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Multiple metal(loid)s bioaccessibility from cooked seafood and health risk assessment

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Abstract Seafood has been generally considered to be the main diet exposure source of metal(loid)s. We evaluated health risk of mercury (Hg), arsenic (As), cadmium (Cd), lead (Pb), chromium (Cr), nickel (Ni), copper (Cu), and zinc (Zn) through consumption of cooked seafood based on bioaccessibility, which was obtained by physiologically based extraction test method. Results showed that cooking practices could decrease metal(loid)s concentration from seafood (by 6.0-45.7%). Metal(loid)s release from seafood in this study followed the descending order of Hg > Zn > Ni > Cd > Pb > As > Cu > Cr. On average,

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W. Liao · W. Zhao · Y. Wu · N. Rong · X. Liu · K. Li · G. Wang (⊠) State Environmental Protection Key Laboratory of Water Environmental Simulation and Pollution Control, South cooking lowered the bioaccessibility of As, Hg, Cd, Pb, Ni, Cr, Cu, and Zn by 15.2, 26.1, 30.9, 30.7, 25.7, 31.2, 17.6, and 22.4%, respectively. Health risk calculation results showed that Cr, Ni, and Zn in seafood species in this study were within the human health benefits range. Hg, Cd, Pb, and Cu exposure from cooked seafood was within the safe dose. However, we found that there is a potential of having cancer (especially bladder and lung cancer) for people exposure to iAs from seafood based on bioaccessible contents the first time.

Keywords Fish · Crustacea · Cooking · Bioaccessibility · Health risk

List of symbols

Concentration of metal(loid)s in food C_i $(\mu g g^{-1})$ Daily consumption of food $(g d^{-1})$ DC BW Average body weight (kg) THO Target hazard quotient (unitless) EF Exposure frequency (days/year) ED Exposure duration (years) The tolerable daily intake ($\mu g d^{-1} k g^{-1}$) RfD AT Average exposure time (days) TR Target cancer risk (unitless) Cancer slope factor (mg kg⁻¹ d⁻¹)⁻¹ CFS LSD Least significant difference SD Standard deviation DL Detected level

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SRM	Standard reference material
HPLC	High-pressure liquid chromatography
ICP	Inductively coupled plasma
MS	Mass spectrometry
dw	Dry weight
WW	Wet weight

Introduction

Metal(loid)s, including heavy metals and metalloid, are commonly found in the environment, and some of them are highly toxic, persistent, and non-degradable and deemed serious pollutants (El-Kady and Abdel-Wahhab 2018). Metal(loid)s in the environment originate from natural processes as well as anthropogenic activities (Ip et al. 2005). Some metal(loid)s (e.g., zinc, nickel) are essential for human health but may be toxic when taken in excess (Chen et al. 2018). However, other metal(loid)s such as arsenic, lead, mercury, and cadmium are extremely detrimental even at low contents and some are even classified as class 1 non-threshold carcinogen (de Filipps et al. 1979; Fowler et al. 2015; Gao et al. 2018).

Except for occupational exposure, dietary intake of heavy metals via food has been believed to be the predominant way of human exposure to environmental metal(loid)s (Chen et al. 2018). It has been reported that metal(loid)s contents are high in certain types of fish although trace levels can be found in nearly all types of seafood (FAO 2014; Gu et al. 2015; Hu et al. 2018; Keshavarzia et al. 2018). Thus, seafood has been generally considered to be the main diet exposure source of metal(loid)s. The health risk assessment of metal(loid)s from food has been widely studied in recent years (Burger et al. 2002; Nfon et al. 2009; Sobhanardakani 2017; Alamdar et al. 2017; Hu et al. 2018; Keshavarzia et al. 2018; Zhong et al. 2018). Researchers usually investigated the health risk of target heavy metals based on their initial contents in food matrices (Islam et al. 2015). However, food is consumed cooked in most cases. It has been revealed that food processing practices affect (lower or increase) the concentration of the elements (Atta et al. 1997; Schmidt et al. 2015; Praveena and Omar 2017; Houlbrèque et al. 2011; Hajeb et al. 2014). For instance, it has been shown that washing could release part of the arsenic from food (Naito et al. 2015). Perelló et al. (2008) reported that the metal(loid)s contents in meat and fish reduced after cooking processes, especially for mercury. Schmidt et al. (2015) found that frying could decrease about 33% of the total mercury from fish.

Bioaccessibility is a maximum bioavailability of the target element and can be conducted in vitro (Moreda-Piñeiro et al. 2011). The bioaccessibility of many metal(loid)s was lower than 100% in general, which has been demonstrated by previous studies (EFSA 2012; Alves et al. 2018). It would be less realistic to use the initial contents of metal(loid)s for health risk assessment. Moreover, in our previous study, we found that methylated As in shellfish could be demethylated and brought about the increase in iAs (Liao et al. 2018). Hence, it is highly necessary to evaluate the risk of health exposure to metal(loid)s from seafood based on bioaccessibility. Besides, it has been also reported that cooking could result in a change of metal bioaccessibility from food. For instance, Amiard et al. (2008) showed that cooking reduced metal(loid)s bioaccessibility in shellfish. Nonetheless, to our knowledge, there was still a lack of health risk assessment of mixed metals from seafood after processing and in vitro gastrointestinal digestion.

Consequently, the present study aimed to solve problems as follows: (1) to investigate the impact of various cooking processes on the contents and bioaccessibility of metal(loid)s in seafood; (2) to assess multiple metal(loid)s bioaccessibility from raw and cooked seafood; (3) to evaluate the health risk of metal(loid)s from seafood in Guangzhou City after cooking and in vitro gastrointestinal digestion on the basis of target hazard quotient as well as target carcinogenic risk.

