ORIGINAL RESEARCH

Glutaraldehyde‑assisted crosslinking in regenerated cellulose flms toward high dielectric and mechanical properties

Meng‑hang Gao · Xu Xie · Ting Huang · Nan Zhang · Yong Wang

Received: 5 February 2022 / Accepted: 28 July 2022 / Published online: 14 August 2022 © The Author(s), under exclusive licence to Springer Nature B.V. 2022

Abstract Developing the green dielectric materials satisfes the requirement of the sustainable development of society and economics. In this work, glutaraldehyde (GA)-assisted crosslinking strategy was developed to prepare the crosslinked regenerated cellulose (CRC) flms, and the efects of diferent crosslinking methods, including crosslinking steps, concentration of GA solution and crosslinking time, on dielectric and mechanical properties of the CRC flms were systematically investigated. Microstructure and morphology characterizations show that compared with the common RC flms, the CRC flms show apparently reduced defects and enhanced intermolecular interaction. At GA concentration of 6 vol%

and crosslinking time of 30 min, the CRC flm shows the lowest dielectric loss (0.03 at 1000 Hz, 92.3% smaller than RC flm) and the highest breakdown strength (336.55 MV m^{-1} , 364.3% higher than RC film), and simultaneously, the film shows the high tensile strength of 76.8 MPa and excellent tensile modulus of 6.08 GPa, about 240.9% and 104% higher than those of the RC flm, respectively. This work provides new insight in tailoring the dielectric and mechanical properties of the cellulose flms through constructing the crosslinking structure, which is of great signifcance for the fabrication of the high-performance cellulose-based dielectric materials.

Supplementary Information The online version contains supplementary material available at [https://doi.](https://doi.org/10.1007/s10570-022-04785-2) [org/10.1007/s10570-022-04785-2.](https://doi.org/10.1007/s10570-022-04785-2)

M. Gao · X. Xie · T. Huang · N. Zhang (\boxtimes) · Y. Wang (\boxtimes)

Key Laboratory of Advanced Technologies of Materials (Ministry of Education), School of Materials Science and Engineering, Southwest Jiaotong University, Chengdu 610031, China e-mail: sanyenan@163.com

Y. Wang e-mail: yongwang1976@swjtu.edu.cn

Graphical abstract

Keywords Cellulose · Crosslinking · Dielectric properties · Mechanical properties

Introduction

Recently, with the rapid development of social economy, energy problems are increasingly serious (Yu et al. [2017\)](#page-16-4). The preparation of green, environmentally friendly, and renewable energy storage materials is of great signifcance to the construction of a sustainable society. Among energy storage devices, dielectric capacitors with the advantages of high-power density, fast charging and discharging speed, etc., have gradually obtained the widespread attention (Liu et al. [2020](#page-15-5); Zhang et al. [2020](#page-16-5)). As the main materials of dielectric capacitors, dielectric materials with good dielectric properties have a wide range of applications in many civil and military felds, such as active vibration control (Sarban et al. [2011;](#page-16-0) Zhao et al. [2019](#page-16-1)), aerospace (Li et al. [2015,](#page-15-0) [2018\)](#page-15-1) and bioelectronics applications (Joyce et al. [2013\)](#page-15-2), etc.

As we all know, traditional dielectric polymers are often synthetic polymers based on petrochemical resources, such as polypropylene (PP) (Yuan et al. [2020\)](#page-16-2), poly(vinylidene fuoride) (PVDF) (Prateek et al. [2016\)](#page-16-3), poly(methyl methacrylate) (PMMA) (Zheng and Wong [2003\)](#page-17-0), polyimide (PI) (Kaltenbrunner et al. [2013\)](#page-15-3), etc. For example, Liu et al. [\(2014](#page-15-4)) prepared the PVDF composites containing high aspect ratio surface-hydroxylated $Ba_{0.6}Sr_{0.4}TiO_3$ nanotubes (BST-NT) and found that the composite with 10 vol% BST-NT showed a dielectric constant of 48.2 at a frequency of 1000 Hz, which was 6.1 times higher than that of the pure PVDF (7.9). However, these synthetic polymers have the common

characteristics, namely, they are non-biodegradable and non-renewable, and the fabrication of such dielectric materials usually leads to serious environmental pollutions, such as the widely existed "plastic microparticles" in water (Wang et al. [2020](#page-16-6)). With the depletion of resources and increasing environmental problems, obviously, it is of great signifcance to develop the green polymers-based dielectric materials.

Cellulose, as one of the most abundant materials in nature, has good biocompatibility, non-toxicity, biodegradability and recyclability (Seddiqi et al. [2021](#page-16-7)). In the microstructure of cellulose, 1,4-glycosidic bonds link β-D-glucopyranosyl groups to form cellulose molecular chains (Croll and Schroeder [2004](#page-15-6)). Due to the regular molecular chain structure and the large number of hydroxyl groups, there are many intermolecular and intramolecular hydrogen bonding interactions, which endow cellulose with promising physicochemical performances, such as good mechanical performances, high solubility resistance and good chemical resistance, etc. (Djahedi et al. [2016\)](#page-15-7). Cellulose is one of the most promising green materials, and the efective use of cellulose resources has become a research hotspot in the felds of chemistry, chemical engineering and materials science. To data, various structural and/or functional materials based on cellulose have been developed, and some of them have gained commercial applications (Moon et al. [2011;](#page-16-8) Movagharnezhad and Moghadam [2017\)](#page-16-9).

Developing the cellulose-based dielectric materials have also attracted much attention of researchers (He et al. [2021](#page-15-8); Zhao et al. [2021\)](#page-16-10). Among these researches, most of them were focused on incorporating nanoparticles into cellulose to prepare the dielectric composites, and these nanoparticles are various, including titanium dioxide $(TiO₂)$ (Madusanka et al. [2016;](#page-16-11) Tao et al. [2019\)](#page-16-12), magnetite (Arantes et al. [2019\)](#page-15-9), barium titanate (BaTiO₃) (Morsi et al. [2019](#page-16-13)), alumina (AI_2O_3) (Yin et al. [2021\)](#page-16-14), MXene (He et al. [2020\)](#page-15-10), Boron nitride (Lao et al. [2018a\)](#page-15-11), graphene oxide (GO) (Wang et al. [2018\)](#page-16-15), montmorillonite (Madusanka et al. [2017\)](#page-16-16), carbon nanotubes (Zeng et al. 2016), etc. For example, Zhang et al. (2019) (2019) prepared regenerated cellulose $(RC)/BaTiO₃$ nanocomposite flms and found that the composite flm with 2 vol $\%$ BaTiO₃ exhibited an ultrahigh discharged energy density of 13.14 J cm^{-3} at a breakdown strength of 370 MV m^{-1} . Lao et al. ([2018b\)](#page-15-11) prepared cellulose/boron nitride nanosheets (BNNS) composites, and the breakdown strength was greatly improved and the composites with 10 wt% BNNS exhibited the energy storage density of 4.1 J cm−3 and breakdown voltage of 370 MV m^{-1} .

