

A NEW NEOLIGNAN GLUCOSIDE FROM THE STEMS OF “BAEKMA” CULTIVAR, *Chrysanthemum morifolium*

Hyoung-Geun Kim,¹ Hyun-Ji Oh,¹ Jung-Hwan Ko,¹
Sun-Woo Joo,¹ Yeong-Geun Lee,¹ Yun-Su Baek,²
Dae Young Lee,³ and Nam-In Baek^{1*}

A new neolignan glucoside has been isolated from Chrysanthemum morifolium ‘Baekma’ cultivar through repeated silica gel and octadecyl silica gel (ODS) column chromatographies. On the basis of spectroscopic data including NMR, MS, and IR, the chemical structure of the new neolignan glucoside was determined to be (7S,8S)-7-O-[1’-(4’,5’-dihydroxy-3’-methoxyphenyl)propanol]guaiaacylglycerol 4-O-β-D-glucopyranoside (1), named baekmaoside A.

Keywords: Baekma, baekmaoside A, *Chrysanthemum morifolium*, neolignan glucoside.

Chrysanthemum morifolium (Compositae) is a herbaceous perennial plant distributed widely in Korea, China, and Japan. *C. morifolium* has mainly been used for ornamental purposes but also as tea or for fragrance in East Asian cultures [1]. *C. morifolium* is classified into standard mums, spray mums, and pollen mums. More than 700 kinds of *C. morifolium* cultivars have been developed in Korea, and standard mums occupy an overwhelming 80% share of the Korean market [2]. The Horticultural Research Institute of the Korea Rural Development Administration recently bred a new cultivar called “Baekma” from the standard type *C. morifolium* [3]. The distribution of *C. morifolium* for ornamental purposes yield a large quantity of by-products, the majority of which are the stems of *C. morifolium*. Alternative uses of by-products from *C. morifolium* lead to cost reduction and increased income for farmers. Also, *C. morifolium* stems have been reported to have antioxidant [4] and antithrombogenic [5] effects. Despite several reports on the pharmacological activities of *C. morifolium* stems, only one phytochemical study has been reported, which listed four phenolic compounds and one lignan as the components of *C. morifolium* stems [6]. This study was carried out to isolate physiologically active substances from the Korean domestic *C. morifolium* cultivar, “Baekma”.

The stems of *C. morifolium* were extracted with aqueous MeOH, and the concentrated extract was fractionated into EtOAc, *n*-BuOH, and water fractions. From the *n*-BuOH fraction, a new neolignan glucoside was isolated through repeated SiO₂ and octadecyl silica gel (ODS) column chromatography.

Compound **1**, a brown amorphous powder, showed UV absorption characteristics at 365 and 254 nm and a black color on TLC by spraying with 10% H₂SO₄ and heating. The molecular formula was determined to be C₂₆H₃₆O₁₃ from the molecular ion peak [M + H]⁺ *m/z* 557.2230 (calcd for C₂₆H₃₇O₁₃, 557.2234) in positive-mode HR-FAB-MS. The IR spectrum suggested the presence of a hydroxyl group (3375 cm⁻¹) and an aromatic double bond (1604, 1580 cm⁻¹).

1) Graduate School of Biotechnology and Department of Oriental Medicinal Biotechnology, Kyung Hee University, 17104, Yongin Republic of Korea, fax: +82 31 204 8116, e-mail: nibaek@khu.ac.kr; 2) Saucos Industrialization Department, Agency for Korea National Food Cluster (AnFC), IPET, 54576, Wanju, Republic of Korea; 3) Department of Herbal Crop Research, National Institute of Horticultural and Herbal Science, RDA, 27709, Eumseong, Republic of Korea. Published in *Khimiya Prirodnikh Soedinenii*, No. 4, July–August, 2019, pp. 527–529. Original article submitted June 11, 2018.

TABLE 1. ^1H (400 MHz) and ^{13}C (100 MHz) NMR Data of Compound **1** (CD_3OD , δ , ppm, J/Hz)

C atom	δ_{H}	δ_{C}	C atom	δ_{H}	δ_{C}
1	–	129.5	6'	6.70 (br.s)	117.8
2	7.02 (d, J = 2.0)	111.1	7'	2.61 (m)	32.9
3	–	150.9	8'	1.80 (m)	35.8
4	–	147.5	9'	3.27 (m)	62.4
5	7.13 (d, J = 8.4)	117.9	1''	4.87 (d, J = 7.6)	102.7
6	6.92 (dd, J = 8.4, 2.0)	119.3	2''	3.42 (m)	74.8
7	5.54 (d, J = 6.0)	88.4	3''	3.43 (m)	78.2
8	3.55 (m)	75.1	4''	3.37 (m)	71.3
9	3.29 (m)	65.0	5''	3.44 (m)	77.8
1'	–	138.3	6''	3.85 (br.d, J = 13.6)	62.2
2'	6.72 (br.s)	114.1		3.73 (dd, J = 13.6, 5.6)	
3'	–	145.2	3-OCH ₃	3.81 (s)	56.6
4'	–	137.0	3'-OCH ₃	3.84 (s)	56.7
5'	–	147.4			