Materials and methods

Samples

Seafood samples, including raw fish (tilapia, 3×5 samples; snapper, 3×5 samples; salmon, 3×5 samples; pomfret, 3×5 samples; croaker, 5×5 samples; turbot, 3×5 samples; tiger grouper, 3×5

samples; and anchovy, 5×5 samples) and raw crustacea (mantis shrimp, 2.0 kg of samples; red shrimp, 2.0 kg of samples; prawn, 2.0 kg of samples; spotted crab, 3×5 samples; red crab, 3×5 samples; clam, 4.0 kg of samples; scallop, 4.0 kg of samples; and oyster, 4.0 kg of samples) were purchased randomly from both markets and grocery from Guangzhou City in China. Shells, skin, and bones of the collected samples were removed. Tools used for cutting and homogenizing tissues were all stainless. In consideration of the initial contents of the analyzed elements, part of seafood samples were chosen for processing, including boiling, steaming, baking, frying. The chosen food tissues for cooking were uniform in shape (radius = 4 cm; thickness = 1 cm). In the boiling process, the seafood tissues were soaked in boiling ultrapure water (100 °C) for 15 min in a Teflon saucepot. For steaming, the seafood tissues were processed for 10 min in a steamer under temperature of 100 °C. The mass ratio of water to seafood ratio was 1:1 for boiling and steaming. Baking was performed in a commercial baking oven without anything for 10 min under temperature of 130 °C. In the process of frying, the seafood tissues was spread in a Teflon pan with peanut oil for 10 min (the ratio of oil to seafood ratio was 0.5:1) and the oil temperature was up to 125 °C. All samples (raw and cooked) were divided into two subsamples: One part was stored in a freezer under temperature of -80 °C immediately, and the other part was freeze-dried firstly (for 48 h at - 80 °C and 0.770 Pa) and afterward stored in the same freezer until further experiments.

Metal(loid) concentrations are shown as $\mu g g^{-1}$ wet weight (ww) or $\mu g g^{-1}$ dry weight (dw). Based on the fresh and dry weights, the moisture contents in seafood could be obtained.

Reagents

High-purity water (resistivity was above 18.2 m Ω cm), which was produced from a water purification system named Milli-Q Element from USA, was used to prepare all solutions. Nitric acid was of chromatographically pure grade. Other chemicals were of guaranteed reagent grade (GR).

In vitro digestion

A physiologically based extraction test (PBET) model was used to study the bioaccessibility of metal(loid)s from raw and cooked homogenized seafood samples. The detailed information of the PBET model was described in our previous study (Liao et al. 2018). Gastric digestion phase: Raw or cooked dried seafood samples (5.0 g) were transferred into a brown bottle, which contained 500 mL of gastric digestion solution (500 mL of ultrapure water with 0.625 g of porcine pepsin, 210 µL of lactic acid, 0.25 mL of glacial acetic acid, and 0.25 g of malic acid, pH value was 2). Then, argon gas was ventilated into the solution in a shaking incubator at 150 rpm with stable temperature (37 $^{\circ}$ C) for 1 h, and the digestion liquid was sampled. Intestinal phase: The pH value was regulated to 5.3 using saturated sodium bicarbonate, and then, porcine bile (mass ratio of seafood samples to bile mass was 1:0.175) and pancreatin (mass ratio of seafood samples mass to pancreatin mass was 1:0.05) were added. Afterward, the pH value was regulated to 7 using $1.0 \text{ mol } \text{L}^{-1}$ of sodium hydroxide and the mixture was digested for 2 h (150 rpm and 37 °C). After gastric and gastrointestinal digestion, the corresponding bioaccessible fractions were centrifuged $(5000 \times g)$ for 20 min and then filtered through a filter (0.22 μ m) for testing.

$$Bioaccessibility (\%) = \frac{Bioaccessible concentration}{Total concentration in seafood \times 100\%}$$
(1)

Determination of metal(loid)s

Extraction of total contents of metal(loid)s in raw and cooked seafood samples followed previously described method (Liao et al. 2018; Liao et al. 2019a, b): 0.2 g of samples were added into Teflon reactors, followed by 5 mL of 14 mol L^{-1} HNO₃ and stand for 6 h. Next, 1 mL H₂O₂ (30% v/v) was added into the mixture, followed by being incubated in a microwave system for 30 min (1600 W, 180 °C).

The metal(loid)s concentrations in seafood samples and the bioaccessible fraction were determined using a quadrupole inductively coupled plasma mass spectrometry unit (ICP-MS; Agilent 7700x, Agilent, USA) 4040

Item	As	iAs	Hg	MeHg	Pb	Cd	Cu	Ni	Cr	Zn
RfD ($\mu g d^{-1} k g^{-1}$)	30	2.1	0.5	0.2	4	1	40	20	1500	300
CFS $(mg kg^{-1} d^{-1})^{-1}$	-	1.5 (skin cancer) 25.7 (bladder and lung cancer)	-	-	-	-	-	-	-	-
DC (g d^{-1})	41.3	(fish) and 31.2 (crustacea)								
BW (kg)	60									
ED (years)	70									

Table 1 The parameters used for health/risk calculation

(Dolan et al. 2003). An environmental calibration standards solution of multi-elements (Agilent, USA) was used for the determination of metal(loid)s in seafood simultaneously by ICP-MS. Concentration of Hg, As, Cd, Pb, Cr, Ni, Cu, and Zn in blank solution (5% HNO₃) was determined for 11 times, and the corresponding mean value was 0.027, 0.0049, 0.015, 0.0096 0.0035, 0.0073, 0.014, and 0.016 μ g L⁻¹, respectively. And the detection limit of the metal(loid)s was the value of three times of the standard deviation of the concentration in 5% HNO₃ solution (Klaue et al. 1999). The detection limit of Hg, As, Cd, Pb, Cr, Ni, Cu, and Zn was 0.0135, 0.0022, 0.0013, 0.0014, 0.0119, 0.0023, 0.0058, and 0.0082 $\mu g \; L^{-1},$ respectively. Internal standard elements were ⁴⁵Sc, ⁷²Ge, ¹¹⁵In, and ²⁰⁹Bi. Extraction and determination procedure of metal(loid)s in the certified reference materials [GBW 10068, GBW 10024, and GBW 10050, China; DORM-4, fish protein, Canada] were the same with samples. Recovery rates of these metal(loid)s ranged from 86 to 112%.

