# ORIGINAL PAPER



# **Emissions monitoring and carcinogenic risk assessment of PM10‑bounded PAHs in the air from Candiota's coal activity area, Brazil**

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**Abstract** The city of Candiota contains a great amount of coal resources. Coal activities, including coal combustion, are major releasers of polycyclic aromatic hydrocarbons (PAHs). The PAHs are considered priority air pollutants regarding their large carcinogenic potential. So, the carcinogenic risk assessment of populations living near areas with PAH sources is mandatory. This study aimed to evaluate the carcinogenic health risk of the PAH inhalation exposure of individuals living in Candiota City. A total of 158 individuals were enrolled in the study. Monitoring of PAH and meteorological parameters were carried out, and the health risk assessment was

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determined through the benzo(a)pyrene equivalent toxic equivalent quotient (BaP-TEQ) and the incremental lifetime cancer risk (ILCR) estimation. The coal activity area of Candiota demonstrated an annual PAH concentration of  $27.7 \text{ ng/m}^3$ , PM10 concentration of 26.3  $\mu$ g/m<sup>3</sup>, SO<sub>2</sub> concentration of 9.5  $\mu$ g/m<sup>3</sup>, a BaP-TEQ value of  $0.3 \text{ ng/m}^3$ , and a daily inhalation of 62.4 ng/day. The comparison among seasons showed no diference in PAH concentration and BaP-TEQ. It was observed ILCR values of  $2.8 \times 10^{-6}$  and  $2.6 \times 10^{-6}$  for estimation based on reference and real values, respectively, and these levels were above the reference limit of  $10^{-6}$ , indicating cancer risk. Therefore, an epidemiological survey of cancer cases in the region and its relationship with environmental exposure and air pollutants levels must be required.

**Keywords** Coal activity · Air pollution · Cancer development · Environmental monitoring

## **Introduction**

The relationship between environmental issues and the abrupt development of the modern world is one of the leading issues of the present time, as its progression has been followed by an accelerated increase in energy demand (Kaygusuz, [2012](#page-11-0)). In present days, the production of energy is mostly by burning fossil fuels such as coal, oil, and natural gas (Veziroglu & Sahin,  $2008$ ). Coal is a relevant resource that is

abundant world widely, providing approximately 40% of the world's electricity, and the cheapest source of energy (Franco & Diaz, [2009](#page-11-1); World Environment Council, [2016\)](#page-12-1). Coal activities, including its pyrolysis, are a great source of polycyclic aromatic hydrocarbons (PAHs) releasing and are linked with several environmental problems (WHO, [2000](#page-12-2); Abdulazeez & Fantke, [2017\)](#page-10-0).

Polycyclic aromatic hydrocarbons (PAHs) consist of two or more fused aromatic rings made entirely from carbon and hydrogen. These substances represent carcinogenic risks to humans, once PAHs are a major group of carcinogens and mutagens in the environment (Straif et al., [2005](#page-12-3), Kelly et al., [2021](#page-11-2)). PAH includes an ample variety of compounds that are emitted from several anthropogenic activities, such as the coal activities, including mining process, combustion, and dust resuspension; also, PAH can be found in the gas phase or well-bonded to particulate matter (PM) (Akhbarizadeh et al., [2020](#page-10-1); Kong et al., [2010](#page-11-3); Zhu et al., [2014\)](#page-12-4). PAHs are persistent and broadly dispersed in the atmosphere, being able to be transported over long distances; also, they are toxic and difficult to degrade. These substances were one of the frst atmospheric pollutants being identifed as a suspected carcinogen when inhaled (Abdulazeez & Fantke, [2017\)](#page-10-0). The US Environmental Protection Agency (USEPA) had listed 16 PAHs as priority pollutants because of the greater chance of exposure by humans and their high concentrations in the environment, in which benzo[a]pyrene (B[a]P) is the best known (USEPA, [2014\)](#page-12-5). B[a]P is a suitable marker due to its stability and relatively constant contribution to the carcinogenic activity of particle-bound PAH, being considered highly carcinogenic and mutagenic (Abdulazeez & Fantke, [2017](#page-10-0); European Communities, [2001\)](#page-11-4). B[a]P is often used as an indicator of human exposure to PAHs, and the toxicity of other PAHs is converted into toxicity equivalency factors (TEFs) to B[a]P to evaluate their relative toxicities (Lee, [2010](#page-12-6)).

Coal activity is a hazardous activity, not restricted to underground miners but extends to surface mining workers and the surrounding communities that live near the mine (Laney & Weissman, [2014](#page-12-7); Perret et al., [2017](#page-12-8)). In the course of the coal mining process, coal dust can spread to the air through mining operations and transport activities. Moreover, generous quantities of mining wastes are produced and stocked around the coal mine (Stout & Emsbo-Mattingly, [2008\)](#page-12-9). Thereby, emissions of coal dust particles from coal mines are absolutely serious, as well as the pyrolysis of coal for energy production. In Brazil, there is the second-largest recoverable coal reserve in the Western Hemisphere, totaling 11.9 billion metric tons (BRASIL, [2008\)](#page-11-5). It comprises the Rio Bonito Formation of the Paraná Basin, located in the southernmost region of Brazil, including the States of Paraná, Santa Catarina, and the Rio Grande do Sul. Still, the greater amount of coal resources occur in the Rio Grande do Sul (89%), more specifcally in the city of Candiota (Chaves et al., [2018\)](#page-11-6). Several studies in Candiota's region have already demonstrated an increased risk of mortality and/or morbidity, alterations in hematological parameters, liver function, and genotoxicity and mutagenicity in human cells among residents in the vicinity of the coal usage (Bigliardi et al., [2021;](#page-11-7) Cortes-Ramirez et al., [2018;](#page-11-8) da Silva Júnior et al., [2018](#page-11-9)). Coal mining and combustion contribute to local PAH emission, with documented genotoxic efects and increased risks for cancer, cardiovascular disease, and respiratory disease, among mineworkers and community populations (Espitia-Pérez et al., [2018](#page-11-10); Ghose & Majee, [2007;](#page-11-11) Kurth et al., [2015](#page-11-12)).

