and No.3H acacia honey samples showed very similar K, Mg, Na and P concentrations. Examining the sunflower honeys, No. 5H sample showed higher macro element concentrations except for Mg; however, major differences were not detected between these two samples. Sunflower honey also tended to have lower P content, which together with the high K in all honey samples, and more in sunflower than acacia honey, confirms our previous studies (Czipa et al. 2015). The macro element concentration orders were the following: Mg<Na<S<P<K for acacia honeys and Na<Mg<S<P<K for sunflower

Table 4 Elem	ent concentrations of s	samples of soil, flowe	ars and hone	ys in aca	icia and sunflower	r-growing regions						
Element	Soil, acacia mean ± SD	Soil, sunflower Mean ± SD	SED	P value	Flower, acacia mean ± SD	Flower, sunflower Mean ± SD	SED	P value	Honey, acacia Mean ± SD	Honey, sunflower Mean ± SD	SED	<i>P</i> value
K (mg kg <sup><math>-1</math></sup> )	$1762 \pm 1466$	$2866\pm 645$	1144	0.406	$17,339 \pm 1176$	$16,953 \pm 715$	954	0.713	$241 \pm 40$	$462 \pm 43$	37.2	0.010
Mg (mg $kg^{-1}$ )	$1391 \pm 1161$	$2965\pm105$	867	0.167	$895\pm52$	$1492\pm243$	174	0.169	$1.80\pm0.88$	$14.5\pm0.9$	0.818	0.001
Na (mg kg <sup><math>-1</math></sup> )	$95.0\pm104$	$94.9\pm11.5$	77.9	0.998	$44.9\pm18.2$	$33.1 \pm 1.8$	13.6	0.450	$2.47\pm0.24$	$4.29\pm0.96$	0.694	0.216
$P (mg kg^{-1})$	$410\pm314$	$717 \pm 165$	249	0.306	$3271\pm849$	$2534\pm543$	694	0.366	$30.3 \pm 1.2$	$57.4\pm6.36$	4.55	0.098
S (mg kg <sup><math>-1</math></sup> )	$344\pm335$	$219 \pm 14$	250	0.651	$2077\pm670$	$1780\pm41$	500	0.594	$16.2 \pm 3.1$	$22.0\pm2.3$	2.64	0.119
As ( $\mu g kg^{-1}$ )	$4588\pm5439$	$7509 \pm 699$	4071	0.525	$120 \pm 16$	$75.4 \pm 19.7$	15.5	0.065	$12.8\pm2.3$	$16.3\pm0.9$	1.48	0.145
B ( $\mu g \ kg^{-1}$ )	$2626\pm511$	$2882\pm383$	431	0.596	$13,457 \pm 1598$	$64,927 \pm 18,574$	9861	0.014	$2873\pm89$	$4010\pm1083$	767	0.376
Ba ( $\mu g \ kg^{-1}$ )	$35,168\pm 32,897$	$298,553 \pm 2196$	24,546	0.002	$3135 \pm 1104$	$7275 \pm 256$	834	0.016	$14.9 \pm 4.1$	$37.3 \pm 3.7$	3.64	0.009
Cd (µg kg <sup>-1</sup> )	$91.5 \pm 74.7$	$556 \pm 54$	62.4	0.005	$3.89\pm2.12$	$361 \pm 46$	32.5	0.058	${<}DL^{a}$	$0.965\pm0.304$	I	I
Co ( $\mu g \ kg^{-1}$ )	$4030\pm3972$	$10,093\pm868$	2996	0.136	$80.3\pm34.4$	$176 \pm 21$	28.0	0.042	<dl< td=""><td><math display="block">1.29\pm0.59</math></td><td>I</td><td>I</td></dl<>	$1.29\pm0.59$	I	I
$Cr (\mu g kg^{-1})$	$17,126 \pm 19,188$	$69,305 \pm 4443$	14,492	0.037	$213 \pm 112$	$88.1\pm2.6$	83.3	0.232	<dl< td=""><td><dl< td=""><td>I</td><td>I</td></dl<></td></dl<>	<dl< td=""><td>I</td><td>I</td></dl<>	I	I
Cu ( $\mu g kg^{-1}$ )	$11,291\pm9695$	$45,674 \pm 560$	7232	0.018	$8701\pm1760$	$16,143\pm84$	1313	0.011	$61.9\pm12.8$	$130 \pm 1$	9.56	0.006
Fe (mg $kg^{-1}$ )	$11,830 \pm 11,786$	$23,490 \pm 1206$	8807	0.277	$110\pm17.2$	$58.8\pm12.9$	14.5	0.039	$0.382\pm0.082$	$0.583\pm0.042$	0.065	0.054
$Mn~(\mu g~kg^{-1})$	$131,\!240\pm 33,\!752$	$147,675 \pm 11,247$	25,846	0.570	$27,154 \pm 2442$	$17,583\pm267$	1826	0.014	$151 \pm 35$	$159\pm18.4$	28.1	0.803
Mo ( $\mu g \ kg^{-1}$ )	$114 \pm 36$	$57.6\pm16.6$	28.5	0.143	$385\pm113$	$209 \pm 3$	84.0	0.127	$0.739\pm0.196$	<dl< td=""><td>I</td><td>Ι</td></dl<>	I	Ι
Ni (µg kg <sup>-1</sup> )	$16,\!286\pm19,\!828$	$60,150 \pm 4743$	14,999	0.061	$7510\pm2606$	$3710\pm54$	1943	0.145	$20.5\pm5.6$	$5.12 \pm 1.78$	3.45	0.036
Pb ( $\mu g kg^{-1}$ )	$10,438 \pm 6816$	$29,179 \pm 2341$	5228	0.037	$50.6 \pm 12.8$	$133 \pm 8$	10.4	0.004	$1.14 \pm 1.14$	$7.12 \pm 0.49$	0.885	0.007
Sr ( $\mu g k g^{-1}$ )	$10,634\pm9930$	$35,987 \pm 3435$	7620	0.045	$3031\pm1169$	$16,818 \pm 971$	1010	0.001	$34.9\pm8.47$	$95.7 \pm 3.54$	6.58	0.003
Zn ( $\mu g kg^{-1}$ )	$31,217 \pm 28,663$	$107,548 \pm 2198$	21,396	0.038	$31,935 \pm 1211$	$42,713 \pm 1904$	1350	0.004	$654\pm282$	$3050\pm 260$	251	0.002
<sup>a</sup> Under detectic	on limit											

