# **9-HYDROXY-, 9-ACETYLOXY-, AND 9-HEXANOYLOXY FATTY ACIDS FROM THE GLANDULAR TRICHOME EXUDATE OF** *Silene latifolia* **subsp.** *alba*

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*Chemical investigation of the glandular trichome exudate from* Silene latifolia *subsp.* alba *(Caryophyllaceae) resulted in the identification of 9-hydroxyoctadecanoic acid (1) and five new fatty acids 9-hydroxyicosanoic acid (2), 9-acetyloxyoctadecanoic acid (3), 9-acetyloxyicosanoic acid (4), 9-hexanoyloxyhexadecanoic acid (5) and 9-hexanoyloxyoctadecanoic acid (6). The oxygenated positions in these fatty acid derivatives were determined to be at C-9 by GC-MS analysis of their trimethylsilyl (TMS) derivatives, while the configurations at C-9 of compounds 1 and 2 were established as* S *by application of the Akasaka-Ohrui method to their methyl esters.*

**Keywords**: *Silene latifolia* subsp. *alba*, Caryophyllaceae, glandular trichome, hydroxy-fatty acid, 9-acetyloxyoctadecanoic acid, 9-hexanoyloxyoctadecanoic acid.

The secondary metabolites secreted by glandular trichomes are suggested to have diverse biological activities, such as protecting the aerial parts of plants against herbivores and pathogens [1, 2]. Glandular trichomes produce secondary metabolites of diverse classes such as terpenoids, phenylpropanoids, polyketides, and fatty acid derivatives [3]. We have recently isolated cyclic fatty acyl glycosides from the glandular trichome exudates of *Silene gallica* [4] and *Cerastium glomeratum* [5] of the Caryophyllaceae family. The occurrence of these unique glycosides prompted us to further investigate secondary metabolites contained in the glandular trichome exudates of other plants that belong to this family.

The genus *Silene* comprises about 700 species found in many areas, mostly in the Northern hemisphere. *Silene latifolia* subsp. *alba* is native to Europe. Nowadays, it is naturalized in worldwide temperate areas, and naturalized in Japan after the Second World War. It can be seen in Hokkaido and a part of Honshu [6]. Flowering stems and calyxes of the plant were found to be rich in glandular trichomes. This article describes the identification and structure elucidation of six fatty acid derivatives from the glandular trichome exudate of *S. latifolia* subsp. *alba.*

The crude exudate sample from flowering stems of the plant showed three major spots on TLC, designated Frs. A, B, and C, in order of decreasing polarity. The three fractions were separated by silica gel column chromatography.

Fraction A was suggested to be a mixture of compounds having the molecular formulas  $C_{18}H_{36}O_3$  and  $C_{20}H_{40}O_3$  on the basis of HR-FAB-MS data. The 1H NMR spectrum of compounds **1** and **2** showed signals assignable to a mono-hydroxylated linear fatty acid: an oxymethine proton at  $\delta$  3.58 (m), methylene protons adjacent to the carboxy group at  $\delta$  2.35 (t, J = 7.0 Hz), a methyl group at  $\delta$  0.88 (t, J = 7.0 Hz), and methylene protons centered at  $\delta$  1.26 (Table 1). The hydroxylated fatty acid structures were supported by the <sup>13</sup>C NMR data, which showed signals of a carboxy carbon ( $\delta$  178.8) and an oxymethine carbon (δ 72.1) in addition to a methylene carbon  $\alpha$  to the carboxyl group (δ 33.8) and two methylene carbons  $\alpha$  to the oxymethine carbon (δ 37.4 and 37.5) (Table 1). These NMR data, taken together with FAB-MS data, suggested that Fr. A could be a mixture of a hydroxy-octadecanoic acid and a hydroxy-icosanoic acid.

The ratio of the two hydroxy-fatty acids in Fr. A was further examined by GC-MS analysis. The total ion chromatogram of the bis-TMS derivative of Fr. A showed two peaks at  $t<sub>R</sub>$  12.83 and 13.92 min in a 49:51 ratio.