Therefore, the knowledge about human health risks related to environmental exposure to hazardous chemical agents is a current and valuable concern (Ghasemi et al., [2020](#page-11-13); Idani et al., [2020](#page-11-14)). Assessing the environmental impact of coal activity areas is a fundamental tool to periodically monitor human health, even more, when considering the potential carcinogenic efects of PAH exposure. Therefore, the present study aimed to evaluate the carcinogenic health risk of the PAH inhalation exposure of individuals living in Candiota City, the biggest coal activity area in Brazil.

#### **Methods**

#### Study area

Candiota city (31 $\degree$  33' 28" S 53 $\degree$  40' 22" W) is located in Campaign Region in the southeast of the state of Rio Grande do Sul, covering an area of 933,834 km<sup>2</sup> (Fig. [1](#page-2-0)). This city has approximately 9.406 inhabitants (9.39 83 people per  $km<sup>2</sup>$ ) and has a human development index (HDI) of 0.698. Candiota is situated



**Fig. 1** The geographical region of the study area located in Candiota City, Rio Grande do Sul State, Brazil

<span id="page-2-0"></span>197 heights above mean sea level and is characterized by a humid subtropical climate (Köppen climate classifcation) (Prefeitura de Candiota, [2013](#page-12-10)). Candiota coalfeld is the largest mineral coal reserve in Brazil, situated in a location with the most favorable geological conditions in the country to superficial coal extraction (up to 50 m deep) (Secretaria do Meio Ambiente e Infraestrutura RS). The extracted coal is used locally to generate electricity at a thermal power complex, called Presidente Medici, which is the largest coal-fred power station. The coal from Candiota´s mine is considered low-quality regarding calorifc power, although it has a content of more than 50% fy ash, which is a fne powder formed from the mineral matter in coal consisting of the noncombustible matter, and a sulfur concentration higher than 1.5%  $(BRASIL, 2008; Pires et al., 2001).$  Therefore, it is assumed that the emission of atmospheric pollutants from this coal is higher than that from similar plants worldwide. A conceptual chart illustrating the methodology approaches is presented in Supplementary material (Fig. S1).

## Data collection of the individuals

Data from individuals living in the study area were used to achieve more realistic and regional parameters, as recommended by USEPA ([2005](#page-12-12)) and Dos Santos et al [\(2021\)](#page-11-15). Data from 158 male workers in the coal sector of the municipality of Candiota. Data of age (y), body mass (kg), and height (m) were collected from 158 individuals working in the coal sector and living in the municipality of Candiota (age:  $44.6 \pm 9.5$  years; body mass:  $85.4 \pm 15.2$  kg; height:  $1.7 \pm 8.1$  m; BMI:  $28.4 \pm 4.1$  kg/m<sup>2</sup>). More detailed information on these subjects is available in previous publications (da Silva Júnior et al., [2018;](#page-11-9) dos Santos et al., [2019\)](#page-11-16). This study comprised all male coal mining workers in the municipality. The body mass was assessed using a digital scale (Model: 210Q, Geratherm®, range 0.1–180 kg), and height was measured using a stadiometer with the individuals instructed to stay in a standing position without shoes. The body mass index (BMI) was calculated by dividing the body mass (kg) by the square of height  $(m<sup>2</sup>)$  (de Onis & Habicht, [1996\)](#page-11-17). This study followed the principles of the Declaration of Helsinki, and it was approved by Research Ethics Committee in Health 036/2013 (CEPAS/FURG). All subjects agreed to be included in our study by signing the informed consent.

# Air pollutants monitoring and meteorological parameters

PAH analysis was assessed in particulate matter with an aerodynamic diameter equal to or less than 10  $\mu$ m (PM<sub>10</sub>). The sampling location was situated in the district of Dario Lassance, which is a continuous monitoring point of the air quality of the Thermal Power Generation Company (*Companhia de Geração Térmica de Energia Elétrica—CGTEE*). This site was chosen because it is located close to the industrial settlements of CGTEE and within an urban area, which has high vehicle-fleet traffic liable to receive the effects of atmospheric contaminants. Monitoring campaigns occurred from July 2012 to March 2013. Fiberglass flters (Energética®—203 mm×254 mm) were used for  $PM_{10}$  sampling, which was balanced in a chamber flter for 24 h to stabilize temperature and humidity. After reaching equilibrium, the flters were weighed on a 5-digit analytical balance. These flters were used in a high-volume air sampler (Hi-Vol) (AGV  $MP_{10}$ , Energética®) placed 1 m from the ground, for 24 h. Figures S2A and S2B illustrate the details of the equipment used in the study (Fig. S1. Supplementary material). After collection, the flters were removed and packed with aluminum foil, and taken to the laboratory where they remained in the same temperature and humidity conditions. The flter was weighed, and the concentration was expressed in  $\mu$ g/m<sup>3</sup>. After, they were stored at 4 °C for further analysis. The analyses of PAHs in  $PM_{10}$  were based on the methodology described by the United States Environmental Protection Agency (USEPA, [1999](#page-12-13)). Before the extraction process,  $100 \mu l$  of an internal standard was added to the flters to measure the recovery of the extraction of hydrocarbons PIR HC (1—Eicosene, 1—Hexadecene, Androstanol, 1—Phenylnanodecene, Paraterphenyl—D14).