 $\Delta$  Springer

SD standard deviation, SED standard error difference

honeys. Comparing the two honey types, higher macro element contents were measured in sunflower honeys than acacia honeys; and significant differences were determined in K and Mg concentrations. Acacia honey samples from Italy, Malaysia and Saudi Arabia have been reported with higher K (719  $\pm$  390, 1277  $\pm$  123 and 429–491 mg kg<sup>-1</sup>, respectively) Mg (70.0  $\pm$  27.0,  $14.1 \pm 5.8$  and  $189-196 \text{ mg kg}^{-1}$ , respectively) and Na  $(91.0 \pm 29.0, 529 \pm 61 \text{ and } 15.9 - 19.0 \text{ mg kg}^{-1}, \text{ respec-}$ tively) concentrations than ours (Di Bella et al. 2015, Chua et al. 2012 and Algarni et al. 2014, respectively). Fermo et al. (2013) found similar concentrations in Italian honeys to those of our samples, but they were not from acacia or sunflower. Oroian et al. (2015) also measured high K (554 and 849 mg kg<sup>-1</sup>), Mg (51.2 and  $63.8 \text{ mg kg}^{-1}$ ) and Na (171 and 154 mg kg $^{-1}$ ) in Romanian acacia and sunflower honeys. However, Atanassova et al. (2012) determined similar K (126 and 247 mg kg<sup>-1</sup>), Mg (6.00 and 14.0 mg kg<sup>-1</sup>), Na  $(8.11 \text{ and } 7.58 \text{ mg kg}^{-1})$ , P (24.0 and 41.0 mg kg<sup>-1</sup>) and S (12.0 and 20.0 mg kg<sup>-1</sup>) concentrations in Bulgarian acacia and sunflower honeys. North Indian sunflower honey showed higher K (176  $\pm$  0 mg kg<sup>-1</sup>) and Na  $(690 \pm 0 \text{ mg kg}^{-1})$  concentrations than ours (Nanda et al. 2003).

Examining the micro element contents of honey samples, the Cr concentrations were under DL in every honey sample and Cd and Co concentrations were both under DL in acacia honeys (Table 3). Micro element contents in No. 2H and No. 3H acacia honey samples were very similar. Acacia honey sample from Békés county (No. 1H) showed higher micro element concentrations than the other two acacia honeys from Szabolcs-Szatmár-Bereg county, except for Mo. In the sunflower honey samples (No.4H and No.5H), the concentrations of examined micro elements were similar; however, much higher B and Zn contents were measured in No. 5H sample.

