

A NEW 9,10-DIHYDROPHENANTHRENE GLYCOSIDE FROM *Dendrobium primulinum*

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A new 9,10-dihydrophenanthrene glycoside named 2,4,5,9S-tetrahydroxy-9,10-dihydrophenanthrene 4-O-β-D-glucopyranoside (1), together with 11 known compounds, was isolated from Dendrobium primulinum. Their structures were determined by analysis of spectroscopic data.

Keywords: *Dendrobium primulinum*, 9,10-dihydrophenanthrene glycoside, 2,4,5,9S-tetrahydroxy-9,10-dihydrophenanthrene 4-O-β-D-glucopyranoside.

Dendrobium, one of the largest genera in the Orchidaceae family with more than 1100 species identified, is widely distributed throughout Asia, Europe, and Australia [1]. There are 75 species and two varieties of *Dendrobium* in China [2], and the stems of many species are used as “Shi-Hu” in traditional Chinese medicines to nourish the stomach, promote the production of body fluid, and reduce fever. *Dendrobium primulinum* Lindl. is distributed in southwest of China, India, Nepal, Burma, Laos, Vietnam, and Thailand and is widely used as a medicinal *Dendrobium* plant in “Dai Medicine.” Previous phytochemical studies on *D. primulinum* have revealed the presence of a series of compounds such as alkaloids, nucleosides, and flavonoids [3, 4]. Our further investigation on this plant species led to the isolation of a novel 9,10-dihydrophenanthrene glycoside (**1**) and 11 known compounds **2–12**.

Compound **1** was obtained as a white amorphous powder from MeOH with $[\alpha]_D^{25} -27^\circ$ (*c* 0.3; MeOH). The molecular formula of C₂₀H₂₂O₉ was deduced from the positive HR-ESI-MS *m/z* 429.1160 [M + Na]⁺ (calcd 429.1162) and its NMR data (Table 1), indicating 10 degrees of unsaturation. The negative-ion ESI-MSⁿ data showed *m/z* 243 [M – H – 162][–], suggesting the presence of one hexose. Its UV maxima absorptions at 211.2, 270.2, and 296.4 nm were similar to those of 9,10-dihydrophenanthrene derivatives [5]. The IR spectrum showed absorption bands for hydroxyl (3396 cm^{–1}) and aromatic (1618, 1451 cm^{–1}) groups. The aromatic region of the ¹H NMR spectrum showed three adjacent aromatic protons at δ 7.14 (1H, t, *J* = 7.8 Hz), 7.05 (1H, d, *J* = 7.8 Hz), and 6.84 (1H, d, *J* = 7.8 Hz) and a pair of *meta*-coupled protons at δ 6.62 (1H, d, *J* = 1.8 Hz) and 6.50 (1H, d, *J* = 1.8 Hz), indicating a 1,2,3-trisubstituted benzene ring and a 1,3,4,5-tetrasubstituted benzene ring, respectively. In the aliphatic region, two methylene protons at δ 2.72 (1H, dd, *J* = 14.4, 3.6 Hz) and 2.62 (1H, dd, *J* = 14.4, 9.6 Hz) and one oxygenated methine proton at δ 4.48 (1H, dd, *J* = 9.6, 3.6 Hz) were attributed to a –OCHCH₂ group by ¹H–¹H COSY correlations, and one anomeric signal at δ 5.13 (1H, d, *J* = 7.8 Hz) and six protons in the range 3.23–3.72 were due to a sugar residue. The ¹³C NMR and DEPT spectrum confirmed the presence of 12 aromatic carbons (three bearing oxygen atoms), one oxygenated methine carbon at δ 67.7, one methylene carbon at δ 39.7, together with six hexose carbons. In addition, based on the IR absorption at 3396 cm^{–1} and its molecular formula and oxygenated C-atoms (δ 157.7, 153.1, 153.0), three hydroxyl groups were assumed. Acid hydrolysis of **1** afforded D-glucose, which was identified by GC analysis, and the *J* value of the anomeric proton δ 5.13 (1H, d, *J* = 7.8 Hz) indicated a β-configuration. In the HMBC spectrum, the anomeric proton (H-1′) showed correlation with C-4, indicating the attachment of β-D-glucose to C-4. The NOESY correlations between H-1′ to H-3 and H-5′ to H-3 confirmed the location of glucose (Fig. 1).