Although the incorporation of nanoparticles leads to the great enhancement of dielectric constant of the cellulose-based dielectric composite, the apparent increase of dielectric loss is usually unavoidable (Song et al. [2022](#page-16-19); Wu et al. [2021](#page-16-20)). Furthermore, because of the poor melt-processing ability, most of the cellulose-based composites reported in literature have been prepared through solution compounding processing. It is well known to all that cellulose is diffcult to dissolve in common organic solvents, and the RC-based flms usually have many defects, which is very unfavorable for the suppression of the intrinsic dielectric loss of the cellulose matrix (Liu et al. [2021;](#page-16-21) Zhang et al. [2016\)](#page-16-22). Therefore, from a viewpoint of declaring microstructure-performance relationship, studying the dependence of dielectric properties of the cellulose-based materials on the intrinsic microstructures is of great signifcance. However, to date, less researches have been carried out to study the efects of molecular chain structure on the dielectric properties of the cellulose-based composites. Some researchers have pointed out that the introduction of functional groups in the molecular chains of cellulose can efectively tailor the dielectric properties (Takechi et al. [2016;](#page-16-23) Yang et al. [2018](#page-16-24)), while others investigated the dielectric performance dependence on molecular weights of cellulose (Yin et al. [2020\)](#page-16-25). However, more information about the fabrication method tailoring the dielectric performances of the cellulose-based materials is still highly required.

Herein, the environmentally friendly cotton cellulose was used to prepare the RC flm. During the RC flm fabrication, the crosslinking reaction was introduced with the aid of glutaraldehyde (GA) (Pereira et al. [2020\)](#page-16-26). Diferent strategies were developed to prepare the GA-crosslinked RC (CRC) flms and the dielectric and mechanical properties of the two kinds of cellulose flms were comparatively investigated. The results show that the crosslinking method exhibits great role in tailoring the comprehensive performances of the CRC flms. At appropriate fabrication conditions, the CRC flm shows largely reduced dielectric loss, highly enhanced breakdown strength

and high mechanical properties. This work provides a good opportunity for further fabricating the cellulosebased dielectric composites with promising dielectric performances.

Experimental section

Materials

Cotton cellulose fbers with degree of polymerization (DP) of 330 were provided by Xinxiang Chemical Fiber Company. Lithium hydroxide (LiOH), urea (Urea), and GA with 50 vol% aqueous solution were provided by Shanghai Aladdin Company, China. Sulfuric acid (98%) was provided by Chengdu Kelong Chemical Reagent Factory. All these chemical reagents have analytic purity.

Preparation of the common regenerated cellulose flms

The fabrication procedures of the RC flm were explained in Fig. [1.](#page-3-0) Cotton cellulose was immersed in 4% NaOH for 4 h under stirring, then it was washed to pH of 7 with water. Urea and LiOH were frstly dissolved in the deionized water to prepare the alkali/ urea (AU) solvent. The solvent was then cooled to −15 °C, and a certain mass of dried cellulose was added into it. The suspension was stirred at 1800 rpm for 5 min to prepare the cellulose solution with a concentration of 4 wt%. After that, the cellulose solution was centrifuged at 5000 rpm for 10 min to remove air bubbles in the solution. Subsequently, the centrifuged solution was coated on a glass plate with a spatula, and this glass plate was subsequently soaked in the 5% H₂SO₄ solution for 5 min. In order to prevent the flm from deforming, the two sides of the flm were clamped with clips. Finally, the flm was washed with deionized water to remove the small molecule impurities, and then placed in an oven to be dried to obtain the RC flm.

Preparation of the crosslinked regenerated cellulose flms

The GA-assisted crosslinking of the RC flms was carried out through diferent strategies. The frst one is crosslinking in solution. Namely, once cellulose was completely dissolved in the solvent system, 10 mL GA solution with a concentration of 2 vol% was added. After being continuously stirred for 20 min to ensure the completely crosslinking by GA, the CRC flm was obtained through centrifugation, flm-forming and drying steps as mentioned above. The sample notation is defned as CRC-I.

The second strategy is crosslinking before gelation. Namely, when the centrifuged solution was coated on the glass plate to obtain the flm, the glass plate was immersed into the GA solution for 20 min

Fig. 1 Schematic diagram showing the preparation procedures of the common RC and CRC flms

Fig. 2 Chemical principle diagram of glutaraldehyde crosslinked regenerated cellulose flm

at 60 °C, after that the glass plate was taken out and immersed into the H_2SO_4 solution for gelation. After being successively washed and dried, the CRC flm was obtained. This kind of flm is labelled as CRC-II.

The third strategy is crosslinking after gelation. Namely, once the RC flm was obtained in the H_2SO_4 solution, it was taken out and immersed into the GA solution for 20 min at 60 °C. After that, the crosslinked RC flm was taken out, washed and dried successively to obtain the target flm. The sample notation was then defned as CRC-III. First, diferent crosslinked RC flms were fabricated through changing the concentrations of GA solution (2, 4, 6, and 8 vol%) while the crosslinking reaction time was kept at 20 min, and the sample notation was named as *a*-CRC-20, where *a* represents the concentration of GA solution. Second, the crosslinking reaction time was varied at 20, 30, 40 and 60 min and the GA

concentration was kept at 6 vol%, and correspondingly, the sample notation was named as 6-CRC-*b*, where *b* represents the crosslinking reaction time. According to the literature (Aburabie et al. [2021](#page-15-12)), GA-assisted crosslinking of cellulose molecular chains can be explained in Fig. [2.](#page-4-0)

Characterization and measurements

The surface morphologies of the RC flms were characterized using a scanning electron microscope (SEM, QUANTA FEG 450, USA) at the operating voltage of 2.7 kV and current of 83 mA and an atomic force microscope (AFM, BRUKER Multimode8, Germany) via the contact mode. An X-ray photoelectron spectrometer (XPS, Thermo Scientifc K-Alpha, America) was used to detect the chemical features of the RC and CRC flms. The chemical structures of the

RC and CRC flms were further characterized using a Fourier transform infrared spectrometer (FTIR, BRUKER TENSOR II, Germany) in the transmission mode, and the wavenumber range was set at 400–4000 cm^{-1} with a resolution of 4 cm^{-1} . Before measurements, the samples were placed in an oven set at 60 °C for 24 h and then storied in a desiccator to reduce the infuence of humidity on the characterization. The crystalline structure of the RC flms was characterized using a wide-angle X-ray difractometer (WAXD, Panalytical Empyrean, Netherlands) with the scanning angle range of 10° -60 $^{\circ}$. The thermal stability of the cellulose flms was evaluated through thermogravimetric analysis (TGA) on a G209F1 Libra (Netzsch, Germany). During the measurements, about 8 mg sample was heated from 30 to 800 °C at a heating rate of 10 $^{\circ}$ C min⁻¹ in nitrogen atmosphere.