Overall, honey from sunflowers had higher B, Ba, Cu, Fe, Pb, Sr and Zn contents and lower Ni than honey from acacia flowers; however, significant differences (*P* value <0.005) were determined only for Ba, Cu, Pb, Sr and Zn contents (Table 4). In relation to micro and trace element content, B, Zn and Fe had the highest values. Mn and As concentrations were similar in both honey types, but Ba, Cu, Sr and Pb were all higher in sunflower than acacia honey samples. However, Mo and Ni contents were higher in acacia than sunflower honey samples. The micro element order was as follows: Mo<Pb<As<Ba<Ni<Sr<Cu<Mn<Fe<Zn<B for acacia honey and Mo<Cd<Co<Ni<Pb<As<Ba<Sr<Cu<Mn<Fe<Zn<B for sunflower honey; thus, the order from Sr to B was the same.

Oroian et al. (2015) measured higher Ba, Cr, Cu, Fe, Mn, Ni, Pb and Sr but lower As concentrations in acacia (28.0; 51.0; 1820; 19,390; 1720; 191; 62.0; 264 and 9.00  $\mu$ g kg<sup>-1</sup>, respectively) and sunflower (349; 37.0; 2390; 24,010; 1000; 183; 40.0; 351 and 5.00  $\mu$ g kg<sup>-1</sup>, respectively) honeys from Romania. Bulgarian acacia and sunflower honeys (Atanassova et al. 2012) had higher Fe (830 and 1930  $\mu g \text{ kg}^{-1}$ ) and Sr (150 and 210  $\mu$ g kg<sup>-1</sup>) but lower Zn (220 and 610  $\mu$ g kg<sup>-1</sup>) content than our samples. Micro and trace element contents of Egyptian honeys from sandy soil measured by Rashed et al. (2009) (5.00–430  $\mu$ g kg<sup>-1</sup> for Cd, 80– 800  $\mu g \ kg^{-1}$  for Co, 650–1600  $\mu g \ kg^{-1}$  for Cr, 1400– 1900  $\mu$ g kg<sup>-1</sup> for Cu, 35,000–64,000  $\mu$ g kg<sup>-1</sup> for Fe, 630–1400  $\mu$ g kg<sup>-1</sup> for Mn, 200–700  $\mu$ g kg<sup>-1</sup> for Ni, 1500–2100  $\mu$ g kg<sup>-1</sup> for Pb and 8800–11,000  $\mu$ g kg<sup>-1</sup> for Zn) were much higher than in our samples and Al Naggar et al. (2013) determined much higher Fe (2800-3730  $\mu$ g kg<sup>-1</sup>), and Pb (110–1590  $\mu$ g kg<sup>-1</sup>), but lower Zn (1020–1430  $\mu$ g kg<sup>-1</sup>), concentrations in their Egyptian honey samples. Conti and Botrè (2000) measured higher Cd ( $<2.00-63.0 \ \mu g \ kg^{-1}$ ) and Cr (8.40- $102 \ \mu g \ kg^{-1}$ ,) concentration in Italian honey samples.

Examining the element concentration of sunflower soil and honey samples, the honey collected from soil with higher element contents also had higher element concentrations. In the case of acacia soil and honey samples, a similar tendency was observed, except for P, Mo and Ni concentrations.

Comparing the element contents of soil, flower and honey samples

Combined with the soil, flower and honey samples confirmed that those from soils with high element concentrations showed high element contents for several examined elements. Because the element uptake and transport is influenced by soil properties and plant type, the samples were analysed separately for the different plant types. Examining the acacia samples, the flower and honey samples followed a tendency that was observed in soil samples, namely the flowers and honeys collected from soils with higher Mg, Na, S, Ba, Cu and Pb contents showed higher concentrations of these elements. In the case of K, Fe, Mn and Zn, the element content of flower samples did not follow the element content of soils; however, the honeys did showed a similar tendency. Flower samples had similarly high concentrations to soil samples in the case of Mo and Sr; however, honey samples did not follow this trend. In the case of P. the order of element content of flower and honeys samples was the same but soils showed a different order. In the case of As, B and Ni relations were not able to be determined. Examining the sunflower soils, the order of examined elements of soil, flower and honey samples was the same except for K, Mg, Ba and Fe. In the case of K, Mg and Fe, the element content of soil samples was followed by honey samples; however, the flower samples did not show this tendency. In the case of Ba, only the flower and honey samples showed the same trends.