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C atom	Fraction A $(1 + 2, 49:51)$		Fraction B $(3 + 4, 74:26)$		Fraction C $(5 + 6, 17:83)$	
	$\delta_{\rm H}$	$\delta_{\rm C}$	$\delta_{\rm H}$	$\delta_{\rm C}$	$\delta_{\rm H}$	$\delta_{\rm C}$
$\mathbf{1}$		178.8		179.1		178.5
$\mathfrak{2}$	2.35 (t, $J = 7.0$ )	33.8	2.34 (t, $J = 7.0$ )	33.9	2.34 (t, $J = 7.0$ )	33.8
$\overline{3}$	$1.64$ (m)	24.7	$1.63$ (m)	24.6	$1.62$ (m)	24.9
$4 - 6$	$1.22 - 1.48$ (m)	$28.9 - 29.7$	$1.22 - 1.36$ (m)	$29.0 - 29.7$	$1.22 - 1.36$ (m)	$29.0 - 29.7$
$\tau$	$1.22 - 1.48$ (m)	25.5	$1.22 - 1.36$ (m)	25.3	$1.22 - 1.36$ (m)	25.2
$8\,$	$1.43$ (m)	37.4	$1.50$ (m)	34.1	1.50(m)	34.1
9	$3.58$ (m)	72.1	$4.85$ (m)	74.4	$4.86$ (m)	74.0
10	$1.43$ (m)	37.5	$1.50$ (m)	34.1	1.50(m)	34.2
11	$1.22 - 1.48$ (m)	25.7	$1,22-1,36$ (m)	25.3	$1.22 - 1.36$ (m)	25.3
$12-15(17a)$	$1.22 - 1.48$ (m)	$28.9 - 29.7$	$1.22 - 1.36$ (m)	$29.0 - 29.7$	$1.22 - 1.36$ (m)	$29.0 - 29.7$
$16(18^a)$	$1.22 - 1.48$ (m)	31.9	$1.22 - 1.36$ (m)	31.9	$1.22 - 1.36$ (m)	31.9
$17(19^a)$	$1.22 - 1.48$ (m)	22.7	$1.22 - 1.36$ (m)	22.7	$1.22 - 1.36$ (m)	22.7
$18(20^a)$	$0.88$ (t, J = 7.0)	14.1	$0.88$ (t, J = 7.0)	14.1	$0.88$ (t, J = 7.0)	14.1
Acyl group						
1'				171.0		173.7
2'			2.04(s)	21.3	2.28 (t, $J = 7.5$ )	34.7
3'					$1.62$ (m)	24.7
4'					$1.22 - 1.36$ (m)	31.4
5'					$1.22 - 1.36$ (m)	22.3
$6^{\prime}$					$0.90$ (t, J = 7.0)	13.9

TABLE 1. <sup>1</sup>H (500 MHz) and <sup>13</sup>C (125 MHz) NMR Spectroscopic Data for Fractions A–C (CDCl<sub>3</sub>,  $\delta$ , ppm, J/Hz)

<sup>a</sup>The atom numbers are for the dodecanoic acid derivatives.



The EI-MS of the former peak showed intense ions at *m*/*z* 317 and 229, which were assignable to the C-1–C-9 fragment arising from C-9/C-10 fission and the C-9–C-18 fragment arising from C-8/C-9 fission, respectively. Similarly, the EI-MS of the latter peak exhibited intense ion peaks at *m*/*z* 317 and 257, which were assignable to C-1–C-9 and C-9–C-20 fragments, respectively. These MS data unequivocally established that the two acids were hydroxylated at the C-9 position.