The optical transmittance was measured by an ultraviolet–visible spectrometer (UV–Vis, SHI-MADZU, Japan). The wavelength range was set at 400–800 nm. The electrical conductivity and the dielectric property measurements were conducted on a broad frequency dielectric spectrometer, Concept 80 (Novocontrol, Germany), and the measurements were carried out at 23 °C and 220 V in the frequency range of 10^2 – 10^7 Hz. The dielectric breakdown strength of the sample was measured using a breakdown voltage tester (BDJC-50 KV, Beiguang Jingyi Instrument, PR China). The voltage was increased at a rate of 1.5 kV s^{-1} using an alternating current, and the highest measured voltage was 30 kV. The D-E loops were measured using a ferroelectric material tester (aixACCT TF 2000E, Germany) at room temperature (25 °C). The surface of sample was coated with a thin layer of gold before measurements. The universal testing machine (XS(08)XT-3, Xusai, Shanghai) was used to measure the tensile properties at room temperature of 25 °C and humidity of 55% . The width of the sample was 5 mm, the gauge length was set at 35 mm, and the cross-head speed was 1 mm min⁻¹. The measurement for each kind of flm was repeated for 5 times and the average value was reported.

Result and discussion

GA-assisted crosslinking in the regenerated cellulose flm

According to the literature (Hou et al. [2019\)](#page-15-13), under the acidic condition, the aldehyde groups of GA can react with the hydroxyl groups of cellulose molecular chains and thereby, GA can be the crosslinking agent of the cellulose, constructing the chemical crosslinking bonds between molecular chains of cellulose. The CRC flm was obtained via crosslinking in 6 vol% GA solution for 30 min after gelation (corresponding to the CRC-III flm). Figure [3](#page-6-0) shows the comparison of the surface morphologies of the RC and CRC flms obtained through SEM and AFM characterizations, and the roughness parameters obtained through AFM characterizations are also provided. Here, the crosslinking reaction was introduced after gelation of cellulose in H_2SO_4 solution. Different from the relative rough surface with many holes of the common RC film (Fig. $3a_1$ $3a_1$, a_2), the CRC film (Fig. $3b_1$ $3b_1$, $b₂$) exhibits smooth surface and fewer holes. Through AFM characterizations, the roughness parameters $(R_a$ and R_a , which represent the root-mean-square roughness and average surface roughness of the flm, respectively) of the flms can be obtained. As shown in Fig. $3a_{4, b4}$ $3a_{4, b4}$, the CRC film shows much more homogeneous surface with lower roughness parameters $(R_a$ of 16.7 nm and R_a of 26.9 nm) compared with the common RC sample, which shows the R_a of 54.6 nm and R_a of 42.8 nm, respectively. The above results clearly confrm that the GA-assisted crosslinking is favorable for improving the flm-forming quality and reducing the defects in the cellulose flms.

To better understand the efects of GA-assisted crosslinking on microstructure and intermolecular interaction of cellulose flm, more characterizations were carried out. Figure [4a](#page-7-0), b show the C 1s spectra of the common RC and crosslinked CRC flms, respectively. For the common RC flm, it shows the characteristic peaks at 287.6, 286 and 284.3 eV, relating to the C=O, C–O and C–C bonds, respectively (Fras et al. [2005](#page-15-14)). The CRC flm shows the completely same characteristic peaks with very small changes of peak positions. However, one can see that the CRC flms show much larger peak area of C–O bonds, this clearly confrms the crosslinking

Fig. 3 SEM (a_1, a_2, b_1, b_2) and AFM (a_3, a_4, b_3, b_4) images showing the surface morphologies of the RC (a_1-a_4) and CRC (b_1-b_4) flms characterized at diferent modifcations. The CRC flm was obtained via crosslinking in 6 vol% GA solution for 30 min

of cellulose molecules by GA, leading to more C–O bonds in the CRC flm. By the way, the presence of C=O bonds can be attributed to the presence of a few urea and GA molecules in the CRC flm.

Fig. 4 a, **b** XPS spectra of C 1s of the common RC and CRC flms, respectively. **c** FTIR spectra, **d** WAXD profles, **e** TGA curves and **f** transmittance of the common RC and CRC flms.

The CRC flm was obtained via crosslinking in 6 vol% GA solution for 30 min

Figure [4](#page-7-0)c shows the FTIR spectra of the common RC and CRC flms. For the common RC flm, there is a strong characteristic absorption band at about 3422 cm−1 relating to the stretching vibration of –OH groups. While for the CRC flm, the absorption band of –OH shifts to higher wavenumbers on the one hand. On the other hand, the intensity of the absorption band is apparently reduced. These changes confrm that crosslinking weakens the hydrogen bonding interaction between cellulose main chains. First, the occurrence of crosslinking consumes some hydroxy groups of cellulose molecular chains because the crosslinking occurs between the hydroxy groups of cellulose and aldehyde groups of GA (Aburabie et al. [2021\)](#page-15-12). Second, the presence of GA molecules possibly increases the distance between chain segments of cellulose, reducing the probability of forming intramolecular hydrogen bonding interaction. Furthermore, one can see that the characteristic absorption bands at 1162 cm^{-1} and 1028 cm^{-1} , which are usually attributed to the stretching vibration of C–O–C and bending vibration of C–O, respectively, becomes more apparent. This further confrms the successful crosslinking of cellulose molecules by GA (Cuba-Chiem et al. [2008\)](#page-15-15).

Figure [4](#page-7-0)d shows the comparison of the WAXD profles of RC and CRC flms. It is well known that the main crystal form of the RC flm obtained by dissolution and regeneration procedures is cellulose II, which exhibits the characteristic difraction peaks at around 12°, 20° and 22° attributed to the $(1\overline{1}0)$, (110) and (020) crystal plane respectively (French [2014](#page-15-16)). The position and intensity of characteristic peaks before and after crosslinking are basically unchanged, indicating that crosslinking has no signifcant infuence on the crystal structure of cellulose (Hou et al. [2019\)](#page-15-13).