From the BCF values of acacia and sunflower samples, it is evident that flower/soil values were greater than 1.00 for K, P, S, B, Cu, Mo, Ni and Zn (acacia) and for K, P, S, B and Mo (sunflower); BCF (honey/flower) values were less than 1.00 in case of all samples; BCF

(honey/soil) values were higher than 1.00 for B in both samples (Table 5). In acacia samples considered separately, BCF (flower/soil) values were much lower for samples from Békés County; samples from Szabolcs-Szatmár-Bereg County showed increased BCF (honey/ flower) values for Na, S, Ba, Cu, Fe, Mo, Ni and Sr. Examining the honey/soil values for acacia samples, those from Békés County had lower values (except Mn, Mo and Pb) than the other two samples. Sunflower samples showed similar BCF (flower/ soil) values for Na, P, Ba, Cd, Co, Cu, Pb, Sr and Zn; however, the samples from Sarkad (No. 5) showed higher values for Mg, As, B and Fe. Examining the BCF (honey/flower) values, the sunflower honeys from Sarkad showed higher values for K, Na, S, Co, Mn, Ni and Zn, and sunflower samples from Sarkadkeresztúr had higher values for Mg, P, As, Ba, Cd and Fe. For other elements, the values were very similar. Examining the BCF (honey/soil) values, the sunflower samples showed similar values for Mg, As, Ba,

Table 5 Bioconcentration factors for the flower/soil, honey/flower and honey/soil transitions

	Flower/so	il		Honey/flo	ower		Honey/soil		
	Acacia	Sunflower	Overall	Acacia	Sunflower	Overall	Acacia	Sunflower	Overall
К	14.6	6.10	11.2	0.014	0.027	0.019 <sup>a</sup>	0.188	0.164	0.178
Mg	0.924	0.505	0.757	0.002	0.010	0.005	0.002	0.005	$0.003^{a}$
Na	0.781	0.351	0.609	0.059	0.129	$0.087^{\rm a}$	0.051	0.045	0.048
Р	11.9	3.54	8.58	0.010	0.023	0.015	0.103	0.081	0.094
S	8.80	8.15	8.54	0.008	0.012	0.010	0.074	0.100	0.085
As	0.059	0.010	0.039	0.108	0.222	0.153 <sup>a</sup>	0.006	0.002	0.004
В	5.32	22.2	12.1	0.216	0.062	0.154 <sup>a</sup>	1.12	1.38	1.22
Ba	0.125	0.024	0.085	0.005	0.005	0.005	0.0006	0.0001	0.0004
Cd	0.050	0.647	0.289 <sup>a</sup>	-	0.003	-	_	0.002	-
Со	0.028	0.017	0.024	-	0.007	-	_	0.0001	-
Cr	0.031	0.001	0.019	-	_	-	_	_	-
Cu	1.09	0.353	0.795	0.007	0.008	0.008	0.008	0.003	0.004
Fe	0.017	0.003	0.011	0.004	0.010	0.006	0.00005	0.00002	0.00004
Mn	0.216	0.119	0.177	0.006	0.009	0.007	0.001	0.001	0.001
Мо	3.41	3.79	3.56	0.002	_	-	0.007	_	-
Ni	1.25	0.062	0.776	0.003	0.001	0.002	0.003	0.0001	0.002
Pb	0.006	0.005	0.005	0.020	0.054	0.033	0.0001	0.0002	$0.0002^{a}$
Sr	0.390	0.468	0.421	0.012	0.006	0.009 <sup>a</sup>	0.005	0.003	0.004
Zn	1.60	0.397	1.12	0.020	0.071	0.041 <sup>a</sup>	0.028	0.028	0.028

<sup>a</sup> Correlation is significant at the 0.05 probability level

Cu, Fe, Mn, Ni and Pb. Samples from Sarkadkeresztúr showed higher values for K, P and Cd.

Considering the samples together (five soils, five flowers and five honeys), the BCF (flower/ soil) values were greater for B, K, P, S and Mo than all other elements. The lowest values were determined for As, Co, Cr, Fe and Pb. BCF (honey/flower) values were the highest for B and As, then all the other elements, except Ni, which was lower than all of these and Cd, Co and Cr, which were not determinable. In the case of honey samples, the BCF (honey/soil) values were low (except B), thus the translocation of examined elements from soil to nectar (honey) was low. The BCF (honey/soil) values were the highest for B, then K, then all other elements, except Cd, Co and Cr, which were non determinable and P and S which were intermediate between K and the other elements. The BCF orders were very similar for acacia flowers and sunflowers.