The configuration at C-9 of the two acids was determined using the Akasaka–Ohrui method [7]. In our previous paper, the <sup>1</sup>H NMR spectrum of  $(1R,2R)$ -2- $(2,3$ -anthracene dicarboximide)cyclohexanecarboxylic acid ester of methyl  $9(R)$ -hydroxydocosanoate showed H<sub>2</sub>-2 and COOMe signals at  $\delta$  2.073 (2H, t, J = 7.5 Hz) and 3.635 (3H, s), while that of the  $(15,25)$ -derivative exhibited the respective signals at  $\delta$  2.147 (2H, t, J = 7.5 Hz) and 3.641 (3H, s) [4]. In the present study, the  $(1R,2R)$ -acid ester of Fr. A methyl ester showed H<sub>2</sub>-2 and COOMe signals at  $\delta$  2.147 (2H, t, J = 7.0 Hz) and 3.641 (3H, s), while the (1*S*,2*S*)-acid ester of Fr. A methyl ester displayed the respective signals at  $\delta$  2.072 (2H, t, J = 7.5 Hz) and 3.636 (3H, s). These data clearly indicated that the two acids have the 9*S* configuration. It is thus concluded that Fr. A is a 49:51 mixture of 9(*S*)-hydroxyoctadecanoic acid (**1**) and 9(*S*)-hydroxyicosanoic acid (**2**).

Fraction B was presumed to be a mixture of compounds that have the molecular formulas  $C_{20}H_{38}O_4$  and  $C_{22}H_{42}O_4$ on the basis of HR-FAB-MS data, both of which gained a C2H2O unit compared to the respective molecular formula of **1** and **2**. The <sup>1</sup>H NMR spectrum of Fr. B was similar to that of Fr. A, except for the presence of an acetyl methyl signal ( $\delta$  2.04) and a downfield shift of the oxymethine proton to δ 4.85 (Table 1). The data suggested that Fr. B could be the 9-*O*-acetyl derivative of Fr. A. The presence of signals ( $\delta$  21.3 and 171.0) assignable to an acetyl group in the <sup>13</sup>C NMR of Fr. B supported the structure assignment (Table 1). GC-MS analysis of the TMS derivative of Fr. B displayed two major peaks at  $t_R$  13.05 and 14.16 min in a 74:26 ratio. The two peaks showed a common fragment ion at *m*/*z* 245, which was assigned to the C-1–C-9 fragment arising from C-9/C-10 fission followed by the elimination of the ketene molecule, thus indicating the C-9 location of the acetate group. Hence, Fr. B was determined to be a 74:26 mixture of 9-acetyloxyoctadecanoic acid (**3**) and 9-acetyloxyicosanoic acid (**4**).

Fraction C was thought to be a compound having the molecular formula  $C_{24}H_{46}O_4$ , which was accompanied by a lesser amount of homologous compound with the molecular formula  $C_{22}H_{42}O_4$  on the basis of HR-FAB-MS data. The <sup>1</sup>H NMR spectrum of Fr. C was similar to that of Fr. B, except for the appearance of signals due to a linear short-chain acyl group (signals of CH<sub>2</sub>  $\alpha$  to a carboxyl group at  $\delta$  2.28 (t, J = 7.5 Hz) and a terminal methyl group at  $\delta$  0.90 (t, J = 7.0 Hz) with concomitant disappearance of the acetyl signal. These data, combined with information on the molecular formula, suggested that the major and minor constituents of Fr. C are 9-hexanoyloxyoctadecanoic acid (**6**) and 9-hexanoyloxyhexadecanoic acid (**5**), respectively. The presence of a hexanoyl group was corroborated by the 13C signals assignable to the group δ 173.7 (C-1′), 34.7 (C-2′), 24.7 (C-3′), 31.4 (C-4′), 22.3 (C-5′), 13.9 (C-6′) [8]. GC-MS analysis of the TMS derivative of Fr. C displayed a minor peak at t<sub>R</sub> 13.85 min and a major peak at t<sub>R</sub> 15.30 min in a 17:83 ratio. Both peaks displayed a common fragment ion at  $m/z$  245 assignable to an ion C-1–C-9 fragment arising from C-9/C-10 fission followed by the elimination of the  $C_4H_9$  CH=C=O molecule as found in the Fr. B TMS derivative. The diagnostic fragment ion indicated the C-9 location of the hexanoyloxy moiety in both acids. Thus, Fr. C was characterized as a 17:83 mixture of 9-hexanoyloxyhexadecanoic acid (**5**) and 9-hexanoyloxyoctadecanoic acid (**6**). The C-9 configurations of acids **3–6** remain to be investigated, although it can be assumed that they are *S* by analogy with compounds **1** and **2**.