HPLC-ICP-MS method was use to detect the contents of inorganic As (arsenious acid, i.e., iAs^{III} and arsenic acid, i.e., iAs^V) and methylmercury (MeHg) (Liao et al. 2018). For determination of inorganic As (iAs), A Prin-cen Specia Fast Column $(4.6 \text{ mm} \times 100 \text{ mm}, \text{Prin-cen Scientific Ltd., China})$ and mobile phase of ammonium nitrate solution (NH₄NO₃) with dual concentrations alternately (solution concentration A: 8 mmol L^{-1} ; concentration B: 20 mmol L^{-1} ; injection volume: 30 µL; flow rate: 1.2 mL min^{-1}) were used. For identification of MeHg, CNW Athena C18 а Column $(4.6 \times 150 \text{ mm}, 5 \mu\text{m}, 120 \text{ Å}, \text{ANPEL Laboratory})$ Technologies (Shanghai) Inc., China) and single mobile phase (solution: $0.6 \text{ mol } L^{-1}$ ammonium acetate with 0.1% L-cystine and 5% methanol; injection volume = $80 \ \mu$ L; flow rate = 0.9 mL min⁻¹) were adopted.

Preparation of calibration standards for the iAs^{III} and iAs^V used stock solution of arsenious acid (GBW 08666) and arsenic acid (GBW 08667), respectively. Detection limits for the iAs^{III} and iAs^V on the ICP-MS instrument were 0.0083 and 0.0052 μ g L⁻¹, respectively. The extraction of iAs was as follows: Raw and cooked seafood samples (0.5 g) were transferred into centrifuged tubes containing 20 mL of 0.01 mol L^{-1} HNO₃ solution, followed by ultrasonic-assisted extraction for an hour. Then, the mixture was centrifuged for 20 min $(5000 \times g)$ and the supernatant was filtered through a cellulose acetate disk filter (0.22 μ m) for testing. Extraction procedure of iAs from the standard reference material (SRM 1568b, rice flour, USA) and blank solutions was the same with samples. The recovery rate for iAs in SRM 1568b was 80-115%.

Preparation of calibration standards for MeHg used MeHg chloride stock solution (o2si smart solutions[®]; 1000 mg L^{-1}). Concerning the extraction of MeHg, seafood samples (0.5 g) were transferred into centrifuged tubes containing 20 mL of 2 mol L^{-1} HCl solution, followed by ultrasonic-assisted extraction for an hour. Then, the mixture was centrifuged for 20 min $(5000 \times g)$. Sampled 2 mL of the supernatant afterward was added to Hg species separation mobile phase carrier solution (2 mL), and the pH was adjusted to 7.0 with 0.310 mL ammonium hydroxide (25%). Next, a cellulose acetate disk filter (0.22 µm) was used to filter the solution for next testing. Extraction procedure of MeHg from the certified reference material (CRM DORM-4, fish protein, Canada) and blank solutions was the same with samples. The recoveries of MeHg in DORM-4 were 90-113%.

The concentration of all the elements in used oil was also determined by ICP-MS, which were below the detected line.

Health benefit/risk calculation

To calculate the resident daily intake of all metal(loid)s through consumption of seafood in Guangzhou City, the established daily intake (EDI) was used in this study and calculated by the following formula:

$$EDI = \frac{C_i \times DC}{BW}$$
(2)

where EDI is the established daily intake of metal(loid)s from seafood (μ g d⁻¹ kg⁻¹), C_i is the concentration of metal(loid)s in food (μ g g⁻¹), and DC is the daily consumption of food (g d⁻¹). Fish and crustacea consumption in Guangzhou was 41.3 g d⁻¹ and 31.2 g d⁻¹ for adults, respectively (Ma 2004). BW is average body weight (60.0 kg for adults in Guangzhou City) (Wang 2005; MEP, 2013).

To calculate the hazardous exposure to metal(loid)s via consumption of seafood by the consumers, the target hazard quotient (THQ) was obtained used the following formulas (Zhong et al. 2018):

$$THQ = \frac{EF \times ED \times EDI}{RfD \times AT}$$
(3)

where THQ is the target hazard quotient (unitless), and EF is the exposure frequency (365 days/year). ED is exposure duration (70 years) (Wang 2005), i.e., the average lifetime. RfD is the tolerable daily intake of metal(loid), and the value for As, iAs, Hg, MeHg, Pb, Cd, Cu, Ni, Cr, and Zn is 30, 2.1, 0.5, 0.2, 4, 1, 40, 20 1500, and 300 μ g d⁻¹ kg⁻¹, respectively (Ke-shavarzia et al. 2018; Sobhanardakani 2017; Hu et al. 2018). AT is the average exposure time (365 days/year × ED).

Besides, we calculated the total THQ (tTHQ, unitless) of metal(loid) in seafood through the following equation (Sobhanardakani 2017):

$$tTHQ = THQ(1) + THQ(2) + \dots + THQ(n)$$
(4)

Among the studied elements, arsenic and inorganic arsenic compounds are identified as carcinogenic to humans (being classified as Group 1 in the IARC Monographs) (IARC 2018). Target cancer risk (TR, unitless) for iAs could be calculated by the following formulas (Zhong et al. 2018):

$$TR = \frac{EF \times ED \times EDI \times CSF}{AT} \times 10^{-3}$$
(5)

where TR is the target cancer risk, and CFS is the cancer slope factor. USEPA proposed a CFS value of 1.5 (mg kg⁻¹ d⁻¹)⁻¹ (skin cancer) and 25.7 (mg kg⁻¹ d⁻¹)⁻¹ (bladder and lung cancer) for iAs (Hu et al. 2018). The acceptable TR ranged from 10^{-4} to 10^{-6} (Gao et al. 2018).

