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# Physical, biological and chemical characterisation of wood treated with silver nanoparticles

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**Abstract** Nowadays, environmentally friendly processes are of great interest and are considerably needed due to the environmental pollution seems to be a problem worldwide. For this reason, in this study, silver nanoparticles were synthesized using environmentally-friendly methods and their effectiveness as wood preservatives was investigated. Scots pine (*Pinus sylvestris L.*) samples were impregnated with an autoxidized soybean oil polymer containing Ag nanoparticles (Agsbox). Samples characterised by Fourier transform infrared spectroscopy (FTIR) were tested against brown rot (*Coniophora puteana*) and wood-destroying insects (*Hylotrupes bajulus*). In addition, decay tests were applied to mini-block samples leached according to the EN 84 standard.

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Department of Aircraft Airframe Engine Maintenance, Kapadokya University, Ürgüp, Nevşehir, Turkey Results demonstrated that Agsbox increased decay resistance in the unleached samples. However, low efficacy was exhibited against newborn *H. bajulus* larvae. As a results of FTIR measurement, impregnated with the nanocomposites showed significant changes at the 2910 cm<sup>-1</sup> (C–H) and 1712 cm<sup>-1</sup> (C=O) peaks.

**Keywords** Autoxidised soybean oil polymer · Silver nanoparticles · *Coniophora puteana* · *Hylotrupes bajulus* 

# Introduction

Wood used in indoor and outdoor applications is exposed to the effects of temperature, humidity and biological agents. In wood with low natural durability, exposure to fungi and insects under appropriate humidity and temperature conditions leads to colour changes and structural damage of the wood (Meyer-Veltrup et al. 2017; Yang et al. 2017). The service life of wood can be extended by treating it with preservative that can be water-borne or solvent-based. However, nowadays, the first is the one most frequently used for construction applications. The most important water-based wood preservative known is copper chromium arsenic (CCA). Over the past 80 years, it has been used extensively worldwide. However, due to the toxic effect on the environment, its use is prohibited in some countries. Copper-based preservatives like alkaline copper quaternary (ACQ) and copper azole (CA) have been developed and utilized as replacements (Zhong et al. 2014) although these preservatives gradually leach out of the wood, causing significant losses in effectiveness (Liu et al. 2018).

Many biological approaches are used to obtain silver nanoparticles via 'green' methods including aqueous extracts of manna from the hedysarum plant and the soap-root (Acanthe phylum bracteatum) plant, culture supernatants of Aspergil lusterreus and leaf extracts of Euphorbia hirta L., Euphorbia milii and Foeniculum vulgare (Kuppusamy et al. 2016). In addition, basic amino acids such as L-lysine or Larginine are also used to obtain silver nanoparticles (Bonde 2011; de Matos et al. 2011; Elumalai et al. 2010; Forough and Fahadi 2010; Li et al. 2011; Singh et al. 2010). In other studies, silver nanoparticles were synthesized using Polyalthia longifolia leaf extract (Kaviya et al. 2011) and Cleome viscosa plant extract (Lakshmanan et al. 2018). In addition, silver nanoparticles of  $13 \pm 3$  nm were obtained using sulphate polysaccharides isolated from Porphyria vietnamensis (Venkatpurwar and Pokharkar 2011).

In general, nanocompounds penetrate into the wood cell walls more easily, while they are more difficult to leach out and have higher bio-durability. In recent years, several studies have been conducted on the antifungal activity of silver nanoparticle compounds (Can et al. 2018; Kim et al. 2009; Paril et al. 2017). However, the antifungal properties, leaching resistance and chemical characterisation of silver are less understood because most studies have been focused on viral pathogens and bacterial in animals (Klasen 2000; Silver 2003) and silver nanoparticle metals have been the subject of intense interest (Baker et al. 2005; Melaiye et al. 2005; Sondi and Salopek-Sondi 2004). Though the biocidal effects and mode of action of silver ions are known (Lakshmanan et al. 2018; Siddiqi et al. 2018), However, the antifungal activity and the mode of action of nano-Ag against fungi have for the most part not been determined and remain unknown. Although some studies have shown that silver nanoparticles are effective against decay fungi (Kim et al. 2009; Pulit et al. 2013), some studies have demonstrated that silver nanoparticles are less effective against decay fungi (Akhtari and Arefkhani 2013; Rezaei et al. 2011).