In addition, careful analysis of the GC-MS data allowed us to identify 9-hydroxyhexadecanoic acid (corresponding to peak at *t <sup>R</sup>* 11.95 min), 9-acetyloxyhexadecanoic acid (corresponding to peak at *t <sup>R</sup>* 12.15 min), and 9-hexanoyloxyicosanoic acid (corresponding to peak at  $t_R$  17.27 min) as very minor constituents in the exudate sample.

In the present study we characterized 9-hydroxyoctadecanoic acid (**1**) and five new fatty acids **2–6** that possess the hydroxy, acetyloxy, or hexanoyloxy group at C-9 from the glandular trichome exudate of *S. latifolia* subsp*. alba*. There was no indication of the occurrence of cyclic glycolipids that were found in *S. gallica* [4] and *C. glomeratum* [5]. Thus, it can be said that Caryophyllaceae plants do not necessarily contain cyclic glycolipids as their glandular trichome exudate constituents. The hydroxy fatty acids can be regarded as a biosynthetic precursor of such cyclic glycolipids. *S. latifolia* subsp*. alba* appears to evolve into an acylation enzyme leading to the *O*-acetate and *O*-hexanoate rather than an *O*-glycosylation enzyme that eventually leads to cyclic glycolipids. It is of note that the configuration at the C-9 stereogenic centers of **1** and **2** was *S*, which is opposite to the *R* configuration at C-9, C-10, C-11, C-12, and C-13 found in the fatty acid moieties of the cyclic glycolipids of *S. gallica* and *C. glomeratum*. These findings further indicate the diversity of plants among the Caryophyllaceae family in the biosynthesis of glandular trichome exudate constituents.

# **EXPERIMENTAL**

**General Methods.** <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Bruker DRX 500 (500 MHz for <sup>1</sup>H and 125 MHz for <sup>13</sup>C) spectrometer in CDCl<sub>3</sub> solution. The tetramethylsilane ( $\delta$  0.00) signal was used as an internal standard for <sup>1</sup>H shifts, and the CDCl<sub>3</sub> ( $\delta$  77.00) signal was used as a reference for <sup>13</sup>C shifts. FAB-MS spectra (3-nitrobenzyl alcohol as the matrix) were obtained on a JEOL JMS-700 spectrometer. IR spectra were recorded on a JASCO-FT/IR-5300 spectrometer. Optical rotations were measured on a JASCO P-2200 polarimeter. TLC analysis was performed using Merck precoated Si gel 60 F254 glass plates, and spots were detected by treating the plates with a 5% ethanolic solution of phosphomolybdic acid followed by heating at 120°C. Silica gel 60 N (spherical neutral, 40–100 μm, Kanto Chemical, Japan) was used for column chromatography (CC). GC-MS was conducted using a mass spectrometer (JMS-AM SUN200, JEOL) connected to a gas chromatograph (6890A, Agilent Technologies) under the following conditions: EI (70 eV); DB-1 capillary column (30 m  $\times$  0.25 mm, 0.25 µm film thickness, J&W Scientific); source temperature 250°C; injection temperature 250°C; column temperature programmed from 80°C to 280°C (held 20 min at the final temperature) at a rate of 20°C; interface temperature 280°C; He carrier gas flow rate 1.0 mL/min with splitless injection.