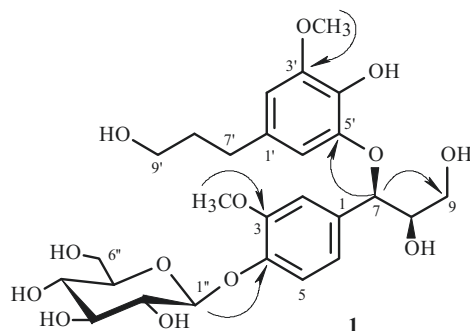


Fig. 1. Chemical structure and key correlations in the gHMBC of compound **1**. gHMBC key correlations are represented by single-headed arrows from H to C.

The ^1H NMR spectrum (Table 1) showed three aromatic methine signals [δ 7.13 (1H, d, J = 8.4 Hz, H-5), 7.02 (1H, d, J = 2.0 Hz, H-2), 6.92 (1H, dd, J = 8.4, 2.0 Hz, H-6)] due to a 1,2,4-trisubstituted benzene ring and two aromatic methine signals [δ 6.72 (1H, br.s, H-2'), 6.70 (1H, br.s, H-6')] due to a 1,2,3,5-tetrasubstituted benzene ring. In the oxygen region, two oxymethines [δ 5.54 (1H, d, J = 6.0 Hz, H-7), 3.55 (1H, m, H-8)], two oxymethylenes [δ 3.29 (1H, overlapped, H-9), 3.27 (1H, overlapped, H-9')], and two methoxyls [δ 3.84 (3H, s, 3'-OCH₃), 3.81 (3H, s, 3-OCH₃)] were observed. In the aliphatic region, two methylenes [δ 2.61 (2H, m, H-7'), 1.80 (2H, m, H-8')] were observed. Therefore, the proton signals indicated the aglycon to be a lignan. The proton signals due to a hemiacetal at δ 4.87 (1H, d, J = 7.6 Hz, H-1''), four oxymethines [δ 3.44 (1H, overlapped, H-5''), 3.43 (1H, overlapped, H-3''), 3.42 (1H, overlapped, H-2''), 3.37 (1H, overlapped, H-4'')], and one oxymethylene [δ 3.85 (1H, br.d, J = 13.6 Hz, H-6''a), 3.73 (1H, dd, J = 13.6, 5.6 Hz, H-6''b)] were interpreted as signaling a hexose moiety.

From the above-mentioned ^1H NMR data, compound **1** was expected to be a lignan monoglycoside with two methoxy groups [7, 8]. The ^{13}C NMR spectrum showed 26 carbon signals including two methoxy groups [δ 56.7 (3'-OCH₃), 56.6 (3-OCH₃)], confirming compound **1** as composed of a lignan and a hexose moiety. In the downfield, five oxygenated olefine quaternary carbons [δ 150.9 (C-3), 147.5 (C-4), 147.4 (C-5'), 145.2 (C-3'), 137.0 (C-4')], two olefine quaternary carbons [δ 138.3 (C-1'), 129.5 (C-1)], and five olefine methines [δ 119.3 (C-6), 117.9 (C-5), 117.8 (C-6'), 114.1 (C-2'), 111.1 (C-2)] were observed. In the oxygen region, two oxymethines [δ 88.4 (C-7), 75.1 (C-8)] and two oxymethylenes [δ 65.0 (C-9), 62.4 (C-9')] were observed. In the aliphatic region, two methylenes [δ 35.8 (C-8'), 32.9 (C-7')] were also observed. The carbon chemical shift of the hexose moiety, a hemiacetal [δ 102.7 (C-1'')], four oxymethines [δ 78.2 (C-3''), 77.8 (C-5''), 74.8 (C-2''), 71.3 (C-4'')], and one oxymethylene [δ 62.2 (C-6'')] revealed the sugar to be a β -glucopyranose, and the coupling constant of

