

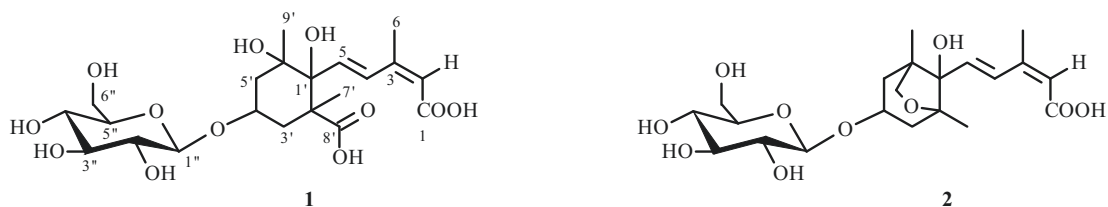
## SESQUITERPENOID GLYCOSIDES FROM THE FRUITS OF *Chaenomeles speciosa*

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A new sesquiterpenoid glycoside, called speciosaoside A (**1**), and a known sesquiterpenoid glycoside, (*1'R,3'R,5'R,8'S*)-*epi*-dihydrophaseic acid  $\beta$ -D-glucoside (**2**), were isolated from the fruits of *Chaenomeles speciosa* (Sweet) Nakai (Rosaceae) using column chromatography and preparative HPLC. The structure of compound **1** was deduced from comprehensive spectroscopic analysis including IR, HR-ESI-MS, 1D NMR, and 2D NMR. The structure of compound **2** was identified by comparison of its spectral data with those reported in the literature.

**Keywords:** *Chaenomeles speciosa*, sesquiterpenoid, glycoside, speciosaoside A.

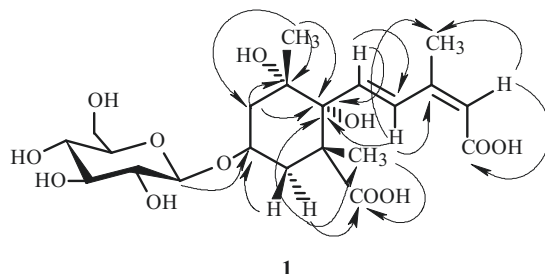
The dried fruit of *Chaenomeles speciosa* (Sweet) Nakai (Rosaceae) is a traditional medicine, widely used for the treatment of inflammatory and infectious diseases, as well as gastrointestinal symptoms, such as rheumatoid arthritis, cholera, diarrhea, dyspepsia, and gastrointestinal spasm [1]. It is also well known in China as a food consumed as an appetizer. It was reported to have a variety of biological activities such as antimicrobial, analgesic [2–5], antidiabetic [6], and antioxidant effects [7]. A previous study showed that the effective constituents of *C. speciosa* included mainly glycosides, flavones, phenolics, tannins, triterpenes, and organic acids [8–14]. Previously, our laboratory had reported that the 80% EtOH extract of the fruits of *C. speciosa* showed positive anti-inflammatory and analgesic activities in carrageenan-induced paw edema in rats [5]. The present study was undertaken to investigate the chemical constituents of the extract, which led to the isolation of a new sesquiterpenoid glycoside called speciosaoside A (**1**) and a known sesquiterpenoid glycoside (*1'R,3'R,5'R,8'S*)-*epi*-dihydrophaseic acid  $\beta$ -D-glucoside (**2**). Their structures were elucidated by chemical and spectroscopic methods. Compound **2** was isolated from the fruits of *C. speciosa* for the first time.



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TABLE 1.  $^1\text{H}$  (600 MHz) and  $^{13}\text{C}$  (150 MHz) NMR, HMBC, NOE of Compound **1** ( $\text{CD}_3\text{OD}$ ,  $\delta$ , ppm, J/Hz)