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TABLE 1. ^1H (600 MHz), ^{13}C NMR (150 MHz) Data, and HMBC Correlations of Compound **1** (DMSO- d_6 + CD_3OD , δ , ppm, J/Hz)

C atom	δ_{H}	δ_{C}	HMBC
1	6.50 (1H, d, $J = 1.8$)	110.4	C-2, C-3, C-4a, C-10
2	–	157.7	
3	6.62 (1H, d, $J = 1.8$)	100.7	C-2, C-4, C-4a
4	–	153.1	
4a	–	112.8	
4b	–	120.0	
5	–	153.0	
6	6.84 (1H, d, $J = 7.8$)	118.2	C-4b, C-5
7	7.14 (1H, t, $J = 7.8$)	127.4	C-5, C-6, C-8a
8	7.05 (1H, d, $J = 7.8$)	117.1	
8a	–	142.9	
9	4.48 (1H, dd, $J = 9.6, 3.6$)	67.7	
10	2.72 (1H, dd, $J = 14.4, 3.6$), 2.62 (1H, dd, $J = 14.4, 9.6$)	39.7	C-4a, C-8a, C-9, C-10a
10a	–	139.2	
1'	5.13 (1H, d, $J = 7.8$)	99.6	C-4
2'	3.28 (1H, m)	73.4	C-1', C-3'
3'	3.34 (1H, m)	76.8	C-2', C-4'
4'	3.23 (1H, m)	69.7	C-3', C-6'
5'	3.39 (1H, m)	77.6	C-6'
6'	3.72 (1H, m)	60.9	C-4', C-5'
	3.52 (1H, m)		

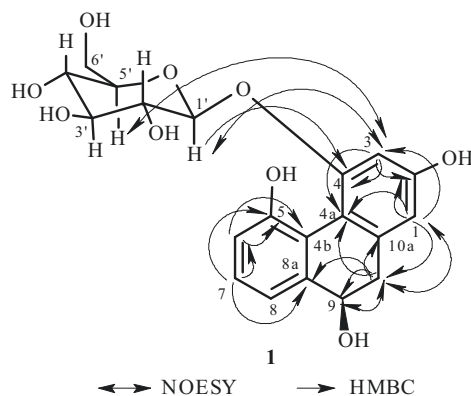


Fig. 1. Structure of compound **1** and its HMBC and NOESY correlations.

The CD spectrum of **1** showed a negative Cotton effect at 233 nm and a positive Cotton effects at 268 nm, which were in agreement with that of (+)-(9*S*)-hydroxy-9,10-dihydrophenanthrene [6]. Therefore, the structure of compound **1** was deduced as 2,4,5,9*S*-tetrahydroxy-9,10-dihydrophenanthrene 4-*O*- β -D-glucopyranoside.

The known compounds were identified as 2,4,7-trihydroxy-9,10-dihydrophenanthrene (**2**) [7], denthyrsinol (**3**) [8], moscatin (**4**) [9], moscatilin (**5**) [10], gigantol (**6**) [11], batatasin III (**7**) [12], tristin (**8**) [13], 3,4',5-trihydroxybibenzyl (**9**) [14], 3,6,9-trihydroxy-3,4-dihydroanthracen-1(2*H*)-one (**10**) [15], β -sitosterol (**11**) [16], and β -daucosterol (**12**) [17] by comparing their spectroscopic data with literature data.

All the above compounds were isolated from this plant for the first time.

EXPERIMENTAL

General Procedures. Optical rotations were taken on a PerkinElmer 341 polarimeter. IR spectra were carried out on a Nicolet-Nexus 670 infrared spectrophotometer with KBr pellet. NMR spectra were run on a Bruker AV-600 NMR spectrometer

using TMS as an internal standard. ESI-MS and HR-ESI-MS were recorded on a Waters Micromass Q-TOF micro mass spectrometer. The CD spectra were obtained on a Chirascan spectrometer. Column chromatography was performed on silica gel (200–300 mesh, Qingdao Marine Chemical Inc., Qingdao, China), RP-18 silicagel (ODS, 50 μ m, YMC), and Sephadex LH-20 (Pharmacia).

Plant Material. The whole plant of *D. primulinum* was collected in Longling, Yunnan Province, China, in November 2012, and identified by Prof. Xinjia Ming, Chongqing Academy of Chinese Materia Medica. A voucher specimen (No. 20121101) was deposited at the Shanghai Institute of Pharmaceutical Industry, Shanghai, China.