The extraction was carried out with the Soxhlet system adding 250 ml of dichloromethane for 12 h, at a temperature ranging  $40-50$  °C. About 2 ml of the samples were concentrated in a rotary evaporator and then placed in tubes. The extraction flask was washed with n-hexane  $(0.5-1$  ml) three times, and the solvent was transferred to the same test tube as the samples. The cleanup was carried out using column adsorption chromatography in a glass column. The column was impregnated with sodium sulfate ( $Na<sub>2</sub>SO<sub>4</sub>$ ), silica (SiO<sub>2</sub>), and alumina  $(A1, O<sub>3</sub>)$ , using n-Hexane as solvent. The silica and alumina were previously heated at a temperature of 160 °C for 3 h. Then, they were deactivated at 5% of the previous weight with MiliQ water, pre-washed with n-hexane. The extract was added to the column and a sequence of solvents was added. First, 25 ml of n-Hexane (F1) was added to remove the fraction of aliphatic hydrocarbons. Then, a mixture of dichloromethane/n-hexane was used to separate the aromatic fraction of hydrocarbons. This extract was then concentrated in a rotary evaporator at 40–50 <sup>o</sup>C and then measured 1 ml with N<sub>2</sub>. Then, 50  $\mu$ l of chromatographic standard was added, containing the following compounds: Acenaphetene—d10, Crysene—d12, Naphthalene—d8, Perylene—d12, Phenanthrene—d10 and 1,4 Dichlorobenzene—d4. For the determination of PAHs, a gas chromatograph (Perkin Elmer Clarus 600®) was used. The limits of detection and quantifcation and the recovery percent of measurement of the 16 PAHs are described in Table S2*—*Supplementary material.

The device was coupled with a mass spectrometer, with Elite-5MS column (5% Diphenyl—95% dimethylsiloxane) with a length of 30 m, 0.25 mm D.I., 0.25 μm DF, and an automatic sample injector. The used heating ramp was at 40 °C, increasing at a rate of 10  $\mathrm{^{\circ}C}$  min<sup>-1</sup> to 60  $\mathrm{^{\circ}C}$ , continuing heating at a rate of 5 °C min−1 to 290 °C, being maintained for 5 min; changing the heating rate from 10 °C min−1 to 300 °C, maintained for 10 min. For the detector, it was used a source 29 in a temperature of 200 °C and a transfer line temperature of 280 °C, using an electronic impact of 70 eV. The samples were injected in Splitless mode, with a flow of 50 ml of helium, after 1 min. The volume of the sample injected was  $1 \mu l$ , and the injector temperature was set at 280 °C with Helium gas as a carrier  $(1.5 \text{ ml } \text{min}^{-1})$ . The detection limit of the equipment was  $1 \mu g kg^{-1}$ . For the identification of PAHs, the SIR method (selected ion recording) was used.

Sulfur Dioxide  $(SO<sub>2</sub>)$  was measured by the Ultraviolet Fluorescence Method using a HORIBA brand ambient analyzer (model APSA-360). This method meets USEPA recommendations—Reference Method EQSA-0506-159.

The meteorological parameters of precipitation (mm), temperature (°C), humidity (%), and wind speed (m/s) were extracted from the National Institute of Meteorology (INMET). The seasons during the monitoring period were defned as Spring (September, October, and November), Summer (December, January, and February), Autumn (March, April, and May), and Winter (June, July, and August).

#### Health risk assessment

The health risk assessment was determined considering the US EPA values of body mass and inhalation rate to male eutrophic adults (reference values) and using the real value for the body mass of individuals living in the coal activity area (USEPA, [2011](#page-12-14)). We calculated the benzo(a)pyrene equivalent toxic equivalent quotient (BaP-TEQ) and the incremental lifetime cancer risk (ILCR) as measures to assess the health risk. The benzo(a)pyrene is the most toxic of the 16 HPAs with a well-established carcinogenic efect, once it is listed as a Group 1 carcinogen by the International Agency for Research on Cancer (IARC), and therefore, it is used to evaluate the risk of exposure to PAHs (IARC, [2013](#page-11-18)) The BaP-TEQ consists of the carcinogenic efect of the PAHs, and it was calculated considering each PAH concentration and the toxic equivalency factor (TEF) of PAHs, according to Eq. [1.](#page-4-0)

$$
BaP - TEQ = C_{HPA} \times TEF
$$
 (1)

in which  $C_{HPA}$  = concentration of PAHs (pg/m<sup>3</sup>); TEF=toxic equivalency factor (Table S1. Supplementary Material).