C atom	$\delta_{\text{C}}$	$\delta_{\text{H}}$	HMBC (H $\rightarrow$ C)	NOE
1	170.8			
2	120.2	5.88 (1H, br.s)	C-1, C-4, C-6	$\text{CH}_3$ -6
3	151.3			
4	133.3	7.83 (1H, d, J = 16.5)	C-3, C-6	$\text{CH}_3$ -9', $\text{CH}_3$ -7'
5	130.1	6.47 (1H, d, J = 16.5)	C-4, C-1'	$\text{CH}_3$ -6, $\text{CH}_3$ -9', $\text{CH}_3$ -7', H-5'ax
6	21.2	2.09 (3H, br.s)	C-2, C-3, C-5	H-2, H-5
1'	83.1			
2'	53.5			
3'ax	38.6	1.96 (1H, m)	C-1', C-2', C-4', C-8'	$\text{CH}_3$ -9', H-5'ax, H-3'eq
3'eq		1.99 (1H, m)		H-3'ax
4'	73.3	4.03 (1H, br.m)		H-1''
5'ax	38.5	2.12 (1H, m)	C-1', C-4', C-6', C-9'	H-5'eq, H-5, H-3'eq
5'eq		2.52 (1H, m)		H-5'ax
6'	90.9			
7'	14.3	1.10 (3H, s)		
8'	182.3			
9'	18.3	1.40 (3H, s)	C-1', C-6', C-5'	
1''	102.6	4.46 (1H, d, J = 7.8)		
2''	74.2	3.20 (1H, dd, J = 7.8, 9.0)		
3''	76.9	3.44 (1H, m)		
4''	70.1	3.37 (1H, m)		
5''	77.1	3.37 (1H, m)		
6''	61.8	3.68 (1H, br.d, J = 12.0) 3.87 (1H, dd, J = 12.0, 4.8)		

Fig. 1. Key HMBC correlations of compound **1**.

Compound **1** was obtained as a brown amorphous powder. The  $[\text{M} + \text{H}]^+$  ion peak at  $m/z$  477.1978 in the HR-ESI-MS corresponded to a molecular formula  $\text{C}_{21}\text{H}_{32}\text{O}_{12}$ , representing six unsaturated degrees. The  $^1\text{H}$  NMR spectrum (Table 1) of compound **1** in  $\text{CD}_3\text{OD}$  contained abundant singlet signals of three methyl groups [ $\delta$  2.09 (3H, br.s,  $\text{CH}_3$ -6), 1.10 (3H, s,  $\text{CH}_3$ -7'), 1.40 (3H, s,  $\text{CH}_3$ -9')], two methylene signals [ $\delta$  1.96 (1H, d, J = 17 Hz, H-3'ax), 1.99 (1H, d, J = 17 Hz, H-3'eq), 2.12 (1H, d, J = 10 Hz, H-5'ax), 2.52 (1H, d, J = 10 Hz, H-5'eq)], and three signals of olefinic protons [ $\delta$  5.88 (1H, br.s, H-2), 6.47 (1H, d, J = 16.5 Hz, H-5), and 7.83 (1H, d, J = 16.5 Hz, H-4)]. The complex oxymethine region integrated for six protons that belong to a glucose unit [ $\delta$  3.20 (1H, dd, J = 7.8, 9.0 Hz, H-2''), 3.44 (1H, m, H-3''), 3.37 (1H, m, H-4''), 3.37 (1H, m, H-5''), 3.68 (1H, br.d, J = 12.0 Hz, H-6''a), 3.87 (1H, dd, J = 4.8, 12.0 Hz, H-6''b)], and resonance 4.46 (1H, d, J = 7.8 Hz, H-1'') was assigned to the anomeric glucose proton. The  $^{13}\text{C}$  NMR spectrum (Table 1) revealed 21 carbon resonances, six of which were assigned to a  $\beta$ -D-glucopyranosyl unit [15] and 15 to the aglycone moiety. From the  $^{13}\text{C}$  and DEPT NMR experiments, these 15 signals were identified as six quaternary carbons: one olefinic carbon C-3 ( $\delta$  151.3), two oxygenated carboxyls C-1 ( $\delta$  170.8) and C-8' ( $\delta$  182.3), and three aliphatic carbons: C-1' ( $\delta$  83.1), C-2' ( $\delta$  53.5), and C-6' ( $\delta$  90.9); three tertiary carbons: C-2 ( $\delta$  120.2), C-4 ( $\delta$  133.3), and C-5 ( $\delta$  130.1); two secondary carbons: C-3' ( $\delta$  38.6) and C-5' ( $\delta$  38.5); and three primary carbons: C-6 ( $\delta$  21.2), C-7' ( $\delta$  14.3), and C-9' ( $\delta$  18.3). The overall data suggested that compound **1** was a phaseic acid derivative [16]. Interestingly, compound **1** has a similar skeleton compared to the known compound: (1'R,3'R,5'R,8'S)-epi-dihydrophaseic acid  $\beta$ -D-glucoside (**2**).