Examining the results, there was little movement of Fe through the soil-flower-honey system. Since Fe can be bound to the cell wall of the root rhizodermis of root (Szabó 1998), the translocation of this element from root to other organs (e.g. flower) is limited. Similarly, the translocation of two potentially toxic elements, Pb and As, was very low. The translocation of Mo was high between the soil and flower; however, this movement was very low to honey. The translocation of Mn and Ba was moderate in this system. In relation to micro elements, the two highest movements were for Zn and B.

Comparing the bioconcentration factors with elements as replicates, these were higher for flower/soil (mean 2.57) than honey/flower and honey/soil (means 0.098 and 0.038, respectively (SED = 0.816, P = 0.005).

Table 6 shows the results of Pearson's correlation between elements of flower and soil, honey and soil or honey and flower system. The elements with significant correlations between honey and soil, in descending order of *P* value, were Cu> Ba>Pb>Sr=Ni>Zn>Mn>As. The elements with significant correlations between

	Flowers and so	pils	Honeys and flo	wers	Honeys and so	ils
	Corr.	P value	Corr.	P value	Corr.	P value
K	-0.422	0.479	-0.374	0.535	0.694	0.194
Mg	0.717	0.173	0.911	0.031	0.784	0.117
Na	0.895 <sup>a</sup>	0.040	-0.235	0.703	0.184	0.767
Р	-0.796	0.107	-0.394	0.512	0.583	0.302
S	0.998 <sup>a</sup>	< 0.001	0.139	0.824	0.166	0.790
As	-0.290	0.636	-0.600	0.285	$0.884^{a}$	0.046
В	0.414	0.488	0.916 <sup>a</sup>	0.029	0.592	0.293
Ba	0.976 <sup>a</sup>	0.004	$0.975^{\rm a}$	0.005	0.982 <sup>a</sup>	0.003
Cd	0.981 <sup>a</sup>	0.003	_	-	_	_
Co	0.962 <sup>a</sup>	0.009	_	-	_	_
Cr	$-0.903^{a}$	0.036	_	-	_	_
Cu	0.964 <sup>a</sup>	0.008	0.981 <sup>a</sup>	0.003	0.993 <sup>a</sup>	0.001
Fe	-0.601	0.284	$-0.972^{a}$	0.006	0.765	0.132
Mn	-0.271	0.659	-0.220	0.722	0.893 <sup>a</sup>	0.041
Мо	0.963 <sup>a</sup>	0.009	0.951 <sup>a</sup>	0.049	0.843	0.157
Ni	-0.827	0.084	0.700	0.188	$-0.954^{a}$	0.012
Pb	0.972 <sup>a</sup>	0.006	0.995 <sup>a</sup>	<.001	0.979 <sup>a</sup>	0.004
Sr	0.935 <sup>a</sup>	0.020	0.996 <sup>a</sup>	<.001	0.954 <sup>a</sup>	0.012
Zn	0.930 <sup>a</sup>	0.022	0.989 <sup>a</sup>	0.001	0.950 <sup>a</sup>	0.013

Table 6 Pearson's correlation coefficients for flowers and soils, honeys and soils, and honeys and flowers

<sup>a</sup> Significant correlation

honey and flower, in descending order of P value, were Pb=Sr>Zn>Cu>Ba>Fe>B>Mo. The elements with significant correlations between flower and soil, in descending order of P value, were S>Cd>Ba>Pb>Cu>Co=Mo>Sr>Zn>Cr>Na.

## Conclusions

In this study, 19 elements were measured in five-five soil, flower and honey samples (acacia and sunflower) from two Hungarian Counties (Békés and Szabolcs-Szatmár-Bereg County) and BCF values were determined using these samples. Soil samples were collected from unpolluted areas and our results showed low contaminant concentrations, with little bioconcentration in the case of Pb and As, and with Cd undeterminable due to low concentrations. The highest bioconcentration from soil to honey was for B, which was the only element in higher concentrations in honey than soil. K, P, S and Na showed higher bioconcentration than other elements. The strongest correlations between soil and honey were for Cu, Ba and Sr. The results have potential for detecting regional deficiencies in soil, for example as suggested by the correlation coefficients of 0.99 and 0.95 for Cu and Zn, respectively, since bees gather pollen from a region of about 7 km<sup>2</sup>, thus avoiding the need to take soil samples over large areas. High Pb and As (CC 0.98 and 0.88, respectively) concentrations in soils may also be successfully determined from their concentrations in honey, but this is yet to be confirmed in contaminated regions. In the literature, there are many studies about honey as a bioindicator; however, the examination of soils, flowers and honeys element content together is very rare. With this study, we are able to verify the relations among the element contents of honeys, flowers and soils.

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