The BaP-TEQ of the annual mean of PAHs and each season of the year was used to calculate the daily inhalation of the BaP-TEQ, as seen in Eq. [2.](#page-4-1)

Daily inhalation = 
$$
(BaP - TEQ \times IR \times ET)/24
$$
 (2)

where BaP-TEQ: benzo(a)pyrene equivalent toxic equivalent quotient (ng/m<sup>3</sup>); IR: inhalation rate (m<sup>3</sup>/ day) of the USEPA value for normal-weight male  $(21–61 \text{ years})$  (USEPA,  $2011$ ); ET: exposure times (hours/day), we considered 24 h.

The incremental lifetime cancer risk provides an estimation of the risk that the population living in the coal area is exposed to. The ILCR was estimated as annual and seasonal values, considering potential air pollutants and meteorological fuctuations. The equation to estimate the ILCR follows (Mo et al., [2019\)](#page-12-15):

values; and inhalation rate  $(m^3/day)$  for normal-body mass (16.68 m<sup>3</sup>/day) and overweight (19.175 m<sup>3</sup>/day) individuals included in our study (USEPA, [2011](#page-12-14)).

 $CF =$ Conversion factor  $10^{-9}$  (mg/pg); ED = Exposure duration (years)—70 years; ET=Exposure times (hours/day)*—*24 h; EF=Exposure frequency (days/year)*—*365 days; CSF=Cancer slope factor (mg/kg-day)−1*—*we adopted the value 3.14 (Chen & Liao, [2006](#page-11-19)); BM=Body mass (kg)*—*we used the body mass of male normal-body mass individuals of USEPA (70 kg) to calculate ILCR with the references values and using the real values of BM collected from the individuals living in the coal activity area; AT=Averaging time of the 365 days of the 70 years of exposure (hours)*—*70 years x 365 days x 24 h.

According to USEPA, the ILCR values above  $10^{-6}$ suggest the point that implicates in potential cancer risk (USEPA, [2005\)](#page-12-12).

## Statistical analysis

<span id="page-4-1"></span><span id="page-4-0"></span>All variables underwent to a Kolmogorov–Smirnov normality test. The comparison among seasons of the PAHs concentration (ng/m<sup>3</sup>), BaP-TEQ, SO<sub>2</sub> concentration, and precipitation levels was analyzed using a nonparametric Kruskal–Wallis test followed by Dunn's post hoc test. The incremental lifetime cancer risk among the seasons was compared using a oneway analysis of variance (ANOVA) followed by Bonferroni's post hoc test. To compare the meteorological parameters of temperature, humidity, and wind speed among the seasons, a one-way analysis of variance (ANOVA) was followed by Bonferroni's post hoc test. A Pearson correlation was performed between the total and individual concentrations of PAHs. A principal component analysis (PCA) was performed considering the meteorological data and the levels of total PAHs and  $PM_{10}$  with the analysis based on the correlation index. The data were plotted considering the two most signifcant axes (factors). Furthermore, bivariate analyzes and multiple linear regression were

 $ILCR = (BaP - TEQ \times IR \times CF \times ED \times ET \times EF) \times (CSF/(BM \times AT))$  (3)

in which BaP - TEQ ∶ benzo(a) pyrene equivalent toxic equivalent quotient  $(pg/m^3)$ ; IR: inhalation rate  $(m<sup>3</sup>/day)$  of the USEPA value for normal-body mass male  $(21-61 \text{ years})$ —16.68 m<sup>3</sup>/day for reference performed between each PAHs (dependent variable) and other environmental variables (PM10, SO2, precipitation, temperature, humidity, and wind speed) (independent variables). Only data from those HPAs that had a signifcant correlation (bivariate analysis) with some environmental parameters were presented. The statistical analysis was carried out using Statistica 8.0 (Statistica Software, Tulsa, Oklahoma, USA) and GraphPad Prism 6.0 (GraphPad Software, San Diego, CA, USA). A value of *p*≥0.05 was considered statistically signifcant.

## **Results**

Descriptive analyses of the epidemiological data of the sample are summarized in Table [1.](#page-5-0) The PAHs concentration  $(ng/m<sup>3</sup>)$ , BaP-TEQ, SO2 concentration, and precipitation levels did not demonstrate a parametric distribution. The incremental lifetime cancer risk, the meteorological parameters of temperature, humidity, and wind speed had a parametric distribution. The annual concentrations of PAHs, PM<sub>10,</sub> and SO<sub>2</sub> were, respectively, 27.7 ng/m<sup>3</sup>, 26.3  $\mu$ g/m<sup>3</sup>, and 9.5  $\mu$ g/m<sup>3</sup>, and the concentrations

<span id="page-5-0"></span>**Table 1** Sample characterization of age and anthropometric measures

Variables	$Mean \pm SD$
Age $(y)$	$44.6 + 9.5$
Body mass (kg)	$85.4 + 15.2$
Height $(m)$	$1.7 + 8.1$
BMI $(kg/m2)$	$28.4 + 4.1$

Descriptive data showed a parametric distribution and is presented as mean  $\pm$  SD