Furthermore, the thermal stability of the RC and CRC flms were also evaluated by TGA measurements, and the results are shown in Fig. [4](#page-7-0)e and Figure S1. Compared with the RC flm, which shows the thermal decomposition temperature at around 329.3 °C, the CRC flm exhibits slightly enhanced decomposition temperature at around 339.7 °C, indicating that crosslinking facilitates the slight enhancement of the

Fig.5 a–**d** Frequency dependence of dielectric constant, dielectric loss and conductivity of the cellulose flms without and with diferent crosslinking methods, **e**, **f** the Weibull distribu-

tion diagram of the breakdown strength of cellulose flms and the corresponding feature breakdown strength

thermal stability to a certain extent. The light transmittance of the RC and CRC flms are shown in Fig. [4](#page-7-0)f. Here, the light transmittance was measured at wavelength range of 400–800 nm, and the ultraviolet transmittance at 550 nm is used to represent the transparency of the flms. The RC flm shows the transmittance of about 79.74%, while for the CRC flm, the transmittance is slightly enhanced to 83.18%. The improvement of the transmittance can be attributed to the more homogenous structure with fewer holes (defects) as well as the smooth surface of the CRC flm as previously confrmed by morphological characterizations. Whatever, the above results clearly show that GA-assisted crosslinking in the CRC flm are successfully achieved, and the CRC flm has fewer defects compared with the common RC flm. This is believed to be favorable for the improvement of the dielectric and mechanical properties.

Furthermore, the effects of crosslinking methods on the microstructures of the CRC flms were comparatively investigated through the above characterization methods, and the results are shown in Figures S2–S5. According to the results of FTIR,

the degree of the crosslinking is calculated, and the results are listed in Table S1. Similarly, the CRC-I and CRC-II samples show the similar crystalline structures with the RC and CRC-III samples as shown in Fig. [4](#page-7-0)d.

The effects of crosslinking methods on the dielectric and mechanical properties of cellulose flms

Then, diferent crosslinking methods were used to crosslink the cellulose, and the dielectric properties were measured, so that the appropriate crosslinking method could be determined. Here, the CRC flm was prepared by diferent strategies as mentioned in 2.3. Figure $5a-c$ $5a-c$ show the frequency dependence of dielectric constant (ε') , dielectric loss $(\tan \delta)$ and comparison of the data collected at 1000 Hz, respectively, and the variations of the conductivity of the RC and CRC flms are shown in Fig. [5d](#page-8-0). The efects of crosslinking on dielectric properties of the RC flm are dramatic at the low frequency ranges. Generally, there are several polarization modes in polymer and/or polymer composites, such as atomic

polarization, electronic polarization, dipole polarization, ionic polarization and interfacial polarization, etc. In the polymer composites, the interfacial polarization mainly determines the dielectric properties of the composites and the polarization occurs at the frequency range of 10^0 – 10^2 Hz, while in the polymer with polar groups, the dipole polarization exhibits the determinable role, which occurs at the larger frequency ranges compared with the polymer composites (Bonardd et al. [2019](#page-15-17)). According to Fig. [5](#page-8-0) it is deduced that the crosslinking mainly afects the dipole polarization of the cellulose flm. For the common RC flm, it exhibits the relatively good dielectric properties with ε' of 16.72 and tan δ of 0.39 at frequency of 1000 Hz. Compared with the common RC film, the CRC film prepared through introducing crosslinking before gelation (CRC-II) exhibits the highest ε' and $\tan \delta$ nearly at all frequency ranges. And at frequency of 1000 Hz, ε' and tan δ achieve 23.95 and 0.65, respectively. However, the CRC flm prepared through introducing crosslinking after gelation (CRC-III) exhibits the lowest ε' and tan δ at all frequency ranges, and in this condition, ε' and tan δ are only 9.65 and 0.17 at 1000 Hz, much smaller than those of the common RC flm. This confrms that crosslinking method has great role in tailoring the dielectric properties of the CRC flm. Figure [5](#page-8-0)d shows the frequency dependence of conductivity. Compared with the common RC flm, the CRC flms show the similar variation trends to those of the dielectric properties.

The diferent variation trends of the dielectric properties are mainly related to the occurrence of the crosslinking reaction and the resultant molecular chain interaction, especially the degree of crosslinking in the flms. Generally, GA-induced crosslinking reaction occurs in the acidic condition (Aburabie et al. [2021](#page-15-12)). For the CRC-I flm, the crosslinking occurred in the LiOH/Urea solvent system and in this condition, to ensure the complete dissolution of cellulose and prevent the gelation of cellulose, the added amount of GA solution was relatively small, only 10 mL and therefore, the degree of the crosslinking is relatively small, about 1.286% as confrmed in Table S1. For the CRC-II flm, the crosslinking reaction was introduced before gelation and in this condition, some impurities (LiOH, Urea, GA and their reaction products with H_2SO_4) could be knocked in the flm during the subsequent gelation process in the H_2SO_4 solution, which possibly contribute to the polarization of the flm in the electric feld on the one hand. On the other hand, the chemical crosslinking structure introduced by GA most likely afects the gelation of cellulose in the H_2SO_4 solution. For example, the skeleton of GA molecules not only increases the distance between main chains of cellulose but also enhances the rigidity of molecular chains, which reduce the formation ability of hydrogen bonds between cellulose molecular chains. Furthermore, one can see that the degree of the crosslinking in the CRC-II flm is the smallest among the CRC flms, only about 0.59% (seen in Table S1). Therefore, in the CRC-II flm, the molecular chains of cellulose have higher mobility and the dipole polarization becomes more available. However, for the CRC-III flm, after gelation in H_2SO_4 , the intermolecular and intramolecular hydrogen bonds already form in the gel and in this condition, some of the hydroxy groups in the cellulose molecular chains have already been consumed and therefore, the subsequent crosslinking reaction with GA only occurs between the free hydroxy groups of cellulose and GA and in this condition, the flms possibly have the highest interaction among molecular chain segments. As shown in Table S1, the CRC-III samples have much higher degrees of crosslinking compared with the CRC-I (1.29%) and CRC-II (0.59%). The 2-CRC-20 and 6-CRC-30 flms show the degrees of crosslinking of 4.97% and 8.69%, respectively. In this condition, the mobility of cellulose molecular chains is greatly suppressed and consequently, the CRC-III flm exhibits the low polarization efect in the electrical feld.