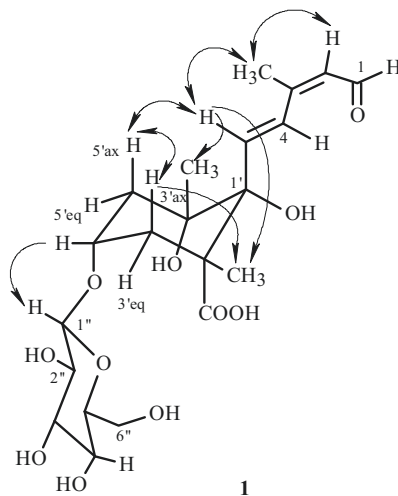


Fig. 2. Structure-relevant NOE resonances in compound **1**.

The specific structure of compound **1** was unambiguously confirmed by heteronuclear correlation experiments such as  $^1\text{H}$ - $^1\text{H}$  COSY and HMBC. The correlations (Fig. 1) of H-5 ( $\delta$  6.47) with C-4/C-3, H-4 ( $\delta$  7.83) with C-2/C-6, and H-2 ( $\delta$  5.88) with C-6/C-3/C-1 were observed in the HMBC spectrum. Besides, in the  $^1\text{H}$ - $^1\text{H}$  COSY spectrum, H-4 showed a correlation with H-5. All the correlations indicated the presence of the  $-\text{C}=\text{C}-\text{C}(\text{CH}_3)=\text{C}-\text{COOH}$  moiety. Meanwhile, H-4 and H-5 showed correlations with the carbon proton C-1' ( $\delta$  83.1), which indicated that the  $-\text{C}=\text{C}-\text{C}(\text{CH}_3)=\text{C}-\text{COOH}$  moiety was at C-1' of the cyclohexene ring. The anomeric sugar proton H-1'' ( $\delta$  4.46) was correlated to the C-4' ( $\delta$  73.3) in the HMBC spectrum, so the sugar moiety was attached to the cyclohexene ring at C-4'. In the HMBC, H-3' was correlated with  $\text{CH}_3$ -7' ( $\delta_{\text{C}}$  14.3), C-2' ( $\delta$  53.5), and C-8' ( $\delta$  182.3), which indicated that the  $-\text{COOH}$  and  $-\text{CH}_3$  were attached at C-2' of the cyclohexene ring. Also, the protons of the methyl group  $\text{CH}_3$ -9' ( $\delta_{\text{H}}$  1.40) showed correlations with C-1' ( $\delta$  83.1), C-5' ( $\delta$  38.5), and C-6' ( $\delta$  90.9), which indicated that  $-\text{CH}_3$  was at C-6' of the cyclohexene ring. The above important correlations are shown in Fig. 1.

Selective NOE irradiation on axial proton H-5'ax ( $\delta$  2.12) resulted in strong NO effects on H-3'ax ( $\delta$  1.96), protons of methyl group  $\text{CH}_3$ -9' ( $\delta_{\text{H}}$  1.40), and the proton of olefinic H-5 ( $\delta$  6.47), clearly indicating that these interfering protons are located on the same side of the cyclohexane plane (Fig. 2). This implied an equatorial orientation for  $\text{CH}_3$ -9', and the olefinic side chain in axial orientation, hence the  $\beta$ -D-glucose moiety was located on the opposite side of the ring system and is also axially oriented. Also, glucosidation at C-4' was evident from the abundant NOE resonances between H-4' ( $\delta$  4.03) and the anomeric proton H-1'' ( $\delta$  4.46) (Fig. 2) in the opposite direction. Confirmation for the stereochemistry at H-6' was obtained by selective irradiation of H-5'b ( $\delta_{\text{H}}$  2.52), which resulted in a NOE resonance to H-5'a ( $\delta_{\text{H}}$  2.12) but not in a through-space interaction to  $\text{CH}_3$ -9'. The double-bond configurations in the olefinic side chain were visible by selective NOE enhancements: the *trans*-configuration of the  $\Delta^4$ -bond was due to the significant NO effect between H-5 ( $\delta$  6.47) and  $\text{CH}_3$ -6 ( $\delta_{\text{H}}$  2.09); also, irradiation of  $\text{CH}_3$ -6 ( $\delta_{\text{H}}$  2.09) resulted in a NOE for H-2 ( $\delta$  5.88) showing a *cis*-configuration of the double bond  $\Delta^2$ . Hence, the new compound was named speciosaoside A.