*BMI* body mass index, *SD* standard deviation

<span id="page-5-1"></span>



along the monitoring campaigns are illustrated in Fig. [2](#page-5-1). The concentration of the 16 PAHs, the toxic equivalent factor of the components, and the respective BaP-TEQ values are described in Table S1 (Supplementary material). During the one-year air pollution monitoring in the coal activity area of Candiota, the BaP-TEQ value was 0.2 ng/  $m<sup>3</sup>$ , the daily inhalation was 62.4 ng/day, the ILCR calculated with reference values of USEPA was  $2.8 \times 10^{-6}$ , while the ILCR estimated using the real values of Candiota's inhabitants was  $2.6 \times 10^{-6}$ (Fig. [3\)](#page-6-0). The  $PM_{10}$  concentration among the seasons did not differ  $(p=0.23)$  $(p=0.23)$  $(p=0.23)$  (Table 2). The SO<sub>2</sub> concentration was higher in Summer and Autumn compared to Spring and Winter, and in Winter compared to Spring  $(p < 0.001)$  (Table [2](#page-6-1)). The ILCR estimated based on reference values of USEPA was  $3.8 \times 10^{-7}$  in Spring, 1.6  $10^{-6}$  in Summer,  $8.7 \times 10^{-6}$ in Autumn, and  $1.7 \times 10^{-6}$  in Winter (Table [2](#page-6-1)).

The comparison among the seasons showed no difference in PAH concentration  $(p=0.49)$  and BaP-TEQ values  $(p=0.50)$  (Fig. [3](#page-6-0)a, b). The daily inhalation of BaP-TEQ was 8.4 ng/day in Spring, 36.5 ng/ day in Summer, 193.2 ng/day in Autumn, and 37.3 ng/day in Winter (Fig. [3c](#page-6-0)). The ILCR using real values showed higher ILCR in the Autumn season compared to Spring, Summer, and Winter, and Summer and Winter had higher ILCR than Spring  $(p<0.001)$  (Fig. [3](#page-6-0)d). The ILCR distribution among male adults from the coal activity area demonstrated a higher number of individuals with ILCR ranging from  $2.2 \times 10^{-6}$  to  $2.8 \times 10^{-6}$ , consisting in 51.3% of our sample, while 86.1% was below the





<span id="page-6-0"></span>**Fig. 3** The seasonal comparison of **a** PAH concentration; **b** BaP-TEQ; **c** Daily inhalation; **d** ILCR. Data of PAH concentration and BaP-TEQ are presented as mean $\pm$ SEM. Difference between groups verifed by the Kruskal–Wallis test followed by Dunn's post hoc  $(p < 0.05)$ . Descriptive data of daily inhalation are presented as mean. Data of ILCR are presented as mean $\pm$ SD. The difference among groups verified by the

ANOVA test followed by  $(p < 0.05)$ . \*Denotes statistical difference compared to Spring. † Denotes statistical diference compared to Winter. ‡ Denotes statistical diference compared to Summer. PAH: polycyclic aromatic hydrocarbons; BaP-TEQ: benzo(a)pyrene equivalent toxic equivalent quotient; ILCR: incremental lifetime cancer risk

<span id="page-6-1"></span>

					<b>Table 2</b> Concentrations of $PM_{10}$ and $SO_2$ and ILCR (reference dose) distribution among the seasons of the year of a coal							
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Data of PM<sub>10</sub> showed a parametric distribution and are presented as mean $\pm$ SD. The difference among groups verified by the ANOVA test followed by Bonferroni's post hoc ( $p$ <0.05). Data of SO<sub>2</sub> concentration showed a non-parametric distribution and is presented as median (IQR). Difference between groups verified by the Kruskal–Wallis test followed by Dunn's post hoc  $(p < 0.05)$ 

 $MP_{10}$  particulate matter with the aerodynamic size of 10  $\mu$ m, *SO<sub>2</sub>* sulfur dioxide, *ILCR* incremental lifetime cancer risk

\*Denotes statistical diference compared to Spring

† Denotes statistical diference compared to Winter

value of ILCR that considered the USEPA reference values for calculation (Fig. [4\)](#page-7-0).

The meteorological parameters during the PAH monitoring period showed no diference in precipitation levels  $(p=0.90)$  among seasons, temperature was lower in Autumn and Winter compared to Spring and Summer  $(p < 0.001)$ . Humidity was higher in Winter compared to Spring  $(p=0.008)$ . The wind <span id="page-7-0"></span>**Fig. 4** Incremental lifetime cancer risk distribution among the male individuals living in the coal activity area of Candiota



speed was lower in Autumn compared to Spring, Summer, and Winter, and Winter had a higher wind speed compared to Summer  $(p < 0.001)$  (Table [3\)](#page-7-1).

There was a signifcant positive correlation between the total concentration of PAHs and Chrysene, Fluoranthene, Pyrene, Indeno(1,2,3)pyrene, Benzo(g, h, i)perylene, Dibenzo(a, h)anthracene, Benzo(a)pyrene, and Benzo(a)anthracene (Fig. [4a](#page-7-0)). The PCA showed a strong association between the concentration of total PAHs and  $PM_{10}$ , and an association between the concentrations of total PAHs,  $PM_{10}$ , and the concentration of  $SO_2$ . Contrarily, the wind speed was negatively associated with the levels of PAHs and  $PM_{10}$ . The significance of the axes was 30.6% (Factor 1) and 23.8% (Factor 2) (Fig. [4](#page-7-0)b).