Figure [5](#page-8-0)e, f show the Weibull-distribution plots of breakdown strengths of the diferent cellulose flms and the comparison of the feature breakdown strengths. The Weibull-distribution function is defned through the equation (Almalki and Nadarajah [2014\)](#page-15-18):

$$
P = 1 - \exp\left[-\left(\frac{E}{\alpha}\right)^{\beta}\right] \tag{1}
$$

where *P* and *E* represent the cumulative probability of electric failure and the breakdown strength obtained during the measurements, respectively. α is usually used to represent the feature breakdown strength (E_{bd}) of the sample, i.e. the breakdown strength obtained at the cumulative failure probability of

Fig. 6 a Tensile stress–strain curves of crosslinking RC flms with diferent crosslinking methods, **b** comparison of the tensile properties among diferent membrane samples as indicated

63.2%. β is a parameter that can be used to describe the reliability of the dielectric materials and larger β indicates the higher reliability. According to Eq. [\(1](#page-9-0)), the plots of log $[-ln(1-p)]$ versus $log E$ can be illustrated, and then the value of α and β are also obtained. The detailed data analysis can be seen in literature (Lu et al. 2019). As shown in Fig. [5e](#page-8-0), f, the common RC flm shows the breakdown strength of about 72.49 MV m^{-1} . The low breakdown strength can be partially attributed to the poor flm quality with many defects (holes) as confrmed by morphological characterizations. It is worth noting that the flm formation of common RC occurs in the H_2SO_4 solution and in this condition, most of the molecular chains are physically crosslinked and the role of chemical crosslinking is very small. The CRC-I and CRC-II shows the similar E_{bd} to that of the common RC film. However, the CRC-III film shows the highest β and the feature breakdown strength achieves 83.66 MV m−1. The enhanced breakdown strength can be attributed to the high quality of the flm with fewer defects and smoother surface. This indicates that the CRC-III flm has high reliability and it is possibly the most appropriate one among these cellulose flms if they are used as the dielectric materials. Therefore, the CRC-III flm is selected in the following section to further declare the efects of crosslinking on dielectric and mechanical properties of the CRC flm.

Figure [6](#page-10-0) shows the typical stress–strain curves and the corresponding mechanical property parameters of the RC and CRC flms obtained after diferent crosslinking methods. The CRC-I and CRC-II flms show the similar tensile strength and tensile modulus to those of the common RC flm, which exhibits the tensile strength and tensile modulus of 22.53 MPa and 2.98 GPa, respectively, but they exhibit lower elongation at break. Among all these CRC flms, the CRC-III flm has much higher tensile strength and tensile modulus, and the elongation at break is also comparable to that of the common RC flm. It has already been reported that higher mechanical properties usually lead to larger breakdown strength in the electric feld (Dan et al. [2019](#page-15-19)). And therefore, the reason why the CRC-III film shows the highest^{β} and E_{bd} among these CRC flms can be partially attributed to their best mechanical properties. Whatever, the above results clearly confrm that the CRC-III flm has the lowest tan δ , highest β and E_{bd} , and best mechanical properties, and it has great potential as the dielectric material.

The effects of GA concentration on the dielectric and mechanical properties of cellulose flm

As mentioned above, constructing crosslinking structure is favorable for improving the comprehensive properties of the CRC flms. It is well known to all

Fig. 7 a–**d** Frequency dependence of dielectric constant, dielectric loss and conductivity of the CRC (CRC-III) flms prepared at diferent GA concentrations, **e**, **f** the Weibull dis-

that the degree of the crosslinking is greatly dependent upon the concentration of GA solution during the crosslinking reaction. Obviously, lower GA concentration leads to smaller degree of crosslinking at the same reaction time while larger GA concentration possibly leads to excessive crosslinking in the flm, which is unfavorable for the polarization of molecules in the electric fled. Most likely, there is an appropriate GA concentration for the CRC flm if it is used as the dielectric material. The following section is to declare the dependence of the dielectric and mechanical properties of the CRC-III flm on the GA

concentration. Figure [7](#page-11-0) shows the dielectric properties of the CRC-III flms prepared at diferent GA concentrations (2, 4, 6 and 8 vol%). The crosslinking reaction time was kept at 20 min. From Fig. [7](#page-11-0)a–c it can be seen that ε' and tan δ firstly decrease with increasing GA concentrations and the lowest values are achieved at GA concentration of 6 vol%. In this condition, ε' of the 6-CRC-20 flm is only 5.77 at 1000 Hz, about 65.5% lower than that of the RC flm (16.72). However, it is also noticed that the tan δ of the 6-CRC-20

tribution diagram of the breakdown strength of these CRC flms and the corresponding feature breakdown strength. The crosslinking reaction time was kept at 20 min

flm is also greatly reduced to 0.05 at 1000 Hz, about 87.2% lower than that of the RC sample. Obviously, the reduction of the tan δ is much more apparent than that of the ε' . Further increasing GA concentration leads to higher ε' and tan δ . The variations of the dielectric properties agree well with the change of the conductivity of the flms as shown in Fig. [7d](#page-11-0). At low frequency ranges, the 6-CRC-20 flm shows the lowest conductivity while the 8-CRC-20 flm shows the highest one. Whatever, all the CRC flms show the lower conductivities compared with the RC flm. As expected, increasing GA concentration leads to higher E_{bd} . Among these films, the 6-CRC-20 film has the highest E_{bd} (124.58 MV m⁻¹), which is 71.9% higher than that of the pure RC film (72.49 MV m⁻¹), while the 8-CRC-20 film has the smallest E_{bd} $(58.23 \text{ MV m}^{-1})$, about 19.7% lower than that of the RC flm.

The effects of GA concentration on the dielectric properties of the CRC flms can be explained as follows. As mentioned above, crosslinking restricts the motion of diploes in the electric feld and obviously, higher degree of crosslinking results in lower mobility

Fig. 8 a Stress–strain curves of the common RC and the *a*-CRC-20 flms and **b** the corresponding mechanical property parameters

of diploes, which leads to lower polarization degree of the CRC flm. But higher degree of crosslinking also leads to stronger intermolecular interaction, which is favorable for the enhancement of the electric breakdown strength of the CRC flm. Here, at relatively low GA concentration (2 vol%), the crosslinking reaction is relatively few in the CRC flm, resulting in relatively lower degree of crosslinking.

Fig. 9 a–**d** Frequency dependence of dielectric constant, dielectric loss and conductivity of the CRC (CRC-III) flms prepared at diferent reaction time, **e**, **f** the Weibull distribution

diagram of the breakdown strength of these CRC flms and the corresponding feature breakdown strength. The crosslinking reaction occurred in the 6 vol% GA solution

Increasing GA concentrations leads to higher degree of crosslinking in the CRC film, and therefore, ε' and tan δ decrease while E_{bd} increases, until the minimum ε' and tan δ and the maximum E_{bd} are achieved at GA concentration of 6 vol%. However, at relatively high GA concentration (8 vol%), possibly some residual GA molecules are present in the CRC flm, and these free GA molecules may contribute to the dramatic increase of ε' and tan δ . The other possibility is that at relative high crosslinking reaction rate resulted by higher GA concentration, a small quantity of water may be knocked in the CRC flm and cannot be easily removed during the subsequent drying treatment.