These days, metal nanoparticles are often obtained through green processes (Ahmed et al. 2016; Bonde 2011; de Matos et al. 2011; Elumalai et al. 2010; Forough and Fahadi 2010; Hazer and Akyol 2016; Li et al. 2011; Singh et al. 2010). Consequently, the present study examined the effect on wood of silver nanoparticles obtained naturally using soybean oil.

This study aimed to increase the leaching resistance of silver nanoparticles produced by the green chemistry approach. Furthermore, the biological durability and chemical properties of Scots pine (*Pinus sylvestris* L.) wood were investigated.

#### **Experimental section**

## Materials

Scots pine (density 0.42 g cm<sup>-3</sup>) (*Pinus sylvestris* L.) specimens were selected from entirely sapwood boards, cut into sizes of 15 (Radial)  $\times$  25 (Tangential)  $\times$  50 (Longitudinal) mm<sup>3</sup> for the decay and insect tests and 20 (R)  $\times$  20 (T)  $\times$  10 (L) mm<sup>3</sup> for the leaching test. Two different sizes were used for the samples in the fungal tests. The first was the standard test size (15  $\times$  25  $\times$  50 mm<sup>3</sup> along the grain), and the second  $(20 \times 20 \times 10 \text{ mm}^3)$  was that applied for the mini-block samples in the leaching test. The AgNO<sub>3</sub> (purity  $\geq$  99.0%) was obtained from Sigma-Aldrich. Soybean oil used was supplied by Çotanak/Altas and was composed of 45.6% polyunsaturated fatty acids (linoleic acid and linolenic acid), 34 wt% monounsaturated fatty acids (oleic acid) and 15.9 wt% saturated fatty acids (stearic acid and palmitic acid).

#### Synthesis of silver nanoparticles (Agsbox)

The same procedure was applied as for the autoxidation, but without silver nanoparticles, on the fatty acids/oil mixed with AgNO<sub>3</sub>. 18 g of soybean oil was poured into a Petri dish ( $\Phi$  = 14.5 cm, oil thickness: 1.0 mm), after which 0.50 g (2.94 mmol) of AgNO<sub>3</sub> was added into this oil and occasionally mixed using a glass rod. The colourless oil layer exposed to the air turned a deep brown colour after 4 weeks due to the formation of silver nanoparticles. In order to obtain a clear nanocomposite liquid, the unreacted inorganic salts were filtered from the oxidized oil containing the silver nanoparticles (Can et al. 2018; Hazer and Kalaycı 2017).

## Impregnation of wood

Two different solutions, chloroform (CL) and toluene (T), were prepared with the soybean oil-Ag nanocomposites for impregnation of the wood specimens. For this purpose (all w/w) 1.5% Agsbox was dissolved in chloroform (Agsbox-CL), and 1.5% Agsbox was dissolved in toluene (Agsbox-T). Samples to be treated with Agsbox were mixed in the chloroform and toluene using a magnetic stirrer for 2 h, followed by an additional 2 h of stirring after changing the chloroform and toluene. Following the stirring process, the samples were held in the lab for 2 h, and then dried in a vacuum oven at  $40 \pm 2$  °C for 4 h subsequent to drying in an oven at  $103 \pm 2$  °C. The control samples were also held in chloroform for 2 h for comparison with the treated samples. After drying at  $103 \pm 2$  °C for 1 day, the samples were weighed and then exposed to the impregnation: vacuum (650 mmHg for 20 min) and atmospheric pressure (1 h) performed in a desiccator. The samples were then oven-dried for 1 day at 103  $\pm$  2 °C and weighed again and the weight percent gain (WPG1) was calculated using Eq. (1).

$$WPG_1 = 100 \times [(M_t - M_o)/M_o]$$
(1)

where  $M_o$  represents oven-dry weights of the untreated samples and  $M_t$  oven dry weights of the treated samples.

Fourier transform infrared (FT-IR) spectroscopy analysis

The FT-IR spectra, with 32 scans for each sample, were collected via a Shimadzu IRAffinity-1 spectrometer equipped with an ATR pike MIRacle accessory (4 cm<sup>-1</sup> resolution). The samples were measured from the earlywood section with an ATR apparatus and spectrum measurements were taken at eight different points from each sample surface (Ext.) and at a depth of 2 mm (Int.). The resulting single spectrum was obtained for each variation by calculating the average of the spectra in the programme of the device.