The parameters used for health/risk calculation are shown in Table 1.

Statistical analyses

SPSS 21.0 (SPSS, Chicago, IL, USA) and OriginLab 9.0 (OriginLab, USA) were used for statistical analysis. A one-way analysis of variance (ANOVA), followed by the least significant difference (LSD) test were served to describe differences in values among treated groups. All data in this study are shown as the means or the mean \pm standard deviation (SD). If p < 0.05, the means were considered significantly different.

Results and discussion

Contents of metal(loid)s in raw and cooked seafood

The concentrations of metal(loid)s ($\mu g g^{-1} ww$) were determined in the raw seafood samples collected in Guangzhou market and are presented in Table 2.

We found a wide range of concentrations of arsenic (As) in the seafood samples, with the highest mean concentration of As in crabs (35.30 μ g g⁻¹ ww) and the lowest in clam (0.15 μ g g⁻¹ ww). It was found that iAs could be detected in almost all seafood samples in this study, and iAs was mainly in the form of iAs^V in raw samples. Nonetheless, iAs contributed far less to the tAs in seafood samples: The contents of iAs in the edible muscle of fish, shrimp, crab, and shellfish samples were < DL-0.018, 0.00078-0.0085,< DL-0.0066, and 0.011-0.022 µg·g⁻¹ ww, respectively. All values of iAs in seafood samples were under the limit recommended by the Chinese National Standard Agency (GB 2762-2017): The limit contents of iAs content in fish and other seafood were 0.1 and $0.5 \ \mu g \ g^{-1}$ ww, respectively. The contents of iAs in

				(p							
Samples amount)	Moisture/ %	$ As \\ (\mu g \ g^{-1}) $	iAs (ng g ⁻¹)	${\rm Hg}_{({\rm ng}~{\rm g}^{-1})}$	$\begin{array}{c} MeHg \\ (ng \ g^{-1}) \end{array}$	Pb (ng g^{-1})	Cd (ng g^{-1})	Cu ($\mu g g^{-1}$)	Cr (µg g ⁻¹)	Ni (µg g ⁻¹)	Zn ($\mu g \ g^{-1}$)
ish											
Tilapia (15)	80	0.19 ± 0.024	18.03 ± 1.5	8.01 ± 0.84	4.81 ± 0.42	11.04 ± 0.46	0.22 ± 0.02	0.16 ± 0.064	0.42 ± 0.054	0.10 ± 0.017	5.59 ± 0.42
Snapper (15)	79	1.13 ± 0.076	6.81 ± 0.82	120.01 ± 3.81	84.08 ± 0.29	0.61 ± 0.12	0.57 ± 0.50	0.17 ± 0.020	0.73 ± 0.044	0.16 ± 0.063	3.19 ± 0.68
Salmon (15)	67	0.97 ± 0.17	< DL	36.11 ± 6.31	29.20 ± 1.81	0.40 ± 0.16	0.20 ± 0.066	0.28 ± 0.15	1.13 ± 0.020	0.19 ± 0.010	4.01 ± 0.20
Pomfret (15)	67	2.01 ± 0.069	16.03 ± 4.00	36.20 ± 1.25	22.10 ± 0.76	90.00 ± 0.76	1.21 ± 0.36	0.30 ± 0.025	0.98 ± 0.011	0.30 ± 0.011	6.01 ± 0.011
Croaker (25)	61	0.89 ± 0.086	13.11 ± 4.31	59.02 ± 5.12	55.00 ± 0.74	3.72 ± 0.47	0.47 ± 0.43	0.20 ± 0.082	1.58 ± 0.010	0.094 ± 0.051	5.01 ± 0.55
Turbot (15)	82	1.57 ± 0.19	2.50 ± 0.47	11.04 ± 1.11	8.61 ± 0.50	0.85 ± 0.12	0.50 ± 0.34	0.12 ± 0.054	0.86 ± 0.055	0.11 ± 0.036	5.31 ± 1.74
Tiger grouper (15)	LL	0.86 ± 0.019	11.22 ± 0.85	60.08 ± 10.11	46.00 ± 2.52	9.40 ± 8.50	1.02 ± 0.11	0.12 ± 0.11	2.04 ± 0.053	0.085 ± 0.015	4.90 ± 2.16
Anchovy (15)	81	0.54 ± 0.019	10.10 ± 0.36	4.62 ± 0.55	3.40 ± 0.61	8.40 ± 0.65	4.04 ± 0.21	0.29 ± 0.082	0.68 ± 0.036	0.036 ± 0.0044	6.28 ± 0.023
Crustacea											
Mantis shrimp (2 kg)	81	9.17 ± 0.061	0.78 ± 0.21	25.14 ± 1.33	25.11 ± 0.21	11.00 ± 0.49	470.09 ± 4.41	11.91 ± 0.080	0.053 ± 0.012	0.15 ± 0.025	23.59 ± 0.44
Red shrimp (2 kg)	80	9.91 ± 0.156	1.30 ± 0.38	20.10 ± 0.68	17.19 ± 0.22	1.90 ± 0.24	40.21 ± 0.38	1.63 ± 0.084	0.043 ± 0.0056	0.044 ± 0.0034	10.43 ± 0.03
Prawn (2 kg)	LL	0.21 ± 0.053	8.50 ± 0.85	17.23 ± 3.11	11.01 ± 0.53	2.10 ± 0.39	4.61 ± 0.56	3.27 ± 0.10	0.032 ± 0.0041	0.014 ± 0.0037	9.96 ± 0.15
Spotted crab (15)	80	35.30 ± 0.25	< DL	70.11 ± 2.04	64.22 ± 0.22	12.04 ± 0.46	58.23 ± 0.36	15.67 ± 0.20	0.21 ± 0.0042	0.12 ± 0.038	118.84 ± 0.16
Red crab (15)	84	33.34 ± 0.32	6.60 ± 0.45	66.19 ± 4.98	56.01 ± 0.54	9.91 ± 0.19	50.00 ± 0.22	7.39 ± 0.12	0.13 ± 0.045	0.080 ± 0.018	77.14 ± 0.90
Clam (4 kg)	85	0.15 ± 0.013	22.05 ± 0.17	4.73 ± 0.51	2.13 ± 0.17	18.00 ± 0.35	16.03 ± 0.38	0.14 ± 0.020	0.041 ± 0.0047	0.032 ± 0.0032	1.52 ± 0.68
Scallop (4 kg)	84	0.76 ± 0.016	16.02 ± 0.37	13.16 ± 0.72	10.02 ± 0.72	180.06 ± 0.56	1900.00 ± 2.11	2.96 ± 0.022	0.24 ± 0.022	0.20 ± 0.024	77.47 ± 0.16
Oyster (4 kg)	88	0.79 ± 0.067	11.30 ± 0.25	6.84 ± 0.25	2.81 ± 0.35	130.09 ± 1.21	1170.05 ± 3.11	17.43 ± 0.054	0.23 ± 0.012	0.13 ± 0.0034	40.20 ± 0.71
Juide lines/ μg g ⁻¹											
GB (2762-2017) (¹	China)	I	0.1	I	0.5/1.0	0.5	0.1	I	2.0	I	I
EC (2006)		I	I	I	I	0.2–2.0	0.1 - 1.0	I	1	I	I
FAO (1983)		I	I	I	I	I	I	30.0	I	I	30.0
ANVISA (2013)		1.0				0.3	0.05-0.3				