The effects of GA concentrations on the mechanical properties of the CRC-III flms are shown in Fig. [8](#page-12-0). Among these flms, the 6-CRC-20 samples shows the largest tensile strength (61.4 MPa), tensile modulus (4.80 GPa) and the tensile ductility (3.47%). Through constructing the double-network structures in the hydrogel to improve the strength and toughness has already been widely reported in literature (Zhao et al. [2016](#page-16-28)). In this work, with the increase of GA concentration, the degree of the crosslinking in the CRC flms increases gradually and consequently, the enhanced intermolecular interaction leads to better strength and toughness. However, the 8-CRC-20 flm exhibits worse mechanical properties, which are even lower than those of the RC flm. The excessive crosslinking with extremely high crosslinking degree may greatly restricts the mobility of cellulose molecules, leading to lower tensile ductility on the one hand. On the other hand, the residual GA molecules may exhibit the plasticizing efect, leading to the reduction of both tensile strength and tensile modulus. Whatever, the above results clearly confrm that among these CRC flms, the 6-CRC-20 flm is possibly the appropriate candidate as the dielectric material from a viewpoint of the relatively good comprehensive physical properties.

The effects of crosslinking reaction time on the dielectric and mechanical properties of cellulose flm

The following section is then clarifying the efect of crosslinking reaction time on the dielectric and mechanical properties of the CRC-III flms. According to the previous results, the concentration of GA was maintained at 6 vol% while the crosslinking reaction time was varied from 20 to 60 min. As shown in Fig. [9](#page-12-1), ε' and tan δ tend to decrease at crosslinking time of 20 and 30 min, and the minimum values are achieved at reaction time of 30 min. Then, further increasing reaction time results in the gradual increase of the ε' and tan δ . The conductivity shows the similar variation trends with ε' and tan δ at low frequency ranges. While the E_{bd} shows the inverse variation trend. Namely, the highest E_{bd} is achieved at reaction time of 30 min. Therefore, from a view point of dielectric properties, the 6-CRC-30 flm is possibly the most appropriate candidate as dielectric material.

Fig. 10 a Stress–strain curves of the common RC and the 6-CRC-*b* flms and **b** the corresponding mechanical property parameters

It shows the lowest tan δ of 0.03 at the frequency of 1000 Hz and the highest E_{bd} of 336.55 MV m⁻¹, which are about 92.3% lower and 363% higher than the tan δ (0.39) and E_{bd} (72.49 MV m⁻¹) of the common RC flm, respectively.

Furthermore, the energy storage performances of the RC and 6-CRC-30 were measured, and the results are shown in Figure S6. It can be seen that the crosslinked flm can withstand a higher electric field, about 350 MV m^{-1} , which is much higher than that of the common RC sample (80 MV m^{-1}). Under the electric field of 50 MV m^{-1} , the maximum polarization of RC and 6-CRC-30 is corresponding to 0.3478 μ C cm⁻² and 0.23482 μ C cm⁻², respectively. On the one hand, the decrease of the maximum polarization in the 6-CRC-30 sample indicates that the crosslinked flm has lower polarization ability due to the restricted movement of the dipole induced by crosslinking. On the other hand, crosslinking makes the cellulose have fewer defects and stronger interaction, which can withstand higher electric feld and have higher energy storage density. From Figure S6c and Figure S6d, when the electric feld intensity is 350 MV m^{-1} , the energy storage density of the 6-CRC-30 is 5.88 J cm^{-3} and the discharge energy storage density is 2.14 J cm⁻³. This clearly confirms that crosslinking is an efficient strategy in enhancing the energy storage ability of the cellulose-based flm.

Similarly, the mechanical properties of the 6-CRC-*b* flms were also measured, and the results are illustrated in Fig. [10.](#page-13-0) Compared with the common RC flm, all the 6-CRC-*b* flms show the greatly enhanced mechanical properties. Similar to the variation trend of the dielectric breakdown strength, the tensile strength and tensile modulus initially increase with increasing crosslinking reaction time, and the maximum values are achieved for the 6-CRC-30 flm, which exhibits the tensile strength of 76.8 MPa and tensile modulus of 6.08 GPa. Compared with the common RC flm, the tensile strength and tensile modulus of the 6-CRC-30 flm are increased by 348.88% and 171.14%, respectively. However, further increasing reaction time leads to the slight reduction of both tensile strength and tensile modulus. The similar variation trends of dielectric breakdown strength and mechanical properties further confrm that there is strong relationship between them, and better mechanical properties usually ensure the higher breakdown strength of the CRC flms.

According to the above results, it can be clearly seen that the dielectric and mechanical properties of the CRC flms are greatly dependent upon the fabrication methods. For the GA-assisted crosslinking, there are most appropriate crosslinking parameters, such as the appropriate GA concentration $(6 \text{ vol}\%)$ and appropriate crosslinking time (30 min). After being crosslinked, the mobility of the chain segments is reduced, which leads to the lower degree of polarization in the electrical field, resulting in lower ε' and tan δ . Due to the enhanced mechanical properties and the reduced defects, the CRC flms show enhanced E_{bd} . However, the above results also indicate at least that there is a contradiction in enhancing the ε' , E_{ba} and simultaneously reducing the tan δ in the present state. More work is still required to further improve the dielectric and mechanical properties of the cellulose-based materials. One possible solution is simultaneously introducing the polar groups in the molecular chains and the crosslinking structure to enhance ε' and suppress tan δ synchronously, and the other possible solution is incorporating nanoparticles but simultaneously restricting the mobility of cellulose chains nearby the nanoparticles. The related work is being carried out in our group and will be reported in the near future.

Conclusion

In summary, diferent crosslinking methods have been developed to prepare the cellulose-based dielectric materials in this work. First, the effects of crosslinking on the morphology and microstructures of the regenerated cellulose flm have been systematically investigated, and the results show that GA-assisted crosslinking improves the flm-forming quality of the RC flm with fewer defects. Then, the effects of crosslinking steps, the concentration of GA solution and the crosslinking time on the dielectric and mechanical properties of the CRC flms have been comparatively studied. The results show that immersing cellulose hydrogel into GA solution is the most efficient way to achieve relatively good comprehensive properties. Furthermore, there are appropriate GA concentration (6 vol%) and crosslinking time (30 min), at which the

CRC flm exhibits relatively good dielectric properties, including the lowest tan δ of 0.03 at frequency of 1000 Hz and the highest E_{bd} of 336.55 MV m⁻¹, and simultaneously, the tensile strength and tensile modulus achieve 76.8 MPa and 6.08 GPa, respectively. This work confrms that introducing crosslinking structure is the highly efficient way to synchronously tailor the dielectric and mechanical properties of the regenerated cellulose flms, which provides signifcant information for the fabrication of the high-performance cellulose-based dielectric materials.