Extraction and Isolation. The air-dried and powdered stems of *D. primulinum* (4 kg) were extracted with 95% ethanol (four times, each 40 L) at room temperature. Evaporation of the solvent under reduced pressure afforded a brown crude extract (300 g). The extract was suspended in water and then partitioned with petroleum ether (60–90°C), EtOAc, and *n*-BuOH successively to obtain four fractions. The EtOAc fraction (40 g) was subjected to silica gel column chromatography and eluted with CH₂Cl₂–MeOH (100:1, 50:1, 30:1, 15:1, 10:1, 0:1) to yield six fractions (Fr. 1–Fr. 6). Fraction 2 (11.8 g) was isolated by silica gel column chromatography with CH₂Cl₂–acetone (40:1–10:1) to give six subfractions (Fr. 2.1–Fr. 2.6). Fraction 2.2 (3.0 g) was rechromatographed on silica gel column with petroleum ether–acetone (5:1–3:1) and then purified over a Sephadex LH-20 column with CH₂Cl₂–MeOH (1:1) and recrystallized (petroleum ether–acetone) to afford compounds **4** (112 mg), **5** (210 mg), **6** (738 mg), and **11** (80 mg). Fraction 2.4 (4.2 g) was further subjected to silica gel column chromatography and eluted with petroleum ether–EtOAc (3:1–1:1) and then purified over a Sephadex LH-20 column with CH₂Cl₂–MeOH (1:1) and preparative TLC to obtain compounds **3** (9 mg) and **8** (170 mg). By the same method, compound **7** (103 mg) was obtained from Fr. 3 (4.2 g), compounds **2** (30 mg) and **9** (20 mg) were obtained from Fr. 4 (4.5 g), and compound **10** (6 mg) was obtained from Fr. 5 (3.3 g). The *n*-BuOH fraction (42 g) was subjected to silica gel column chromatography and eluted with CH₂Cl₂–MeOH–H₂O (30:1:0, 15:1:0, 10:1:0, 4:1:0.1, 7:3:0.5, 0:1:0) to yield six fractions (Fr. I–VI). Fraction III (7.4 g) was subjected to Sephadex LH-20 column chromatography and eluted with EtOH and recrystallized (MeOH) to give compound **12** (19 mg). Fraction IV (10.5 g) was subjected to ODS column chromatography and eluted with a step gradient of MeOH–H₂O (1:9–6:4) to yield nine subfractions (Fr. IV.1–Fr. IV.9). Fraction IV. 5 was purified over a Sephadex LH-20 column with EtOH and silica gel column chromatography with CH₂Cl₂–MeOH–H₂O (6:1:0.1) to obtain compound **1** (43 mg).

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REFERENCES

1. J. Xu, Q. B. Han, S. L. Li, X. J. Chen, X. N. Wang, Z. Z. Zhao, and H. B. Chen, *Phytochem. Rev.*, **12**, 341 (2013).
2. Editorial Board of Flora of China of Chinese Academy of Sciences, *Flora of China*, Science Press, Beijing, 1999, 111 pp.
3. B. Luning and K. Leander, *Acta Chem. Scand.*, **19**, 1607 (1965).
4. Y. Mei, Q. H. Ye, P. M. Yang, D. Y. Kong, and L. Cheng, *Chin. J. Pharm.*, **45**, 224 (2014).
5. Y. Lin, F. Wang, L. J. Yang, Z. Chun, and J. K. Bao, *Phytochemistry*, **95**, 242 (2013).
6. S. M. Resnick and D. T. Gibson, *Appl. Environ. Microbiol.*, **62**, 3355 (1996).
7. H. Yang, Z. T. Wang, L. S. Xu, and Z. B. Hu, *J. Chin. Pharm. Univ.*, **33**, 367 (2002).
8. G. N. Zhang, L. Y. Zhong, and S. W. Bligh, *Phytochemistry*, **66**, 1113 (2005).
9. P. L. Majumder and R. C. Sen, *India J. Chem.*, **26B**, 18 (1987).
10. P. L. Majumder and R. C. Sen, *Phytochemistry*, **26**, 2121 (1987).
11. R. K. Juneja, S. C. Sharma, and J. S. Tandon, *Phytochemistry*, **24**, 321 (1985).
12. Y. W. Leong, C. C. Kang, and L. Harrison, *Phytochemistry*, **44**, 157 (1997).
13. P. L. Majumder and S. Pal, *Phytochemistry*, **32**, 1561 (1993).
14. S. A. Adesanya, S. K. Ogundana, and M. F. Roberts, *Phytochemistry*, **28**, 773 (1989).
15. J. M. Hu, Y. X. Zhao, Z. H. Miao, and J. Zhou, *Bull. Korean Chem. Soc.*, **30**, 2098 (2009).
16. V. S. P. Chaturvedula and I. Prakash, *Int. Curr. Pharm. J.*, **1**, 239 (2012).
17. F. T. Chen, F. H. Ge, and L. H. Xie, *Zhongyaocai*, **11**, 37 (1988).