**Plant Material**. The seeds of *Silene latifolia* subsp. *alba* (Caryophyllaceae) were collected in Asahikawa City, Hokkaido, Japan in July 2010, and the plants were kept cultivated in a planter at the campus of Tokyo Institute of Technology. The plant was identified by one of the author (Y. F). A voucher specimen (CMS-2202) was deposited in the Department of Chemistry and Materials Science, Tokyo Institute of Technology.

**Extraction and Isolation**. In a preliminary investigation, the heads of the glandular trichomes were directly pressed to a TLC plate, and the plate was developed with hexane–EtOAc (2:1). The analysis revealed the presence of three major spots designated Frs. A–C with *R<sub>f</sub>* values 0.24, 0.38, and 0.53, respectively. A large-scale sampling was carried out in June 2014 by dipping flower stems (30 stems, ca. 15 cm long) into ether (400 mL) for 1 min. The ether was dried over  $\text{Na}_2\text{SO}_4$  and filtered. Concentration of the filtrate under reduced pressure yielded a yellow semi-solid (140 mg). This was subjected to silica gel (5 g) column chromatography eluting with increasing polarities of hexane–EtOAc. Elution with hexane–EtOAc (10:1) removed a surface wax (mainly hydrocarbon, 109 mg). Elution with hexane–EtOAc (4:1) afforded Fr. C (12 mg). Further elution with  $CHCl<sub>3</sub>$ –MeOH (3:1) furnished Fr. B (5 mg), and further elution with hexane–EtOAc (2:1) furnished Fr. A (4 mg).

Fraction A (49:51 mixture of compounds 1 and 2. Colorless oil.  $[\alpha]_D^{25}$  –4.7° (*c* 0.07, CHCl<sub>3</sub>). IR (CHCl<sub>3</sub>, v<sub>max</sub>, cm<sup>-1</sup>): 3690, 3600, 3520, 2930, 2855, 1715. For 1H and 13C NMR data, see Table 1. HR-FAB-MS *m/z*: 299.2567 [M – H]– (calcd for  $C_{18}H_{35}O_3$ , 299.2586) and 327.2875 [M – H]<sup>–</sup> (calcd for  $C_{20}H_{39}O_3$ , 327.2899). Two peaks were observed at 12.83 and 13.92 min in a 49:51 ratio. EI-MS of the shorter  $t_R m/z$ : 429 ( $[M-Me]^+$ , 13%), 413 ( $[M-MeO]^+$ , 26%), 339 ([M – Me – TMSOH]+, 26%), 317 (C-9/C-10 cleavage, 100%), 288 (19%), 229 (C-8/C-9 cleavage, 82%), 129 (43%), 73 (100%). EI-MS of the longer t<sub>*R</sub> m/z:* 457 ( $[M-Me]^+$ , 5%), 441 ( $[M-Me0]^+$ , 7%), 367 ( $[M-Me-TMSOH]^+$ , 10%), 317 (C-9/C-10</sub> cleavage, 96%), 288 (7%), 257 (C-8/C-9 cleavage, 41%), 129 (18%), 73 (100).