#### Colour measurement

Colour measurement was applied with a Konica Minolta spectrophotometer (Osaka, Japan) by measuring the L, a and b values on the samples. For each sample, four colour measurements were made and colour evaluation was carried out according to ISO 7724-2 standard (ISO 7724 1984). The changes in colour coordinates ( $\Delta$ L\*,  $\Delta$ a\* and  $\Delta$ b\*) were determined by the difference between the initial and the final values. Hereafter, the total colour changes ( $\Delta$ E\*) were calculated according to the following equations:

Total colour change ( $\Delta E^*$ ) was calculated according to Eq. (2) below.

$$(\Delta E*) = \left[ (\Delta a*)^2 + (\Delta b*)^2 + (\Delta L*)^2 \right] \frac{1}{2}$$
(2)

# Leaching test

The leaching procedure was similar to the EN 84 (1997) with wood size modifications. Wood specimens with dimensions of  $20 \times 20 \times 10 \text{ mm}^3$  were placed into a 300-mL beaker filled with deionized water, and subjected to a vacuum (650 mmHg) to impregnate the blocks with the leaching solution. A set of six replicates was used for each treatment. At the end of each period, the leaching water was collected and its silver content was determined by ICP analysis. The silver content of the leached water was measured via inductively coupled plasma (ICP) analysis, using an ICP spectrometer (ICP-AES, Spectro Genesis), according to the AWPA standard (AWPA A7 1993). After leaching and conditioning, WPG<sub>2</sub> was then calculated using Eq. (3).

$$WPG_2 = 100 \times [(M_1 - M_o)/M_o]$$
(3)

where  $M_1$  is the oven dry weight of the leached wood blocks.

## Decay test

The fungal test was conducted with the decay fungus *Coniophora puteana* (Schumacher ex Fries) Karsten strain BAM Ebw according to EN 113 (2006). In a glass container, where the fungal strain had been previously grown on a 4–2.5% malt-agar medium, one treated and one untreated sample were incubated at a constant condition of  $20 \pm 2$  °C and  $70 \pm 5\%$  R.H

for 16 weeks. Six replicates were performed for each treatment.

# Insect test

The insect test was performed based on the EN 47 (2005) standard to determine the effectiveness of the Agsbox against the larvae of *Hylotrupes bajulus*. Prior to the test, 24 sample with dimensions of  $15 \times 25 \times 50 \text{ mm}^3$  along the grain were conditioned at  $20 \pm 2 \text{ °C}$  and  $65 \pm 5\%$  RH. After exposure to the larvae, the test specimens were placed in jars and stored in a controlled chamber at  $20 \pm 2 \text{ °C}$  and  $70 \pm 5\%$  RH for 8 weeks. After the exposure, each sample was examined by X-ray analysis to check the status of the inserted larvae (dead, living, not recovered).

# Statistical analysis

A descriptive analysis was developed (mean and standard deviation) for decay test. ANOVA was applied to verify the effect of treatment with the silver nanoparticles. Duncan's test was set at 99% confidence level to determine the statistical difference between the means.

# **Results and discussion**

As seen in previous publications, transmission electron microscopy (TEM) images of Agsbox showed a 'bunch of grapes' cluster of Ag nanoparticles approximately 5 nm in size (Hazer and Kalaycı 2017).

Fourier transform infrared spectroscopy (FTIR-ATR)

The assignments of characteristic absorption IR bands of the wood samples are given in Table 1 and the FTIR spectra of the treated and untreated samples are shown in Fig. 1.