Table 2 The concentrations of metal(loid)s in raw food (wet weight) (average \pm standard deviation) ($n \ge 3$)

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Table 3 Target cancer risk (TR) of iAs from seafood $(n \ge 3)$ after in vitro digestion

Samples	TR of ski	n cancer				TR of bladder and lung cancer					
(amount)	Raw	Boiled	Steamed	Baked	Fried	Raw	Boiled	Steamed	Baked	Fried	
Fish											
Tilapia (15)	2.3E-05	1.2E-05	1.8E-05	1.8E-05	9.1E-06	4.0E-04	2.1E-04	3.2E-04	3.0E-04	1.6E-04	
Snapper (15)	8.9E-06	4.5E-06	6.6E-06	6.5E-06	3.6E-06	1.5E-04	7.7E-05	1.1E-04	1.1E-04	6.2E-05	
Salmon (15)	0.0E + 00	0.0E+00	0.0E+00	0.0E+00	0.0E+00	0.0E+00	0.0E+00	0.0E+00	0.0E+00	0.0E+00	
Pomfret (15)	2.1E-05	1.1E-05	1.7E-05	1.5E-05	8.7E-06	3.6E-04	1.9E-04	2.9E-04	2.6E-04	1.5E-04	
Croaker (25)	1.7E-05	8.7E-06	1.3E-05	1.2E-05	6.5E-06	2.9E-04	1.5E-04	2.2E-04	2.1E-04	1.1E-04	
Turbot (15)	3.3E-06	2.0E-06	2.7E-06	2.5E-06	1.4E-06	5.6E-05	3.4E-05	4.6E-05	4.3E-05	2.3E-05	
Tiger Grouper (15)	1.4E-05	8.7E-06	1.1E-05	1.1E-05	6.5E-06	2.5E-04	1.5E-04	1.9E-04	1.9E-04	1.1E-04	
Anchovy (15)	1.3E-05	8.1E-06	1.1E-05	1.0E-05	6.1E-06	2.2E-04	1.4E-04	1.8E-04	1.7E-04	1.0E-04	
Crustacea											
Mantis shrimp (2 kg)	1.0E-06	5.6E-07	7.7E-07	7.6E-07	4.7E-07	1.7E-05	9.6E-06	1.3E-05	1.3E-05	8.1E-06	
Red shrimp (2 kg)	1.7E-06	8.3E-07	1.2E-06	1.2E-06	6.4E-07	2.9E-05	1.4E-05	2.0E-05	2.1E-05	1.1E-05	
Prawn (2 kg)	1.1E-05	5.4E-06	8.0E-06	7.8E-06	4.4E-06	1.9E-04	9.3E-05	1.4E-04	1.3E-04	7.5E-05	
Spotted crab (15)	0.0E+00	0.0E+00	0.0E+00	0.0E+00	0.0E+00	0.0E+00	0.0E+00	0.0E+00	0.0E+00	0.0E+00	
Red crab (15)	8.6E-06	4.3E-06	6.2E-06	6.2E-06	3.3E-06	1.5E-04	7.4E-05	1.1E-04	1.1E-04	5.7E-05	
Clam (4 kg)	3.2E-05	3.9E-05	7.0E-05	1.4E-04	2.9E-04	5.6E-04	6.7E-04	1.2E-03	2.4E-03	5.0E-03	
Scallop (4 kg)	2.3E-05	2.8E-05	5.0E-05	1.0E-04	2.1E-04	4.0E-04	4.8E-04	8.6E-04	1.7E-03	3.6E-03	
Oyster (4 kg)	1.6E-05	1.9E-05	3.5E-05	6.9E-05	1.5E-04	2.7E-04	3.3E-04	5.9E-04	1.2E-03	2.5E-03	

most kinds of seafood were comparable with the results reported by previous research which concluded that iAs is at very low contents (< 0.01 μ g g⁻¹) in seafood (El-Kady and Abdel-Wahhab 2018).