Besides the new sesquiterpenoid glycoside speciosaoside A (**1**), a known sesquiterpenoid glycoside, (1'*R*,3'*R*,5'*R*,8'*S*)-epi-dihydrophaseic acid  $\beta$ -D-glucoside (**2**) [16], was identified by comparison of its physical and spectroscopic data with those reported in the literatures. The obtained spectral data (MS,  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra) and optical rotation of this known compound were similar to those reported in the previous literature.

## EXPERIMENTAL

**General Procedures.** Silica gel (Jiangyou Company of Yantai, 100–200 and 200–300 mesh), RP-C18 (43–60  $\mu\text{m}$ ; Merck, Darmstadt, Germany), Sephadex LH-20 (Pharmacia, Uppsala, Sweden), and MCI gel (Mitsubishi Chemical Corporation, Tokyo, Japan) were used for column chromatography. Preparative HPLC was performed using an Agilent (Palo Alto, CA, USA) 1200 liquid chromatography system equipped with a quaternary solvent delivery system and an ultraviolet detector. A Platisil ODS C18 column was used for analytical (4.6 mm  $\times$  250 mm; Waters, USA) and preparative (250 mm  $\times$  20 mm; Waters, USA) purposes. Infrared (IR) spectra were recorded on a Bruker Vector 22 spectrometer with a KBr pellet. NMR, including COSY, HMBC, and HSQC experiments, were recorded on a Bruker Avance 600 NMR spectrometer operating at 600 MHz ( $^1\text{H}$ ) and

150 MHz ( $^{13}\text{C}$ ), with chemical shifts given in ppm ( $\delta$ ), using tetramethylsilane (TMS) as an internal standard. HR-ESI-MS spectra were recorded on an Agilent Technologies 6538 UHD Accurate-Mass Q-TOF LC/MS spectrometer (Agilent Technologies, Santa Clara, CA with an electrospray interface) and Agilent 1290 Infinity modules (G4220A binary pump, G4212A photodiode array detector, and G4226A autosampler).

**Plant Material.** The dried fruits of *C. speciosa* were purchased from Linyi of Shandong Province, China, in July 2006. The plant was identified by Prof. Hanming Zhang at the School of Pharmacy, Second Military Medical University, China. A voucher specimen (No. TD20060720) was deposited in the Department of Pharmacognosy, Second Military Medical University, China.

**Extraction and Isolation.** A total of 10 kg of dried and powdered fruits of *C. speciosa* was soaked in 10 L of 80% ethanol for 2 days and percolated with 160 L of 80% ethanol for 7 days at room temperature. The percolate was then collected and concentrated at 60°C under vacuum to obtain a crude extract. The aqueous supernatant was chromatographed on D101 macroporous resin eluted with water, 10% ethanol, 30% ethanol, 50% ethanol, and 95% ethanol in series to yield fractions 1–5. Fraction 2 was evaporated under reduced pressure, and a portion (150 g) was dissolved in  $\text{H}_2\text{O}$  (300 mL), then applied on an MCI gel column chromatograph eluted with  $\text{H}_2\text{O}$ , followed by increasing concentrations of MeOH (10, 20, 50, 100%) to give five fractions: Fr. 2-1 to 2-5. Fraction 2-4 was applied on an ODS column with MeOH– $\text{H}_2\text{O}$  (0:1, 2:8, 5:5, 1:0) to obtain four fractions (Fr. 2-4-1 to Fr. 2-4-4). Fractions 2-4-2 (3.6 g) and 2-4-3 (4.7 g) were repeatedly purified by Sephadex LH-20 and semipreparative HPLC to yield compounds **1** (4.8 mg) and **2** (10.8 mg).

**Compound 1.** Brown amorphous powder,  $[\alpha]_{\text{D}}^{20} -5.0^\circ$  (*c* 0.16, MeOH). HR-ESI-MS  $m/z$  477.1978  $[\text{M} + \text{H}]^+$ ; calcd 477.1972. For  $^1\text{H}$  and  $^{13}\text{C}$  NMR, see Table 1.

## ACKNOWLEDGMENT

The authors are grateful to Prof. Gen-Jin Yang (School of Pharmacy, Second Military Medical University) for his assistance with the measurement of NMR spectra. This project was financially supported by a grant from the National Science Fund for Distinguished Young Scholars (No. 81325024) and the National Specific Project of New Drugs Innovation (No. 2011ZX09102-006-03).

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