Additionally, the acenaftilene had a negative correlation with temperature  $(r=-0.7, p=0.003)$ and a positive correlation with humidity  $(r=0.6,$  $p=0.01$ ), and benzo(a)pyrene had negative correlation with temperature  $(r=-0.5, p=0.04)$  and wind speed  $(r=-0.7, p=0.009)$ . Then, the multiple linear regression demonstrated a relation of acenaftilene with the temperature ( $\beta$ = − 0.6, *p* = 0.05), and humidity (β=0.2,  $p=0.4$ ), and a relation of benzo(a)pyrene with temperature ( $\beta$ = − 0.4, *p* = 0.39), and wind speed (β =  $-$  0.6,  $p$  = 0.01) (Table [4\)](#page-8-0) (Fig. [5](#page-8-1)).

#### **Discussion**

In the present study, the air pollution monitoring in the coal activity area of Candiota demonstrated an annual PAH concentration of 27.7 ng/m<sup>3</sup>, PM<sub>10</sub> concentration of 26.3  $\mu$ g/m<sup>3</sup>, SO<sub>2</sub> concentration of 9.5  $\mu$ g/  $m<sup>3</sup>$ , a BaP-TEQ value of 0.2 ng/m<sup>3</sup>, a daily inhalation of 62.4 ng/day, and ILCR values of  $2.8 \times 10^{-6}$ and  $2.6 \times 10^{-6}$  for estimation based on reference and real values, respectively. The comparison among seasons showed no diference in PAH concentration and BaP-TEQ. However, it was observed a higher daily

<span id="page-7-1"></span>**Table 3** Meteorological parameters during PAHs monitoring period (July of 2012–May of 2013) in a coal activity area

Meteorological parameters	Spring	Summer	Autumn	Winter	$p$ -value
Precipitation (mm)	$4.2 + 12.3$	$4.8 + 13.9$	$3.4 + 9.4$	$3.1 + 8.7$	0.9029
Temperature $(^{\circ}C)$	$20.9 \pm 3.1$	$21.5 \pm 3.1$	$16.6 + 3.6$ * <sup>*</sup>	$14+5.1$ **	< 0.001
Humidity $(\%)$	$70.8 + 11.7$	$73 + 7.5$	$75.4 + 10.4$	$75.8 + 8.1*$	0.008
Wind speed $(m/s)$	$3.6 + 1.4$	$3.2 \pm 1.1$	$2.5 + 1.4$ <sup>***</sup>	$3.9 + 1.6$ <sup>#</sup>	< 0.001

Data of temperature, humidity, and wind speed showed a parametric distribution and are presented as mean $\pm$ SD. The difference among groups verified by the ANOVA test followed by  $(p<0.05)$ . Precipitation levels showed a non-parametric distribution and were analyzed using the Kruskal–Wallis test followed by Dunn's post hoc  $(p < 0.05)$ 

\*Denotes statistical diference compared to Spring

† Denotes statistical diference compared to Winter

‡ Denotes statistical diference compared to Summer

<span id="page-8-0"></span>**Table 4** Bivariate analysis and multiple linear regression between Acenaftilene and Benzo(a)pyrene with environment variables

Variables		Bivariate analysis	Multiple linear regression		
	r	$p$ -value	β	$p$ -value	
Acenaftilene					
$PM_{10}$	$-0.1$	0.7			
SO <sub>2</sub>	$-0.3$	0.4			
Precipitation	$-0.1$	0.8			
Temperature	$-0.7$	0.003	$-0.6$	0.05	
Humidity	0.6	0.01	0.2	0.39	
Wind speed	0.4	0.2			
Benzo(a)pyrene					
$PM_{10}$	0.1	0.9			
SO <sub>2</sub>	$-0.2$	0.6			
Precipitation	0.07	0.8			
Temperature	$-0.5$	0.04	$-0.4$	0.05	
Humidity	0.5	0.1			
Wind speed	$-0.7$	0.009	- 0.6	0.01	

Bold values represents the results with statistical diference  $PM_{10}$  particulate matter 10,  $SO_2$  sulfur dioxide



<span id="page-8-1"></span>**Fig. 5** Analysis of total PAH with each of the 16 individuals HPA and with environmental variables. **a** Correlation plot between total PAHs and each individual HPA. The respective correlation (*r*) and *p*-values were: PAH<sub>total</sub>:Chrysene:  $r=0.7808$ ;  $p=0.0006$ ; PAH<sub>total</sub>:Fluoranthene:  $r=0.6655$ ;<br> $p=0.0068$ ; PAH<sub>total</sub>:Pyrene:  $r=0.7668$ ;  $p=0.0009$ ; *PAH*<sub>total</sub>:Pyrene: PAH<sub>total</sub>:B(b)f: *r*=0.3573; *p*=0.1911; PAH<sub>total</sub>:Indeno(1.2.3) pyrene:  $r=0.9853$ ;  $p=0.0000$ ; PAH<sub>total</sub>:Benzo(g, h, i)perylene:  $r = 0.9598$ ;  $p = 0.00000002$ ; PAH<sub>total</sub>:Dibenzo(a, h)anthracene:<br> $r = 0.9082$ ;  $p = 0.000003$ ; PAH<sub>total</sub>:Benzo(k)fluoranthene: PAH<sub>total</sub>:Benzo(k)fluoranthene:  $r=0.1216$ ;  $p=0.6660$ ; PAH<sub>total</sub>:Acenaftilene:  $r=0.0808$ ;

inhalation of BaP-TEQ and incremental lifetime cancer risk in the autumn season, with the majority of the ILCR of individuals included in this study ranging from  $2.2 \times 10^{-6}$  to  $2.8 \times 10^{-6}$ . In the Autumn season, the temperature and wind speed were lower than the levels detected in the seasons' Spring and Summer. Also, the total PAHs were associated with  $PM_{10}$  and  $SO<sub>2</sub>$  concentrations, the wind speed was negatively associated with the levels of PAHs and  $PM_{10}$ , while acenaftilene had a negative correlation with temperature and a positive correlation with humidity, and benzo(a)pyrene had a negative correlation with temperature and wind speed.