As described in Table 2, mean concentration of mercury (Hg) (μ g g⁻¹ ww) in seafood followed the descending order of: crab (0.068 ± 0.0028) > fish (0.042 ± 0.038) > shrimp

 $(0.021 \pm 0.0040) >$ shellfish (0.0082 ± 0.0043) . The highest mean concentration of Hg was observed in snapper (0.12 ± 0.038) , and the lowest was found in anchovy (0.0046 ± 0.00055) . The concentration ratios of MeHg to the total Hg in seafood samples were in the range of 40.2–100%. Correspondingly, the contents of MeHg in seafood samples were from 0.0021 to 0.084 µg g⁻¹ ww, which were under the limit recommended by the GB (2762-2017) in China (carnivorous fish: 1.0 µg g⁻¹ ww; other seafood:

0.5 μ g g⁻¹ ww) and similar to the previous research (EFSA 2012).

The content ($\mu g g^{-1}$ ww) of lead (Pb) in seafood samples in this study ranged from 0.00040 \pm 0.00016 to 0.18 ± 0.00056 . And we found that the highest level of Pb was in scallop (0.18 \pm 0.00056), followed by the oyster (0.13 ± 0.0012) and silvery pomfret (0.090 ± 0.00076) . The overall average contents of (Cd) cadmium across all samples were 0.00020 ± 0.000066 (salmon) to 1.90 ± 0.0021 (scallop). And the detected Cd levels in all seafood samples in this study followed the descending order of shellfish > shrimp > crab > fish. The range of concentration for chromium (Cr) was ranged (0.032 ± 0.0041) - (2.04 ± 0.053) µg g⁻¹ ww in all studied seafood species, which were substantially lower than the limit of 2.0 μ g g⁻¹ in aquatic foods set by China. The maximum mean level of nickel (Ni) detected was $0.20 \ \mu g \ g^{-1}$ ww in scallop, and the



Fig. 1 The metal(loid)s contents in various seafood samples $(n \ge 3)$ before and after cooking (mean \pm standard deviation). Lowercase letters indicate significant differences at p < 0.05 among each type of seafood. Capital letters indicate significant differences at p < 0.05 among these types of raw and cooked

minimum was 0.014 μ g g⁻¹ ww in prawn. The concentration ($\mu g g^{-1}$ ww) of copper (Cu) ranged from 0.12 to 17.43, with the overall average value of 3.88. Zinc (Zn) was the most abundant metal in all seafood samples analyzed, whose average content was highest in spotted crab (118.84 \pm 0.16 µg g⁻¹ ww) and lowest in clam (1.52 \pm 0.068 µg g⁻¹ ww). There are no limits of Ni, Cu, and Zn for seafood in Chinese standards. FAO (1983) set that the maximum limits for Cu and Zn were both 30 μ g g⁻¹ ww, and we found that only Zn levels in some species (spotted crab, red crab, scallop, and oyster) were higher than the limit. Overall, the results of metal(loid)s had no conflict with previous studies. For instance, Ip et al. (2005) reported the level of Cd in fish, shrimp, crab, and shellfish collected from the Pearl River Estuary was (0.01-2.1)

seafood. (a: turbot A; b: turbot B; c: tiger grouper A; d: tiger grouper B; e: anchovy; f: snapper; g: yellow croaker; h: mantis shrimp; i: prawn; j: red shrimp; k: spotted crab; l: red crab; m: clam; n: scallop)

 $\mu g g^{-1}$ ww, which was comparable with results in this study. Gu et al. (2015) reported the contents of metal(loid)s ($\mu g g^{-1}$ ww) collected from the South China Sea: 0.00054–0.027 (Pb), 0.00051–0.115 (Cd), 0.02–1.26 (Cr), 0.00083–0.057 (Ni), 0.12–1.13 (Cu), and 2.34–6.88 (Zn), respectively. The results of Pb levels in this study were similar to those of previous studies of fish collected from South California (Burger et al. 2002) and Sweden (Nfon et al. 2009), though lower than the investigations of fish from river Chenab (Alamdar et al. 2017).

To exclude the effect of moisture, we study the influence of cooking based on the dry weight. As shown in Fig. 1, cooking practices were found to have some effect on the metal(loid)s concentration. The four types of cooking methods could decrease As



Fig. 2 Bioaccessibility of metal(loid)s in raw seafood muscles (mean \pm standard deviation) ($n \geq 3$). Least significant difference test demonstrated that there were significant differences at p < 0.05 among the bioaccessibility for different types of seafood

contents from seafood, and boiling released As at the most extent (25.8%), followed by frying (24.1%), steaming (18.5%), and baking (16.6%). Correspondingly, the contents of iAs lowered after cooking (by an average of 24.5%). In regard to Hg (total Hg), the trends of average decreased percent of Hg in cooked seafood were in the order of steaming (14.8%) < boiling (16.6%) < baking (38.6%) < frying (45.7%). In respect of Ni, Cu, and Zn levels in cooked seafood, there existed the same decreasing trends with Hg after cooking. And, the decreased percent in contents of Ni, Cu, and Zn was in the range of 14.8–39.0%, 3.1-36.8%, and 9.1-43.9%, respectively. Cooking practices also brought about a decrease in Pb concentration in seafood samples in this study (a mean reduction of 21.7%). In relation to Cd concentration in seafood muscles, cooking could decrease it on average of 22.0% (frying: 36.5%; boiling: 24.3%; baking: 17.1%; steaming: 10.1%). Moreover, results



Fig. 3 Bioaccessibility of metal(loid)s in cooked seafood muscles (mean \pm standard deviation) (n \geq 3). Least significant difference test demonstrated that there were significant differences at p < 0.05 among the bioaccessibility for each cooking methods