Acknowledgments SEM characterizations were supported by the Analytical and Testing Center of Southwest Jiaotong University. Thanks to Mr. Zhen-jie Lu (Southwest Jiaotong University) for the help with our AFM tests.

Author contribution MG: Conceptualization, Methodology, Data curation, Formal analysis, Writing—original draft, Writing—review & editing. XX: Formal analysis, Investigation, Visualization. TH: Investigation, Discussion. NZ Investigation, Discussion, Funding acquisition, Writing—review & editing. YW: Conceptualization, Investigation, Funding acquisition, Project administration, Supervision, Writing—review & editing.

Funding This work was fnancially supported by the National Natural Science Foundation of China (51673159), the Youth Science and Technology Innovation Team of Sichuan Province of Functional Polymer Composites (2021JDTD0009) and the Sichuan Science and Technology Program (2020YFG0099).

Declarations

Confict of interest The authors declare that they have no known competing fnancial interests or personal relationships that could have appeared to infuence the work reported in this paper.

References

- Aburabie J, Lalia B, Hashaikeh R (2021) Proton conductive, low methanol crossover cellulose-based membranes. Membranes 11:539
- Almalki SJ, Nadarajah S (2014) Modifcations of the Weibull distribution: a review. Reliab Eng Syst Saf 124:32–55
- Arantes ACC, Silva LE, Wood DF, Almeida CD, Tonoli GHD, de Oliveira JE, da Silva JP, Williams TG, Orts WJ, Bianchi ML (2019) Bio-based thin flms of cellulose nanofbrils and magnetite for potential application in green electronics. Carbohydr Polym 207:100–107
- Bonardd S, Moreno-Serna V, Kortaberria G, Diaz DD, Leiva A, Saldias C (2019) Dipolar glass polymers containing polarizable groups as dielectric materials for energy storage applications. A minireview. Polymers 11:317
- Croll DC, Schroeder LR (2004) Synthesis of a ring-rigid disaccharide model for studies of alkaline chain cleavage in cellulose. J Wood Chem Technol 24:27–38
- Cuba-Chiem LT, Huynh L, Ralston J, Beattie DA (2008) In situ particle flm ATR FTIR spectroscopy of carboxymethyl cellulose adsorption on talc: binding mechanism, pH efects, and aadsorption kinetics. Langmuir 24:8036–8044
- Dan ZK, Jiang JY, Qian JF, Shen ZH, Li M, Nan CW, Shen Y (2019) A ferroconcrete-like all-organic nanocomposite exhibiting improved mechanical property, high breakdown strength, and high energy efficiency. Macromol Mater Eng 304:1900433
- Djahedi C, Bergenstråhle-Wohlert M, Berglund LA, Wohlert J (2016) Role of hydrogen bonding in cellulose deformation: the leverage effect analyzed by molecular modeling. Cellulose 23:2315–2323
- Fras L, Johansson LS, Stenius P, Laine J, Stana-Kleinschek K, Ribitsch V (2005) Analysis of the oxidation of cellulose fbres by titration and XPS. Colloid Surface A 260:101–108
- French AD (2014) Idealized powder difraction patterns for cellulose polymorphs. Cellulose 21:885–896
- He P, Cao MS, Cai YZ, Shu JC, Cao WQ, Yuan J (2020) Selfassembling fexible 2D carbide MXene flm with tunable integrated electron migration and group relaxation toward energy storage and green EMI shielding. Carbon 157:80–89
- He J, Yin Y, Xu M, Wang P, Yang Z, Yang Q, Shi Z, Xiong C (2021) Regenerated Cellulose/NaNbO₃ nanowire dielectric composite flms with superior discharge energy density and efficiency. ACS Appl Energy Mater 4:8150-8157
- Hou T, Guo K, Wang Z, Zhang X-F, Feng Y, He M, Yao J (2019) Glutaraldehyde and polyvinyl alcohol crosslinked cellulose membranes for efficient methyl orange and Congo red removal. Cellulose 26:5065–5074
- Joyce DM, Venkat N, Ouchen F, Singh KM, Smith SR, Grote JG (2013) DNA hybrid dielectric flm devices for energy storage and bioelectronics applications. Paper presented at: conference on nanobiosystems-processing, Characterization, and Applications VI San Diego
- Kaltenbrunner M, Sekitani T, Reeder J, Yokota T, Kuribara K, Tokuhara T, Drack M, Schwodiauer R, Graz I, Bauer-Gogonea S et al (2013) An ultra-lightweight design for imperceptible plastic electronics. Nature 499:458–463
- Lao J, Xie H, Shi Z, Li G, Li B, Hu G-H, Yang Q, Xiong C (2018a) Flexible regenerated cellulose/boron nitride nanosheehigh-temperature dielectric nanocomposite flms with high energy density and breakdown strength. ACS Sustain Chem Eng 6:7151–7158
- Li Q, Chen L, Gadinski MR, Zhang SH, Zhang GZ, Li HY, Haque A, Chen LQ, Jackson TN, Wang Q (2015) Flexible high-temperature dielectric materials from polymer nanocomposites. Nature 523:576–579
- Li Q, Yao FZ, Liu Y, Zhang GZ, Wang H, Wang Q (2018) High-temperature dielectric materials for electrical energy storage. Annu Rev Mater Sci 48:219–243
- Liu SH, Xue SX, Zhang WQ, Zhai JW, Chen GH (2014) Signifcantly enhanced dielectric property in PVDF nanocomposites fexible flms through a small loading of surface-hydroxylated $Ba_{0.6}Sr_{0.4}TiO_3$ nanotubes. J Mater Chem A 2:18040–18046
- Liu LM, Qu JL, Gu AJ, Wang BH (2020) Percolative polymer composites for dielectric capacitors: a brief history,

