

COMPOSITION AND LARVICIDAL ACTIVITY OF THE ESSENTIAL OILS OF *Lantana camara* AND *Lantana montevidensis*

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Lantana camara Linn. and *Lantana montevidensis* Briq. (Verbenaceae) are native species from America's rain forest. However, these species are cultivated as ornamental plants [1]. They might be poisonous to ruminants, but are largely used in popular medicine all over the world to treat fever, flu, asthma, bronchitis, and a variety of other illnesses [2–4]. Specially in Brazil they are used to treat rheumatism and pulmonary diseases [5]. Their leaves are rich in essential oils and phenolic compounds like aesculin, quercetin, isorhamnetine, fisetine, gossypetine, tricine and aesculetine, and triterpenoids. These chemical constituents are also found in some weeds presenting allelopathic and toxic properties [6]. *Aedes aegypti* is responsible for yellow fever transmission and is also a vector for hemorrhaging dengue (breakbone fever), an endemic disease in Southeast Asia, Pacific islands, Africa, and America [7]. Few published works associating larval infestation levels and socioeconomical characteristics, together with dengue endemic reality and urban infestation by *Aedes aegypti*, are available. As larvae radication is much easier than elimination of the adult mosquito, new strategies are necessary to control larvae proliferation. The most efficient and ideal way to control dengue would be the clearance of spots favorable to growth of the mosquito. The usual way to do it involves application of synthetic products, which results in resistance development by the mosquito [8]. Several studies have targeted natural products as insecticidal agents in order to control *A. aegypti* larvae due to the large number of compounds and medicinal plant essential oils exhibiting larvicidal activity [9].

The results of analysis of the chemical composition of the essential oils are presented in Table 1. The volatile components of both species revealed the predominance of terpenoids and sesquiterpenoids. The main component of *L. camara* essential oil was found to be bicyclogermacrene (19.51%), while in *L. montevidensis* it was β -caryophyllene (29.81%).

Twenty-one components (99.11%) were identified for *L. camara* leaves oil, and eighteen (95.83%) for *L. montevidensis*, with nine chemical constituents not common to both species. The oils were assessed against *Aedes aegypti* larvae at the third development stage to test the larvicidal potential. Results showed that both species have larvicidal potential: *Lantana camara* with LD₅₀ of 42.3 ± 0.85 µg/mL, and *Lantana montevidensis* with LD₅₀ of 117 ± 0.5 µg/mL. Additionally, both values are below the standard limit of LD₅₀ < 1000 µg/mL.

On the face of preliminary bioassay results against *A. aegypti* larvae, it might be suggested that the essential oils from these species, especially *L. camara*, could be regarded as natural alternatives to eradicate *A. aegypti*.

Lantana camara and *Lantana montevidensis* leaves were collected in January, 2008 from the Small Aromatic and Medicinal Plants Garden of the Natural Products Research Laboratory (LPPN) at Cariri Regional University (URCA), City of Crato, Ceara State, Brazil. Exsiccates were deposited at the Dardaro de Andrade Lima Caririensis Herbarium, Biology Department, under registry No. 1662 and No. 1619, respectively for *L. camara* and *L. montevidensis*.

Fresh leaves (200 g) were submitted to the hydrodistillation process in a Clevenger-like apparatus for two hours, resulting in an essential oil yield of 0.05% for *L. camara* and 0.07% for *L. montevidensis*, which were subsequently dried by anhydrous sodium sulfate (Na₂SO₄) and kept refrigerated until they were analyzed.

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TABLE 1. Chemical Composition (%) of *Lantana camara* and *Lantana montevidensis* Fresh Leaves Essential Oils

Compound	RI	<i>L. camara</i>	<i>L. montevidensis</i>	Compound	RI	<i>L. camara</i>	<i>L. montevidensis</i>
3,5-Dimethylene-1,4,4-trimethylcyclopentene	938	–	0.71	<i>E</i> -Caryophyllene	1416	16.71	–
Thuja-2,4(10)-diene	960	–	0.48	α -Humulene	1451	2.17	2.68
Sabinene	971	4.80	0.55	Alloaromadendrene	1459	2.26	1.41
β -Pinene	983	0.31	–	β -Selinene	1489	–	1.56
β -Myrcene	989	0.41	–	Bicyclogermacrene	1500	19.51	14.14
β -Ocimene	1019	1.14	–	γ -Cadinene	1502	–	2.61
<i>p</i> -Cimene	1024	5.21	–	Valencene	1503	12.35	25.21
1,8-Cineole	1026	1.35	–	Germacrene A	1507	12.98	–
γ -Terpinene	1064	0.61	–	β -Cadinene	1518	0.51	–
Linalool	1096	–	1.85	Germacrene D	1557	–	1.56
Camphor	1142	–	0.34	Spathulenol	1571	2.61	3.28
Terpin-4-ol	1180	0.59	–	Caryophyllene oxide	1579	7.75	2.61
α -Copaene	1371	0.41	2.61	Humulene oxide	1609	–	1.31
β -Elemene	1389	1.20	3.11	Cubenol	1648	5.48	–
<i>Z</i> -Caryophyllene	1412	–	29.81	Total identified		99.11	95.83

Volatile constituent analysis was carried out on a Shimadzu GC-17A/MS QP5050A system (GC/MS system) employing a nonpolar capillary column, DB-1, of fused silica (30 m \times 0.25 mm i.d., 0.25 μ m pellicle), pumped by helium gas at 0.8 mL/min and with split mode. Injector and detector temperatures were set to 260°C and 270°C, respectively. Column temperature was programmed from 60°C (1.0 min) to 180°C (for 4 min) at 4°C/min, then from 180°C to 260°C at 6°C/min. Mass spectra were recorded from 43–450 *m/z*. Individual components were identified by comparison of their mass spectra at 70 eV to the Wiley built-in database and two additional computers using retention indices as pre-selection parameters [10], as well as by visually comparing standard fragmentation to those in the literature [11, 12].

GC analyses were performed on a Hewlett-Packard 5890 SERIES II instrument equipped with a flame ionization detector (FID) and a J & W Scientific DB-5 fused silica capillary column (30 m \times 25 μ m \times 0.25 μ m); column temperatures were programmed from 60°C for 1 min, raised to 180°C (for 4 min) at 4°C/min, then increased to 260°C at 10°C/min for integrating purposes. Injector and detector temperatures were 260°C and 270°C, respectively. Hydrogen was used as carrier gas, flow rate 1.5 mL/min, split mode (1:10). Injection volume, 1.5 μ L, solution of about 10 mg of oil in ethyl acetate. The retention indices were obtained by co-injecting the oil sample with a C₉–C₂₄ linear hydrocarbon mixture.

Larvicidal assessment was carried out according to the recommended methodology [13]. Samples from the essential oils were tested, in concentrations ranging from 25 to 500 μ g/mL and diluted in Tween 80/H₂O, against *Aedes aegypti* larvae in the third development stage. The experiment was triply run, accompanied by a Temephos® (*O,O'*-thiodi-4,1-phenylene) positive control and Tween 80/H₂O negative control. After 24 h, dead larvae percentages were calculated and used to obtain the LD₅₀ by a linear regression method, in which a sample is considered active when LD₅₀ < 1000 μ g/mL.

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