Fig. 4 Total target hazard quotient (tTHQ) of hazardous metal(loid)s and through consumption of seafood in Guangzhou City based on bioaccessible contents (A) and initial contents (B), respectively. Least significant difference test demonstrated that there were significant differences at p < 0.05 among the THQ values for different cooking processes



demonstrated that cooking could also decrease Cr contents from seafood (on average of 6.0%) and the difference between cooking methods was not significant (p > 0.05). There have been some studies on the cooking effect on metal(loid)s in food. Our results were comparable with the previous studies, which believed that cooking resulted in a decrease in total As, Hg, Cd, Ni, Pb, Zn, Cu, and Cr in food matrices (Atta et al. 1997; Schmidt et al. 2015; Praveena and Omar 2017). However, results in some previous researches also gave various results. For instance, a study of Houlbrèque et al. (2011) said that cooking led to a rise in total Cd content in Chilean mussels. In our previous studies, we have verified that soluble As dissolved into the cooking water caused the reduction of As in shellfish after cooking (Liao et al. 2018). Besides, we determined the contents of all mentioned metals in raw food, cooked food, and discarded liquid and found that As, Pb, Cd, Ni, Cr, Cu, and Zn could be detected in the fried oil, boiled water, and steamed water after cooking. According to the boiling point (for instance, the boiling point of AsCl₃ and As₂O₃ is 130 °C and 456 °C, respectively; the boiling point of Cd is 765 °C), those metal(loid)s were hardly volatile during the thermal treatments performed in this study (< 130 °C) (Chem YQ). Heating procedure could promote protein degradation, resulting in the dissociation of As in water as free salts, soluble amino acids, and bound to proteins. Other metal(loid)s also have a high affinity for metallothionein, a protein with low molecular weight and high cysteine residue content (de Filipps 1979). Thus, cooking resulted in a decrease in As, Pb, Cd, Ni, Cr, Cu, and Zn contents in seafood muscles maybe due to the solubilization of them into the cooking liquid.

However, Hg release into cooking liquid from the seafood, especially as MeHg, is negligible. Further, we found that cooking affects the release of metal(loid)s from seafood in this study which followed the descending order of Hg > Zn > Ni > Cd > Pb >As > Cu > Cr. It is known that the affinity ability for metals is Hg > Cu > Pb > Cd > Ni > Zn (de Filipps 1979). Results in our study showed that after cooking, the trend of decrease percent of the target metal(loid) reversed that of its affinity ability with metallothionein, which was except for Hg. Some studies have reported that cooking could decrease contents of Hg in fish muscle, which was attributed to evaporation (Hajeb et al. 2014; Mieiro et al. 2016). Consequently, volatilization of Hg in seafood muscles during heating may be dominant.

Bioaccessibility of metal(loid)s in seafood

We selected some seafood samples (turbot, snapper, yellow croaker, mantis shrimp, and clam) for performing in vitro gastrointestinal digestion tests to estimate the metal(loid) bioaccessibility from raw and cooked seafood. The results are shown in Fig. 2.

In general, the bioaccessible concentration of all metal(loid)s was lower than the initial concentration for all selected seafood samples. For raw fish, the average bioaccessible content of As, Hg, Pb, Cd, Cr, Ni, Cu, and Zn was 1.02, 0.042, 0.031, 0.0010, 1.05, 0.13, 0.21, and 5.04 μ g g⁻¹ ww, respectively. And for raw shellfish, corresponding mean content of each element was 11.20, 0.028, 0.046, 0.46, 0.12, 0.096, 7.55, and 44.89 μ g g⁻¹ ww, respectively. Based on bioaccessible concentration, bioaccessibility of metal(loid)s was given here.

Results showed that As was highly bioaccessible in raw seafood, whose bioaccessibility ranged from $87.4 \pm 2.4\%$ (yellow croaker) to $98.4 \pm 3.3\%$ (snapper), with the average value of 92.8%. In regard to Hg, the highest bioaccessible percent in raw seafood was found in turbot (96.8 \pm 3.3%) and the lowest was in clam (64.7 \pm 3.5%). It was observed that the mean bioaccessibility of MeHg had a wide range (45.2–100.0%). As a whole, the highest bioaccessibility of Cd, Pb, and Ni was all observed in raw turbot, while the lowest was found in raw mantis shrimp. Cd, Pb, and Ni bioaccessibility was 60.0-99.4%, 78.9-93.8%, and 75.9-94.3%, respectively. In relation to Cr, it was revealed that bioaccessible fraction in raw seafood samples was highly variable between species in this study, which was from $20.2 \pm 1.1\%$ (snapper) to $87.6 \pm 3.3\%$ (clam). As for Cu, there was an average of 71.5% (70.4–89.1%) of the initial concentrations in raw seafood muscles that were bioaccessible. Moreover, above 90% of Zn in seafood were bioaccessible, which ranged from 93.2 to 100.0% in this study.

In general, cooking processes (boiling, steaming, baking, and frying in this study) could affect the bioaccessibility of all metal(loid)s in selected seafood muscles in this study (Fig. 3). For all analyzed seafood species, all cooking practices decreased the bioaccessibility of Hg and MeHg significantly (p < 0.01). On average, Hg bioaccessibility in seafood after cooking reduced by 14%, 14%, 22%, and 39% for boiling, steaming, baking, and frying, respectively. The average reduced percentage for MeHg was 24%, and the

maximum value was 44% (after frying). The bioaccessibility of As from seafood decreased by 2.7-18.6%, 3.5-14.0%, 6.5-26.8%, and 12.9-35.2% after boiling, steaming, baking, and frying, respectively. On average, cooking lowered the bioaccessibility of Cd, Pb, Ni, Cu, and Zn by 30.9%, 30.7%, 25.7%, 17.6%, and 22.4%, respectively. And we found that bioaccessibility reduction of As, Cd, Pb, Ni, Cu, and Zn after cooking was all in this order: steaming < boiling < baking < frying. To mention specially, the reduction of bioaccessibility of Cu in seafood after steaming (-0.4%) was not as obvious as that after boiling (12.5%), baking (20.0%), and frying (38.3%). Alves et al. (2018) also found that bioaccessible Cu in seafood seemed not to be affected by steam. In relation to Cr in seafood, baking (by 37.7%) and frying (by 57.6%) decreased its bioaccessibility greatly, while the effect of boiling (by 11.5%) and steaming (by 18.0%) was much smaller. Our results were similar to Amiard et al. (2008), who reported that cooking led to a decrease in Cd, Cu, Pb, and Zn bioaccessibility in shellfish.