materials, and multilayer interface design. J Mater Chem A 8:18515–18537

- Liu QL, Yu HT, Mu TC, Xue ZM, Xu F (2021) Robust superbase-based emerging solvents for highly efficient dissolution of cellulose. Carbohydr Polym 272:118454
- Lu Y, Xie X, Wang WY, Qi XD, Lei YZ, Yang JH, Wang Y (2019) ZnO nanoparticles-tailored GO dispersion toward fexible dielectric composites with high relative permittivity, low dielectric loss and high breakdown strength. Compos Part A Appl Sci Manuf 124:105489
- Madusanka N, Shivareddy SG, Hiralal P, Eddleston MD, Choi Y, Oliver RA, Amaratunga GAJ (2016) Nanocomposites of TiO₂/cyanoethylated cellulose with ultra high dielectric constants. Nanotechnology 27:321
- Madusanka N, Shivareddy SG, Eddleston MD, Hiralal P, Oliver RA, Amaratunga GAJ (2017) Dielectric behaviour of montmorillonite/cyanoethylated cellulose nanocomposites. Carbohydr Polym 172:315–321
- Moon RJ, Martini A, Nairn J, Simonsen J, Youngblood J (2011) Cellulose nanomaterials review: structure, properties and nanocomposites. Chem Soc Rev 40:3941–3994
- Morsi MA, Abdelaziz M, Oraby AH, Mokhles I (2019) Structural, optical, thermal, and dielectric properties of polyethylene oxide/carboxymethyl cellulose blend flled with barium titanate. J Phys Chem Solids 125:103–114
- Movagharnezhad N, Moghadam PN (2017) Hexamethylene diamine/carboxymethyl cellulose grafted on magnetic nanoparticles for controlled drug delivery. Polym Bull 74:4645–4658
- Pereira MBB, França DB, Araújo RC, Silva Filho EC, Rigaud B, Fonseca MG, Jaber M (2020) Amino hydroxyapatite/ chitosan hybrids reticulated with glutaraldehyde at diferent pH values and their use for diclofenac removal. Carbohydr Polym 236:116036
- Prateek T, V.K., and Gupta, R.K. (2016) Recent progress on ferroelectric polymer-based nanocomposites for high energy density capacitors: synthesis, dielectric properties, and future aspects. Chem Rev 116:4260–4317
- Sarban R, Jones RW, Mace BR, Rustighi E (2011) A tubular dielectric elastomer actuator: fabrication, characterization and active vibration isolation. Mech Syst Signal Proc 25:2879–2891
- Seddiqi H, Oliaei E, Honarkar H, Jin J, Geonzon LC, Bacabac RG, Klein-Nulend J (2021) Cellulose and its derivatives: towards biomedical applications. Cellulose 28:1893–1931
- Song Y, Wu T, Bao J, Xu M, Yang Q, Zhu L, Shi Z, Hu GH, Xiong C (2022) Porous cellulose composite aerogel flms with super piezoelectric properties for energy harvesting. Carbohydr Polym 288:119407
- Takechi S, Teramoto Y, Nishio Y (2016) Improvement of dielectric properties of cyanoethyl cellulose via esterifcation and flm stretching. Cellulose 23:765–777
- Tao J, Cao SA, Liu W, Deng YL (2019) Facile preparation of high dielectric fexible flms based on titanium dioxide and cellulose nanofbrils. Cellulose 26:6087–6098
- Wang FJ, Wang MH, Shao ZQ (2018) Dispersion of reduced graphene oxide with montmorillonite for enhancing dielectric properties and thermal stability of cyanoethyl cellulose nanocomposites. Cellulose 25:7143–7152
- Wang S, Chen H, Zhou X, Tian Y, Lin C, Wang W, Zhou K, Zhang Y, Lin H (2020) Microplastic abundance, distribution and composition in the mid-west Pacifc Ocean. Environ Pollut 264:114125
- Wu T, Song Y, Shi Z, Liu D, Chen S, Xiong C, Yang Q (2021) High-performance nanogenerators based on fexible cellulose nanofibril/MoS₂ nanosheet composite piezoelectric flms for energy harvesting. Nano Energy 80:105541
- Yang Q, Zhang C, Shi Z, Wang J, Xiong C, Saito T, Isogai A (2018) Luminescent and transparent nanocellulose flms containing europium carboxylate groups as fexible dielectric materials. ACS Appl Nano Mater 1:4972–4979
- Yin YN, Zhang CG, Yu WC, Kang GH, Yang QL, Shi ZQ, Xiong CX (2020) Transparent and fexible cellulose dielectric flms with high breakdown strength and energy density. Energy Storage Mater 26:105–111
- Yin YN, He JC, Zhang CG, Chen JS, Wu JX, Shi ZQ, Xiong CX, Yang QL (2021) Flexible cellulose/alumina (Al_2O_3) nanocomposite films with enhanced energy density and efficiency for dielectric capacitors. Cellulose 28:1541–1553
- Yu H, Duan J, Du W, Xue S, Sun J (2017) China's energy storage industry: develop status, existing problems and countermeasures. Renew Sustain Energy Rev 71:767–784
- Yuan MX, Zhang G, Li B, Chung TCM, Rajagopalan R, Lanagan MT (2020) Thermally stable low-loss polymer dielectrics enabled by attaching cross-linkable antioxidant to polypropylene. ACS Appl Mater Interfaces 12:14154–14164
- Zeng X, Deng L, Yao Y, Sun R, Xu J, Wong C-P (2016) Flexible dielectric papers based on biodegradable cellulose nanofbers and carbon nanotubes for dielectric energy storage. J Mater Chem C 4:6037–6044
- Zhang C, Li PP, Zhang YJ, Lu F, Li WW, Kang HL, Xiang JF, Huang Y, Liu RG (2016) Hierarchical porous structures in cellulose: NMR relaxometry approach. Polymer 98:237–243
- Zhang CG, Yin YN, Yang QL, Shi ZQ, Hu GH, Xiong CX (2019) Flexible cellulose/BaTiO₃ nanocomposites with high energy density for flm dielectric capacitor. ACS Sustain Chem Eng 7:10641–10648
- Zhang X, Zhang Y, Zhou Q, Zhang XL, Guo SY (2020) Symmetrical "sandwich" polybutadiene flm with highfrequency low dielectric constants, ultralow dielectric loss, and high adhesive strength. Ind Eng Chem Res 59:1142–1150
- Zhao D, Huang JC, Zhong Y, Li K, Zhang LN, Cai J (2016) High-strength and high-toughness double-cross-linked cellulose hydrogels: a new strategy using sequential chemical and physical cross-linking. Adv Funct Mater 26:6279–6287
- Zhao YH, Guo QW, Wu S, Meng G, Zhang WM (2019) Design and experimental validation of an annular dielectric elastomer actuator for active vibration isolation. Mech Syst Signal Proc 134:106367
- Zhao DW, Zhu Y, Cheng WK, Chen WS, Wu YQ, Yu HP (2021) Cellulose-based fexible functional materials for emerging intelligent electronics. Adv Mater 33:2000619

Zheng W, Wong SC (2003) Electrical conductivity and dielectric properties of PMMA/expanded graphite composites. Compos Sci Technol 63:225–235

Publisher's Note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

Springer Nature or its licensor holds exclusive rights to this article under a publishing agreement with the author(s) or other rightsholder(s); author self-archiving of the accepted manuscript version of this article is solely governed by the terms of such publishing agreement and applicable law.