The spectrum of pine wood (Fig. 1) shows the same basic structure as wood samples.  $3300-4000 \text{ cm}^{-1}$  peaks shows strong main OH stretching (1),  $2800-3000 \text{ cm}^{-1}$  shows C–H stretching in methyl and methylene groups (2–3). Two new characteristic peaks appeared in the samples treated with the Agsbox at 2910 cm<sup>-1</sup> (2) and 2840 cm<sup>-1</sup> (3). These peaks

were more prominent for the surface areas of the samples. 2919 cm<sup>-1</sup> and 2849 cm<sup>-1</sup> peaks intensity increased after impregnation in the samples treated with the Agsbox. These peak values were higher with the Agsbox dissolved in chloroform than for that dissolved in toluene. Agsbox-CL-Ext. and Agsbox-CL-Int. gave similar peaks. This showed that the Agsbox-CL material exhibited behaviour similar to that of the wood surface even at depths of 2 mm. However, the same was not the case for Agsbox-T because lower peak values were obtained at a depth of 2 mm for the samples impregnated with Agsbox-T.

No distinctive peak was observed in the range between 2700 and 1800  $\text{cm}^{-1}$ . The region from 1800 to 700  $\text{cm}^{-1}$  included the sharp and discrete functional groups characterised for wood components (cellulose, hemicellulose, and lignin) (Liu et al. 2018). The modification process made with Agsbox caused significant changes in this region. The band at  $1712 \text{ cm}^{-1}$ (4) indicating an unconjugated C=O stretch in the xylan. This peak increased sharply. Modification of acetyl and carbonyl groups is the main reason for the increase in peak values. Agsbox-Ext. showed a more salient band than the other groups at  $1712 \text{ cm}^{-1}$ . The peak at 1224  $\text{cm}^{-1}(8)$  showed the CO and OH groups in the hemicellulose, cellulose and lignin (Faix 1991). The CO and OH peaks of the silver-impregnated Agsbox-CL specimens increased compared to the control samples, whereas there were no significant changes for the Agsbox-T specimens.

The IR bands at  $1016 \text{ cm}^{-1}$  (11) on behalf of cellulose and hemicellulose changed after Agsbox impregnation.  $1016 \text{ cm}^{-1}$  peak was increased after impregnation with Agsbox-CL, while no change was seen in samples impregnated with Agsbox-T.

Weight percent gain (WPG %) and leachability of nanoparticles

The WPG values for the impregnated specimens are given in Table 2. In addition, Duncan's homogeneity groups are indicated with letters (p < 0.05) for the WPG results. In the homogeneity groups, each column is evaluated within itself.

Higher WPG values were obtained in the samples impregnated with Agsbox-CL than in those impregnated with Agsbox-T in both the mini-block and the standard test samples. Both the nano-Ag and the soybean oil in the material used caused the high WPG

	Wavenumber $(cm^{-1})$	Peak assignment	Structural polymer
1	3336	O-H stretching of bonded hydroxyl groups	-
2–3	2910-2840	C-H stretching absorption band	_
4	1712	C=O stretching	Acetyl group in wood, soybean oil
5	1506	Aromatic skeletal vibrations in lignin	Lignin
6	1415	C-H in plane deformations + C-H deformation in lignin and carbohydrates	Lignin
7	1320	C-H vibration in cellulose	Cellulose
8	1224	CO and OH stretching	Hemicellulose and Lignin
9	1157	C–O–C vibration	
10	1045	C–O Stretch	Cellulose and hemicellulose
11	1016	C–O Stretch	Cellulose and hemicellulose
12	891	C–H deformation	Cellulose
13	802	-	Glucomannan

Table 1 IR spectrum bands (Pandey 2005; Yilgor et al. 2013; Can and Sivrikaya 2017)





Table 2	Weight percent	gain (WPC	%) for	mini-block	and	standard	test	samples
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	Mini-block test			Standard te	
	WPG <sub>1</sub> (%)	WPG <sub>2</sub> (%)	Amount of leached silver (%)	WPG <sub>1</sub> (%)	
Agsbox-CL	4.99a <sup>a</sup> (0.26)	- 0.46a (0.50)	0.0023	4.17a (0.11)	
Agsbox-T	3.45b (0.10)	- 0.72a (0.11)	0.0025	3.64b (0.15)	

 $WPG_1$  weight percent gain after impregnation,  $WPG_2$  weight percent gain after leaching; (in parentheses): SD

<sup>a</sup>The letters indicate Duncan's homogeneity groups in the column

values, which resulted from the anatomical structure and porosity of the Scots pine wood species. The high WPG values caused colour changes on the wood surface of the specimens, which attained a dark-brown colour when treated with the silver nanoparticles (Fig. 2 and Table 3).