the anomer proton signal ($J = 8.0$ Hz) confirmed the anomer hydroxyl as having a β -configuration. In the gHMBC spectrum (Fig. 1), the oxymethine proton signal (δ 5.54, H-7) showed a cross-peak with the oxygenated olefine quaternary carbon signal (δ 147.5, C-5'), and the anomer proton signal (δ 4.87, H-1'') showed a cross-peak with the oxygenated olefine quaternary carbon signal (δ 147.5, C-4). Also, two methoxy proton signals [δ 3.84 (3H, s), 3.81 (3H, s)] showed cross peaks with their respective oxygenated olefine quaternary carbon signals [δ 145.2 (C-3'), 150.9 (C-3)]. Taken together, the planar structure of compound **1** was determined to be 7-*O*-[1'-(4',5'-dihydroxy-3'-methoxyphenyl)propanol]guaiaacylglycerol 4-*O*- β -D-glucopyranoside. The large coupling constant between H-7 and H-8 ($J = 6.0$ Hz) suggested the threo conformation of C-7/C-8 [9]. Because the CD spectrum of compound **1** showed a positive cotton effect at 297 nm and a negative cotton effect at 266 nm, the absolute configurations of chiral centers were assigned to be 7*S* and 8*S* [10]. Therefore, the chemical structure of compound **1** was determined to be (7*S*,8*S*)-7-*O*-[1'-(4',5'-dihydroxy-3'-methoxyphenyl)propanol]guaiaacylglycerol 4-*O*- β -D-glucopyranoside, which was revealed to be a new compound, named baekmaoside A.

EXPERIMENTAL

General Methods. The materials and methods used for this study were the same as those in the previous study [11, 12].

Plant Material. Stems of *C. morifolium* "Baekma" were purchased from Yangjae Flower Market Center, Seoul, Korea, in 2017 and identified by Prof. Ha-Seung Pak, Flower Research Institute, Chungcheongnam-do ARES, Yesan, Korea. Unlike other chrysanthemum varieties where the center of the flower is yellowish, "Baekma" has green centers, so it can be easily identified. A voucher specimen (NPCL-20170430) has been deposited at the Natural Products Chemistry Laboratory, Kyung Hee University, Yongin, Korea.

Extraction of *C. morifolium* "Baekma" Stems and Isolation of a Neolignan Glycoside. Three kilograms of *C. morifolium* "Baekma" stems were cut into pieces, dried, and extracted with 80% methanol (MeOH, 30 L \times 3) for 24 h at room temperature. The concentrated methanol extract (177 g) was suspended in water (3.0 L) and then consecutively extracted with EtOAc (3.0 L \times 3) and *n*-BuOH (2.4 L \times 3). The fractions were concentrated *in vacuo* to produce EtOAc (CBAE, 19.3 g), *n*-BuOH (CBAB, 18.9 g), and H₂O (CBAW, 138.8 g) residues. The CBAB fraction was subjected to silica gel (SiO₂) column chromatography (CC) (5.5 cm \times 17.0 cm) and eluted with CHCl₃-MeOH (3:1, 2.0 L) and CHCl₃-MeOH-H₂O (7:3:1 \rightarrow 65:35:10, 2.0 L of both). The eluting solutions were monitored by TLC to produce eight fractions (CBAB-1 to CBAB-8). Fraction CBAB-3 [1.4 g, elution volume/total volume (Ve/Vt) 0.312-0.454] was applied to an octadecyl silica gel (ODS) CC (3.5 cm \times 8.0 cm) and eluted with MeOH-H₂O (1:3 \rightarrow 1:2, 1.3 L of both) to yield 12 fractions (CBAB-3-1 to CBAB-3-12). Fraction CBAB-3-11 (54.1 mg, Ve/Vt 0.840-0.880) was subjected to SiO₂ CC (1.5 cm \times 16 cm) and eluted with EtOAc-*n*-BuOH-H₂O (25:3:1 \rightarrow 16:3:1, 1.0 L of both) to produce eight fractions (CBAB-3-11-1 to CBAB-3-11-8) along with a purified compound **1** [CBAB-3-11-5, 19.6 mg, Ve/Vt 0.397-0.469, TLC (SiO₂ F₂₅₄) R_f 0.66, CHCl₃-MeOH-H₂O, 20:3:1].

Baekmaoside A (1), yellow amorphous powder (MeOH); $[\alpha]_D^{23} +43.0^\circ$ (c 0.10, MeOH). IR (CaF₂ plate, cm⁻¹): 3375, 1604, 1580. HR-FAB/MS m/z 557.2230 [M + H]⁺ (calcd for C₂₆H₃₇O₁₃, 557.2234). ¹H (400 MHz, CD₃OD, δ , ppm) and ¹³C NMR (100 MHz, CD₃OD, δ , ppm), see Table 1.

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