Studies investigating the PM-bounded PAHs showed large heterogeneity of results related to the concentration of these air pollutants. This variability seems to be connected with several factors, as the geographic region, meteorological conditions and aerodynamic size of the pollutant at the moment of collection. A study conducted in Bangkok (Thailand) demonstrated that the annual mean of  $PM_{10}$ concentration of 3 polluted areas was  $49.1 \text{ ug/m}^3$ . Despite the higher levels of  $PM_{10}$ , the concentration of bounded PAH and BaP-TEQ values (3.7 ng/



*p*=0.7746; PAH<sub>total</sub>:Anthracene: *r*=− 0.1302; *p*=0.6436; PAH<sub>total</sub>:Fluorene:  $r = -0.0230$ ;  $p = 0.9352$ ; PAH<sub>total</sub>:Benzo(a) pyrene:  $r=0.7050$ ;  $p=0.0033$ ; PAH<sub>total</sub>:Benzo(a)anthracene: *r*=0.7982; *p*=0.0004; PAH<sub>total</sub>:Naphthalene: *r*=− 0.1989;  $p=0.4772$ ; PAH<sub>total</sub>:Phenanthrene:  $r=0.2570$ ;  $p=0.3552$ ; PAH<sub>total</sub>:Acenaphene:  $r = -0.4039$ ;  $p = 0.1354$ . **b** PCA of total PAHs and  $PM_{10}$  levels and meteorological data with the analysis based on the correlation index. The data were plotted considering the two most signifcant axes (factors), the signifcance of the axes was 30.6% (Factor 1) and 23.8% (Factor 2)

 $m<sup>3</sup>$  and 0.4 ng/m<sup>3</sup>, respectively) was lower than that found in our study (Tadsanaprasittipol et al., [2021](#page-12-16)). Previously, another study also realized in Bangkok observed a PAH annual concentration of  $12.6 \text{ ng/m}^3$ , approximately half of the concentration found in our study, but a BaP-TEQ value of  $2.1 \text{ ng/m}^3$  (Norramit et al., [2005\)](#page-12-17). Still, in the same region, an investigation showed a  $PM_{10}$  annual concentration of 66.0 ug/  $m<sup>3</sup>$  followed by a PAH concentration of 60.8  $\text{ng/m}^3$ (Thongsanit et al., [2003\)](#page-12-18). Regarding PAH bounded in  $PM<sub>2.5</sub>$ , a study from China showed a mean annual concentration of  $PM_{2.5}$  of 204  $\mu$ g/m<sup>3</sup> and PAH concentration of 66.2 ng/m<sup>3</sup>. As expected, the BaP-TEQ value also demonstrated a high value  $(16.6 \text{ ng/m}^3)$  at the expense of the great  $PM_{2.5}$  concentration (Chao et al., [2019](#page-11-20)).

Regarding the variability factors infuencing the concentration of PM-bounded PAH, it is common to fnd a higher concentration of PAH in urban areas; however, other areas containing great sources of PAHs could equally demonstrate augment levels. This trend was observed by Tadsanaprasittipol et al. [\(2021](#page-12-16)), showing a higher BaP account in roadside sites than in urban and industrial ones, as observed in the geolocation of area Candiota City. Additionally, PAH emissions are closely related to the socialeconomic population level. In Candiota city, the per capita income was 2221\$ in the period of study development (2013), which is substantially lower compared to the capital of Rio Grande do Sul State, Porto Alegre (6922\$) (IBGE, [2021\)](#page-11-21), so it indicates that Candiota City is a vulnerable area considering its social-economic status. Meteorological conditions also infuence air pollutants levels, as signifcantly lower concentrations were observed in hot and humid periods, while higher average total carcinogenicity was observed during the winter (Tadsanaprasittipol et al., [2021](#page-12-16)). Several studies also found higher PAH concentrations in the coldest season, contributing to higher daily inhalation of BaP-TEQ and incremental lifetime cancer risk (Hsu, [2019](#page-11-22); Jyethi et al., [2014](#page-11-23); Stracquadanio et al., [2007;](#page-12-19) Tadsanaprasittipol et al., [2021\)](#page-12-16). In our study, PAH concentration, daily inhalation of BaP-TEQ, and incremental lifetime cancer risk were higher in cold period, once the thermal inversion phenomenon increase the concentration of  $PM_{10}$ and PAH bounded to it.