Table 3 shows the colour changes ( $\Delta L^*$ ,  $\Delta a^*$ ,  $\Delta b^*$ ,  $\Delta E^*$ ) of wood after treatment. Negative lightness stability ( $\Delta L^*$ ) indicate a tendency of wood surface to become darker. The original colour of wood samples was noticeably changed by Agsbox impregnation. Positive values show the tendency of a wood surface to become reddish for  $\Delta a^*$ , yellowish for  $\Delta b^*$  and negative values represent greenish for  $\Delta a^*$ , bluish for  $\Delta b^*$  (Sivrikaya and Can 2016). Agsbox treatment caused the surface to become reddish. Agsbox caused a yellowing of the surface, as evinced by increased  $\Delta b^*$  values. The maximum total colour change ( $\Delta E^*$ ) was obtained in the samples impregnated with Agsbox. Chloroform and toluene are capable of breaking down the polymeric components such as tannin, lignin, and cellulose present in tree bark, sawdust and pinecones. Fragmentation of tannin and lignin causes colour change on wood surfaces. The colourless oil layer exposed to the air turned a deep brown colour in 3-4 weeks due to the formation of the silver nanoparticles. An intensive bright-green colour was observed for the nanocomposite samples under UV illumination, while the samples appeared brown in daylight. In previous studies, it has been reported that silver had a dark brown appearance (Hazer and Kalaycı 2017; Paril et al. 2017; Lakshmanan et al. 2018).

Negative WPG values were obtained in the leached mini-block test samples. This indicated that silver and

also some wood components had been leached from the wood (Fig. 3). According to the results, 0.0023%of the silver was leached out when chloroform was used and 0.0025% when toluene was used. After the leaching test, most of the -0.46% and -0.72%WPG<sub>2</sub> values obtained were from other substances which were leached from the wood.

After the leaching test, the final amount of Ag released was 0.156 ppb for Agsbox-CL and 0.182 ppb for Agsbox-T, which represented approximately 1.8% and 2.2% of the silver content, respectively, that was in the unleached samples. This finding was satisfactory because silver compounds such as silver nitrate have a water-soluble structure.

The highest leaching rate was realized after the initial leaching period. The amount of leached substances gradually decreased after 48 h. A similar leaching progression was presented by Paril et al. (2017) for silver- and copper-based preservatives. The accessibility of the silver deposited at a higher concentration on the wood surface layer was considered as the main reason for this. The results indicated that the silver nanoparticles were more sensitive to leaching, but many factors, including size, type of stabilizers or species treated, could have influenced the final behaviour.

There is a global movement to ban silver nanoparticles in antimicrobials. Several health maintenance organizations have already banned the use silver nanoparticle antibiotics in many of the products they use. However, the leaching of the nano silver was very low due to our method to obtain nano silver. According to the leaching results, only 0.0023–0.0025% nano silver was leached. It is necessary to conduct further research on toxicity of low amounts of nano-silver



Fig. 2 Colour of treated and untreated specimens (*Control* without impregnation, *CL* impregnated with chloroform, *T* impregnated with toluene)

	ΔL	Δа	$\Delta b$	ΔΕ	
CL	$0.12a^{a}(0.10)$	- 0.89a (0.05)	0.20a (0.02)	0.91a (0.19)	
Т	1.10b (0.24)	- 0.85a (0.12)	2.10b (0.32)	2.52b (0.32)	
Agsbox-CL	- 11.11c (2.58)	4.56b (1.32)	11.78c (2.35)	16.82c (3.25)	
Agsbox-T	- 10.21c (1.59)	3.12b (1.10)	10.12c (2.56)	14.71c (2.69)	

 Table 3
 Colour changes for treated Scots pine

In parentheses: SD

<sup>a</sup>The letters indicate Duncan's homogeneity groups in the column



Fig. 3 Silver leaching (ppb) from Scots pine samples treated with Agsbox-CL and Agsbox-T

(Faunce and Watal 2010; Pulit-Prociak and Banach 2016).

#### Decay test

Figure 4 shows the weight loss values in the samples after the fungal test. In the Duncan test results, the unleached samples were evaluated among themselves, and the leached samples were evaluated among themselves. When the unleached samples were



Fig. 4 Mass loss of unleached and leached silver nanoparticletreated Scots pine after exposure to *C. puteana* (n = 6)

evaluated among themselves, it was seen that all test samples were in the same homogeneity group. In the leached samples, all other variations except Agsbox-T were in the same homogeneity group.