Thermal treatment could change the nutritional structure of seafood samples and influence the dissolvability of these metal(loid)s as a consequence (Wu et al. 2018). Metals, as well as As, have a high affinity for sulfhydryl groups in peptides and proteins, and heat deactivates proteins and enhances the breaking of bonds between them and seafood proteins, which makes their solubilization easier (Cheyns et al. 2017; Matos et al. 2015; de Filipps 1979). In a word, we believed two mechanisms influence the bioaccessibility of these metal(loid)s in seafood after cooking: The one is protein denaturation and the other one is the liberation of soluble chemicals into the cooking solution, or of volatile forms into the air (Liao et al. 2018; Liao et al. 2019a, b). Hence, the determined process affecting metal(loid)s bioaccessibility in seafood during cooking is the freeing of soluble portions, which decreases bioaccessibility as a result. Consequently, more severe heating conditions (e.g., baking and frying) are more conducive to decreasing bioaccessible fractions of metal(loid)s in seafood.

Assessment of health risk

Till now, health risk assessment of mixed metal(loid)s from seafood based on bioaccessibility was still very less. Thus, daily intake of metal(loid)s was calculated based on the bioaccessible concentration of raw and cooked seafood in this study. For an adult in Guangzhou City (average body weight is 60 kg), the estimation of Cr and Ni intake through consumption of seafood species in this study was in the range of 0.24–51.3 and 0.18–15.3 μ g d⁻¹, respectively. The recommended daily intake of Cr and Ni was 50–200 and 100–300 μ g, respectively (Gu et al. 2015). The estimated intake of Zn from crab collected in this study was up to 60.13 μ g kg⁻¹ d⁻¹, followed by scallop (39.20 μ g kg⁻¹ d⁻¹), which were much lower than the tolerable daily intake (300 μ g kg⁻¹ d⁻¹). Thus, these metals (Cr, Ni, and Zn) in seafood species in this study were within the appropriate range, which is a benefit for human health.

Daily intake of As, Hg, MeHg, Pb, Cd, and Cu from cooked seafood was 0.027-13.24, raw and 5.5×10^{-5} -0.078. 0.0002-0.065. 0.0003 - 0.10. 0.0001–0.67, and 0.012–5.86 μ g kg⁻¹ d⁻¹, respectively, based on bioaccessibility. These values were well below the maximum tolerable ones. As shown in Fig. 4, all values of tTHQ of the detected metal(loid)s were below 1.0 based on bioaccessible concentration while higher than 1.0 for spotted crab and scallop when calculated based on initial contents. Gu et al. (2015) also reported that the hazardous quotient values of Cd, Cr, Ni, Cu, and Zn were all less than 1, which revealed no significant adverse health effects with consumption of wild fishes captured from the South China Sea.

Table 3 shows that TR values of skin cancer risk and TR values of bladder and lung cancer via exposure of iAs were in the range of $0-2.9 \times 10^{-4}$ and $0-5.0 \times 10^{-3}$, respectively. Range of acceptable risk levels for carcinogens is $10^{-4}-10^{-6}$. Thus, results demonstrated that there is a potential of having bladder and lung cancer for adults via iAs exposure from some seafood species (e.g., clam, scallop, and oyster) in this study. In our previous study, we have verified that there existed As demethylation reaction during gastrointestinal digestion (Liao et al. 2018). Concerns should be focused particularly on those people who exposed to iAs from shellfish which can harm their bladder and lung.

This paper investigated the health risk of Hg, As, Cd,

Pb, Cr, Ni, Cu, and Zn in seafood based on

Conclusion

bioaccessibility in Guangzhou, China. The cooking effect was considered here. Results showed that cooking practices could decrease the metal(loid)s concentration from seafood. Decreased percent in contents of As, Hg, Cd, Pb, Ni, Cr, Cu, and Zn was in the range of 16.6–25.8, 14.8–45.7, 10.1–36.5, 1.6-49.4, 14.8-39.0, -5.4-21.6, 3.1-36.8, and 9.1-43.9%, respectively. On average, the release of metal(loid)s from seafood in this study followed the descending order of Hg > Zn > Ni > Cd > Pb > As > Cu > Cr. Cooking also brought about reduction of bioaccessibility of these metal(loid)s from seafood, and frying process decreased greatest. On average, cooking lowered the bioaccessibility of As, Hg, Cd, Pb, Ni, Cr, Cu, and Zn by 15.2, 26.1, 30.9, 30.7, 25.7, 31.2, 17.6, and 22.4%, respectively. Health risk calculation assessment results showed that Cr, Ni, and Zn in seafood species in this study were within the human health benefits range. And Hg, Cd, Pb, and Cu exposure from cooked seafood was within the safe dose. According to the total target hazardous quotient of mixed metal(loid)s, it seems the risk is tolerable. However, target cancer risk (TR) value of inorganic As from seafood showed that there is a potential of having cancer (especially bladder and lung cancer) for people exposure from iAs in seafood in this study.

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Compliance with ethical standards

Conflict of interest The authors declare no conflict of interest.

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