In our study, a signifcant positive correlation was observed between the total PAHs and Chrysene,

Fluoranthene, Pyrene, Indeno(1,2,3)pyrene, Benzo(g, h, i)perylene, Dibenzo(a, h)anthracene, Benzo(a)pyrene, and Benzo(a)anthracene. Those compounds had a high molecular weight and present 4 to 6 rings in their composition compared to other PAHs. Moreover, they are classifed with increased carcinogenic potency (Norramit et al., [2005;](#page-12-17) USEPA, [2005](#page-12-12)). Likewise, Chrysene, Fluoranthene, Benzo(a)pyrene, and Benzo(a)anthracene are considered markers of coal combustion and were associated with the total PAH in our study, ensuring that the activities related to coal in Candiota area were the sources of the PAHs (Larsen & Baker, [2003](#page-12-20), Abdulazeez & Fantke, [2017](#page-10-0)). Rekefu et al. (2018) demonstrated that similar annual concentrations of different PM sizes (PM $_{2.5}$  20.9 ng/  $m^3$  and PM<sub>2.5</sub>–10 20.9 ng/m<sup>3</sup>) resulted in very distinct BaP-TEQ values  $(10.5 \text{ ng/m}^3 \text{ and } 1.2 \text{ ng/m}^3 \text{, respectively.}$ tively). Once each PAH has a diferent carcinogenic potential, the carcinogenic risk of PAH is not necessarily associated with total pollutant concentration showing the relevance of dosing isolated compounds (Norramit et al., [2005\)](#page-12-17).

The individuals living in the coal activity area of Candiota are exposed to potential cancer risk, once the ILCR values exceed the limit of 10–6 (USEPA, [2005\)](#page-12-12). Individuals living in areas with coal extraction and combustion are more susceptible to developing cancer, once PAHs concentration, which is carcinogenic compounds, is higher because of the coal-related activities. Epidemiological studies, systematic reviews, and geospatial analysis of cancer risk showed that the PAH emissions from coal activity contributes to the etiology of cancer once it was observed a positive association between all cancer mortality, lung cancer incidence, and mortality with proximity to a coal mine. The carcinogenic efect caused by PAH emitted by coal activity could afect population due to occupational and environmental exposure, including the geographical proximity as a risk factor for the residents in these areas (Bigliardi et al., [2021;](#page-11-7) Cortes-Ramirez et al., [2018](#page-11-8); da Silva Júnior et al., [2018](#page-11-9); Fernández-Navarro et al., [2012;](#page-11-24) Jenkins et al., [2013;](#page-11-25) Mueller et al., [2015\)](#page-12-21). The ILCR values estimated considering the reference values of USEPA for inhalation rate and body mass were higher than the ILCR using the real values (USEPA, [2011](#page-12-14)).

Our results showed that 86.1% of our sample was below the value of ILCR that considered the USEPA reference values for calculation. This diference

indicates the importance to consider information from the local population. We used the body mass index of each individual included in our study to estimate their inhalation rate to calculate the ILCR. Overweight individuals have a higher inhalation rate than lean individuals, as defned by USEPA, [2011](#page-12-14). However, overweight individuals did not present a higher exposure because the equation is relativized by the body mass.

As a limitation of our study, we cited the monitoring campaigns frequency, which did not have a homogeneous scenario during all the seasons of the one-year monitoring period. Moreover, there was a lack of data concerning coal mining and power plant activities. Although it was an advance to include local epidemiological data, it was limited to the only data available, a male adult population. Moreover, we did not investigated the dietary intake of the population enrolled in the study; therefore, it is not included the PAHs intake in the exposure. The technique to meal preparation with grilling and/or frying would produce high levels of PAHs, due to the pyrolysis of fat at higher temperature and adsorption of PAHs emitted from combustion process. Additionally, the PAHs concentration in raw foods may come from the deposition of ambient particles, contaminated soils and water (Alomirah et al., [2011;](#page-10-2) Duan et al., [2016](#page-11-26)).

### **Conclusion**

Our study demonstrated that Candiota city area had substantial emissions of air pollutants, including toxic components with carcinogenic potentials, like  $PM_{10}$ and PAHs. This emission could be related to the coal activity and burning activities performed in this area, once these pollutants are emitted by coal combustion. Consequently, the studied population had ILCR levels above the reference limit of  $10^{-6}$ , indicating cancer risk. The observed risk in the population sets an alert for the harmful consequences of the use of this energy source. Therefore, an epidemiological survey of cancer cases in the region and its relationship with environmental exposure and air pollutants levels must be mandatory. Especially because there is a lack of public policies in Brazil focusing on air pollutants emissions monitoring, highlighting the need for free disclosure and interpretation of data collected by air

quality monitoring stations administered by public or private companies.

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**Authors' contribution** BM involved in conceptualization, methodology, formal analysis, writing—original draft, and writing—review and editing. RBC involved in conceptualization, methodology, formal analysis, writing—original draft, and writing—review and editing. ALM-B involved in methodology. PRMB involved in methodology. MS involved in methodology. EMG involved in methodology. CRR involved in conceptualization, writing—original draft, writing—review and editing. FMRSJ involved in conceptualization, methodology, formal analysis, writing—original draft, and writing—review and editing.

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**Availability of data and materials** The datasets used and/or analyzed during the current study are available from the corresponding author on reasonable request.

#### **Declarations**

**Confict of interest** The authors declare that they have no known competing fnancial interests or personal relationships that could have appeared to infuence the work reported in this paper.

**Ethical approval** This study followed the principles of the Declaration of Helsinki and it was approved by the Research Ethics Committee in Health 036/2013 (CEPAS/FURG).

**Consent to participate** Informed consent was obtained from all individual participants included in the study.

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