According to Fig. 4, at the end of the 8 weeks decay test, unleached control specimens suffered mass loss at the rate of 37.39%. The decay test standard (EN 113 2006) accounted for the mass loss of 20% in the untreated samples. At the end of 8 weeks, the test was terminated because a weight loss of more than 20% had occurred. Compared to the high weight losses in the unleached control samples, the weight loss was improved in the modified samples at the rate of 50.54% for Agsbox-CL. The lowest weight loss obtained in the leached Agsbox-T samples was also seen in the unleached samples. The control leached specimens exhibited mass loss at the rate of 32.30%. Lower weight losses were obtained in the test samples compared to the controls. However, these weight loss values were above the value of 3% specified by the EN 113 standard.

The results indicated that the decay resistance of wood modified by Agsbox was improved against the brown-rot fungus (*C. puteana*) when compared to that of untreated Scots pine. Many studies have investigated the efficacy of silver material in the wood protection field. In these studies, it has been emphasized that nano silver material is effective against white-rot fungi, but has less effect on brown-rot fungi (Moya et al. 2014, 2017; Bak and Nemeth 2018).

Vegetable oils, due to their chemical formulations, have no fungicidal and/or insecticidal effects on the structure. For this reason, the introduction of a fungicide and insecticide into the wood structure would increase the effectiveness of the oil against fungi and insects. In our study, the fungal resistance in the wood samples was enhanced by the effect of the extensive use of the silver nanoparticles in the soybean oil. In the study by Tomak et al. (2011), by adding boron compounds to oils obtained from vegetable sources (hazelnut, sunflower, canola, soybean, etc.), weight loss in the fungal decay test was 7-13%after the impregnation process. These weight losses were reported to be due to the fact that the oils obtained from plant materials were not effective against fungi, and only prevented water from entering the cells so that the fungus could not get the moisture it required.

## Insect test

Figure 5 shows the mortality rates after the insect test. The 4-week and 2-month results were compared and the Duncan test results were calculated from the final (2-month) results only.

The larvae mortality is presented in Fig. 5. According to the figure, 4 weeks after the first inclusion of the larvae into the treated wood, the chloroform formulations had the same probability of causing the death of H. bajulus larvae as the treatment with toluene formulations. In the first 4 weeks, the mortality rates with the silver-based variations were higher than that of the control. At the end of the 8-week test, high-level rates of larval death had occurred only in the samples impregnated with chloroform and toluene. Mortality rates obtained with Agsbox-CL and Agsbox-T were higher than in the control, while they were lower than the mortality rates when chloroform and toluene were used alone. The study was terminated after 8 weeks. For this reason, it can be said that mortality rates were low in the samples containing nano silver. In other words, Agsbox exhibited low activity against the H. *bajulus* beetle in the 8-week period.

It can be said that the soybean oil used in the production of Agsbox was a nutritive additive for the larvae. In order to survive, insects require proteins, carbohydrates, fat and vitamins, although there are variations in the specific requirements of different insect species (Nair 2007). The statistical analysis revealed that chloroform and toluene were in the same homogeneity class while there were significant differences from the control (p < 0.05).

# Conclusion

This study was undertaken to characterize wood impregnated with Agsbox and to determine the performance of it against *C. puteana* fungus and *H. bajulus* larvae. Further, it was investigated the leaching of Ag nanoparticles from wood after the impregnation. By the FTIR spectra a significant increase in the asymmetric and the symmetric stretching vibrations for C–H after the Agsbox impregnation was observed.

Regarding the insect test, the soybean oil could have provided a nutritive effect and therefore, the Agsbox was not significantly effective against the *H. bajulus* larvae. In the decay tests, no significant differences was found on non leached samples treated with Agbox and two solvents, the only significant difference was found in the control as we can expected. By the decay results seemed that even the solvents alone can gives some effectiveness against fungi, probably due to extraction of some component into the wood. Further investigation on the effect of solvents such as toluene and chloroform on wood resistance against fungus *C. puteana* could be performed in the future.



Fig. 5 Insect test results against *H. bajulus* larvae

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