Fraction B (74:26 mixture of compounds 3 and 4). Colorless oil.  $[\alpha]_D^{25}$  –2.6° (*c* 0.17, CHCl<sub>3</sub>). IR (CHCl<sub>3</sub>, v<sub>max</sub>, cm<sup>-1</sup>): 3675, 3514, 2930, 2860, 1720. For <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 1. HR-FAB-MS  $m/z$ : 343.2831 [M + H]<sup>+</sup> (calcd for C<sub>20</sub>H<sub>39</sub>O<sub>4</sub>, 343.2848) and 371.3183  $[M + H]^{+}$  (calcd for  $C_{22}H_{43}O_{4}$ , 371.3161). Two peaks were observed at 13.05 and 14.16 min in a 74:26 ratio. EI-MS of the shorter t<sub>*R</sub> m*/*z*: 371 ([M – Ac]<sup>+</sup>, 2.0%), 354 ([M – AcOH]<sup>+</sup>, 6.7%), 339 ([354 – Me]<sup>+</sup>, 76.2%), 281 ([354 – TMS]<sup>+</sup>,</sub> 5%), 264 ([354 – TMSOH]<sup>+</sup>, 8%), 245 (C-9/C-10 cleavage-CH<sub>2</sub>CO, 54%), 155 (53%), 117 (C-1/C-2 fission, 78%), 73 (100%). EI-MS of the longer t<sub>*R</sub>* m/z: 399 ([M – Ac]<sup>+</sup>, 1%), 382 ([M – AcOH]<sup>+</sup>, 8%), 367 ([382 – Me]<sup>+</sup>, 86%), 309 ([382 – TMS]<sup>+</sup>, 6%),</sub> 292 ([382 – TMSOH]<sup>+</sup>, 14%), 245 (C-9/C-10 cleavage-CH<sub>2</sub>CO, 57%), 155 (61%), 117 (C-1/C-2 fission, 100%), 73 (100%).

Fraction C (17:83 mixture of compounds 5 and 6). Colorless oil.  $\left[\alpha\right]_D^{25} + 0.9^\circ$  (*c* 0.27, CHCl<sub>3</sub>). IR (CHCl<sub>3</sub>, v<sub>max</sub>, cm<sup>-1</sup>): 3690, 3520, 2930, 2860, 1715. For 1H and 13C NMR data, see Table 1. HR-FAB-MS *m/z*: 399.3463 [M + H]+ (calcd for  $C_{24}H_{47}O_4$ , 399.3474) and 371.315 [M + H]<sup>+</sup> (calcd for  $C_{22}H_{43}O_4$ , 427.3787). Two peaks were observed at 13.85 and 15.30 min in an 17:83 ratio. EI-MS of the shorter  $t_R m/z$ : 343 ( $[M - C_5H_{11}CO]^+$ , 5%), 326 ( $[M - C_5H_{11}COOH]^+$ , 28%), 311  $([326 - Me]^+, 74\%)$ , 253  $([326 - TMS]^+, 12\%)$ , 236  $([326 - TMSOH]^+, 14\%)$ , 245  $(C-9/C-10$  cleavage-CH<sub>2</sub>CO, 91%), 99  $([C_5H_{11}CO]^+, 100\%), 73 (100\%).$  EI-MS of the longer  $t_R$  *m/z*: 371 ( $[M - C_5H_{11}CO]^+, 5\%$ ), 354 ( $[M - C_5H_{11}COOH]^+, 28\%$ ), 339 ([354 – Me]<sup>+</sup>, 74%), 281 ([354 – TMS]<sup>+</sup>, 12%), 264 ([354 – TMSOH], 14%), 245 (C-9/C-10 cleavage-CH<sub>2</sub>CO, 91%), 99 ( $[C_5H_{11}CO]^+$ , 100%), 73 (100%).

**Trimethylsilylation and GC/MS Analysis**. A portion (ca. 0.1 mg) of Fr. A, B, or C was heated with *N*-methyl-*N*trimethylsilyltrifluoroacetamide (50 μL) at 80°C for 0.5 h, and then cooled to room temperature. A portion (1–2 μL) of the solution was subjected to GC/MS analysis.

**Determination of the C-9 Configuration of Fr. A.** Fraction A (3 mg) was reacted with trimethylsilyldiazomethane (0.6 M hexane solution, 0.2 mL) in EtOAc (0.3 mL)–MeOH (0.2 mL) at room temperature for 30 min. The crude methyl ester was purified by *p*-TLC to give an oily methyl ester (3 mg). The methyl ester (1 mg each) was further converted to the 9-*O*-(1*R*,2*R*)- and (1*S*,2*S*)-2-(2,3-anthracene dicarboximide)-cyclohexanecarboxylic acid derivatives in the same manner as described in our previous paper [4]. Pertinent  ${}^{1}$ H NMR data are described in the text.

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